

Solutions Introduction Key

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3. Solutions are said to be in a ____ phase even though the components may have been in different phases before the solution was formed. 4. Pure gold is 24 carat. 14-carat gold contains 14 parts gold and 10 parts other metals. 14-carat gold is said to be a(n) ____, which is a type of ____ solution.

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Worksheet: Solutions Introduction Name

ANSWER KEY Worksheet – Introduction to Solutions Part I- The Basics: 1. Define the following terms: a. Solute – substance being dissolved (lower concentration) b. Solvent – substance doing the dissolving (higher concentration) c. Solution – homogenous mixture of 2 or more substances (solute + solvent) d.

Worksheet Introduction to Solutions

Introduction to Solutions Most chemical reactions occur in solutions . This is because a substance dissolved in a solvent , the solute , will be in its smallest state of subdivision, existing as individual molecules or ions that will increase their ability to react with other molecules or ions.

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answers from students. such as such as Matter Energy State Physical changes in forms such as given off absorbed always include changes in

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Chapter 7: "Solutions" Worksheet and Key Define the following terms: mixture aqueous solutions colloids suspensions heterogeneous homogenous solvent solute saturated solution solubility molarity molality Henry's law Osmosis Osmotic pressure Osmolarity Osmolality Diffusion SEE LECTURE NOTES FOR DEFINITION ANSWERS 1.

Chapter 7: "Solutions" Worksheet and Key

SOLUTIONS Date ____ Period ____ Solutions: An Introduction Aim • To explain why substances dissolve Notes Definition: Solution = homogeneous mixture n Nature of mixtures q consists of two or more kinds of matter q each substance in a mixture retains its own properties p sugar and water - sweet and wet p brine (salt water) - salty liquid

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New York State Department of Transportation Request for ...

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