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Balancing Chemical Equations Answer Key Vocabulary: coefficient, combination, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript Prior Knowledge Questions (Do these BEFORE using the Gizmo.) [Note: The purpose of these questions is to activate prior knowledge and get students thinking.

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Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass. Launch Gizmo

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Gizmo Warm-up In a chemical reaction, reactants interact to form products. This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo $^{\text{m}}$, look at the floating molecules below the initial reaction: H 2 + O 2 H 2O. 1. How many atoms are in a hydrogen molecule (H 2)?

2. How many atoms are in an oxygen molecule ...

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Activity A: Interpreting chemical formulas Get the Gizmo ready: Erase the chemical formulas in each text box. Check that the Visual displays are selected. Introduction: To balance a chemical equation, you first need to be able to count how many atoms of each element are on each side of

the equation. In this activity, you will practice counting the atoms that are represented in chemical formulas.

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