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Q(1)

Zirlo-clad.

$\delta = ?$

$T = 625 \text{ K}$

$t = 400 \text{ days}$

initial $\delta_0 = 500 \text{ } \mu\text{m}$
thickness

16/16

* find the critical time for transition

$$t^*(d) = 6.62 \times 10^{-7} \exp\left(\frac{11949}{T}\right)$$

$$t^*(d) = 6.62 \times 10^{-7} \exp\left(\frac{11949}{625}\right) = 133 \text{ day} < 400 \text{ days.}$$

which means we passed the transition

* Critical oxide thickness for transition.

$$\delta^* = 5.1 \exp\left(\frac{-550}{625}\right) = 2.115 \text{ } \mu\text{m}$$

$$* \rightarrow K_L \left(\frac{\mu\text{m}}{d}\right) = 7.48 \times 10^6 \exp\left(\frac{-12500}{625}\right)$$

$$K_L = 0.0154$$

$$\Rightarrow \delta = \delta^* + K_L (t - t^*)$$

$$\delta = 2.115 + 0.0154 (400 - 133)$$

$$\delta = 6.23 \text{ } \mu\text{m}$$

B)

H - pick up fraction = 18%
weight PPM of H in the
cladding after 1 year.

PBR = 1.56

$\rho_{Zr} = 6.5 \text{ g/cc}$

$\rho_{ZrO_2} = 5.68 \text{ g/cc}$

$$C_H^{\text{clad}} (\text{wt. ppm}) = \frac{2 \times \delta \times \rho_{\text{oxide}} \times \frac{M_H}{M_O} \times 10^6}{\left(t - \frac{\delta}{\text{PBR}}\right) \times \rho_{\text{metal}}}$$

$$\rho_{ZrO_2} = \frac{2 \times 16}{2 \times 16 + 41} = 0.26$$

$$C_H^{\text{clad}} (\text{wt. ppm}) = \frac{2 \times 0.18 \times 6.23 \text{ } \mu\text{m} \times 5.68 \times 0.26 \times \frac{1}{16} \times 10^6}{\left(500 - \frac{6.23}{1.56}\right) 6.5}$$

$$C_H (\text{wt. ppm}) = 64.2 \text{ wt. ppm.}$$

Q(2)

$$\Delta V_{total} = ?$$

$$\alpha_{th} = 11 \times 10^{-6}$$

$$\dot{p} = 3.5 \times 10^{13} \text{ Pa/s/cm}^3$$

$$T = 1200 \text{ K}$$

$$T_{ref} = 300 \text{ K}$$

$$\phi_0 = 0.01$$

$$B_D = 5 \text{ MW D/kg u}$$

$$\rho(UO_2) = 10.97 \text{ g/cm}^3$$

$$t = 85 \text{ days}$$

$$\Delta V_{total} = \epsilon_{th} + \epsilon_D + \epsilon_{sfp} + \epsilon_{gfp}$$

due to thermal expansion.

$$\epsilon_{th} = \alpha \Delta T = 11 \times 10^{-6} (1200 - 300)$$

$$\epsilon_{th} = 9.9 \times 10^{-3}$$

$$\beta(FIMA) = \frac{\dot{F}t}{N_u}$$

$$N_u = ?$$

$$\rightarrow M_{UO_2} = 269.9 \text{ g/mol}$$

$$N_u = \frac{N_A \rho_{UO_2}}{M_{UO_2}} = \frac{6.023 \times 10^{23} \times 10.97}{269.9} = 2.45 \times 10^{22}$$

$$\beta = \frac{3.95 \times 10^{23} \times 85 \times 24 \times 60 \times 60}{2.45 \times 10^{22}} = 0.01 \text{ FIMA}$$

$$\epsilon_D = \phi_0 \left(e^{\frac{\beta \ln(\phi_0)}{\phi_0 B_D}} - 1 \right)$$

$$G_D = 1$$

$$B_D = \frac{5}{950} = 0.0053 \text{ FIMA}$$

$$\epsilon_D = 0.01 \left(e^{\frac{0.01 \ln(0.01)}{0.0053}} - 1 \right) = -1 \times 10^{-2} \Rightarrow 9.9 \times 10^{-3}$$

$$\epsilon_{sfp} = 5.577 \times 10^{-2} \beta$$

$$\epsilon_{sfp} = 5.577 \times 10^{-2} \times 10.97 \times 0.01 = 6.1 \times 10^{-3}$$

$$\epsilon_{gfp} = 1.96 \times 10^{-28} \times 10.97 \times 0.01 (2800 - 1200) e^{11.73 - 0.0162(2800 - 1200)} = 17.8 \times 10.97 \times 0.01$$

$$\epsilon_{gfp} = 6.48 \times 10^{-4}$$

Change in Volume

$$\epsilon_{total} = (6.48 \times 10^{-4}) + (6.1 \times 10^{-3}) + (9.9 \times 10^{-3}) + 9.9 \times 10^{-3}$$

$$\epsilon_{total} = 6.79 \times 10^{-3}$$

Q(3) total creep = ?

$$\sigma_m = 200 \text{ MPa}$$

$$T = 600 \text{ K}$$

$$\text{LHR} = 150 \text{ W/cm}$$

$$t = 1.5 \text{ year}$$

total creep = thermal + irradiation creep.

