## 1- Given parameters:

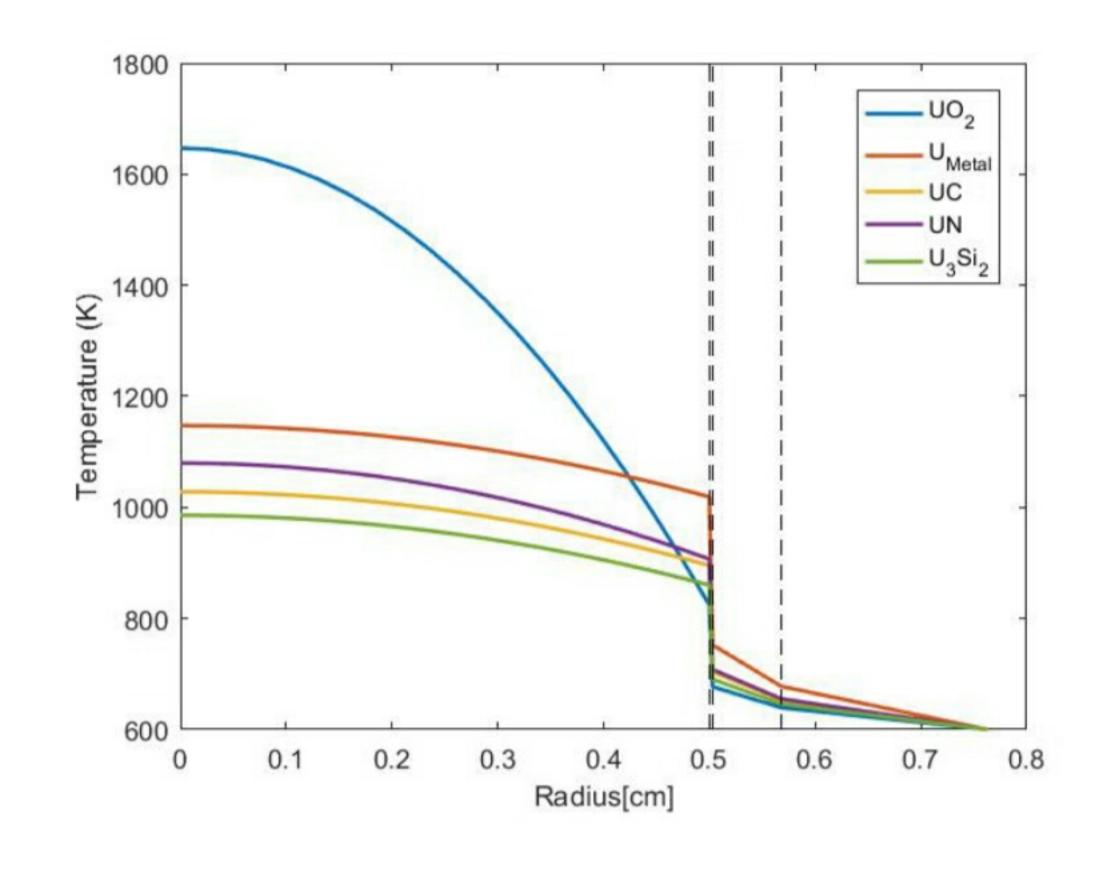
Fuel Material	kfven[W/cmk]	O[W/cm3]	Treeting [K]
Umetal	0.38	3 <b>9</b> 5	1405
UO2	0.03	779	3138
UC	0.25	531	3123
UN	0.2	553	3133
U <sub>3</sub> Si <sub>2</sub>	0.23	463	1938

