

EECS 16B

Designing Information Devices and Systems II

Lecture 9

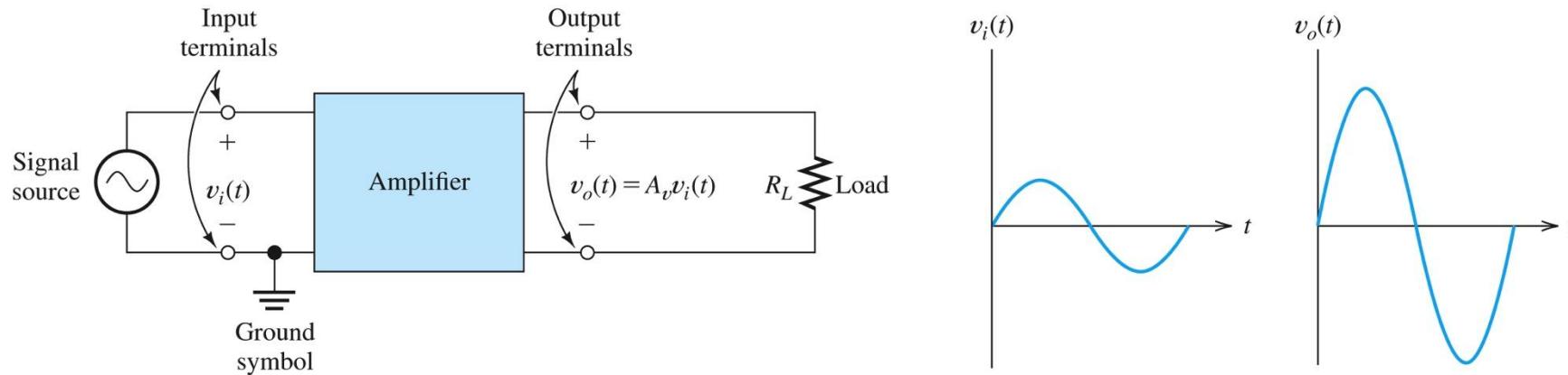
Prof. Sayeef Salahuddin

Department of Electrical Engineering and Computer Sciences, UC Berkeley,
sayeef@eecs.berkeley.edu

Devices

- Outline
 - Amplifiers and Devices
- Reading-slides

Active Devices



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- Active devices are made of semiconductors
 - Semi-conductors are materials whose resistance is in between a metal and insulator
- Half**
- More interestingly, one is able to change the resistance of the semiconductor materials by using external control such as voltage or current

Semiconductors

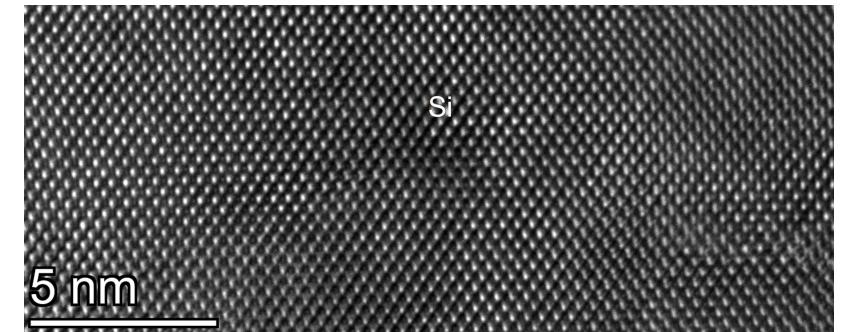
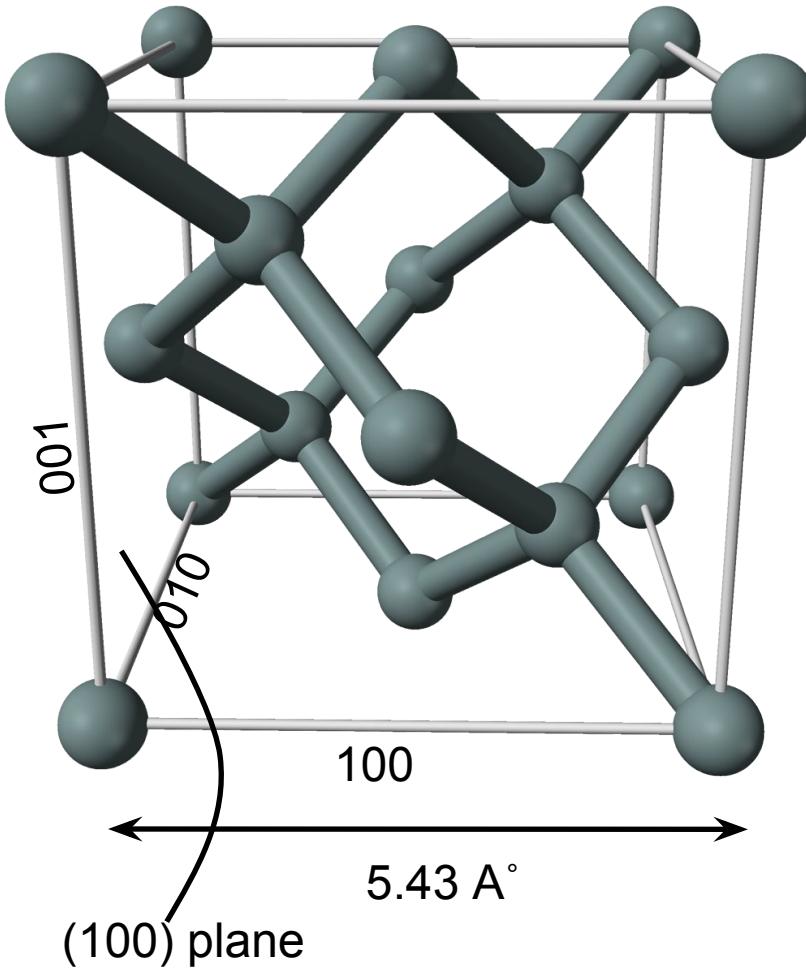
- Semiconductors are **usually** made of group IV elements- atoms that contain, on average, four valence electrons
- Most Common semiconductor used in electronic devices is **Silicon**

The image shows a standard periodic table of elements. The table is organized into seven horizontal rows (periods) and 18 vertical columns (groups). The groups are color-coded: IA (green), IIA (blue), IIIA (purple), IVA (light green), VA (bright green), VIA (yellow-green), VIIA (yellow), and O (orange). The first two periods have only two groups each. From period 3 onwards, there are three groups in the first section (IIIB, IVB, VB), one group in the middle section (VIB), and two groups in the last section (VIIIB, IB, IIB). The table includes element symbols, atomic numbers, and some additional elements at the bottom.

		Periodic Table of the Elements																									
		IA		IIA		VIA						VIIA						O									
1	H	3	Li	4	Be							5	B	6	C	7	N	8	O	9	F <th>10</th> <td>He</td>	10	He				
2	Na	11	Mg	IIIB		IVB	VB	VIB	VIIIB	VII		IB	IIB	13	Al	14	Si	15	P	16	S	17	Cl	18	Ar		
3	K	19	Ca	21	Sc	22	Ti	23	V	24	Cr	25	Mn	26	Fe	27	Co	28 <td>Ni</td> <th>29<td>Cu</td><th>30</th><td>Zn</td></th>	Ni	29 <td>Cu</td> <th>30</th> <td>Zn</td>	Cu	30	Zn				
4	Rb	37	Sr	38	Y	39	Zr	40	Nb	41	Mo	42	Tc	43	Ru	44	Rh	45	Pd	46	Ag	47	Cd	48	In		
5	Cs	55	Ba	56	*La	57	Hf	72	Ta	73	W	74	Re	75	Os	76	Ir	77	Pt	78 <td>Au</td> <th>79</th> <td>Hg</td> <th>80</th> <td>Tl</td> <th>81</th> <td>In</td>	Au	79	Hg	80	Tl	81	In
6	Fr	87	Ra	88	+Ac	89	Rf	104	Ha	105	Sg	106	Ns	107	Hs	108	Mt	109	110	111	112	113	113	113	113		
7																											

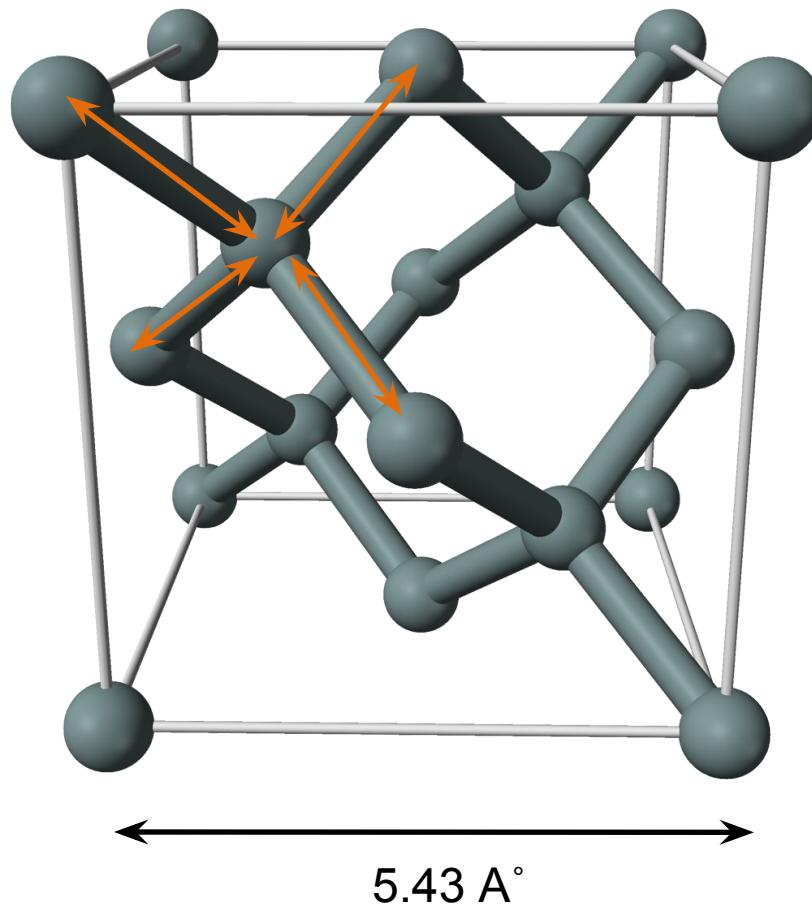
Crystal Structure of Si

Often known as the diamond lattice



Transmission Electron Microscopy
Image of Si taken at Lawrence
Berkeley National Laboratorys

Crystal Structure of Si



Each atom has 4 nearest neighbors

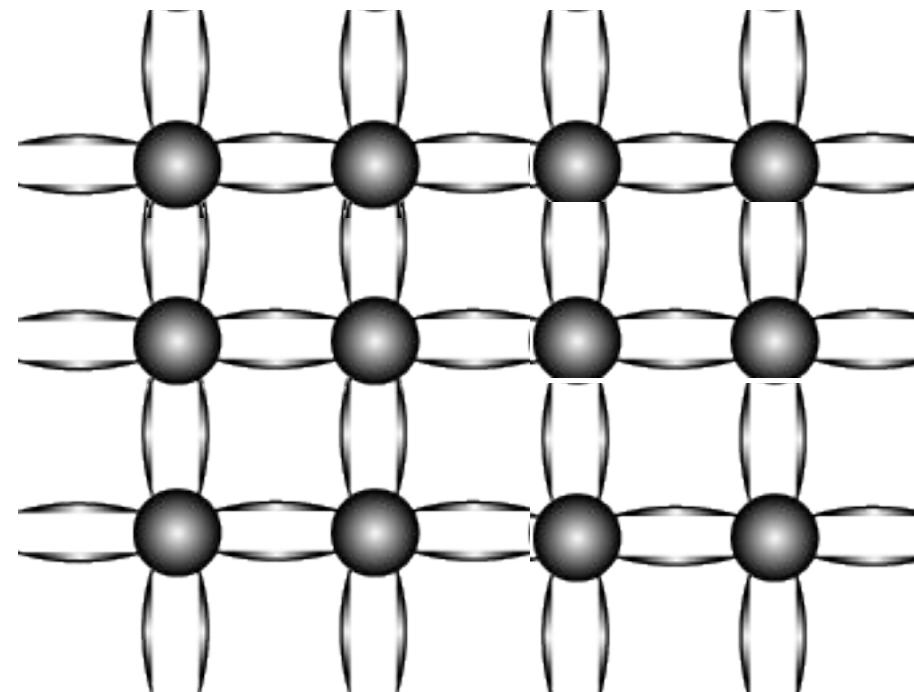


4 valence electrons

Each atom shares 2 electrons
with 4 nearest neighbors to form
a covalent bond

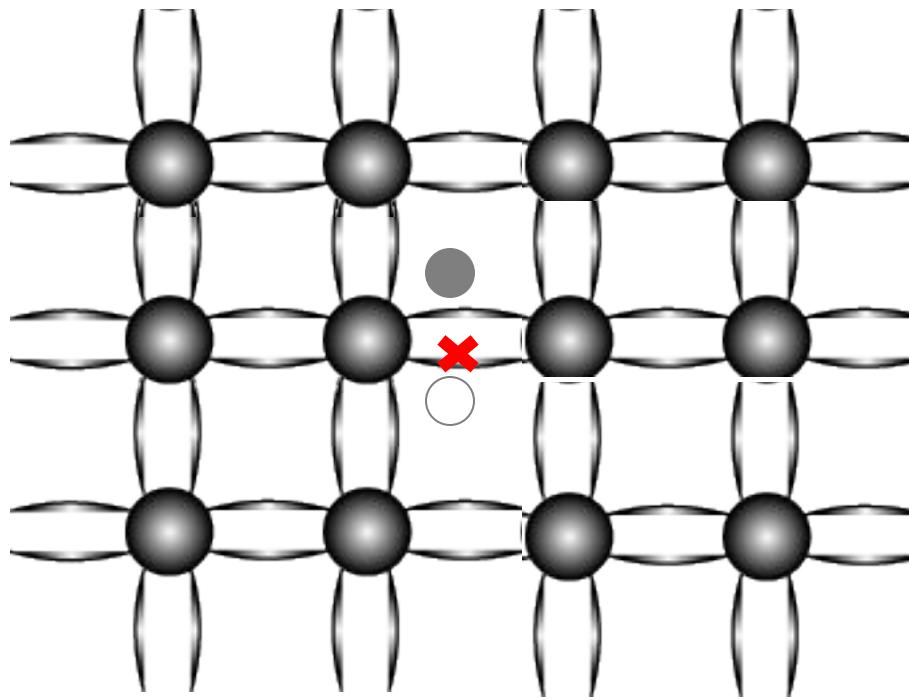
The Bond Model

Each atom shares 2 electrons with 4 nearest neighbors to form a covalent bond



At T=0K, all bonds are satisfied, there are no **free** carriers, no current flows,
looks like an insulator

