

- ① Given: Zr10 clad $T = 625K$
 $t = 400 \text{ days}$
 $t_{\text{wall}} = 500 \mu\text{M}$

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Find a) t_{ox} @ $t = 400 \text{ days}$

12/16 $t^*(d) = 6.62 \times 10^{-7} \exp\left(\frac{11949}{T}\right) = 133.01 \text{ days}$ ✓

$t > t^*$ thus, 550

$$\delta^* = 5.1 \exp\left(\frac{-500}{T}\right) = 2.29 \mu\text{M}$$

$$\delta = \delta^* + k_L (t - t^*)$$
 ✓

$$\hookrightarrow k_L = 7.48 \times 10^{-6} \exp\left(\frac{-12500}{T}\right) = 0.0154 \frac{\mu\text{M}}{\text{day}}$$

$$= 2.29 + 0.0154 (400 - 133)$$

$$\boxed{= 6.4018 \mu\text{M}}$$
 ✓

b) $f_{\text{HP}} = 0.18$

$t = 1 \text{ year}$

$\text{PBR} = 1.58$

$\rho_{\text{Zr}} = 6.59 \text{ g/cm}^3$

$\rho_{\text{ZrO}_2} = 5.68 \text{ g/cm}^3$

$$C_{\text{H}}^{\text{wt, PPM}} = \frac{2 f_{\text{HP}} \cdot \delta_{\text{oxL}} \cdot P_{\text{ox}} \cdot f_{\text{ZrO}_2} \cdot \frac{1}{16} \cdot 10^6}{\left(t_{\text{HP}} - \frac{\delta_{\text{oxL}}}{\text{PBR}}\right) \rho_{\text{Zr}}}$$

$$f_{\text{ZrO}_2} = \frac{16 \cdot 2}{16 \cdot 2 + 91} = 0.26$$

$$\boxed{= 66.034, \text{ wt. PPM H}}$$

② Given: $\alpha_{th} = 11 \cdot 10^{-6}$
 $\dot{r} = 3.5 \times 10^{13} \text{ f/s-cm}^3 \rightarrow \beta_{E,MA} = \frac{(3.5 \times 10^{13})(7344000 \text{ s})}{2.45 \times 10^{22} \text{ u/cc}}$
 $T_{fuel} = 1200 \text{ K}$
 $T_{ref} = 300 \text{ K}$
 $\Delta P_0 = 0.01$
 $B_0 = 5 \text{ MWD/kgU} / 950 = 0.0053 \text{ FIMA}$
 $\rho_{uor} = 10.97 \text{ g/cc}$
 $t = 85 \text{ days} = 7344000 \text{ s}$

Find $E_{tot} : ?$

$$E_{th} = \alpha \Delta T = (11 \times 10^{-6})(1200 - 300) = 0.0099$$

$$E_D = \Delta P_0 \left(\exp \left(\frac{\beta_{E,MA} \cdot \ln(0.01)}{\beta_0 B_0} \right) \right) = -0.0099$$

$$E_{SFP} = 5.57 \times 10^{-2} \rho_{uor} \beta_{F,MA} = 0.0064$$

$$E_{GFP} = (1.96 \times 10^{-28}) (\rho_{uor} \beta_{F,MA}) (T_F - T_{ref}) \exp(-0.0162(-800 - T_F)) \exp(-17.8 \rho_{uor} \beta_{F,MA})$$

- hard to tell what went wrong, but too large

$$= 0.00380 \rightarrow 2.59 \times 10^{-5}$$

$$E_{tot} = E_{th} + E_D + E_{SFP} + E_{GFP} = 0.0102$$

= 1.02 %

③ Given: $\sigma_{vm} = 200 \text{ MPa}$; $T = 600 \text{ K}$; $LHR = 150 \text{ W/cm}$ ^{3/6}
 $t = 1.5 \text{ years} = 4.73 \times 10^7 \text{ s}$

