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The Wurtz reaction, named after Charles Adolphe Wurtz, is a coupling reaction in organic chemistry, organometallic chemistry and recently inorganic main group polymers, whereby two alkyl halides are reacted with sodium metal in dry ether solution to form a higher alkane: $2 R-X + 2 Na \rightarrow R-R + 2 Na + X -$. Other metals have also been used to effect the Wurtz coupling, among them silver ...

Wurtz reaction - Wikipedia

Dextrorotation and levorotation (also spelled as laevorotation) are terms used to describe the rotation of plane-polarized light. From the point of view of the observer, dextrorotation refers to clockwise rotation while levorotation refers to counterclockwise rotation. A compound that causes dextrorotation is called dextrorotatory or dextrorotary, while a compound that causes levorotation is ...

Dextrorotation and levorotation - Wikipedia

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Aim: The objective of this exercise is to prepare cyclohexene from cyclohexanol and determine the efficiency of this conversion. Experimental learning objectives: How to; set up a distillation apparatus and perform a distillation

Preparation of cyclohexene from cyclohexanol.

Aniline, phenylamine, or aminobenzene is an organic compound with the formula $C_6H_5NH_2$. It is an organic chemical compound, specifically an aryl amine, consisting of a phenyl group attached to an amino group. The chemical structure of aniline is shown at the right. It is now used mainly in the manufacture of polyurethane, although it previously was mainly used more for dyes and drugs.

Aniline - New World Encyclopedia

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Acetic acid, also known as ethanoic acid, is an organic chemical compound best recognized for giving vinegar its sour taste and pungent smell. It is one of the simplest carboxylic acids (the second-simplest, after formic acid) and has the chemical formula CH_3COOH . In its pure, water-free state, called glacial acetic acid, it is a colorless, hygroscopic liquid that freezes below $16.7^\circ C$ (62 ...

Acetic acid - New World Encyclopedia

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En chimie, les métaux sont des matériaux dont les atomes sont unis par des liaisons métalliques. Il s'agit de corps simples ou d'alliages le plus souvent durs, opaques, brillants, bons conducteurs de la chaleur et de l'électricité. Ils sont généralement malléables, c'est-à-dire qu'ils peuvent être martelés ou pressés pour leur faire changer de forme sans les fissurer ni les briser.

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