Intrinsic Si: The Bond Model: Electrons

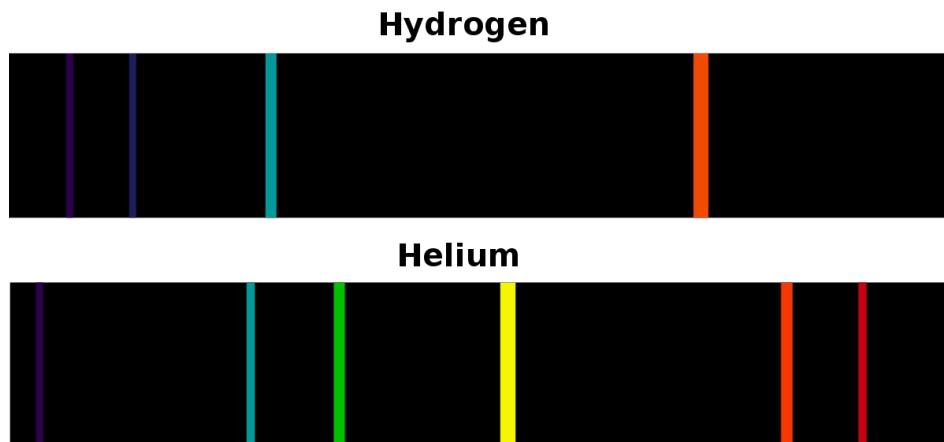


At finite temperature, an electron may gain enough energy to break the covalent bond, become **free** and move around.

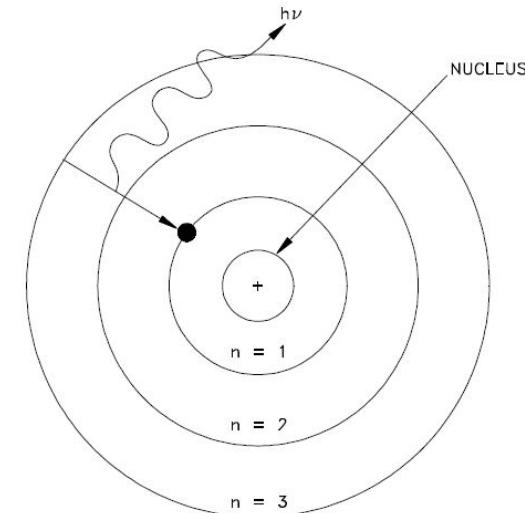
Energy Band Model

Electrons around nucleus

It was known from John Herschel's experiment in (1826) that heated gas emits a unique combination of colors

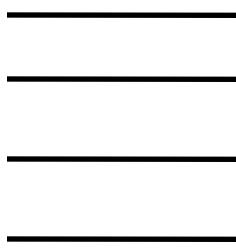


- In 1913 Niels Bohr proposed an atomic model that assumes electrons are orbiting around a positively charged nucleus in **specific shells**

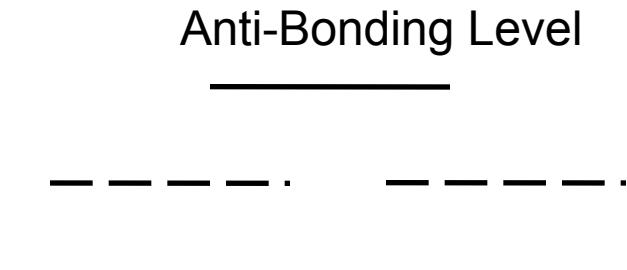


- When heated electrons can absorb the energy and go from shell 1 to 2. When cooling down, it comes down to 1, **emitting** the specific energy difference between 2 and 1 giving a specific color of light.

Energy Levels and Formation of a Molecule



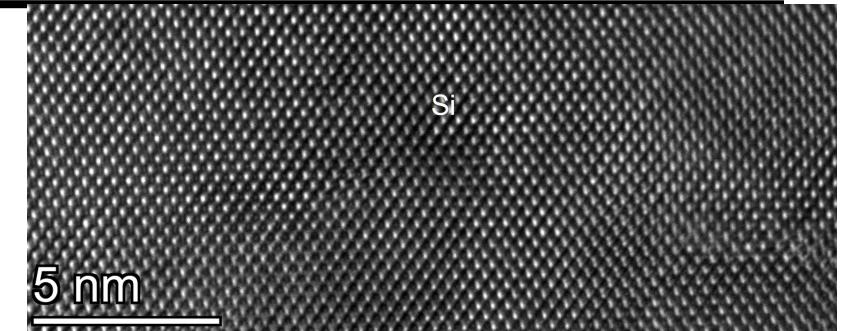
Discrete energy levels in an atom



When energy levels of two atoms interact, they create one bonding and one anti-bonding level

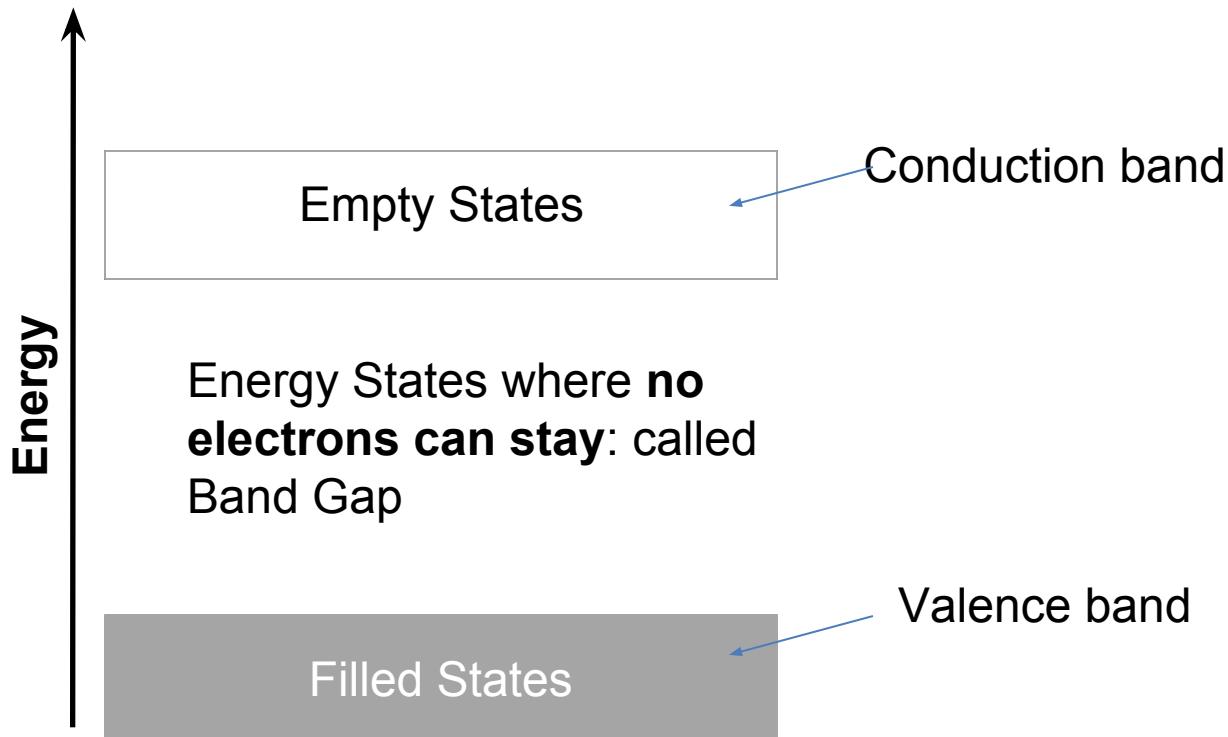
Energy Bands

In a solid as many atoms are brought close to each other they create many many bonding and anti-bonding levels

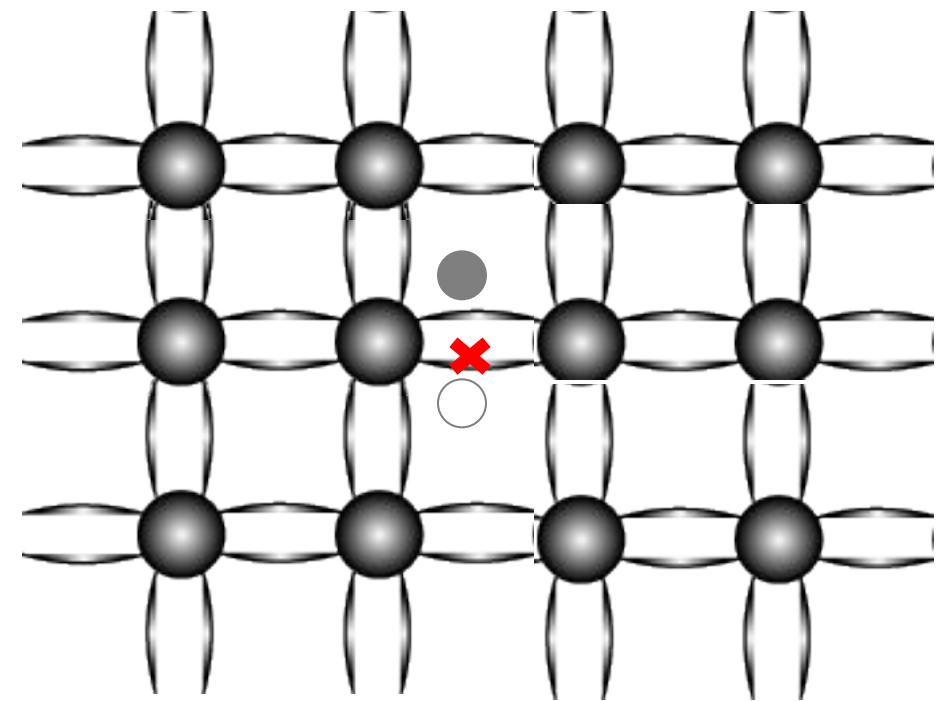
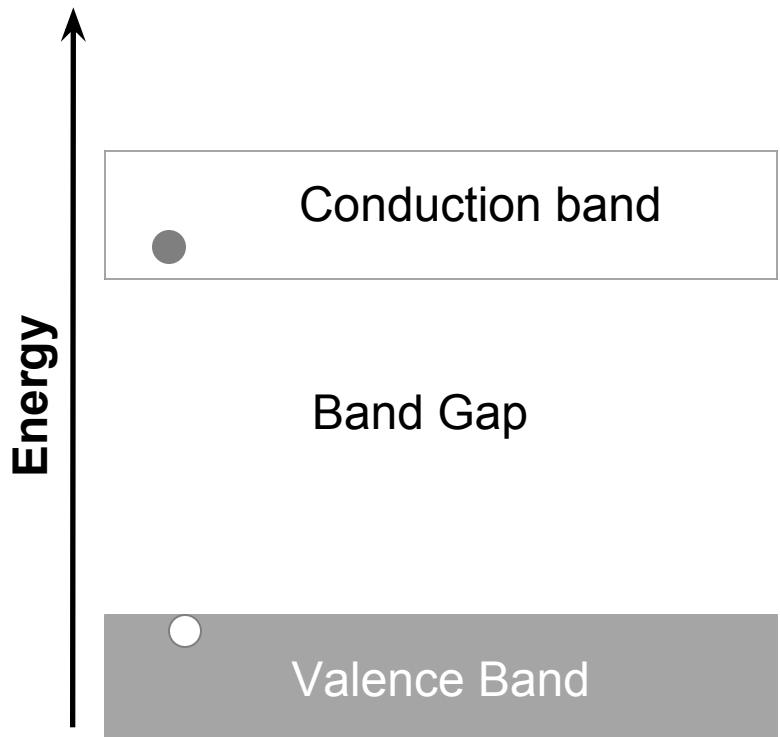


