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NE SSS Test 3

1. $t_{total} = ?$ $\alpha_{th} = 10 \times 10^{-6}$ $\dot{F} = 6 \times 10^{12} \frac{1/25}{cm^2 s}$ $T_g = 1600h$ $T_{ref} = 500h$ $\Delta g_0 = 0.015$

$b_0 = 5 MWDP / \ln u \approx 0.005 FMA$ $g(1002) = 10.97 \frac{g}{cc}$ $t > 300 \text{ days}$

$t_{th} = \Delta T = (10 \times 10^{-6})(1600 - 500) = 0.011$

$t_p: \beta = \frac{\dot{F} t}{N_u}$ $N_u = 10.97 \frac{g}{cc} \cdot \frac{1 \text{ mol}}{270g} \cdot \frac{0.022 \times 10^{23}}{1 \text{ mol}} \cdot \frac{1 \mu}{1002} = 2.447 \times 10^{22} \frac{u}{cc}$

$\Rightarrow \beta = \frac{(6 \times 10^{12} \frac{1/25}{cm^2 s})(300 \cdot 24 \cdot 3600)}{2.447 \times 10^{22} \frac{u}{cc}} = 0.063555$

$T > 750^\circ C \Rightarrow \rho = 1$

$t_0 = \Delta g_0 \left(\exp\left(\frac{\rho \ln(\alpha_{th})}{\rho_0 \beta_0}\right) - 1 \right) = 0.015 \left(\exp\left(\frac{(0.063555) \ln(0.011)}{(1)(0.005)}\right) - 1 \right) = -0.015$

$t_{exp} = 5.577 \times 10^7 g \beta = 5.577 \times 10^7 (10.97)(0.063555) = 0.03888$

$t_{gfp} = 1.96 \times 10^{-28} g \beta (2800 - T)^{11.73} \exp(-0.0162(2800 - T)) \exp(-17.8 g \beta)$
 $= (1.96 \times 10^{-28})(10.97)(0.063555)(2800 - 1600)^{11.73} \exp(-0.0162(2800 - 1600)) \exp(-17.8(10.97)(0.063555))$
 $= 2.643 \times 10^{-6}$

$t_{total} = t_{th} + t_0 + t_{exp} + t_{gfp} = 0.011 - 0.015 + 0.03888 + 2.643 \times 10^{-6} = 0.0348876 \text{ or } 3.49\%$

2. Total creep: $R \times A$, $\sigma_m = 85 MPa$, $T = 650h$, $LHR = 200 \frac{u}{cm}$, $t = 200 \text{ day}$

$A_0 = 3.14 \times 10^{24}$ $G = 4.259 \times 10^{10} - 2.2185 \times 10^7 T$, $n = 5$, $Q = 2.7 \times 10^5$, $C_0 = 1.654 \times 10^{-24}$, $C_1 = 0.85$

$C_2 = 1$ $R = 8.314 \frac{J}{mol K}$

$\dot{\epsilon}_{ss} = A_0 \left(\frac{\sigma_m}{G} \right)^n \exp\left(\frac{-Q}{RT}\right) = 3.14 \times 10^{24} \left(\frac{85000000 Pa}{4.259 \times 10^{10} - 2.2185 \times 10^7 (650)} \right)^5 \exp\left(\frac{-2.7 \times 10^5}{(8.314)(650)}\right)$
 $= 1.593 \times 10^{-10} \frac{1}{s} \approx 3 \times 10^{-11}$

$\dot{\epsilon}_{ir} \approx b = 5 \times 10^{-11} LHR = 5 \times 10^{-11} (200) = 1 \times 10^{-14} \frac{u}{cm^2 s}$

$\dot{\epsilon}_{ir} = C_0 b^n \sigma_m^{C_1} = (1.654 \times 10^{-24})(1 \times 10^{-14})^{0.85} (85000000)^1 = 1.167 \times 10^{-10} \frac{1}{s}$

$t_{tot} = (\dot{\epsilon}_{ss} + \dot{\epsilon}_{ir})t = (1.593 \times 10^{-10} + 1.167 \times 10^{-10})(200 \times 24 \times 3600) = 0.00108 \text{ or } 0.468\%$

3. F6 release: $t = 60 \text{ day}$; $T = 1500h$; $\dot{F} = 3 \times 10^{13} \frac{1/25}{cm^2 s}$ $a_1 = 10 \mu m$ $a_2 = 25 \mu m$

A) $\gamma = 0 \frac{u}{a^2}$; $D = D_1 + D_2 + D_3$ $b_0 = 8.6173 \times 10^{-5} \frac{eV}{K}$

$D_1 = 7.6 \times 10^{-14} \exp\left(\frac{-3.03eV}{b_0 T}\right) = 5.01675 \times 10^{-16} \frac{cm^2}{s}$

