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polyatomic ion. Your answer should include a discussion of subatomic particles. Somewhere in the ion there are either extra (in the case of an ion) or fewer electrons (in the case of a cation) compared to the total number of protons in the atoms involved. 7. What are the similarities and differences between the nitrate and nitrite ions in Model 1?

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The Ion Names, Formulas and Charges Chart for chemistry classrooms comprehensively lists the names and formulas of cations and anions, allowing students to quickly determine formulas for various combinations of cations and anions.

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