# Course Project: Structural Analysis of the Hemoglobin Complex

**Proteomes Interactomes and Biological Networks** 

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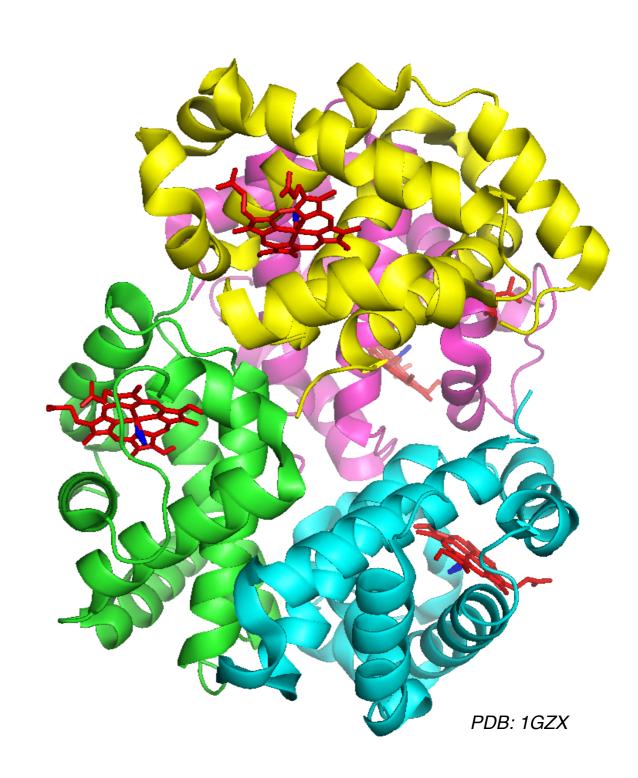


## Hemoglobin

Hemoglobin is an oxygen-transport protein. It is a tetramer composed by two subunits designated  $\alpha$  and  $\beta$ , with stoichiometry  $\alpha 2\beta 2$ .

The four subunits of hemoglobin sit roughly at the corners of a tetrahedron, facing each other across a cavity at the center of the molecule. Each of the subunits contains a heme prosthetic group.

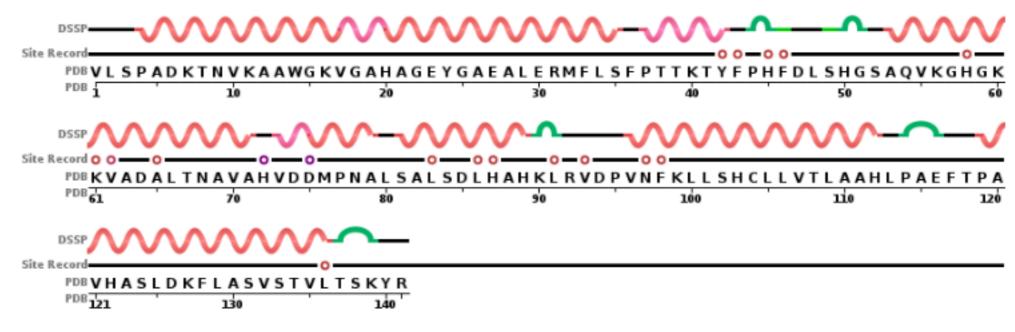
Each individual heme molecule contains one Fe2+ atom. The heme group binds oxygen while still attached to the hemoglobin monomer.



## Subunit Alpha

All-alpha chain A and C of the PDB structure 1GZX composed by 141 residues

#### Sequence Chain View



#### Site Record Legend

BINDING SITE FOR RESIDUE HEM B1290 (Software)
 BINDING SITE FOR RESIDUE HEM A1142 (Software)

#### BINDING SITE FOR RESIDUE OXY A1143 (Software)

empty: no secondary structure assigned

S: bend

T: turn

DSSP Legend

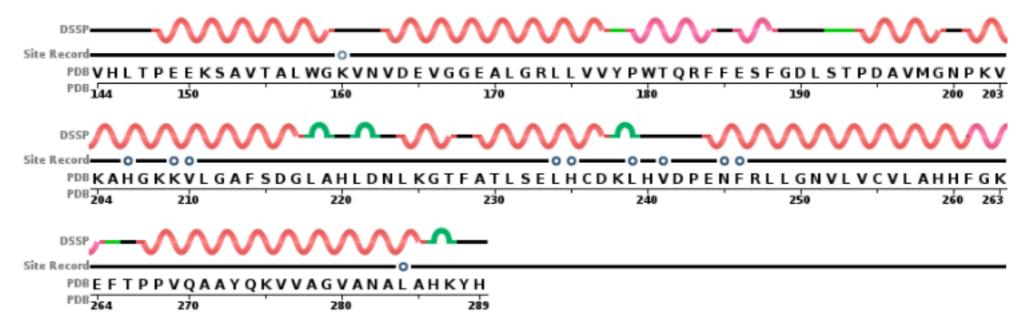
G: 3/10-helix

H: alpha helix

## Subunit Beta

All-alpha chain B and D of the PDB structure 1GZX composed by 146 residues

#### Sequence Chain View



#### Site Record Legend

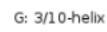
- BINDING SITE FOR RESIDUE HEM C1542 (Software)
- BINDING SITE FOR RESIDUE HEM B1290 (Software)

