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Chemistry - Wikipedia

An acid is a molecule or ion capable of donating a proton (hydrogen ion H +), or, alternatively, capable of forming a covalent bond with an electron pair (a Lewis acid).. The first category of acids is the proton donors or Brønsted acids. In the special case of aqueous solutions, proton donors form the hydronium ion H 3 O + and are known as Arrhenius acids. ...

Acid - Wikipedia

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Insights into biochar and hydrochar production and ...

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