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#### PRENTICE HALL CHEMISTRY CHEMICAL CALCULATION ANSWER KEY

• Simply put: the mass of a molecule. • Molecular mass is specifically for molecules. • Ionic compounds don't exist as molecules; for them we use ... • Formula mass: sum of the masses of the atoms or ions present in a formula unit. Calculate the molecular mass of glycerol (1,2,3-propanetriol).

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In chemical calculations, moles ratios are used to convert between moles of reactant(s) and moles of product(s) In a typical stiochiometric problem, the given quantity is usually converted to moles. Then the mole ratio from the balanced chemical equation is used to the moles of the wanted substance.

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the ratio of the actual yield to the theoretical yield expressed as a percent. Stoichiometry. that portion of chemistry dealing with numerical relationships in chemical reactions; the calculation of quantities of substances involved in chemical equations.

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Chemical Equations As reactants are converted to products we observe:

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