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Properties Of Metals Yahoo Answers

Answers. It belongs to the transition metals, and is hard and ductile. It occurs most usually in combination with sulfur and iron in pentlandite, with sulfur in millerite, with arsenic in the mineral niccolite, and with arsenic and sulfur in nickel glance. Because of its permanence in air and its inertness to oxidation, it is used in coins,...

Properties of metals? | Yahoo Answers

Which metals have these properties? and please write the use next to the metal with that property. ps. please try to use different metals each time. -metals are solid -metals are hard (scratch proof) -metals are strong -metals are shiny -metals bend -metals are tough (hard wearing) -metals feel cold -metals feel cold -metals conduct electricity metals are dense -some metals are magnetic ...

Properties of Metals? | Yahoo Answers

Best Answer: Properties of metals : Metals are solid at room temp. They are generally hard and strong. They have metallic lustre. They have high melting and boiling points. They are good conductors of heat and electricity. Metals are sonorous. Most metals have high tensile strength. They are malleable. They are ductile. They have high densities.

properties of metals and non-metals? | Yahoo Answers

Answers. Many of the properties of metals, including large atomic radius, low ionization energy, and low electronegativity, are due to the fact that the electrons in the valence shell of a metal atoms can be removed easily. One characteristic of metals is their ability to be deformed without breaking.

what are metals and what are the properties of metals? - Yahoo

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PROPERTIES OF METALS YAHOO ANSWERS

Best Answer: Physical Properties of Metals and Non-metals Physical Properties of Metals Physical State Metals are solids at room temperature with the exception of mercury and gallium, which are liquids at room temperature. Lustre Metals have the quality of reflecting light from its surface and can be ...

What are the properties of metals and non-metals? | Yahoo ...

Answers. Metals - good conductors of heat and electricity, good reflectors of light, malleable, ductile, tend to lose electrons in forming bonds, low electronegativity, low ionization energy Nonmetals - poor conductors of heat and electricity, poor reflectors of light, soft or brittle, tend to gain electrons in forming bonds,...

what are the general properties of metals? | Yahoo Answers

Best Answer: This is a main topic in chemistry and can be found in any book fcking canadian here u go -They are poor conductors of heat and electricity. -Solid nonmetals are brittle, with little or no metallic luster -Gain electrons quickly -Solids, liquids or gases at room temperature. there r more but ...

List 3 properties of non-metals.? | Yahoo Answers

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