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stoichiometry review \*\*you must show all work and your answers must include the proper number of sig figs and complete units in order to receive credit for the problem.\*\* balance the following equations to use in questions 5 - 14: ...

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### **11 1 Review And Reinforcement Answer Key**

Stoichiometry Review Assignment Answer Key Example 1: Calculate the mass of a magnesium, Mg, atoms in grams.  $24.035 \text{ g Mg} \cdot 1 \text{ mol Mg} \cdot 1 \text{ molecule Mg} = 4.04 \times 10^{-23} \text{ g/Mg atom}$   $1 \text{ mol Mg} \cdot 6.02 \times 10^{23} \text{ molecules} \cdot 1 \text{ atom Mg}$  Example 2: Calculate the number of atoms in one-millionth of a gram of magnesium, Mg.

**Stoichiometry Review Assignment Answer Key**

11—2 Review and Reinforcement Solving Stoichiometry Problems If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided. The major types of stoichiometry problems are mass-mass, mass-volume, and volume-volume.

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I really doubt that this is a "review". If this was a review, then why should you ask these questions in the first place? Shouldn't you already have the answers? If you really want an answer, then you should either ask one question at a time, or show what work you have done so far. Have a nice evening!

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368 Chapter 11 • Stoichiometry Section 11.11.1 Objectives Describe the types of relationships indicated by a balanced chemical equation. State the mole ratios from a balanced chemical equation. Review Vocabulary reactant: the starting substance in a chemical reaction New Vocabulary stoichiometry mole ratio Defining Stoichiometry

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