

Paul S. J. Verburg

## Soil solution and extractable soil nitrogen response to climate change in two boreal forest ecosystems

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**Abstract** Several studies show that increases in soil temperature result in higher N mineralization rates in soils. It is, however, unclear if additional N is taken up by the vegetation or accumulates in the soil. To address this question two small, forested catchments in southern Norway were experimentally manipulated by increasing air temperature (+3°C in summer to +5°C in winter) and CO<sub>2</sub> concentrations (+200 ppmv) in one catchment (CO<sub>2</sub>T-T) and soil temperature (+3°C in summer to +5°C in winter) using heating cables in a second catchment (T-T). During the first treatment year, the climate treatments caused significant increases in soil extractable NH<sub>4</sub> under *Vaccinium* in CO<sub>2</sub>T-T. In the second treatment year extractable NH<sub>4</sub> in CO<sub>2</sub>T-T and NO<sub>3</sub> in T-T significantly increased. Soil solution NH<sub>4</sub> concentrations did not follow patterns in extractable NH<sub>4</sub> but changes in soil NO<sub>3</sub> pools were reflected by changes in dissolved NO<sub>3</sub>. The anomalous behavior of soil solution NH<sub>4</sub> compared to NO<sub>3</sub> was most likely due to the higher NH<sub>4</sub> adsorption capacity of the soil. The data from this study showed that after 2 years of treatment soil inorganic N pools increased indicating that increases in mineralization, as observed in previous studies, exceeded plant demand and leaching losses.

**Keywords** Climate change · Soil extracts · Soil solution · Nitrogen

P. S. J. Verburg  
Laboratory of Soil Science and Geology,  
Department of Environmental Sciences,  
Wageningen Agricultural University,  
PO Box 37, 6700 AA Wageningen, The Netherlands

P. S. J. Verburg (✉)  
Division of Earth and Ecosystem Sciences,  
Desert Research Institute,  
2215 Raggio Parkway, Reno, NV, 89512, USA  
e-mail: paul.verburg@dri.edu  
Tel.: +1-775-6737425  
Fax: +1-775-6737485

### Introduction

Increased emissions of CO<sub>2</sub> and other greenhouse gases due to combustion of fossil fuels and changes in land use may contribute to increases in global temperatures (IPCC 2001). An increase in atmospheric temperature may stimulate decomposition of soil organic matter and nutrient mineralization in terrestrial ecosystems (Kirschbaum 1995; MacDonald et al. 1995). A meta-analysis of data from a wide range of soil/ecosystem warming and gradient studies showed that warming led to significant overall increases in N mineralization, nitrification and litter decomposition (Rustad et al. 2001). Despite these general patterns, effects of elevated temperature on soil nutrient pools are less clear. Few data are available addressing the effects of warming on soil extractable N. Peterjohn et al. (1994) measured increases in soil inorganic N pools in response to warming. Shaw and Harte (2001) measured no increase despite increases in N mineralization rates, which was ascribed to increased plant uptake. Effects of warming on soil solution concentrations of dissolved inorganic N (DIN; NH<sub>4</sub> and NO<sub>3</sub>) have been mixed as well with studies reporting increases (Joslin and Wolfe 1993; McHale and Mitchell 1996), decreases (Ineson et al. 1998) or no change (Peterjohn et al. 1994) in soil solution DIN concentrations upon soil heating. Rustad et al. (2000) hypothesized that soil solution DIN concentrations will increase only in N-saturated ecosystems, whereas most N will be taken up by vegetation in N-limited ecosystems.

To study potential impacts of climate change on nutrient cycling in a boreal ecosystem, the Climate Change Experiment (CLIMEX) was initiated in southern Norway (Van Breemen et al. 1998). Previous studies have shown that mineralization rates increased in response to climate manipulations (Verburg et al. 1999; Verburg and Van Breemen 2000). This paper presents 3 years of data (one pre-treatment year and two treatment years) describing the effects of ecosystem warming on extractable soil inorganic N pools and soil solution DIN concentrations.

