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Solutions & Their Behavior. Solution concentration can often be expressed in units of molarity, molality, weight percent, or as ppm. Polar solutions are miscible with other polar solutions, and nonpolar solutions are miscible with other nonpolar solutions. Polar and nonpolar solutions do not mix and form heterogeneous mixtures.

16. [Solutions & Their Behavior] | General ... - Educator

Solutions and . Their Behavior. By. Dr. Andrea Wallace. CGCC . Edited by. John Taylor. FCCJ-North Campus . Chapter 14 : Solutions and Their Behavior . 14.1 Units of Concentration, p. 656 . A solution consists of a _____ and a _____. Examples: Concentration is a way to express the amount of solute present. Equations:

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