Ionisation energy and sub-shells

Pre-lesson assignment p.56 + 99-100

Make notes on ionisation energy and sub-shells

Use the following questions as guidance:

- 1. State the trend of ionisation energy across a group
- 2. Explain why this general trend is observed
- 3. State the rules for the filling of atomic orbitals (p56)
- 4. Sketch figure 4, and explain
 - a. the fall in first ionisation energy from Be to B
 - b. the fall in first ionisation energy from N to O