## Materials and methods

The CLIMEX site is located at Risdalsheia ( $58^{\circ}23'N$ ,  $8^{\circ}19'E$ ) in southern Norway. The site is 300 m above sea level on a large biotite granite plateau representative of large upland areas in southern Norway. Mean annual precipitation is 1,400 mm, and mean annual temperature is  $5^{\circ}C$  ( $-3^{\circ}C$  in January and  $+16^{\circ}C$  in July). Depressions in the granite surface are filled with post-glacial soil material in which acidic, peaty podsolic soils have developed. Mean soil depth is 20 cm with a maximum of 70 cm. Between 30% and 50% of the bedrock is exposed. Vegetation is dominated by shrubs—*Calluna vulgaris* (L.) Hull, *Vaccinium myrtillus* L., *V. uliginosum* L. and *V. vitis-idaea* L. Overstory trees include Scots pine (*Pinus sylvestris* L.) and birch (*Betula pubescens* Ehrh.).

In 1983, two small headwater catchments were covered with transparent roofs as part of the Reversing Acidification in Norway (RAIN) project (Wright et al. 1993). Precipitation collected from the roof of one catchment was treated to remove all acidifying components. In 1993, this catchment was completely enclosed with transparent walls and the greenhouse was separated into two parts using another transparent wall. Starting in June 1994, 80% of the greenhouse ( $CO_2T-T$ ) was manipulated by increasing  $CO_2$  concentrations by 200 ppmv during the growing season and increasing air temperature by  $5^{\circ}C$  in January and  $3^{\circ}C$  in July with intermediate temperature increases in the intervening months. The average annual soil temperature increase was  $2.6^{\circ}C$  at 5-cm and  $1.9^{\circ}C$  at 30-cm depth. The upper 20% of the greenhouse ( $CO_2T-C$ ) served as an untreated control.

The second covered catchment received ambient acid rain. Starting in April 1994, soil temperature was increased by  $5^{\circ}C$  in January and  $3^{\circ}C$  in July with intermediate temperature increases in the intervening months using heating cables in the lower 80% of the catchment ( $T-T$ ; Lükelwille and Wright 1997). Average annual soil temperature increases were  $3.4^{\circ}C$  at 5 cm and  $2.5^{\circ}C$  at 30 cm. The upper 20% of the catchment ( $T-C$ ) served as an untreated control. A third, uncovered catchment served as outside control (Out-C). An overview of the experimental treatments is given in Table 1.

In all treated and control areas ten subplots of  $0.1\text{ m}^2$  were established both in *Calluna*- and *Vaccinium*-dominated areas. In  $T-C$  and  $T-T$ , no *Vaccinium* was present. Soils were sampled in each subplot four times between

**Table 1** Experimental treatments in the studied catchments

Catchment	Area ( $\text{m}^2$ )	Enclosure	N deposition	Climate treatment
$CO_2T-C$	170	Roof	Low	0
$CO_2T-T$	690	Roof	Low	$+4^{\circ}C$ (air), $+200$ ppmv $CO_2$
$T-C$	80	Roof	High	0
$T-T$	320	Roof	High	$+4^{\circ}C$ (soil)
Out-C	650	No roof	High	0

April and October during 1993 (pre-treatment year), 1994 and 1995 (treatment years). Winter sampling was impossible due to snow cover and limited site access. Soil cores (2.8-cm diameter) were taken to a depth of 10 cm unless bedrock was shallower. Samples were weighed and stored overnight at  $4^{\circ}C$ . The following day, 20 g field moist soil was extracted by shaking for 1 h with 50 ml 1 M KCl. The KCl extracts were filtered (Whatman GF C) and analyzed colorimetrically for  $\text{NH}_4\text{-N}$  and  $\text{NO}_3\text{-N}$  on a Technicon auto analyzer. A subsample of the soil was dried at  $105^{\circ}C$  for 24 h to determine soil moisture content.

