

NE 591 - 010

Nuclear Fuel Performance

Exam 1

$$1) \frac{\partial}{\partial x} \left(k \frac{\partial T}{\partial x} \right) + Q = 0$$

$$T'(x_0) = 0 \quad x_0 = 0$$

$$T(x_1) = T_1 \quad x_1 = X$$

$$\int \frac{\partial}{\partial x} \left(k \frac{\partial T}{\partial x} \right) = -Q \quad dx$$

$$k \frac{\partial T}{\partial x} = -Qx + C_1$$

$$\rightarrow k(0) = Q(0) + C_1$$

$$C_1 = 0$$

$$\int k \frac{\partial T}{\partial x} = -\int Qx \quad dx$$

$$T(x) = -\frac{Q}{2k} x^2 + C_2$$

$$T_1 = -\frac{Q}{2k} X^2 + C_2$$

$$C_2 = T_1 + \frac{Q}{2k} X^2$$

$$T(x) - T_1 = \frac{Q}{2k} X^2 - \frac{Q}{2k} x^2 \rightarrow \frac{Q}{2k} (X^2 - x^2)$$

Steady-state

1-D

K is constant

$$2) \quad T_o? \quad T(r=0.4)?$$

$$LHR = Q_{in} \pi R_p^2$$

$$LHR = (400 \text{ W/cm}) \pi (0.6^2)$$

$$LHR = 452 \text{ W/cm}$$

$$T_{co,1} - T_{co,1} = \frac{LHR}{2\pi R_F h_{co,1}}$$

$$T_{co,1} - 800 = \frac{452}{2\pi (0.6)} \frac{1}{5.5} = 21.8 \text{ K} \quad \frac{LHR}{2\pi R_F} = 120$$

$$T_{co,1} = 821.8 \text{ K}$$

$$T_{co} - T_{co,1} = \frac{LHR}{2\pi R_F} \frac{t_{co,1}}{K_{co,1}} = (120) \left(\frac{0.01}{0.05} \right) = 24 \text{ K}$$

$$T_{co} = 845.8 \text{ K}$$

$$T_{c,1} - T_{co} = \frac{LHR}{2\pi R_F} \frac{t_c}{K_c} = (120) \left(\frac{0.05}{0.15} \right) = 40 \text{ K}$$

$$T_{c,1} = 885.8 \text{ K}$$

$$T_F - T_{c,1} = \frac{LHR}{2\pi R_F} \frac{t_F}{K_F} = (120) \left(\frac{0.2}{0.25} \right) = 96 \text{ K}$$

$$T_F = 981.8 \text{ K}$$

$$T_o - T_F = \frac{LHR}{4\pi K_F} = \frac{452}{4\pi (0.5)} = 72 \text{ K}$$

$$T_F = 1053.7 \text{ K}$$

$$T(r=0.4) = \frac{Q}{4K} (R^2 - r^2) + T_F = \frac{400}{4(0.5)} (0.6^2 - 0.4^2) + 981.8 \text{ K}$$

$$T(r=0.4) = 1021.8 \text{ K}$$

3) U_3Si_2 $K = 14.5 \frac{W}{m^2 \cdot K}$ enrich = 19.5%

$\rho = 15.67 \frac{g}{cc}$ $\phi = 2 \times 10^{14} \frac{1}{cm^2 \cdot s}$

Q? $Q = E_f N_f \sigma_f \phi$

$N_f?$ $N(U) = 235 \times 0.195 + 238 \times 0.805 = 237.415$
 $n(U_3Si_2) = 3(237.415) + 2(28) = 768.245 \frac{g}{mole}$

$15.67 \frac{g}{cc} \frac{(mole)}{768.245 g} \frac{6.022 \times 10^{23} \frac{1}{mole}}{1 mole} + \frac{34}{U_3Si_2} \times 0.195 = 7.19 \times 10^{21}$

$Q = (200 \times 10^6 eV) \left(1.602 \times 10^{-19} \frac{J}{eV} \right) (7.19 \times 10^{21}) (570 \times 10^{-24}) (2 \times 10^{14})$

$Q = 265.6 \frac{W}{cm^2}$

match N_f for UO_2 $x = \text{enrich}$

$n(UO_2) = 235x + 238(1-x) + 32 = 270 - 3x$

$10.97 \frac{g}{cc} \frac{(mole)}{270-3x} \frac{6.022 \times 10^{23}}{1 mole} + \frac{14}{UO_2} x = 7.19 \times 10^{21}$

