

①

$$\frac{\partial}{\partial x} \left(k \frac{\partial T}{\partial x} \right) + Q = 0$$

$$T'(x_0) = 0 \quad x_0 = 0$$

$$T(x_1) = T_1 \quad x_1 = X$$

$$\frac{\partial}{\partial x} \left(k \frac{\partial T}{\partial x} \right) = -Q$$

$$k \frac{\partial T}{\partial x} = -Qx + C_1$$

$$0 = -Q \cdot 0 + C_1 \Rightarrow C_1 = 0$$

$$k \frac{\partial T}{\partial x} = -Qx$$

$$\frac{\partial T}{\partial x} = -\frac{Q}{k} x$$

$$T = -\frac{Q}{k} \frac{x^2}{2} + C_2$$

$$T_1 = -\frac{Q}{k} \frac{X^2}{2} + C_2$$

$$C_2 = T_1 + \frac{Q}{k} \frac{X^2}{2}$$

$$T = -\frac{Q}{2k} x^2 + T_1 + \frac{Q}{k} \frac{X^2}{2}$$

$$T - T_1 = \frac{Q}{2k} (X^2 - x^2)$$

To get the provide equation it was assumed:

- ① steady state
- ② axisymmetric
- ③ Temperature is constant in z
- ④ the thermal conductivity is constant

② $T_0 = ?$
 $t(r=0,4 \text{ cm}) = ?$

$$K_{cot} = 5 \frac{\text{W}}{\text{m} \cdot \text{K}}$$

$$K_{clad} = 15 \frac{\text{W}}{\text{m} \cdot \text{K}}$$

$$K_f = 0,5 \text{ W/cm} \cdot \text{K}$$

$$K_g = 25 \text{ W/m} \cdot \text{K}$$

$$h_{cool} = 5,5 \text{ W/cm}^2 \cdot \text{K}$$

$$T_{cool} = 800 \text{ K}$$

$$Q = 400 \text{ W/cm}^3$$

$$r_1 = r_f = 0,6 \text{ cm}$$

$$r_2 = r_{c1} = 0,8 \text{ cm}$$

$$r_3 = r_{c0} = 0,85 \text{ cm}$$

$$r_4 = r_{cot} = 0,86 \text{ cm}$$

$$T_{cot} = \frac{Q}{2h_{cool}} R_f + T_{cool}$$

$$= \frac{400 \text{ W/cm}^3}{2 \cdot 5,5 \frac{\text{W}}{\text{cm}^2 \cdot \text{K}}} \cdot 0,6 \text{ cm} + 800 \text{ K}$$

$$= 821,81 \text{ K}$$

$$T_{co} - T_{cot} = \frac{Q \cdot t_{cot} \cdot R_f}{2 \cdot K_{cot}}$$

$$T_{co} = \frac{400 \text{ W/cm}^3 [0,86 \text{ cm} - 0,85 \text{ cm}] \cdot 0,6 \text{ cm}}{2 \cdot 5 \frac{\text{W}}{\text{m} \cdot \text{K}} \cdot \frac{1 \text{ m}}{10^2 \text{ cm}}}$$

$$+ 821,81 \text{ K}$$

$$= 845,81 \text{ K}$$

$$T_g - T_{co} = \frac{Q \cdot t_{clad} \cdot R_f}{2K_{clad}}$$

$$T_g = T_{ci} = \frac{400 \text{ W/cm}^3 \cdot (0,85 - 0,8) \text{ cm} \cdot 0,6 \text{ cm}}{2 \cdot 15 \frac{\text{W}}{\text{m} \cdot \text{K}} \cdot \frac{1 \text{ m}}{10^2 \text{ cm}}} + 845,81 \text{ K}$$

$$T_g = 885,81 \text{ K} = T_{ci}$$

$$T_f - T_{ca} = \frac{Q}{2h_{gap}} R_f$$

$$T_f = \frac{400 \text{ W/cm}^3 [0,8 - 0,6] \text{ cm} \cdot 0,6 \text{ cm} + 885,81 \text{ K}}{2 \cdot 25 \frac{\text{W}}{\text{mK}} \cdot \frac{10 \text{ m}}{10^2 \text{ cm}}}$$

