

## *Nuclear Chemistry Topic 12 Answer Key*

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ANSWERS To TOPIC 6 columns of elements with similar chemical properties.  $\text{Mg}^{2+}$  and  $\text{Al}^{3+}$  are isoelectronic and have the electron configuration of Ne.  $\text{Br}^{-}$ ,  $\text{Cl}^{-}$ , and K are also isoelectronic and have the electron configuration of Ar.  $\text{Ba}^{2+}$  has the electron configuration of Xe. 13 Ge and AS Group 1 metals all become ions with a 1+ charge.

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