■ <u>Table 1.5</u> Approximate Physical Properties of Some Common Liquids (BG Units)

Liquid	Temperature (°F)	Density, $\rho \left(\text{slugs} / \text{ft}^3 \right)$	$\begin{array}{c} \textbf{Specific} \\ \textbf{Weight,} \\ \boldsymbol{\gamma} \left(\text{lb} / \text{ft}^3 \right) \end{array}$	$\begin{array}{c} \textbf{Dynamic} \\ \textbf{Viscosity,} \\ \boldsymbol{\mu} \ \left(\text{lb} \cdot \text{s/ ft}^2 \right) \end{array}$	Kinematic Viscosity, $\nu \left(\mathrm{ft}^2 / \mathrm{s} \right)$	Surface Tension, $\frac{a}{\sigma}$ (lb / ft)	$egin{aligned} \mathbf{Vapor\ Pressure,}\ oldsymbol{p_v}\ \left[\mathrm{lb\ /in.^2\ (abs)} ight] \end{aligned}$	$egin{aligned} \mathbf{Bulk} \\ \mathbf{Modulus,}^{\mathbf{b}} \\ \mathbf{E}_{oldsymbol{v}} & \left(\mathrm{lb} / \mathrm{in.}^2 ight) \end{aligned}$
Carbon tetrachloride	68	3.09	99.5	$2.00~\mathrm{E}-5$	$6.47~\mathrm{E}-6$	$1.84~{ m E} - 3$	$1.9 ext{ E} + 0$	$1.91 \mathrm{~E} + 5$
Ethyl alcohol	68	1.53	49.3	$2.49 ext{ E} - 5$	$1.63 \mathrm{\ E} - 5$	$1.56 \mathrm{\ E} - 3$	$8.5~\mathrm{E}-1$	$1.54 \mathrm{~E} + 5$
Gasoline	60	1.32	42.5	$6.5 \mathrm{E} - 6$	$4.9 \mathrm{E} - 6$	$1.5 \mathrm{E} - 3$	$8.0 ext{ E} + 0$	$1.9 \mathrm{E} + 5$
Glycerin	68	2.44	78.6	$3.13 ext{ E} - 2$	$1.28~\mathrm{E}-2$	$4.34 \mathrm{\ E} - 3$	$2.0~\mathrm{E}-6$	$6.56~\mathrm{E}+5$
Mercury	68	26.3	847	$3.28 \to -5$	$1.25~\mathrm{E}-6$	$3.19~\mathrm{E}-2$	$2.3~\mathrm{E}-5$	$4.14 ext{ E} + 6$
SAE 30 oil ^c	60	1.77	57.0	$8.0 \mathrm{E} - 3$	$4.5 \mathrm{E} - 3$	$2.5 \mathrm{E} - 3$	_	$2.2 \mathrm{E} + 5$
Seawater	60	1.99	64.0	$2.51~\mathrm{E}-5$	$1.26~\mathrm{E}-5$	$5.03 \; \mathrm{E} - 3$	$2.56~\mathrm{E}-1$	$3.39 \to 5$
Water	60	1.94	62.4	$2.34~\mathrm{E}-5$	$1.21~\mathrm{E}-5$	$5.03~\mathrm{E}-3$	$2.56~\mathrm{E}-1$	$3.12 \mathrm{~E} + 5$

^a In contact with air.

■ <u>Table 1.6</u> Approximate Physical Properties of Some Common Liquids (SI Units)

