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GAS VARIABLES POGIL Proportionality Directly Proportional As \square one increases, so does the other $= \square * \square$ Inversely Proportional As \square one increases, the other decreases $= \square / \square$ Thermodynamic Observables Aspects of a system that we can observe, measure, or calculate in some way. Pressure - P kilopascals (kPa) Temperature - T Kelvin (K) Volume - V Liters (L) moles - n ...

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Gas Variables POGIL

Chem 116 POGIL Worksheet - Week 1 Gas Laws - Part 1 Why? Of the three principal states of matter (gas, liquid, solid), gases show behavior that is most easily connected to molecular motion. The observed behavior of gases, embodied in the empirical gas laws, leads to a series of equations that can be summarized by a single equation of state, called

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