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Concentration of an acid depends on the ratio between the solute (acidic compound) and the solvent (water). The greater the amount of solute, the higher the concentration. Molarity is the measure of concentration of acids and bases. How can you use a solubility curve to tell if a solution is saturated, dilute, or supersaturated?

### **Solutions, Acids, Bases Study Guide**

14.2Solution Concentration. In your textbook, read about expressing concentration and using percent to describe concentration. Data related to aqueous solutions of sodium chloride (NaCl) and aqueous solutions of ethanol (C<sub>2</sub>H<sub>5</sub>OH) are provided in the table below. Use the table to answer the following questions.

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