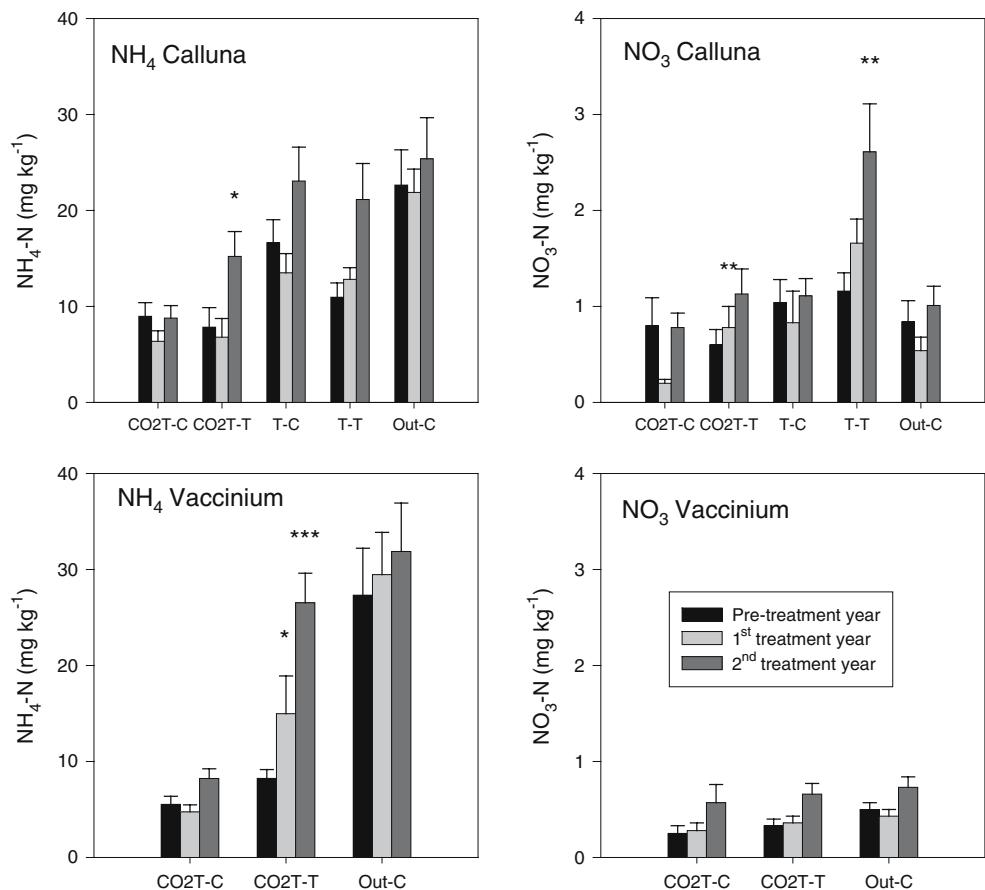
To measure treatment effects on soil solution DIN concentrations, Rhizon mini-lysimeters (Eijkelkamp Agri-search, The Netherlands) were installed in two locations in  $CO_2T-C$  and  $T-C$  and three locations in  $CO_2T-T$ ,  $T-T$  and Out-C in October 1992. Vegetation was dominated by *Vaccinium* (80% of the vegetated surface area; soils under *Calluna* were too shallow to install lysimeters) in Out-C,  $CO_2T-C$  and  $CO_2T-T$ . Subplots in  $T-C$  and  $T-T$  were dominated by *Calluna*. The lysimeters were installed horizontally in the walls of a small pit ( $30 \times 30 \times 40\text{ cm}$ ,  $L \times W \times D$ ) at 5-, 15- and 25-cm depths unless soils were too shallow. After installation, the pits were backfilled with the same soil. Soil water was collected overnight using pre-evacuated (50 kPa), 100-ml glass bottles connected to the lysimeters. Volume-weighted samples were pooled by depth for each pit. Soil solution was sampled within 2 days of the soil sampling dates. The samples were stored at  $4^{\circ}C$  prior to analysis. Samples were analyzed colorimetrically for  $\text{NH}_4$  and by ion chromatography for  $\text{NO}_3$  and Cl. Soil solution Cl concentrations were used as a proxy for soil moisture assuming that Cl behaves as a conservative ion (e.g. Johnson et al. 2001). Under this assumption, a decrease in soil moisture would be reflected by an increase in soil solution Cl concentration.

