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Orbitals and Quantum Numbers Practice Questions

QUANTUM NUMBERS WORKSHEET 1. State the four quantum numbers, then explain the possible values they may have and what they actually represent. n - Pricipal Quantum Number: represents the energy level the electron is in, linked to the

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7. Answer the following questions: a) The quantum number n describes the	of an atomic
orbital. b) The shape of an atomic orbital is given by the quantum number _	d) The maximum
number of orbitals that may be associated with the set of quantum numbers	s n=4 and I=3 is

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You can print this interactive quiz and worksheet to use alongside the lesson on principal, angular momentum, magnetic, and spin quantum numbers....

Four Quantum Numbers: Principal, Angular ... - Study.com

1. The magnetic quantum number describes the: number of electrons. average distance of the most electron-dense regions from the nucleus. spatial orientation of the orbital. shape of the orbital. 2. There are ____ types of quantum numbers 2 5 7 4. 3. The principle quantum number is related to: the shape of the orbital. the spatial orientation of ...

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