Supporting information

# Maud’s kinetic model

## Parameters

[Table 1](#tbl-params) shows all of Maud’s unknown parameters along with their dimensions

Note that Maud’s metabolic model includes some quantities that are not treated as parameters in its statistical model, including temperatures, compartment volumes and the formation energy of water. Maud treats these quantities as if they were known precisely: they can be configured by the user or default values can be used. Although in practice there can be considerable uncertainty regarding these quantities, we chose to disregard this uncertainty in the interest of simplicity.

Table 1: Parameters of Maud’s statistical model

| Parameter | Modelled quantity | Dimensions |
| --- | --- | --- |
|  | Formation energy | metabolites |
|  | Michaelis Menten constants | Substrates of all enzyme/reactions and products of reversible enzyme/reactions |
|  | Inhibition constants | Inhibiting metabolite/compartments of enzyme/reactions exhibiting competitive inhibition |
|  | Rate constants | Enzyme/reactions |
|  | Transfer constants | Allosteric interactions |
|  | T dissociation constants | Modifying metabolites of allosteric inhibitions |
|  | R dissociation constants | Modifying metabolites of allosteric activations |
|  | Rate constants of phosphorylation modifying enzymes | Phosphorylation modifying enzymes |
|  | Drain fluxes | Drains, experiments |
|  | Enzyme concentrations | Enzymes, experiments |
|  | Unbalanced metabolite/compartment concentrations | Unbalanced metabolite/compartments, experiments |
|  | Phosphorylation modifying enzyme concentrations | Phosphorylation modifying enzymes, experiments |
|  | Membrane potentials | Experiments |

Solving the steady state problem for a given set of parameters in an experiment yields a vector of balanced metabolite concentrations. These are combined with the balanced metabolite concentrations to produce a vector with a concentration for each metabolite/compartment combination.

parameters can optionally be fixed; this can be useful for computational purposes, as for example to avoid estimating the formation energy of a metabolite about which there is no available information due to it only participating in irreversible reactions.

## Rate equations

As discussed in the main text, Maud’s kinetic model decomposes into factors contributing to the flux in a metabolic network in an experiment as shown in equation . For succinctness, and since Maud’s model assumes that there are no interactions between experiments, we omit any notation referring to experiments below. We also omit any reference to the network’s drain reactions: these are modelled as being exactly determined by the values of the parameter vector .

The term in equation is a vector of non-negative real numbers representing the concentration of the enzyme catalysing each reaction.

The term in equation is a vector of non-negative real numbers representing the amount of flux carried per unit of saturated enzyme.

The term in equation is a vector of real numbers capturing the impact of thermodynamic effects on the reaction’s flux, as shown in equation .

The terms in have the following meanings:

* is the temperature in Kelvin (a number),
* is the gas constant (a number),
* is a vector representing the Gibbs free energy change of each reaction in standard conditions,
* is a vector representing the standard condition Gibbs free energy change of each metabolite’s formation reaction, or in other words each metabolite’s ‘formation energy’.
* is a vector representing the number of charges transported by each reaction.
* is the Faraday constant (a number)
* is a vector representing each reaction’s membrane potential (these numbers only matter for reactions that transport non-zero charge)

Note that, for reactions with zero transported charge, the thermodynamic effect on each reaction is derived from metabolite formation energies. This formulation is helpful because, provided that all reactions’ rates are calculated from the same formation energies, they are guaranteed to be thermodynamically consistent.

The term accounts for both the charge and the directionality. For instance, a reaction that exports 2 protons to the extracellular space in the forward direction would have -2 charge. If a negatively charged molecule like acetate is exported in the forward direction, would be 1.

Note that this way of modelling the effect of transported charge does not take into account that the concentration gradient used by the transport is that of the dissociated molecules. Thus, this expression is only correct for ions whose concentration can be expressed in the model only in the charged form; e.g., protons, , , , etc.

The term in equation is a vector of non-negative real numbers representing, for each reaction, the fraction of enzyme that is saturated, i.e. bound to one of the reaction’s substrates. To describe saturation we use equation , which is taken from Liebermeister, Uhlendorf, and Klipp (2010). Additionally, this term captures competitive inhibition: as competitive inhibitor concentration increases, the saturation denominator increases, effectively decreasing the saturation of the substrate on the total enzyme pool. Conversely, as the substrate concentration increases this term approaches 1.

The term in equation is a vector of non-negative numbers describing the effect of allosteric regulation on each reaction. Allosteric regulation happens when binding to a certain molecule changes an enzyme’s shape in a way that changes its catalytic behaviour. We use equation to describe this phenomenon, following the generalised MWC approach described in Monod, Wyman, and Changeux (1965), Changeux (2013), Popova and Sel’kov (1975) and Popova and Sel’kov (1979).

The parameter in equation is called the transfer constant, and the parameter vectors and are called tense and relaxed dissociation constants respectively.

Finally, the term in equation captures the important effect whereby enzyme activity is altered due to a coupled process of phosphorylation and dephosphorylation. This description achieves a similar behaviour to the MWC formalism for describing allosteric regulation, but using the rates of phosphorylation and dephosphorylation rather than concentrations of metabolites.

# Methionine case study

## Dataset generation

## Prior distributions compared with true parameter values

[Table 2](#tbl-case-study-params) shows the prior distributions we used for independent parameters.