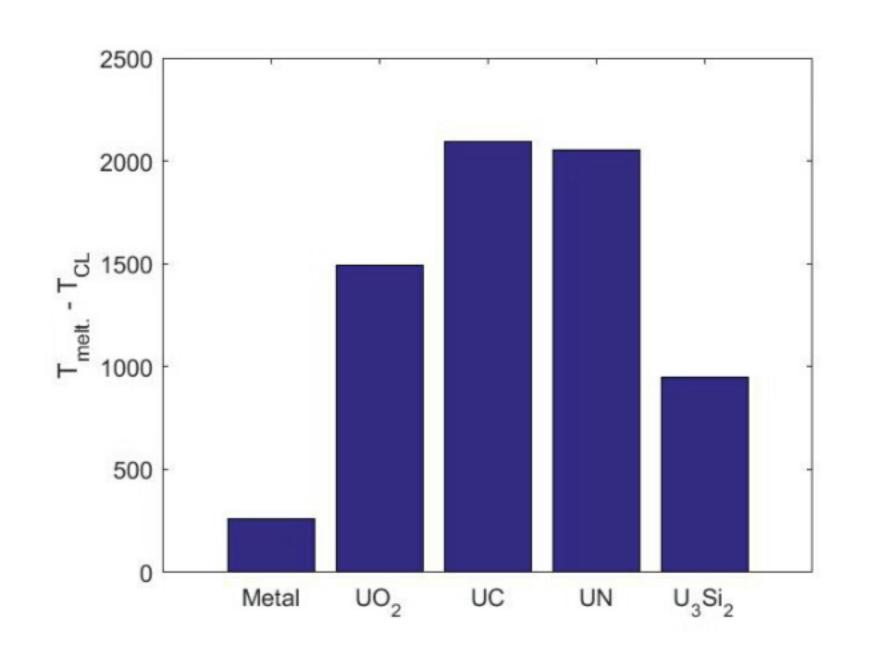
Values taken from slide 18 of Lec 2.

## Relevant equations:

$$T_{m}-T_{s}=\frac{Q}{4k_{f}}$$

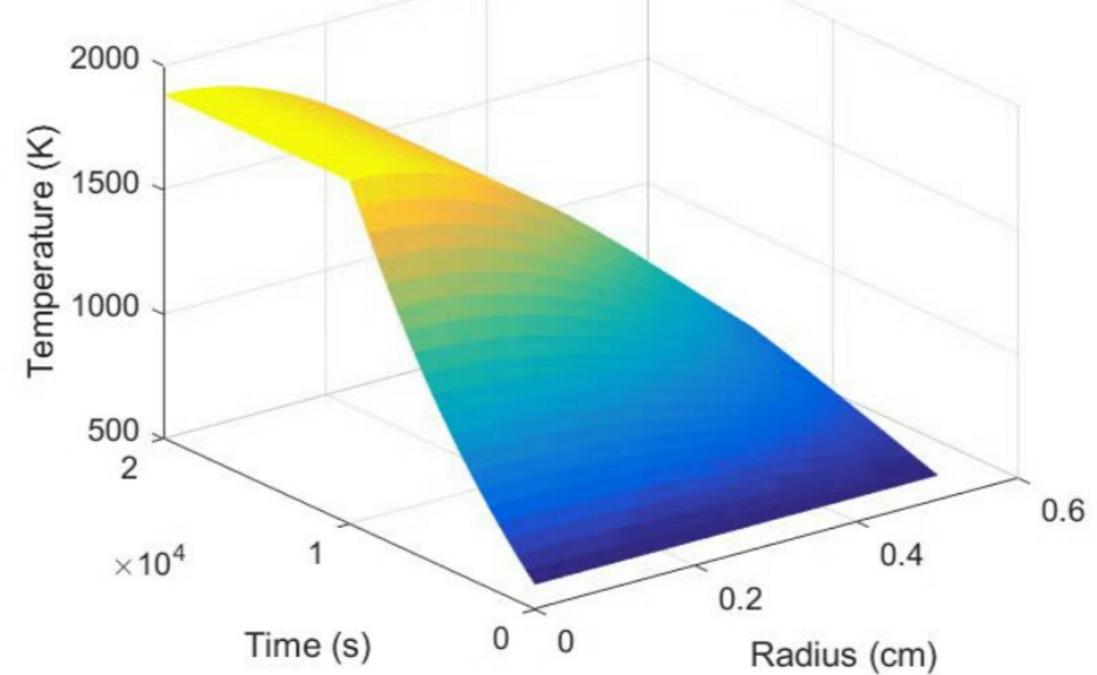
Fuel Material	Tmesty (K)	Tooler kne [K]
Umetal	1405	1147
UO2	3138	1647
UC	3123	1028
UN	3 133	( <b>68</b> 0
U <sub>3</sub> Si <sub>2</sub>	1938	986