A) thermal creep.

$$\dot{\epsilon}_t = \left(A_0 \left(\frac{\sigma_m}{G} \right)^n e^{-\frac{Q}{RT}} \right)$$

$$\dot{\epsilon}_t = \left(3.14 \times 10^{24} \times \left(\frac{200}{2.92 \times 10^4} \right)^5 e^{-\frac{2.7 \times 10^5}{8.314 \times 600}} \right)$$

$$\dot{\epsilon} = 1.47 \times 10^{-10} \quad \checkmark$$

$$A_0 = 3.14 \times 10^{24} \text{ s}^{-1}$$

$$G = 4.2519 \times 10^{10} - 2.2 \times 10^7 \text{ Pa}$$

$$G = 2.92 \times 10^{10} \text{ Pa}$$

$$Q = 2.7 \times 10^5 \text{ J/mol}$$

$$n = 5$$

$$R = 8.314$$

B) irradiation creep.

$$\dot{\epsilon}_{irr} = C_0 \phi \sigma_m^{C_1} C_2$$

$$C_0 = 3.557 \times 10^{-24}$$

$$C_1 = 0.85$$

$$C_2 = 1 \quad \checkmark$$

$$\phi = 3 \times 10^{11} \text{ LHR}$$

$$\phi = 3 \times 10^{11} \times 150 = 4.5 \times 10^{13} \text{ n/cm}^2\text{-s} \quad \checkmark$$

$$\dot{\epsilon}_{irr} = 3.557 \times 10^{-24} \times (4.5 \times 10^{13})^{0.85} \times (200)^1$$

$$\dot{\epsilon}_{irr} = 2.866 \times 10^{-10}$$

$$\dot{\epsilon}_{total} = (2.866 \times 10^{-10} + 1.47 \times 10^{-10}) = 4.34 \times 10^{-10} \text{ s}^{-1}$$

$$\epsilon_{total} = 4.34 \times 10^{-10} \times 1.5 \times 365 \times 24 \times 3600$$

$$\epsilon_{total} = 2.05 \% \quad \checkmark$$

Q(4)

fission products types.

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- 1 - Soluble ~~oxides~~ (Y, La and rare earth etc -)
- 2 - Insoluble ~~oxides~~ (Zr, Ba., Sr)
- 3 - Metals (Mo, Ru, Pd and Tc)
- 4 - Volatiles (Br, Rb, Te, I. and Cs)
- 5 - Noble ~~gases~~ (Xe, Kr)

Q(5)

Instead of depending on temp and burn up only in modelling, it ~~depends~~ on the state of the microstructure and the evolution of microstructure based on a set of variables

6/6

⇒ takes into account many variables like temp.
displacement
stoichio —

⇒ Utilize the updated microstructure to inform a number of structure/property relationships.

Q(6)

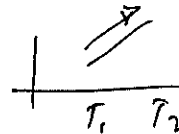
benefits of using Zr cladding.

- low neutrons absorption.
- corrosion resistance $T < 300^\circ\text{C}$
- resistance to void swelling
- good mechanical properties
- good thermal conduct —
- cost / availability. ✓

6/6

Q(7)

- Zr diffuse via Soret diffusion
- Zr. has different solubility in each phase.
- distinct zones of Zr. content in radial rings.



δ -phase \rightarrow high Zr
 β -phase \rightarrow low Zr
 $\alpha/\delta \rightarrow$ as sintered

5/5

Q(8)

Differences in MOX Fuel

behavior.

- Restructuring takes place due to high temp.
- Redistribution.
- Gap Closure.
- JOG

performance.

operate at high linear heat generation rate
higher power density and heat fluxes
highly radiation tolerant.

environment.

very intense neutron flux.

Na at 400°C $\xrightarrow[\text{bottom}]{\text{input from}}$ average coolant temp above the core is 550°C

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A) Corrosive environment.

due to chemically aggressive FP accumulation in the fuel clad gap.

B) Susceptible material.

- the susceptibility is influenced by composition, micro structure texture

- Zr-alloys are prone to PCI

C) Sufficient stress

stress coming from coolant pressure, creep, internal stress.

D) Sufficient time.

time is required for SCC to develop in the clad.

⇒ stages of SCC

- development of corrosive environment.
- initiation of SCC
- propagation of SCC
- failure.

in PCI

- environment.

FP accumulation

- susceptible material

Zr- is prone to SCC

- Stress.

mechanical stress from the contact between P-C

- time

sufficient time for contacting between P-C.



(10)

trapped gas in bubbles heats up and becomes overpressurized.

Cracking initiates at these overpressurized bubbles.

Current model.

↓
empirical.
based on T
and burn up.

Scientists are working on developing a physics based. criterion in B/B_0 that account for microstructure. Focus on HBS.

(11)

RIA

8/8 leads to a fast rise in fuel power and temp. ✓

power ramp lead to failure of fuel rods.

rapid steam generation ✓

pressure pulse.

types (CREA, CIRDA) ✓

LOCA

the coolant is reduced or lost ✓

pressure drop ✓

SCRAMS.

$T \uparrow$, $P \downarrow$

clad ballooning, burst. ✓

similar to RIA but more slow ✓

$T \uparrow \rightarrow$ gas bubbles $P \uparrow$, fuel fragmentation, FCMI

Example of RIA

Chernobyl RIA ✓

~~accident occur due to fault of reactor~~

(12)

6/6

- 1 - Improved reaction kinetics with steam.
- 2 - Improved fuel properties.
- 3 - Improved cladding properties.
- 4 - Enhanced fission product retention.

ATF options.

⇒ Cladding coatings.
protect Zr.

⇒ Alternate claddings ✓
SiC, FeCrAl

⇒ UO₂ dopants
Cr, SiC, BeO.

⇒ Alternate fuels.
U₂S₃, UN, UC.

(13) Limiting phenomena

1 - PCMT ✓

2 - Cladding elongation and assembly bow

3 - Cladding oxidation and H pick up.

4 - clad wear

5 - Power to melt

6 - internal pressure.

7 - DNBR.

4/4