Energy Bands

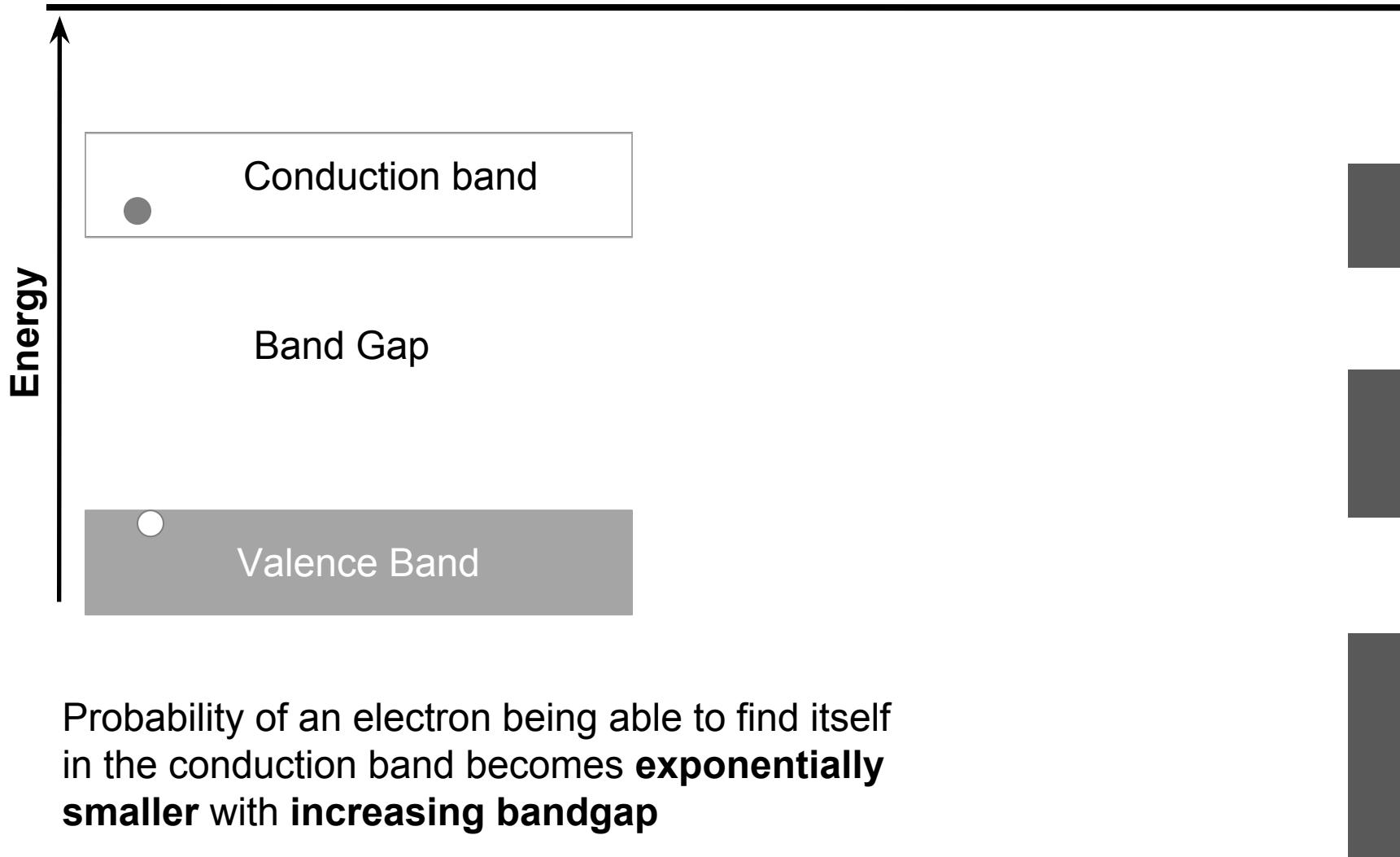
Energy Bands



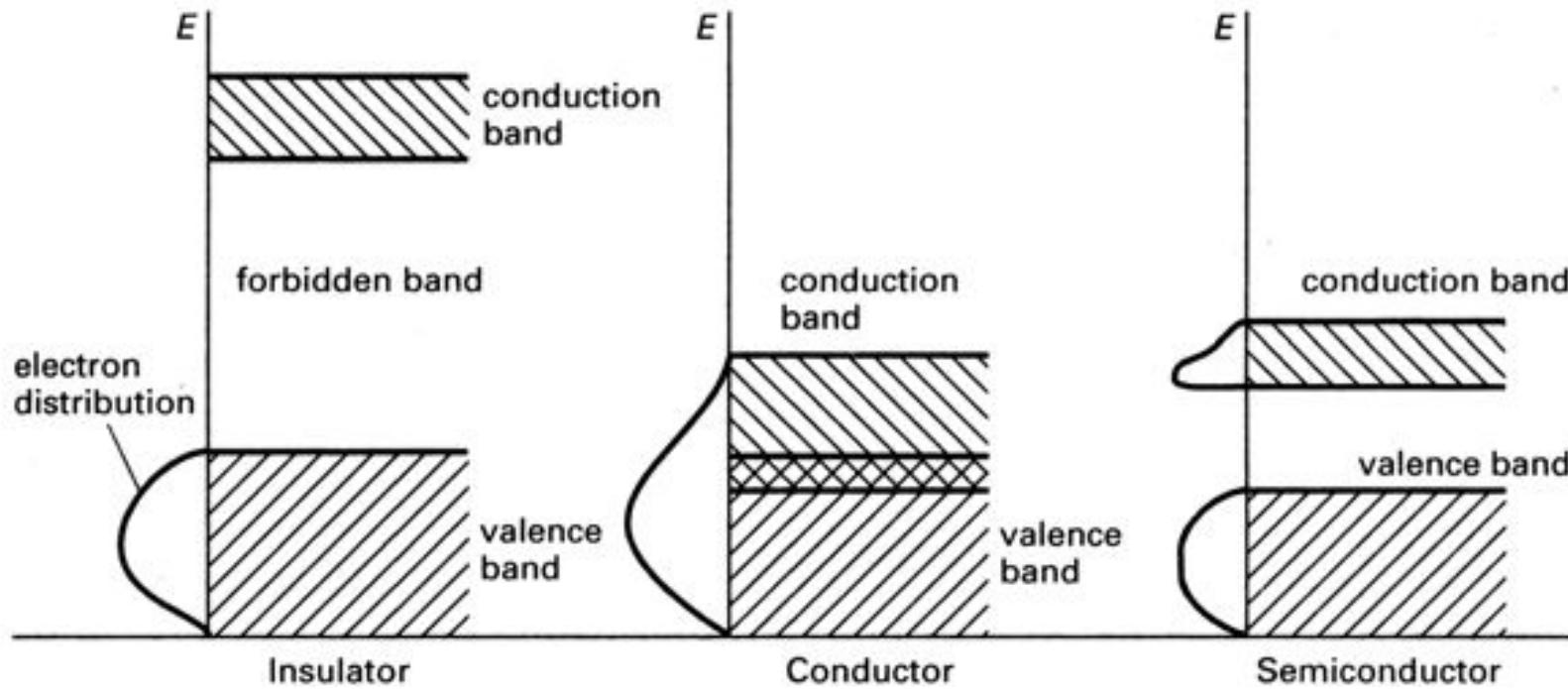
Energy Bands



Probability of an electron being free

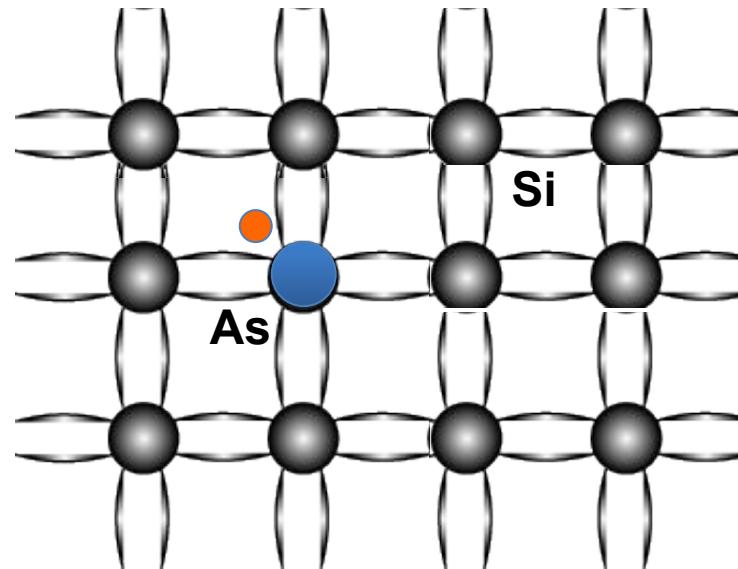
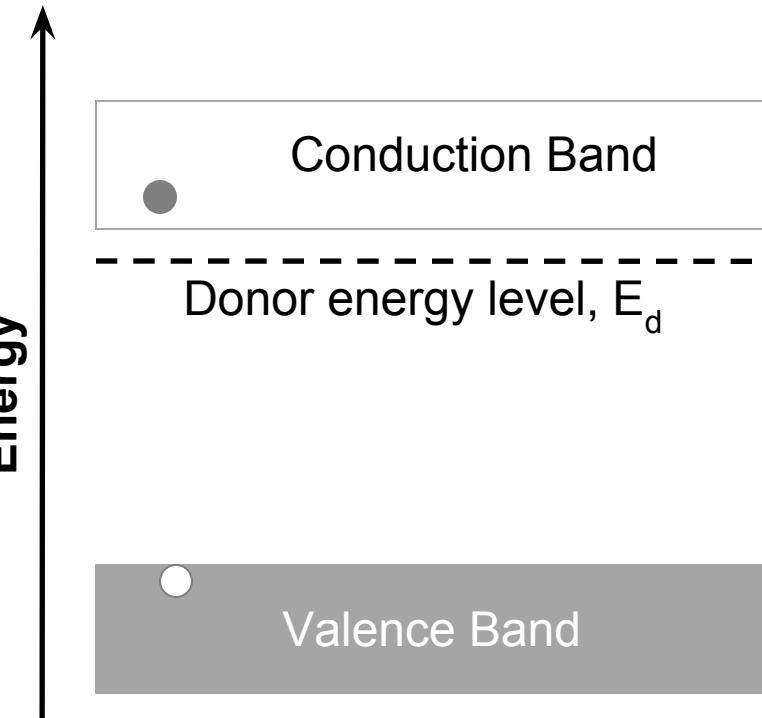


Semiconductors, Insulators and Conductors



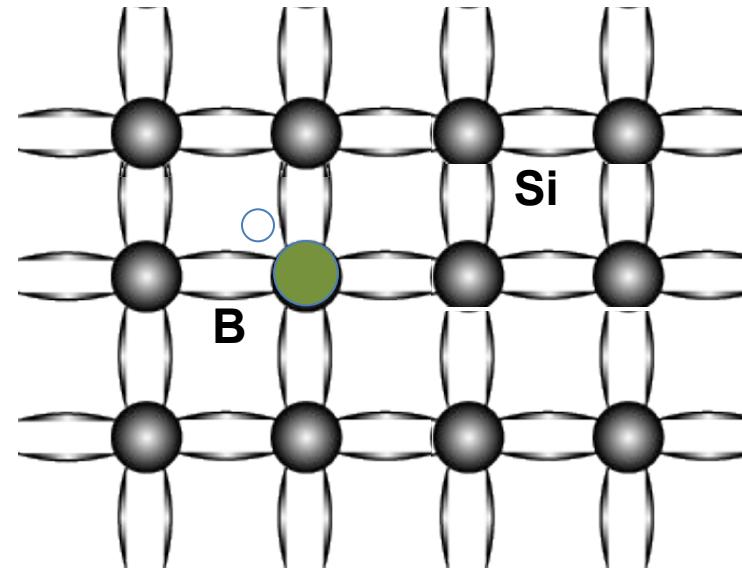
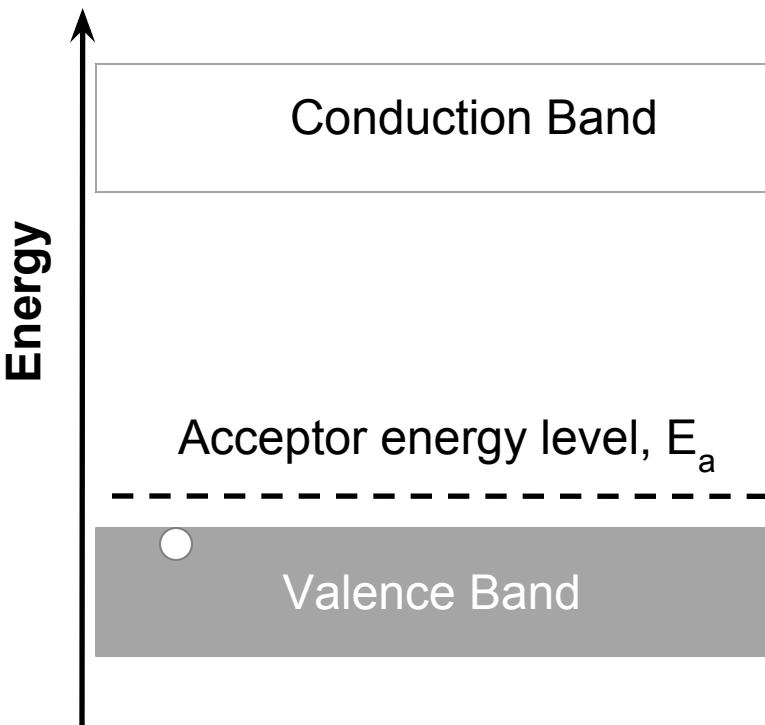
- ✓ Conductors have half filled bands
- ✓ Semiconductors have lower energy gap compared to insulators and can be doped

