



UNIVERSITY OF VIENNA  
FACULTY OF PHYSICS

QUANTUM OPTICS PRACTICAL COURSE

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KDTLI

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## Abstract

# 1 Molecule Interference

The aim of this experiment is to show interference patterns using  $C_{60}$  molecules and the quantum mechanical description of matter waves. The following review will give you an introduction to the theory of the experiment as well as a description of the experimental setup. At the end, the obtained results are presented and discussed.

## 2 Theory

### 2.1 Matter Waves

According to quantum theory, also massive particles can be described as a wave, where the wavelength is given by the de Broglie relation.

$$\lambda = h/p$$

Thus, interference and defraction of these massive particles can be observed. The Schrödinger equation describes the propagation of such a matter wave. For time independent problems, this equation can be reduced to the Helmholtz equation. This means, that matter waves can be described as electromagnet waves and also exhibit the same effects as electromagnet waves. Thus, one can use the Talbot and the Lau effect, which are needed to perform the experiment.

### 2.2 Talbot Effect

The Talbot effect is observed in the optic near feld, or Fresnel regime, where the

curvature of the wavefront can not be neglected. It occurs when coherent light impingin on a periodic grating. Then, self images of the grating can be seen at multiples of a charactersitic distance from the grating, usually referred to as Talbot lenght. The Talbot lenght is wavelength dependend and can be calculatet by  $L_T = d^2/\lambda$ , where d is the grating period.

### 2.3 Lau Effect

This effect is similar to the Talbot effect but the crucial difference is that one can use an incoherent (light) source. Incoherent light is impinging on a grating. Every slit of this grating can be seen as the startig point of a cylindrical wave. If there is a second grating with a period similar to the first one, these waves produce talbot images after the second, which are interfering constructivley. This effect is used to get a spatial coherent molecule beam.

### 2.4 Kapitza-Dirac Effect

The Kapitza-Dirac effect describes the phenomenon that the electric field of a standing light wave can induce a dipole moment in the molecules passing through. This leads to a phase boost in the molecules depending on the position they are passing through. The standig light wave acts as a grating, or better spoken, as a phase grating. Using a non-matter grating has some advantages. One of them is that Van der Waals forces can be neglected. Furthermore, the passing molecules do not damage the grating and the lattice constant can easily be modified since it is the half of the wavelength.

### 3 Experimental assembly

The following figure shows a sketch of the experimental setup.

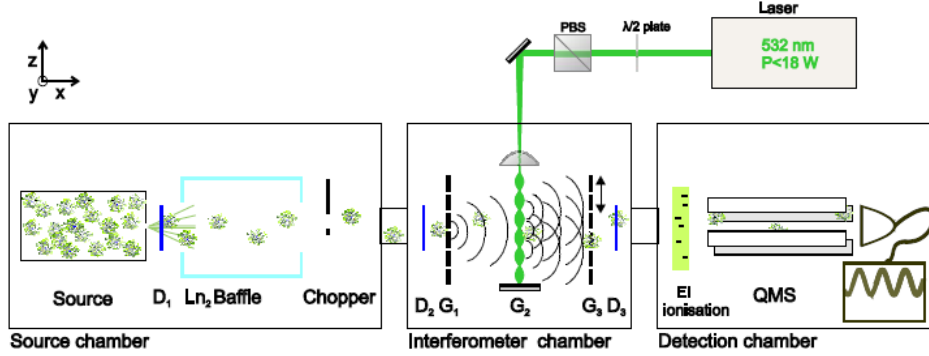


Figure 1: Experimental Setup. Taken from the Quantum Optic Course manual.

The setup is divided into 3 compartments. The first part contains the oven (which is the source of the molecules) as well as the chopper and delimiters which are needed for the velocity selection. After the source chamber, the molecule beam enters the interferometer chamber. In this part, the 3 gratings are situated.  $G_1$  and  $G_3$  are silicon nitride gratings with periodicity of 266nm.  $G_2$  is a standing light wave produced by a 532nm Laser (which results in a 266nm periodicity, too). The third and last compartment contains an electron ionization quadrupole mass spectrometer where the interference patterns are detected. In order to get the intended result, the whole experiment takes place in vacuum. This ensures that the molecules do not collide with atoms of the background gas.

