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6. Circle the letter that identifies what c represents in Einstein's mass-energy equation, $E = mc^2$. a. the charge on a proton b. the speed of light c. the charge on an electron 7. Is the following sentence true or false?

Chapter 10 Nuclear Chemistry Section 10.4 Fission and Fusion

Section 24.6; Core Teaching Resources, Section 24.6; Interpreting Graphics, Chapter 24 TECH: Interactive Textbook with ChemASAP, Chapter 24; Assessment 24.6; Go Online, Section 24.6; Go Online (web code: cdn-1246); Transparencies, T285 H.1E.1 Classify the bodies in our solar system based on properties and composition. Describe

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CHAPTER 5, Atomic Structure and the Periodic Table(continued) An atom of neon-22 has two more neutrons in its nucleus than an atom of neon-20. isotopes true b c a carbon-12; 12 amu false The values of atomic masses measured in grams are inconveniently small and impractical to work with.

5 ATOMIC STRUCTURE AND THE PERIODIC TABLE - Bergen

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