#### DSSP Legend

empty: no secondary structure assigned



S: bend



H: alpha helix

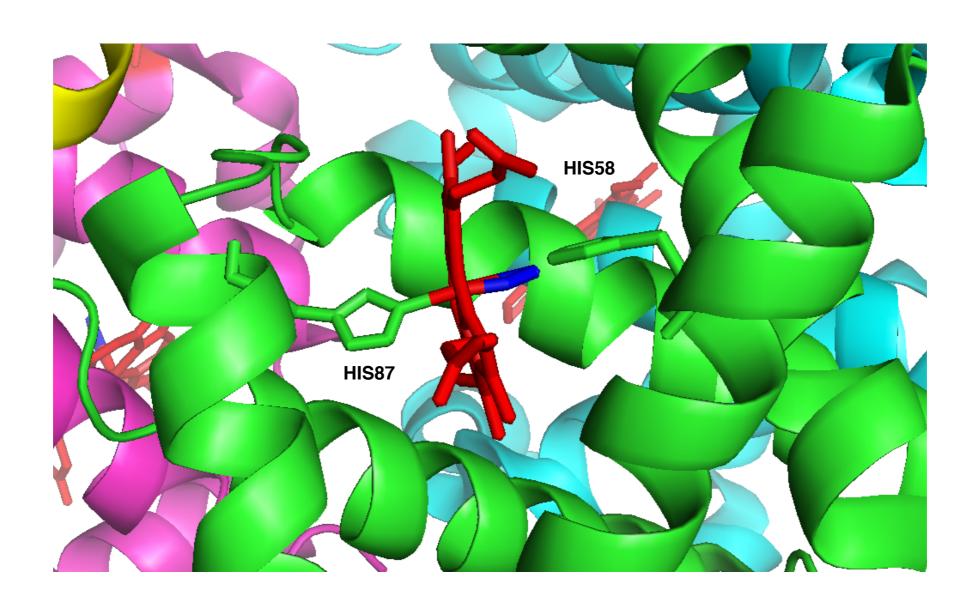
# Sequence Comparison

The alignment of the two sequences shows that they share ~44% of the residues

```
n-w bits E(1)
The best scores are:
                                                    ( 147) 373 65.3 1.4e-161
HBB HUMAN 147 bp
>>HBB HUMAN 147 bp
                                                           (147 aa)
n-w opt: 373 Z-score: 320.6 bits: 65.3 E(1): 1.4e-161
global/local score: 373; 43.6% identity (74.5% similar) in 149 aa overlap (1-142:1-147)
HBA HU MV-LSPADKTNVKAAWGKVGAHAGEYGAEALERMFLSFPTTKTYFPHF-DLS----HGS
       :: :.: .:. : : :::: .. : :.::: :... .: :. .: : :::
HBB HU MVHLTPEEKSAVTALWGKV--NVDEVGGEALGRLLVVYPWTQRFFESFGDLSTPDAVMGN
               10
                           20
                                      30
                                                40
                      70
                                80
                                           90
                                                    100
HBA HU AQVKGHGKKVADALTNAVAHVDDMPNALSALSDLHAHKLRVDPVNFKLLSHCLLVTLAAH
        .::.:::: : :...... :..... ::.... ::.... ::.... ::.... ::.... :. .:. ::
HBB HU PKVKAHGKKVLGAFSDGLAHLDNLKGTFATLSELHCDKLHVDPENFRLLGNVLVCVLAHH
       60
                 70
                           80
                                      90
                                               100
                                                         110
           120
                     130
HBA HU LPAEFTPAVHASLDKFLASVSTVLTSKYR
       . :::: :.:. .: .:.:...:. ::.
HBB HU FGKEFTPPVOAAYOKVVAGVANALAHKYH
      120
                130
                          140
```

## Heme and Oxygen

Anchoring of the heme is facilitated by a histidine nitrogen that binds to the iron. A second histidine is near the bound oxygen.



### Problem 1

Given the PDB structure 1GZX for each one of the monomers:

- Considering a minimum atom distance of 3.5 Å, identify the residues in proximity of the heme and oxygen group.
- Using the same procedure identify the possible interacting residues among the monomers.
- Generate the dssp file of the whole structure and all substructure and calculate the surface of interaction between all the monomers.
- Evaluating the differences in relative solvent accessibility, identify the interacting residues among all the monomers.
- Compare the results obtained for each monomer.

## Suggestions

- Modify the program for parsing a PDB file including the file 'HETATM' that includes the coordinate of the Heme and Oxy groups
- The distance between two group of atoms is minimum distance between all possible pairs combinations
- The functions used to calculate the distance between prostetic groups and monomers can be modified to calculate the distance between monomers.