Treatment effects on average annual soil extractable  $\text{NH}_4$  and  $\text{NO}_3$  concentrations for *Calluna* and *Vaccinium* as well as soil solution concentrations were analyzed by analysis of co-variance (ANCOVA) for both manipulated catchments using control vs treated plots (“treatment”) as main factor. Data from the pre-treatment year were used as co-variable to account for pre-treatment differences between control and treated plots. All soil solution data were log transformed prior to the ANCOVA analysis. To include element concentrations below the detection limit (reported as “zero”),  $1\text{ }\mu\text{Eq l}^{-1}$  was added to all values before the log transformation. All statistical analyses were carried out using DataDesk version 6.0 and treatment effects were considered significant if  $P < 0.05$ .

## Results and discussion

The effects of the climate manipulations on soil extractable N varied by nutrient, vegetation, and plot. During the first treatment year the treatments did not affect soil  $\text{NH}_4$  concentrations under *Calluna* in  $CO_2T-T$  and  $T-T$  (Fig. 1). During the second treatment year soil  $\text{NH}_4$  concentrations increased by 100% in  $CO_2T-T$  compared to the pre-

**Fig. 1** Average annual extractable soil  $\text{NH}_4$  and  $\text{NO}_3$  concentrations under *Calluna* and *Vaccinium* in control ( $\text{CO}_2\text{T-C}$ ,  $\text{T-C}$ ,  $\text{Out-C}$ ) and heated ( $\text{CO}_2\text{T-T}$ ,  $\text{T-T}$ ) parts of the studied catchments during pre-treatment and treatment years (see Table 1 for catchment details). Error bars represent standard errors of the mean. The asterisks indicate significant treatment effects as determined by ANCOVA using pre-treatment data as co-variable (\*  $P < 0.05$ , \*\*  $P < 0.01$ , \*\*\*  $P < 0.001$ )



treatment year while concentrations remained constant in  $\text{CO}_2\text{T-C}$  and  $\text{Out-C}$ . Soil  $\text{NH}_4$  concentrations doubled in  $\text{T-T}$  in the second treatment year as well but a similar increase was observed in  $\text{T-C}$  indicating that the increase in  $\text{T-T}$  reflected interannual variability in ambient conditions. During the first treatment year soil  $\text{NO}_3$  concentrations under *Calluna* remained constant in  $\text{CO}_2\text{T-T}$  but decreased by 75% in  $\text{CO}_2\text{T-C}$ . Soil  $\text{NO}_3$  concentrations also tended to decrease in  $\text{Out-C}$  ( $P = 0.052$ ) suggesting that the treatment effect in  $\text{CO}_2\text{T-T}$  was caused by the climate manipulations. No treatment effect was observed during the second treatment year in  $\text{CO}_2\text{T-T}$ . Soil warming caused an increase in soil  $\text{NO}_3$  concentrations by 110% in  $\text{T-T}$  during the second treatment year while concentrations remained unchanged in  $\text{T-C}$  and  $\text{Out-C}$ . Soil  $\text{NH}_4$  concentrations under *Vaccinium* increased by 80% in  $\text{CO}_2\text{T-T}$  during the first treatment year and by 200% during the second treatment year but remained constant in  $\text{CO}_2\text{T-C}$  and  $\text{Out-C}$  in both treatment years. Soil  $\text{NO}_3$  concentrations under *Vaccinium* in  $\text{CO}_2\text{T-T}$  doubled during the second treatment year but similar increases were found in the control plots  $\text{CO}_2\text{T-C}$  and  $\text{Out-C}$  indicating that the increase in  $\text{CO}_2\text{T-T}$  was due to interannual variability. Prior to the start of the treatments  $\text{NH}_4$  under *Calluna* and *Vaccinium* and  $\text{NO}_3$  under *Vaccinium* were significantly lower ( $\text{NH}_4$ ,  $P < 0.001$ ;  $\text{NO}_3$ ,  $P = 0.02$ ) in the reduced-N plots ( $\text{CO}_2\text{T-C}$  and  $\text{CO}_2\text{T-T}$ ) than in the plots receiving ambient N inputs ( $\text{T-C}$ ,  $\text{T-T}$ ,  $\text{Out-C}$ ). The 2 years of climate

manipulation in  $\text{CO}_2\text{T-T}$  caused soil  $\text{NH}_4$  and  $\text{NO}_3$  concentrations to become similar to those in the ambient deposition plots.

