# Lecture 11: Dielectrics and the Equation of Continuity

#### ECE221: Electric and Magnetic Fields



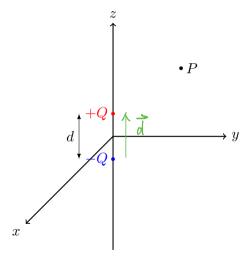
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#### Outline

- 1 Dipole Description of Materials
- Polarization and Dielectric Constant
- 3 Dielectric Breakdown
- 4 Conservation of Charge and the Equation of Continuity

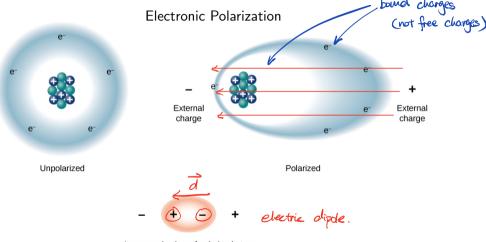
#### Review: Electric Dipole



$$E = \frac{Qd}{4\pi\epsilon_0 R^3} (2\cos\theta \hat{R} + \sin\theta \hat{\theta})$$

$$\vec{P} = \text{dipole moment} = \vec{Q} \vec{d}$$

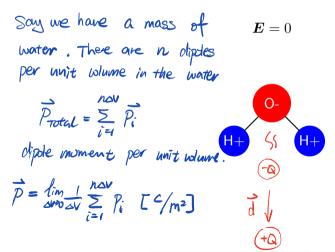
#### The Effect of an Applied Electric Field on Dielectrics



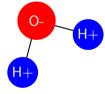
Large-scale view of polarized atom

Source: phys.libretexts.org

### Effect of an Applied Electric Field on Polar Molecules

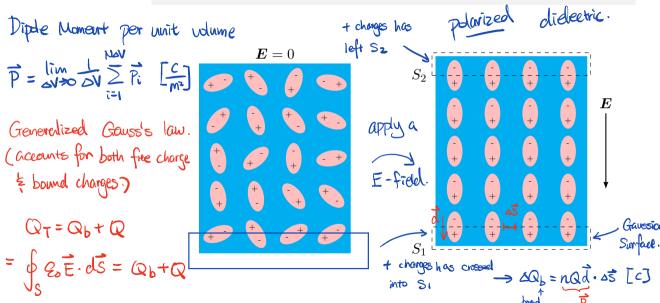








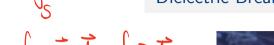
#### Polarization of a Dielectric



$$Q_b = -\oint \overrightarrow{p} \cdot d\overrightarrow{s} \qquad |odes| \text{ kind of like}$$

$$Gass's law.$$

$$Q = \oint \overrightarrow{D} \cdot d\overrightarrow{s} \qquad |ector | |ector$$



$$= \oint_{S} \mathcal{E} \cdot ds + \oint_{S} \mathbf{P} \cdot ds$$

$$= \oint_{S} \left( \mathcal{E} \cdot \mathbf{F} \right) \cdot ds \quad \text{non-Jeno}$$





let P= Xe SoE

D= SE +P

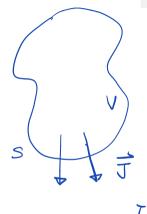
The dielectric strength is the maximum electric field that a dielectric can withstand without electrical breakdown.

where  $\chi_{e}$  electric susceptibility!

$$\overline{D} = \mathcal{C}_{\infty}(1+\chi_e)\overline{E}$$
 Table of Dielectrics

0 - 1 2 - 1 5	Material	Dielectric constant $\epsilon_r$	Dielectric strength ( $\times 10^6$ V/m)
Z= Folatine permittiving	Vacuum	1	$\infty$
Du halaus	Dry air (1 atm) Teflon <sup>TM</sup>	1.00059	3.0
of the dieletric.	Teflon <sup>TM</sup>	2.1	60 to 173
1	Paraffin	2.3	11
= 9 F	Silicon oil	2.5	10 to 15
ことに	Polystyrene	2.56	19.7
	Nylon	3.4	14
	Paper	3.7	16
E= 60E+	Fused quartz	3.78	8
	Glass	4 to 6	9.8 to 13.8
	Concrete	4.5	_
	Diamond	5.5	2,000
	Mica	6.0	118
	Water	80	_
	Titanium dioxide	86 to 173	_
	Strontium titanate	310	8
	Barium titanate	1,200 to 10,000	_
	Calcium copper titanate	> 250,000	_

## Conservation of Charge and the Equation of Continuity



#### Charges cannot be created or destroyed.

What is the circuit equivalent of this law?

Total current leaving closed surface

S is 
$$I = \oint \vec{J} \cdot d\vec{s} = \text{Sutward flow}$$
of positive charge.

Div. theorem.

the total change in S is Qi 
$$\Rightarrow \nabla \cdot \vec{J} = -\frac{d\vec{l}}{dt}$$

Equation of continuity (of current) potent

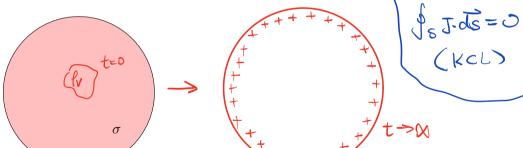
$$\vec{J} \cdot d\vec{s} = -\frac{dQ_i}{dt}$$

Lecture 11



#### Relaxation Time

Consider a problem where charges are introduced into the interior of a conductor during the time t < 0. What happens for t > 0?



#### Relaxation Time

$$\frac{\partial \ell_{\nu}}{\partial t} + \sqrt{\frac{\ell_{\nu}}{20}} = 0 \rightarrow \frac{\partial \ell_{\nu}}{\ell_{\nu}} = \frac{-\sqrt{3}}{20} \delta t$$

**Relaxation time** is the time it takes a charge density placed in the interior of a material to drop to  $e^{-1}=36.8\%$  of its original value.

$$\ln(fv) - \ln(fvo)$$

$$= \frac{-\tau}{40} t$$

$$T_r = \frac{\varepsilon}{4}$$
 relaxation time.