

## *Stoichiometry Balloon Race Lab Answers*

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**Stoichiometry Balloon Race Lab Answers**

Stoichiometry quantitatively describes relationships between substances in a reaction. It is used for balancing chemical equations. The ratio between reactants is an integer ratio. A stoichiometric number counts molecules that are consumed (negative) and produced (positive) in a reaction. Stoichiometry is based on the law of mass conservation, where amounts of reagent and product are equal.

**My PBL Project - STOICHIOMETRY BALLOON RACE**

Stoichiometry Balloon Races continued 2 21 linn Scientific nc ll Rights Reserved 4. Allow the reactions to proceed until the bubbling stops. Compare the size of the inflated balloons and whether the solid has dissolved in each case. (The balloon size should increase fairly uniformly for flasks 1-5, and then stay constant. It may be hard to tell

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Balloon Stoichiometry. Add to Favorites. 23 teachers like this lesson. Print Lesson. ... I then go over the answers with students and answer their questions. ... I begin by passing out the Balloon Lab paper to students. I have students look over the lab paper for 2-3 minutes.

**Ninth grade Lesson Balloon Stoichiometry | BetterLesson**

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**Stoichiometry Balloon Races Lab Answers**

Stoichiometry 3 weeks- TEKS: 8E What students need to know? TEACH THIS!!! § Calculation of moles of reactants or products from given moles Stoichiometry 5E lesson Balloon Race Activity Introduction to Stoichiometry Worksheet with Ans Key § Calculation of mass of reactants or products from given moles or mass (Moles to Mass, Mass to Mass Calculations) Stoichiometry 5E lesson Flinn ...

**2. Stoichiometry - HARMONY CHEMISTRY**

Stoichiometry Balloon Races (A Lab about Limiting Reactants) Purpose: Why are we doing this? Record in your lab notebook Safety: What safety precautions need to be taken? Procedure: 1. Label six large test tubes 1-6. Using a graduated cylinder, add 5 ml of 2 M acetic acid to each flask. 2. Add a few drops of Universal Indicator to each test ...

**Stoichiometry Balloon Races (A Lab about Limiting Reactants)**

If a little is good, more is better, right? Increasing the mass of a reactant in a chemical reaction may not increase the amount of product that can be formed—the yield may be “limited” based on the mole ratio of reactants. Demonstrate the concept of limiting and excess reactants by comparing the amount of carbon dioxide obtained when varying amounts of sodium bicarbonate react with a ...

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“Balloon Race” is a general experiment that expands upon the scientific principles of stoichiometric relationships, limiting and excess reactants, chemical equations and chemical reactions through the use of sodium bicarbonate and acetic acid.

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