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Source: 3-4 Practice Problems, Chemistry Connections supplement (Prentice-Hall) Alpha & Beta Decay 1. Write a nuclear equation for the alpha decay of ^{231}Pa . $^{91}\text{Pa} \rightarrow ^{89}\text{Ac} + ^4_2\text{He}$ 2. Write a nuclear equation for the beta decay of ^{223}Fr . $^{87}\text{Fr} \rightarrow ^{88}\text{Ra} + ^0_{-1}\text{e}$ 3. Write a nuclear equation for the alpha decay of ^{149}Sm . $^{62}\text{Sm} \rightarrow ^{145}\text{Dy} + ^4_2\text{He}$

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Radioactivity •Radioactivity is the process by which nuclei emit particles and rays as they break down. •The name of the penetrating rays emitted by a radioactive source is called radiation. •A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or

Chapter 25 - Nuclear Chemistry

Nuclear Chemistry Kinetics of Radioactive Decay A wooden object from an archeological site is subjected to radiocarbon dating. The activity of the sample that is due to ^{14}C is measured to be 11.6 disintegrations per second.

Chapter 21 Nuclear Chemistry - University of Massachusetts ...

Chapter 25.1 and 25.2 in Prentice Hall Chemistry Text Book. STUDY. PLAY. Unstable Nucleus. Loses energy by emitting radiation during the process of radioactive decay. ... a process used in nuclear reactors to slow down neutrons so the reactor fuel captures them to continue the chain reaction.

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