

## *Section 43 Modern Atomic Theory Answers*

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Chapter 4 Atomic Structure. Section 4.3 Modern Atomic Theory. (pages 113–118) This section focuses on the arrangement and behavior of electrons in atoms. Reading Strategy (page 113) Sequencing After you read, complete the description in the flow chart below of how the gain or loss of energy affects electrons in atoms.

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4.3 Modern Atomic Theory Bishop section 43 modern atomic pdf In atomic theory and quantum mechanics, an atomic orbital is a mathematical function that describes the wave-like behavior of either one electron or a pair of electrons in an atom. Section 43 Modern Atomic Theory Answers

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Chapter 4 43 orbitals for the one electron in a hydrogen atom and to know how these orbitals can be arranged into energy levels and sublevels. Section 4.3 Multi-Electron Atoms. Goals. □ To show how the knowledge of the atomic orbitals of hydrogen can be applied to atoms of the other elements.

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Chapter 4.3 Modern Atomic Theory Vocabulary: 1. Energy Level 2. Electron Cloud 3. Orbital 4. Electron Configuration 5. Ground State Modern Atomic Theory Bohr's Model of the Atom Atoms that look like a solar system are based on a model of the atom developed by Niels Bohr, a Danish physicist who worked for a while with Rutherford. Energy Levels

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