## Naming Chemicals 2

Pre-lesson assignment – Textbook page 168-172

## You will need to learn to name these functional groups

	To harrie these farietic			
Structure	Name	Prefix	Suffix	Example
R	Alkene		-ene	
i i				
				Propene
Ŕ				Fropelie
R	Alkyne		-yne	/H
				II_C_C_H
R R				H_C
K				🗆
				Ethyne
$\wedge$	Arene	Phenyl-	-benzene	
		,		
				Phenylethane
				(
				Methylbenzene
	Dla a s a l			
ОН	Phenol		-phenol	4-methylphenol
				ОН
				]
11. 11				
-X	Haloalkane			
-Cl	Chloroalkane	Chloro-		CL CL
				CI-CH <sub>3</sub>
-F	Fluoroalkane	Fluoro-		Chloromethane
-Br	Bromoalkane	Bromo-		
-l	Iodoalkane	Iodo-		
-OH	Alcohol	Hydroxy-	-ol	
311	711001101	TTYGTOXY		/ `OH
				Ethanol
				OH
				2-Hydroxypropene
-CHO	Aldehyde		-al	<b>\</b> 0
0				
0   0				Propanal
R <sup>C</sup> H				
O II R R R	Ketone		-one	Ο
"				
R / C / B				
				Propanone
				1 Topullone
R-O-	Ether	(alkyl)-oxy-		
CH₃-O-		Methoxy		<b>-</b> 0
		,		Methoxyethane
				conoxycenane
		<del></del>	<del></del>	
-COOH	Carboxylic acid	Carboxy-	-oic acid	O
O				[]
"				OH
R <sup>C</sup> OH				Ethanoic acid
	<del>-  </del>			EUIAHOIC ACIU
-COO-R	Ester	(Alkyl)	-oate	l Ö
-COO-C <sub>2</sub> H <sub>5</sub>		Ethyl		
O				
				Ethyl propagata
R'O-R				Ethyl propanoate
O II R CI	Acid chloride		-oyl chloride	Ο
			<u> </u>	II
K U				CI
				Ethanoyl chloride
•	i.	1		

$O$ $C$ $NH_2$ $O$ $R$ $N$ $R$ $NR_2$	Acid amide		-amide	NH <sub>2</sub> Ethanamide
R O R	Acid anhydride		-oic anhydride	O O O O O O O O O O O O O O O O O O O
-CN	Nitrile		-Nitrile	N=/ Propanenitrile
-NH <sub>2</sub> -NHR -NR <sub>2</sub>	Amine	Amino-	-amine	NH <sub>2</sub> Propylamine NH <sub>2</sub> 2-aminopropane

## Make notes on naming chemicals with functional groups

Use the following questions as guidance

- 1. Briefly explain how alkanes can be named.
- 2. Briefly explain how to name a chemical with functional groups.