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Chloroform (CHCl₃) is produced by a reaction between methane and chlorine. Use the balanced chemical equation for this reaction to determine the mass of CH₄ needed to produce 50.0 g of CHCl₃. Balanced Equation: CH₄ + 3Cl₂ → CHCl₃ + 3HCl

Ultimate Quiz On Stoichiometry Quiz - ProProfs Quiz

Stoichiometry S Class ate FcE1ôhJ Section Quiz: Introduction to Stoichiometry In the space

provided, write the letter of the term or phrase that best completes each sentence or best answers each question. 1. Stoichiometry is the branch of chemistry that deals with elements in compounds and with reactants and products in chemical reactions,

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2. d Which of the following would not be studied within the topic of stoichiometry? (a) the mole ratio of Al to Cl in the compound aluminum chloride (b) the mass of carbon produced when a known mass of sucrose decomposes

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To solve mole-mole problems requires a balanced chemical equation and a mole ratio. Use the coefficients from the balanced equation and multiply it by the appropriate mole ratio to get an answer. This quiz will cover simple mole-mole problems. You will need a calculator. Select the best answer from ...

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