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Quantitative Analysis Solution Stoichiometry

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general remarks. Magnesium can be easily determined by EDTA titration in the pH 10 against Eriochrome Black T. If the solution initially contains also different metal ions, they should be removed or masked, as EDTA react easily with most cations (with the exception of alkali metals).

Complexometric determination of magnesium - Titration

Chemistry 101 Class Notes Professor N. De Leon: TAKE AN ON-LINE EXAM Survey Results Spring 2001

C101 index - Indiana University Northwest

general remarks. Calcium can be determined by EDTA titration in solution of 0.1 M sodium hydroxide (pH 12-13) against murexide. Just like during determination of magnesium all metals other than alkali metals can interfere and should be removed prior to titration. Magnesium in that high pH precipitates as $\text{Mg}(\text{OH})_2$ and is not complexed by EDTA, thus its presence can be ignored.

Complexometric determination of calcium - Titration

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Potentiostatic coulometry. Potentiostatic coulometry is a technique most commonly referred to as "bulk electrolysis". The working electrode is kept at a constant potential and the current that flows through the circuit is measured. This constant potential is applied long enough to fully reduce or oxidize all of the electroactive species in a given solution.

Coulometry - Wikipedia

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Solubility equilibrium is a type of dynamic equilibrium that exists when a chemical compound in the solid state is in chemical equilibrium with a solution of that compound. The solid may dissolve unchanged, with dissociation or with chemical reaction with another constituent of the solvent, such as acid or alkali.

Solubility equilibrium - Wikipedia

Novel GC-MS-based approach for ¹³C metabolic flux analysis of pseudomonads and other microbes with cyclic hexose metabolism.. All fluxes in the recently discovered EDEMP cycle can be resolved.

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