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The two components of a conversion factor are equivalent measurements with different units. They are two ways of expressing the same quantity.

3.3 Solving Conversion Problems > - Scarsdale Middle School

170 Guided Reading and Study Workbook 4. Look at Figure 16.8 on page 481. Circle the letter of the best procedure for making a 0.50-molar (0.50M) solution in a 1.0-L volumetric flask. a. Add distilled water exactly to the 1.0-L mark, add 0.50 mol of solute, and

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Chapter 14

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11.2 Types of Chemical Reactions> 30 A halogen can also replace another halogen from a compound. •The activity of halogens decreases as you go down Group 7A of the periodic table—fluorine, chlorine, bromine, and iodine. •Bromine is more active than iodine, so this reaction occurs: $\text{Br}_2(\text{aq}) + 2\text{NaI}(\text{aq}) \rightarrow 2\text{NaBr}(\text{aq}) + \text{I}_2(\text{aq})$

11.2 Types of Chemical Reactions> - Quia

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