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describe atomic structure in terms of number of protons, neutrons and electrons in the atom; state the relative masses and relative charges of protons, neutrons and electrons; represent isotopes using atomic numbers and mass numbers, and use these together with ionic charges to calculate the number of protons, neutrons and electrons in atoms and ions

### **Atomic structure**

Dr. Tro has received Westmont's outstanding teacher of the year award three times. He is the successful author of several bestselling texts including, Introductory Chemistry, 6e (Pearson), Chemistry: A Molecular Approach, 4e (Pearson), and Chemistry: Structure and Properties, 2e (Pearson).

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### Atomic Structure Chemistry Quiz - ThoughtCo

There are three isotopes of silicon. Si-28 has an atomic mass of 27.98 amu and an abundance of 92.21%, Si-29 has an atomic mass of 28.98 amu and an abundance of 4.70%, and Si-30 has an atomic mass of 29.97 amu and an abundance of 3.09%. Calculate the average atomic mass of silicon.

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