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STRUCTURAL ISOMERISM - INTRODUCTION COMPOUNDS HAVE THE SAME MOLECULAR FORMULA BUT DIFFERENT STRUCTURAL FORMULA Chain different arrangements of the carbon skeleton similar chemical properties slightly different physical properties more branching = lower boiling point Positional same carbon skeleton same functional group functional group is in a ...

Isomerism Power point - SlideShare

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• What are the general structural and chemical characteristics of polymer molecules? • What are some of the common polymeric materials, and how do they differ chemically? • How is the crystalline state in polymers different from that in metals and ceramics? Chapter 14:Polymer Structures

Chapter 14: Polymer Structure

Correct Answer: A) They are both polymers of glucose. Incorrect Answers: (C) Neither starch nor cellulose can be digested by humans (D) Only starch is used for energy storage in plants (E) Only cellulose is a structural component of the plant cell wall (B) They are not geometric isomers of each other.

CAMPBELL BIOLOGY CHAPTER 5 Flashcards | Quizlet

The Structure and Properties of Polymers Also known as Bonding + Properties What is a polymer? A long molecule made up from lots of small molecules called monomers.

The Structure and Properties of Polymers

CHAPTER 5: STRUCTURE OF POLYMERS "The time has come," the Walrus said, "To talk of many things: Of shoes--and ships--and sealing-wax-- Of cabbages--and kings--" Lewis Carroll, Through the Looking Glass (1872) shoes, ship, sealing wax, cabbage, and a king The "many things" listed by the Walrus are actually very similar in chemical composition

CHAPTER 5: STRUCTURE OF POLYMERS

understanding of how polymer liquids difier from other liquids and from other forms of condensed matter. Ex-cellent treatments exist already (Jannink et al. 1992), notably Scaling Concepts in Polymer Physics by P. G. deGennes (DeGennes 1979). Here we adopt a geometric approach, exploiting the \fractal" structure of the poly-mers.

Polymer Solutions: a Geometric Introduction

Introduction to polymers . 2.3.3 Geometrical isomerism ... But there is an alternative structure with exactly the same formula: the polymer is gutta percha, and its structural formula is shown below that of NR in Figure 19. Here, the two parts of the chain are on opposite sides of the double bond, giving the overall chain a zig-zag appearance. ...

Introduction to polymers: 2.3.3 Geometrical isomerism ...

Polymers that contain only a single type of monomer are known as Homopolymers. Example Poly(styrene) Polymers containing a mixture of monomers are known as Copolymers. Example Ethylene-vinyl acetate. A polymer molecule containing ionizable subunits is known as a Polyelectrolyte.

Polymers | authorSTREAM

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3.Find out the appropriate VSEPR geometry for the specified number of electron pairs, both bonding and lone pairs. 4.Use the positions of atoms to establish the resulting molecular geometry. Multiple Bonds and Molecular Geometry Multiple bonds count as one - e.g. 4 bonding pairs around C, but trigonal planar instead of tetrahedral.

Chemical Bonding II: Molecular Geometry and Hybridization ...

s geometric isomers positional isomers chiral ... Compounds with the same molecular formula but different structural formula ... | PowerPoint PPT presentation | free to view Isomers and polymers - Monomer: the smallest repeating unit of a polymer (propene in polypropylene).

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