

Periodic Trends Worksheet Answers

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Periodic Trends Worksheet Answers

Periodic Trends Worksheet (Answer Key 1. Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:)) Cu) K) Ni) Br))) Largest - K Smallest - Br) Explain why you made these choices.) Atomic radius decreases as you go left to right across a period. Potassium is in the far

Periodic Trends Worksheet Answers

Periodic Trends. Showing top 8 worksheets in the category - Periodic Trends. Some of the worksheets displayed are Periodic trends work, Periodic trends work answers, Name work periodic trends period, Work periodic table trends name, Periodic trends work, Chem 115 pogil work, Periodic trends work, Periodic trends.

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Answers for Comparing Atomic Sizes Here are answers for the questions above. a. Li, C, F All are in the same period and thus have the same number of energy levels. Therefore, the important factor is the nuclear charge. Li is the largest because it has the smallest nuclear charge and pulls the ... Microsoft Word - Periodic Trends Worksheet Author:

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Periodic Trends Worksheet ,ctions: Use your notes to answer the following questions. 1. Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium. 2. Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum. 3. Why does fluorine have a higher ionization energy than iodine ...

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Periodic Trends Worksheet. Directions: Use your notes to answer the following questions. Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium. Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.

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ANSWER KEY Trends on the Periodic Table Part 1: Identifying chemical properties/characteristics of the elements For each of the following elements, draw a caricature or cartoon above the description to show its

ANSWER KEY Trends on the Periodic Table

Name_____ Worksheet: Periodic Trends Period_____1. Which statement best describes Group 2 elements as they are considered in order from top to bottom of the Periodic Table? (A) The number of principal energy levels increases, and the number of valence electrons increases.

Name Worksheet: Periodic Trends Period - kmacgill.com

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To reinforce the trends I have students perform a practice worksheet. The worksheet contains two sections. 1. For the first section students explain characteristics of the trends ... trends in the periodic table, ... As students are done I make sure that they have the correct trends. This is the answer key for expected answers. Graphing ...

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