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P44881A0324 3 Turn over 4 In which series of compounds does covalent character increase when going from left to right? A KI, KBr, KCl B NaI, KI, RbI C NaCl, MgCl 2, AlCl 3 D SO 2, P 4O 10, SiO 2 (Total for Question 4 = 1 mark) 5 Hydrogen bromide has a lower boiling temperature than hydrogen iodide. This is because A hydrogen bromide has a smaller permanent dipole than hydrogen iodide.

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