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How does nuclear radiation affect atoms? Nuclear radiation can ionize atoms.15. Is the following sentence true or false? One potential danger of radon gas is that prolonged exposure to it can lead to lung cancer. trueDetecting Nuclear Radiation (page 297)16. Name two devices that are used to detect nuclear radiation.a. Geiger counters b.

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Answer Key to "Nuclear Chemistry Practice" Problems 1. Predict the type of radioactive decay expected for each nuclide I made predictions first, and then checked on the web to see the decay process that actually has been

### **Answer Key to "Nuclear Chemistry Practice" Problems 1 ...**

Answer Key for Nuclear Chemistry Worksheet #1: Nuclear Decay Processes Chem 160 - K. Marr Key Questions 1. What does the symbol " e" tell the reader? (i.e., What does the superscript 0 mean? What can a subscript -1 possibly mean? Why is the beta particle symbolized with an e?)

### **Answer Key for Nuclear Chemistry Worksheet #1: Nuclear ...**

strong nuclear force: the powerful attractive force that binds protons and neutrons together in the nucleus: fission: a nuclear reaction in which an atomic nucleus is split into 2 smaller parts: chain reaction: a series of fission reactions triggered by neutrons released during the fission of a nucleus: critical mass

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Chapter 10-1 Chapter 10 Nuclear Chemistry Solutions to In-Chapter Problems 10.1 Refer to Example 10.1 to answer the question. • The atomic number (Z) = the number of protons. • The mass number (A) = the number of protons + the number of neutrons. • Isotopes are written with the mass number to the upper left of the element symbol and the

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