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2) The transition elements are missing from the abbreviated periodic table in this lab. Why do you think they were not included? 3) What is periodicity? How does it apply to this lab? 4) List three physical properties which distinguish metals from nonmetals. 5) Give a reason for the location of sodium in the periodic table.

CHEMISTRY LAB: THE PERIODIC LAW - Kwanga.net

CH 5 LAB: PERIODIC LAW (CHEM.A.2.3) Purpose: In this experiment, you will use your knowledge of periodic properties and a list of clues to correctly arrange the elements from a scrambled periodic table. You will also predict values for any

CH 5 LAB: PERIODIC LAW (CHEM.A.2.3) Purpose

THE PERIODIC LAW LAB CHAPTER 6/7 The present organization of the elements is a product of the first periodic table published by Dmitri Men- deleev in 1869. The amazing accuracy of his predictions has been very important to chemists the last 150 years. However, the basis of his arrangement was the atomic masses of the elements.

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Tags: Answers Assignment Element 101 was named in his honour Iodine belongs to this group Most nonmetals are in this state Nonmetal that is a liquid at SATP Number of elements known in 1800 Periodic Table Silicon is in this group Silicon is in this period Synthetic elements with atomic numbers greater than 93 The first alkali metal The general ...

Periodic Table Assignment Answers - SchoolWorkHelper

Periodic Table Mystery. By Chuck Roser, Retired Chemistry Instructor North Carolina School of Science and Mathematics. Objective. Using coded symbols for the main group elements in the first 4 periods of the periodic table, students will generate a periodic table from pertinent clues.

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