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6 The answer is B. The important step that determines the rate of reaction is that between the halogen radical and the alkane. e Option A is not correct because the reaction is between chlorine radicals and the alkane not chlorine molecules. Options C and D are incorrect — the Cl-Cl bond is the stronger. 7 The answer is B. The initial attack is by a  $\delta^+$  + Br atom.

### **6 The answer is B The important step that determines the ...**

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