

Equilibrium And Pressure Gizmo Answer Key

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Equilibrium And Pressure Gizmo Answer

Observe how reactants and products interact in reversible reactions. The amounts of each substance can be manipulated, as well as the pressure on the chamber. This lesson focuses on partial pressures, Dalton's law, and Le Chatelier's principle.

Equilibrium and Pressure Gizmo : ExploreLearning

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Equilibrium and Concentration Answer Key Vocabulary: chemical equilibrium, concentration, equilibrium, equilibrium constant, reaction quotient, reversible reaction Prior Knowledge Questions (Do these BEFORE using the Gizmo.) [Note: The purpose of these questions is to activate prior knowledge and get students thinking.

Equilibrium and Concentration Answer Key

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EQUILIBRIUM AND CONCENTRATION GIZMO ANSWER KEY PDF

collisions creates pressure on the walls of the container. In theory, any amount of gas can be squeezed into a container if the container is strong enough to withstand the gas pressure. The Equilibrium and Pressure Gizmo shows a mixture of gases in a chamber. The lid of the chamber can move up or down. 1. Check that Reaction 1 is selected.

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Partial pressure and equilibrium Get the Gizmo ready: Select Reaction 2. Move the Sim. speed . slider all the way to the right. Introduction: In the Equilibrium and Concentration Gizmo, you found that reversible reactions eventually result in chemical equilibrium.

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At a particular temperature, $K_p = 0.240$ for the reaction $\text{N}_2\text{O}_4(\text{g}) \rightleftharpoons 2\text{NO}_2(\text{g})$ A flask containing only $\text{N}_2\text{O}_4(\text{g})$ at an initial pressure of 4.50 atm is allowed to reach equilibrium. Calculate the total pressure in this flask at equilibrium. With no change in the amount of material in the flask, the volume of the container in question is decreased to 0.400 times the original volume.

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