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Chem 3322 -- Physical Chemistry II

In chemistry, colligative properties are properties of solutions that depend on the ratio of the number of solute particles to the number of solvent molecules in a solution, and not on the nature of the chemical species present. The number ratio can be related to the various units for concentration of solutions. The assumption that solution properties are independent of nature of solute ...

Colligative properties - Wikipedia

In chemistry, an ideal solution or ideal mixture is a solution with thermodynamic properties analogous to those of a mixture of ideal gases. The enthalpy of mixing is zero as is the volume change on mixing by definition; the closer to zero the enthalpy of mixing is, the more "ideal" the behaviour of the solution becomes. The vapor pressure of the solution obeys Raoult's law, and the activity ...

Ideal solution - Wikipedia

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Colloidal cesium lead halide perovskite nanocrystals exhibit unique photophysical properties including high quantum yields, tunable emission colors, and narrow photoluminescence spectra that have marked them as promising light emitters for applications in diverse photonic devices.

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Fibromyalgia is currently understood to be a disorder of central pain processing or a syndrome of central sensitivity. Clauw describes the syndrome as a diffuse problem of sensory "volume control"

such that patients have a lower threshold of pain and of other stimuli, such as heat, noise, and strong odors.

Fibromyalgia: Practice Essentials, Background, Pathophysiology

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The methods of molecular biology includes: 1)Hemacytometer for cell count, 2)Restriction enzyme digest(a process of cutting DNA molecules into smaller pieces with special enzymes called restriction endonucleases), 3) DNA ligation (using DNA ligase enzyme that helps in joining the DNA strands together by catalysing the formation of a phosphodiester bond), 4)Transfection(a process of ...

Methods in Molecular Biology | List of High Impact ...

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