

## *Flame Test Lab Answer Key*

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**Flame Test Lab Answer Key**

The responding variable was the color of the flame. The flame was orange at the end of each test because the wood splint started burning. The reason each of the elements produced a different color (or series of wavelengths) is because of the make up of the atom. ... Flame Test Lab Questions Answer Key: ...

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**Flame Test Pre Lab Answer Key - examget.net**

10. Hold the splint in the flame and record the color of the flame that is produced. 11. Using your data, identify the metal ion in your unknown solution. Flame Test Lab Activity Key Note: If chloride compounds are not available, metal nitrate compounds may be substituted. Use dilute or approximately 0.1 M solutions Unknowns: Number the beakers ...

**Flame Test Lab Activity Key - University of South Florida**

Chemistry flame test lab answer key. eXam Aswers Search Engine . Submit. More "Chemistry Flame Test Lab Answer Key" links. Emission Spectroscopy: Element identification. An itroduction to emission spectra and flame tests for the identification of elements.

**Chemistry Flame Test Lab Answer Key - fullexams.com**

Flame Test Lab Questions Answer Key: 1. You could readily identify the elements that had obvious colors different form all the others- such as copper that gave off a blue/green color and lithium that gave off a bright red color. 2. It was difficult to identify a couple of the elements that had colors that were similar. 3.

**Flame Test 2.pdf - Flame Test Lab Questions Answer Key 1 ...**

Flame Test Kit Student Laboratory Kit Introduction Just as a fingerprint is unique to each person, the color of light emitted by metals heated in a flame is unique to each metal. In this laboratory activity, the characteristic color of light emitted for calcium, copper, lithium, potassium, sodium, and strontium ... CF#5607 Flame Test Kit SLK ...

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Pre Lab Questions (answer on separate paper) 1. Describe the common characteristics of flame tests that would be observed if a spectroscope were used. 2. List . at least five. other means of qualitative analysis that are used to identify metals. (Use the index in your text or other reference sources.)

**LAB FIVE - site.lssc.edu**

Burning metals reveal themselves from unexpected, awesome colors in a simple flame test. Pour a little powder to test into each small dish. Dip one end of a popsicle stick into the water to moisten it. It helps to dim the lights in your lab so the colors are easier to see. Light the flame and place ...

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