

90

Exam 1 NE 553

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13/15

1) UN enrichment 19.50%,  $\rho = 12.30 \text{ g/cm}^3$ ,  $d = 5 \times 10^{12} \text{ n/cm}^2 \text{ s}$   
 $\sigma_f = 587 \text{ barns}$   
 a)  $Q = E_f N_f \phi$   
 $\text{Mass U} = 0.195 \times 235 + (1 - 0.195) \times 238 = 237.415 \text{ g/mol}$

assume UN not  $\text{U}_2\text{O}_8$  or  $\text{U}_2\text{N}_8$   
 1 uranium atom per UN = 1 nitrogen per UN  
 $\text{Mass UN} = 237.415 + 14 = 251.415 \text{ g/mol}$

$$N_f = 12.30 \frac{\text{g}}{\text{cm}^3} \times \frac{1 \text{ mol}}{251.415 \text{ g/mol}} \times \frac{6.022 \times 10^{23}}{1 \text{ mol}} \times \frac{10}{1 \text{ UN}} \times (0.195)$$

$$N_f = 5.75 \times 10^{21} \checkmark$$

$$Q = (5.75 \times 10^{21}) \left( 587 \times 10^{-24} \right) \left( 5 \times 10^{12} \right) \left( 200 \times 10^6 \right) \left( 1.602 \times 10^{-19} \right)$$

$$Q = 540.7 \text{ W/cm}^3 \checkmark$$

b) for  $\text{UO}_2$   $\rho = 10.97 \text{ g/cm}^3$

$$N_f = 10.97 \times \frac{1 \text{ mol}}{269.14 \text{ g/mol}} \times \frac{6.022 \times 10^{23}}{1} \times \frac{10}{1 \text{ UO}_2} (e) = ?$$

Assume  $\sigma_f$  &  $\phi$  are same.  $\checkmark$

So  $N_f =$

$$N_f = 5.75 \times 10^{21} \text{ (calculated above)} \checkmark$$

$$e = 0.234$$

$$\text{Enrichment needed} = 23.4\% \checkmark$$

- would like to see work here for dup part

$$2) \frac{\partial}{\partial x} \left( k \frac{\partial T}{\partial x} \right) + Q = 0$$

11/12

$$\int \frac{\partial}{\partial x} \left( k \frac{\partial T}{\partial x} \right) = \int -Q$$

← assume  $Q$  doesn't change w/ respect to  $x$

$$k \frac{\partial T}{\partial x} = -Qx + C_1$$

$$\frac{\partial T}{\partial x} = \frac{-Qx + C_1}{k}$$

assume  $k$  doesn't change w/ respect to  $x$

$$\frac{\partial T}{\partial x} \Big|_{x=0} \Rightarrow 0 = \frac{C_1}{k} \Rightarrow C_1 = 0$$

$$\int \frac{\partial T}{\partial x} = \int \frac{-Qx}{k}$$

$$T(x) = \frac{-Qx^2}{2k} + C_2$$

$$T(x) \Rightarrow T_1 = \frac{-Qx^2}{2k} + C_2 \Rightarrow C_2 = T_1 + \frac{Qx^2}{2k}$$

$$T(x) = \frac{Q}{2k} (x^2 - x^2) + T_1$$

Assumptions  $Q, k$  are constant w/ respect to  $x$

- Steady state, 1-D



3)  $k_{\text{cond}} = 0.15 \frac{\text{W}}{\text{cm} \cdot \text{K}}$   $k_{\text{fuel}} = 0.03$   $h_{\text{cool}} = 25 \frac{\text{W}}{\text{cm}^2 \cdot \text{K}}$   $Q = 350 \frac{\text{W}}{\text{cm}^2}$

17/20

$R_F = 0.6 \text{ cm}$   $t_g = 0.003 \text{ cm}$   $t_{\text{clad}} = 0.05 \text{ cm}$   $T_{\text{cool}} = 550 \text{ K}$

