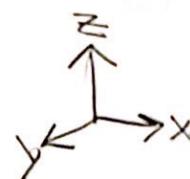


ii.

$$\textcircled{1} \quad \frac{\partial T}{\partial x}(x_0) = 0, \quad x_0 = 0$$

$$\textcircled{2} \quad x_1 = x, \quad T(x_1) = T_1$$



CARTESIAN COORDINATES

$$\frac{\partial^2 T}{\partial x^2} (k \frac{\partial T}{\partial x}) + Q = 0$$

Assume:

- * k = Constant i.e. $k \neq$ func of x
- * Temp does not vary along y, z (i.e. symmetric)
- * Steady State $\rightarrow \frac{\partial^2 T}{\partial x^2} = 0$
- * Q is invariant with x (i.e. constant)
- * fuel is rectangular or box shaped

$$\frac{\partial^2 T}{\partial x^2} (k \frac{\partial T}{\partial x}) = -Q$$

$$\int \frac{\partial^2 T}{\partial x^2} (k \frac{\partial T}{\partial x}) = - \int Q dx$$

$$k \frac{\partial T}{\partial x} = -Qx + C_1$$

$$\text{using BC } \#1 \rightarrow C_1 = 0$$

$$\int k \frac{\partial T}{\partial x} = \int [-Qx] dx$$

$$kT(x) = -\frac{Qx^2}{2} + C_2$$

$$\text{use BC } \#2 - T(x_1) = T_1$$

$$kT_1 = -\frac{Qx_1^2}{2} + C_2$$

$$C_2 = kT_2 + \frac{Qx_1^2}{2}$$

$$kT(x) = -\frac{Qx^2}{2} + kT_1 + \frac{Qx_1^2}{2}$$

$$k(T(x) - T_1) = \frac{Q}{2} [x_1^2 - x^2]$$

$$T(x) = \frac{Q}{2k} [x_1^2 - x^2] + T_1$$

② Calc centerline T & $T(r=0.4\text{ cm})$

$$k_{coat} = 0.05 \frac{\text{W}}{\text{cmK}}, k_{clad} = 0.15 \frac{\text{W}}{\text{cmK}}, k_{gap} = 0.25 \frac{\text{W}}{\text{cmK}}$$

$$k_{fuel} = 0.005 \frac{\text{W}}{\text{cmK}}$$

$$h_{cool} = 5.5 \frac{\text{W}}{\text{cm}^2 \cdot \text{K}}$$

$$1\text{ cm} = 0.01\text{ m}$$

$$T_{oo} = 800\text{ K}$$

$$Q = 400 \frac{\text{W}}{\text{cm}^3}$$

$$T_{coat} = T(r_4) = \frac{Q}{2h_{cool}} R_{fuel} + T_{oo}$$

$$T_4 = \frac{400 \frac{\text{W}}{\text{cm}^3} (0.6)}{(2 \times 5.5 \frac{\text{W}}{\text{cm}^2 \cdot \text{K}})} + 800\text{ K}$$

$$\underline{T_4 = 821.82\text{ K}}$$

$$T_3 = \frac{Q t_{coat} R_f}{2 k_{coat}} = \frac{(400)(0.101)(0.6)}{2 \times 0.05} + T_4$$

$$\boxed{T_3 = 845.82}$$

$$T(r_2) = T_2 = \frac{Q t_{clad} R_f}{2 k_{clad}} + T_3$$

$$= 847.32\text{ K}$$

$$T_1 = T(r_i) = \frac{Q t_{qip} R_f}{2 k_{qip}} + T_2$$

$$T_1 = 1007.32 \text{ K}$$

$$T_0 = \frac{Q}{4 k_{fuel}} R_{fuel}^2 + T_1$$

$$T_0 = 8207.32 \text{ K}$$

at $r = 0.4 \text{ cm}$

$$T(0.4 \text{ cm}) = \frac{Q (0.6^2 - 0.4^2)}{4 \times k_{fuel}} + T_1$$
$$= 5007.32 \text{ K}$$

Calculating heat generation rate for a given fuel

~ about 200 MeV of energy is available due to a fission (210 MeV)

~ section of the target nuclide (tabulated)

$\cdot (\dot{\phi})$, Temp, fuel, etc.

$\rightarrow \sigma$ is the cross section

heat system

$$0.145 \text{ W/cm K}$$



$$\textcircled{3} \quad k_{\text{fuel}} = 145 \text{ W/mK} \quad E = 0.195$$

$$\rho_{\text{fuel}} = 15.67 \text{ g/cm}^3$$

$$\Omega_f = 5.7 \times 10^{-22} \text{ cm}^2$$

$$\text{a) } \dot{\phi} = 2 \times 10^{12} \text{ n/cm}^2 \cdot \text{s}$$

$$Q = E_f N_f \frac{\Omega_f}{\rho_{\text{fuel}}} \frac{\dot{\phi}}{N_A} E$$

$$N_f = \frac{235}{235E + 238(1-E)} + 1.2 \times 10^{-16}$$

$$= \frac{15.67 \times 6.023 \times 10^{23} \times 0.195}{269.77}$$

$$= 6.82 \times 10^{21} \frac{U^{235} \text{ atoms}}{\text{cm}^3}$$

$$E_f = \frac{200 \times 10^6 \text{ eV}}{\text{fission}} \times 1.602 \times 10^{-19} \text{ J/eV}$$

