## Naming Chemicals 2

Pre-lesson assignment – Textbook page 174-178

## You will need to learn to name these functional groups

Structure	Name	Prefix	Suffix	Example
R	Alkene		-ene	Propene
R	Alkyne		-yne	H C C H
	Arene	Phenyl-	-benzene	Phenylethane  Methylbenzene
OH OH	Phenol		-phenol	4-methylphenol OH
-X -Cl -F -Br -I	Haloalkane Chloroalkane Fluoroalkane Bromoalkane Iodoalkane	Chloro- Fluoro- Bromo- Iodo-		CI-CH <sub>3</sub> Chloromethane
-ОН	Alcohol	Hydroxy-	-ol	OH Ethanol OH 2-Hydroxypropene
-CHO O II R C H	Aldehyde		-al	Propanal
O R C R	Ketone		-one	O Propanone

R-O- CH₃-O-	Ether	(alkyl)-oxy- Methoxy		—O Methoxyethane
-cooh O II C OH	Carboxylic acid	Carboxy-	-oic acid	O OH Ethanoic acid
-COO-R -COO-C <sub>2</sub> H <sub>5</sub> O	Ester	(Alkyl) Ethyl	-oate	O O Ethyl propanoate
O R C CI	Acid chloride		-oyl chloride	CI Ethanoyl chloride

O II R NH <sub>2</sub>	Acid amide		-amide	NH <sub>2</sub>
R N R				Ethanamide
R NR <sub>2</sub>				
$R \longrightarrow R$	Acid anhydride		-oic anhydride	Ethanoic anhydride
-CN	Nitrile		-Nitrile	N=
-NH <sub>2</sub> -NHR -NR <sub>2</sub>	Amine	Amino-	-amine	NH <sub>2</sub> Propylamine NH <sub>2</sub> 2-aminopropane

## Make notes on naming chemicals with functional groups

Use the following questions as guidance

- 1. Briefly explain how alkanes can be named.
- 2. Briefly explain how to name a chemical with functional groups.

Now complete the summary questions (green box) on each of the pages you studied (pgs. 173, 178, 181).