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Water molecules are polar and are capable of forming hydrogen bonds with other polar or charged molecules. As a result, water has the following properties: 20 molecules are cohesive; they form hydrogen bonds with each other. 20 molecules are adhesive; they form hydrogen bonds with polar surfaces.

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Write your answers on a separate piece of paper. Properties of water. 1. Write the chemical equation for water. 2. Draw a model of a water molecule. Show polarity by labeling positive and negatively charged regions of the molecule. 3. How is a water molecule like a magnet? ... Properties of water worksheet ...

Properties of water worksheet - Central Bucks School District

Those relatively strong bonds are the reason we see water's unusual properties: high surface tension, high boiling temperature, adhesion, cohesion, low vapor pressure, high specific heat, "universal solvent," the density of the solid form being less than that of the liquid form, and being a liquid at room temperature.

Lesson 2: The Science of Water Teacher Materials

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Properties of water worksheet. Name ____ 1. Write the chemical equation for water. 2. Draw a Stick model of a water molecule. 3. Sketch a space filling model of a water molecule, show polarity by label positive and negatively charged regions of the molecule. 4. How is a water molecule like a magnet? 5. What causes surface tension in water? 6.

Properties of water worksheet - somerset.k12.ky.us

Properties of Water 2.2 Lesson Objectives Discuss the unique properties of water. Class Date Differentiate between solutions and suspensions Explain what acidic solutions and basic solutions are. Lesson Summary Water molecules (H₂O) are polar because of an uneven The Water Molecule

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