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1) The molality of hydrochloric acid, HCl, in an aqueous solution is 8.80 mol/kg. What is the mole fraction of hydrochloric acid in the solution? 2) An aqueous solution is 6.00 PERCENT by mass hydrochloric acid, HCl, and has a density of 1.03 g/mL. The molality of hydrochloric acid in the solution is -----m

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