

Lecture 4 - de Broglie waves

Monday, August 28, 2023 9:04 AM

- Warm-up Quiz
- HW2 Due Friday
- office hours

M - 3-4

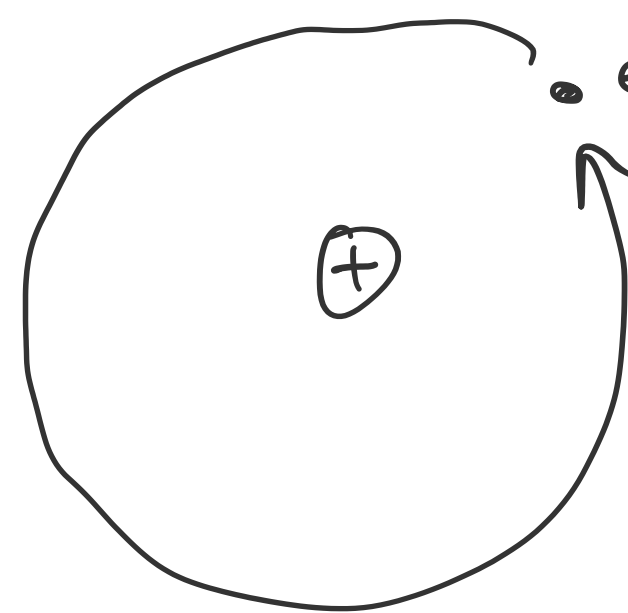
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R - 5-6

Today :
• de Broglie waves
• Schrödinger eqn

Last time : Rutherford \rightarrow planetary model



Bohr

• orbits \rightarrow energy level

$$E_n = -\frac{hcR_H}{n^2} \quad n = 1, 2, 3, \dots$$

$$r_n = a_0 n^2$$

$a_0 \equiv$ Bohr radius

Light emission from orbital jumps $\Delta E_{n \rightarrow m}$

But what makes e^- choose certain orbits, not allowing others?

de Broglie \rightarrow if Light has wave/particle properties
Maybe e^- 's are too.

particle \leftrightarrow wave
stationary state \leftrightarrow standing wave

For light:

$$E = h\nu = cp$$

$$\frac{h}{\lambda} = p \Rightarrow \boxed{\lambda = h/p} \quad \star$$

$$\vec{p} = \hbar \vec{k}$$

momentum

wave vector

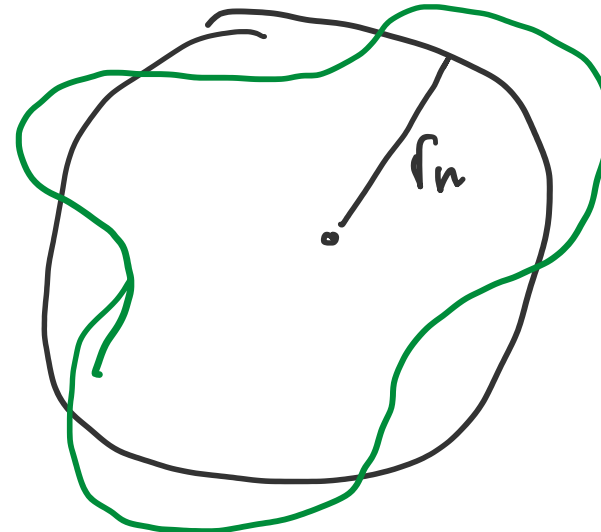
Bohr in terms of R_H

$$r_n = a_0 n^2$$

Circumference $2\pi r_n$

idea - fit integer number of wavelengths in circumference \rightarrow standing wave

$n=3$



$$\therefore 2\pi r_n = n\lambda = \frac{nh}{p} = \frac{nh}{mv}$$

Classical circular orbit

$$\frac{mv^2}{r_n} = \frac{Ze^2}{4\pi\epsilon_0 r_n^2} \Rightarrow v_n = \sqrt{\frac{Ze^2}{4\pi\epsilon_0 r_n m}}$$

centrifugal force Coulomb force

$$2\pi r_n = \frac{nh}{mv} = nh \sqrt{\frac{4\pi\epsilon_0 r_n}{Ze^2 m}}$$

$$\text{solve for } r_n \quad r_n^2 = \frac{n^2 h^2}{4\pi^2} \frac{4\pi\epsilon_0}{Ze^2 m}$$

$$r_n = \left(\frac{\hbar^2 \epsilon_0}{\pi Ze^2 m} \right) n^2 \quad \boxed{\text{as before}}$$

a_0 in terms of fundamental quantities

Notice angular momentum is quantized

$$L_n = m v_n r_n = \hbar n$$

\hookrightarrow assumption of wave boundary conditions

Plausibility argument for the Schrödinger equation

$$\vec{p} = \hbar \vec{k} \quad (\text{de Broglie})$$

$$E = \hbar \omega \quad (\text{Einstein})$$

$$E = \frac{p^2}{2m} + V \quad (\text{Newton}) \quad [\text{non-relativistic}]$$

We know we need a wave equation
Wave equation solutions $\Psi(x,t)$ have standing wave
sol \rightarrow and linear superpositions.

$$\Psi(x,t) = C_1 \Psi_1(x,t) + C_2 \Psi_2(x,t) + \dots$$

if Ψ_1, Ψ_2 are solutions of the equation.

\Rightarrow only linear terms in Ψ (no Ψ^2 , but e.g. $\frac{\partial \Psi}{\partial t}$ ok)

\Rightarrow Look for a wave equation for a massive particle w/ const E, p
 $\rightarrow \omega, k$ have to be constant

Look @ Classical wave equation

$c \equiv$ Velocity $\rightarrow 1$

$$\frac{\partial^2 y}{\partial x^2} = \frac{1}{c^2} \frac{\partial^2 y}{\partial t^2}$$

$$\text{recall } y = E_0 \cos(kx - \omega t)$$

$$\frac{\partial \Psi}{\partial t} = \omega \sin(kx - \omega t)$$

$$\frac{\partial^2 \Psi}{\partial t^2} = -\omega^2 \cos(kx - \omega t)$$

$$\frac{\partial \Psi}{\partial x} = -k \sin(kx - \omega t)$$

$$\frac{\partial^2 \Psi}{\partial x^2} = -k^2 \cos(kx - \omega t)$$

sub in

$$-k^2 \cos(kx - \omega t) = -\frac{\omega^2}{c^2} \cos(kx - \omega t)$$

$$\Rightarrow \omega^2 = c^2 k^2 \quad \text{or} \quad \omega = c|k|$$

Equiv. to

$$E = \hbar c|k|$$

Linear dispersion of Light

$$E \leftrightarrow p$$

problem: need $E \propto k^2 \rightarrow$ need to end up w/ $E = \frac{p^2}{2m} + V$

$$E = \frac{\hbar^2 k^2}{2m} + V$$

each $\frac{\partial}{\partial x} \rightarrow k$; $\frac{\partial}{\partial t} \rightarrow \omega$

$$E \Psi = \frac{p^2}{2m} \Psi + V \Psi$$

$$\omega \Psi = \frac{\hbar^2}{2m} k^2 \Psi + \frac{V}{\hbar} \Psi$$

single derivative w/ t

2 der. w/ x

$$\boxed{\text{postulate} \quad \alpha \frac{\partial \Psi}{\partial t} = \beta \frac{\partial^2 \Psi}{\partial x^2} + V \Psi}$$