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Chemistry--Unit 6: States of Matter Practice Problems 1. Convert the following Celsius temperatures to Kelvin. a. $450 \,^{\circ}\text{C}$ b. $-117 \,^{\circ}\text{C}$ c. $-12 \,^{\circ}\text{C}$ d. $25 \,^{\circ}\text{C}$ 2. Convert the following Kelvin temperatures to Celsius. a. $450 \,^{\circ}\text{K}$ b. $2 \,^{\circ}\text{K}$ c. $373 \,^{\circ}\text{K}$ d. $298 \,^{\circ}\text{K}$ 3. Convert 566 mmHg to atm. To kPa. 4. Convert 0.85 atm to mmHg. To kPa. 5. Convert 112.5 kPa to atm. To mmHg.

Chemistry--Chapter 10: States of Matter

www.njctl.org AP Chemistry States of Matter 6. Examine the phase diagram below and answer the following questions: a. Sublimation and deposition b. liquid c. 373 K d. 340 K e. Condensation and freezing 7. a. 12,000 Joules b. 5.71 C 8. a. Br2 is a covalent molecular solid held by intermolecular forces

PSI AP Chemistry: States of Matter Name

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all matter consists of tiny particles that are in constant motion; Gas particles are small spheres with an insignificant volume; motion is rapid, random and constant; collisions between particles are elastic. Inelastic collision. loses kinetic energy with each collison, slows down. Gas pressure.

Chemistry Chapter 13 Test: States of Matter Flashcards ...

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Unit 3: States of Matter Practice Exam Multiple Choice. Identify the choice that best completes the statement or answers the question. ____ 1. Two gases with unequal masses are injected into opposite ends of a long tube at the same time and allowed to diffuse toward the center. They should begin to mix a. in approximately five minutes.

Chem Unit 3 Practice Exam - Annville-Cleona School District

A) change 1 g of the substance from the solid to the liquid state. B) raise the temperature of 1 g of the substance by 1 °C. C) change 1 g of the substance from the liquid to the solid state.

Energy and Matter Practice Problems - Utah State University

CHAPTER 13 PRACTICE TEST Matching: Match the lettered elements with the numbered elements. Answers may be used more than once. Only one answer may be used for each question. Place all answers on answer sheet provided. (2 point each = 30) a. decrease(s) f. plasma k. vapor pressure b. increase(s) g. evaporation l. thermometer c. no h. phase change m.

Chapter 13 Practice Test 2014 + Answers - CHAPTER 13 ...

States of matter follow-up (Opens a modal) Specific heat and latent heat of fusion and vaporization (Opens a modal) Specific heat, heat of fusion and vaporization example (Opens a modal) Chilling water problem (Opens a modal) Change of state example (Opens a modal) Vapor pressure (Opens a modal) ... Practice. Molarity calculations. 4 questions ...

States of matter and intermolecular forces - Khan Academy

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