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Kinetic Molecular Theory. Model 2 POSTULATES OF THE KINETIC MOLECULAR THEORY 1. Gases consist of tiny particles (atoms or molecules). 2. These particles are so small, compared with the distances between them that the volume (size) of the individual particles can be assumed to be negligible (zero).

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The Kinetic Molecular Theory states that the pressure of a gas results from. the mass of particles. the collisions of particles with the walls of a container. the forces of repulsion between the particles. the energy loss experienced by each particle.

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