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1. lithium chlorate: 2. ammonium carbonate: 3. barium sulfite: 4. molybdenum(VI) chloride: 5. diarsenic trisulfide: 6. cesium bromide: 7. copper(I) chloride

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Write the formula from the names for each: 21. cobalt(III) nitride. 22. sodium hydrogen sulfite. 23. aluminum carbide. 24. zinc sulfate. 25. disulfur dichloride

Nomenclature 2

Start studying Chemistry Chapter 7 Nomenclature Part 2 Test Review. Learn vocabulary, terms, and more with flashcards, games, and other study tools.

Chemistry Chapter 7 Nomenclature Part 2 Test Review - Quizlet

Nomenclature Worksheet 2: Simple Binary Ionic Compounds Please complete the following table: Formula of Ionic Compound s 2. 3. 4. 5. 6. 8. 9. Name of Ionic Compound

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Nomenclature, a collection of rules for naming things, is important in science and in many other situations. This module describes an approach that is used to name simple ionic and molecular compounds, such as NaCl, CaCO 3, and N 2 O 4. The simplest of these are binary compounds, those containing only two elements, but we will also consider how to name ionic compounds containing polyatomic ions ...

2.7 Chemical Nomenclature - Chemistry

WS 10.4 ORGANIC NAMING _Name (yours): _ Name Complete structural formula Lineformula Condensed structural formula Mol. form 1) butane IIII CH3CH2CH2CH3 C4H10

WS 10.7 Organic Nomenclature, part 2

View Notes - Worksheet #2 Answer Key from SCIENCE Accelerate at Cold Spring Harbor High School. ionic compounds worksheet #2

Worksheet #2 Answer Key - ionic compounds worksheet#2

lonic Nomenclature Worksheet I. Write the formula for each of the compounds. 1) copper(II) sulfate 10) barium hydroxide 2) ammonium nitrate 11) calcium chlorate 3) lithium dichromate 12) aluminum sulfite 4) magnesium acetate 13) chromium(II) oxide 5) sodium arsenate 14) potassium periodate

Ionic Nomenclature Worksheet

Basic Organic Nomenclature Packet 2-1 Name: _____ 1105:_Date ____ Use this packet and your book to answer the questions throughout this packet. Organic Nomenclature - Alkanes, Alkenes, Alkynes These are the rules of the IUPAC system used to name alkanes, alkenes and alkynes. These groups are hydrocarbons, compounds made of the elements carbon and hydrogen.

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