

```

In[1]:= (* we will do everything in terms of h-bar to make our life easier *)
hbar = 1;
P = 2;
psiX[x_] := (1 / x) * Sqrt[2 * hbar / (Pi * P)] * Sin[P * x / (2 * hbar)];
probX[x_] := psiX[x] * Conjugate[psiX[x]];
psiP[p_] :=
  Piecewise[{{0, p < -P / 2}, {1 / Sqrt[P], -P / 2 < p < P / 2}, {0, p > P / 2}}];
probP[p_] := psiP[p] * Conjugate[psiP[p]];

```

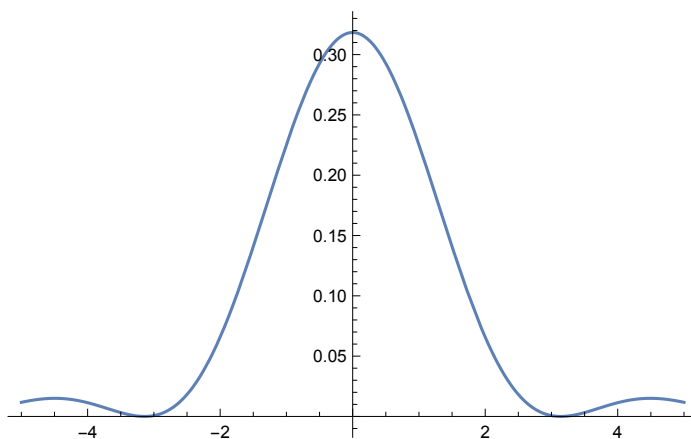
In[7]:=

```

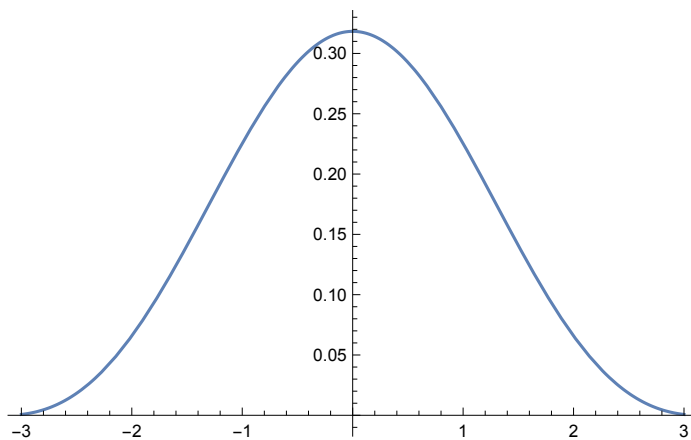
In[8]:= Plot[probX[x], {x, -5, 5}, PlotRange -> All]
Plot[probX[x], {x, -3, 3}]

