

KEY

Relevant information:

Thermal neutron cross section for U-235: 570 barns

1 eV = 1.602E-19 J

Density of UO<sub>2</sub>: 10.97 g/cc

Si (Z=14, A=28)

Pay attention to units!

Total 105 pts

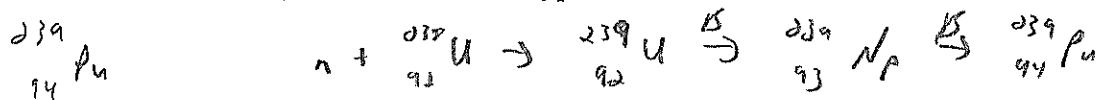
Explain the difference between fertile, fissile and fissionable. 5 pts

fertile: can absorb a neutron to become a fissile nuclide

fissile: can readily undergo fission w/ neutron absorption

fissionable: can undergo fission w/ absorption of high E neutrons

What is the secondary fissile element in typical commercial fuel? How is it formed? 4 pts



What does TRISO stand for? Identify the layers in a TRISO particle. Provide an example of a reactor that utilizes TRISO-based fuel. 5 pts

Tr: structural Isotropic

high temperature gas reactor

kernel

buffer

IPyC

SiC

OPyC

What is the compound that uranium is converted into for enrichment purposes? 2 pts



Which noble gas has a lower thermal conductivity, He or Xe? 2 pts

Xe

Outline and describe the entire fabrication process of the fuel. 10 pts

mining  $\rightarrow$  conversion  $\rightarrow$  enrichment  $\rightarrow$  pellets  $\rightarrow$  rods

- U ore is mined & processed to make  $U_3O_8$
- $U_3O_8$  is converted to  $UF_6$
- $UF_6$  is utilized to increase  $U^{235}$  fraction
- $UF_6$  converted back to  $UO_2$ ,  $UO_2$  powder compacted & sintered into pellets
- $UO_2$  pellets assembled into fuel rods/elements

Name two primary fission product species. Provide justification. 5 pts

Mo + Xe

Double hump fission product yield with  
peaks @  $A = 90$  +  $A = 135$

What is the role of cladding? 2 pts

primary containment of the fuel

What are the three ways that space is discretized for numerical solutions? 4 pts

finite difference

finite volume

finite element

$U_3Si_5$  is a uranium silicide fuel being considered for use in light water reactors. It has a thermal conductivity of  $12.5 \text{ W/(m-K)}$ , an enrichment of 5% and a density of  $8.97 \text{ g/cm}^3$ . Answer the following questions:

What is the heat generation rate for  $U_3Si_5$  given a neutron flux of  $3E13 \text{ n/cm}^2\text{-s}$ ? 10 pts

$$Q = \sigma N_f \phi E_f \quad N_f = 8.97 \frac{\text{g}}{\text{cm}^3} \frac{1 \text{ mol}}{853.5 \text{ g}} \frac{6.022 \times 10^{23}}{1 \text{ mol}} \frac{34}{145.5} \times 0.05 = 9.5 \times 10^{20} \frac{\text{atoms}}{\text{cm}^3}$$

$$\text{next, } U = 0.05 \times 235 + 0.95 \times 238 = 237.85$$

$$\text{next, } U_{3Si_5} = 3(237.85) + 5(28) = 853.55$$

$$Q = (570 \times 10^{24} \text{ cm}^3) (9.5 \times 10^{20} \frac{\text{atoms}}{\text{cm}^3}) (3 \times 10^{13} \frac{\text{n}}{\text{cm}^2\text{-s}}) (200 \times 10^6 \text{ eV}) (1.602 \times 10^{-19} \frac{\text{J}}{\text{eV}})$$

$$Q = 520 \frac{\text{J}}{\text{s-cm}^3} = \boxed{520 \frac{\text{W}}{\text{cm}^3}}$$

Given a fuel radius of 0.45 cm, what is the temperature drop over the fuel pellet? 6 pts

$$T_o - T_f = \frac{LHR}{4\pi K_f} \quad LHR = Q \pi R_p^2 = (520)(\pi)(0.45)^2 = 331 \frac{\text{W}}{\text{cm}}$$

$$T_o - T_f = \frac{331}{4\pi(0.125)} = \boxed{211 \text{ K}}$$

$$K = 12.5 \frac{\text{W}}{\text{m-K}}$$

$$K = 0.125 \frac{\text{W}}{\text{cm-K}}$$

What enrichment of  $UO_2$  would be required to obtain the same heat generation rate? 10 pts

$$N_f = 9.5 \times 10^{20} \frac{\text{atoms}}{\text{cm}^3} \quad \text{next, } U = 235x + (1-x)238 = 238 - 3x$$

$$\text{next, } UO_2 = 238 - 3x + 2(16) = 270 - 3x$$

$$9.5 \times 10^{20} \frac{\text{atoms}}{\text{cm}^3} = 10.97 \frac{\text{g}}{\text{cm}^3} \frac{1 \text{ mol}}{270 - 3x \text{ g}} \frac{6.022 \times 10^{23}}{1 \text{ mol}} \frac{14}{140} \times x$$

$$1.44 \times 10^{-4} = \frac{x}{270 - 3x} \rightarrow 0.0388 - 4.3 \times 10^{-4} x = x$$

