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Molar solutions. Divide the mass of acid by its density (1.049 g/mL) to determine the volume (114

mL). Use either 120.1 g or 114 mL acetic acid to make the solution. Swirl the flask gently to mix the solution. Once the solution is at room temperature, dilute to the mark and invert the flask several times to mix.

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3: What is the relationship between the compounds in #1 and #2. Since they have the same formula, they are the same. geometric isomers. conformers

EXERCISE 16-1 Organic Chemistry

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