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# Naming Ionic Pounds POGIL Answer Key | School. | Key, Ap ...

4 POGIL™ Activities for High School Chemistry 16. For each of the following compounds, indicate whether or not your naming rules from Question 15 will apply. If not, explain why the naming rules do not apply. Fel 3 ICl 5 HBrO 4 17. Using the rules your group developed in Question 15, name each of the following molecular compounds.

#### **Pogil Naming Molecular Compounds**

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Chem 115 POGIL Worksheet - Week 3 Compounds, Naming, Reaction Equations, and Formula Weights Answers to Key Questions and Exercises Key Questions & Exercises 1. Do all compounds contain molecules? No. Only molecular compounds contain molecules of the compound. (Ionic compounds may contain molecular ions, but these compounds are not composed of ...

# Compounds, Naming, Reaction Equations, and Formula Weights ...

Naming Ionic Compounds 1 ... 6. Use the table of ions in Model 1 to answer the following questions: a. In the compound zinc phosphide, what is the charge on the zinc ion? ...  $4 \,^{\text{TM}}$  POGIL Activities for High School Chemistry 21. Fill in the table below using what you've learned from Model 3.

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Chem 115 POGIL Worksheet - Week 3 Compounds, Naming, Reaction Equations, and Formula Weights Why? Compounds are generally classified as molecular, ionic, or (more rarely) network. Knowing the classification allows us to name the compound correctly and to understand the microscopic organization of it.

#### Chem 115 POGIL Worksheet - Week 3 Compounds, Naming ...

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chemistry became a real science, chemists realized that all mater was made of atoms and that chemical processes were simply a rearrangement of these atoms. Chemists needed some simple, shorthand way to show this fact, and thus created chemical formulas. Success criteria You should be able to write the correct formula for any ionic compound

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Naming Acids Worksheet Key Block\_\_\_\_ Name\_\_\_\_ Name the following acids 1. HCL hydrochloric acid 2. HClO4 perchloric acid 3. HIO3 iodic acid 4. HI hydroiodic acid 5. H2SO4 sulfuric acid 6. H2S hydrosulfuric acid 7. HCN hydrocyanic acid 8. H2CO3 carbonic acid 9.

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