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In a reaction, $A + B \rightarrow Product$, rate is doubled when the concentration of B is doubled, and rate increases by a factor of 8 when the concentrations of both the reactants (A and B) are doubled, rate law for the reaction can be written as [CBSE AIPMT 2012]

Chemical Kinetics MCQ | Questions - Paper 1

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KINETICS Practice Problems and Solutions

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Chemistry Review Activities - ScienceGeek.net

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A practice analysis of toxicology - ScienceDirect

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1. A full-term female infant failed to gain weight and showed metabolic acidosis in the neonatal period. A physical examination at 6 months showed failure to thrive, hypotonia, small muscle mass, severe head lag, and a persistent acidosis (pH 7.0 to 7.2).

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