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This activity will explore whether or not there is a limit to how much of one substance can dis-solve in another. Model 1 – Saturated and Unsaturated Solutions • All beakers contain 10.0 g of water. • All beakers are kept at 20 °C. • All solutions are stirred for 2 hours. • Solute is the same substance in all beakers.

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Eidell/Scott POGIL – Chem 504 6. Looking at the 3-dimensional drawings of a saturated and unsaturated fat on the right, why do you think saturated fats cause a build-up of fat/plaque in arteries? SATURATED FATS LIE VERY FLAT AND CAN PACK DENSLY WITH OTHER SATURATED FATS; THEY TEND TO BE SOLIDS AT ROOM TEMPERATURE AND CAN

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• Saturated and Unsaturated Solutions • Solubility • Molarity Thermochemistry • Calorimetry • Bond Energy Equilibrium • Equilibrium Acids and Bases ... 6 POGIL™ Activities for High School Chemistry 20. For each experiment in Model 2, determine the relationship between the independent and

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