

Science Section 2 Answer States Of Matter

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SCIENCE SECTION 2 ANSWER STATES OF MATTER

Chapter 2 States of Matter SECTION 1 THREE STATES OF MATTER 1. the physical forms of a substance 2. They move about the most in the gas state and the least in the solid state. 3. They vibrate. 4. They move past each other. 5. a force that acts on the particles at the sur-face of a liquid 6. water, cream, syrup 7. There is more space between particles. Review 1.

CHAPTER States of Matter SECTION 2 Behavior of Gases

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An alpha particle is the same as a helium-4 nucleus. After releasing an alpha particle, the nucleus' neutron number N and atomic number Z have both decreased by 2, and its mass number A has decreased by 4. This is represented with a reaction "equation" analogous to that for a chemical reaction: $A-4 \quad Z \quad X \rightarrow Z-2 + 2 \text{ He}$.

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2. freezing a. Solid to gas 3. sublimation b. Liquid to gas 4. condensation c. Gas to solid 5. melting d. Liquid to solid 6. deposition e. Gas to liquid 7. vaporization f. Solid to liquid 8. What happens to the temperature of a substance during a phase change? 9. Is the following sentence true or false? The temperature at which a substance

Chapter 3 States of Matter Section 3.3 Phase Changes

Interactive Reader and Study Guide 62 States of Matter SECTION 2 Name Class Date Changes of State continued What Is Melting? When energy is added to a solid, it can melt. Melting is the change of state from a solid to a liquid. For exam-ple, an ice cube in a glass of lemonade melts as it absorbs heat from the lemonade.

CHAPTER 4 States of Matter SECTION 2 Changes of State

STATES OF MATTER71 Figure 2 Common States of Matter Here, the element sodium is shown as the solid metal. A This is sodium melted as a liquid. B Sodium exists as a gas in a sodium-vapor lamp. C Figure 4 The particles in an amorphous solid do not have an orderly arrangement. Figure 3 Particles in a crystalline solid have an orderly arrangement.

CHAPTER 3 States of Matter - Grygla Public School

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Section Quizzes and Chapter Tests - Glencoe

States of Matter SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. a Liquids possess all the following properties except (a) relatively low density. (c) relative incompressibility. (b) the ability to diffuse. (d) the ability to change to a gas. 2. a. Chemists distinguish between intermolecular and intramolecular forces.

10 States of Matter - Website

prior knowledge. There is a transparency for each section, or lesson, in the Student Edition. The reproducible student masters are located in the Transparency Activitiessection. The teacher

material, located in the Teacher Guide and Answers section, includes Transparency Teaching Tips, a Content Background section, and Answers for each transparency.

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2. 3. Students explain how asking and answering questions helps them remember what they have learned. Section 2 Before You Read (p. 9) Students' responses will vary, but students should indicate how they would get the information they need from their science book. For example, they may be able to go to a friend's house to look at the ...

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5 6 7 082 11 10 09 08 07 CHAPTER 1 The World of Physical Science SECTION 2 The Electromagnetic Spectrum. After you read this section, you should be able to answer Section 1 Review. Holt Science Spectrum Physical Science Chapter 5 Review Answers Holt Science Spectrum You will learn that waves

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2. d, i SECTION: FLUIDS 1. less; The buoyant force cannot equal the gold's weight, so the gold sinks. 2. greater; The buoyant force exceeds the balloon's weight, so the balloon rises. 3. equal; The weight of the boat is balanced by the buoyant force. 4. equal; The weight of the submarine is balanced by the buoyant force at the

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