Doping



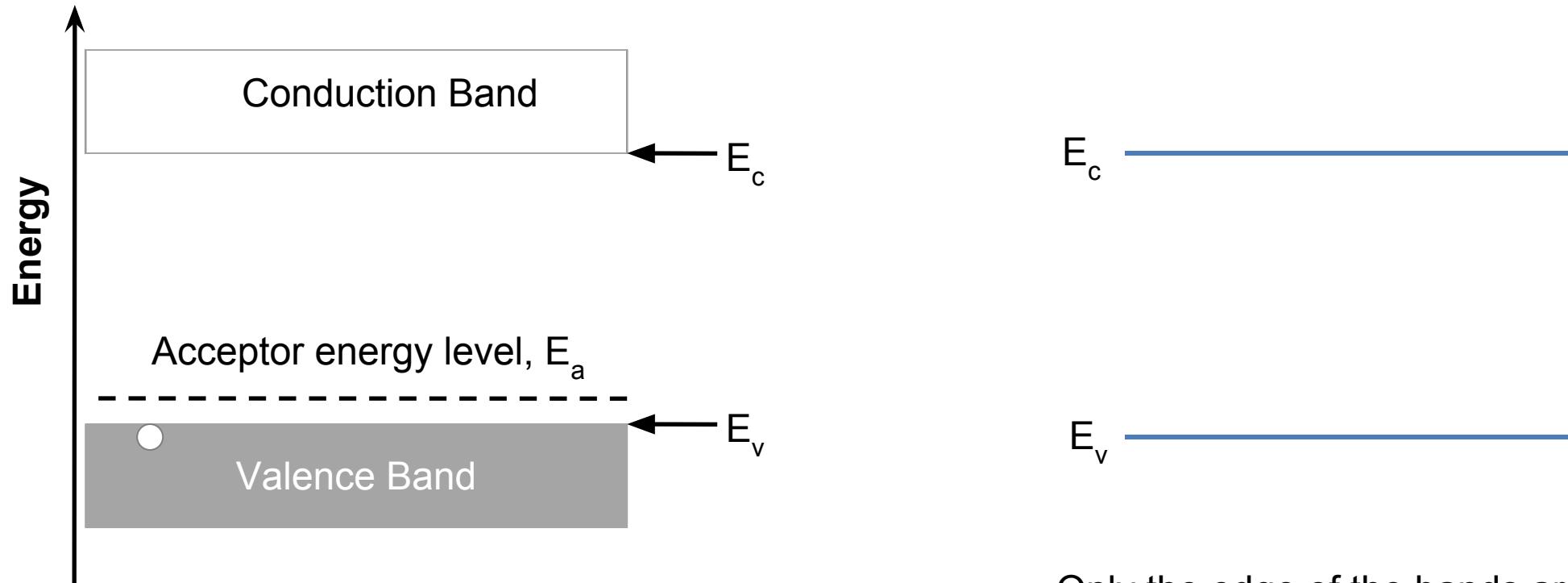
	III	IV	V	
	Boron (B)	Carbon (C)		
...	Aluminum (Al)	Silicon (Si)	Phosphorous (P)	...
	Galium (Al)	Germanium (Ge)	Arsenic (As)	
		•	•	•

Doping



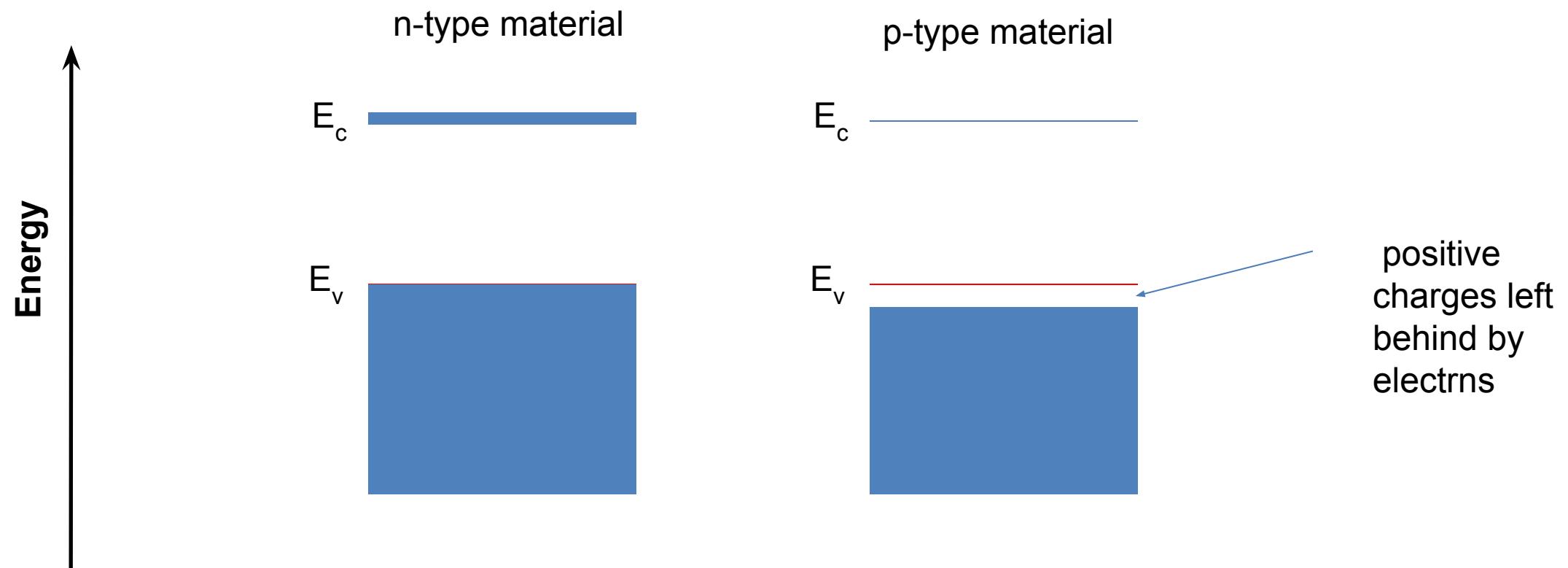
	III	IV	V	
	Boron (B)	Carbon (C)		
...	Aluminum (Al)	Silicon (Si)	Phosphorous (P)	...
	Galium (Al)	Germanium (Ge)	Arsenic (As)	
		• • •		

A convention about energy bands



Only the edge of the bands are shown where the difference between the two edges is the bandgap

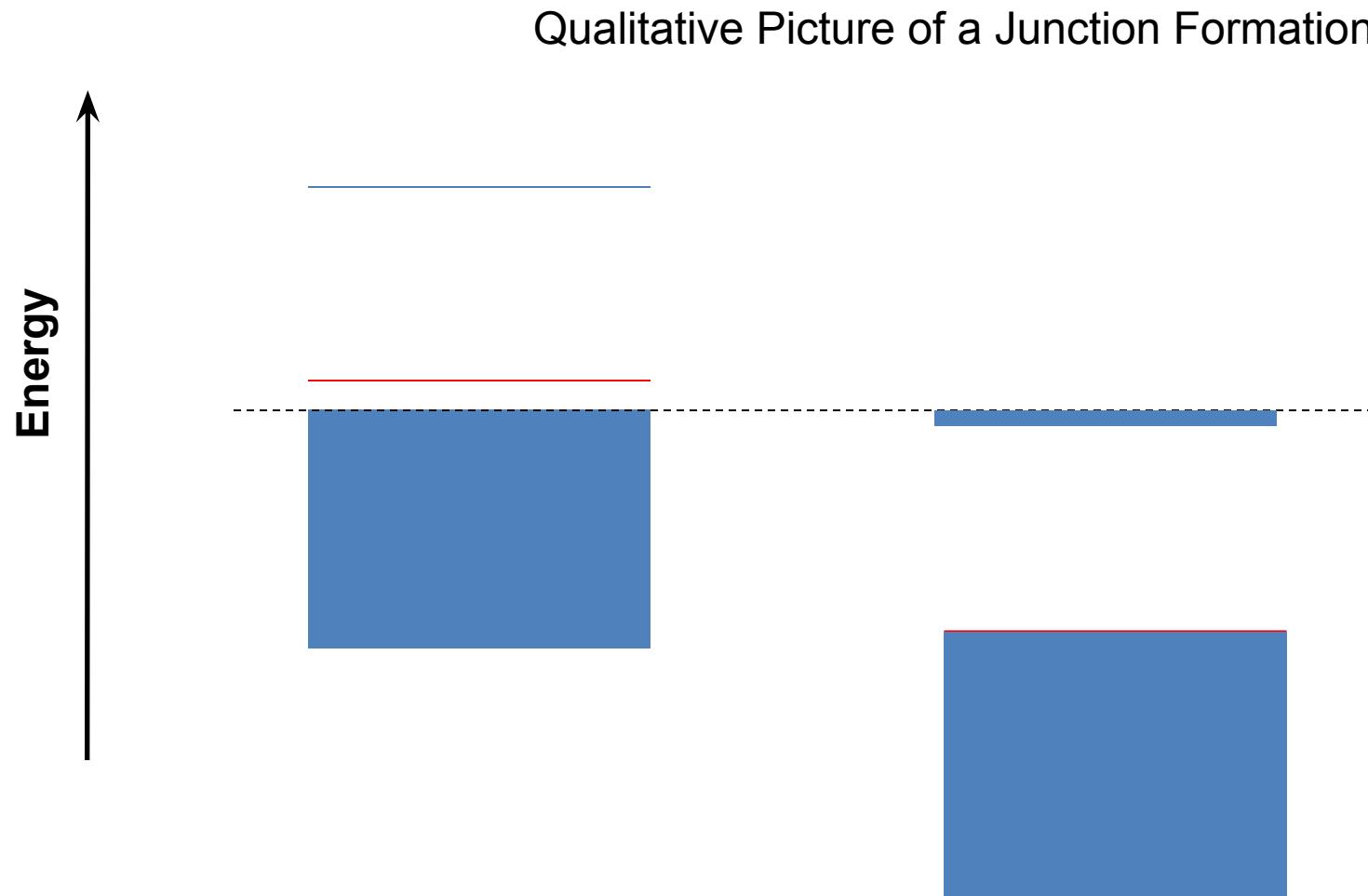
N and P type Materials, Junctions and Devices



