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When a nuclear undergoes nuclear decay by gamma rays, the atomic number of the element a. remains the same b. decreases by one d. increases by one d. increases by two

Chapter 10: Nuclear Changes (TEST ANSWERS) Flashcards ...

Chapter 10: Nuclear Changes. a positively charged particle that consists of two protons and two neutrons and that is emitted from a nucleus during radioactive decay; it is identical to the nucleus of a helium atom and has a charge of +2.

Chapter 10: Nuclear Changes Flashcards | Quizlet

Nuclear decay causes changes in the nucleus of an atom. When an unstable nucleus releases an alpha or beta particle, the number of protons and neutrons changes. For instance, when radium-226 emits an alpha particle, it changes to radon-222. Nuclear decay changed the number of protons, so the atom becomes a different element. Nuclear-decay equations are written in a way similar to

CHAPTER 10 SECTION 1 What Is Radioactivity?

Interactive Reader 14 Answer Key Science Spectrum Answer Keycontinued 3. Strong acids and bases form strong electrolytes in water. Weak acids and bases form weak electrolytes in water. 4. acidic 5. soft drink, vinegar, stomach acid 6. Graph of acidity and pH: x-axis, "pH"; y-axis, "Acidity." Line should start from top left

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Interactive Reader 227 Nuclear Changes SECTION 3 Name Class Date Nuclear Radiation Today continued STORING NUCLEAR WASTES Nuclear wastes, such as the fuel rods that were used in the reactor, cannot simply be thrown away. Radioactive wastes must be stored safely for thousands of years until they have decayed to less harmful elements. Storage facil-

CHAPTER 10 SECTION 13 Nuclear Radiation Today

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Concept Review: Energy. 1. energy 2. physical 3. chemical 4. endothermic 5. exothermic 6. kinetic 7. transferred 8. In any chemical or physical change, the total quantity of energy remains constant. Energy cannot be created or destroyed. 9. Heat is the energy transferred between objects that are at different temperatures.

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