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Chemistry - Wikipedia

The lattice energy of a crystalline solid is a measure of the energy released when ions are combined to make a compound. It is a measure of the cohesive forces that bind ions. Lattice energy is relevant to many practical properties including solubility, hardness, and volatility.

Lattice energy - Wikipedia

The bond energy in the gaseous diatomic species FF is 158.78 kJ mol⁻¹.. Fluorine: bond enthalpies in gaseous diatomic species. The following values refer to neutral heterodiatom molecules in the gas phase. These numbers may well differ considerably from, say, single bond energies in a solid.

Fluorine: properties of compounds - Periodic table

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Periodiske system - Wikipedia, den frie encyklopædi

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Complexe (chimie) — Wikipédia

Le dioxyde de silicium, ou silice, est un composé chimique de formule SiO₂. Il s'agit d'un solide incolore présent en abondance dans le milieu naturel et chez divers êtres vivants. Il existe à l'état libre sous différentes formes cristallines ou amorphes, et combiné chimiquement avec d'autres oxydes dans les silicates, qui sont les principaux constituants de l'écorce terrestre et du ...

Dioxyde de silicium — Wikipédia

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