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This turned out to be good, because my teacher was an idiot. He was wrong about stoichiometry, particularly stoichimetry involving solutions. Fortunately, I'm less of an idiot than he is, so I'm able to share with you the magic of solutions stoichiometry.

#### Solutions Stoichiometry | The Cavalcade o' Chemistry

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Answers – Naming Chemical Compounds . Name the following chemical compounds: 1) NaBr sodium bromide. 2) Ca(C 2H 3O 2) 2 calcium acetate. 3) P 2O 5 diphosphorus pentoxide. 4) Ti(SO 4) 2 titanium(IV) sulfate. 5) FePO 4 iron(III) phosphate. 6) K 3N potassium nitride. 7) SO 2 sulfur dioxide. 8) CuOH copper(I) hydroxide. 9) Zn(NO

#### **Answers - Naming Chemical Compounds**

Practice Problems: Solutions (Answer Key) What mass of solute is needed to prepare each of the following solutions? a. 1.00 L of 0.125 M K 2 SO 4 21.8 g K 2 SO 4 b. 375 mL of 0.015 M NaF 0.24 g NaF c. 500 mL of 0.350 M C 6 H 12 O 6 31.5 g C 6 H 12 O 6; Calculate the molarity of each of the following solutions:

#### **Practice Problems: Solutions (Answer Key)**

For chemistry help, visit www.chemfiesta.com © 2000 Cavalcade Publishing, All Rights Reserved 7) How many liters of a 0.88 M solution can be made with 25.5 grams of

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#### Adel Solutions LLC. - Our Clients

If you're anything like me (and pray that you aren't), one of your favorite things in the whole world is to name chemical compounds. I just sit and name compounds all day long, happy in the knowledge that one day the world will need a compound naming guru to save our species. With these worksheets,...

# Naming Worksheets | The Cavalcade o' Chemistry

Practice Problems: Solutions (Answer Key) 1. ... Calculate the molality of each of the following solutions: a. 2.89 g of NaCl dissolved in 0.159 L of water (density of water is 1.00 g/mL) 0.311 molal NaCl b. 1.80 mol KCl in 16.0 mol of H2O 6.25 molal KCl c. 13.0 g benzene, C6H6 in 17.0 g ...

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