Liquid	Temperature	Density, $\rho \left(\log / \mathbf{m}^3 \right)$	$\begin{array}{c} \textbf{Specific} \\ \textbf{Weight,} \\ \boldsymbol{\gamma} \ \left(\text{kN} / \textbf{m}^3 \right) \end{array}$	$\begin{array}{c} \textbf{Dynamic} \\ \textbf{Viscosity,} \\ \boldsymbol{\mu} \ \left(\mathbf{N} \cdot \mathbf{s} / \mathbf{m}^2 \right) \end{array}$	Kinematic Viscosity, $ u\left(\mathbf{m^2/s}\right)$	Surface Tension, ^a σ (N/m)	Vapor Pressure, $p_v \left[\mathbf{N}/\mathbf{m}^2 (\mathrm{abs}) \right]$	$egin{aligned} \mathbf{Bulk} \\ \mathbf{Modulus,}^{\mathbf{b}} \\ \mathbf{E}_{v} \ \left(\mathbf{N}/\mathbf{m}^{2} ight) \end{aligned}$
Carbon tetrachloride	20	1,590	15.6	$9.58~\mathrm{E}-4$	$6.03~\mathrm{E}-7$	$2.69~\mathrm{E}-2$	$1.3~\mathrm{E}+4$	$1.31 \mathrm{~E} + 9$
Ethyl alcohol	20	789	7-74	$1.19~\mathrm{E}-3$	$1.51~\mathrm{E}-6$	$2.28~\mathrm{E}-2$	$5.9~\mathrm{E}+3$	$1.06 \mathrm{~E} + 9$
Gasoline ^c	15.6	680	6.67	$3.1~\mathrm{E}-4$	$4.6 \mathrm{E} - 7$	$2.2~\mathrm{E}-2$	$5.5~\mathrm{E}+4$	$1.3 \mathrm{E} + 9$
Glycerin	20	1,260	12.4	$1.50 \mathrm{~E} + 0$	$1.19~\mathrm{E}-3$	$6.33~\mathrm{E}-2$	$1.4~\mathrm{E}-2$	$4.52~\mathrm{E}+9$
Mercury	20	13,600	133	$1.57~\mathrm{E}-3$	$1.15~\mathrm{E}-7$	$4.66~\mathrm{E}-1$	$1.6~\mathrm{E}-1$	$2.85~\mathrm{E}+10$
SAE 30 oil ^c	15.6	912	8.95	$3.8 \mathrm{E} - 1$	$4.2~\mathrm{E}-4$	$3.6 \mathrm{E} - 2$	_	$1.5 \mathbf{E} + 9$
Seawater	15.6	1,030	10.1	$1.20~\mathrm{E}-3$	$1.17~\mathrm{E}-6$	$7.34~\mathrm{E}-2$	$1.77 ext{ E} + 3$	$2.34 \mathrm{~E} + 9$
Water	15.6	999	9.80	$1.12~\mathrm{E}-3$	$1.12~\mathrm{E}-6$	$7.34~\mathrm{E}-2$	$1.77 ext{ E} + 3$	$2.15~\mathrm{E}+9$

a In contact with air.

b Isentropic bulk modulus calculated from speed of sound.

^c Typical values. Properties of petroleum products vary.

 $[\]underline{{}^{\underline{b}}}$ Is entropic bulk modulus calculated from speed of sound.

^c Typical values. Properties of petroleum products vary.

■ Table 1.7 Approximate Physical Properties of Some Common Gases at Standard Atmospheric Pressure (BG Units)