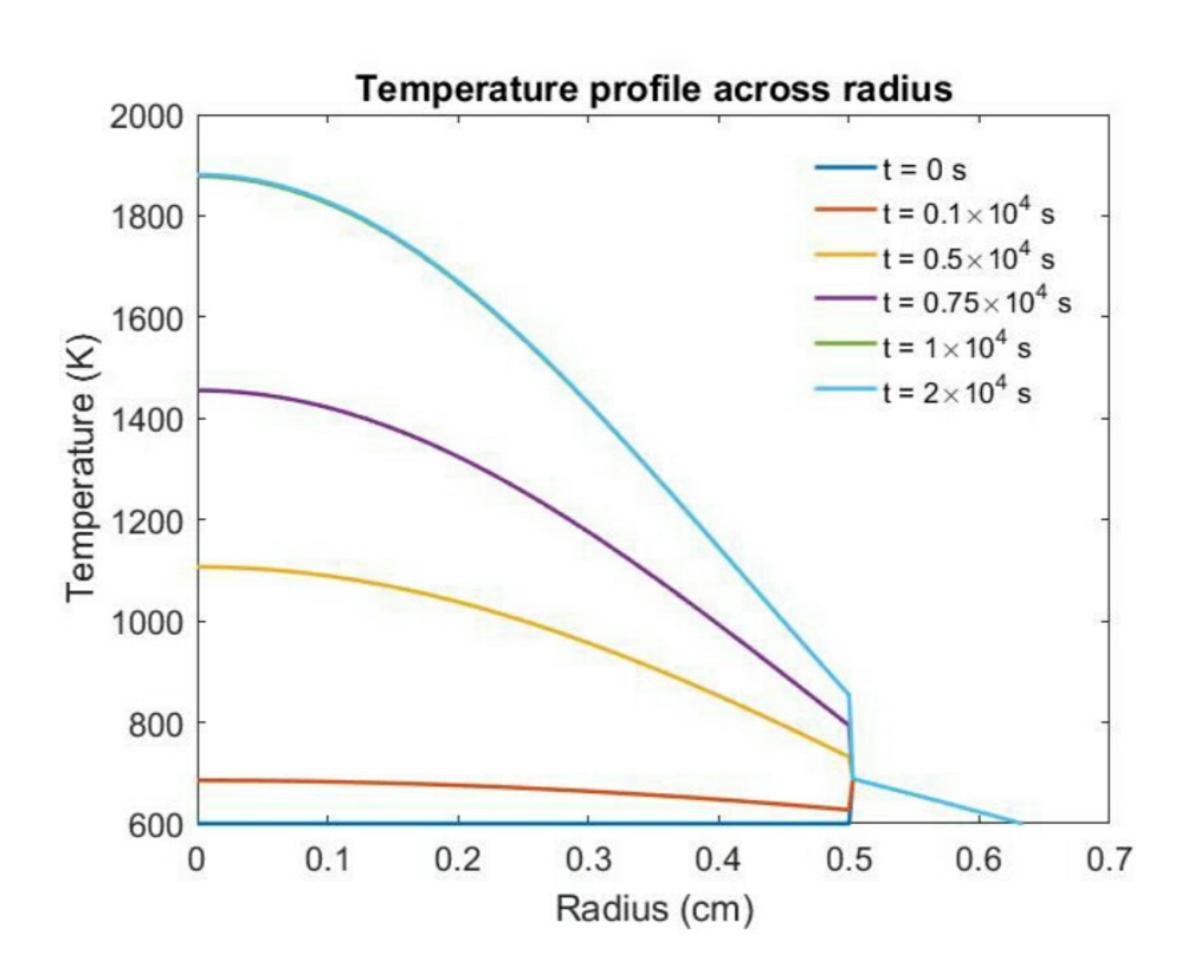




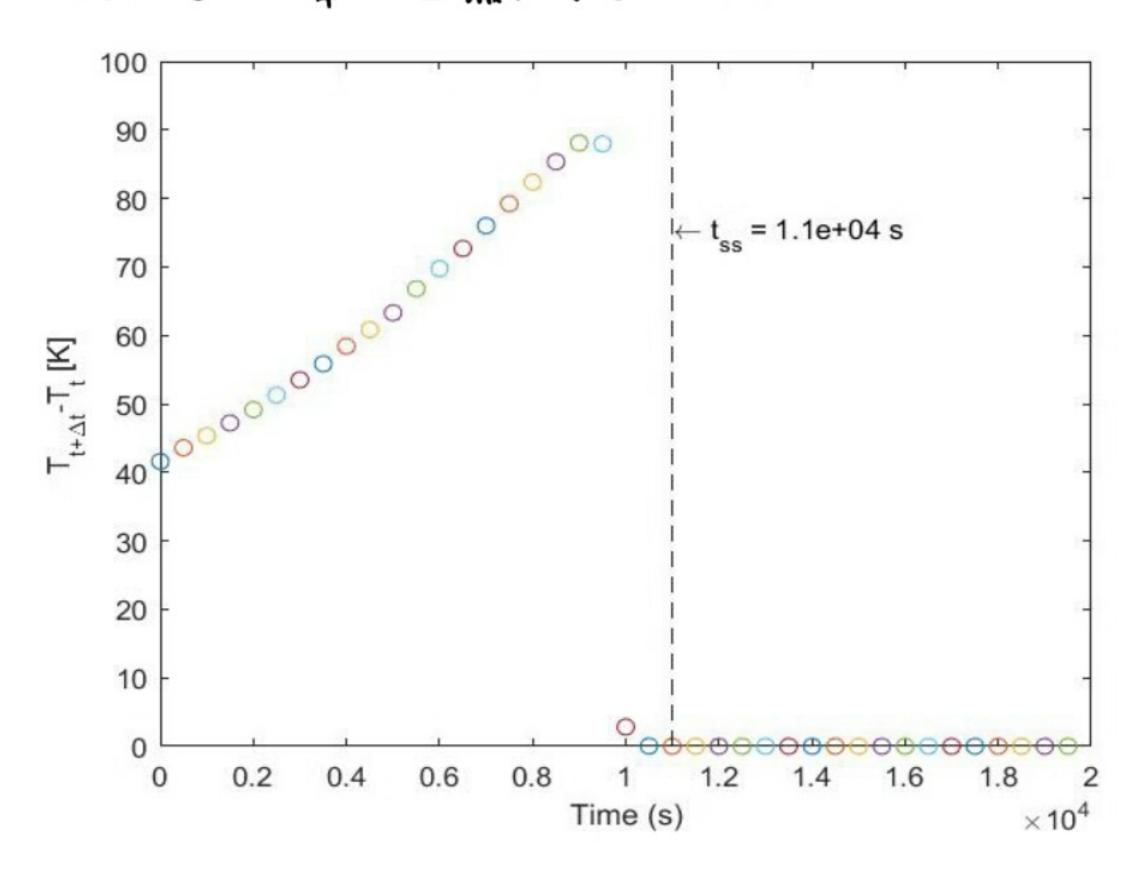
Calculated in HW-1

