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Holt ChemFile: Problem-Solving Workbook 49 Mole Concept Name Class Date Problem Solving continued PROBLEMS INVOLVING ATOMS AND ELEMENTS Sample Problem 1 A chemist has a jar containing 388.2 g of iron filings. How many moles of iron does the jar contain? Solution ANALYZE What is given in the problem? mass of iron in grams

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Holt ChemFile: Problem-Solving Workbook 99 Stoichiometry Name Class Date Problem Solving continued Sample Problem 1 Ammonia is made industrially by reacting nitrogen and hydrogen under pressure, at high temperature, and in the presence of a catalyst. The equation is $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$. If 4.0 mol of H_2 react, how many moles of NH_3 will ...

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Holt ChemFile: Problem-Solving Workbook 99 Stoichiometry Name Class Date Problem Solving continued Sample Problem 1 Ammonia is made industrially by reacting nitrogen and hydrogen under pressure, at high temperature, and in the presence of a catalyst. The equation is $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$. If 4.0 mol of H_2 react, how many moles of NH_3 will ...

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Holt Chemistry 12 Gases. Name Class Date Problem Solving continued Boyle's law can be expressed by the following mathematical equation. = ... Problem Solving continued Sample Problem 1 A sample of nitrogen collected in the laboratory occupies a volume of 725 mL at a pressure of 0.971 atm. What volume will the gas occupy at a

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Holt Chemistry 15 Solutions Name Class Date Problem Solving continued Sample Problem 2 What is the molarity of a solution prepared by dissolving 37.94 g of potassium hydroxide in some water and then diluting the solution to a volume of 500.00 mL? Solution ANALYZE What is given in the problem? the mass of the solute, KOH, and the final volume ...

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Holt ChemFile: Problem-Solving Workbook 127 Percentage Yield Name Class Date Problem Solving continued Sample Problem 2 Acetylene, C_2H_2 , can be used as an industrial starting material for the production of many organic compounds. Sometimes, it is first brominated to form 1,1,2,2-tetrabromoethane, $\text{CHBr}_2\text{CHBr}_2$, which can then be

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Holt ChemFile: Problem-Solving Workbook 5 Conversions Name Class Date Problem Solving continued Sample Problem 2 A metallurgist is going to make an experimental alloy that requires adding 325 g of bismuth to 2.500 kg of molten lead. What is the total mass of the mixture in kilograms? Solution ANALYZE

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Holt Chemistry 36 Solutions Name Class Date Problem Solving continued General Plan for Solving Problems Involving Freezing-Point Depression and Boiling-Point Elevation Mass of solute Convert using the molar mass of the solute. 1 Molal concentration of particles in solution, m 4 Boiling-point elevation, t b 5b Freezing-point depression, t f 5a ...

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Holt ChemFile: Problem-Solving Workbook 58 Mole Concept Name Class Date Problem Solving continued CONVERTING NUMBER OF ATOMS OF AN ELEMENT TO MASS Sample Problem 4 uses the progression of steps 1 → 2 → 3 to convert from the mass of an element to the number of atoms.

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