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### Chemistry - Wikipedia

Nitrous acid is useful for destroying sodium azide solutions, which are toxic and potentially explosive. It can also be used to prepare diazonium salts, which combine with anilines and phenols to form brightly colored azo compounds. The latter type of reaction can be used to produce azo-dyes, as well as serving as a qualitative test for aromatic amines.

### Nitrous acid - New World Encyclopedia

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Potassium is a chemical element with the symbol K (from Neo-Latin kalium) and atomic number 19. Potassium is a silvery-white metal that is soft enough to be cut with a knife, with little force. Potassium metal reacts rapidly with atmospheric oxygen to form flaky white potassium peroxide in only seconds of exposure. It was first isolated from potash, the ashes of plants, from which its name derives.

### **Potassium - Wikipedia**

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История. На международном съезде химиков в Карлсруэ в 1860 году были приняты определения понятий молекулы и атома. Молекула была определена как наименьшая частица химического вещества, обладающая всеми его ...

### **Молекула — Википедия**

Cato T. Laurencin, M.D., Ph.D., is a University Professor at the University of Connecticut (the 8th to be designated in the institution's over 135 year history).

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