Index of Refraction Data

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-- mode: org --

Setup

Using Snell's Law, we computed the refraction index for liquids through the equation

 $n_2 = \frac{n_1 sin(\theta_1)}{sin(\theta_2)}$ derived from $n_1 sin(\theta_1) = n_2 sin(\theta_2)$ n_2 is the refraction index for the liquid.

Needed in our report:

- \bullet The Equation formatted nicely with LATEX (Anthony or I can help you with that)
- Why we tried soda and not just water or other clearer liquids
- The data collection papers Anthony created.
- Why we used two lasers instead of one
- Some explanation of the code we used to compute the refaction indices and propagate error
- The problems in data collection we encountered (ask Anthony)
- The problems in code we encountered (ask Benjamin
- What we would have done if we could have collected data the second week

The following measurements use a subscript g or r to denote a measurement from the green laser or the red laser, respectively. Each measurement shone the light first through air into the substance, so a the n_{1g} is the known index of refraction for air (1.000277). We will list during each trial experiment to avoid confusion. We have assumed a $\pm 1*10^{-6}$ uncertainty in the refraction index for air.

Mountain Dew

Green Laser

```
n_{1g} = 1.000277
\theta_{1g} = 30^{\circ}
\theta_{2g} = 21^{\circ}
n_{2g} = 1.3956
dn_{1g} = 1 * 10^{-6}
d\theta_{1g} = 1^{\circ}
d\theta_{2g} = 1^{\circ}
dn_{2g} = 4.36591
```

Red Laser

```
\begin{split} n_{1r} &= 1.000277 \\ \theta_{1r} &= 38^{\circ} \\ \theta_{2r} &= 27^{\circ} \\ n_{2r} &= 1.35649 \\ dn_{1r} &= 1 * 10^{-}6 \\ d\theta_{1r} &= 1^{\circ} \\ d\theta_{2r} &= 1^{\circ} \\ dn_{2r} &= 3.17838 \end{split}
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Diet Mountain Dew

Green Laser

```
n_{1g} = 1.000277
\theta_{1g} = 30^{\circ}
\theta_{2g} = 16^{\circ}
n_{2g} = 1.81448
dn_{1g} = 1 * 10^{-}6
```

$$d\theta_{1g}=1^{\circ} \\ d\theta_{2g}=1^{\circ} \\ dn_{2g}=7.06531$$

Red Laser

$$\begin{split} n_{1r} &= 1.000277 \\ \theta_{1r} &= 37^{\circ} \\ \theta_{2r} &= 29^{\circ} \\ n_{2r} &= 1.24169 \\ dn_{1r} &= 1*10^{-6} \\ d\theta_{1r} &= 1^{\circ} \\ d\theta_{2r} &= 1^{\circ} \\ dn_{2r} &= 2.78083 \end{split}$$

Water

Green Laser

$$\begin{split} n_{1g} &= 1.000277 \\ \theta_{1g} &= 22^{\circ} \\ \theta_{2g} &= 18^{\circ} \\ n_{2g} &= 1.21259 \\ dn_{1g} &= 1*10^{-6} \\ d\theta_{1g} &= 1^{\circ} \\ d\theta_{2g} &= 1^{\circ} \\ dn_{2g} &= 2.75429 \end{split}$$

Red Laser

$$n_{1r} = 1.000277$$

$$\theta_{1r} = 42^{\circ}$$

$$\theta_{2r} = 30^{\circ}$$

$$n_{2r} = 1.33863$$

$$dn_{1r} = 1 * 10^{-}6$$

$$d\theta_{1r} = 1^{\circ}$$

$$d\theta_{2r} = 1^{\circ}$$

$$dn_{2r} = 2.75429$$