Gas	Temperature (°F)	$\begin{array}{c} \textbf{Density,} \\ \boldsymbol{\rho} \ \left(\text{slugs} / \text{ft}^3 \right) \end{array}$	Specific Weight, $\gamma \left(\operatorname{lb} / \operatorname{ft}^3 \right)$	Dynamic Viscosity, $\mu \left(\text{lb} \cdot \text{s/ft}^2 \right)$	Kinematic Viscosity, $oldsymbol{ u}\left(ext{ft}^2/ ext{s}\right)$	Gas Constant, $\frac{a}{R}$ (ft · lb /slug · $^{\circ}$ R)	Specific Heat Ratio, b k
Air (standard)	59	$2.38~\mathrm{E}-3$	$7.65~\mathrm{E}-2$	$3.74~\mathrm{E}-7$	$1.57~\mathrm{E}-4$	$1.716~\mathrm{E} + 3$	1.40
Carbon dioxide	68	$3.55~\mathrm{E}-3$	$1.14 ext{ E} - 1$	$3.07\mathrm{E}-7$	$8.65~\mathrm{E}-5$	$1.130~\mathrm{E}+3$	1.30
Helium	68	$3.23~\mathrm{E}-4$	$1.04~\mathrm{E}-2$	$4.09 \; \mathrm{E} - 7$	$1.27~\mathrm{E}-3$	$1.242 ext{ E} + 4$	1.66
Hydrogen	68	$1.63~\mathrm{E}-4$	$5.25~\mathrm{E}-3$	$1.85~\mathrm{E}-7$	$1.13~\mathrm{E}-3$	$2.466~\mathrm{E}+4$	1.41
Methane (natural gas)	68	$1.29 \; \mathrm{E} - 3$	$4.15~\mathrm{E}-2$	$2.29~\mathrm{E}-7$	$1.78~\mathrm{E}-4$	$3.099~\mathrm{E}+3$	1.31
Nitrogen	68	$2.26~\mathrm{E}-3$	$7.28~\mathrm{E}-2$	$3.68~\mathrm{E}-7$	$1.63~\mathrm{E}-4$	$1.775 ext{ E} + 3$	1.40
Oxygen	68	$2.58~\mathrm{E}-3$	$8.31~\mathrm{E}-2$	$4.25~\mathrm{E}-7$	$1.65~\mathrm{E}-4$	$1.554~\mathrm{E} + 3$	1.40

^a Values of the gas constant are independent of temperature.

■ Table 1.8 Approximate Physical Properties of Some Common Gases at Standard Atmospheric Pressure (SI Units)

Gas	$\begin{array}{c} \textbf{Temperature} \\ (°\mathbf{C}) \end{array}$	Density, $ ho \; \left(\log / \mathbf{m}^3 \right)$	Specific Weight, $\gamma \left({f N}/{f m}^3 ight)$	Dynamic Viscosity, $\mu \ \left(\mathbf{N}\cdot\mathbf{s}/\mathbf{m}^2\right)$	Kinematic Viscosity, $ u$ (m ² /s)	$\begin{array}{c}\textbf{Gas Constant,}^{\underline{\mathbf{a}}}\\ \boldsymbol{R}\ (\mathbf{J}/\mathrm{kg}\cdot\mathbf{K})\end{array}$	Specific Heat Ratio, b k
Air (standard)	15	$1.23 \mathrm{~E} + 0$	$1.20~\mathrm{E}+1$	$1.79~\mathrm{E}-5$	$1.46~\mathrm{E}-5$	$2.869~\mathrm{E}+2$	1.40
Carbon dioxide	20	$1.83 \mathrm{~E} + 0$	$1.80 ext{ E} + 1$	$1.47~\mathrm{E}-5$	$8.03~\mathrm{E}-6$	$1.889~\mathrm{E}+2$	1.30
Helium	20	$1.66 \; \mathrm{E} - 1$	$1.63 ext{ E} + 0$	$1.94~\mathrm{E}-5$	$1.15~\mathrm{E}-4$	$2.077 \mathrm{~E} + 3$	1.66
Hydrogen	20	$8.38~\mathrm{E}-2$	$8.22 \; \mathrm{E} - 1$	$8.84 \: \mathrm{E} - 6$	$1.05~\mathrm{E}-4$	$4.124~\mathrm{E} + 3$	1.41
Methane (natural gas)	20	$6.67~\mathrm{E}-1$	$6.54~\mathrm{E}+0$	$1.10\mathrm{E}-5$	$1.65~\mathrm{E}-5$	$5.183~\mathrm{E}+2$	1.31
Nitrogen	20	$1.16 \mathrm{~E} + 0$	1.14 E + 1	$1.76~\mathrm{E}-5$	$1.52~\mathrm{E}-5$	$2.968~\mathrm{E}+2$	1.40
Oxygen	20	$1.33 \mathrm{~E} + 0$	$1.30~\mathrm{E}+1$	$2.04~\mathrm{E}-5$	$1.53~\mathrm{E}-5$	$2.598~\mathrm{E} + 2$	1.40

^a Values of the gas constant are independent of temperature.

