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presented by Mapo ORGANIC CHEMISTRY ANET L. MAXWELL sapling learning For the following reaction, show the product of one mole of reagent adding across the triple bond, and then show the product of a second mole of reagent. one mole of reagent CI csecond mole of reagent Previous & Give Up & View Solution e Check Answer Next∏Exit- Hint

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be the base, because the reaction is carried out in strongly acidic solution; hydroxide cannot survive in strong acid because it protonates instantaneously to form water. Nor is hydroxide necessary as a base, because the carbocation is a very strong acid; the pK a of a carbocation b-proton is -8 to -10.

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