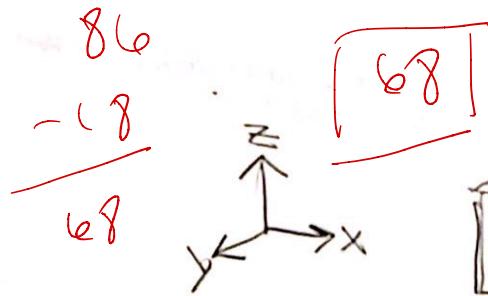


Late - 18



12/12

i.

① $\frac{\partial}{\partial x} T(x_0) = 0$, $x_0 = 0$
② $x_1 = x$, $T(x_1) = T_1$

CARTESIAN
COORDINATES

$$\frac{\partial^2 T}{\partial x^2} \left(k \frac{\partial T}{\partial x} \right) + Q = 0$$

Assume:

- * $k = \text{constant}$ i.e. $k \neq \text{func of } x$
- * Temp does not vary along y, z (i.e. symmetric)
- * Steady state $\Rightarrow \frac{\partial T}{\partial x} = 0$
- * Q is invariant with x (i.e. constant)
- * fuel is rectangular or box shaped

$$\frac{\partial^2 T}{\partial x^2} \left(k \frac{\partial T}{\partial x} \right) = -Q$$

$$\int \frac{\partial^2 T}{\partial x^2} \left(k \frac{\partial T}{\partial x} \right) dx = - \int Q dx$$

$$k \frac{\partial T}{\partial x} = -Qx + C_1$$

Using BC #1 $\rightarrow C_1 = 0$

$$\int k \frac{\partial T}{\partial x} dx = \int -Qx dx$$

$$kT(x) = -\frac{Qx^2}{2} + C_2$$

use BC #2 $\rightarrow T(x_1) = T_1$

$$kT_1 = -\frac{Qx_1^2}{2} + C_2$$

$$C_2 = kT_2 + \frac{Qx_1^2}{2}$$

$$kT(x) \stackrel{SO}{=} -\frac{Qx^2}{2} + kT_2 + \frac{Qx_1^2}{2}$$

$$k(T(x) - T_1) = \frac{Q}{2} [x_1^2 - x^2]$$

$$T(x) = \frac{Q}{2k} [x_1^2 - x^2] + T_1$$

10/18

② Calc centerline T & $T(r=0.4\text{ cm})$

$$k_{coat} = 0.05 \frac{\text{W}}{\text{cm}\cdot\text{K}}, k_{clad} = 0.15 \frac{\text{W}}{\text{cm}\cdot\text{K}}, k_{gap} = 0.25 \frac{\text{W}}{\text{cm}\cdot\text{K}}$$
$$k_{fuel} = 0.005 \frac{\text{W}}{\text{cm}\cdot\text{K}} \quad K_F = 0.5 \frac{\text{W}}{\text{cm}\cdot\text{K}}$$
$$h_{cool} = 5.5 \frac{\text{W}}{\text{cm}^2\cdot\text{K}}, 1\text{ cm} = 0.01\text{ m}$$
$$\overline{T}_{coo} = 800\text{ K}$$
$$Q = 400 \frac{\text{W}}{\text{cm}^3}$$

$$T_{coat} = T(r_4) = \frac{Q}{2h_{cool}} R_{fuel} + \overline{T}_{coo}$$

$$T_4 = \frac{400 \frac{\text{W}}{\text{cm}^3} (0.6)}{(2 \times 5.5 \frac{\text{W}}{\text{cm}^2\cdot\text{K}})} + 800\text{ K}$$

$$\underline{T_4 = 821.82\text{ K}}$$

$$T_3 = \frac{Q t_{coat} R_f}{2 k_{coat}} = \frac{(400)(0.101)(0.6)}{2 \times 0.05} + T_4$$

$$\boxed{T_3 = 845.82} \quad \checkmark$$

$$T(r_2) = T_2 = \frac{Q t_{clad} R_f}{2 k_{clad}} + T_3$$

$$= 847.32\text{ K}$$

Some thing
here, but
you didn't show all
 $\Delta T_{cl-co} = 40\text{ K}$

$$T_1 = T(r_i) = \frac{Q}{2k_{fuel}} R_f + T_2$$

also here something off, not sure

$$\Delta T_f < c_1 = 96 K$$

$$T_1 = 1007.32 K$$

$$T_0 = \frac{Q}{4k_{fuel}} R_{fuel}^2 + T_1$$

$$T_0 = 8207.32 K$$

at $r = 0.4 \text{ cm}$

$$T(0.4 \text{ cm}) = \frac{Q (0.6^2 - 0.4^2)}{4 \times k_{fuel}} + T_1$$

$$= 5007.32 K$$

procedure correct

Calculating heat generation rate for a given fuel

~ about 200 MeV of energy is available due to a fission (210 MeV)

section of the target nuclide (tabulated)

σ , Temp, fuel, etc.

σ is the

heat system

$$0.145 \text{ W/cm K}$$

$$8/4$$

$$\textcircled{3} \quad k_{\text{fuel}} = 145 \text{ W/mK} \quad E = 0.195$$

$$\rho_{\text{fuel}} = 15.67 \text{ g/cm}^3$$

$$S_i = 28 \text{ mm}$$

$$\sigma_f = 5.7 \times 10^{-22} \text{ cm}^2$$

$$\text{a) } \dot{\phi} = 2 \times 10^{12} \text{ n/cm}^2 \cdot \text{s}$$

$$Q = E_f N_f \sigma_f \frac{\dot{\phi}}{N_A E}$$

$$N_f = \frac{(235 E + 238(1-E))}{2 \times 16}$$

$$= \frac{15.67 \times 6.023 \times 10^{23} \times 0.195}{269.77}$$

$$= 6.82 \times 10^{21} \frac{U^{235} \text{ atoms}}{\text{cm}^3}$$

$$E_f = \frac{200 \times 10^6 \text{ eV}}{\text{fission}} \times 1.602 \times 10^{-19} \text{ J/eV}$$

$$= 3.228 \times 10^{-11} \text{ J}$$

$$Q = 250.6 \text{ W/cm}^3$$

b) Using excel to change enrichment

$$\rho = 10.979 \text{ g/cm}^3$$

$$E = 0.2784 \text{ or } 27.84\%$$

→ need to solve for x w/ same N_f
System work

Calculating heat generation rate for a given fuel

.. of energy is available due to a fission (210 MeV
insulated)

Q) $LHR(z = 1.8 \text{ m})$

$\downarrow z/z$

LHR is not per K

$$LHR^o = 150 \text{ W/cm}^2 \text{ K} = 1.5 \times 10^4 \text{ W/m}^2 \text{ K}$$

$$z_0 = \frac{L}{2} = 1.5 \text{ m}$$

$$LHR^o = LHR(1.5 \text{ m})$$

$$LHR(1.8/1.5) = (1.5 \times 10^4 \text{ W/m}^2 \text{ K}) \left(\cos\left(\frac{\pi}{2.2} \left(\frac{1.8}{1.5} - 1\right)\right) \right)$$

$$\boxed{LHR(1.8 \text{ m}) = 1.44 \times 10^4 \text{ W/m}^2 \text{ K}}$$

assuming radians

b) $\Delta T = \frac{1}{1.43} \frac{z_0 LHR^o}{m \text{ CPW}} (\sin(1.43) + \sin(1.43))$

$$\Delta T_i = \frac{17.3}{16.86} \times \left\{ \begin{array}{l} \sin(1.43) \\ \sin(1.43) \end{array} \right\}$$

$$\Delta T_{ii} = \left\{ \begin{array}{l} 93.39 \\ = 92.47 \end{array} \right\}$$

$$\boxed{\Delta T_{ii} > \Delta T_i}$$

would prefer
you show
work

⑤ $y(t_0) = 6 = y(1)$ $t_n = 2$
 $y'(t) = 4t - 3t^2$ 16/16
 Fwd Euler
 $y(t_1) = y(t_0) + \Delta t \cdot y'(t_0)$ ✓

