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Detailed solutions to these problems, many of which require the use of geochemical computer models such as MINTEQA2 (Allison et al. 1991), PHREEQC (Parkhurst 1995)'+ or SOLMINEQ.88 (Kharaka et al. 1988), are available from the publisher in a companion volume to this text, titled Solutions Manual - Aqueous Environmental Geochemistry.

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Appendix 1 DEFINITIONS AND REFERENCES FOR AQUEOUS GEOCHEMISTRY Laurie Balistrieri Acid: a substance containing hydrogen that gives free hydrogen ions (H+) when dissolved in water; e.g., hydrochloric acid (HCl), nitric acid (HNO3). Strong acids release a greater proportion of their protons than weak acids.

## **Appendix 1 DEFINITIONS AND REFERENCES FOR AQUEOUS GEOCHEMISTRY**

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