### Project Report

**Proteomes Interactomes and Biological Networks** 

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#### Report Outlines

- Title
- Abstract: Summary of the work
- Introduction: Description of the Hemoglobin function
- Methods: Detailed information about the methodologies used for the analysis
- Results: Quantitative results of the analysis
- Discussion: Short summary of the results
- References: List of articles and web pages
- Supplementary Materials: Information not included in the main report

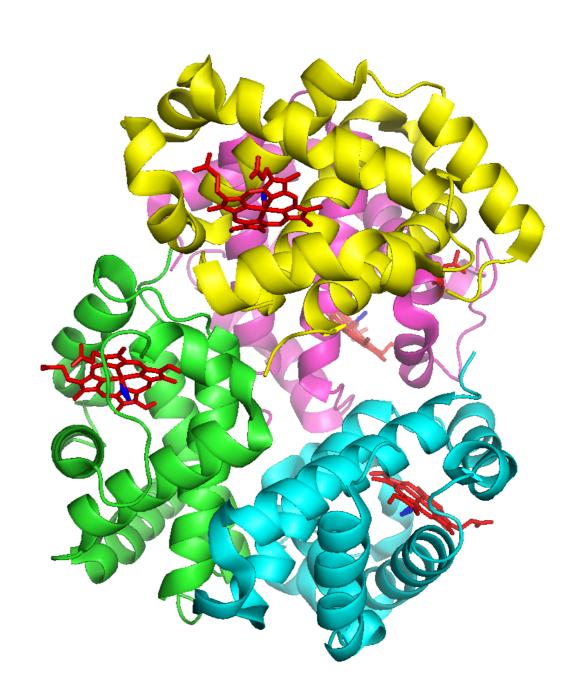
#### Introduction

Description of the hemoglobin function and the protein complex

Oxygen transport

• Tetramer composed by 2 types of monomers ( $\alpha$  and  $\beta$  subunits)

Each monomer interact with a Heme group



#### Methods

Detailed description of the data and methodologies used for the development of the project.

 Details about the protein structure used for the analysis of the Hemoglobin complex (1GZX).

Programs used for the analysis of complex.

 Procedure used for the analysis of the hemoglobin complex: Calculation of the physical interactions (heme-monomers and between monomers) and the surface of interaction between monomers.

#### Results (I)

Quantitative results of the analysis divided in two main parts:

• Analysis of the physical interactions heme and oxygen groups and monomers and between monomers. What are the atoms and residues below 3.5 Å?

The mean donor-acceptor distances in protein secondary structure elements are close to 3.0 Å. Since many pdb files lack hydrogen atoms, the presence of an energetically significant hydrogen bond can be inferred when a probable donor and acceptor are within 3.5 Å of each other (<a href="https://proteopedia.org/wiki/index.php/Hydrogen\_bonds">https://proteopedia.org/wiki/index.php/Hydrogen\_bonds</a>).

The distance between the residues participating in the salt bridge is less than 4 Å (https://proteopedia.org/wiki/index.php/Salt\_bridges).

# Table (I)

Heme - monomer interactions:

| Chain | Residue | Hetero  | Atoms (≤3.5Å) |
|-------|---------|---------|---------------|
| Α     | HIS58   | OXY1143 | NE2-O2,       |
|       | HIS87   | HEM1142 | NE2-FE,       |

Interactions between monomers:

| Chair | n1 | Residue1 | Chain2 | Residue2 | Atoms (≤3.5Å) |
|-------|----|----------|--------|----------|---------------|
| Α     |    | ARG141   | С      | ASP526   | NH2-OD2,      |
|       |    |          |        |          |               |

Highlights the salt bridges that stabilizes the interactions and show some figures

### Results (II)

Quantitative results of the analysis divided in two main parts:

 Analysis of the surface of interaction between monomers and the lost of accessibility of the single residues

Calculate the surface of interaction for each pair of chains to calculate which chains has stronger interaction.

Determine the possible interaction hot-spots considering the hydrophobic residues with large value of relative solvent accessibility lost.

## Table (II)

Surface of interaction between monomers:

| Chain1 | Chain2 | SA (Ų) |  |
|--------|--------|--------|--|
| Α      | В      | 994    |  |
|        |        |        |  |

Lost relative solvent accessibility for each residue

| Chain | Residue | RSA(M) | RSA(C) | RSA(M)-RSA(M) |
|-------|---------|--------|--------|---------------|
| Α     | LEU34   | 0.74   | 0.44   | 0.31          |
|       |         |        |        |               |

Show the residues with more than 10% of difference and highlight the hydrophobic residues with high difference

#### Project Submission

- The report of the project should be a PDF file named lastname\_pibn.pdf
- A directory containing the supplementary materials should be send as a unique zipped file named lastname\_supmat.zip.
- All the report should be send by email to emidio.capriotti@unibo.it by December 31, 2020.
- The subject of the mail should be "lastname PIBN project".