$$= 981,81 \text{ K}$$

$$T_0 = \frac{Q}{4K} R_f^2 + T_f = \frac{400 \text{ W/cm}^3 \cdot (0,6 \text{ cm})^2}{9 \cdot 0,5 \frac{\text{W}}{\text{cmK}}} + 981,81 \text{ K} = \underline{\underline{1053,81 \text{ K}}}$$

$$T(r_1 = 0,4) = \frac{Q}{4K} (R_f^2 - r^2) + T_f$$

$$= \frac{400 \text{ W/cm}^3}{9 \cdot 0,5 \frac{\text{W}}{\text{cmK}}} (0,6^2 - 0,4^2) \text{ cm}^2 + 981,81 \text{ K}$$

$$= \underline{\underline{1021,81 \text{ K}}}$$

$$\textcircled{3} K_f = 15,5 \frac{\text{W}}{\text{mK}}$$

$$e = 0,195$$

$$\rho_f = 15,67 \frac{\text{g}}{\text{cm}^3}$$

$$b_f = 570 \cdot 10^{-24} \text{ cm}^2$$

$$\phi_n = 2 \cdot 10^{12} \frac{n}{\text{cm}^2 \cdot \text{s}}$$

$$1) Q = E_f \cdot A_f \cdot b_f \cdot \phi$$

$$Q = 235 \cdot 2195 + 238 \cdot 9805 = 237,415 \text{ aum}$$

$$\text{mass}(U_3Si_2) = 3 \cdot 237,415 + 2 \cdot 28 = 768,24 \text{ aum}$$

$$N_{U-235} = 15,67 \frac{\text{g}}{\text{cm}^3} \cdot \frac{\text{mol}}{768,24 \text{ g}} \cdot 6,022 \cdot 10^{23} \frac{\text{atom}}{\text{mol}}$$

$$\cdot \frac{34}{1 U_3Si_2} \cdot 0,195$$

$$= 7,18 \cdot 10^{21} \text{ atom/cm}^3$$

$$Q = 200 \text{ kW} \cdot \frac{10^6 \text{ eV}}{\text{MeV}} \cdot 1,002 \cdot 10^{-19} \frac{\text{J}}{\text{eV}} \cdot 7,18 \cdot 10^{21} \frac{\text{atom}}{\text{cm}^3} \cdot 570 \cdot 10^{-24} \text{ cm}^2 \cdot 2 \cdot 10^{12} \frac{\text{n}}{\text{cm}^2 \text{s}} \quad (49)$$

$$Q = 202,25 \text{ W/cm}^3$$

b) $\rho_{\text{O}_2} = 10,97 \frac{\text{g}}{\text{cm}^3}$ $\text{mass}(\text{O}_2) = 237,41 + 2 \cdot 16 = 269,41 \text{ g/mol}$

$$7,18 \cdot 10^{21} \frac{\text{atom}}{\text{cm}^3} = 10,97 \frac{\text{g}}{\text{cm}^3} \cdot \frac{\text{mol}}{269,41 \text{ g}} \cdot 6,022 \cdot 10^{23} \frac{\text{atom}}{\text{mol}} \cdot \frac{19}{100} \cdot \text{enrich.}$$

$$\text{enrich.} = 0,108\%$$

9) $\text{CHR} = 150 \text{ W/cm}$ a)
 $z_0 = 1,15$
 $n = 1,1$

$$\text{CHR} = 1,8 \text{ m} = ?$$

$$\text{CHR}(1,8 \text{ m}) = \text{CHR}^0 \cos \left(\frac{n}{2} \left(\frac{z}{z_0} - 1 \right) \right)$$

$$= 150 \frac{\text{W}}{\text{cm}} \cos \left(\frac{n}{2 \cdot 1,1} \left(\frac{1,8}{1,15} - 1 \right) \right)$$