(1ac)

$$T_{\text{co}} - T_{\text{cool}} = \frac{Q}{2h_{\text{cool}}} R_{\text{fuel}}$$

$$\text{LHR} = \pi R^2 Q$$

$$= 395.8 \frac{\text{W}}{\text{cm}^2}$$

$$T_{\text{co}} = \frac{Q R_F}{2h_{\text{cool}}} + T_{\text{cool}} \Rightarrow T_{\text{co}} = 542 \text{ K}$$

$$T_{\text{ci}} - T_{\text{co}} = \frac{\text{LHR}}{2\pi R_F} \frac{t_c}{k_c}$$

$$\frac{\text{LHR}}{2\pi R_F} = 105$$

$$T_{\text{ci}} = 105 \left( \frac{0.05}{0.15} \right) + 542 \text{ K}$$

$$T_{\text{ci}} = 627 \text{ K}$$

$$T_s - T_{\text{ci}} = \frac{\text{LHR}}{2\pi R_F} \left( \frac{t_g}{k_g} \right)$$

$$k_g(H_2) = 16 \times 10^{-6} \text{ }^{279}$$

$$k_g(T_{\text{ci}}) = 0.00259$$

$$T_s = 105 \left( \frac{0.003}{0.00259} \right) + T_{\text{ci}}$$

$$T_s = 748.6 \text{ K}$$

$$T(r) = \frac{\text{LHR}}{4\pi k_f} \left( 1 - \frac{r^2}{R_F^2} \right) + T_s$$

$$T_{\text{center}} = 1027.1 \text{ K}$$

$$\left( \frac{105}{4\pi(0.05)} \right) \left( 1 - \frac{0}{0.6^2} \right) + T_s = 1027.1 \text{ K}$$

$$\text{LHR} = 396$$

Accounting for K temp dependence

$$\frac{1}{B} \ln \left( \frac{A + BT_0}{A + BT_5} \right) = \frac{LHR}{4\pi}$$

- I assume you used  
LHR wrong again  
here

$$A = 3.8 + 200 \text{ J/mol}$$

$$B = 0.0217$$

first five

$$\ln \left( \frac{\frac{LHR}{4\pi} B}{A + BT_5} \right) - A$$

$$= T_0$$

B

$$T_{\text{center}} = 932.23 \text{ K}$$

Temp goes down ✓



4)  $L = 3.2$   $LHR^0 = 360 \text{ W/cm}$   $(\gamma = 1)$

Find  $LHR$  @  $z = 0.3 \text{ m}$   $\frac{z}{z_0}$

7/12

$$LHR(z) = LHR^0 \cos\left(\frac{\pi}{2\delta} \left(\frac{z_0}{z} - 1\right)\right)$$

$$z_0 = \frac{L}{2} = 1.6 \text{ m}$$

$$LHR(0.3) = 360 \cos\left(\frac{\pi}{2(1.1)} \left(\frac{1.6}{0.3} - 1\right)\right)$$

$$\frac{0.3}{z_0} = \frac{0.3}{1.6}$$

$$LHR(0.3) = 42.69 \text{ W/cm}$$

- this is for too low to make sense  $\frac{z}{z_0}$

$$b) T_{in} - T_{cool} = \frac{2(1.1)}{\pi} \left( \frac{1.6(360 \text{ W/cm})}{(0.3)(42.69)} \right) \left[ \sin\left(\frac{\pi}{2(1.1)}\right) + \sin\left(\frac{\pi}{2(1.1)} \left(\frac{1.6}{0.3} - 1\right)\right) \right]$$

$$500 - T_{cool} = 0.528$$

$$T_{cool} = 499.47 \text{ K}$$

- temp. increases as  $z$  increases  
 $T(z) - T_{cool}$

5) There is very little U-235 naturally occurring, it must be enriched to obtain fissile material ✓

8/8

UF<sub>6</sub> is used during enrichment ✓

Centrifuge:

UF<sub>6</sub> is supplied to the centrifuge, heavier molecules move to the outside of the rotating drum ✓ U-238 is heavier so U-235 congregates in the center ✓



7/8 6) Departure from nucleate boiling is when a thin film of vapor is formed at the interface and the system begins transitioning to film boiling

Critical heat flux is the max heat flux observed - where system dries out + have gas film

3/3 7) Fertile: material not fissionable but that can be converted to be fissile

Fissile: capable of sustaining a chain reaction  
Fissionable: capable of undergoing fission w/ high E neutrons

3/3 8) - Pure U has many phases, difficult to control  
- Drastic thermal swelling

3/4 9) Ratio of fuel volume to total volume. It is used to help determine cladding interactions w/ fuel + predict behavior  
- allows fuel swelling, decrease PCM!

3/3 10) Layers: Kernel | PYC, SiC layer, OPYC, shell  
- better

3/3 11) AL ATF cladding example is FeCrAl

0/3 12) U-235 - rarely used Plutonium fission products? Xe, Ce, etc.

3/4 13) Contain fuel + fission products

3/3 14) Fuel, cladding, coolant gap

4/4 15) - that generation (amounts)  
- Length of time fuel operates w/out problems  
- How well it performs during an accident