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Ch 17 Thermochemistry Practice Test Matching Match each item with the correct statement below. a. calorimeter d. enthalpy b. calorie e. specific heat c. joule f. heat capacity ____ 1. quantity of heat needed to raise the temperature of 1 g of water by 1°C 2. SI unit of energy 3.

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Chapter 17 Thermochemistry183 SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages

505–510) This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. Energy Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that

SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505-510)

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Chapter 17 - Thermochemistry - 17.4 Calculating Heats of ...

Thermochemistry Test Preview Matching Match each item with the correct statement below. a. calorimeter d. enthalpy b. calorie e. specific heat c. joule f. heat capacity ____ 1. quantity of heat needed to raise the temperature of 1 g of water by 1 C 2.___ SI unit of energy 3.___ quantity of heat needed to change the temperature of 1 g of a ...

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05 CTR ch17 7/12/04 8:15 AM Page 429 THE FLOW OF ENERGY ...

Heat = Transfer of Energy The 3 methods that heat can be transferred: Conduction - heat transfers by direct contact Convection - is the process in which heat is carried from one place to another by the bulk movement of a fluid. Radiation- heat transfers by electromagnetic waves.

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05 CTR ch17 7/12/04 8:15 AM Page 431 MEASURING AND ...

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