$$= 3.218 \times 10^{-11} \text{ J}$$

$$\boxed{Q = 250.6 \text{ W/cm}^3}$$

b) Using excel to change enrichment

$$\rho = 10.979 \text{ g/cm}^3$$

$$\boxed{20.2784 \text{ or } 27.84\%}$$

Calculating heat generation rate for a given fuel
 ... of energy is available due to a fission (210 MeV
 insulated)

q) $LHR(z = 1.8 \text{ m})$

$$LHR^o = 150 \text{ W/cm}^2 K = 1.5 \times 10^4 \text{ W/m}^2 K$$

$$z_0 = \frac{L}{2} = 1.5 \text{ m}$$

$$LHR^o = LHR(1.5 \text{ m})$$

$$LHR(1.8/1.5) = (1.5 \times 10^4 \text{ W/m}^2 K) \left(\cos\left(\frac{\pi}{2.2} \left(\frac{1.8}{1.5} - 1 \right)\right) \right)$$

$$\boxed{LHR(1.8 \text{ m}) = 1.44 \times 10^4 \text{ W/m}^2 K}$$

assuming
radians
↓

b) $\Delta T = \frac{1}{1.43} \frac{z_0 LHR^o}{m CPW} (\sin(1.43) + \sin(1.43))$

$$\Delta T_i = \frac{17.3}{16.86} \times \left\{ \begin{array}{l} " \sin(1.43) \\ " \end{array} \right\}$$

$$\Delta T_{ii} = \left\{ \begin{array}{l} 93.39 \\ = 92.47 \end{array} \right\} "$$

$$\boxed{\Delta T_{ii} > \Delta T_i}$$

$$⑤ \quad y(t_0) = 6 = y(1) \quad t_n = 2$$

$$y'(t) = 4t - 3t^2 \quad dt = 0.33$$

Fwd Euler

$$y(t_1) = y(t_0) + \Delta t \cdot y'(t_0)$$

$$t_0 = 1$$

$$t_1 = t_0 + dt = 1.33$$

$$t_2 = t_1 + dt = 1.66$$

$$t_3 = t_2 + dt = 1.99 \approx 2$$

$$y(t_0) = 6$$

$$y(t_1) = 6.33$$

$$y(t_2) = 6.3343$$

$$y(t_3) = 5.798$$

$$y(t_1) = 6 + 0.33 \times (4(1) - 3(1)^2)$$

$$y(t_2) = y(t_1) + 0.33 \times (4 \times (1.33) - 3(1.33)^2)$$

$$y(t_3) = y(t_2) + 0.33 (4 \times 1.66 - 3(1.66)^2)$$

$$y(t_4) = y(t_3) + 0.33 (4 \times 1.99 - 3 \times 1.99^2)$$

Backward Euler

$$y(t_1) = y(t_0) + dt * y'(t_1)$$

or

$$y_{n+1} = y_n + dt * y'_n$$

$$y_1 = 6 + 0.33 \times (4 \times 1.33 - 3 \times (1.33)^2)$$

Backward

$$y_0 = 6$$

$$y_1 = 6.0043$$

$$y_2 = 5.467$$

$$y_3 = 4.174$$

⑥

Fissionable - can undergo fission @ high or low neutron energy

Fertile can become fissile via transmutation

fissile - able to undergo fission w/ low energy neutron

⑦

① many phases α, γ, β

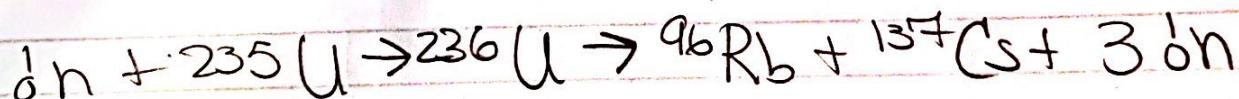
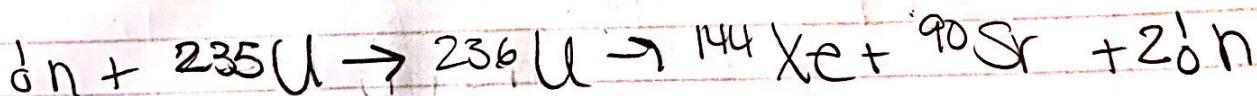
② α -U shrinks in 1-D & expands along other 2-directions (thermally unstable)

⑧

Smear density is ratio of fuel mass to total volume w/in fuel rod. It is used to compare the relative swelling of fuels

⑩

^{96}Rb & ^{144}Xe



⑪

3 space discretization methods

Finite element

Finite difference

Finite volume

all 3 used in state of art codes

Finite difference was the 1st method developed

Finite volume is very useful when flux is approx. equal @ interfaces

Finite volume is most flexible & useful in complex geometries but very intricate so not used when not needed

- ⑨ We enrich U to increase fissile atoms in fuel. UF_6 is used in centrifuge process. The difference in isotopic mass allows separation by centrifugal force i.e. ^{238}U gets pushed to outside of centrifuge container & allows ^{235}U to be drawn out from center.