# Student Exploration Chemical Equations Answer Key

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# **Student Exploration Chemical Equations Answer**

Chemical Equations. Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass. Use for 5 minutes a day.

# Chemical Equations Gizmo: Lesson Info: ExploreLearning

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Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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#### Balancing Chemical Equations Worksheet Answer Key Gizmo ...

Question: How are chemical equations balanced? 1. Balance: Turn on Show histograms. The equation is balanced when there are equal numbers of each type of atom represented on each side of the equation. In the Gizmo, use the up and down arrows to adjust the numbers of hydrogen, oxygen, and water molecules until the equation is balanced.

#### Student Exploration: Balancing Chemical Equations

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Chemical equations gizmo - Name Arianna Noble Date Student... To set up an equation in the Chemical Equations Gizmo™, type the chemical formulas into the text boxes of the Gizmo. First, type in "H2+O2" in the Reactants box and "H2O" in the Products box. This represents the reaction of hydrogen and oxygen gas to form water.

# Chemical equations gizmo - Name Arianna Noble Date Student ...

Therefore, a balanced chemical equation will show the same number of each type of atom on each side of the equation. To set up an equation in the Chemical Equations  $Gizmo^{\mathsf{TM}}$ , type the chemical formulas into the text boxes of the Gizmo. Products box. This represents the reaction of hydrogen and oxygen gas to form water.

# **Student Exploration: Chemical Equations**

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In a chemical reaction, reactants interact to form products. This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo $^{\text{TM}}$ , look at the floating molecules below the initial reaction: H 2 + O 2 H 2 O. 1. How many atoms are in a hydrogen molecule (H 2)? \_\_\_\_ 2. How many atoms are in an oxygen molecule (O 2)? \_\_\_\_ 3.

#### Student Exploration: Balancing Chemical Equations

Balancing Chemical Equations Answer Key Vocabulary: coefficient, combination, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript Prior Knowledge Questions (Do these BEFORE using the Gizmo.) [Note: The purpose of these questions is to activate prior knowledge and get students thinking.

# BalancingChemEquationsSE Key-1 (1) - Balancing Chemical ...

This skill is important because if they are unable to count atoms in chemical formulas then they will probably not be able to balance a chemical equation. If a student does struggle with counting it tends to be on the more difficult formulas with polyatomic ions and subscripts outside of the parenthesis. These students will require extra practice.

# **Tenth grade Lesson Gizmo: Modeling Chemical Reactions**

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Student Exploration: Balancing Chemical Equations. DEFINE Vocabulary: coefficient. Combination- a chemical reaction when two reactions combine to make one. Compound- a substance consisting of two or more elements. Decomposition- the separation of a substance into a simpler form. Double replacement- multiple elements that are getting swapped

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Exploration Balancing Chemical Equations Worksheet Answer Key section. Here you will find all we have for Student Exploration Balancing Chemical Equations Worksheet Answer Key.

# Student Exploration Balancing Chemical Equations Worksheet ...

Balancing Chemical Equations Gizmo. STUDY. PLAY. Coefficient. a number that multiplies a term in an equation. Combination. a chemical reaction in which two or more reactants form a single product. Compound. a pure substance composed of two or more elements chemically combined. Decomposition.

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