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With the Flame Tests Chemistry Laboratory Kit, students observe the characteristic color of light emitted by metallic salts when placed in a flame. This basic lab activity is ideal for physical science and introductory chemistry classes.

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Flame Test Kit Student Laboratory Kit Introduction Just as a fingerprint is unique to each person, the color of light emitted by metals heated in a flame is unique to each metal. In this laboratory activity, the characteristic color of light emitted for calcium, copper, lithium, potassium, sodium, and strontium will be observed. Chemical Concepts

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Flame Tests of Metals - De La Salle High School

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Such analysis is known as a flame test. To do a flame test on a metallic element, the metal is first dissolved in a solution and the solution is then held in the hot, blue flame of a Bunsen burner. This test works well for metal ions, and was perfected by Robert Bunsen (1811 - 1899).

Flame Tests Lab - Barberton High School

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