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What is the oxidation number for $\text{Rh}_3(\text{PO}_4)_2$? Upload failed. Please upload a file larger than 100x100 pixels; We are experiencing some problems, please try again.

Oxidation Number ? | Yahoo Answers

It shows the charge of the atom or molecule. A change in oxidation number can be used to determine which species in the reaction has changed and which is the redundant (undergoes oxidation, so it changes to a higher oxidation number) or the oxidation (undergoes reduction, so it changes to a lower oxidation number).

what is an oxidation number? | Yahoo Answers

Best Answer: Determination of the oxidation numbers is accomplished by using the rules given below: 1. The oxidation number of an atom in an element or in a molecule of an element is zero. For example, the oxidation number of chlorine in Cl_2 or phosphorus in P_4 is zero. Similarly the atoms of Na, Zn, Mg, Fe ...

oxidation number? | Yahoo Answers

Best Answer: No matter what the coefficient of the compound be but the oxidation state is seen between the elements among the compound here in NaOH oxidation number of sodium(Na^+) is +1 its better we call oxidation state rather than oxidation number this oxidation state doesnt depend on normality i.e ...

Oxidation Number? | Yahoo Answers

What's the oxidation number of 2I^{-} (2I negative sign on top; & thats an Iodine) and the oxidation number of I_2 is 0, right?

What is the oxidation number of 2I^{-} ? | Yahoo Answers

Best Answer: Carbon is +2/3 in $\text{C}_6\text{H}_8\text{O}_6$ and it is +1 in $\text{C}_6\text{H}_6\text{O}_6$. This compound (and many others) exhibit what is called an "average oxidation number." Remember, the assignment of numerical charges to atoms in a compound is our convenience. But with the varieties of covalent bondings, many times the average ...

Oxidation Number Calculation? | Yahoo Answers

For the best answers, search on this site <https://shorturl.im/avgBM> No matter what the coefficient of the compound be but the oxidation state is seen between the elements among the compound here in NaOH oxidation number of sodium(Na^+) is +1 its better we call oxidation state rather than oxidation number this oxidation state doesnt depend on normality i.e concentration or coefficient in a ...

what is oxidation number? | Yahoo Answers

Oxidation number is simply an "assigned number", much like the concept of "valence". Oxidation number is useful for explaining redox reactions and for figuring out chemical formulas. In addition, the oxidation number can also be the ionic charge when a compound dissociates into ions. Take HCl, which is a molecular compound.

what is the oxidation number of...? | Yahoo Answers

A. determine the oxidation numbers of the following: 1.chlorine in - ClO_2 - HClO_4 - Cl_2O_7 2. nitrogen in - NH_3 - HNO_2 - N_2O_5 3. carbon in - CCl_4 - H_2CO_3 4. chromium in - H_2CrO_4 - $\text{K}_2\text{Cr}_2\text{O}_7$ B. what is the oxidation number of the atom with quotation mark in the following formulas?

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