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KINETICS Practice Problems and Solutions d. Write the rate law for the overall reaction.  $\text{rate} = k [\text{A}]^2 [\text{B}]^2$  9. Consider the following mechanism.  $\text{O}_3 \rightarrow \text{O}_2 + \text{O}$  (fast)  $\text{O}_3 + \text{O} \rightarrow 2 \text{O}_2$  (slow) a. Write the overall balanced chemical equation.  $2 \text{O}_3 \rightarrow 3 \text{O}_2$  b. Identify any intermediates within the mechanism. O c. What is the order with respect to each reactant? O

### **KINETICS Practice Problems and Solutions**

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### **Elements of Chemical Reaction Engineering**

2 Asymptotic Solution to diffusion-reaction problems For a single non-linear reaction problem, a solution can be obtained analytically for large values of Thiele modulus. This asymptotic solution can then be used to define a generalized Thiele modulus and can be extended to obtain approximate analytical solutions even for low Thiele modulus. This

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