*Blue color indicates electrons

Combining N and P materials

N and P type Materials, Junctions and Devices

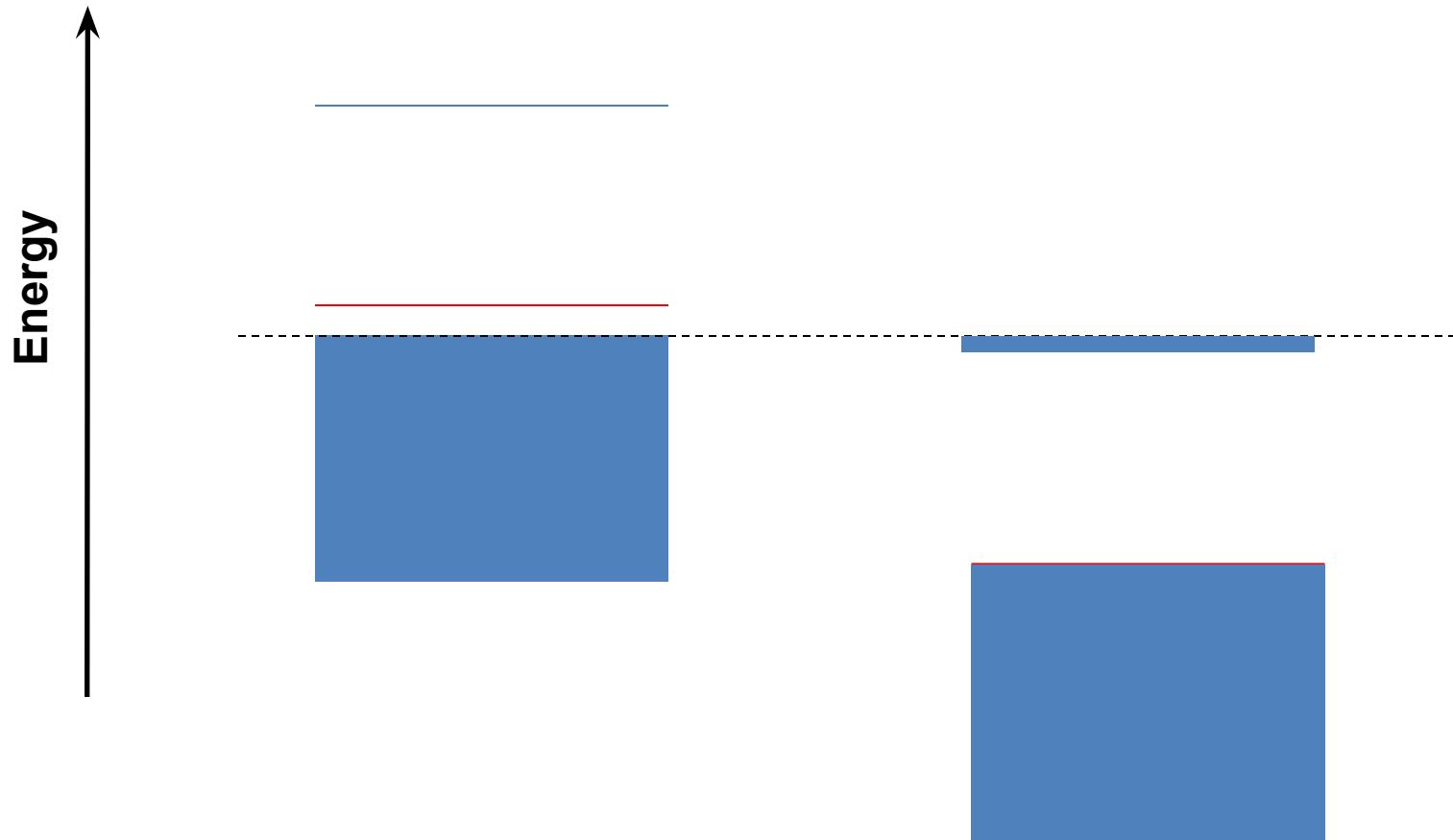


- When a n and p are put together, they form a p-n junction diode
Symbol:

p-side n-side
- Electron densities align in energy so that there is no difference in concentration
- Technically what aligns is the energy level where probability of finding an electron is $\frac{1}{2}$ to be discussed in more details in EE130

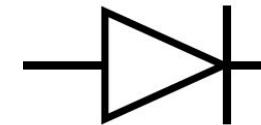
What does a voltage do?

Qualitative Picture of a Junction Formation



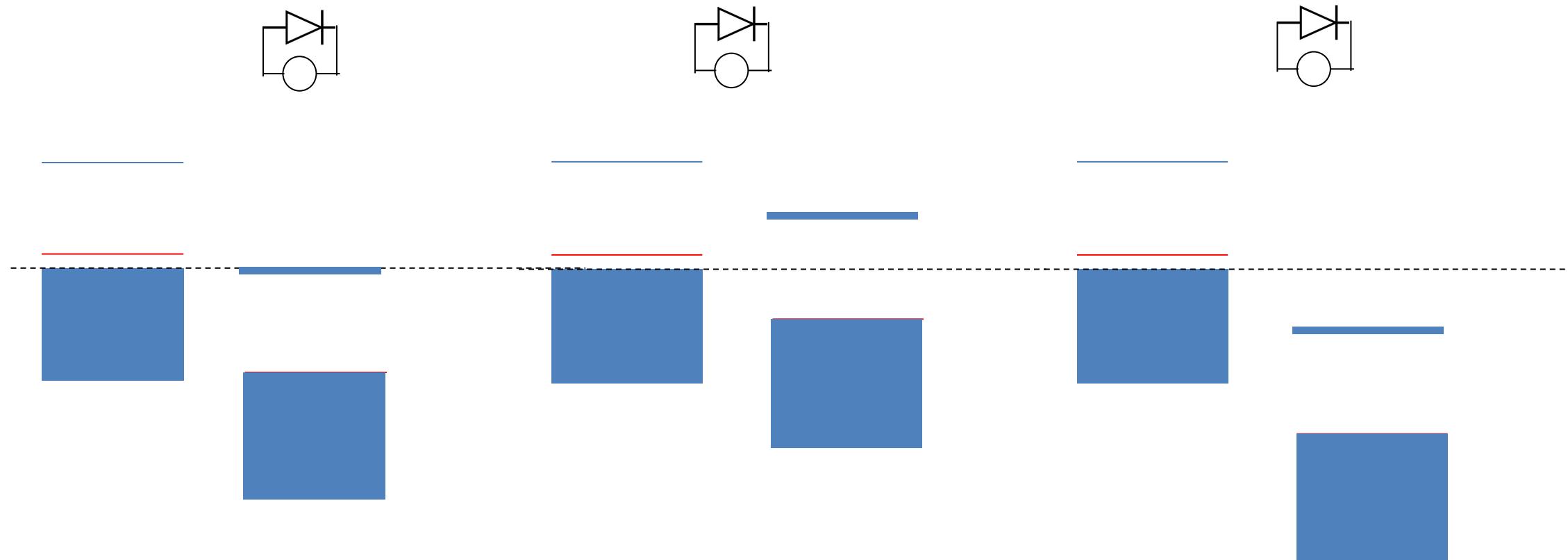
p-side

n-side



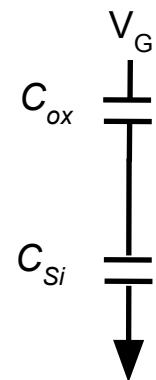
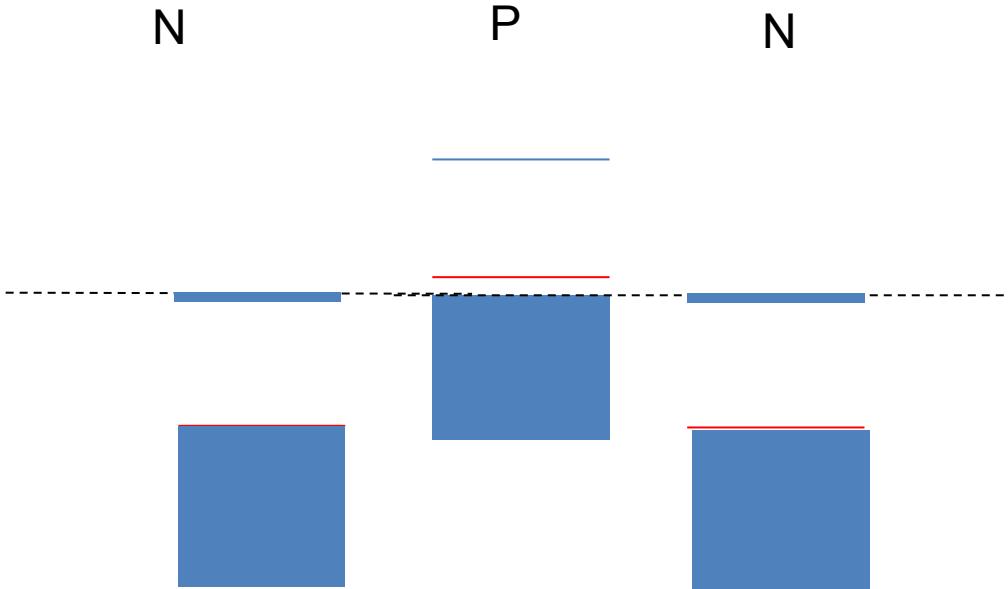
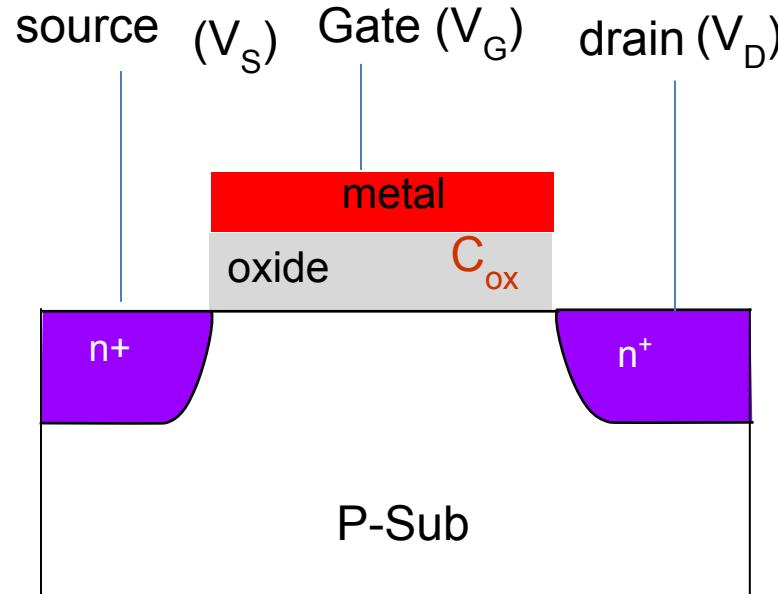
Negative terminal of a battery
brings electrons and thereby
increases energy.

N and P type Materials, Junctions and Devices



I-V of a PN junction Diode

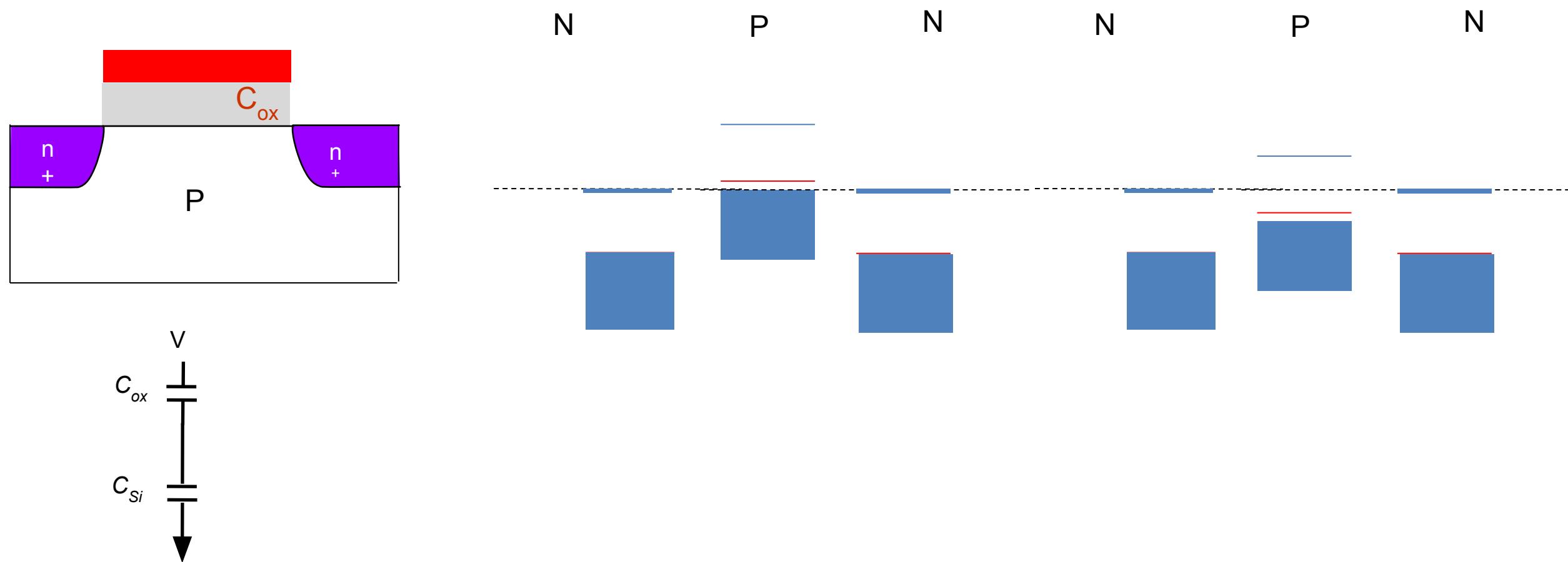
Metal-Oxide-Semiconductor Field Effect Transistor (MOSFET)



