Periodic Law Scramble Lab Answer Key

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2) The transition elements are missing from the abbreviated periodic table in this lab. Why do you think they were not included? 3) What is periodicity? How does it apply to this lab? 4) List three physical properties which distinguish metals from nonmetals. 5) Give a reason for the location of sodium in the periodic table.

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CH 5 LAB: PERIODIC LAW (CHEM.A.2.3) Purpose: In this experiment, you will use your knowledge of periodic properties and a list of clues to correctly arrange the elements from a scrambled periodic table. You will also predict values for any

CH 5 LAB: PERIODIC LAW (CHEM.A.2.3) Purpose

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The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided.

1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass.

(b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based

on when they were discovered.

5 The Periodic Law - Office of Instructional Technology

CHAPTER The Periodic Table and Periodic Law Section 6.1 Development of the Modern Periodic Table 1 session 1/2 block P LS 6 ... students answer the accompanying questions using Section Focus Transparency Master 20. ... 6.1 Development of the Modern Periodic Table 151 DISCOVERY LAB Materials tape samples of copper light socket with bulb, wires ...

Chapter 6: The Periodic Table and Periodic Law - irion-isd.org

THE PERIODIC LAW LAB CHAPTER 6/7 The present organization of the elements is a product of the first periodic table published by Dmitri Men- deleev in 1869. The amazing accuracty of his predictions has been very important to chemists the last 150 years. However, the basis of his arrangement was the atomic masses of the elements.

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Tags: Answers Assignment Element 101 was named in his honour lodine belongs to this group Most nonmetals are in this state Nonmetal that is a liquid at SATP Number of elements known in 1800 Periodic Table Silicon is in this group Silicon is in this period Synthetic elements with atomic numbers greater than 93 The first alkali metal The general ...

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