

## *Stoichiometry Questions And Answer*

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### **Stoichiometry Questions And Answer**

Multiple Choice Questions (MCQ) and Answers on Stoichiometry Question 1 : The weight fraction of methanol in an aqueous solution is 0.64. The mole fraction of methanol  $X_M$  satisfies  $X_M < 0.5$   $X_M = 0.5$   $0.5 < X_M < 0.64$   $X_M \geq 0.64$  Answer : 4 Question 2 : On addition of 1 c.c. of dilute hydrochloric acid (1% concentration) to 80 c.c. of a buffer solution of pH = 4, the pH of the solution becomes 1 8 ...

### **Stoichiometry Questions and Answers - Q for Questions**

Problem :  $2\text{Al} + 3\text{Cl}_2 \rightarrow 2\text{AlCl}_3$  When 80 grams of aluminum is reacted with excess chlorine gas, how many formula units of  $\text{AlCl}_3$  are produced?

### **SparkNotes: Stoichiometric Calculations: Problems**

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### **Chemistry and More - Practice Problems with Answers**

This is the chemical engineering questions and answers section on "Stoichiometry" with explanation for various interview, competitive examination and entrance test. Solved examples with detailed answer description, explanation are given and it would be easy to understand.

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Answers.com is the place to go to get the answers you need and to ask the questions you want. Go. ... Stoichiometry is a branch of chemistry which deals with the ratios of the reactants and ...

### **503 Questions Asked In Stoichiometry - Answers**

Stoichiometry Chapter Exam Instructions. Choose your answers to the questions and click 'Next' to see the next set of questions. You can skip questions if you would like and come back to them ...

### **Stoichiometry - Practice Test Questions & Chapter Exam ...**

Practice Problems: Stoichiometry. Balance the following chemical reactions: Hint a.  $\text{CO} + \text{O}_2 \rightarrow \text{CO}_2$  b.  $\text{KNO}_3 \rightarrow \text{KNO}_2 + \text{O}_2$  c.  $\text{O}_3 \rightarrow \text{O}_2$  d.  $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + \text{H}_2\text{O}$  e.  $\text{CH}_3\text{NH}_2 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{N}_2$  Hint f.  $\text{Cr}(\text{OH})_3 + \text{HClO}_4 \rightarrow \text{Cr}(\text{ClO}_4)_3 + \text{H}_2\text{O}$  Write the balanced chemical equations of each reaction:

### **Practice Problems: Stoichiometry**

Practice Problems: Stoichiometry (Answer Key) Balance the following chemical reactions: a.  $2\text{CO} + \text{O}_2 \rightarrow 2\text{CO}_2$  b.  $2\text{KNO}_3 \rightarrow 2\text{KNO}_2 + \text{O}_2$  c.  $2\text{O}_3 \rightarrow 3\text{O}_2$  d.  $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + 2\text{H}_2\text{O}$  e.  $4\text{CH}_3\text{NH}_2 + 9\text{O}_2 \rightarrow 4\text{CO}_2 + 10\text{H}_2\text{O} + 2\text{N}_2$  f.  $\text{Cr}(\text{OH})_3 + 3\text{HClO}_4 \rightarrow \text{Cr}(\text{ClO}_4)_3 + 3\text{H}_2\text{O}$  Write the balanced chemical equations of each reaction:

### **Practice Problems: Stoichiometry (Answer Key)**

1. d It might be easiest to balance the equation with mostly whole numbers:  $2\text{NH}_3 + \frac{7}{2}\text{O}_2 \rightarrow 2\text{NO}_2 + 3\text{H}_2\text{O}$ . The question asks about the amount of oxygen reacting with ONE mole of ammonia, thus cut the  $\frac{7}{2}$  (3.5) of oxygen in half to 1.75

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