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Review Sheet: Unit 6 Name_____ CHEMISTRY: A Study of Matter © 2004, GPB 6.38a KEY I. Fill in the blanks with the most appropriate term:

Review Sheet: Unit 6 Name KEY - Mr Kiddey's Classroom

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'93013 MO

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REVIEW SHEET UNIT 6 CHEMISTRY ANSWERS

Review Sheet: Unit 6 Name_____ CHEMISTRY: A Study of Matter © 2004, GPB 6.34 I. Fill in the blanks with the most appropriate term:

Review Sheet: Unit 6 Name - Georgia Public Broadcasting

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Unit 6 Review Sheet KEY - Course Hero

Review Sheet: Unit 6 Name_____ KEY I. Fill in the blanks with the most appropriate term: Common names of substances like "milk of magnesia" or "lime" usually give no information about the chemical composition of a compound. Consequently, chemists rely on a chemical _____ formula when representing a chemical compound.

6-38a,b,c,d,e-Review%20Sheet-Key - Yumpu

Review Sheet: Unit 5 Name_____ CHEMISTRY: A Study of Matter © 2004, GPB 5.24a KEY I. For each of the following statements, write "I" for ionic, "C" for covalent

CHEMISTRY: A Study of Matter - Thomas County Schools

Review Sheet: Unit 2 Name_____ CHEMISTRY: A Study of Matter © 2004, GPB 2.20a KEY I. Fill in the blanks _____ properties can be observed without chemically changing ...

CHEMISTRY: A Study of Matter - Thomas County Schools

CHEMISTRY: A Study of Matter © 2004, GPB 1.28 6. We have learned that numbers without units are meaningless. The unit for volume in the metric system is the _____.

Review Sheet: Unit 1 Name - Georgia Public Broadcasting

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Review Sheet: unit 10 Fill in the blanks using the most appropriate word or phrase. 10. For most solutes to be dissolved in liquid solvents: as temperature increases the rate of solution as surface area increases, the rate of solution stirring or agitating the mixture S the rate of solution.

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Semester 1 Review Sheet Unit 1: Matter and Measurement Topics: classification of matter, properties of matter, units of measure, uncertainty, dimensional analysis. 1) When 10.0g of marble chips (calcium carbonate) is treated with 50.0 ml. of hydrochloric acid (density = 1.096 g/mL), the marble dissolves, giving a solution

www.erhsnyc.org

Chemistry 20 – Chapter 3,4 Review Worksheet page 3 10. Do the following calculations. (Don't forget working, units, significant digits.) a) Calculate the number of moles of methane gas in 6.0 L at SATP.

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