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NULE HQT

Exam # 2

Q1) Solution:

-10, 15/25

-4, no discussion of temperature

of cladding

- 1) Quick change in a gap, due to thermal expansion.
Point 1 is when gap start changing due from initial stage due to the thermal expansion.
- 2) Increase in gap due to densification. In this point we will start seeing the increase in cladding gap due to densification.
- 3) Gap getting smaller and smaller and when it finally gets to contact at point 4.
-2 Why is the gap getting smaller?
- 4) Gap initially comes to contact with pellet and the cladding. ~~the cladding~~ Gap start getting nearer and near to contact with fuel pellet.
-2, fission gas release cause T increase before gap closure
- 5) Gap started getting connected and we started getting PCI. (Pellet cladding interaction).

-2, T increases due to fuel k decrease with burnup

92) Solution:

-29, 1/30

avg: grain size = 8 microns:

volumetric neutron flux = 2.0×10^{13} fission / $\text{cm}^3 \cdot \text{s}$.

Assume: uniform temperature = 900°C .

a) To find = fission gas diffusion coefficient = ?

-5

is at
100

b) $\gamma = 0.3017$ To find = # of gas atoms / cm^3 released.
 $t = 2$ years.

of gas atoms released after 2 years is given by.

$$N_{\text{released}} = f N_{\text{gas}}$$

-9

-15

Q3) Solution:

$T = 600\text{K}$, for 1 year

-11, 19/30

Initial thickness = 0.6mm .

critical time for transition =

$$t^* = 6.62 \times 10^{-7} \exp\left(\frac{11949}{600\text{K}}\right) = 295 \text{ days.}$$

critical thickness for transition.

$$\delta^* = 5.1 \exp\left(\frac{-530}{600}\right) = 2.04 \mu\text{m.}$$

oxide thickness after 1 year.

$$\delta(\mu\text{m}) = \delta^* + K_L(t - t^*)$$

$$K_L = 7.48 \times 10^6 \exp\left(\frac{-12500}{600}\right) = 6.7 \times 10^{-3}$$

$$\delta = 2.04 + 6.7 \times 10^{-3} (365 - 295)$$

$$\delta = 2.509 \mu\text{m.}$$

total weight gain (mg/dm^2)

$$\delta(\mu\text{m}) = \frac{W \left(\frac{\text{mg}}{\text{dm}^2} \right)}{14.7}$$

$$W = 2.509 \times 14.7$$

$$W = 36.88 \text{ mg}/\text{dm}^2$$

\therefore The oxide weight gain is mg/dm^2 after this

$$\text{time} = \boxed{36.88 \frac{\text{mg}}{\text{dm}^2}}$$

b) To find ZIRLO wall thickness after this time = ?

$$\text{initial} = 0.6 \text{ mm}$$

$$\text{after transition} = 2.509 \text{ } \mu\text{m}$$

-2, Metal lost = oxide thickness/1.56
-2, ZIRLO thickness is lost, not gained

$$\begin{aligned} \therefore \text{Total} &= 0.6 \text{ mm} + 2.509 \text{ } \mu\text{m} \\ &= 600 \text{ } \mu\text{m} + 2.509 \text{ } \mu\text{m} \\ &= \boxed{602.509 \text{ } \mu\text{m}} \text{ or } \underline{\underline{0.602509 \text{ mm}}} \end{aligned}$$

c) Solution:

Assume hydrogen pickup fraction = 15%.

To find = weight ppm of hydrogen in cladding after 1 year.

$$\delta_{\text{ZrO}_2} = \frac{W(\text{mg/dm}^2)}{14.7} = W = 36.88 \text{ mg/dm}^2$$

$$N_o = \frac{W/N_{\text{av}}}{M_o} = \frac{0.036 \times 0.6022 \times 10^{24}}{16} = N_o = 1.388 \times 10^{21} \text{ } \overset{21}{\text{oxygen}} \text{ } \overset{\text{hydrogen}}{\text{atoms/dm}^2}$$

$$N_H = 2N_o = 2 \times 1.388 \times 10^{21} = 2.77 \times 10^{21} \text{ hydrogen atoms/dm}^2$$

$$N_{H-Zr} = 2fN_o = 2 \times 0.15 \times 1.388 \times 10^{21} = 4.16 \times 10^{20} \text{ hydrogen atoms/dm}^2$$

