Acides-bases et pH

<https://www.khanacademy.org/science/chemistry/acids-and-bases-topic/modal/v/autoionization-water>

<https://www.khanacademy.org/science/chemistry/acids-and-bases-topic/modal/v/introduction-to-definition-of-ph>

<https://www.khanacademy.org/science/chemistry/acids-and-bases-topic/modal/v/ph-poh-of-strong-acids-and-bases>

<https://www.khanacademy.org/science/chemistry/acids-and-bases-topic/modal/v/strong-acids-and-strong-bases>

<https://www.khanacademy.org/science/chemistry/acids-and-bases-topic/modal/v/pka-and-pkb-relationship>

<https://www.khanacademy.org/science/chemistry/acids-and-bases-topic/modal/v/relationship-between-ka-and-kb>

<https://www.khanacademy.org/science/chemistry/acids-and-bases-topic/modal/v/ka-and-acid-strength>

<https://www.khanacademy.org/science/chemistry/acids-and-bases-topic/modal/v/weak-acid-equilibrium>

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