Naming Chemicals 2

Pre-lesson assignment – Textbook page 174-178

You will need to learn to name these functional groups

			0 (6)	T- 1
Structure	Name	Prefix	Suffix	Example
R	Alkene		-ene	
				Propene
Ŕ				
	Allama			1.1
R	Alkyne		-yne	L_C_C_H
R				H H
				Ethyne
	Arene	Phenyl-	-benzene	
' ()				(
				Phenylethane
				Methylbenzene
OH	Phenol		-phenol	4-methylphenol
ĭ''			1	OH
[]				
L、リ				[~]
				l
-X	Haloalkane			
-Cl	Chloroalkane	Chloro-		CI-CH ₃
-F	Fluoroalkane	Fluoro-		
				Chloromethane
-Br	Bromoalkane	Bromo-		
-I	Iodoalkane	lodo-		
-OH	Alcohol	Hydroxy-	-ol	
	7.1001101	11,010,01,	J	/ \OH
				Ethanol
				OH
				2-Hydroxypropene
-CHO	Aldehyde		-al	√ 0
0				·
0::				Propanal
R ^C \H				
O II R ^C R	Ketone		-one	Ο
"				
P\C\D				
				Propanone
				1 Topanone
R-O-	Ether	(alkyl)-oxy-		
CH₃-O-		Methoxy		- 0
				Methoxyothana
				Methoxyethane
-СООН	Carboxylic acid	Carboxy-	-oic acid	O
0				
0 ::				OH
R~~OH				Ethanoic acid
-COO-R	Ester	(Alkyl)	-oate	0
-COO-C ₂ H ₅		Ethyl		<u> </u>
O				• 0—
				\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \
R´``O-R				Ethyl propanoate
	Acid chloride		-oyl chloride	0
0 	Acid chloride		- Oyl Cilloride	
_ ,C,				
R CI				/ `CI
				Ethanoyl chloride
				Ethanoyi amoriae

O R O NH_2 O R NR_2	Acid amide		-amide	NH ₂ Ethanamide
R O R	Acid anhydride		-oic anhydride	O O O Ethanoic anhydride
-CN	Nitrile		-Nitrile	N=
-NH ₂ -NHR -NR ₂	Amine	Amino-	-amine	NH ₂ Propylamine NH ₂ 2-aminopropane

Make notes on naming chemicals with functional groups

Use the following questions as guidance

- 1. Briefly explain how alkanes can be named.
- 2. Briefly explain how to name a chemical with functional groups.