The effects of the climate treatments on extractable N concentrations were consistent with patterns in net mineralization as measured in previous studies. The increase in net mineralization in  $\text{CO}_2\text{T-T}$  was mainly caused by an increase in ammonification while in  $\text{T-T}$  only nitrification increased (Verburg et al. 1999; Verburg and Van Breemen 2000). These studies also showed that contribution of nitrification to the total mineralization was higher for *Calluna* (10%) than for *Vaccinium* (2%). The 2 years of climate manipulations caused an increase in vegetation N uptake in  $\text{CO}_2\text{T-T}$  (Van Breemen et al. 1998) while DIN leaching increased by approximately  $0.07 \text{ g m}^{-2} \text{ year}^{-1}$  in both catchments (Lükewille and Wright 1997; Wright 1998). Still, soil extractable N pools increased showing that the increases in mineralization exceeded vegetation demand and leaching. The increase in inorganic soil N pools represented a significant N flux. For instance, assuming a bulk density of  $0.67 \text{ g cm}^{-3}$  for topsoils at the site (Lotse and Ottabong 1985), soil  $\text{NH}_4$  increased from 0.6 to  $1.0 \text{ g N m}^{-2}$  under *Vaccinium* in  $\text{CO}_2\text{T-T}$  during the first treatment year. In this period net ammonification in these plots increased from  $4.1$  to  $5.7 \text{ g N m}^{-2} \text{ year}^{-1}$  (Verburg and Van Breemen 2000) suggesting that 25% of the additionally mineralized N was transferred to the extractable inorganic N pool.

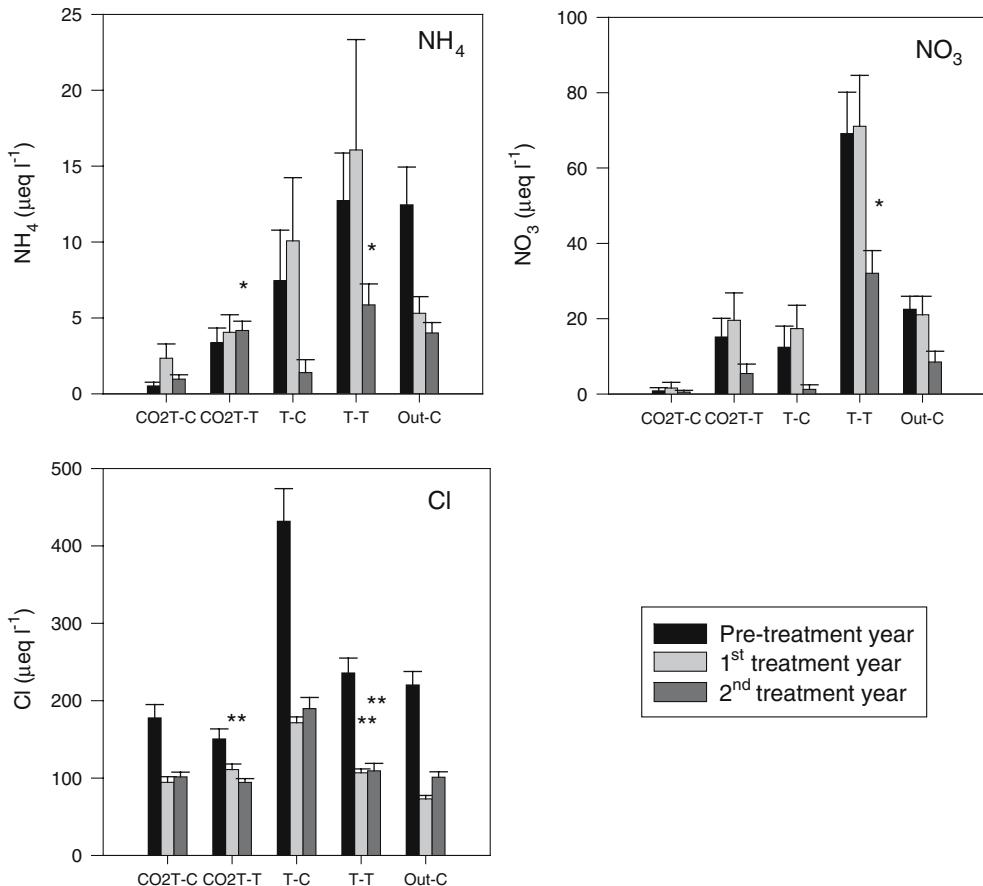
During the first treatment year, soil solution  $\text{NH}_4$  and  $\text{NO}_3$  concentrations did not show any significant treatment effects in  $\text{CO}_2\text{T-T}$  and  $\text{T-T}$  even though soil extractable  $\text{NH}_4$  under *Vaccinium* and  $\text{NO}_3$  under *Calluna* significantly increased in  $\text{CO}_2\text{T-T}$  (Fig. 2). During the second treatment year, average annual soil solution  $\text{NH}_4$  concentration increased by 86% in  $\text{CO}_2\text{T-C}$  but only by 24% in  $\text{CO}_2\text{T-T}$  compared to the pre-treatment year indicating that soil solution  $\text{NH}_4$  concentrations decreased in response to the treatments. In  $\text{T-T}$  soil solution  $\text{NH}_4$  and  $\text{NO}_3$  concentrations increased during the second treatment year. Soil solution  $\text{NH}_4$  concentration decreased by 81% in  $\text{T-C}$  and by 54% in  $\text{T-T}$  and  $\text{NO}_3$  concentration decreased by 90% in  $\text{T-C}$  and by 54% in  $\text{T-T}$  compared to the pre-treatment year suggesting a relative increase in DIN concentrations in  $\text{T-T}$ .