$$= 143,92 \text{ W/cm}$$

b) $T_{\text{cool}} - T_{\text{cool}}^0 = \frac{1}{1,2} \cdot \frac{1,5 \cdot 150 \text{ W/cm}}{\dot{m} \text{ cp}} \left[\sin \left(\frac{n}{2 \cdot 1,1} \right) + \sin \left(\frac{n}{2 \cdot 1,1} \left(\frac{1,8}{1,15} - 1 \right) \right) \right]$

water $\Delta T = \frac{1}{1,2} \cdot \frac{1,5 \cdot 150 \text{ W/cm}}{0,22 \cdot \frac{\text{kg}}{\text{s rod}} \cdot 4200 \frac{\text{J}}{\text{kg}}}$ $\cdot 1,2715 = 1,2715$

$$= 0,2580$$

$$\text{Solro } \Delta T = \frac{1,5 \cdot 150}{1,2 \cdot 0,12 \cdot 1404} = 1,2715$$

#5:

$$= 1,415$$

Sodium see larger change.

⑤ $dt = 0,33$ $t_n = 2$ $y(t_0) = 4$ $t_0 = 1$ $y' = 4t - 3t^2$

$t_0 = 1$ $y(t_0) = 4$ Forward

$$t_1 = t_0 + \Delta t = 1 + 0,33 = 1,33 \quad y_1 = y_0 + dt y'_1 = 4 + 0,33 \cdot (4 - 3) = 4,33$$

$$t_2 = t_1 + \Delta t = 1,33 + 0,33 = 1,66$$

$$y_2 = y_1 + dt y'_1 = 4,33 + 0,33 \cdot (4 \cdot (1,33) - 3(1,33)^2) = 4,334$$

$$t_3 = t_2 + \Delta t = 2$$

$$y_3 = y_2 + dt y'_2 = 4,334 + (4(1,66) - 3(1,66)^2) = \underline{\underline{4,7072}}$$

Backward.

$$t_0 = 1 \quad y(t_0) = 4$$

$$t_1 = 1,33 \quad y_1 = y_0 + dt y'_1 = 4 + 0,33 \cdot [4(1,33) - 3(1,33)^2] = 4,004$$

$$t_2 = 1,66 \quad y_2 = y_1 + dt y'_2 = 4,004 + 0,33 \cdot [4(1,66) - 3(1,66)^2] = 5,4671$$

$$t_3 = 2 \quad y_3 = y_2 + dt y'_3 = 5,4671 + 0,33 [4 \cdot 2 - 3 \cdot 2^2] = \underline{\underline{4,1471}}$$

⑤ A fissile isotope can undergo fission by absorption of thermal neutron. On the other hand and fissionable isotope can undergo fission as well but by energetic neutrons, Finally a fertile isotope does not undergo fission by thermal or energetic neutron but can ~~absorb~~^{get} a neutron and be converted into a fissile isotope.

- ⑥ ① Because it drastically swells
② Have anisotropic thermal expansion and irradiation growths

⑦ ① smear density is the ratio of the volume of fuel pellet to fuel element.

$$\text{Smear density} = r_1^2 / r_2^2$$

it is necessary because the fuel swell so an smear density less than 1 is important to avoid fuel-cladding mechanical and chemical interactions

⑧ Uranium need to be enriched because U-235 is an small percent of natural uranium and U-238 does not undergo fission by thermal neutron.

UF₆ is utilize in the enrichment process.

In the centrifuge method the gas is placed in a centrifuge cylinder and rotate at high speed because U-235 and U-238 have different mass, the gas molecules of the U-238 will move towards the outside of the cylinder, then ~~the~~^{more} U-235 ~~can~~ enriched can be obtained.

#7

(12) $Mo, Cs \rightarrow$ fission products yields are effectively the same (double hump distribution)

(11) ① finite difference.

② finite volume

③ finite element.

finite element is more correctly used due to more flexibility with geometry and boundary conditions than finite difference which have been traditionally used. Finite volume is not used because it cannot solve for stress.