Temperature profile across radius with time

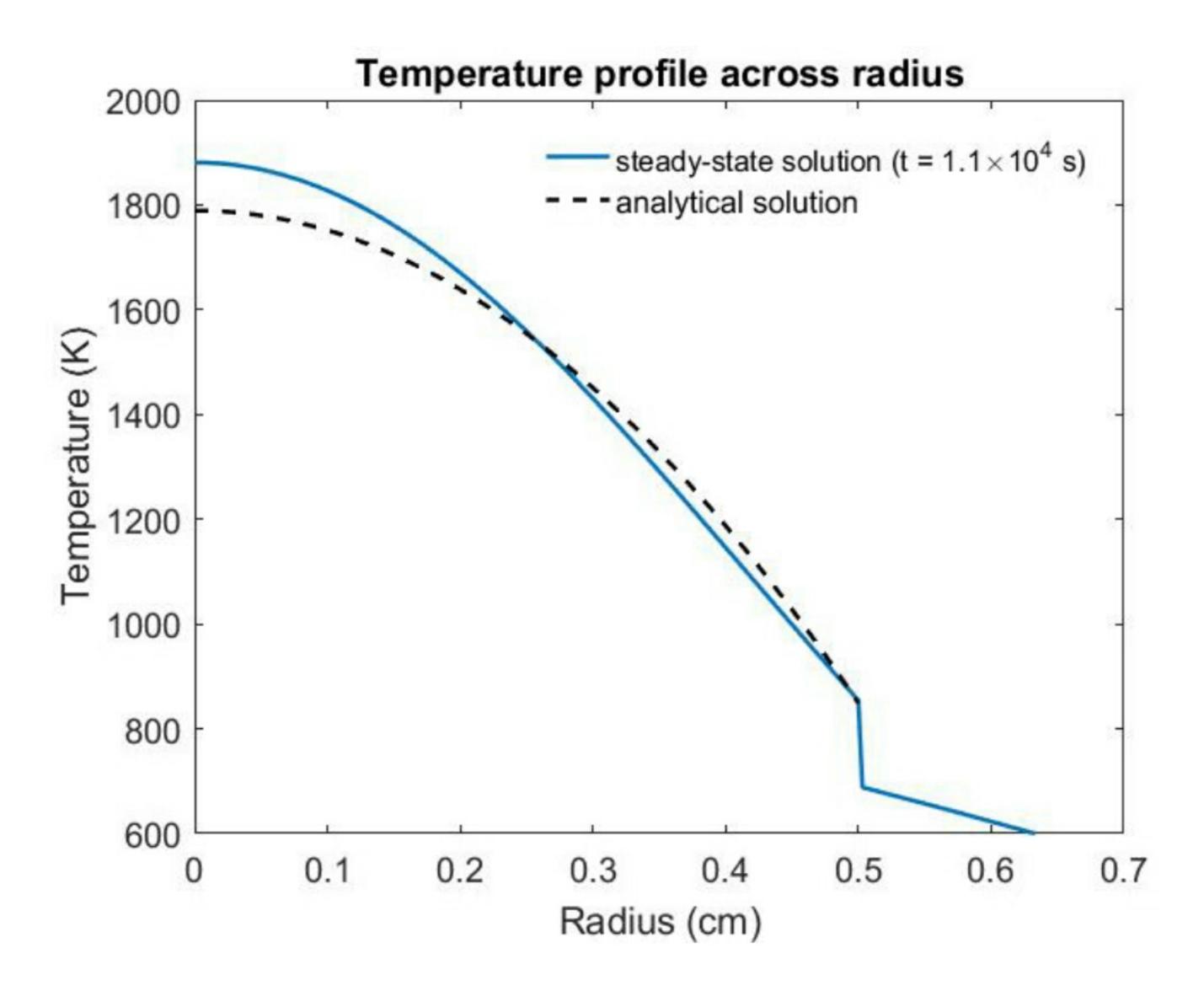




temperature difference was calculated for each time step. Steady-state solution was obtained if DTmox < & = 0.001.

