

## *Solubility Curve Practice Problems Answers*

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### **Solubility Curve Practice Problems Answers**

A solution of potassium chlorate,  $\text{KClO}_3$ , has 20 grams of the salt dissolved in 100 grams of water at  $70^\circ\text{C}$ . Approximately how many more grams of the salt can be added to the solution before reaching the saturation point?

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Once microorganisms have started to grow and become physiologically active they usually produce water as an end product of respiration. Thus they increase the water activity of their own immediate environment so that eventually micro-organisms requiring a high  $a_w$  are able to grow and spoil a food which was initially considered to be microbiologically stable.

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