

## *Phase Change Diagram Answer Key*

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### Phase Change Diagram Answer Key

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Phase Change Diagram Answer Key. Directions: Label the phase change of each arc. Brainstorm at least one example for each phase change and write it under each phase change. In the boxes under the phases draw a small picture of how the molecules are arranged. Melting ex. Ice to water) Energy Absorbed Solid Vibratin Energy Released Freezing (ex.

### Phase Change Diagram Answer Key - pspb.org

Part D – Phase Diagram for Tastegudum. On Crosbia, bolonium (Bg) and manasium (Ma) react together to form the compound tastegudum. For each of the following questions (16-28), refer to the phase diagram for tastegudum. See Miss Scott for answer key with labels. Label the regions of the diagram that correspond to the solid, liquid, and vapor phases.

### Phase Diagram Worksheet

Phase Change Diagram Answer Key. Directions: Label the phase change of each arc. Brainstorm at least one example for each phase change and write it under each phase change. In the boxes under the phases draw a small picture of how the molecules are arranged. Melting ex. Ice to water) Energy Absorbed Solid Vibratin

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Phase Changes Model 2: Temperature of a Substance as Heat is Added Over Time boiling B melting lid Time increasing -¥ Key Questions 1. What is plotted on the x-axis and what is plotted on the y-axis of the graph in Model 2? 2. Over which line segments does temperature increase? 3. Over which line segments is there no change in temperature? 4.

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The Model: Phase Diagrams A phase diagram shows you the combinations of pressure and temperature that will result in a substance being a solid, a liquid, or a gas. The following is a sketch of a phase diagram of the typical substance (but not of water). The thick curves represent the combinations of pressure and temperature that result in an

### The Model: Heat and Phase Changes - Green River College

Unit 3: Phases of Matter-key Regents hemistry 14- Z15 Mr. Murdoch Page 15 of 63 Website upload 2014 Phase Change Diagrams (Heating/Cooling Curve) Page 14: When applying CONSTANT HEAT to a solid (point A), the temperature of the solid increases (segment AB) until the melting point (point B) is reached, for example of water ice at 0° or 273 K.

### Unit 3: Phases of Matter-key Regents hemistry 14 Mr ...

Students may have some prior knowledge for this lesson. What may differentiate it from middle school content is the vocabulary, and the accuracy of their phase change diagram. The materials you will need to do this lab are: a heat source (Bunsen burner or hot plate) ice. beakers. water. thermometers.

### Eleventh grade Lesson Phase Change | BetterLesson

Identify the choice that best completes the statement or answers the question.. 19. Rank the states of matter from weakest to strongest intermolecular forces. A. gas, solid, liquid C. solid, liquid, gas B. gas, liquid, solid D. solid, gas, liquid 20. Rank the states of matter from slowest to fastest particle speed.

### Unit 3 Test: States of Matter, Heat, Phase Changes 45 pts

A phase change is the reversible physical change that takes place when a substance changes from one state of matter to another. d a e f c b The temperature of a substance remains constant during

a phase change. 11. A substance absorbs energy from its surroundings during a(n) change.

### Chapter 3 States of Matter Section 3.3 Phase Changes

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Explore the relationship between molecular motion, temperature, and phase changes. Compare the molecular structure of solids, liquids, and gases. Graph temperature changes as ice is melted and water is boiled. Find the effect of altitude on phase changes. The starting temperature, ice volume, altitude, and rate of heating or cooling can be adjusted.

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