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using a device or system in a practical application. Usually, however, technology only permits a
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Electronics Section 1-1 Atomic Structure 1. An atom with an atomic number of 6 has 6 electrons
and 6 protons. 2. The third shell of an atom can have $2n^2 = 2(3)^2 = 18$ electrons. Section 1-2
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Prob. 2.7 A particle is described in 1D by a wavefunction: $\Psi = Be^{-2x}$ for $x \geq 0$ and Ce^{+4x} for $x < 0$, and B and C are real constants. Calculate B and C to make Ψ a valid wavefunction.

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