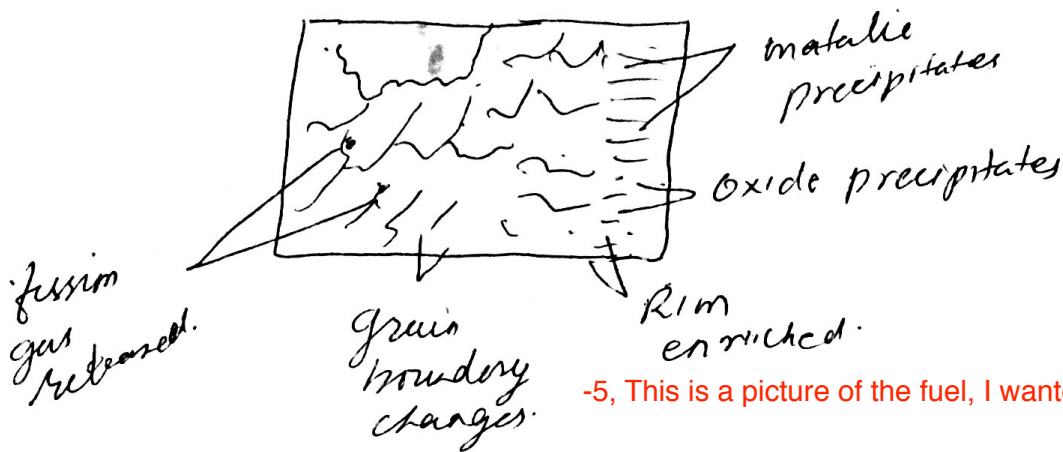
$$W_H = \frac{N_{H-Zr} M_H}{N_{\text{av}}} = \frac{4.16 \times 10^{20} \times 1}{0.6022 \times 10^{24}} = 6.9 \times 10^{-4} \text{ g of H in the cladding.}$$

$$C_H = \frac{W_H}{W_{\text{Zr}}} = \frac{W_H}{\rho_{\text{Zr}} V_{\text{Zr}}} = \frac{W_H}{\rho_{\text{Zr}} (t - \frac{\delta_{\text{ZrO}_2}}{\rho_{\text{OR}}})_A}$$

$$C_H = \frac{6.9 \times 10^{-4}}{6.5 \left(600 - \frac{2.50}{1.56} \right) (10 \times 10 \times 10^{-4})} = \boxed{1.77 \times 10^{-5} \text{ wt. ppm}}$$

-2, Easier to use eqn, answer should be 17.87 wt.ppm

d) Solution:



-5, This is a picture of the fuel, I wanted a picture of the cladding

Q4) Solution:

-4.5, 10.5/15

a) In Reactivity Initiated Accident (RIA) due to the large and rapid reactivity insertion; fission rate and fuel temperature rapidly increases. Due to change in fuel temperature, fuel expands in milliseconds and hits the cladding. The sudden change or jump in temperature also causes a pressure jump in fission gas which results in cracking on cladding, causing it to break.

→ But in loss of coolant accident (LOCA), the fuel and cladding experiences: decrease in temperature ^{and} decrease in coolant pressure.

-2, Both have high temperatures in the fuel

b) The similarities between the fuel and cladding behavior in a RIA and a LOCA is that both the accident will cause the fuel and cladding failure in nuclear reactors. In both cases cladding ~~temperature and fuel temperature~~ and fuel have high chances of bursting of clad and fuel breaking into pieces.

c) One potential accident tolerant fuel concept is concept of improving fuel properties. If we ~~increase~~ improve fuel properties than:

- Lower fuel operating temperature.
- minimized cladding internal oxidation
- minimized fuel relocation / dispersion.
- Higher fuel melt temperature.

-2.5, How would you improve fuel properties?

∴ If we improve fuel properties it can tolerate loss of active cooling for a considerably longer period.