Table 2: Parameter specification, marginal prior distributions and true parameter values used in our case study.

| parameter name | 1% prior quantile | 99% prior quantile | true value | prior Z-score of true value |
| --- | --- | --- | --- | --- |
|  | 3.430e-06 | 0.002480 | 9.3e-05 | 0.004 |
|  | 3.000e-05 | 0.002000 | 2.000e-05 | -2.787 |
|  | 1.000e-04 | 0.001000 | 3.170e-04 | 0.003 |
|  | 4.500e-04 | 0.000800 | 6.000e-04 | 0.000 |
|  | 1.120e-07 | 0.000081 | 2.000e-06 | -0.101 |
|  | 1.120e-05 | 0.008050 | 2.290e-04 | -0.136 |
|  | 1.120e-07 | 0.000081 | 1.500e-05 | 0.549306 |
|  | 1.200e+02 | 400.000000 | 2.340e+02 | 0.179861 |
|  | 6.000e+00 | 35.000000 | 1.380e+01 | -0.135 |
|  | 1.000e+01 | 188.000000 | 7.020e+00 | -2.887 |
|  | 7.000e-01 | 60.000000 | 1.050e+01 | 0.352083 |
|  | 8.200e-02 | 59.100000 | 7.900e+00 | 0.44375 |
|  | 5.890e-01 | 424.000000 | 1.990e+01 | 0.080556 |
|  | 4.840e-01 | 349.000000 | 1.160e+00 | -1.209 |
|  | 1.000e+00 | 3.300000 | 1.770e+00 | -0.091 |
|  | 1.300e+00 | 4.200000 | 3.170e+00 | 0.183333 |
|  | 1.590e-01 | 0.222000 | 2.650e-01 | 0.41875 |
|  | 2.000e-06 | 0.001400 | 5.300e-05 | 0.010 |
|  | 3.000e-04 | 0.000400 | 3.470e-04 | 0.014 |
|  | 1.000e-06 | 0.000030 | 6.000e-06 | 0.021 |
|  | 5.220e-05 | 0.037600 | 2.320e-05 | -2.050 |
|  | 1.670e-07 | 0.000120 | 5.660e-06 | 0.081944 |
|  | 1.580e-07 | 0.000114 | 1.060e-05 | 0.318056 |
|  | 1.200e-05 | 0.000032 | 1.980e-05 | 0.049 |
|  | 4.720e-05 | 0.034000 | 8.460e-03 | 0.659028 |
|  | 1.000e-06 | 0.000025 | 4.240e-05 | 3.090 |
|  | 2.000e-06 | 0.000004 | 2.830e-06 | 0.004 |
|  | 1.300e-05 | 0.009400 | 5.200e-04 | 0.1375 |
|  | 4.100e-07 | 0.000295 | 1.100e-05 | 0.000 |
|  | 5.480e-05 | 0.039500 | 2.540e-03 | 0.189583 |
|  | 3.730e-09 | 0.000003 | 1.000e-07 | 0.000 |
|  | 1.400e-05 | 0.000720 | 1.070e-04 | 0.074 |
|  | 5.270e-05 | 0.038000 | 2.030e-03 | 0.125694 |
|  | 4.470e-05 | 0.032200 | 1.130e-03 | -0.029 |
|  | 5.270e-05 | 0.038000 | 2.370e-03 | 0.179167 |
|  | 7.000e-06 | 0.000013 | 9.370e-06 | -0.135 |
|  | 3.320e-06 | 0.002390 | 6.940e-05 | -0.124 |
|  | 1.000e-06 | 0.000003 | 1.710e-06 | -0.054 |
|  | 7.500e-05 | 0.000088 | 8.080e-05 | -0.158 |
|  | 1.600e-05 | 0.000028 | 2.090e-05 | -0.105 |
|  | 4.500e-05 | 0.000085 | 4.390e-05 | -2.507 |
|  | 3.730e-02 | 26.800000 | 1.030e+00 | 0.017 |
|  | 3.730e-02 | 26.800000 | 1.310e+02 | 0.3875 |
|  | 3.730e-03 | 2.680000 | 1.080e-01 | 0.037 |
|  | 1.120e-01 | 80.500000 | 3.920e-01 | -1.018 |

parameters for most metabolites were fixed; those that were modelled as unknown had a multivariate normal prior distribution derived from eQuilibrator (Beber et al. 2021).

The values for parameters, as well as all other model parameters, can be found by inspecting the file priors.toml which is online at <https://github.com/biosustain/Methionine_model/blob/main/data/methionine/priors.toml>.

The results of all reported Maud runs can be found at <https://github.com/biosustain/Methionine_model/blob/main/results>.

# References

Beber, Moritz E., Mattia G. Gollub, Dana Mozaffari, Kevin M. Shebek, and Elad Noor. 2021. “eQuilibrator 3.0 – a Platform for the Estimation of Thermodynamic Constants.” *arXiv:2103.00621 [q-Bio]*, February. <http://arxiv.org/abs/2103.00621>.

Changeux, Jean-Pierre. 2013. “50 Years of Allosteric Interactions: The Twists and Turns of the Models.” *Nature Reviews. Molecular Cell Biology* 14 (12): 819–29. <https://doi.org/10.1038/nrm3695>.

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