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Organic Chemistry Lab Report Reference Part 2 for Williamson Macroscale and Microscale. The 1-Step, 2-Step Hoedown: SN 1 and SN2 Nucleophilic Substitution and the Reactions of Alkyl Halides

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A polar aprotic solvent does not hydrogen bond to nucleophiles to a significant extent, meaning that the nucleophiles have greater freedom in solution. Under these conditions, nucleophilicity correlates well with basicity – and fluoride ion, being the most unstable of the halide ions, reacts fastest with electrophiles.

What Makes A Good Nucleophile? - Master Organic Chemistry

Chirality / k aɪ ' r æ l ɪ t i / is a geometric property of some molecules and ions. A chiral molecule/ion is non-superposable on its mirror image. The presence of an asymmetric carbon center is one of several structural features that induce chirality in organic and inorganic molecules. The term chirality is derived from the Ancient Greek word for hand, $\chi \epsilon \tilde{\iota} \rho$ (kheir).

Chirality (chemistry) - Wikipedia

Hydrogen iodide (H I) is a diatomic molecule and hydrogen halide. Aqueous solutions of HI are known as hydroiodic acid or hydriodic acid, a strong acid. Hydrogen iodide and hydroiodic acid are, however, different in that the former is a gas under standard conditions, whereas the other is an aqueous solution of said gas.

Hydrogen iodide - Wikipedia

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En chimie, l'équation de Henderson-Hasselbalch est une équation donnant le pH d'un système tamponné. Elle est notamment utilisée en physiologie et en médecine pour déterminer le pH sanguin à partir des concentrations en ion bicarbonate et en acide carbonique

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