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the ejection of electrons by metals when light shines on them the region around the nucleus of an atom where an electron is likely to be moving An atomic orbital may describe at most two electrons. the regions within which electrons have the highest probability of being found When electrons occupy orbitals of equal energy, one

Name Date Class ELECTRONS IN ATOMS 5

The atom is a ball of positive charge with electrons stuck into the ball. Electrons are arranged in concentric circular paths around the nucleus. Most of an atom's mass is concentrated in the small, positively charged nucleus. The electrons surround the nucleus and the rest of the atom is empty space. false farther It is called a quantum.

SECTION 5.1 MODELS OF THE ATOM (pages 127-132)

atoms. • electron configuration: the arrangement of various orbitals around the nuclei of atoms

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Chapter 5

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The lowest energy level electron orbitals are filled first and if there are more electrons after the lowest energy level is filled, they move to the next orbital. The order of the electron orbital energy levels, starting from least to greatest, is as follows: 1s, 2s, 2p, 3s, 3p, 4s, 3d, 4p, 5s, 4d, 5p, 6s, 4f, 5d, 6p, 7s, 5f, 6d, 7p.

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