^b Values of the specific heat ratio depend only slightly on temperature.

b Values of the specific heat ratio depend only slightly on temperature.

■ Table B.1 Physical Properties of Water (BG/EE Units)ª

Temperature (°F)	Density, ρ (slugs/ft ³) ^b	Specific Weight [£] , γ (lb/ft ³)	Dynamic Viscosity, μ (lb·s/ft²)	Kinematic Viscosity, <i>v</i> (ft²/s)	Surface Tension ^d , σ (lb/ft)	Vapor Pressure, p_v [lb/in.² (abs)]	Speed of Sound ^e , c (ft/s)
32	1.940	62.42	$3.732 ext{ E} - 5$	$1.924 ext{ E} - 5$	$5.18 ext{ E} - 3$	$8.854 ext{ E} - 2$	4603
40	1.940	62.43	$3.228 ext{ E} - 5$	$1.664 ext{ E} - 5$	5.13 $E - 3$	1.217 $E-1$	4672
50	1.940	62.41	$2.730 ext{ E} - 5$	1.407 $E-5$	$5.09 ext{ E} - 3$	1.781 E – 1	4748
60	1.938	62.37	$2.344 ext{ E} - 5$	1.210 $E-5$	5.03 $E - 3$	$2.563 ext{ E} - 1$	4814
70	1.936	62.30	$2.037 ext{ E} - 5$	$1.052 ext{ E} - 5$	4.97 $E - 3$	3.631 E-1	4871
80	1.934	62.22	1.791 $E-5$	$9.262 ext{ E} - 6$	$4.91 ext{ E} - 3$	$5.069 ext{ E} - 1$	4819
90	1.931	62.11	1.500 E -5	8.233 $E-6$	$4.86 ext{ E} - 3$	6.979 E-1	4960
100	1.927	62.00	1.423 $E-5$	7.383 $E-6$	4.79 $E - 3$	9.493 E-1	4995
120	1.918	61.71	1.164 $E-5$	6.067 $E-6$	$4.67 ext{ E} - 3$	1.692 E+0	5049
140	1.908	61.38	$9.743 ext{ E} - 6$	5.106 $E-6$	$4.53 ext{ E} - 3$	2.888 E+0	5091
160	1.896	61.00	8.315 $E-6$	$4.385 ext{ E} - 6$	$4.40 ext{ E} - 3$	4.736 E+0	5101
180	1.883	60.58	7.207 $E-6$	$3.827 ext{ E} - 6$	$4.26 ext{ E} - 3$	7.507 E + 0	5195
200	1.869	60.12	$6.342 ext{ E} - 6$	$3.393 ext{ E} - 6$	$4.12 ext{ E} - 3$	1.152 E + 1	5089
212	1.860	59.83	$5.886 ext{ E} - 6$	$3.165 ext{ E} - 6$	4.04 E – 3	1.469 E+1	5062

^a Based on data from *Handbook of Chemistry and Physics*, 69th Ed., CRC Press, 1988. Where necessary, values obtained by interpolation.

b To obtain EE units (lbm / ft³) multiply by 32.174.

 $^{^{} extstyle 2}$ Density and specific weight are related through the equation $\gamma=
ho$ g. For this table, $g=32.174\, ext{ ft}/ ext{s}^2$.

d In contact with air.

^e Based on data from R. D. Blevins, *Applied Fluid Dynamics Handbook*, Van Nostrand Reinhold Co., Inc., New York, 1984.

