# Finding Solutions Kool Aid Dilution Lab

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# **Finding Solutions Kool Aid Dilution**

Finding Solutions Kool Aid Dilution The first one I want to show you is the cool aid spray! Yep, cool aid is notorious for not coming out once spilled, so that's just what we need.

## Finding Solutions Kool Aid Dilution Lab - 3babak.com

Lab Activity- Kool-Aid Concentration. Introduction: This activity introduces you to solutions and allows you to experience making different concentrations of Kool- aid solution. There are many ways to calculate the concentration of a substance including: molarity (M), parts per million (ppm), percent composition (% comp), and grams per liter (g/L).

#### **ANSWER KEY Lab Activity- Kool-Aid Concentration**

Weigh out the calculated mass of Kool-Aid to make your 4 M solution. Place it in the plastic cup, and add enough water to dissolve the solute. Use your stir rod to mix the solution until all of the Kool-Aid is dissolved. Now, add water until the solution volume reaches the 100 mL mark.

#### Concentration of Kool-Aid® - ScienceGeek.net

The Famous Kool-Aid Lab 48 t-spoons = 1 cup, 1 t-spoon = 5 milliliters, 1.0 liter = 1.057 quart, ... If the kool-aid was originally 0.33 M, find the new molarity.

#### **Finding Solutions Kool Aid Dilution Lab**

Make three solutions of Kool-Aid with different concentrations and taste them to decide which is the correct concentration. You will prepare 0.1 L of each of the following Kool-Aid solutions: 0.1 M, 0.4 M, and 0.7 M. ... Lab 10: What's the Concentration of Kool-Aid ...

#### Lab 10: What's the Concentration of Kool-Aid

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#### Name: Per: Lab Activity- Kool-Aid Concentration

In this chemistry experiment, students will investigate dilution with Kool-Aid. The students will use their five senses to explore the solution.

#### **Kool Aid Chemistry | Curriki**

One mole of sodium chloride in one liter of water is 1 molar NaCl. This is the molarity of the solution and abbreviated as M. A dilution is taking a specific amount of a solution and adding it to a larger volume. As an example, think of making a cup of Kool-Aid. Now take  $\frac{1}{2}$  of that cup and put it in a new cup.

#### **Molarity and Serial Dilutions Teacher Handout**

Procedure. Stir the solution with the popsicle stick until all of the Kool-Aid powder has dissolved. Place the large plastic cup, containing the Kool-Aid solution on the scale and record the total mass in the data table below. Record your observations, including appearance, smell and taste of the solution in your data table.

#### Classroom Resources | The Perfect Kool-Aid Concentration ...

Lab Activity- Kool-Aid Concentration Introduction: This activity introduces you to solutions and allows you to experience making different concentrations of Kool- aid solution. In chemistry, concentration is usually measured by the number of moles of substance dissolved in a liter of liquid.

#### Name Date Lab Activity- Kool-Aid Concentration

Made with Explain Everything. STOP! YOU ARE MAKING A BIG MISTAKE DRINKING LEMON WATER LIKE THAT, HOW TO MAKE LEMON WATER K.Beauty - Duration: 4:20. Khichi Beauty 7,233,599 views

#### koolaid dilution lab

MATERIALS AND EQUIPMENT: Balance Kool-aid  $\square$  (Solute) in 300 mL vial 120 mL graduated vial concentrated juice solution paper cups Cold water (solvent) SAFETY: This activity is designed to teach concentration. When mixing chemicals in the lab, the student would not normally be asked to taste a solution.

# koolaid lab short form - Molarity of a Solution A cool aid ...

MOLARITY BY DILUTION – Making a weak solution from a stronger one. M1V1 = M2V2. PROBLEMS: 1. a. ... A student is making popsicles from Kool-Aid. He finds that the popsicles made with a strong Kool-Aid solution take longer to freeze than the popsicles made with a weak solution. Use colligative properties to explain why.

# **SOLUTIONS - Livingston Public Schools**

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~Data Table~ Show all the work for your calculations below: ~KOOL-AID CONCENTRATION ACTIVITY~ ~PROCEDURE ~ Go to a lab table with your group. At the lab table you will find a pitcher, a packet of Kool-Aid, and an index card which will tell you the amount of sugar you need to put in YOUR Kool-Aid solution.

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