

## ***Gizmo Chemical Equations Answers***

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This is the fourth lesson in the chemical reaction unit that covers HS-PS1-7, the uses of mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction; and DCI-PS1.B, the fact that atoms are conserved, together with knowledge of the chemical properties of the elements involved, can be used to describe and predict chemical reactions.

### **Tenth grade Lesson Gizmo: Modeling Chemical Reactions**

To set up an equation in the Chemical Equations Gizmo™, type the chemical formulas into the text boxes of the Gizmo. First, type in "H<sub>2</sub>+O<sub>2</sub>" in the Reactants box and "H<sub>2</sub>O" in the Products box. This represents the reaction of hydrogen and oxygen gas to form water.

### **Student Exploration: Chemical Equations**

Question: How are chemical equations balanced? 1. Balance: Turn on Show histograms. The equation is balanced when there are equal numbers of each type of atom represented on each side of the equation. In the Gizmo, use the up and down arrows to adjust the numbers of hydrogen, oxygen, and water molecules until the equation is balanced.

### **Student Exploration: Balancing Chemical Equations**

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### **Balancing Chemical Equations Worksheet Answer Key Gizmo**

Balancing Chemical Equations Gizmo. STUDY. PLAY. Coefficient. a number that multiplies a term in an equation. Combination. a chemical reaction in which two or more reactants form a single product. Compound. a pure substance composed of two or more elements chemically combined. Decomposition.

### **Balancing Chemical Equations Gizmo Flashcards | Quizlet**

Balancing Chemical Equations Answer Key Vocabulary: coefficient, combination, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript Prior Knowledge Questions (Do these BEFORE using the Gizmo.) [Note: The purpose of these questions is to activate prior knowledge and get students thinking.]

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