

Formulas And Nomenclature Answers

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Formulas And Nomenclature Answers

Example #1 - write the formula for copper(II) chlorate. Step #1 - the first word tells you the symbol of the cation. In this case it is Cu. Step #2 - the Roman numeral WILL tell you the charge on the cation. In this case it is a positive two.

ChemTeam: Nomenclature-Polyatomic

Example #6 - write the name for $\text{Fe}(\text{OH})_3$. Iron is an element with two possible oxidation states. The iron is a +3 charge because (1) there are three hydroxides, (2) hydroxide is a minus one charge, (3) this gives a total charge of negative three and (4) there is only one iron, so it must be a +3.

ChemTeam: Nomenclature-Polyatomic

Recommended software downloads: Below are links to general freeware programs that I highly recommend for learning chemistry.

Honors Chemistry - Darrell Feebeck

Chemical Foundations measurement, significant figures, precision & accuracy, conversion factors, matter: Atoms, Molecules and Ions atomic theory, intro to Periodic Table, formulas & names of compounds

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Acids and Bases - VDOE

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-OH is a functional group. It's not at all clear what you're trying to ask. The functional group is called the hydroxyl group. Its compounds are called "alcohols".

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Chemistry Homepage - ScienceGeek.net Homepage

Remarkably, the first 6 terms (!) of the power series for this formula (with respect to e^2) coincide with the corresponding terms in the exact expansion given above. It's only with the coefficient of e^{12} that things start to differ slightly: The correct coefficient of e^{12} is $-4851/2^{20}$ whereas Ramanujan's formula gives $-9703/2^{21}$, for a discrepancy approximately equal to $-e^{12}/2^{21}$.

Circumference/Perimeter of an Ellipse: Formula(s ...

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