

## *Nuclear Chemistry And Answer Key*

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### **Piersa, Amanda / Unit 5: Nuclear Chemistry**

Nuclear decay with no mass and no charge An electron Least penetrating nuclear decay Most damaging nuclear decay to the human body Nuclear decay that can be stopped by skin or paper. 3. Owl phcx bescx 10. Nuclear decay that can be stopped by aluminum. Complete the following nuclear equations. 12. 14.  $^{235}_{92}\text{Pu} + ^4_2\text{He} \rightarrow ^{239}_{94}\text{Cm} + ^0_{-1}\text{e}$  11. 13. 15 ...

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Nuclear Chemistry Nuclear Transformations • Rutherford in 1919 performed the first nuclear transformation. • The transmutations are sometimes represented by listing in order, the target nucleus, the bombarding particle, the ejecting particle and the product nucleus. • The above equation becomes:  ${}^{14}_2\text{N} + {}^4_2\text{He} \rightarrow {}^{17}_8\text{O} + {}^1_1\text{H}$

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GCC CHM152LL Nuclear Chemistry Summer Practice Worksheet p.1 of 4 CHM152LL: Nuclear Chemistry Summer Worksheet This worksheet is a summary of Nuclear Chemistry concepts and questions – you will not turn it in for a grade. An answer key will be available in PS149 – please check your answers before the final exam. I. Radioactive Isotopes and Nuclear Equations

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Nuclear Fission •With a high number of neutrons, radioisotopes split into smaller fragments, called nuclear fission. •Uranium-235 and Plutonium-239 are the only fissionable atoms. •Nuclear fission releases an enormous amount of energy. E.G.: 1 kg of U-235 yields a release of energy equal to 20,000 lbs of dynamite. (Atom bomb)

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806 Chapter 25 Nuclear Chemistry Figure 25-2 Both Pierre and Marie Curie played important roles in founding the field of nuclear chemistry. Marie Curie went on to show that unlike chemical reactions, radioactivity is not affected by changes in physical conditions such as temperature and pressure. She is the only person in history to receive

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