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POGIL | Chemistry

The molecular weight of O_2 is 32.00 u, so a mole of O_2 would have a mass of 32.00 g and would contain 6.022×10^{23} O_2 molecules. Each O_2 molecule is composed of two oxygen atoms, so one mole of O_2 contains two moles of oxygen atoms. The mole is sometimes described as "the chemist's dozen".

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