

NE 533

Nuclear Fuel Performance

Exam 1

$$1) \frac{\partial}{\partial x} \left(k \frac{\partial T}{\partial x} \right) + Q = 0$$

$$T'(x_0) = 0 \quad x_0 = X$$

$$x_0 = 0$$

$$T(x_1) = T_1$$

$$\int \frac{\partial}{\partial x} \left(k \frac{\partial T}{\partial x} \right) dx = \int -Q dx$$

$$k \frac{\partial T}{\partial x} = -Qx + C_1 \quad \rightarrow \quad k(0) = -Q(0) + C_1$$

$$C_1 = 0$$

$$\int k \frac{\partial T}{\partial x} = -Qx \quad dx$$

$$T(x) = -\frac{Q}{2k} x^2 + C_2$$

$$T(x_1) = -\frac{Q}{2k} X^2 + C_2 = T_1$$

$$C_2 = T_1 + \frac{Q}{2k} X^2$$

$$T(x) = -\frac{Q}{2k} x^2 + T_1 + \frac{Q}{2k} X^2$$

$$T(x) - T_1 = \frac{Q}{2k} (X^2 - x^2)$$

- Steady State

- 1-D

- k is constant

2) T_o w/ + w/o coating

$$R_F = 0.6 \text{ cm}$$

$$K_F = 0.05 \frac{\text{W}}{\text{cm} \cdot \text{K}}$$

$$t_j = 0.005 \text{ cm}$$

$$K_j = 0.004 \frac{\text{W}}{\text{cm} \cdot \text{K}}$$

$$t_{\text{clad}} = 0.05 \text{ cm}$$

$$K_{\text{clad}} = 0.15 \frac{\text{W}}{\text{cm} \cdot \text{K}}$$

$$t_{\text{coat}} = 0.01 \text{ cm}$$

$$K_{\text{coat}} = 0.015 \frac{\text{W}}{\text{cm} \cdot \text{K}}$$

$$T_{\text{coat}} = 600 \text{ K}$$

$$Q = 250 \frac{\text{W}}{\text{cm}^2}$$



$$LHR = \pi R_F^2 Q = \pi (0.6)^2 (250) = 283 \frac{\text{W}}{\text{cm}}$$

$$\frac{LHR}{2\pi R_F} = \frac{283}{2\pi (0.6)} = 75 \frac{\text{W}}{\text{cm}^2}$$

$$T_{C0} = T_{\text{clad}} + \frac{LHR}{2\pi R_F} \frac{t_{\text{coat}}}{K_{\text{coat}}} = 600 + 75 \left(\frac{0.01}{0.015} \right) = 650 \text{ K}$$

$$T_{C1} = T_{C0} + \frac{LHR}{2\pi R_F} \frac{t_{\text{clad}}}{K_{\text{clad}}} = 650 + 75 \left(\frac{0.05}{0.15} \right) = 675 \text{ K}$$

$$T_F = T_{C1} + \frac{LHR}{2\pi R_F} \frac{t_j}{K_j} = 675 + 75 \left(\frac{0.005}{0.004} \right) = 769 \text{ K}$$

$$T_o = T_F + \frac{LHR}{4\pi K_F} = 769 + \frac{283}{4\pi (0.05)} = \underline{\underline{1169 \text{ K}}}$$

$$\Delta T \text{ over coating} = 50 \text{ K}$$

$$T_o \text{ w/o coating} = \underline{\underline{1169 \text{ K}}}$$

$$3) \text{ UN} \quad x_e = 19.5\% \quad \rho = 12.3 \text{ g/cc} \quad \sigma_f = 570 \text{ b}$$

$$Q = E_f \phi \sigma_f N_f$$

$$A(u) = 235 \times 0.195 + 238 \times 0.805 = 237.4$$

$$N_f ?$$

$$A(\text{UN}) = 237.4 + 14 = 251.4$$

$$N_f \Rightarrow 12.3 \text{ g/cc} \frac{1 \text{ mol}}{251.4 \text{ g}} \frac{6.022 \times 10^{23} \text{ mol}^{-1}}{1 \text{ mol}} \frac{14}{1 \text{ UN}} \times 0.195$$

$$N_f = 5.75 \times 10^{21} \text{ U}^{235} \text{ / cc}$$

$$Q = (200 \times 10^6) (1.602 \times 10^{-19} \text{ J/eV}) (5 \times 10^{12} \text{ cm}^{-2} \text{ s}^{-1}) (570 \times 10^{-24} \text{ cm}^2) (5.75 \times 10^{21})$$

