

P.1

Que.1

Given that, The boundary conditions are, $T'(x_0) = 0$

$$x_0 = 0$$

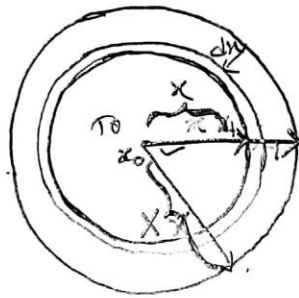
$$x_1 = X$$

$$T(x_1) = T_1$$

Now, Solving the temperature profile with boundary condition we can write,

$$\frac{d}{dx_0} (xk \frac{dT}{dx}) = -Qx$$

$$\therefore xk \frac{dT}{dx} = \frac{-Qx^2}{2} + C_1 \quad \left[\begin{array}{l} \text{differentiating} \\ \text{them with} \\ \text{r.t. } x \end{array} \right]$$



$$\text{Or, } 0 = \frac{-Q \cdot 0^2}{2} + C_1$$

$$\text{Now, } \frac{dT}{dx} = - \frac{Qx}{2k}$$

$$\text{Or, } T(x) = - \frac{Qx^2}{4k} + C_2 \quad \text{--- (1)}$$

$$\text{Or, } C_2 = \frac{QX^2}{4k} + T_1$$

Putting the value of C_2 in eqⁿ (1) we can get,

$$T(x) = \frac{-Qx^2}{4k} + \frac{QX^2}{4k} + T_1$$

$$\text{Or, } T(x) = \frac{Q(X^2 - x^2)}{4k} + T_1$$

$$\text{Or, } T(x) - T_1 = \frac{Q(X^2 - x^2)}{4k}$$

$$\text{Or, } T_0 - T_1 = \frac{Q(X^2 - x^2)}{4k}$$

$$\text{Or, } T_0 - T_1 = \frac{QX^2}{4k} \quad \because x = x_0 = 0$$

Assumptions:

The assumptions that are made to get to this equations are:

- (i) Steady state
- (ii) Axisymmetric
- (iii) Constant $c_p \rightarrow$
- (iv) Constant thermal conductivity.

Q82

Given that, Coating $k = 0.015 \text{ W/cm-K}$

Aluminum, $k = 0.15 \text{ W/cm-K}$.

Gap $k = 0.004 \text{ W/cm-K}$.

Fuel $k = 0.05 \text{ W/cm-K}$.

$P_s = 600 \text{ K}$.

$Q = 250 \text{ W/cm}^3$

$R_F = 0.8 \text{ cm}$

$t_g = 0.005 \text{ cm}$

$t_{\text{coat}} = 0.05 \text{ cm}$

$t_{\text{cool}} = 0.01 \text{ cm}$

$T_{\infty} = 600 \text{ K}$.

∴ Centroidal temperature without coating,

$$\begin{aligned}
 T_0 &= \frac{QR^2}{4k} + T_s \\
 &= \frac{250 \text{ W/cm}^3 \times (0.8 \text{ cm})^2}{4 \times 0.05 \text{ W/cm-K}} + 600 \text{ K} \\
 &= 1050 \text{ K}.
 \end{aligned}$$

Centroidal temperature with coating,

$$\begin{aligned}
 T_{cl} &= \frac{LHR \cdot t_{\text{cool}}}{2 \pi R_F k_{\text{cool}}} + T_{\infty} \\
 &= \frac{\cancel{LHR}}{2 \pi} k_{\text{gap}} \frac{t_{\text{cool}}}{k_{\text{cool}}} + T_{\infty} \quad \left[\because \frac{LHR}{2 \pi R_F} = k_{\text{gap}} \right] \\
 &= 0.004 \text{ W/cm-K} \times (0.05 \text{ cm} / 0.15 \text{ W/cm-K}) + 600 \text{ K}
 \end{aligned}$$

$$= 600.001 \text{ K}.$$

Now, ~~For~~ full centerline temperatures with

$$\text{coating} = T_{C1} + T_{C0}$$

$$= 600.001 \text{ K} + 600 \text{ K}$$

$$= 1200.001 \text{ K}$$

Ans:

Q.6

Fertile: In fertile isotopes can convert into fissile isotopes.

Fissile: In that isotopes the atoms can undergo fission reactions.

Fissionable: In that nuclide is capable of undergoing fission after capturing high energy neutrons or low energy neutrons.

Q.7Reasons:

- (i) During normal cycling, pure uranium dramatically swells.
- (ii) ^{238}U has both anisotropic thermal expansion and anisotropic irradiation growth.

Q.8Stowage density:

The percentage of actual fuel weight contained in the volume enclosed by a unit

p.4

Length of abiding relative to the weight which could be contained if the same volume were occupied by 100% dense fuel.

Necessary because

Because of \rightarrow

\Rightarrow It provides information on the number and shape of the cells.

Q: 9

\Rightarrow Enrich U because it needs to have a higher concentration rather than the U exists in nature.

\Rightarrow UF₆

\Rightarrow Centrifugal works by \rightarrow

~~rot~~ rotating at rapid speed

\downarrow

separating substances using centrifuge

\downarrow

then work as final work

Q.10

METHOD → Euler method

- Finite difference method
- Taylor series method

Q.11

Re DEPARTURE from nucleate boiling and
 increased heating flux. because of increase
 the heat transfer due to mixing
 and turbulence near to the heater
 surface.

Q.12

Layer → Light bridge → fuel kernel
 → kdinos → Buffer.
 → NABA/LANC → IPYC
 → SIC
 → OPYC.

Ex: ~~UC~~

Ex: UC.