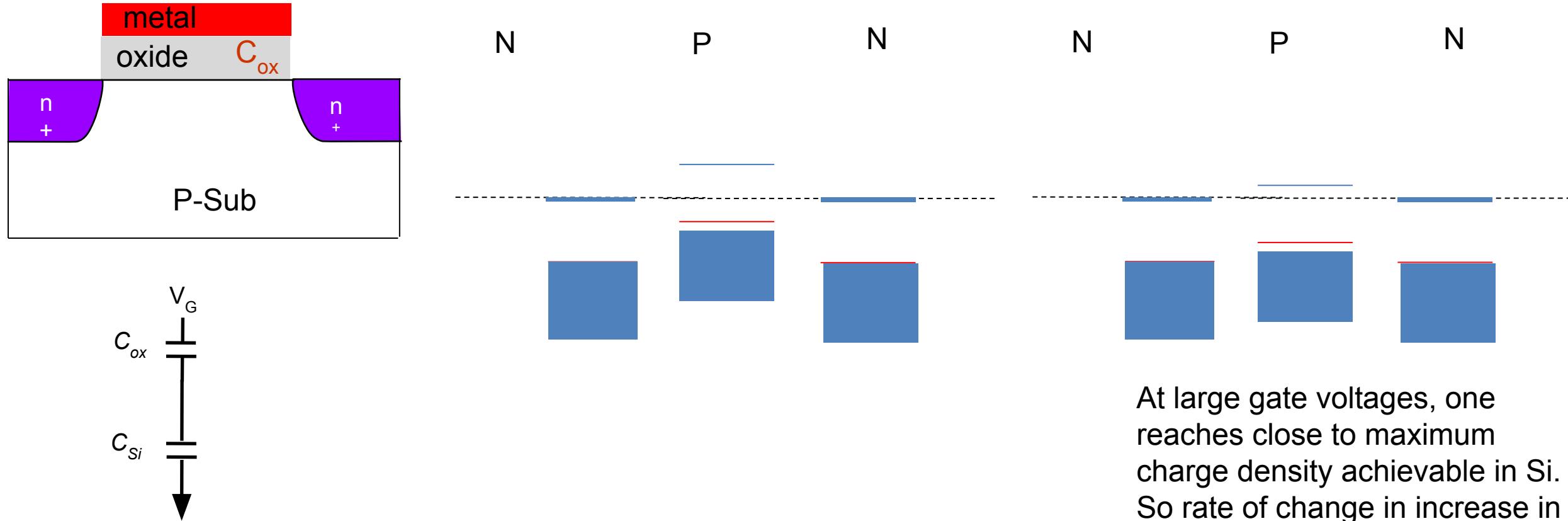
- + or – in the name of n or p type material indicates extent of doping. N+ means doped **heavily** to n type.
- In common MOSFET source and drain voltages are interchangeable

P-type semiconductor in the middle with little to no electrons on the conduction band acts like an insulator

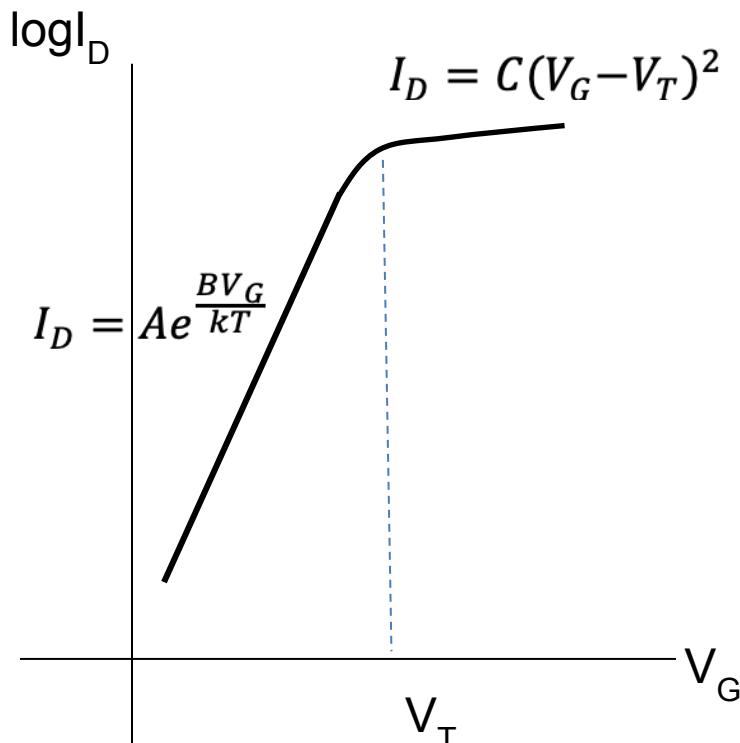
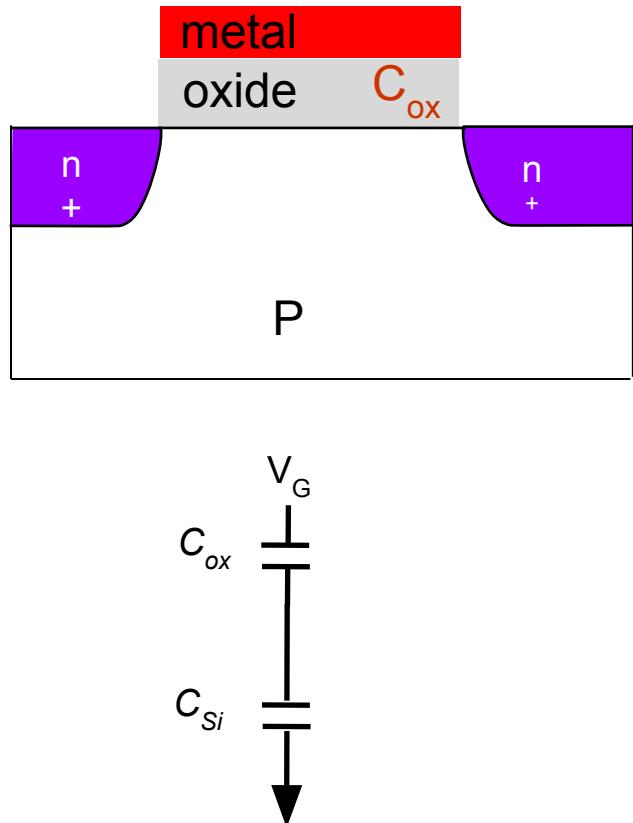
MOSFET



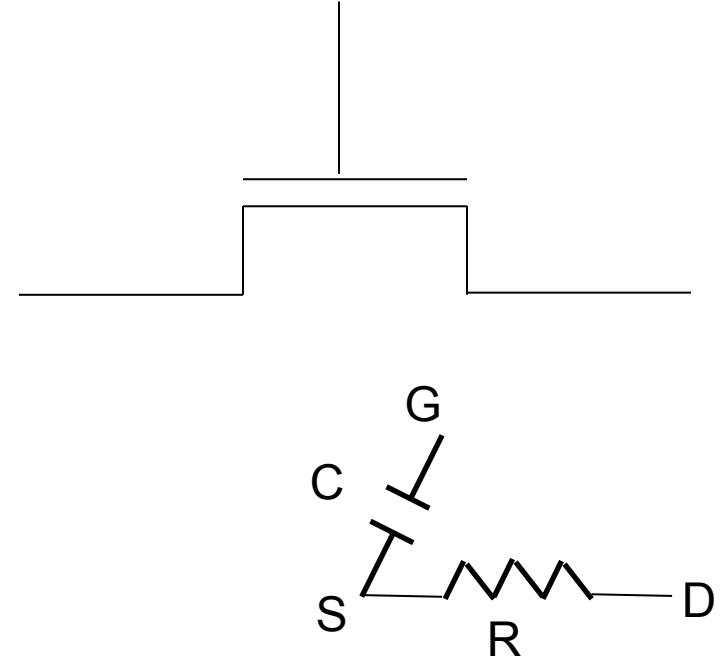
MOSFET



MOSFETs



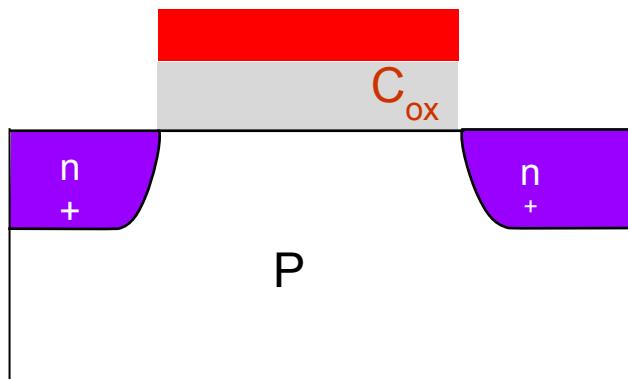
Assuming large V_{DS} is present



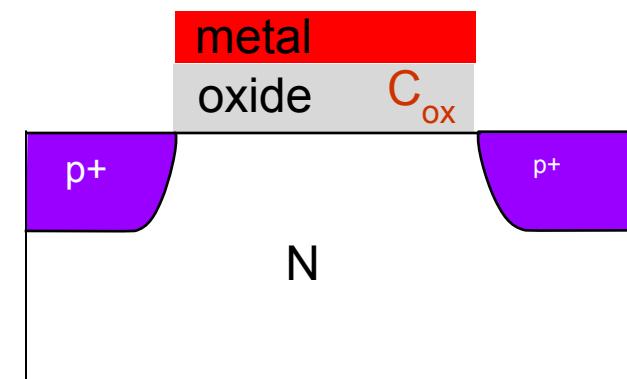
- C is the series combination of C_{ox} and C_{Si}
- $R = I_{DS}/V_{DS}$

