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In chemistry, the linear molecular geometry describes the geometry around a central atom bonded to two other atoms (or ligands) placed at a bond-angle of 180°.Linear organic molecules, such as acetylene (HC≡CH), are often described by invoking sp orbital hybridization for their carbon centers.. According to the VSEPR model, linear geometry occurs at central atoms with two bonded atoms and ...

Linear molecular geometry - Wikipedia
Valence shell electron pair repulsion (VSEPR) theory is a model used in chemistry to predict the geometry of individual molecules from the number of electron pairs surrounding their central atoms. It is also named the Gillespie-Nyholm theory after its two main developers, Ronald Gillespie and Ronald Nyholm.The acronym "VSEPR" is pronounced either "ves-pur": 410 or "vuh-seh-per".

VSEPR theory - Wikipedia
In this lesson we'll define coordinate geometry and learn about formulas that are commonly used in coordinate geometry. You can then take a brief quiz to see what you learned.

Coordinate Geometry: Definition & Formulas - Study.com
The lesson will explore the history and nature of Euclidean geometry, including its origins in Alexandria under Euclid and its five postulates.

Euclidean Geometry: Definition, History & Examples - Video ...
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Random Number -- from Wolfram MathWorld
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Probably the most famous of the organometallic compounds is the exceedingly stable orange ferrocene, [Fe(C 5 H 5) 2].Its discovery in 1951 triggered an explosion in coordination chemistry , and the subsequent preparation of thousands of new organometallic compounds.Ferrocene consists of two planar, five-membered rings sandwiching an iron atom between them.

Coordination Compounds - Chemistry Encyclopedia ...
Laplace Transform. The Laplace transform is an integral transform perhaps second only to the Fourier transform in its utility in solving physical problems. The Laplace transform is particularly useful in solving linear ordinary differential equations such as those arising in the analysis of electronic circuits.. The (unilateral) Laplace transform (not to be confused with the Lie derivative ...

Laplace Transform -- from Wolfram MathWorld
원자가껍질 전자쌍 반발 이론(Valence shell electron pair repulsion theory : VSEPR theory)이란 화학에서 중심원자의 배위수와 전자쌍 반발 원리를 통해 분자의 구조를 예측, 나타내는 모형이다. (루이스 구조식에서 비공유 전자쌍을 나타내주어야 한다) 이 이론은 루이스 구조에서 나타나는 중심원자의 각 ...

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