■ Table B.2 Physical Properties of Water (SI Units)^a

Temperature (°C)	Density, ρ (kg/m³)	Specific Weight ^b , γ (kN/m³)	Dynamic Viscosity, μ (N·s/m²)	Kinematic Viscosity, <i>v</i> (m ² /s)	Surface Tension ^c , σ (N/m)	Vapor Pressure, p _v [N/m ² (abs)]	Speed of Sound ^d , c (m/s)
0	999.9	9.806	1.787 $E-3$	1.787 $E-6$	$7.56 ext{ E} - 2$	$6.105 ext{ E} + 2$	1403
5	1000.0	9.807	$1.519 ext{ E} - 3$	1.519 $E-6$	7.49 $E-2$	$8.722 ext{ E} + 2$	1427
10	999.7	9.804	$1.307 ext{ E} - 3$	$1.307 ext{ E} - 6$	7.42 $E-2$	$1.228 ext{ E} + 3$	1447
20	998.2	9.789	$1.002 ext{ E} - 3$	$1.004 ext{ E} - 6$	7.28 $E-2$	2.338 E+3	1481
30	995.7	9.765	7.975 $E-4$	8.009 $E-7$	7.12 $E-2$	$4.243 ext{ E} + 3$	1507
40	992.2	9.731	$6.529 ext{ E} - 4$	6.580 $E-7$	6.96 $E-2$	$7.376 ext{ E} + 3$	1526
50	988.1	9.690	$5.468 ext{ E} - 4$	$5.534 ext{ E} - 7$	6.79 $E-2$	1.233 $E+4$	1541
60	983.2	9.642	$4.665 ext{ E} - 4$	4.745 $E-7$	6.62 $E-2$	1.992 E+4	1552
70	977.8	9.589	$4.042 ext{ E} - 4$	4.134 $E-7$	6.44 $E-2$	$3.116 ext{ E} + 4$	1555
80	971.8	9.530	$3.547 ext{ E} - 4$	$3.650 ext{ E} - 7$	6.26 $E-2$	$4.734 ext{ E} + 4$	1555
90	965.3	9.467	$3.147 ext{ E} - 4$	$3.260 ext{ E} - 7$	6.08 $E-2$	7.010 $E + 4$	1550
100	958.4	9.399	$2.818 ext{ E} - 4$	$2.940 ext{ E} - 7$	$5.89 ext{ E} - 2$	1.013 E + 5	1543

a Based on data from *Handbook of Chemistry and Physics*, 69th Ed., CRC Press, 1988. Density and specific weight are related through the equation $\gamma = \rho g$. For this table, $g = 9.807 \text{ m/s}^2$.

^c In contact with air.

d Based on data from R. D. Blevins, *Applied Fluid Dynamics Handbook*, Van Nostrand Reinhold Co., Inc., New York, 1984.

■ Table B.3 Physical Properties of Air at Standard Atmospheric Pressure (BG/EE Units)^a