$$x = 0.0388 \rightarrow \boxed{3.9\%}$$

Which coolant sees a larger change in outlet to inlet temperature? 1) water:  $C_p = 4200 \text{ J/kg-K}$ ,  $\dot{m} = 0.25 \text{ kg/s-rod}$ ,  $Z_0 = 1.5 \text{ m}$ ; 2) sodium:  $C_p = 1404 \text{ J/kg-K}$ ,  $\dot{m} = 1.2 \text{ kg/s-rod}$ ,  $Z_0 = 0.5 \text{ m}$ . (10 pts)

$$T_{\text{cool}} - T_{\text{cool}}^{\text{in}} = \frac{1}{1.2} \frac{Z_0 \text{ LHR}^0}{\dot{m} C_p} \left\{ \sin(1.2) + \sin\left[1.2\left(\frac{Z}{Z_0} - 1\right)\right] \right\}$$

$$Z = 2Z_0 \rightarrow \text{convert to cm}$$

$$2) \Delta T_{\text{cool}} = \frac{1}{1.2} \frac{150 \text{ LHR}^0}{(0.25)(4200)} \left\{ \sin(1.2) + \sin(1.2) \right\}$$

$$\Delta T_{\text{cool}} = 0.22 \text{ LHR}^0$$

$$2) \Delta T_{\text{cool}} = \frac{1}{1.2} \frac{50 \text{ LHR}^0}{(1.2)(1404)} \left( \sin(1.2) + \sin(1.2) \right)$$

$$\Delta T_{\text{cool}} = 0.046 \text{ LHR}^0$$

(#2)

Given a rod of 2 m in length and an  $\text{LHR}^0 = 150 \text{ W/cm}$ , what is the LHR at  $z=1.6 \text{ m}$ ? (6 pts)

$$Z_0 = 1 \quad \text{LHR}(z=1.6)$$

$$\text{LHR}\left(\frac{z}{Z_0}\right) = \text{LHR}^0 \cos\left[\frac{\pi}{2} \left(\frac{z}{Z_0} - 1\right)\right]$$

$$\text{LHR} = 150 \frac{\text{W}}{\text{cm}} \cos\left[\frac{\pi}{2} \left(\frac{1.6}{1} - 1\right)\right]$$

$$\boxed{\text{LHR} = 112 \frac{\text{W}}{\text{cm}}}$$

Perform a forward Euler time stepping to approximate the following function:  $y(t) = e^{-2t}$  12 pt  
 Compute with timesteps  $dt=0.5$  and  $dt=0.25$ , expanding to  $t_0=1$ .

$$y(t) = e^{-2t} \quad y'(t) = -2e^{-2t}$$

$$y_{n+1} = y_n + h y'_n$$

$$y(0) = 1 \quad dt = 0.5, 0.25$$

$$dt = 0.5$$

$$t_0 = 0$$

$$t_1 = 0.5$$

$$t_2 = 1$$

$$y_1 = y_0 + h y'_0 = 1 + (0.5)(-2e^{-2(0)}) = 0$$

$$y_2 = y_1 + h y'_1 = 0 + (0.5)(-2e^{-2(0.5)}) = -0.37$$

$$dt = 0.25$$

$$t_0 = 0$$

$$t_1 = 0.25$$

$$t_2 = 0.5$$

$$t_3 = 0.75$$

$$t_4 = 1$$

$$y_1 = y_0 + h y'_0 = 1 + (0.25)(-2e^{-2(0)}) = 0.5$$

$$y_2 = y_1 + h y'_1 = 0.5 + (0.25)(-2e^{-2(0.25)}) = 0.2$$

$$y_3 = y_2 + h y'_2 = 0.2 + (0.25)(-2e^{-2(0.5)}) = -0.02$$

$$y_4 = y_3 + h y'_3 = -0.02 + (0.25)(-2e^{-2(0.75)}) = -0.92$$

Consider a metallic (UZr) fuel slug with a radius of 0.3 cm, a sodium gap of 0.1 cm, an HT9 cladding thickness of 0.05 cm. UZr thermal conductivity = 0.22 W/cm-K; Na thermal conductivity = 0.5 W/cm-K;  $h_{cool} = 5.5$  W/cm-K; HT9 thermal conductivity = 20 W/m-K;  $T_{cool} = 400$  K;  $Q = 550$  W/cm<sup>3</sup> 12 pt

What is the temperature at  $r=0.15$  cm?

$$LHR = Q \pi R_F^2 = (550)(\pi)(0.3^2) = 155 \frac{W}{cm} \checkmark$$

$$T_{fuel} - T_{cool} = \frac{LHR}{2\pi R_F h_{cool}} = \frac{155}{2\pi(0.3)(5.5)} = 15 \text{ K} \quad T_{cool} = 415 \text{ K}$$

$$T_{gap} - T_{clad} = \frac{LHR}{2\pi R_F h_{clad}} = \frac{155}{2\pi(0.3)} \frac{0.05}{0.2} = 20.6 \text{ K} \quad T_{gap} = 435.6 \text{ K}$$

$$T_{fuel} - T_{gap} = \frac{LHR}{2\pi R_F h_{gap}} = \frac{155}{2\pi(0.3)} \frac{0.1}{0.5} = 16.4 \text{ K} \quad T_{fuel} = 452 \text{ K}$$

$$T(r) - T_F = \frac{LHR}{4\pi k} \left(1 - \frac{r^2}{R_F^2}\right)$$

$$T(0.15) = \frac{155}{4\pi(0.22)} \left(1 - \frac{0.15^2}{0.3^2}\right) + 452$$

$$T(0.15) = 494 \text{ K}$$