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Pearson Physical Chemistry Ch 13

P51460RA0324 3 Turn over DO NOT RITE IN THIS AREA DO NOT RITE IN THIS AREA DO NOT RITE IN THIS AREA 2 This question is about the molar volume of gases. (a) (i) Calculate the volume of one mole of an ideal gas, A, at 60 °C and 500 kPa pressure. Give your answer to two significant figures and include units.

Pearson Edexcel GCE Chemistry

P44881A0524 5 Turn over 8 Which of the following properties decreases on descending Group 2 of the Periodic Table? A Solubility of the sulfates. B Solubility of the hydroxides. C Reactivity of the elements. D Ionic character of the oxides. (Total for Question 8 = 1 mark) 9 Compound X is an anhydrous, white solid which decomposes on heating to form a white solid residue, a colourless gas, and ...

p44881a gce chemistry advsub unit 2 6ch02 01 jun15

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International Union of Pure and Applied Chemistry - Wikipedia

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for the molal (m) case, where a_{H^+} is the hydrogen ion activity, γ_{H^+} and f_{H^+} are activity coefficients of the hydrogen species in their respective proton scales. pH can be measured based on multiple assumptions, which we will discuss below. Also, chemical equilibrium models can be used to calculate activity coefficients and solution concentrations, which allow estimation of pH.

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