

Pogil Bond Energy Answer Key

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Chem 115 POGIL Worksheet - Week 11 - Solutions Chemical Bonds Key Questions 1. Using the electronegativity values in the table above, calculate $\delta\chi$ for the pairs of elements forming bonds in each of the following species, and describe the type of bonding as either pure covalent, polar covalent, or ionic: H_2 , HCl , PbBr_2 , NaCl , NCl_3 .

Chem 115 POGIL Worksheet - Week 11 - Solutions Chemical Bonds

A good answer should state that as the distance between a proton (nucleus) and electron increases,

the attractive force decreases. This is followed by a quick class discussion to summarize the previous day's lesson and leads quickly into model 3 and 4 of the POGIL activity.

answer key - BetterLesson

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Bond energy is defined as the amount of energy required to break a bond. These values are positive, indicating that bond breaking is endothermic. Bond energies are reported in kilojoules per mole (kJ/mol). The energy for breaking a hydrogen-hydrogen bond is 436 kJ/mol so when a hydrogen-hydrogen bond is formed the process releases 436 kJ/mol.

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