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CHAPTER 6 REVIEW Chemical Bonding SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.

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REVIEW CHEMICAL BONDING SECTION 5 ANSWER KEY

5. What levels of differences in electronegativity correspond to polar covalent, nonpolar covalent, and ionic bonds? 6. If Sodium, Na, was to bond with Chlorine, Cl, what type of bond would they form? 7. If Bromine, Br was to bond with Chlorine, Cl, what type of bond would they form? 8.

CHAPTER 6 TEST: CHEMICAL BONDING REVIEW SHEET

Review Module / Chapters 5-8 31 ... Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number. ... INTRODUCTION TO CHEMICAL BONDING SECTION REVIEW 6.1

6.1 INTRODUCTION TO CHEMICAL BONDING SECTION REVIEW - LPS

Review Previous Concepts Chemical Bonding CHAPTER 6 Section 1 Introduction to Chemical Bonding What is a chemical bond and why does it form? Section 2 Covalent Bonding and Molecular Compounds What is a molecular formula? What are the characteristics of a covalent bond?

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CHAPTER 6 REVIEW Chemical Bonding SECTION 5 SHORT ANSWER Answer the following questions in the space provided. 1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms. Pairs of valence electrons repel one another.

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