Temperature (°F)	Density, ρ (slugs/ft³) ^b	Specific Weight ^c , γ (lb/ft ³)	Dynamic Viscosity, μ (lb·s/ft²)	Kinematic Viscosity, v (ft²/s)	Specific Heat Ratio, k (—)	Speed of Sound, of (ft/s)
-40	$2.939 ext{ E} - 3$	$9.456 ext{ E} - 2$	$3.29 ext{ E} - 7$	$1.12 ext{ E} - 4$	1.401	1004
-20	$2.805 ext{ E} - 3$	$9.026 ext{ E} - 2$	$3.34 ext{ E} - 7$	1.19 E – 4	1.401	1028
О	$2.683 ext{ E} - 3$	8.633 $E-2$	$3.38 ext{ E} - 7$	$1.26 ext{ E} - 4$	1.401	1051
10	$2.626 ext{ E} - 3$	8.449 $E-2$	$3.44 ext{ E} - 7$	1.31 E – 4	1.401	1062
20	$2.571 ext{ E} - 3$	8.273 $E-2$	$3.50 ext{ E} - 7$	1.36 E – 4	1.401	1074
30	$2.519 ext{ E} - 3$	8.104 $E-2$	$3.58 ext{ E} - 7$	1.42 E – 4	1.401	1085
40	$2.469 ext{ E} - 3$	7.942 $E-2$	$3.60 ext{ E} - 7$	1.46 E-4	1.401	1096
50	$2.420 ext{ E} - 3$	7.786 $E-2$	$3.68 ext{ E} - 7$	1.52 E – 4	1.401	1106
60	$2.373 ext{ E} - 3$	7.636 $E-2$	$3.75 ext{ E} - 7$	1.58 E-4	1.401	1117
70	$2.329 ext{ E} - 3$	7.492 $E-2$	$3.82 ext{ E} - 7$	1.64 E-4	1.401	1128
80	$2.286 ext{ E} - 3$	7.353 $E-2$	$3.86 ext{ E} - 7$	1.69 E-4	1.400	1138
90	$2.244 ext{ E} - 3$	7.219 $E-2$	$3.90 ext{ E} - 7$	1.74 E-4	1.400	1149
100	$2.204 ext{ E} - 3$	7.090 $E-2$	$3.94 ext{ E} - 7$	1.79 $E-4$	1.400	1159
120	$2.128 ext{ E} - 3$	6.846 $E-2$	$4.02 ext{ E} - 7$	1.89 E-4	1.400	1180
140	$2.057 ext{ E} - 3$	6.617 $E-2$	4.13 $E-7$	$2.01 ext{ E} - 4$	1.399	1200
160	1.990 $E - 3$	6.404 $E-2$	$4.22 ext{ E} - 7$	$2.12 ext{ E} - 4$	1.399	1220
180	1.928 $E-3$	6.204 $E-2$	4.34 $E-7$	$2.25 ext{ E} - 4$	1.399	1239
200	1.870 $E - 3$	6.016 $E-2$	4.49 $E-7$	$2.40 ext{ E} - 4$	1.398	1258
300	$1.624 ext{ E} - 3$	$5.224 ext{ E} - 2$	4.97 $E-7$	$3.06 ext{ E} - 4$	1.394	1348
400	$1.435 ext{ E} - 3$	$4.616 ext{ E} - 2$	$5.24 ext{ E} - 7$	$3.65 ext{ E} - 4$	1.389	1431
500	$1.285 ext{ E} - 3$	$4.135 ext{ E} - 2$	$5.80 ext{ E} - 7$	4.51 $E-4$	1.383	1509
750	$1.020 ext{ E} - 3$	$3.280 ext{ E} - 2$	6.81 $E-7$	6.68 $E-4$	1.367	1685
1000	8.445 $E-4$	$2.717 ext{ E} - 2$	$7.85 ext{ E} - 7$	$9.30 ext{ E} - 4$	1.351	1839
1500	6.291 $E-4$	$2.024 ext{ E} - 2$	$9.50 ext{ E} - 7$	1.51 E – 3	1.329	2114

 $[\]frac{a}{b}$ Based on data from R. D. Blevins, *Applied Fluid Dynamics Handbook*, Van Nostrand Reinhold Co., Inc., New York, 1984. $\frac{b}{b}$ To obtain EE units (lbm / ft³) multiply by 32.174.

 $[\]subseteq$ Density and specific weight are related through the equation $\gamma = \rho g$. For this table g = 32.174 ft $/s^2$.

■ Table B.4 Physical Properties of Air at Standard Atmospheric Pressure (SI Units)^a

