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(2) Mass from amount : The key mathematical equation needed here is - mass (g) = relative formula mass (g mol⁻¹) x amount (mol). Using the triangular relationship from above if the mass section is covered over then the amount multiplied by the relative formula mass gives the mass -

GCSE Chemistry, Year 10, Amount of Substance page

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About This Edition. v. About This Edition. This edition of the . AP Chemistry Course and Exam Description. includes the following changes, which take effect in fall 2014:

AP Chemistry Course and Exam Description - College Board

§112.31. Implementation of Texas Essential Knowledge and Skills for Science, High School. (a) The provisions of this subchapter shall be implemented by school districts.

19 TAC Chapter 112, Subchapter C - Texas Education Agency

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Equilibrium Constant Expressions. Reactions don't stop when they come to equilibrium. But the forward and reverse reactions are in balance at equilibrium, so there is no net change in the concentrations of the reactants or products, and the reaction appears to stop on the macroscopic scale.

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