nFET vs pFET

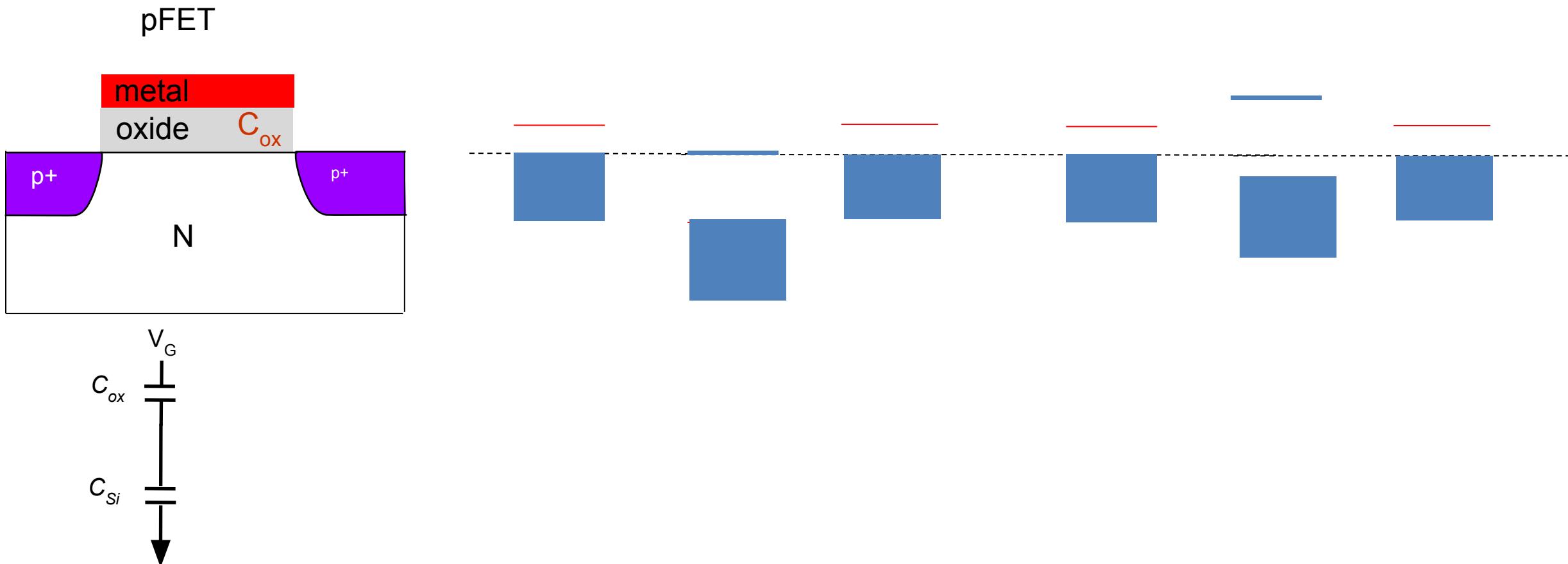
nFET



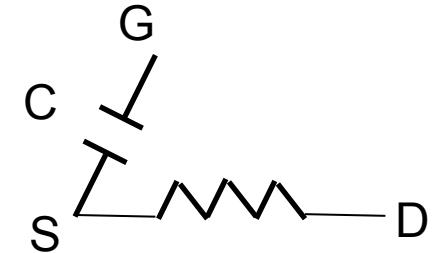
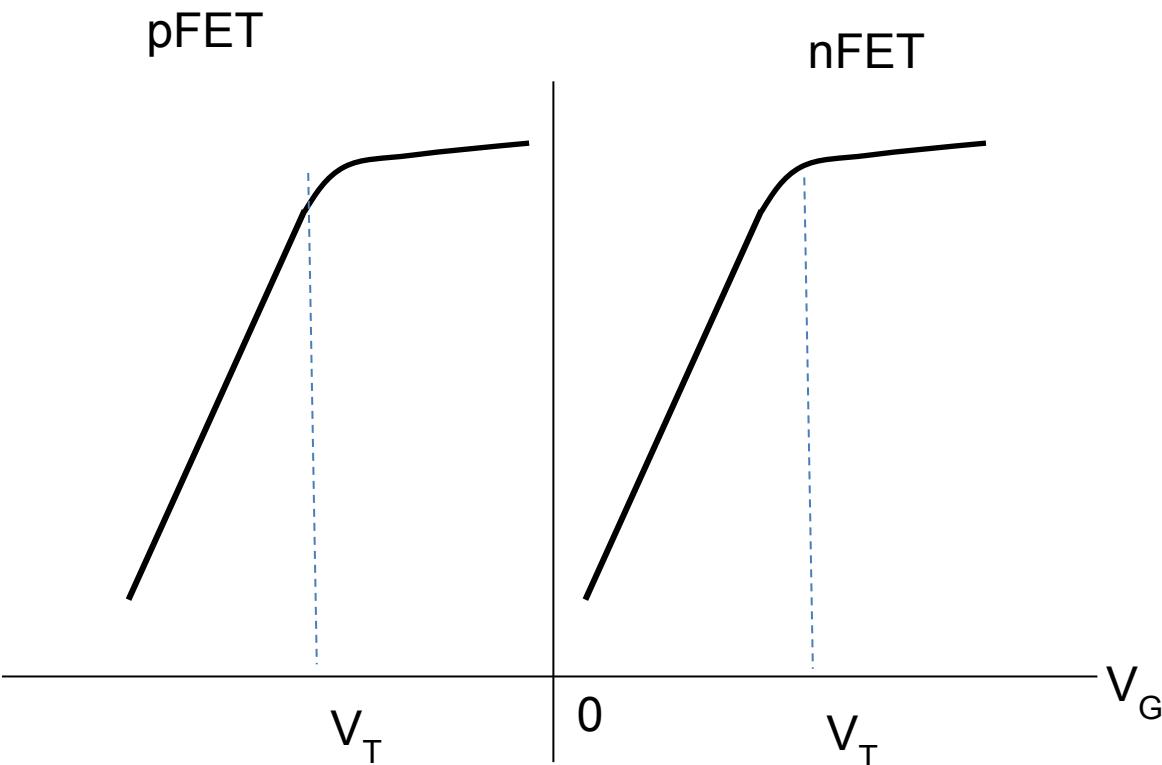
pFET



nFET vs pFET

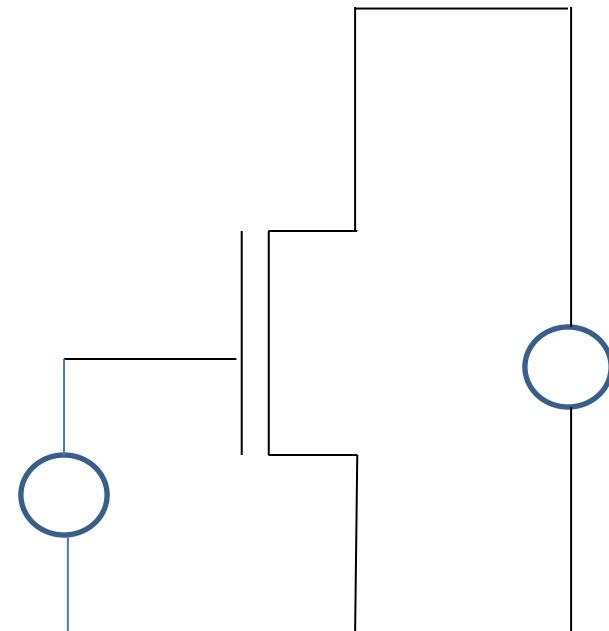


nFET vs pFET



- nFET, V_{GS} and V_T are positive
- pFET, V_{GS} and V_T are negative

FET as an analog amplifier



When $V_{GS} < V_T$

$$I_D = A e^{\frac{BV_G}{kT}}$$

Small change in V_{GS} changes I_D exponentially

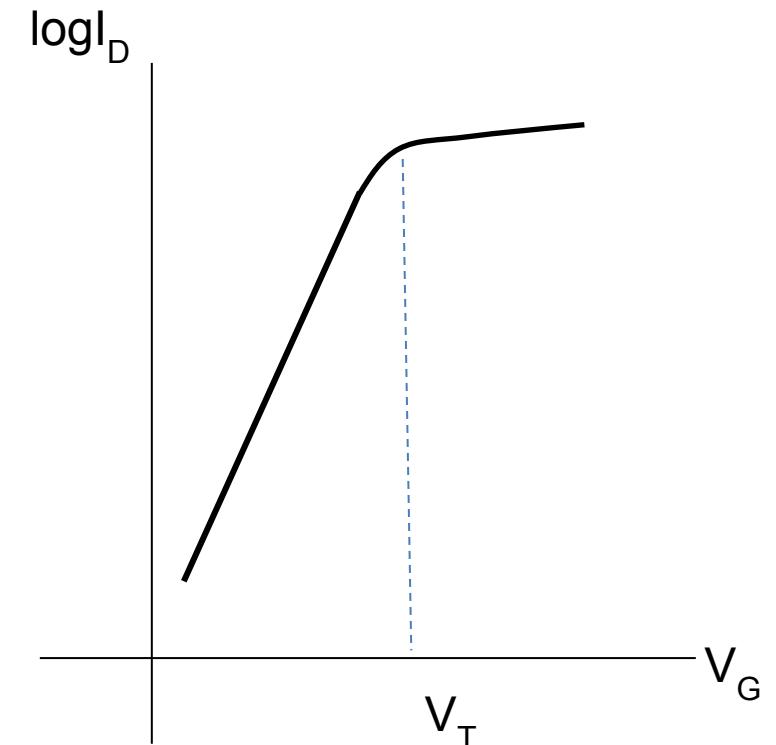
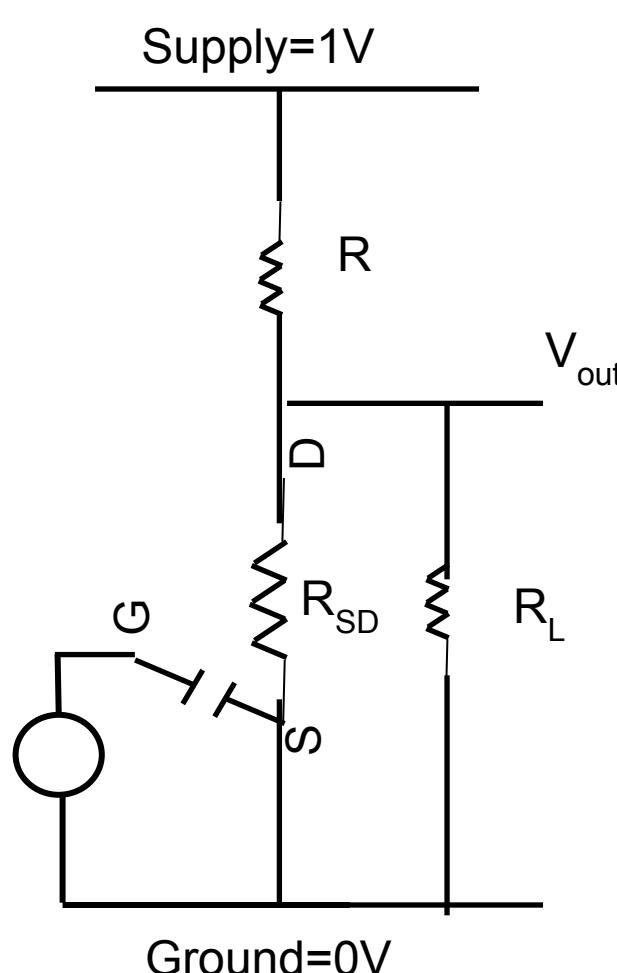
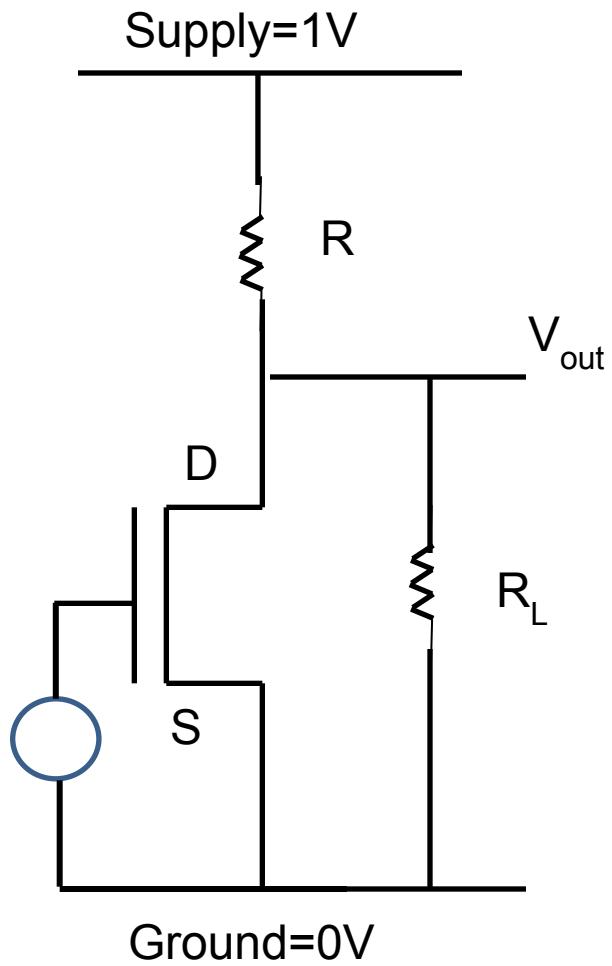
When $V_{GS} > V_T$

$$I_D = C(V_G - V_T)^2$$

Small change in V_{GS} changes I_D quadratically

Overall, Large changes in the Drain current can be achieved by changing Gate Voltage

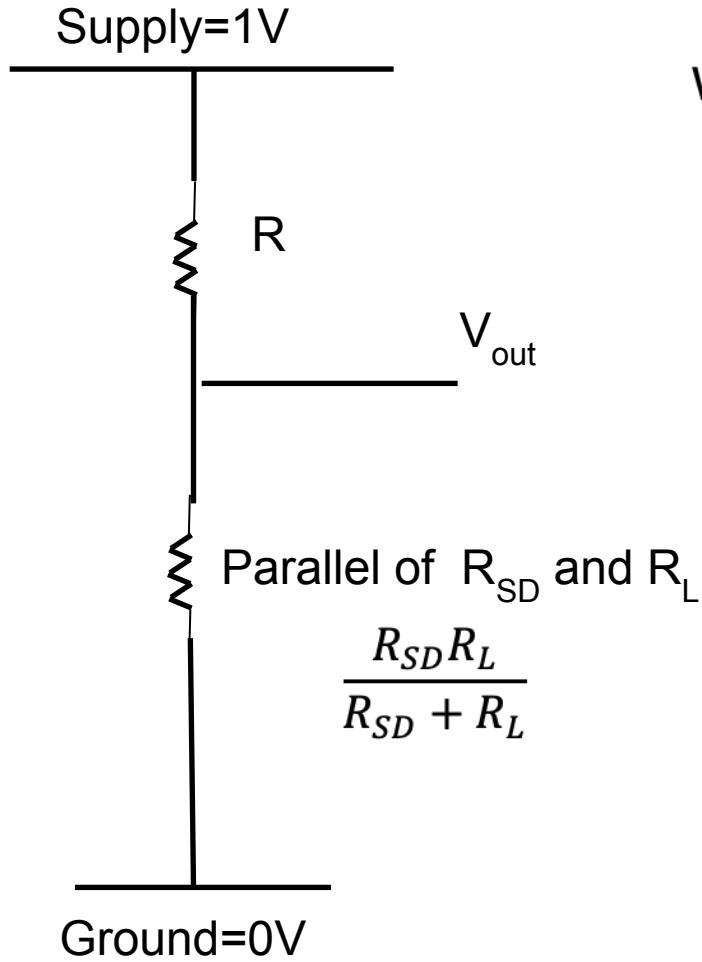
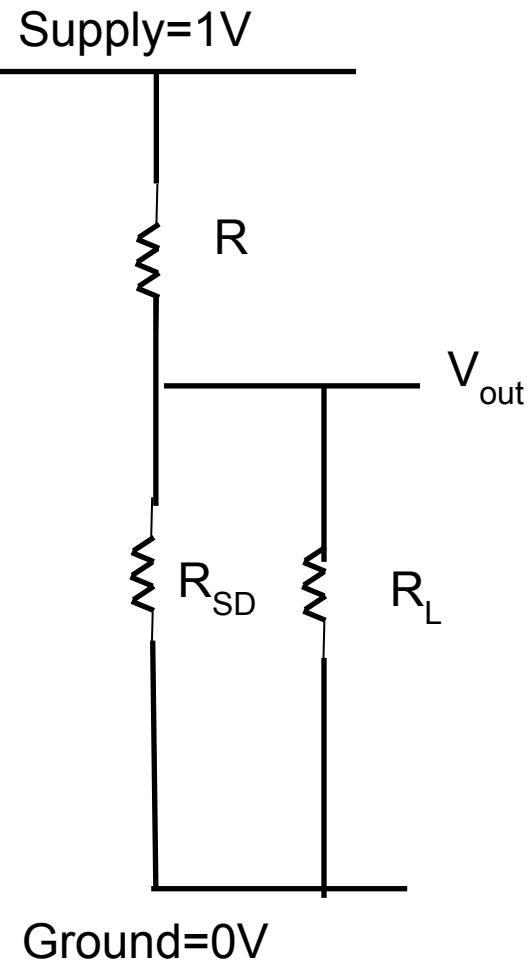
FET in digital logic



