

Gas Variables Answer Key

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In your answer, consider the external and internal pressure data given in Model 1. Read This! Pressure is caused by molecules hitting the sides of a container or other objects. The pressure changes when the molecules change how often or how hard they hit. A nonflexible container is needed if the gas ... Gas Variables 7 Extension Questions

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CHEMISTRY POGIL ANSWER KEY GAS VARIABLES PDF

CH301 Worksheet 8—Gases (Answer Key) 1. What do we assume about ideal gases? What is the ideal gas law? Give the units for each variable. Ideal gases are infinitely small, hard spheres that do not interact with each other. They are essentially "blind" to other gas molecules and will bounce off of each other just as they would bounce off a wall.

CH301 Worksheet 8—Gases (Answer Key) 0.08206 (L*atm)/(K*mol).

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The ideal gas law is an equation that relates the volume, temperature, pressure and amount of gas particles to a constant. The ideal gas constant is abbreviated with the variable R and has the value of 0.0821 atm·L/mol·K. The ideal gas law can be used when three of the four gas variables are known.

Ideal Gas Law Name Chem Worksheet 14-4

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Chem 116 POGIL Worksheet - Week 1 Gas Laws - Part 1 Why? Of the three principal states of matter (gas, liquid, solid), gases show behavior that is most easily ... Connecting these variables together is a ... and pressure relationships for a fixed amount of gas. Key Questions 11. Carry out the following steps to derive equations for the Combined ...

Chem 116 POGIL Worksheet - Week 1 Gas Laws - Part 1

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Gas Variables Answer Key

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