Project 1 report: Hartree Fock with Integrals

Benjamin Peyton

December 20, 2017

1 Overview

For this project, a code was developed according to the directions given in the Final Project 1

Description pdf. The code for the project spans across three files: ints.py, hf.py, and opt.py.

The integral evaluation code is split into functions within ints.py, the Hartree Fock method is

split into functions (and a total HF function is also implemented) within hf.py which calls on

the integral functions from ints.py for the required integral matrices, and finally opt.py calls

the HF procedure and uses it to perform a geometry optimization on \mathcal{H}_2 using Newton-Raphson

steps with approximated derivatives. The complete code can be found on my github page at

github.com/bgpeyton/HF_Int along with a README which outlines the code and describes

dependencies (math, numpy, and scipy packages).

2 Derivations and Procedure

2.1 Integral Evaluation

Before integral evaluation, a closed expression for the integral in Eq (1) was derived to be used

later for Gaussian-type integrals (derivations separate).

 $R_k = \int_{-\infty}^{\infty} x^k e^{-\alpha x^2} dx \tag{1}$

Next, using the Gaussian product theorem, a closed expression for the overlap integral

between two primitive non-normalized Gaussians was derived. The function gen_S() was coded

in ints.py to evaluate these overlap integrals, and another function S_mat() returns a matrix

of such integrals given a molecule and basis set. A normalize() function was also implemented,

1

which returns the inverse square root of the self-overlap of a given basis function to be used as a normalization constant for S and all future matrices.

For kinetic energy integrals, again a closed form expression was derived. This expression was implemented in gen_K() within ints.py and then K_mat() generates a matrix of these integrals given a molecule and basis set.

An expression for nuclear attraction integrals was given, and this was implemented as gen_V() within ints.py and followed by V_mat(). Similarly, the expression for the electron repulsion integrals was given and implemented as gen_eri(), followed by eri_mat() which returns the later-discussed G-matrix necessary for the Hartree Fock procedure.

2.2 Hartree Fock Procedure

Evaluation of the Hartree Fock equations (Eq (2)) is made relatively simple using the functions defined in ints.py. The core Hamiltonian matrix \mathbf{H} is directly solved from the \mathbf{T} and \mathbf{V} matrices solved using the corresponding generating functions from ints.py. An orthogonalization matrix $\mathbf{S}^{-1/2}$ is generated and used to diagonalize the Hamiltonian, and the eigenvectors c are used to form the density matrix according to Eq (3).

$$\mathbf{H}\mathbf{c}_i^{(0)} = e_i \mathbf{S}\mathbf{c}_i^{(0)} \tag{2}$$

$$P_{ij} = \sum_{m=1}^{N/2} c_{mi} c_{mj} \tag{3}$$

The G matrix is then computed using the before-mentioned eri_mat() and the Fock matrix F is computed as the sum of G and H, to be used in the next HF step. An initial energy is computed (first using H, then using F in subsequent iterations) using Eq (4), and this process is repeated until convergence of the energy is reached. A single-HF-loop function loop_hf() was designed to take a molecule, basis, H, and F matrices. This function is then called in a loop within the full_hf() function, which computes the entire HF energy from start to finish with only a molecular coordinate array and basis set array as arguments (with optional convergence argument, the default being 10^{-12}).

$$E = \sum_{i=1}^{M} \sum_{j=1}^{M} (2H_{ih} + G_{ij}) P_{ij}$$
(4)

2.3 Geometry Optimization

A geometry optimization was implemented in opt.py. The first (E') and second (E'') derivatives of the energy are approximated using small 0.001au steps in the geometry (R), then used to take Newton-Raphson steps according to Eq (5). This is repeated until the absolute value of E'' was less than $10^{-6}au$.

$$R^{new} = R - \frac{E'(R)}{E''(R)} \tag{5}$$

3 Results

The Hartree Fock geometry optimization procedure produced values comparable (though not exactly equivalent to) a Psi4 output using similar methods and starting values. The input and output are included as input.dat and output.dat. The bond distance, energies, derivatives, and second derivatives of each step are printed by opt.py and are included in Table (1) for an H_2 molecule starting with the atoms 1au apart and using the basis set provided in the project pdf.

Table 1: Energies and derivatives

R (au)	E (au)	E' (au)	E" (au)
1.0	-1.07263860249	-0.32892177304	1.53767876387
1.21390798961	-1.11524236275	-0.097629478193	0.735004021202
1.34673648493	-1.12257966082	-0.0185942960007	0.474442480858
1.38592837172	-1.12295919264	-0.00114918519656	0.416997325114
1.3886842292	-1.1229607803	-5.20953924443e-06	0.413221389106

Psi4 predicts an equilibrium bond distance of 0.734828003532Å, or roughly 1.388623575au, with an energy of -1.122965336026au. These are in good (though not perfect) agreement with the calculated values from opt.py, with discrepancies small enough to be attributed to slight differences in the basis set and procedure used in the calculation.

The NIST webbook lists the equilibrium bond distance of H_2 as 0.74144Å, or roughly 1.40112au. This discrepancy is not surprising- Hartree Fock ignores a lot of contributions,

including electron correlation, and this is hardly an ideal HF calculation considering the small uncontracted basis used. The literature shows a bond dissociation energy (BDE) of -1.12296au, while this calculation predicts BDE = 2(-0.5)au - (-1.12296au) = 0.12296au, using the analytical solution of 0.5au per hydrogen atom. Again, this discrepancy is most likely due to shortcomings in the HF procedure.