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CH301 Worksheet 8—Gases (Answer Key) 1. What do we assume about ideal gases? What is the ideal gas law? Give the units for each variable. Ideal gases are infinitely small, hard spheres that do not interact with each other. They are essentially "blind" to other gas molecules and will bounce off of each other just as they would bounce of a wall.

CH301 Worksheet 8—Gases (Answer Key) 0.08206 (l*atm)/(K*mol).

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In your answer, consider the external and internal pressure data given in Model 1. Read This! Pressure is caused by molecules hitting the sides of a container or other objects. The pressure changes when the molecules change how often or how hard they hit. A nonflexible container is needed if the gas ... Gas Variables 7 Extension Questions

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The ideal gas law is an equation that relates the volume, temperature, pressure and amount of gas particles to a constant. The ideal gas constant is abbreviated with the variable R and has the value of $0.0821 \text{ atm}\cdot\text{L/mol}\cdot\text{K}$. The ideal gas law can be used when three of the four gas variables are known.

Ideal Gas Law Name Chem Worksheet 14-4

description for each experiment and answer the questions. 1 - Patty Power Mr. Krabbs wants to make Bikini Bottoms a nicer place to live. He has created a new sauce that he thinks will reduce the production of body gas associated with eating crabby patties from the Krusty Krab. He recruits 100 customers with a history of gas problems.

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