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4 *P46937A0424* DO NOT WRITE IN THIS AREA DO NOT WRITE IN THIS AREA DO NOT WRITE IN THIS AREA 6 The table shows the mean bond enthalpies for some covalent bonds. Covalent bond C C C C Br Br C Br C H Mean bond enthalpy / kJ mol⁻¹ 347 612 193 290 413 What is the approximate enthalpy change, in kJ mol⁻¹, for the reaction shown? H C C H

Advanced Level Chemistry - Pearson qualifications

is spontaneous. A solution containing K₂Cr₂O₇ and H₂SO₄ is poured into one beaker, and a solution of KI is poured into another. A salt bridge is used to join the beakers. A metallic conductor that will not react with

Sample Exercise 20.1 Identifying Oxidizing and Reducing Agents

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