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1. solution. A well-mixed mixture containing a solvent and at least one solute that has the same properties throughout. 2. solvent. The part of a solution that is present in the largest amount and dissolves a solute. 3. solute. The part of a solution present in a lesser amount and dissolved by the solvent.

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chemical properties of a substance remain unchanged . Chemical Building Blocks SE/TE: 22-23, 40-41, 42-47, 48-53, 54 6.PS.NM.4. Describe that chemical and physical changes occur all around us (e.g., in the human body, cooking and industry). Chemical Building Blocks SE/TE: 9, 22-27, 32-33, 40-41, 42-47, 48-53, 54 Grade Seven Nature of Energy

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