$D_2 = 1.41 \times 10^{-18} \exp\left(\frac{-1.19eV}{b_0 T}\right) \sqrt{\dot{F}} = 7.754 \times 10^{-16} \frac{cm^2}{s}$

$D_3 = 2.0 \times 10^{-30} \dot{F} = 6 \times 10^{-17} \frac{cm^2}{s}$

$D = 5.01675 \times 10^{-16} + 7.754 \times 10^{-16} + 6 \times 10^{-17} = 1.337675 \times 10^{-15} \frac{cm^2}{s}$

$\Rightarrow \gamma = 1.337675 \times 10^{-15} \left(\frac{(60 \times 24 \times 3600)^{1/2}}{(10 \times 10^{-4} cm)^2} \right) = 0.0093$ $C \# \approx 0.109$

$S = 4 \sqrt{\frac{D t}{\pi a^2}} - \frac{3}{2} \frac{R t}{a^2} = 0.1775 \text{ or } 17.75\%$

$$\alpha = 25 \times 10^{-4} \text{ cm}$$

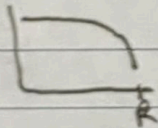
$$B) \gamma = 1.337675 \times 10^{-15} \cdot \frac{(20 + 24 + 3600)}{(25 \times 10^{-4})^2} = 0.001109 \text{ } \epsilon \pi^{-2} \approx 0.101$$

$$f = 4 \sqrt{\frac{D_F}{D_A^2}} - \frac{3}{2} \frac{D_F}{D_A^2} = 0.07349 \text{ or } 7.349\%$$

4. U^{4+} is the typical charge state in UO_2 because O^{2-} is usual oxygen state. This changes as fission products build up and change electronegativity system state. U^{3+} , U^{4+} , U^{5+} , and U^{6+} are viable states and form U_4O_9 , U_3O_8 , UO_3 . O/M increases w/ burnup.

5. Oxygen concentration decreases with radius and drops sharply at the fuel/cladding interface. ZrO_2 is more favorable than the ~~more~~ more compound.

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→ fission product

Mo is a fuel additive to help stabilize ~~the~~ O/M ratios in fuel. It functions as a oxygen sink for oxygens released after a Uranium fission. ~~the~~ Slows O/M increase w/ burnup.

6. Melting temperature, thermal conductivity, and creep vary as a function of stoichiometry in UO_2 .

7. Fission product types:

- Soluble Oxides (Y, La, rare earths): Dissolution within sublattice. Chemically inert w/ FP's.
- Insoluble Oxides (Zr, Ba, Sr): Form insoluble oxides in fluorite lattice. Take O places in lattice.
- Metals (Mo, Ru, Pd, Te): Form metallic precipitates, cause swelling.
- Volatiles (Br, Rb, Te, I, Cs): Gas at intem, solid at extem, some are corrosive.
- Noble gas (Xe, Kr): Insoluble, form voids decrease thermal conductivity.

8. Fission gas release:

9/9 Stage 1: Produce FG from fission + diffuse to GB or from ^{intra} intergranular gas bubble (Size limited num)

Stage 2: Gas bubbles nucleate on GB, bubbles grow + interconnect

Stage 3: Gas travels through interconnected bubbles to free surface

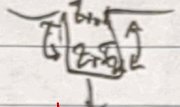
9. Creep is material deformation at stresses below the yield stress that happen over time caused by defect diffusion. Thermal creep, or both diffusion (Nabarro-Herring creep), is defect diffusion at high temperatures.

10. For cladding benefits include: High corrosion resistance, high thermal conductivity, cheap. Helps retain FPs, maintain core geometry, and contain pellets

11. Loops form on the pyramidal plane (a), causing growth in a and contraction in c plane.
- vacancy loops?

12. Stage 1: cladding contracts from water pressure and gets larger longer
2: Partial clad/fuel contact presses out at contact causing elongation w/ fuel growth
3: Full clad/fuel contact, cladding expands out and contracts in length dimension
- axial stresses from fuel?

13. Time: Build environment + FeO_2 branch in internal Corrosive environment: I move to Fe cladding + remove material
14/16 Tensile stress: Crack line fuel apply stress after clad/fuel contact from creep Susceptible material: Fe metal exposed. - more...
- wanted a bit more for each

14. Zr gas more to sorption + corrode Zr metal. ZrO_2 bands w/
expose Zr . Pits in the metal until thin ligament left and
the ductile tear creates failure. But PCI heal after
 $\frac{4}{5}$ pressure failure.  cyclic degradation.

- got most pieces, but not fully

$$\bar{E} = \frac{C \sigma^m}{D_{ac}} e^{-\frac{Q}{RT}}$$
$$t_{\text{total}} = 0.0121 + 0.004 + 5.16e^{-9} + 0.002 = 0.008 \text{ or } 0.8\% \text{ max!}$$