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In thermodynamics, work performed by a system is energy transferred by the system to its surroundings, due solely to macroscopic forces exerted by the system on its surroundings, where those forces, and their external effects, can be measured. Such work is the only kind by which a thermodynamic system can be made to lift a weight. The externally measured forces and external effects may be ...

Work (thermodynamics) - Wikipedia

of Fundamentals ُ(2012 سال 2012) و سانتاگ ویرایش هشتم (سال 2012) Thermodynamics 8th Edition Borgnakke & Sonntag

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In thermodynamics, an isentropic process is an idealized thermodynamic process that is both adiabatic and reversible. The work transfers of the system are frictionless, and there is no transfer of heat or matter. Such an idealized process is useful in engineering as a model of and basis of comparison for real processes. The word "isentropic" is occasionally, though not customarily, interpreted ...

Isentropic process - Wikipedia

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Coefficiente di prestazione - Wikipedia

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A Simple Model of Global Average Surface Temperature April 30th, 2013 by Roy W. Spencer, Ph. D.

A Simple Model of Global Average Surface Temperature

Calor específico é uma grandeza física intensiva que define a variação térmica de determinada substância ao receber determinada quantidade de calor. Também é chamado de capacidade térmica mássica. A unidade no SI é / (.) (joule por quilograma e por kelvin). Uma unidade usual bastante utilizada para calores específicos é cal/(g.°C) (caloria por grama e por grau Celsius).

Calor específico - Wikipédia, a enciclopédia livre

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Dampmaskin - Wikipedia

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