When V_{GS} is High, R_{SD} is low

When V_{GS} is Low, R_{SD} is High

FET in digital logic

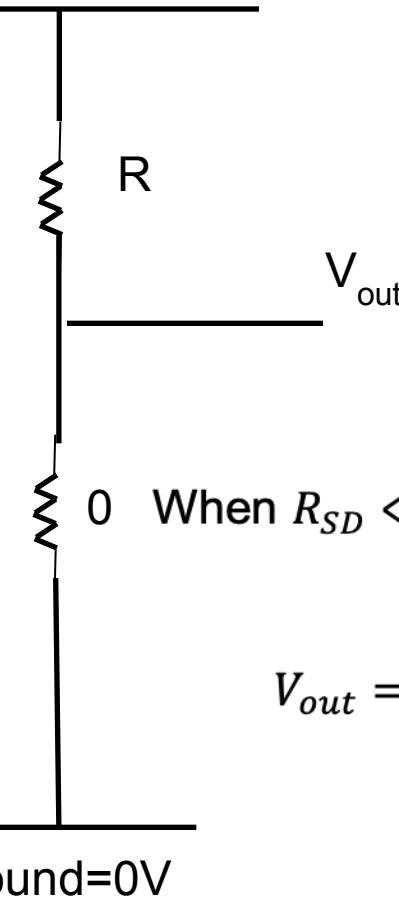


When $R_{SD} \ll R_L$; $\frac{R_{SD}R_L}{R_{SD} + R_L} \approx \frac{R_{SD}}{R_L} \approx 0$

When $R_{SD} \gg R_L$; $\frac{R_{SD}R_L}{R_{SD} + R_L} \approx R_L$

FET in digital logic

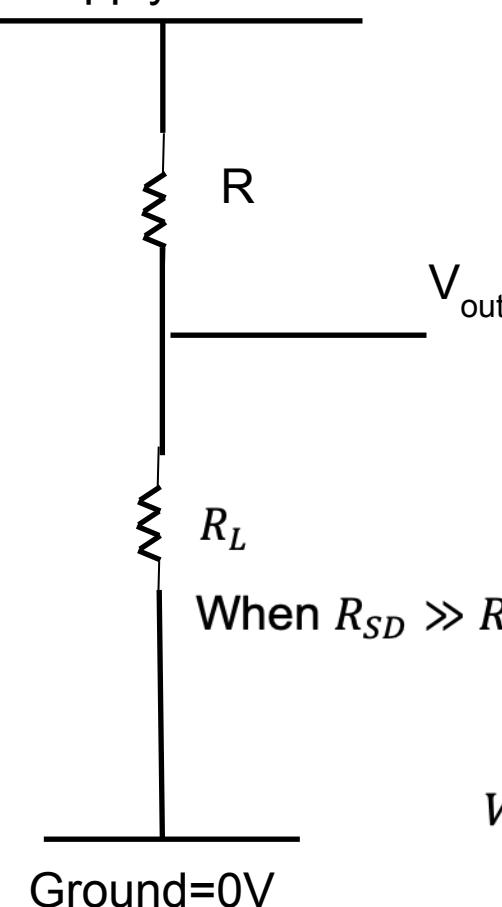
Supply=1V



0 When $R_{SD} \ll R_L$ i.e., V_{GS} is high

$$V_{out} = \frac{0}{R + 0} = 0$$

Supply=1V

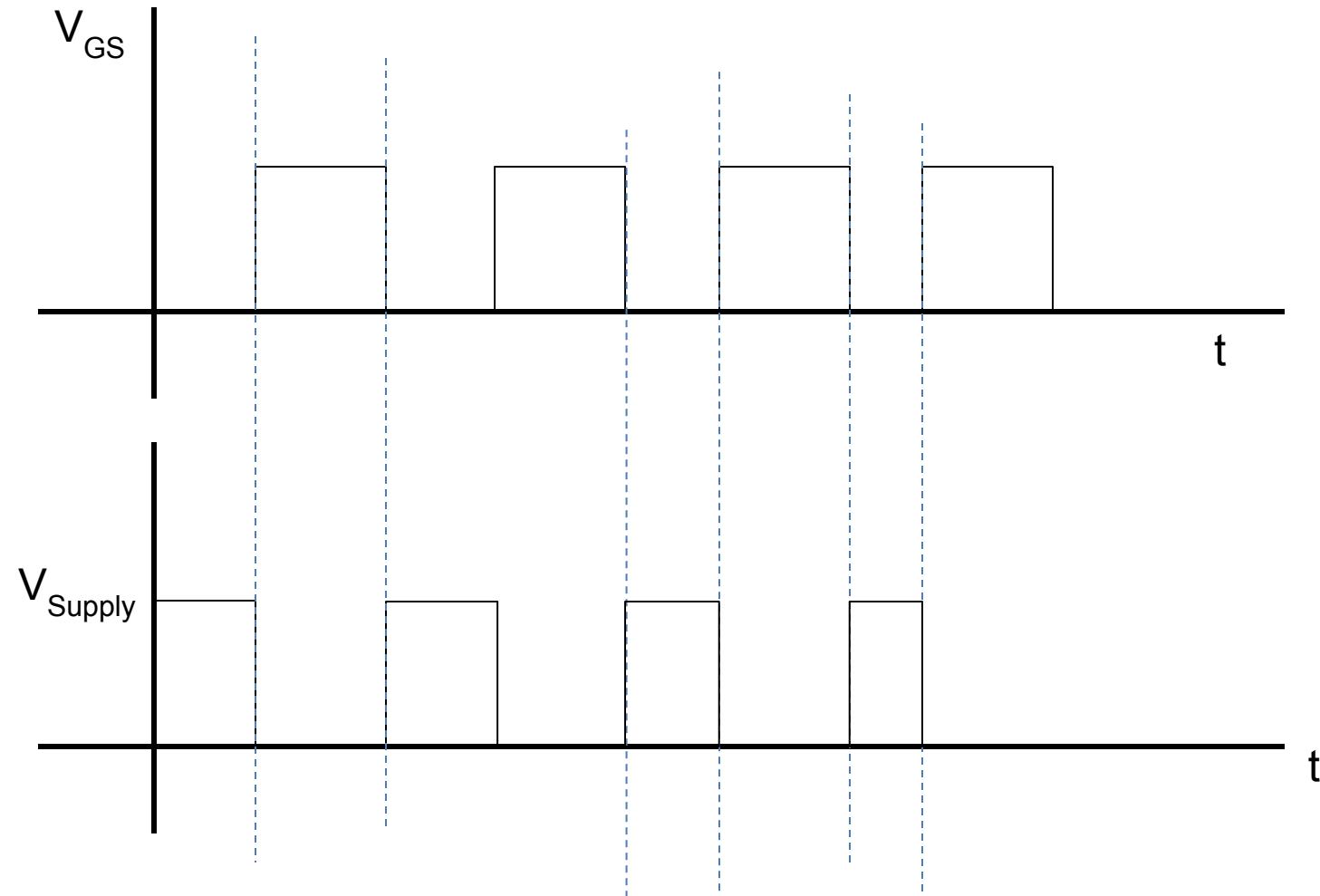


When $R_{SD} \gg R_L$; when V_{GS} is low

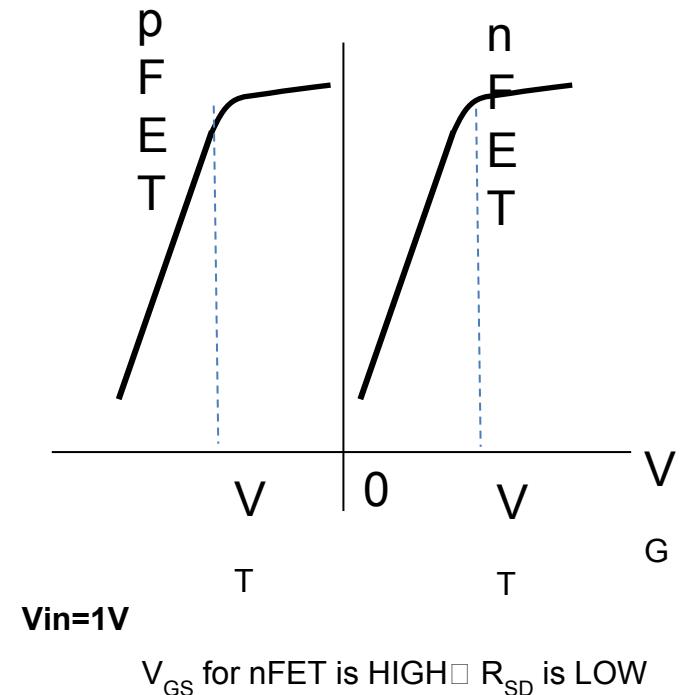
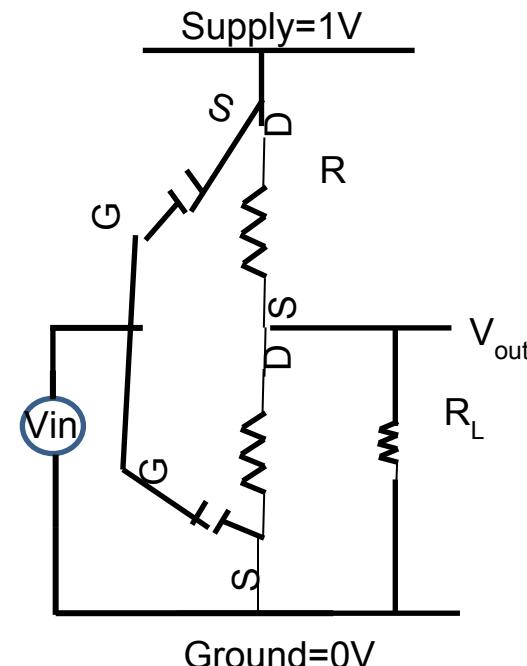
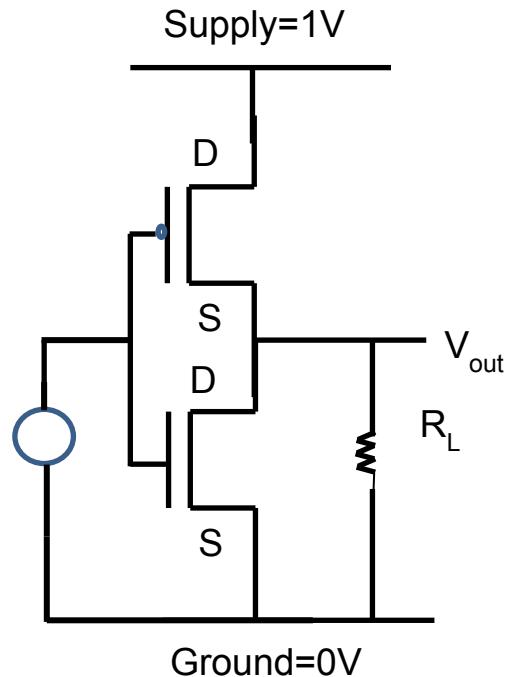
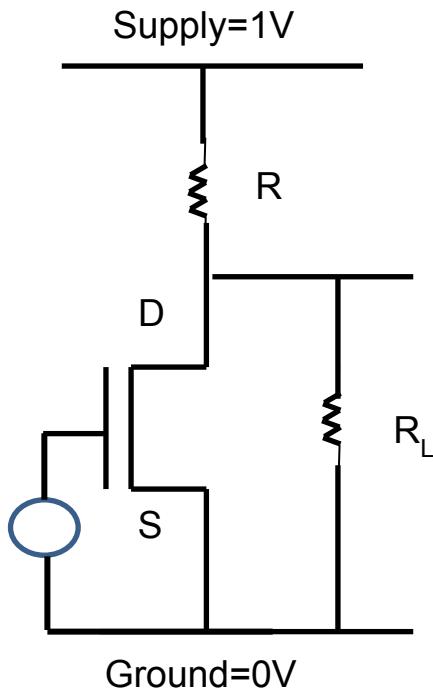
$$V_{out} = \frac{R_L}{R + R_L} V_{supply}$$

$\approx V_{supply}$ if we design $R \ll R_L$

FET in digital logic



CMOS



$V_{in}=0V$

V_{GS} for nFET is LOW \square R_{SD} is HIGH

V_{GS} for pFET is HIGH **NEGATIVE** \square R_{SD} is LOW