

(2)  $K_{coat} = 5 \text{ W/mK} = 0.05 \text{ W/cmK}$   
 $K_{lead} = 15 \text{ W/mK} = 0.15 \text{ W/cmK}$   
 $K_F = 0.5 \text{ W/cmK}$   
 $K_{gap} = 25 \text{ W/mK} = 0.25 \text{ W/cmK}$

$h_{cool} = 5.5 \text{ W/cm}^2 \text{K}$   
 $T_{cool} = 200 \text{ K}$   
 $Q = 400 \text{ W/cm}^2$   
 $r_1 = 0.6 \text{ cm} = r_4$   
 $r_2 = 0.2 \text{ cm}$   
 $r_3 = 0.25 \text{ cm}$   
 $r_4 = 0.25 \text{ cm}$

$$LHR = Q \pi R_F^2 = 400 \pi (0.6)^2 = 452.4 \text{ W/cm}$$

$$T_{coat} = T_{cool} + \frac{LHR}{2 \pi R_F h_{cool}} = 200 + \frac{452.4}{2 \pi (0.6) (5.5)} = 821.8 \text{ K}$$

$$T_{lead} = T_{coat} + \frac{LHR \cdot t_{coat} \rightarrow (\lambda_4 - \lambda_3)}{2 \pi R_F K_{coat}} = 821.8 + \frac{452.4 (0.25 - 0.2)}{2 \pi (0.6) (0.05)} = 845.8 \text{ K}$$

$$T_{gap} = T_{lead} + \frac{LHR \cdot t_{lead} \rightarrow (\lambda_3 - \lambda_2)}{2 \pi R_F K_{lead}} = \frac{452.4 (0.25 - 0.2)}{2 \pi (0.6) (0.15)} + 845.8 = 885.8 \text{ K}$$

$$T_{fuel} = T_{gap} + \frac{LHR \cdot t_{gap} \rightarrow (\lambda_2 - \lambda_1)}{2 \pi R_F K_{gap}} = 885.8 + \frac{452.4 (0.2 - 0.6)}{2 \pi (0.6) (0.25)} = 981.8 \text{ K}$$

$$T_o = T_{fuel} + \frac{LHR}{4 \pi K_F} = 981.8 + \frac{452.4}{4 \pi 0.5} = \boxed{1053.8 \text{ K}}$$

@  $r = 0.4 \text{ cm}$ ,

$$T(r) = \frac{Q (R^2 - r^2)}{4K} + T_F \Rightarrow T(0.4) = \frac{400 (0.6^2 - 0.4^2)}{4(0.5)} +$$

③  $L = 145 \text{ W/mK}$   $\text{U}_3\text{Si}_2$   
enrichment: 19.5%  
 $\rho = 15.67 \text{ g/cm}^3$   
 $\sigma_f = 570 \text{ b}$

a)  $\phi = 2 \times 10^{12} \text{ n/cm}^2\text{s}$

mass U:  $0.195(235) + 0.805(238) = 237.415 \text{ a.m.u.}$

mass Si:  $28 \text{ a.m.u.}$

total mass:  $3(237.415) + 2(28) = 768.245 \text{ a.m.u.}$

$$N_f = 15.67 \frac{\text{g}}{\text{cm}^3} \cdot \frac{\text{mol}}{768.245} \cdot \frac{6.022 \times 10^{23} \text{ at.}}{\text{mol}} \cdot \frac{30}{103.84} \cdot 0.195 = 7.19 \times 10^{21} \frac{\text{at.}}{\text{g}}$$

$$Q = E_f N_f \phi \sigma_f$$

$$Q = (200 \times 10^6 \text{ eV}) \left( 1.602 \times 10^{-19} \frac{\text{J}}{\text{eV}} \right) \left( 7.19 \times 10^{21} \frac{\text{U}^{235}}{\text{g}} \right) \left( 2 \times 10^{12} \frac{\text{n}}{\text{cm}^2\text{s}} \right) \left( 570 \times 10^{-24} \text{ cm}^2 \right)$$

$$Q = 262.62 \text{ W/cm}^3$$

b)  $\rho_{\text{UO}_2} = 10.97 \text{ g/cc}$

$$N_f = 7.19 \times 10^{21} = \left( 10.97 \frac{\text{g}}{\text{cc}} \right) \frac{1}{M} \cdot \frac{6.022 \times 10^{23} \text{ at.}}{\text{mol}} \cdot 1x$$

