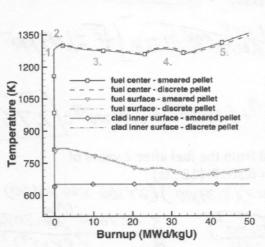
NucE 497 Fuel Performance Exam 2 covering modules 4 - 6

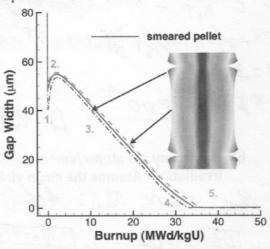
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-5, 20/25

Question 1 (25 points):

The temperature and gap width of a fuel pellet, as predicted by a fuel performance code, is shown below. Using the plots as your guide, determine what is currently occurring within the cladding, gap, and pellet at each number. Note that the numbers are at the same burnups on the two plots.





For each number, describe what is occurring in the cladding, gap, and pellet. Also, describe what features in the plots indicated these behaviors.

1. Classing! Instruct the small creep -0.5, thermal expansion in the cladding gap! The small expansion causes gap to become tess. pellet! thermen expension

2. Classing, Modration creep begins. gop! Increases in size due to Pellet Shrinking pellet! Shrinks and becomes more dense.

-0.5, creep has nothing to do with hydride formation

3. Classing; Irradiation and thermal creep may begin to cause hydride termotion

gop; tecreuses as fuel swells and clusting creeps

Pellet! Swelling begins. -3, No hydride cracking, T goes up due to fission gas release 4. Classing: creep continues, hydridethen back down due to continued swelling

gof: continues to decreuse and eventually becomes o Pellet: Swelling towards cluffing cent completly closes get.

5. classing : Mekes contact with fuel, cousing temp increase, gop! non existent.

Pellet! begins to pish on clushing and fromthe or crock

-1, PCMI does not cause the pellet to crack

Question 2 (30 points)

A fuel pellet with an average grain size of 8 microns is irradiated with a volumetric neutron flux of 2.0e13 fissions/(cm³ s). Assume the pellet is at a uniform temperature of 900 °C.

a) What is the fission gas diffusion coefficient at this temperature? (5 pts) where K = 8,617 X10 Da= 1,41×10-18 exp[-1,AeN/KoT] F=1,41×10-18 exp[-1,19eN] 2,0×10-30 F= 6,78×10-19 D=7,28×10-19+6.35×10-18+6,78×10-19=/7.76 ×10 b) How many gas atoms/cm3 are released from the fuel after 2 years of irradiation? Assume the chain yield y = 0.3017. (10 pts) Neleuse = + Ngos where, Ngus = YFt/x= (0,3617)(3,39×10") (2×365 ×24 ×3600) = 6,451 ×0 2,300 f=4\fra2 - 3Dt = 4\fra2 - 3Dt = 4\frac{7.76\text{x10}^48(6.3072\text{x10}^3)}{11 (8\text{x10}^4)^2} - \frac{3}{2} (7.76\text{x10}^4)^2 (8\text{x10}^4)^2 -0, Fdot = 2e13 f = 0.0624 - 0.00/147 = 0.06/253 -1, Check tau to determine which equation? Neleuse = (0,061253)(6,451×10") = 3,95×10 =3 c) After 2 years of irradiation, the pellet is removed from the reactor and from its cladding, venting all released gas. It is then moved to a furnace and annealed at 2000 °C. Estimate how long before 10% of the gas trapped in the $= 8.21 \times 10^{-13}$ $D = 2.27 \times 10^{-12} \text{ cm}^2$ -1, Math error pellet is released. How many gas atoms/cm³ will have been released during this time? (15 pts)

 $f = 6\sqrt{\frac{Dt}{Ha^{2}}} - \frac{3Dt}{a^{2}} = \frac{3D$

Nreleuse = fyft = (0,1) (0,3017) (3,39 X10") (246,04) & 2,52 X10" cm3)

-4, No more gas is produced in post irradiation annealing. Should be fraction of gas remaining from previous irradiation

A ZIRLO cladding tube is in reactor at 600 K for one year. The initial wall thickness is 0.6 mm.

a) What is the oxide weight gain in mg/dm² after this time? (10 pts)

$$\int = \int_{K_{2}}^{*} + K_{L}(t - t^{*})$$

$$K_{L} = 7.40 \times 10^{6} \exp\left[-\frac{12500}{500}\right] = 7.48 \times 10^{6} \exp\left[-\frac{12500}{600}\right] = 0.0067$$

$$\int_{L}^{*} = 5.1 \exp\left[-\frac{550}{5}\right] = 5.10 \times 10^{6} \exp\left[-\frac{12500}{600}\right] = 2.039 \text{ Mm}$$

$$\int_{L}^{*} = 6.62 \times 10^{6} \exp\left[\frac{11940}{500}\right] = 6.62 \times 10^{6} \exp\left[\frac{11940}{600}\right] = 295.01 \text{ days}$$

$$\int_{L}^{*} = (2.039) + (6.0067)(365 - 295) = 2.508 \text{ Mm}$$

$$\int_{L}^{*} = W_{L}^{*} = W_{L}^{*} = S(14.7) = 36.87 \frac{m_{d}}{4m^{2}}$$

b) What is the ZIRLO wall thickness after this time? (5 pts)

tg=tg+ d= 0,6mm + 2,508 Mm = [6.02,508 Mm]

-2, ZIRLO lost is oxide thickness / 1.56

Galking from PCMI Issues

a) What are the primary differences between a loss of coolant accident and a reactivity insertion accident, regarding the fuel and cladding behavior? (5 pts)

The primary difference in the Cludding is there in Local you have alpha to better transition as that temp increases, of for a LOCA relocation accours in the fuel where as RIA fuel dispersed occurs.

-2, RIA is much faster

-2, RIA increases Q while LOCA increases coolant T

b) What are similarities between the fuel and cladding behavior in a RIA and a LOCA? (5 pts)

Some of the Smillorities include the doldling Swelling and bursting, the fuel breaking into Pieces, fission gas release, Both cuise themas stresses on fuel cent clubbing.

c) List a potential accident tolerant fuel concept and describe how it could meet the primary goal of the accident tolerant fuel program. (5 pts)

Some potential accident toterant fuel concepts are to Change the Claffing muterical 1.2. I silican Corbite composite, advanced steel, or refractery meteurs. The Gentites of such materials white resilence to class frecture, uncreased steellists, higher lasting melting temperatures, and minamitany fuel to classing intercedion.

-2, Primary cladding benefit is reduced oxidation kinetics