

## ***Pressure Unit Conversions Worksheet Answer Key***

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### **Pressure Unit Conversions Worksheet Answer**

PRESSURE UNIT CONVERSIONS WORKSHEET 1 atm = 760 mm Hg = 101325 Pa = 14.7 lb/in<sup>2</sup> = 1.013 bar 1. The air pressure for a certain tire is 109 kPa. What is this pressure in atmospheres? (Answer: 1.08 atm). 2. The air pressure inside a submarine is 0.62 atm. What would be the height of a column of mercury balanced by this pressure? (Answer: 470 mm Hg ...

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### **Pressure Conversion Worksheet Answer Key | Gas Laws Unit ...**

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### **Pressure Conversion Worksheets - Printable Worksheets**

Quiz & Worksheet - Units and Conversions of Pressure Quiz; ... receive your score and answers at the end. ... - access the knowledge you have gained regarding units that are used to measure pressure

### **Quiz & Worksheet - Units and Conversions of Pressure ...**

Convert this pressure to psi. 15.2 psi 101.3 kPa 14.7 psi 1 ... A gas pressure results from the many collisions between gas particles and their container. One common unit for pressure is the newton per square meter (N/m<sup>2</sup>). This unit is called a pascal ... Pressure Conversions Name \_\_\_\_ Chem Worksheet 13-1 FORCE AREA AREA FORCE

### Pressure Conversions Name Chem Worksheet 13-1

Chemistry: Unit Conversions for the Gas Laws Directions: Complete the following tables, showing your work for each lettered box beside the corresponding letter below. Include units on your work, and write your final answers in the tables. TEMPERATURE PRESSURE K oC mm Hg kPa atm 373 K (D) 100 oC 890 mm Hg (K) 118.6 kPa (O) 1.17 atm

### Unit Conversions for the Gas Laws - teachnlearnchem.com

Resource Pressure Conversion Worksheet Answer Key . Pressure Conversion Worksheet Answer Key . Created By laura\_webb; In 3 Playlist(s) Resource Playlists. Gas Laws; Gas Laws; Gas Laws Unit; Description: To provide easy access to answers for teachers. Purpose:

### Pressure Conversion Worksheet Answer Key | Gas Laws ...

The same bag of potato chips from problem 5 is shipped to Denver, Colorado, where the atmospheric pressure is 99.82 kPa. What is the difference (in Pa) between the pressure in the bag and the atmospheric pressure?  $99.82 \text{ kPa} \times 1000 \text{ Pa} = 99820 \text{ Pa}$  .  $99820 \text{ Pa} - 763.1 \text{ Pa} = 1678 \text{ Pa}$  . 1 kPa. Pressure Unit Conversions Worksheet

## Pressure Unit Conversions Worksheet Answer Key

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