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CHAPTER 21 REVIEW Nuclear Chemistry. Modern Chemistry 171 Nuclearchemistry CHAPTER 21 REVIEW Nuclear Chemistry SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. \_\_\_\_ The nuclear equation is an example of an equation that represents (a) alpha emission. (b) beta emission. (c) positron emission. (d) electron capture. 2.

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Modern Chemistry 175 Nuclearchemistry CHAPTER 21 REVIEW Nuclear Chemistry SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. Match each of the following statements with the process(es) to which they apply, using one of the choices below: (1) fission only (3) both fission and fusion

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Chapter 18 - Nuclear Chemistry 289 Key Ideas Answers 14. Because protons and neutrons reside in the nucleus of atoms, they are called nucleons. 16. There are two forces among the particles within the nucleus. The first, called the electrostatic force, is the force between electrically charged particles. The second force,

### **Chapter 18 Nuclear Chemistry**

Nuclear Chemistry Kinetics of Radioactive Decay A wooden object from an archeological site is subjected to radiocarbon dating. The activity of the sample that is due to  $^{14}\text{C}$  is measured to be 11.6 disintegrations per second.

### **Chapter 21 Nuclear Chemistry - University of Massachusetts ...**

Chapter 25 Nuclear Chemistry267 SECTION 25.1 NUCLEAR RADIATION (pages 799-802) This section describes the nature of radioactivity and the process of radio-active decay. It characterizes alpha, beta, and gamma radiation in terms ... Look at Table 25.3 on page 805 to help you answer the

following questions. a.

**SECTION 25.1 NUCLEAR RADIATION (pages 799-802)**

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**Piersa, Amanda / Unit 5: Nuclear Chemistry**

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