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0 3L

A sample of neon gas occupies a volume of 2.8 L at 1.8 atm. What would its volume be at 1.2 atm? A balloon full of air has a volume of 2.75 L at a temperature of 18oC.

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Unit 6 - Gas Laws

Gas Laws Worksheet atm = 760.0 mm Hg = 101.3 kPa = 760.0 torr Boyle's Law Problems: 1. If 22.5 L of nitrogen at 748 mm Hg are compressed to 725 mm Hg at constant temperature. What is the new volume? 2. A gas with a volume of <math>4.0L at a pressure of 205kPa is allowed to expand to a volume of 12.0L.

Gas Laws Worksheet

Gas Laws STUDY GUIDE Due: February 12th. Units of Measurement: For the following questions, use the following answer choices to indicate what each unit of measurement is. used to measure. A. Pressure B. Volume. C 1. K \ddot{y} 4. kPa. A 2. atm \ddot{y} 5.

Gas Laws STUDY GUIDE Due: February 12th

Honors Chemistry: Unit 7- Physical State Theory and Gas Laws - PRACTICE TEST KEY Students are REQUIRED to do this practice test. Answers must be complete and thorough. Answers to the questions on this test AND all the work to support those answers should be written neatly and clearly on a separate sheet of paper (or multiple sheets of paper, if

Honors Chemistry: Unit 7 Physical State Theory and Gas Laws

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