The patterns in soil solution DIN concentrations were not caused by changes in soil moisture. During the first treatment year the average annual Cl concentration showed a significantly smaller decrease in  $\text{CO}_2\text{T-T}$  (26%) than in  $\text{CO}_2\text{T-C}$  (47%; Fig. 2) compared to the pre-treatment year while soil warming caused a slight but significantly smaller decrease in Cl concentration in  $\text{T-T}$  (54%) than in  $\text{T-C}$  (58%) in both treatment years. Data from both catchments suggest a reduction in soil moisture in the treated plots ( $\text{CO}_2\text{T-T}$  and  $\text{T-T}$ ) but the changes in soil moisture based on Cl data were too small to explain the changes in DIN concentrations. Neither in  $\text{CO}_2\text{T-T}$  nor in  $\text{T-T}$  did soil solution

$\text{NH}_4$  concentrations reflect patterns in soil extractable  $\text{NH}_4$ ; in  $\text{CO}_2\text{T-T}$   $\text{NH}_4$  concentrations decreased despite an increase in extractable  $\text{NH}_4$  while in  $\text{T-T}$  soil solution  $\text{NH}_4$  increased even though extractable  $\text{NH}_4$  did not change. The patterns found in  $\text{CO}_2\text{T-T}$  were similar to those observed in Harvard forest (Peterjohn et al. 1994). Behavior of dissolved  $\text{NH}_4$  is more complicated than that of  $\text{NO}_3$  since  $\text{NH}_4$  more easily adsorbs to the soil than  $\text{NO}_3$ ; soil solution  $\text{NO}_3$  concentrations were four to five times higher than  $\text{NH}_4$  concentrations even though soil extractable  $\text{NO}_3$  concentrations were ten times lower. Consequently, dissolved  $\text{NH}_4$  concentrations will not only depend on extractable soil  $\text{NH}_4$  pool size but also on soil chemical parameters including pH and cation exchange capacity. As a result, dissolved  $\text{NH}_4$  concentrations are probably less sensitive to changes in extractable pool size than  $\text{NO}_3$ . Soil solution and soil extractable  $\text{NO}_3$  responses appear to support the hypothesis of Rustad et al. (2000) since the increase in soil and soil solution  $\text{NO}_3$  concentrations was smaller in the plots receiving reduced N inputs. However, the increase in nitrification rate was lower in  $\text{CO}_2\text{T-T}$  than in  $\text{T-T}$  (Verburg et al. 1999; Verburg and Van Breemen 2000), which could also account for the difference in  $\text{NO}_3$  response between  $\text{CO}_2\text{T-T}$  and  $\text{T-T}$ .

