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States of Matter Class 11: Overview In our everyday life, we can observe four different states of matter, namely solid, liquid, gas and plasma. However, there are numerous other states that can be seen to exist but only under extreme conditions. The ones worth mentioning are glass and Bose-Einstein condensates.

States of Matter for Class 11: Different States ...

CBSE Class 11 Chemistry Notes: States of Matter. (ii) Boiling point This is the temperature at which the vapour" pressure of a liquid becomes equal to the atmospheric pressure. It increases in the presence of impurity and with rise in pressure. Boiling point of water is 100°C. (iii) Freezing point At this temperature,...

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in three states is depicted as follows: We have already learnt the cause for the existence of the three states of matter. Now we will learn more about gaseous and liquid states and the laws which govern the behaviour of matter in these states. We shall deal with the solid state in class XII. 5.4 THE GASEOUS STATE This is the simplest state of matter.

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existence of the three states of matter. Now we will learn more about gaseous and liquid states and the laws which govern the behaviour of matter in these states. We shall deal with the solid state in class XII. 5.4 THE GASEOUS STATE This is the simplest state of matter. Throughout our life we remain immersed in the ocean of air which is a ...

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States of matter We know matter exists in three states Solid, liquid and gas. In this chapter, we will learn more about States of matter from perspective of intermolecular forces, Thermal interaction Intermolecular Forces These are the forces of attraction and repulsion between interacting particles (atoms and molecules)

States of Matter | Intermolecular Forces | Class 11 Chemistry

States Of Matter Class 11 Notes - Chapter 5 The intermolecular forces run between the particles of matter. There exists a pure electrostatic force between two ions which are oppositely charged. The intermolecular forces are different from these pure electrostatic forces.

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