Stoichiometry Worksheet Mole Answer Key

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Answer Key. Stoichiometry: Mole-Mole Problems. N2 + 3H2 → 2NH3. How many moles of hydrogen are needed to completely react with 2.0 moles of nitrogen? 6.0 moles of hydrogen . 2. 2KClO3 → 2KCl + 3O2. How many moles of oxygen are produced by the decomposition of 6.0 moles of potassium chlorate? 9.0 moles of oxygen .

Stoichiometry: Mole-Mole Problems

Stoichiometry Worksheet 2 Answer Key 1. a. 2 / 13 b. 13 / 8 c. 13 / 10 d. 2 / 8 (or 1 / 4) e. 2 / 10 (or 1 / 5) 2. The KClO 3 / O 2 molar ratio is 2/3. 2 mol KClO 3 / 3 mol. O 2 ... 12.00 moles of NaClO 3 will produce how many grams of O 2? How many grams of NaCl are produced when 80.0 grams of O 2

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Compiled by Ema Gluckmann 8/92, revised 12/03, 1/06, 2/13 Page 1 STOICHIOMETRY WORKSHEET ANSWER KEY 1. One mole of nitrogen combines with one mole of oxygen according to the equation: N

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Answer Key Mole/Stoichiometry.Test.Review 1. 6.022x1023particles((atoms,(molecules)((2. 1mole(=6.022x1023particles((1mole=molar(mass(1mole=22.4L(3. Calculate(the ...

Answer Key Mole/Stoichiometry.Test.Review

Stoichiometry Worksheet and Key 1.65 mol KClO 3 mol KClO 3 mol O 2 = molO 2 3.50mol KCl ... How many moles of KCl will be formed from 2.73 moles of KClO 3? ... Microsoft Word - stoichiometry_1_worksheet_and_key.docx Created Date:

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Practice Problems: Stoichiometry (Answer Key) Balance the following chemical reactions: a. 2 CO + O 2 2 CO 2 b. 2 KNO 3 2 KNO 2 + O 2 c. 2 O 3 3 O 2 d. NH 4 NO 3 N 2 O + 2 H 2 O e. 4 CH 3 NH 2 + 9 O 2 4 CO 2 + 10 H 2 O + 2 N 2 f. Cr(OH) 3 + 3 HClO 4 Cr(ClO 4) 3 + 3 H 2 O Write the balanced chemical equations of each reaction:

Practice Problems: Stoichiometry (Answer Key)

Chem 115 POGIL Worksheet - Week 4 Moles & Stoichiometry Answers Key Questions & Exercises 1. The atomic weight of carbon is 12.0107 u, so a mole of carbon has a mass of 12.0107 g.

Moles & Stoichiometry Answers Key Questions & Exercises

Chapter 6 Balancing and Stoichiometry Worksheet and Key Topics: • Balancing Equations • Writing a chemical equation • Stoichiometry Practice: 1. In the reaction: 4Li (s) + O 2 (g) ® 2Li 2O (s) a. what is the product? b. what are the reactants? c. what does the "(s)" after the formula of lithium oxide signify?

Stoichiometry Worksheet Mole Answer Key

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