The soil solution data presented in this paper are consistent with changes in runoff observed by Lükewille and Wright (1997) and Wright (1998). During the 2 years of

**Fig. 2** Average annual soil solution  $\text{NH}_4$ ,  $\text{NO}_3$  and Cl concentrations in control ( $\text{CO}_2\text{T-C}$ ,  $\text{T-C}$ , Out-C) and heated ( $\text{CO}_2\text{T-T}$ ,  $\text{T-T}$ ) parts of the studied catchments during pre-treatment and treatment years (see Table 1 for catchment details). Error bars represent standard errors of the mean. The asterisks indicate significant treatment effects as determined by ANCOVA using pre-treatment data as co-variable (\*  $P < 0.05$ , \*\*  $P < 0.01$ )



treatment, soil warming in T-T increased the runoff  $\text{NH}_4$  concentration by  $3 \mu\text{Eq l}^{-1}$  and  $\text{NO}_3$  concentration by  $4 \mu\text{Eq l}^{-1}$  (Lükewille and Wright 1997). Increases in runoff  $\text{NH}_4$  and  $\text{NO}_3$  concentrations in response to climate treatments in  $\text{CO}_2\text{T-T}$  were smaller (Wright 1998), as were soil solution responses. Changes in runoff and soil solution concentrations in response to the climate treatments were relatively small compared to changes due to long-term manipulation of N inputs. Eleven years of reduced N inputs in  $\text{CO}_2\text{T-C}$  and  $\text{CO}_2\text{T-T}$  resulted in significantly lower soil solution  $\text{NH}_4$  and  $\text{NO}_3$  concentrations than in the plots receiving ambient N inputs (Out-C, T-C and T-T,  $P < 0.001$ ). During this period, runoff  $\text{NH}_4$  concentrations in  $\text{CO}_2\text{T}$  decreased by  $15 \mu\text{Eq l}^{-1}$  and  $\text{NO}_3$  concentrations by  $30 \mu\text{Eq l}^{-1}$  (Wright 1998). Therefore, it is very likely that if changes in N cycling in response to the climate treatments persist, soil solution responses may become more pronounced.

The results of this study showed that the climate manipulations in both catchments caused soil inorganic N pools to increase because of increases in N mineralization rates. Since KCl-extractable N is assumed to be readily exchangeable (e.g. Johnson et al. 2000) the observed increases in inorganic N pools most likely represented “leftover N” after plant and microbial demands were satisfied. Still, some competition between plant microbial uptake and adsorption may occur especially for  $\text{NH}_4$  enhancing accumulation of extractable  $\text{NH}_4$ . In addition to N uptake by the vegetation and microbes, several laboratory studies have demonstrated abiotic incorporation of  $\text{NH}_4$  and  $\text{NO}_3$  into soil organic matter as a potential removal mechanism of inorganic N (e.g. Johnson et al. 2000; Fitzhugh et al. 2003). Although these studies indicate that abiotic N immobilization is common in forest soils, its importance on an ecosystem scale has yet to be determined (Johnson et al. 2000). Finally, it is unknown if the increased mineralization rates and thus increase in soil inorganic N pools will be sustained indefinitely. In Harvard forest soil warming initially resulted in an increase in N mineralization, but this stimulation decreased after several years most likely due to a decrease in labile N pools (Melillo et al. 2002). Despite the importance of these potential acclimation effects little experimental evidence exists that allows for evaluation of long-term impacts of environmental change on N cycling.

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