

## 1 unit conversion

see <http://physics.nist.gov/cuu/Units/units.html> and [1] page T-5

| Description  | Name    | Symbol   | convert               | SI                                   |
|--|---------|----------|-----------------------|--------------------------------------|
| Force  | Newton  | N        |                       | $\frac{m \cdot kg}{s^2}$             |
| energy, work, quantity of heat                     | joule   | J        | $N \cdot m$           | $\frac{m^2 \cdot kg}{s^2}$           |
| power  | Watt    | W        | $\frac{J}{s}$         | $\frac{m^2 \cdot kg}{s^3}$           |
| pressure, stress                                   | pascal  | Pa       | $\frac{N}{m^2}$       | $\frac{kg}{m \cdot s^2}$             |
| electric charge, quantity of electricity           | coulomb | C        |                       | $s \cdot A$                          |
| electric potential difference, electromotive force | volt    | V        | W/A                   | $\frac{m^2 \cdot kg}{s^3 \cdot A}$   |
| capacitance  | farad   | F        | C/V                   | $\frac{s^4 \cdot A^2}{m^2 \cdot kg}$ |
| electric resistance                                | ohm     | $\Omega$ | V/A                   | $\frac{m^2 \cdot kg}{s^3 \cdot A^2}$ |
| magnetic field                                     | Tesla   | T        | $\frac{V \cdot s}{m}$ | $\frac{m \cdot kg}{s^2 \cdot A}$     |

$$1\text{Joule} = 6.24150974 \cdot 10^{18} \text{ electron volts}$$

$$k_{Boltzmann} T_{room} = \frac{1}{40} \text{ eV}$$

$$T_{room} = 293.15 K$$

## 2 constants

see [http://en.wikipedia.org/wiki/Physical\\_constant](http://en.wikipedia.org/wiki/Physical_constant) and <http://physics.nist.gov/cuu/Constants/>

Universal constants :

| Description                       | Symbol                   | SI value                                 |
|-----------------------------------|--------------------------|--|
| speed of light in vacuum          | $c$                      | $299792458 \frac{m}{s}$                  |
| Newtonian constant of gravitation | $G$                      | $6.67428(67) 10^{11} \frac{m^3}{kg s^2}$ |
| Planck's constant                 | $h$                      | $6.62606896(33) 10^{34} J s$             |
| reduced Planck constant           | $\hbar = \frac{h}{2\pi}$ | $1.054571628(53) 10^{34} J s$            |

| Description                                    | Symbol, definition   | SI value  |
|--|--|---|
| magnetic constant (vacuum permeability)        | $\mu_0$  | $4\pi 10^7 \frac{N}{A^2}$                         |
| electric constant (vacuum permittivity)        | $\epsilon_0 = \frac{1}{\mu_0 c^2}$                                     | $8.854187817 10^{12} \frac{F}{m}$                 |
| elementary charge                              | $e$  | $1.602176487(40) 10^{19} C$                       |
| Bohr radius                                    | $a_0 = \frac{\alpha}{4\pi R_\infty}$                                   | $0.5291772108(18) 10^{10} m$                      |
| classical electron radius                      | $r_e = \frac{e^2}{4\pi\epsilon_0 m_e c^2}$                             | $2.8179402894(58) 10^{15} m$                      |
| electron mass                                  | $m_e$  | $9.10938215(45) 10^{31} kg = 511 \frac{KeV}{c^2}$ |
| fine-structure constant                        | $\alpha = \frac{\mu_0 e^2 c}{2h} = \frac{e^2}{4\pi\epsilon_0 \hbar c}$ | $7.2973525376(50) 10^3$                           |
| proton mass                                    | $m_p$  | $1.672621637(83) 10^{27} kg$                      |
| Rydberg constant                               | $R_\infty = \frac{\alpha^2 m_e c}{2h}$                                 | $10973731.568525(73) \frac{1}{m}$                 |
| atomic mass unit<br>(unified atomic mass unit) | $m_u = \frac{1}{u}$  | $1.66053886(28) 10^{27} kg$                       |
| Avogadro's number                              | $N_{Av}$   | $6.0221415(10) 10^{23} \frac{1}{mol}$             |
| Boltzmann constant                             | $k_B = \frac{R}{N_{Av}}$   | $1.3806505(24) 10^{23} \frac{J}{K}$               |
| gas constant                                   | $R$  | $8.314472(15) \frac{J}{K mol}$                    |
| standard atmosphere                            | $atm$  | $101325 Pa = 14.7 Psi$                            |

Miscellaneous:

$$1 eV = 1.602 10^{-19} J \quad (1)$$

standard temperature is 293.15 K

$$\frac{m_{proton}}{m_e} = 1836 \quad (2)$$

### 3 Symbol notations

See page 631-642 of [5]

$C_v \equiv$  specific heat for constant volume

$k_b \equiv$  Boltzmann constant

$\vec{E} \equiv$  electric field

$\vec{B} \equiv$  magnetic field

$\vec{D} \equiv$  electric displacement, equ 4.21, page 175 [2]

$\vec{N} \equiv$  torque

$W \equiv$  work

$\vec{v} \equiv$  velocity

$V \equiv$  electrostatic potential [Volts]

$V \equiv$  volume

$U \equiv$  potential energy [Volts]

$\vec{P} \equiv$  polarization, page 166 [2]

$I \equiv$  current [Amps]

$\vec{F} \equiv$  force [Newtons,  $\frac{kg \cdot m}{s^2}$ ]

$E \equiv$  energy

$T \equiv$  kinetic energy

$T \equiv$  temperature

$L \equiv$  lagrangian

$L \equiv$  capacitance

$\vec{L} \equiv$  classical (orbital) angular momentum

$l \equiv$  quantum (orbital) angular momentum

$\vec{p} \equiv$  linear momentum

$\omega \equiv$  angular frequency

$z \equiv$  single particle partition function

$Z \equiv$  total partition function

### References

[1] Sears, Zemansky, Young *University Physics, Fifth edition*, (1976)

[2] Griffith *Intro to Electrodynamics*, (1999)

[3] Jackson *Classical Electrodynamics*, (1999)

[4] Goldstein *Classical Mechanics*, (1980) (second edition)  
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[5] Reif *Fundamentals of statistical and thermal physics*, (1965)

[6] Parris's book on *Quantum mechanics*, (2008)  
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[7] Liboff *Introduction to Quantum mechanics*, (2003) Fourth edition

[8] Marion *Classical Dynamics of Particles and systems*, (1970) Second edition.  
Available from Prof Waddill, Elizabeth

[9] Tipler, Llewellyn *Modern Physics*, (1999)