$$Q = 525 \text{ W/cc}$$

$$b) \quad x_e \text{ for } \text{UO}_2 ?$$

$$N_f(\text{UN}) = N_f(\text{UO}_2)$$

$$N_f = 5.75 \times 10^{21} = 10.97 \text{ g/cc} \frac{1 \text{ mol}}{A(\text{UO}_2)} \frac{6.022 \times 10^{23}}{1 \text{ mol}} \frac{14}{1 \text{ UO}_2} x_e$$

$$\frac{x_e}{A(\text{UO}_2)} = 8.7 \times 10^{-4}$$

$$A(\text{UO}_2) = 235 x_e + 238(1-x_e) + 32$$

$$x_e = (8.7 \times 10^{-4}) (270 - 3 x_e)$$

$$1.0026 x_e = 0.2349$$

$$x_e = 0.243$$

$$\Rightarrow \underline{\underline{24.3\%}}$$

$$4) \quad L = 3.5 \text{ m} \quad z_0 = 1.75 \text{ m} \quad LHR^0 = 350 \frac{\text{W}}{\text{cm}^2} \quad \gamma = 1.3$$

$$a) \quad LHR(z = 1.4 \text{ m})$$

$$\frac{\pi}{2\gamma} = 1.2$$

$$LHR = LHR^0 \cos \left[\frac{\pi}{2\gamma} \left(\frac{z}{z_0} - 1 \right) \right]$$

$$LHR = 350 \cos \left(1.2 \left(\frac{1.4}{1.75} - 1 \right) \right) = \underline{\underline{340 \frac{\text{W}}{\text{cm}^2}}}$$

$$b) \quad \Delta T ?$$

$$\Delta T = \frac{2\gamma}{\pi} \frac{z_0 LHR^0}{\dot{m} c_p} \left[\sin \frac{\pi}{2\gamma} + \sin \left(\frac{\pi}{2\gamma} \left(\frac{z}{z_0} - 1 \right) \right) \right]$$

$$\Delta T = \frac{1}{1.2} \frac{(1.75)(350)}{(0.22)(4200)} \left[\sin(1.2) + \sin(1.2(2-1)) \right]$$

$$\underline{\underline{\Delta T = 103 \text{ K}}}$$

5) Backward Euler $\gamma_{n+1} = \gamma_n + dt \gamma'_{n+1}$

$\gamma' = t \exp(-2t)$ $t_0 = 0$ $\gamma_0 = 4$ $dt = 0.5$

$t_1 = 0.5$ $\gamma_1 = \gamma_0 + dt \gamma'_1 = 4 + 0.5(0.5 \exp(-2 \times 0.5))$

$\gamma_1 = 4.09$

$t_2 = 1$ $\gamma_2 = \gamma_1 + dt \gamma'_2 = 4.09 + 0.5(1 \exp(-2 \times 1))$

$\gamma_2 = 4.16$

$t_3 = 1.5$ $\gamma_3 = \gamma_2 + dt \gamma'_3 = 4.16 + 0.5(1.5 \exp(-2 \times 1.5))$

$\gamma_3 = 4.20$

b) Fertile \rightarrow can be converted into a fissile atom via neutron capture

Fissile \rightarrow can undergo fission w/ a neutron of any energy

Fissionable \rightarrow can undergo fission w/ a high energy neutron

- 7) Low melting point
Anisotropic thermal expansion
Anisotropic irradiation growth
Poor oxidation behavior
Extreme swelling

- 8) Linear density is the relative volume of fuel compared to the maximum possible volume inside the cladding
Necessary because of fuel swelling and thermal expansion

- 9) We enrich U to increase ρ_f , number density of U^{235}
Results in higher heat generation rate.
 UF_6 is the compound used for enrichment

Centrifuge

- spinning cylinder w/ a feed of UF_6 gas
- centrifugal forces preferentially push U^{238} to outside of cylinder
- enriched product removed from center of cylinder

- 10) Finite difference, finite volume, finite element
- FD is only known at discrete points, typically restricted to rectangular geometries
 - other answers are acceptable

11) DNBR

- ratio of the heat flux that causes dryout to the actual heat flux
- minimum DNBR is taken in the hottest channel
- greater than 1.15

CHT

- the heat flux at which dryout occurs, where the rod is coated in steam
- beyond CHT cladding has large temperature increase

12) Fuel Kernel, buffer, IPyC, SiC, OPyC

- high temperature gas reactor