

Elements And Macromolecules Answer Key

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Uranium is a chemical element with the symbol U and atomic number 92. It is a silvery-grey metal in the actinide series of the periodic table. A uranium atom has 92 protons and 92 electrons, of which 6 are valence electrons. Uranium is weakly radioactive because all isotopes of uranium are unstable, with half-lives varying between 159,200 years and 4.5 billion years.

Uranium - Wikipedia

Topic 1 Key concepts in chemistry (Edexcel 9-1 GCSE Chemistry 1, Paper 1, Topic 1 "Key concepts in chemistry"). A page of important ideas, concepts and definitions for chemistry students (Topic 1 – Overarching concepts in chemistry)

Edexcel GCSE 9-1 Chemistry 1 Paper 1 separate science past ...

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Overview. A chemical formula identifies each constituent element by its chemical symbol and indicates the proportionate number of atoms of each element. In empirical formulas, these proportions begin with a key element and then assign numbers of atoms of the other elements in the compound, by ratios to the key element.

Chemical formula - Wikipedia

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Masses of proteins are often expressed in daltons. For example, a protein with a molecular weight of 64,000 g mol⁻¹ has a mass of 64,000 daltons or 64 kDa in short.

Is g/mol or dalton the same? - ResearchGate

A precipitation reaction occurs. Yellow precipitates of lead iodide are formed. The equation for this reaction is $\text{Pb}(\text{NO}_3)_2 + 2\text{KI} \rightarrow \text{PbI}_2 + 2\text{KNO}_3$

Precipitates can be formed in what kind of reaction?

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