

## *Principles Of Modern Chemistry 6th Edition Solutions Manual*

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### **Principles Of Modern Chemistry 6th**

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In order for NaCl to be soluble, the Na<sup>+</sup> and Cl<sup>-</sup> ions must break free from the crystal-lattice structure of the solid. When the ions are in solution, they are surrounded by water molecules, and the ions are said to be solvated, or dissolved in an aqueous solution, denoted (aq). Hence, the process of dissolving a NaCl crystal can be described by the following chemical equation (Equation 1):

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CHAPTER XI ASTROLOGY AS AN AID TO HEALING ASTROLOGICAL BASIS FOR HEALING: It is well known to the modern physician that the condition of the blood, and therefore the condition of the whole body, changes in sympathy with the state of mind of the patient, and the more the physician uses suggestion as an adjunct to medicine the more successful he is.

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