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Workbook Chapter 9 – The Mole Concept. W9.1 (a) 1 mole of Li atoms = 7 g. (b) 1 mole of Na atoms = 23 g. (c) 1 mole of Ca atoms = 40 g. (d) 1 mole of Fe atoms = 56 g. (e) 1 mole of Ag atoms = 108 g. (f) 1 mole of Pb atoms = 207 g. W9.2 (a) 1 mole of Cl atoms = 35.5 g 10 moles of Cl atoms = $35.5 \times 10 = 355$ g.

Workbook - Worked Solutions Chapter 4 - The Periodic ...

Chapter 7 (workbook) how hot or cold something is; a measure of the average energy of motion of the particles of a substance; the measure of the average kinetic energy of the particles of a substance.

Chapter 7 (workbook) Flashcards | Quizlet

Chapter 7 energy and Chemical reactions 249 energy...it makes things happen. To get an idea of the role energy plays in our lives, let's spend some time with John, a college student in one of the coastal towns in California. He wakes up in the morning to a beautiful sunny day and decides to take his chemistry book to the beach.

Chapter 7 - An Introduction to Chemistry: Energy and ...

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