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Nuclear Transformations •Protons and Neutrons are held together in the Nucleus of an atom by the nuclear force. •Without the nuclear force, the like-charged protons would repel and fly apart. •Atoms undergo radioactive decay due to the neutron-proton ratio. -If the number of neutrons are much greater than the

Chapter 25 - Nuclear Chemistry

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268 Guided Reading and Study Workbook CHAPTER 25, Nuclear Chemistry (continued) Types of Radiation (pages 800–802) 6. Complete the following table showing some characteristics of the main types of radiation commonly emitted during radioactive decay. 7. Look at Figure 25.2a on page 801. It shows the alpha decay of uranium-238 to thorium-234. a.

SECTION 25.1 NUCLEAR RADIATION (pages 799-802)

How does nuclear radiation affect atoms? Nuclear radiation can ionize atoms.15. Is the following sentence true or false? One potential danger of radon gas is that prolonged exposure to it can lead to lung cancer. trueDetecting Nuclear Radiation (page 297)16. Name two devices that are used to detect nuclear radiation.a. Geiger counters b.

Chapter 10 Nuclear Chemistry Section 10.1 Radioactivity ...

Nuclear Chemistry Worksheet K Directions: Identify the following as alpha, beta, gamma, Name: Period: or neutron. Date: 231 90 91 Kr + 3 36 5. 6. 8. 9. Nuclear decay with no mass and no charge

An electron Least penetrating nuclear decay Most damaging nuclear decay to the human body Nuclear decay that can be stopped by skin or paper. 3. Owl phcx bescx 10.

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