Temperature (°C)	Density, ρ (kg/m³)	Specific Weight ^b , γ (N/m³)	Dynamic Viscosity, μ (N·s/m²)	Kinematic Viscosity, v (m²/s)	Specific Heat Ratio, k (—)	Speed of Sound, c (m/s)
-40	1.514	14.85	1.57 $E-5$	$1.04 ext{ E} - 5$	1.401	306.2
-20	1.395	13.68	1.63 $E-5$	1.17 E – 5	1.401	319.1
О	1.292	12.67	1.71 $E-5$	1.32 E -5	1.401	331.4
5	1.269	12.45	1.73 $E-5$	1.36 E – 5	1.401	334.4
10	1.247	12.23	1.76 $E-5$	1.41 E – 5	1.401	337.4
15	1.225	12.01	1.80 E -5	1.47 E – 5	1.401	340.4
20	1.204	11.81	$1.82 ext{ E} - 5$	1.51 E - 5	1.401	343.3
25	1.184	11.61	1.85 E -5	1.56 E - 5	1.401	346.3
30	1.165	11.43	$1.86 ext{ E} - 5$	1.60 E - 5	1.400	349.1
40	1.127	11.05	1.87 $E-5$	1.66 E – 5	1.400	354-7
50	1.109	10.88	1.95 E -5	1.76 E – 5	1.400	360.3
60	1.060	10.40	1.97 $E-5$	1.86 E – 5	1.399	365.7
70	1.029	10.09	$2.03 ext{ E} - 5$	1.97 E – 5	1.399	371.2
80	0.9996	9.803	$2.07 ext{ E} - 5$	$2.07 ext{ E} - 5$	1.399	376.6
90	0.9721	9.533	$2.14 ext{ E} - 5$	$2.20 ext{ E} - 5$	1.398	381.7
100	0.9461	9.278	$2.17 ext{ E} - 5$	$2.29 ext{ E} - 5$	1.397	386.9
200	0.7461	7.317	$2.53 ext{ E} - 5$	3.39 $E-5$	1.390	434.5
300	0.6159	6.040	$2.98 ext{ E} - 5$	4.84 $E-5$	1.379	476.3
400	0.5243	5.142	$3.32 ext{ E} - 5$	6.34 E – 5	1.368	514.1
500	0.4565	4.477	$3.64 ext{ E} - 5$	7.97 $E-5$	1.357	548.8
1000	0.2772	2.719	$5.04 ext{ E} - 5$	$1.82 ext{ E} - 4$	1.321	694.8

^a Based on data from R. D. Blevins, *Applied Fluid Dynamics Handbook*, Van Nostrand Reinhold Co., Inc., New York, 1984. b Density and specific weight are related through the equation $\gamma = \rho g$. For this table $g = 9.807 \text{ m/s}^2$.

■ Table C.1 Properties of the U.S. Standard Atmosphere (BG/EE Units)^a

Altitude (ft)	Temperature (°F)	Acceleration of Gravity, g (ft/s ²)	Pressure, p [lb/in.²(abs)]	Density, $ ho$ (slugs/ft³) $^{ extbf{b}}$	Dynamic Viscosity, μ (lb · s/ft ²)
-5,000	76.84	32.189	17.554	$2.745 ext{ E} - 3$	$3.836 ext{ E} - 7$
0	59.00	32.174	14.696	$2.377 ext{ E} - 3$	3.737 $E-7$
5,000	41.17	32.159	12.228	$2.048 ext{ E} - 3$	$3.637 ext{ E} - 7$
10,000	23.36	32.143	10.108	1.756 $E-3$	$3.534 ext{ E} - 7$
15,000	5.55	32.128	8.297	$1.496 ext{ E} - 3$	$3.430 ext{ E} - 7$
20,000	-12.26	32.112	6.759	1.267 $E-3$	$3.324 ext{ E} - 7$
25,000	-30.05	32.097	5.461	$1.066 ext{ E} - 3$	$3.217 ext{ E} - 7$
30,000	-47.83	32.082	4.373	8.907 $E-4$	$3.107 ext{ E} - 7$
35,000	-65.61	32.066	3.468	7.382 $E-4$	$2.995 ext{ E} - 7$
40,000	-69.70	32.051	2.730	$5.873 ext{ E} - 4$	$2.969 ext{ E} - 7$
45,000	-69.70	32.036	2.149	$4.623 ext{ E} - 4$	$2.969 ext{ E} - 7$
50,000	-69.70	32.020	1.692	$3.639 ext{ E} - 4$	$2.969 ext{ E} - 7$
60,000	-69.70	31.990	1.049	$2.256 ext{ E} - 4$	$2.969 ext{ E} - 7$
70,000	-67.42	31.959	0.651	$1.392 ext{ E} - 4$	$2.984 ext{ E} - 7$
80,000	-61.98	31.929	0.406	8.571 $E-5$	$3.018 ext{ E} - 7$
90,000	-56.54	31.897	0.255	$5.610 ext{ E} - 5$	$3.052 ext{ E} - 7$
100,000	-51.10	31.868	0.162	$3.318 ext{ E} - 5$	$3.087 ext{ E} - 7$
150,000	19.40	31.717	0.020	$3.658 ext{ E} - 6$	$3.511 ext{ E} - 7$
200,000	-19.78	31.566	0.003	$5.328 ext{ E} - 7$	$3.279 ext{ E} - 7$
250,000	-88.77	31.415	0.000	6.458 E – 8	$2.846 ext{ E} - 7$