$$\frac{x}{M} = 1.09 \times 10^{-3} \Rightarrow M = \frac{x}{1.09 \times 10^{-3}}$$

$$M = (x \cdot 235 + (1-x) \cdot 238) + 2 \cdot 16 = \frac{x}{1.09 \times 10^{-3}}$$

$$x = 0.2557x + 0.259 - 0.259x + 3.48 \times 10^{-2}$$

$$x = 29.3\% \text{ enrichment}$$

ok 2.2.25

④  $z_0 = 3 \text{ m}$  ;  $LHR^0 = 150 \text{ W/cm}^2$  ;  $\delta = 1$  !

a)  $LHR @ z = 1.8 \text{ m}$

$$LHR\left(\frac{z}{z_0}\right) = 150 \cos\left[\frac{\pi}{2(1.1)}\left(\frac{1.8}{3} - 1\right)\right] = 46.66$$

b) i) Water  $c_p = 4200 \text{ J/kgK}$  ,  $\dot{m} = 0.22 \text{ kg/s cool}$

$$T(z)_{\text{cool}} - T_{\text{cool}}^{\text{in}} = \frac{2 \cdot (1.1)}{\pi} \frac{3(150)}{0.22(4200)} \left[ \sin\left(\frac{\pi}{2(1.1)}\right) + \sin\left(\frac{\pi}{2(1.1)}\left(\frac{1.8}{3} - 1\right)\right) \right]$$

$$T(z)_{\text{cool}} - T_{\text{cool}}^{\text{in}} = 0.1532 \text{ K}$$

$$= 0.54$$

ii) sodium  $c_p = 1404 \text{ J/kgK}$   $\dot{m} = 0.12 \text{ kg/s cool}$

$$T(z)_{\text{cool}} - T_{\text{cool}}^{\text{in}} = 0.1532 \times \frac{0.22(4200)}{0.12(1404)} = 0.8402 \text{ K}$$

⑤ Forward Euler:  $f(t_{n+1}) = f(t_n) + \Delta t f'(t_n)$

$$t_1 = 0.33 \quad f_1 = 6 + 0.33 [4(0) - 3(0)^2] = 6$$

$$t_2 = 0.66 \quad f_2 = 6 + 0.33 [4(0.33) - 3(0.33)^2] = 6.33$$

$$t_3 = 0.99 \quad f_3 = 6.33 + 0.33 [4(0.66) - 3(0.66)^2] = 6.77$$

$$t_4 = 1.32 \quad f_4 = 6.77 + 0.33 [4(0.99) - 3(0.99)^2] = 7.11$$

$$t_5 = 1.65 \quad f_5 = 7.11 + 0.33 [4(1.32) - 3(1.32)^2] = 7.13$$

$$t_6 = 1.98 \quad f_6 = 7.13 + 0.33 [4(1.65) - 3(1.65)^2] = 6.61$$

Backward Euler

$$f(t_{n+1}) = f(t_n) + \Delta t f'(t_{n+1})$$

$$t_1 = 0.33 \quad f_1 = 6 + 0.33 [4(0.33) - 3(0.33)^2] = 6.33$$

$$t_2 = 0.66 \quad f_2 = 6.33 + 0.33 [4(0.66) - 3(0.66)^2] = 6.78$$

$$t_3 = 0.99 \quad f_3 = 6.78 + 0.33 [4(0.99) - 3(0.99)^2] = 7.12$$

$$t_4 = 1.32 \quad f_4 = 7.12 + 0.33 [4(1.32) - 3(1.32)^2] = 7.14$$

$$t_5 = 1.65 \quad f_5 = 7.14 + 0.33 [4(1.65) - 3(1.65)^2] = 6.62$$

$$t_6 = 1.98 \quad f_6 = 6.62 + 0.33 [4(1.98) - 3(1.98)^2] = 5.35$$

⑧ → 3 phases in operation temperature cause swelling  
→ low melting point (1400K)

⑨ Ratio of fuel volume per total volume in fuel cell.  
Important due to fuel swelling during operation

⑩ We enrich U to increase fissile quantity (U-235) in the fuel. Enrichment is possible due to mass difference between U-235 to U-238.