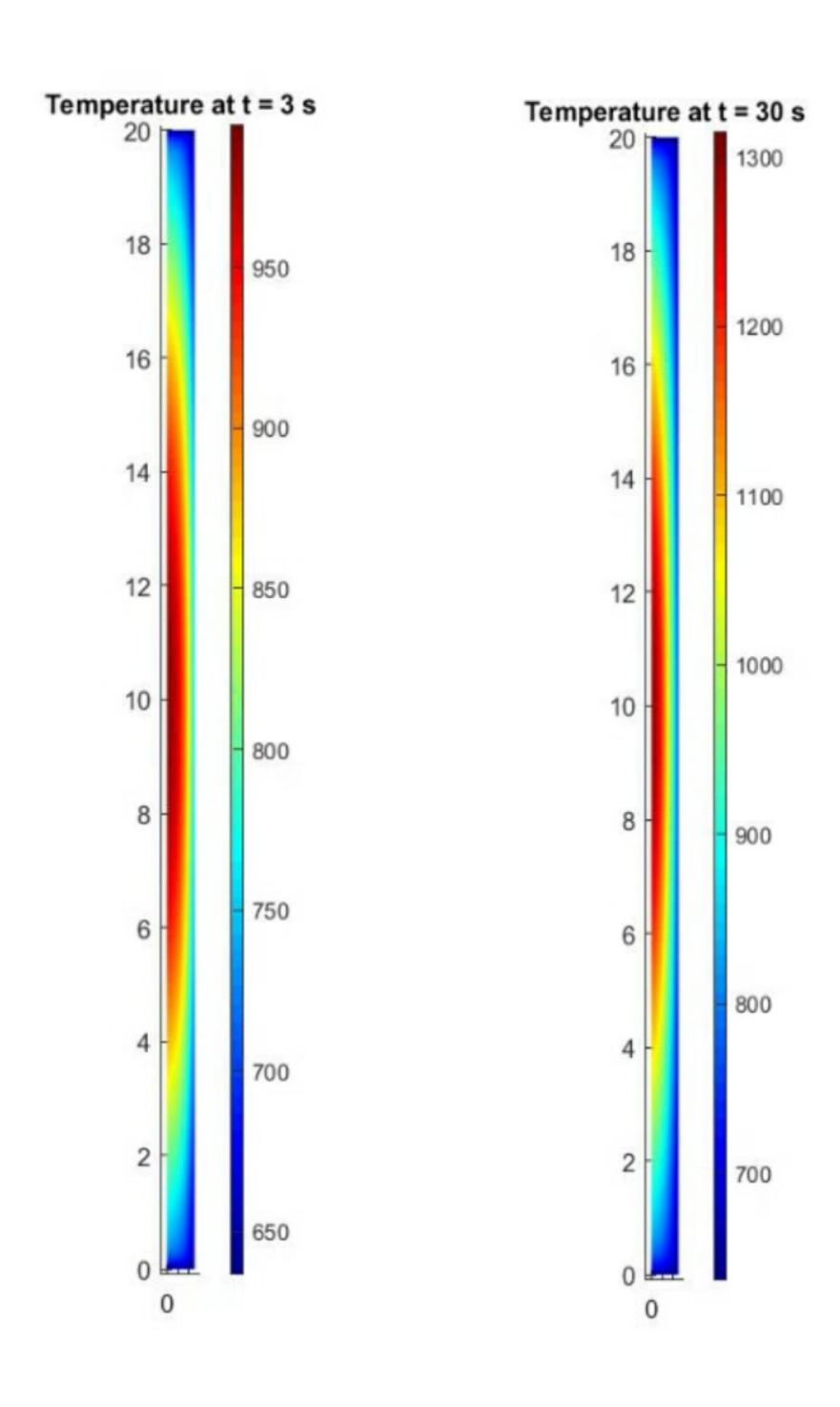


c )



Analytical solution and steady-state solution show slight variation. The main reason of this difference is the usage of constant k in analytical solution.

0-3)



We can rewrite this relation as fellows!

By using the results obtained in O-1, we can calculate DTMox for each first material:

