Exam 3: NE533: Nuclear Fuel Performance

Show your work. Point values are indicative of the depth of expected response. Check units.

1. A ZIRLO cladding tube is in reactor at 625 K for 400 days. The initial wall thickness is 500 μm.

a) Estimate the oxide thickness after this time? (8 pts)

b) Assuming the hydrogen pickup fraction is 18%, what is the weight PPM of hydrogen in the cladding after one year? Assume PBR = 1.56,  $\rho_{Zr}$  = 6.5 g/cc,  $\rho_{ZrO2}$  = 5.68 g/cc. (8 pts)



2. Determine the total change in the fuel volume given:  $\alpha_{th}$ =11x10<sup>-6</sup>, fission rate = 3.5x10<sup>13</sup> fiss/cm<sup>3</sup>-s, T=1200 K, T<sub>ref</sub>=300 K,  $\Delta \rho_0$ =0.01, B<sub>D</sub>=5 MWD/kgU,  $\rho$ (UO2)=10.97 g/cc, t=85 days. (16 pts)

 $\mathcal{E}_{tot} = \sum_{i} \mathcal{E}_{i}$   $\mathcal{E}_{th} = \mathcal{A}\Delta T = 11.16^{6} \cdot (1200 - 300) = 9.9.16^{3}$ Ed = APO [e^{\left(\frac{\beta\left(n(0,01)\right)}{\left(\frac{\beta\left(0,01)\right)}{\left(\frac{\beta\left(0,01)\right)}}\right] \beta\left(\frac{5}{950} = 5,3\right)\beta\beta\right)F.m.

2 45:10:22 Esfp = 5.577.10-2 PB = 0.00642 EGFP = 1,98.10<sup>-28</sup> PB (2890-T) 1.73 -0,0 162(2800-T) -17.888 4.42.10<sup>36</sup> 5.5346.10<sup>-12</sup> 0,1287

16/le

3. What is the total creep in a zirconium cladding given a von mises stress of 200 MPa, a temperature of 600 K, a LHR of 150 W/cm, and a time of 1.5 years? Use nominal values from lactures (12 pts)

/0/

from lectures. (12 pts)

$$Sm = 200 \text{ MB} \quad T = 600 \text{ k} \quad LHR = 160$$

$$t = 1.5 \text{ yr} \quad Q = 2.7 \cdot 10^{5} \quad n = 8 \text{ Reg. 3H}$$

$$Ao = 318 \cdot 18^{4} \cdot 5^{2} \cdot G = 4.1 \cdot 10^{6} - 2.3 \cdot 10^{7} = 2.93 \cdot 10^{7} \text{ MB}$$

$$Ess = Ao \left(\frac{200}{4}\right)^{6} e^{-4} \left(\frac{200}{4}\right)^{6} = 1.48 \cdot 10^{10} \text{ s}$$

$$EIrr = 2.87 \cdot 10^{10} \cdot 5^{2}$$

$$Etot = (2.87 + 1.48) \cdot 10^{10} \cdot 36.5 \cdot 1.5 \cdot 24 \cdot 36.00$$

$$Etot = 0.007 = 7 \quad 0.7 \cdot 70$$

$$(2.87 + 1.48) \cdot 10^{4} \cdot 10 = 4.35 \cdot 10^{4} \cdot 10$$

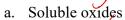
$$1.5 \cdot 365 \cdot 24 \cdot 3600 = 47304000$$

$$47304000 \cdot 4.35 \cdot 10^{4} \cdot 10 = 0.020577$$
etot = 2.06%

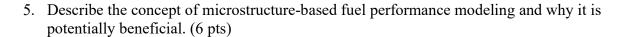
My mistake.

$$- What value \left( 3 + C_{0} \right)$$

4. What are the five types of fission products that form in the fuel? (5 pts)



- b. Insoluble oxides
- c. Metals
- d. Volatiles
- e. Noble Gases





Examination of the microstructural behavior such as grain growth change in thermal conductivity, intragranular porosity, Precipitated fission products, etc. Is being conducted to produce mechanistic models of the behavior of the fuel over the course of a fuel cycle to create more accurate prediction capabilities.

The potential benefits are many, but a reduction of the necessary experimentation is a large one.

6. List three benefits of using Zr cladding. (6 pts)

- a. Resistance to void swelling
- b. Good thermal conductivity
- c. Available in large quantities

7. Why does metallic fuel undergo constituent redistribution? (5 pts)

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oes metallic fuel undergo constituent redistribution? (5 pts)

Redistribution occurs in metallic fuel more than oxide fuel because the metallic fuel substantially disturbs the constituent.

8. What are some of the key differences in MOX fuel compared to LWR fuel? Emphasize differences on in-reactor behavior/performance/environment. (8 pts)

a. Higher power density

b. Higher neutron flux intensity

c. Higher burnup

-higher temperatures

- relonstourtion

9. What are the four conditions that must be met for SCC? Briefly describe how each is met in PCI. (8 pts)

- a. Four Conditions
  - i. Corrosive environment
  - ii. Susceptible material
  - iii. Sufficient stress
  - iv. Sufficient time