■ Table C.2 Properties of the U.S. Standard Atmosphere (SI Units)ª

Altitude	Temperature	Acceleration of Gravity,	Press	ure, p		ity, ρ	Dynamic Viscosity, μ		
(m)	(°C)	$g (\mathrm{m/s^2})$	[N/m ²	(abs)]	(kg/	/m³)	(N·s	s/m²)	
-1,000	21.50	9.810	1.139	E+5	1.347	E + 0	1.821	$\mathrm{E}-5$	
О	15.00	9.807	1.013	E+5	1.225	E + 0	1.789	$\mathrm{E}-5$	
1,000	8.50	9.804	8.988	$\mathrm{E}+4$	1.112	E + 0	1.758	$\mathrm{E}-5$	
2,000	2.00	9.801	7.950	$\mathbf{E} + 4$	1.007	E + 0	1.726	$\mathrm{E}-5$	
3,000	-4.49	9.797	7.012	$\mathrm{E}+4$	9.093	$\mathrm{E}-1$	1.694	$\mathrm{E}-5$	
4,000	-10.98	9.794	6.166	$\mathrm{E}+4$	8.194	$\mathrm{E}-1$	1.661	$\mathrm{E}-5$	
5,000	-17.47	9.791	5.405	E+4	7.364	$\mathrm{E}-1$	1.628	$\mathrm{E}-5$	
6,000	-23.96	9.788	4.722	E+4	6.601	$\mathrm{E}-1$	1.595	$\mathrm{E}-5$	
7,000	-30.45	9.785	4.111	$\mathrm{E}+4$	5.900	$\mathrm{E}-1$	1.561	$\mathrm{E}-5$	
8,000	-36.94	9.782	3.565	$\mathbf{E} + 4$	5.258	E-1	1.527	$\mathrm{E}-5$	
9,000	-43.42	9.779	3.080	E+4	4.671	$\mathrm{E}-1$	1.493	$\mathrm{E}-5$	
10,000	-49.90	9.776	2.650	$\mathbf{E} + 4$	4.135	$\mathrm{E}-1$	1.458	$\mathrm{E}-5$	
15,000	-56.50	9.761	1.211	$\mathrm{E}+4$	1.948	$\mathrm{E}-1$	1.422	$\mathrm{E}-5$	
20,000	-56.50	9.745	5.529	$\mathrm{E}+3$	8.891	$\mathrm{E}-2$	1.422	$\mathrm{E}-5$	
25,000	-51.60	9.730	2.549	$\mathrm{E}+3$	4.008	$\mathrm{E}-2$	1.448	$\mathrm{E}-5$	
30,000	-46.64	9.715	1.197	$\mathrm{E}+3$	1.841	$\mathrm{E}-2$	1.475	$\mathrm{E}-5$	
40,000	-22.80	9.684	2.871	$\mathrm{E}+2$	3.996	$\mathrm{E}-3$	1.601	$\mathrm{E}-5$	
50,000	-2.50	9.654	7.978	$\mathbf{E}+1$	1.027	$\mathrm{E}-3$	1.704	$\mathrm{E}-5$	
60,000	-26.13	9.624	2.196	E+1	3.097	$\mathrm{E}-4$	1.584	$\mathrm{E}-5$	
70,000	-53.57	9.594	5.221	$\mathbf{E} + 0$	8.283	$\mathrm{E}-5$	1.438	$\mathrm{E}-5$	
80,000	-74.51	9.564	1.052	$\mathbf{E} + 0$	1.846	$\mathrm{E}-5$	1.321	$\mathrm{E}-5$	

^a Data abridged from *U.S. Standard Atmosphere*, 1976, U.S. Government Printing Office, Washington, D.C.