

Section 92 Substituted Hydrocarbons Answers

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Section PG Go. Assessment. 1. The functional groups in an alcohol, organic acid and an organic base are hydroxyl group(-OH), carboxyl group(-COOH) & an amino group(-NH₂) respectively. ... A substituted hydrocarbon is an atom or group of atoms substituted in place of a hydrogen atom on the parent chain of a hydrocarbon. 1 answ. 4. When a ...

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9.2 Substituted Hydrocarbons • A hydrocarbon in which one or more hydrogen atoms have been replaced by an atom or group of atoms is a substituted hydrocarbon. • The substituted atom or group of atoms is called a functional group. The functional group in an alcohol is a hydroxyl group, -OH.

Chapter 9 Carbon Chemistry - Amazon S3

Substituted Hydrocarbons and Their Reactions Substituted Hydrocarbons and Their Reactions Section 23.1 Functional Groups In your textbook, read about functional groups. Circle the letter of the choice that best completes the statement or answers the question. 1. In hydrocarbons, carbon atoms are generally linked to

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786 Chapter 22 • Substituted Hydrocarbons and Their Reactions Section 22.12.1 Figure 22.1 All of these items contain at least one of the functional groups that you will study in this chapter. For example, the fruit and flowers have sweet-smelling aromas that are due to ester molecules. Objectives Define functional group, and give examples.

Chapter 22: Substituted Hydrocarbons and Their Reactions

The pH of rain containing sulfuric acid and nitric acid can be as low as 2.7. Quick Lab: Comparing Isomers p. 265 Section Assessment Questions 1-1-8 p. 269 9.2 Substituted Hydrocarbons Organic compounds containing chlorine or other halogens are halocarbons.

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Section 9.2 Substituted Hydrocarbons (pages 272–274) This section discusses organic compounds

that contain atoms of elements other than carbon and hydrogen. ... Carbon and Hydrocarbons
SECTION 20-4 SHORT ANSWER Answer the following questions in the space provided. 1.
Hydrocarbons that contain double bonds are referred to as .

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synthetic rubber,using hydrocarbons from petroleum.Natural rubber and synthetic rubbers contain different monomers and have different properties. The tires in Figure 14A will resist wear and be less likely to leak if they are made of synthetic rubber. Rubber is used as an adhesive.The How It Works box on page 277 explains how adhesives work.

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