

Temperature and Thermometer B.H.

## Temperature and Thermometer

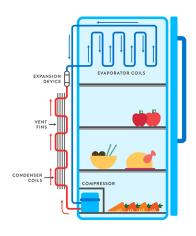
## Thermal Physics

Instructor: Ben Huang



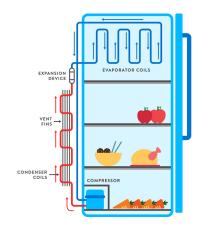


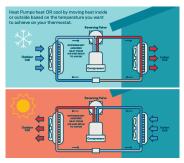
## Where do we apply thermal physics?





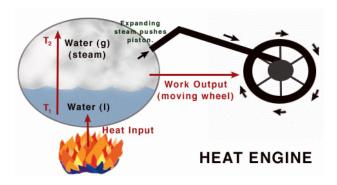
## Where do we apply thermal physics?







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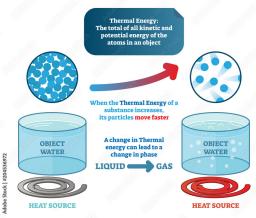


### Thermal Energy

Temperature and Thermometer

B.H.

## **THERMAL ENERGY**





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What is temperature?



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What is temperature? Mental Experiments:

1. How do you feel?





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## What is temperature? Mental Experiments:

#### 1. How do you feel?







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What is temperature? Mental Experiments:

2. Which one will boil quicker?





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What is temperature? Mental Experiments:

3. What will happen to the water?



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#### In summary:

Temperature is the property that determines whether or not energy will transfer between two objects when they are in thermal contact.

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#### In summary:

- Temperature is the property that determines whether or not energy will transfer between two objects when they are in thermal contact.
- For a single object, the higher its temperature is, the more thermal energy it has.



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What is the ice point and the steam point of water at atmospheric pressure?

Celsius Fahrenheit

ice point

steam point



Temperature and Thermometer B.H.

What is the ice point and the steam point of water at atmospheric pressure?

|             | Celsius | Fahrenheit    |
|-------------|---------|---------------|
| ice point   |         | 32° <i>F</i>  |
| steam point |         | 212° <i>F</i> |
|             |         |               |

Temperature and Thermometer B.H.

What is the ice point and the steam point of water at atmospheric pressure?

|             | Celsius       | Fahrenheit    |
|-------------|---------------|---------------|
| ice point   | 0° <i>C</i>   | 32° <i>F</i>  |
| steam point | 100° <i>C</i> | 212° <i>F</i> |

Temperature and Thermometer B.H.

What is the ice point and the steam point of water at atmospheric pressure?

|             | Celsius       | Fahrenheit    |
|-------------|---------------|---------------|
| ice point   | 0° <i>C</i>   | 32° <i>F</i>  |
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**Exercise.** Convert the following temperatures to their values on the Fahrenheit scale: (a) the sublimation point of dry ice,  $-78.5^{\circ}C$ ; (b) human body temperature,  $37.0^{\circ}C$ .



Pressure v.s. Temperature

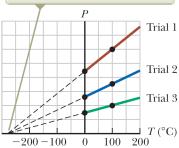




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The Kelvin Scale

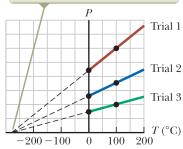
For all three trials, the pressure extrapolates to zero at the temperature  $-273.15^{\circ}$ C.



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The Kelvin Scale

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Kelvin = Celsius + 273.15



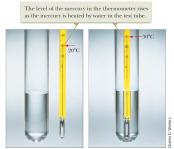
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Examine the following report, extrapolate the temperature when the pressure is zero, and express the result in Kelvin scale.

Report: Pressure-Volume-Temperature Data for Oxygen



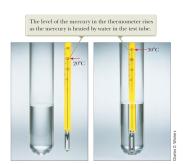
### Thermometer





#### Thermometer

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The volume of gas in the flask is kept constant by raising or lowering reservoir B to keep the mercury level in column A constant.

