

Oxidation Reduction Reaction Answer Key

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Oxidation Reduction Reaction Answer Key

Oxidation Reduction Reactions- Answer Key. 4.51. If nitric acid is a strong oxidizing agent and zinc is a strong reducing agent, then zinc metal will probably reduce nitric acid when the two react; that is, N will gain electrons and the oxidation number of N must decrease.

Oxidation Reduction Reactions- Answer Key

Practice Problems: Redox Reactions (Answer Key) ... Write balanced equations for the following redox reactions: a. $2 \text{NaBr} + \text{Cl}_2 \rightarrow 2 \text{NaCl} + \text{Br}_2$ b. $\text{Fe}_2\text{O}_3 + 3 \text{CO} \rightarrow 2 \text{Fe} + 3 \text{CO}_2$ in acidic solution ... Write the balanced half reactions of the following reactions: a. $\text{NiO} + 2 \text{H}_2\text{O} + \text{Fe} \rightarrow \text{Ni} + \dots$

Practice Problems: Redox Reactions (Answer Key)

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Which is an oxidation-reduction reaction? A. $4\text{Na} + \text{O}_2 \rightarrow 2\text{Na}_2\text{O}$ B. $3\text{O}_2 \rightarrow 2\text{O}_3$ C. $\text{AgNO}_3 + \text{NaCl} \rightarrow \text{AgCl} + \text{NaNO}_3$ D. $\text{KI} + \text{K}^+ + \text{I}^- \rightarrow 2\text{I}$ 21. Which is a redox reaction? ... Which statement best describes the key? A. It acts as the cathode and is negative. B. It acts as the cathode and is positive. ... Base your answer(s) to the following question(s) on the diagram below ...

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there any features of the redox reactions that would allow you to identify them as redox reactions? If y, specify es m Model 1 to support your answer. 6. Identify the following reactions as either redox or non ox using oxidation numbers as evidence. a. $2\text{NaI}(\text{aq}) + \text{PbI}_2(\text{s}) \rightarrow 2\text{NaNO}_3(\text{aq}) + \dots$ Redox Intro Key ...

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For each reaction in problem 13, identify the oxidizing agent and reducing agent. 15. Write half-reactions for the oxidation and reduction process for each of the following. ... Chapter 20 Worksheet: Redox ANSWERS I. Determine what is oxidized and what is reduced in each reaction. Identify the oxidizing agent and the reducing agent, also. 1 ...

Chapter 20 Worksheet Redox - bhhs.bhusd.org

Worksheet 25 - Oxidation/Reduction Reactions Oxidation number rules: Elements have an oxidation number of 0 Group I and II - In addition to the elemental oxidation state of 0, Group I has an oxidation state of +1 and Group II has an oxidation state of +2. Hydrogen -usually +1, except when bonded to Group I or Group II, when it forms hydrides, -1. ...

Worksheet 25 - Oxidation/Reduction Reactions 0 II +1 +2 -2 -1

Oxidation Reduction Reactions- Answer Key. 4.43. Even though the problem doesn't ask you to assign oxidation numbers, you need to be able to do so in order to determine what is being oxidized or reduced. (i) Half Reactions (ii) Oxidizing Agent (iii) Reducing Agent (a)

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