$\frac{x}{270-3x} = 1.08 \times 10^{-3}$

$x = -3.26 \times 10^{-3} x + 0.294$

$x = 0.293$ 29.3%

$$4) \quad LHR\left(\frac{z}{z_0}\right) = LHR^0 \cos\left[\frac{\pi}{2\delta}\left(\frac{z}{z_0} - 1\right)\right]$$

$$a) \quad z_0 = 1.5 \text{ m} \quad LHR^0 = 150 \text{ W/cm} \quad \delta = 1.1$$

$$LHR(z = 1.8) = 150 \times \cos\left[\frac{\pi}{2(1.1)}\left(\frac{1.8}{1.5} - 1\right)\right]$$

$$= \underline{143.9 \text{ W/cm}}$$

$$b) \quad \Delta T_{cool} = \frac{2\delta}{\pi} \frac{z_0 LHR^0}{m c_p} \left(\sin \frac{\pi}{2\delta} + \sin\left(\frac{\pi}{2\delta}\left(\frac{z}{z_0} - 1\right)\right) \right)$$

$$\frac{\pi}{2\delta} = 1.43$$

$$\Delta T_{cool}(\text{water}) = \frac{1}{1.43} \frac{(150)(150)}{(0.22)(4200)} \left(\sin(1.43) + \sin(1.43(z-1)) \right)$$

$$\Delta T_{cool}(\text{water}) = \underline{33.7 \text{ K}}$$

$$\Delta T_{cool}(N_2) = \frac{1}{1.43} \frac{(150)(150)}{(0.12)(1404)} \left(2 \times \sin(1.43) \right)$$

$$\Delta T_{cool}(N_2) = \underline{184.9 \text{ K}} \quad \leftarrow$$

$$5) \quad y(t_0) = 6 \quad \frac{dy}{dt} = 4t - 3t^2 \quad \Delta t = 0.33 \\ t_0 = 1 \quad t \rightarrow 2$$

Forward $y_{n+1} = y_n + \Delta t y'_n$

$$t_1 = 1.33 \quad y_1 = y_0 + \Delta t y'_0 = 6 + 0.33(4(1) - 3(1)^2) = 6.33$$

$$t_2 = 1.67 \quad y_2 = y_1 + \Delta t y'_1 = 6.33 + 0.33(4(1.33) - 3(1.33)^2) \\ = 6.334$$

$$t_3 = 2 \quad y_3 = y_2 + \Delta t y'_2 = 6.334 + 0.33(4(1.67) - 3(1.67)^2) \\ = \underline{5.78}$$

Backward $y_{n+1} = y_n + \Delta t y'_{n+1}$

$$t_1 = 1.33 \quad y_1 = y_0 + \Delta t y'_1 = 6 + 0.33(4(1.33) - 3(1.33)^2) = 6.004$$

$$t_2 = 1.67 \quad y_2 = y_1 + \Delta t y'_2 = 6.004 + 0.33(4(1.67) - 3(1.67)^2) \\ = 5.447$$

$$t_3 = 2 \quad y_3 = y_2 + \Delta t y'_3 = 5.447 + 0.33(4(2) - 3(2)^2) \\ = \underline{4.127}$$

- ⑥. Fertile: can be converted into fissile via neutron capture
 Fissile: capable of fission w/ any neutron energy
 Fissionable: Capable of undergoing fission w/ high E neutron capture

- ⑦ Low melting point
 Anisotropic thermal expansion
 Anisotropic irradiation growth
 Poor behavior in water
 Heterogeneous phase mixture
 Extreme Swelling

- ⑧ Relative volume of fuel compared to maximum possible volume inside cladding $\frac{r_1^2}{r_2^2}$
 Necessary because of inevitable fuel swelling

- ⑨. - We enrich U to increase N_f (fissile atom number density), to have higher heat generation rate, thus higher power output of fuel.
- UF_6 is the compound that undergoes enrichment
 - Spinning cylinder filled w/ UF_6
 - heavier mass of $U^{238}F_6$ preferentially moves to the outside of the canister due to centrifugal forces
 - enriched $U^{235}F_6$ withdrawn from center of canister

(10) Mo + Xe

Mo has $Z=42$ and $A=96$

Xe has $Z=54$ and $A=131$

- Double hump fission product distribution has peaks at around $A \approx 95$ and $A \approx 135$
- other examples valid

(11) Finite difference

Finite volume

Finite element

FEM are used in S-O-T-A simulations because of the flexible geometry, ability to handle any boundary condition, and continuous representation of quantities of interest