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Find $\epsilon_{tot} = ?$

$$\dot{\epsilon}_{ss} = A_0 \left(\frac{\sigma_{vm}}{G} \right)^n \exp \left(\frac{-Q}{RT} \right) =$$

$$A_0 = 4 \times 10^{+24} \text{ s}^{-1}; G = 4.1 \times 10^{10} - 2.3 \times 10^7 \text{ T} = 27200 \text{ MPa}$$

$$n = 5; Q = 2.7 \times 10^5 \text{ J/mol}$$

$$\Rightarrow 4 \times 10^{+24} \left(\frac{200}{27200} \right)^5 \exp \left(\frac{-2.7 \times 10^5}{8.314 \cdot 600 \text{ K}} \right) =$$

$$= 8.597 \times 10^{13} \cdot 3.116 \times 10^{-24} = 2.679 \times 10^{-10} \text{ s}^{-1}$$

$$\dot{\epsilon}_{ir} = C_0 \Phi^{C_1} \sigma_{vm}^{C_2}$$

$$C_0 = 2.714 \times 10^{-24}$$

$$C_1 = 0.85$$

$$C_2 = 1$$

$$\Phi = 3 \times 10^{11} (150) = 4.5 \times 10^{13}$$

$$\rightarrow 2.714 \times 10^{-24} (4.5 \times 10^{13})^{0.85} \cdot (200)^1$$

$$= 2.187 \times 10^{-10} \text{ s}^{-1}$$

$$\dot{\epsilon}_{tot} = \dot{\epsilon}_{ss} + \dot{\epsilon}_{ir} = 4.857 \times 10^{-10}$$

$$t = 4.73 \times 10^7 \text{ s}$$

$$\epsilon_{tot} = (\dot{\epsilon}_{tot}) t = 0.023 = 2.3\%$$

4) Five types of FPs in fuel: ① Solid solution

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- ② Oxide Precipitates
- ③ Metallic precip.
- ④ Volatile Gases
- ⑤ Noble Gases

5) Microstructure-based fuel perf modeling and why it's beneficial:

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Lower scale models provide insight into microscale mechanisms that explain macroscale behaviors. Microstructure fuel perf. modeling takes Grain growth/porosity/defect transport/densification, and model fission gas behavior to predict evolution of microstructure, to inform property or structure relationships. Benefit: It can provide a structure/property relationship that can replace burnup dependant models, and can develop things like k as a function of their microstructure evolution (grain boundary bubbles/porosity/fission precipitates) to parameterize the model and get better correlation to data as it provides a more prototypic model with better accuracy.

6) 3 benefits of using Zr clad: ① Low ② cross-section

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- ③ Good therm. conductivity
- ③ Corrosion & void swell resistant

7) Metallic-fuel const. redistribution: Zr diff up thermal gradient, has different solubilities in each phase, thus goes out radially into distinct radial zones. Different Zr-melt-temps @ diff phases/context

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8) Difference in MOX fuel vs. LWR fuel

MOX fuels like PuO_2 can be used in fast reactors, allows ability to burn weapons grade Pu. Has diff. neutronics, FGR, therm cond, higher operation LHR (400-500 W/cm) vs. LWR (250 W/cm). Smaller than classical rod dia in MOX, thus Power dens. and heat flux much higher. Much higher neutron flux, higher Burnup. Provides reliable/robust fuel @ higher Powers.
- redistribution, higher temperatures

9) SCC requires:

- ① Corrosive env. Fission Products in fuel-ded gap diffusion/crack penetration.
- ② Susceptible Mat. All Zr-alloys are prone to PCI
- ③ Sufficient stress. Fuel expansion/swelling, creep, and pressures due to env.
- ④ Sufficient time. Long operation of terrestrial Power reactors ... on order of years.

⑤ Fuel Pulverization in HBS can occur during LOCA, trapped gas bubbles heat up and over-pres, causing cracking and has been seen in regions with partial/completely formed HBS, as grain size decr. Being addressed by modeling in BISON, to develop criteria req'd to form. Addressed also by reducing LOCA risk.

⑪ RIA - Radiation induced ~~while~~ LOCA is loss of coolant $\frac{6}{6}$ induced. RIA caused by fast rise in power, while LOCA caused by slower ballooning of clad and rapid oxidation. $\frac{8}{8}$ RIA causes high rise in power and temp. RIA occurred at Chalk River, SL-1, and Chernobyl caused by operator errors and exceedance of ops. guidelines and safety systems turned off.

⑫ Can make fuel more tol by: ① Improve fuel perf. by operating $\frac{6}{6}$ at lower temp or raise melt temps. ② Improve reactor kinetics w/ steam or better retention of GFP & SFP

⑬ LWR Ops is limited by: ① PCI, clad oxidation $\frac{4}{4}$ ② Power to melt, fuel rod int. Pres or dept. from DNBR.