$t_0 = 1$	$y(t_0) = 6$
$t_1 = t_0 + \Delta t = 1.33$	$y(t_1) = 6.33$
$t_2 = t_1 + \Delta t = 1.66$	$y(t_2) = 6.3343$
$t_3 = t_2 + \Delta t = 1.99 \approx 2$	$y(t_3) = 5.798$

$$y(t_1) = 6 + 0.33 \times (4(1) - 3(1)^2)$$

$$y(t_2) = y(t_1) + 0.33 \times (4 \times (1.33) - 3(1.33)^2)$$

$$y(t_3) = y(t_2) + 0.33 (4 \times 1.66 - 3(1.66)^2)$$

$$y(t_4) = y(t_3) + 0.33 (4 \times 1.99 - 3 \times 1.99^2)$$

Backward Euler

$$y(t_1) = y(t_0) + \Delta t * y'(t_1) \quad \checkmark$$

$$\text{or} \\ y_{n+1} = y_n + \Delta t * y'_n$$

$$y_1 = 6 + 0.33 \times (4 \times 1.33 - 3 \times (1.33)^2)$$

Backward

$$y_0 = 6$$

$$y_1 = 6.0043$$

$$y_2 = 5.467$$

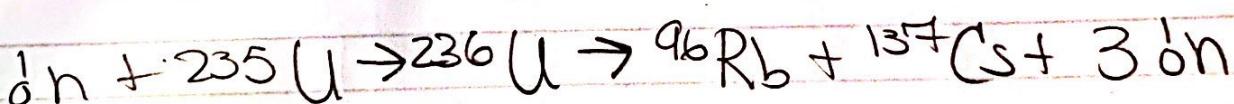
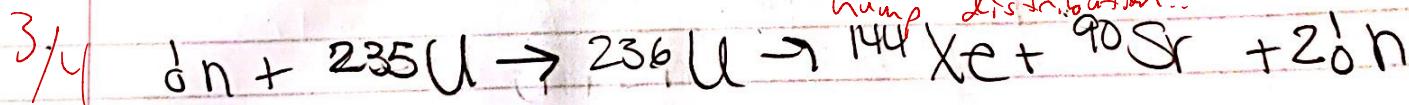
$$y_3 = 4.174$$

- ⑥ ~~4/5~~ FISSIONABLE - can undergo fission ^{only w/ high E} high or low neutron energy
 FERTILE can become fissile ~~via~~ transmutation
 fissile - able to undergo fission w/ low energy neutron ✓

- ⑦ ~~4/4~~ ① many phases α, γ, β ✓
 ② α -U shrinks in 1-D & expands along other 2-directions (thermally unstable) ✓

- ⑧ ~~3/4~~ smear density is ratio of fuel mass to total volume w/in fuel rod. It is used to compare the relative swelling of fuels → required s/c of swelling

- ⑩ ^{96}Rb & ^{144}Xe ✓ wanted something about double hump distribution..



- ⑪ 3 space discretization methods

~~6/8~~ Finite element ✓ → used in S.O.T.A. codes

Finite difference ✓

Finite volume ✓

all 3 used in state of art codes

Finite difference was the 1st method developed

Finite volume is very useful when flux is approx. equal @ interfaces

Finite ~~volume~~^{element} is most flexible & useful in complex geometries but very intricate so not used when not needed

- ⑨ We enrich U to increase fissile atoms in fuel. UF₆ is used in centrifuge process. The difference in isotopic mass allows separation by centrifugal force i.e. ²³⁸U gets pushed to outside of centrifuge container & allows ²³⁵U to be drawn out from center.