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Electron Energy and Light 5 20. Label the remaining drawings in Model 3 with the electron transitions that are occurring ($n=?$ to $n=?$), the wavelengths and corresponding colors as given in example A in Model 3. See Model 2 in order to identify the color of spectral lines produced in each of the hydrogen atom electron transitions shown in Model 3.

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Electron Energy And Light. STUDY. PLAY. violet. high energy, high frequency, short wavelengths. red. low energy, low frequency, long wave lengths. ... for an electron to move from an energy level far from the nucleus to an energy level close to the nucleus it would need to ____ energy. lose. electron moves up.

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Electron Energy and Light 1 Electron Energy and Light How does light reveal the behavior of electrons in an atom? Why? From fireworks to stars, the color of light is useful in finding out what's in matter. The emission of light by hydrogen and other atoms has played a key role in understanding the electronic structure of atoms.

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POGIL | POGIL Activities for High School Chemistry

This video is for the Westlake Chemistry-1 classes. It is meant as a supplementary resource for the "Electron Energy and Light" POGIL we worked on in class. Please attempt the POGIL on your own ...

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