```

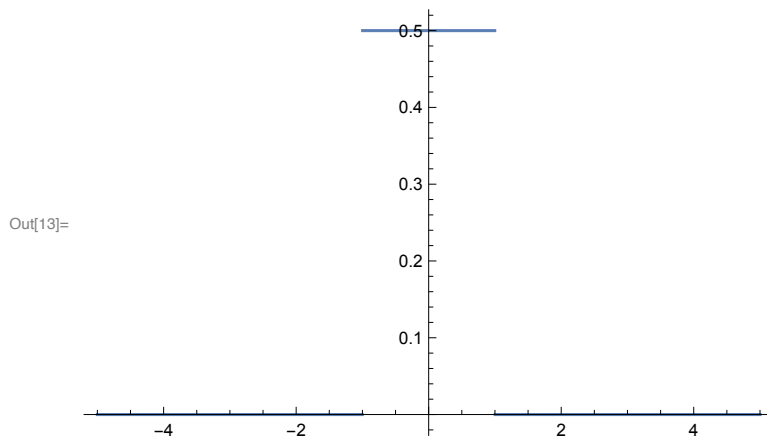
Out[8]=



Out[9]=



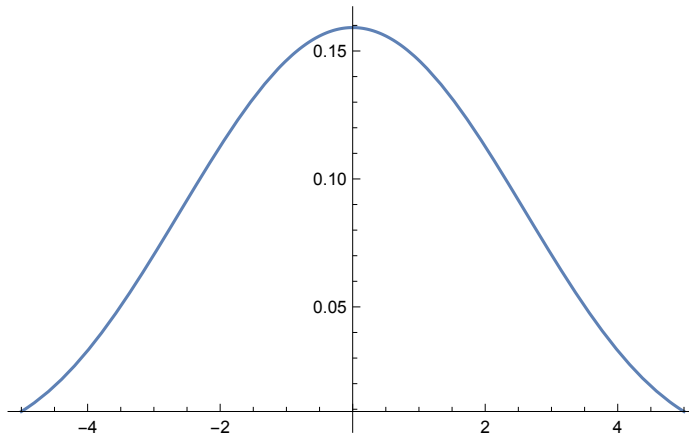
```
In[13]:= Plot[probP[x], {x, -5, 5}, PlotRange -> All]
```



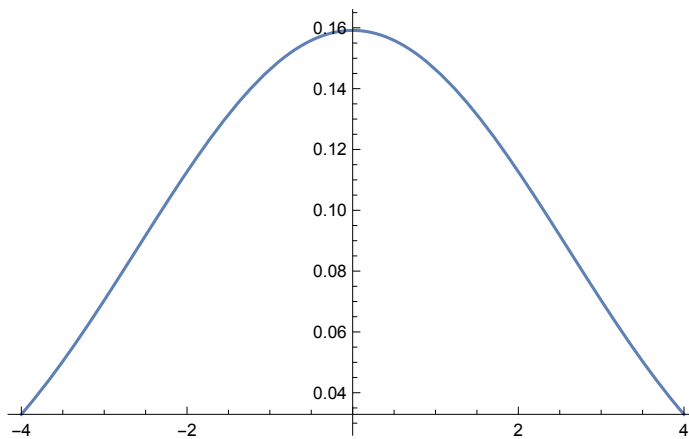
(* we can see from Out[9] that the first 0s will be at $\pm 3\hbar$ therefore we can guess that Δx is about $3\hbar$. This is reasonable to do because this big curve is the majority of the probability so the particle will be found to have some x-value between the first 0s and thus the uncertainty in x is the width up to the 0s. This aligns with Heisenberg since the principle states the uncertainties have to be greater than or equal to $\hbar/2$ and we can see with the plots above that $\Delta x = 3\hbar$ and $\Delta p = \hbar$ and so we do not violate the principle.*)

```
In[30]:= P = 1;
psiX[x_] := (1/x) * Sqrt[2 * hbar / (Pi * P)] * Sin[P * x / (2 * hbar)];
probX[x_] := psiX[x] * Conjugate[psiX[x]];
psiP[p_] :=
  Piecewise[{{0, p < -P/2}, {1/Sqrt[P], -P/2 < p < P/2}, {0, p > P/2}}];
probP[p_] := psiP[p] * Conjugate[psiP[p]];
Plot[probX[x], {x, -5, 5}, PlotRange -> All]
Plot[probX[x], {x, -4, 4}]
Plot[probP[p], {p, -5, 5}, PlotRange -> All]
```

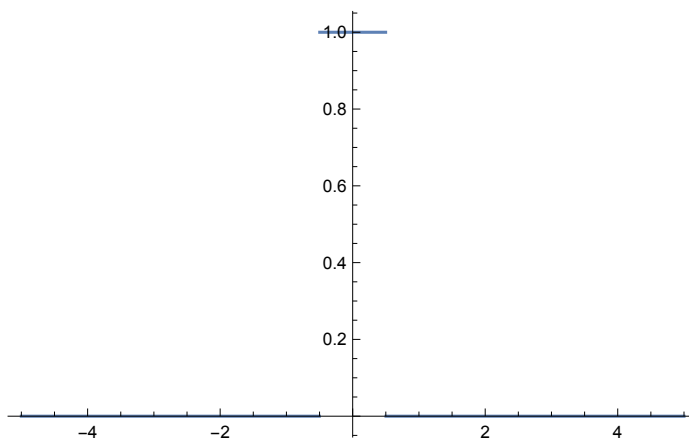
Out[35]=



Out[36]=



Out[37]=



(* we can see as we decrease P ,
 we increase Δx (seen in Out[35] and Out[36]) and therefore
 decrease Δp (seen in Out[37]. This behavior is expected due
 to Heisenberg's uncertainty principle *)