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SAMPLE ANSWER: Many parts of an organism's body contain water. SAMPLE ANSWER: Some of the most important biological fluids contain water. Also, water's high heat capacity protects living things both internally and externally from drastic temperature changes. SAMPLE ANSWER: Carbon is the primary element found in living things.

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Student answers should include blood. Many sub-stances in the environment are colloids, such as smoke, fog, and dust. 2. Particles in a colloid are usually much smaller than particles in a suspension. Suspension particles tend to settle out after a period of time, and colloid particles will continue to mix with the liquid or gas that they are suspended in.

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The two main types of chemi- cal bonds are ionic and covalent bonds. Ionic bonds are formed when a transfer of electrons takes place from one atom to another, and covalent bonds are formed when electrons are shared between atoms. 7. An atom becomes an ion when it gains or loses electrons.

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The Chemistry of Life ANSWER KEY Section 2–1 The Nature of Matter (pages 35–39) This section identifies the three particles that make up atoms. It also explains how atoms of the same element can have a different number of neutrons and describes the two main types of chemical bonds. Atoms (page 35) 1. The basic unit of matter is called a(an) atom.

Chemistry of Life packet 1 ANSWER KEY (1) - The Chemistry ...

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Chapter 2 The Chemistry of Life Chapter Test B Multiple Choice Write the letter that best answers the question or completes the statement on the line provided. ____ 1. The three particles that make up an atom are a. protons, neutrons, and isotopes. b. neutrons, isotopes, and electrons. c. positives, negatives, and electrons.

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