The oven itself is a ceramic cylinder which is secluded by a stainless steel top cover and heated by a heating wire. It is used to evaporate the molecules. In

our case, it was filled with  $C_{60}$  molecules. The apparatus is mounted on a translation stage, thus, the direction of the beam can be modulated in the X/Y axis via joystick. In order to get a temporal coherent beam, a velocity selection has to be made. At the beginning of the experiment, the oven is heated up until it has reached a stable temperature of about 620°. Then, the joystick was used to optimize the count rate. After that, the velocity selection can be made.

Since there is no background gas and only gravitation acting on them, the molecules travel along a parabola. To select a specific one, 3 delimiters can be inserted into the beam, which are sufficient due to the fact that a parabola is well defined by 3 points. The width of the delimiters can be modified by using Piezo effects, thus applying voltage. For the actual measurement of the velocity, the so called chopper is used. The chopper is a rotating disc which can be inserted into

the beam. All over the chopper, there are certain patterns of openings. When the beam is passing through one of the holes, a signal is sent to a computer which measures the time of flight. By knowing the time and the distance, one can calculate the velocity and then plot it as function of the countrate. A plot of such an obtained distribution can be seen in figure 2. One must not forget to take the chopper out of the beam after the velocity measurement due to the fact that it blocks some part of the beam and it is rotating which causes vibrations that can affect the visibility of the interference.

After the source chamber, the molecules enter the interferometer chamber. It is crucial that the light grating is nearly perfect adjustet to the molecules. This can be done by measuring the visibility as function of the light grating height. Thus, the position of the laser grating is changed and measurements are done to determin the best position.

In the last chamber, the measurement takes place. Therefore, a quadrupol mass spectrometer is installed. Before the

molecules enter the QMS, they get bombarded by elecetrons and ionized. As the name indicates, the QMS is made of 4 ion rods. Only molecules with a certain  $m/z$  are passing down the rods into the detector.

## 4 Results

At the beginning, the oven was heated until a stable temperature of  $628^\circ$ . With this temperature, the velocity measurement was taken. Therefore, the three delimiters and the chopper were put into the beam. A software program calculated the time of flight. The distance between the oven and the detector was 1,74m, with this, the following results can be obtained.

$$v = 228,35 \pm 0,02$$

To aligne the laser as perfectly as possilbe to the molecule beam, the interference was recorded as function of the light grating height. Maximum visibility was measured at a height of 4,26mm.

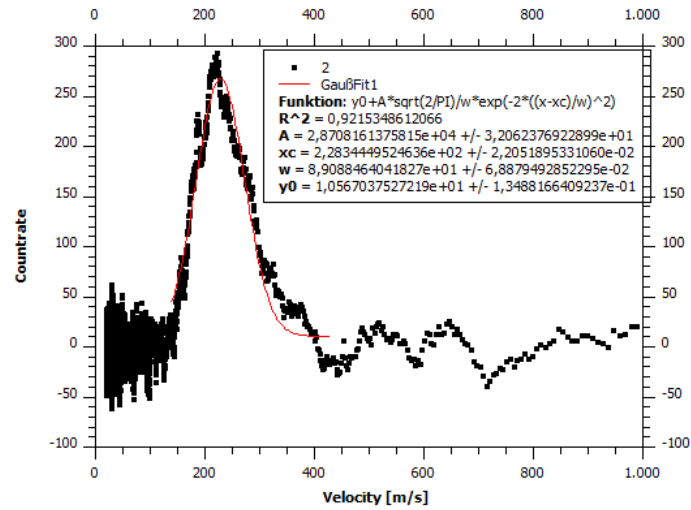


Figure 2: Velocity Spectrum with Gaus Fit

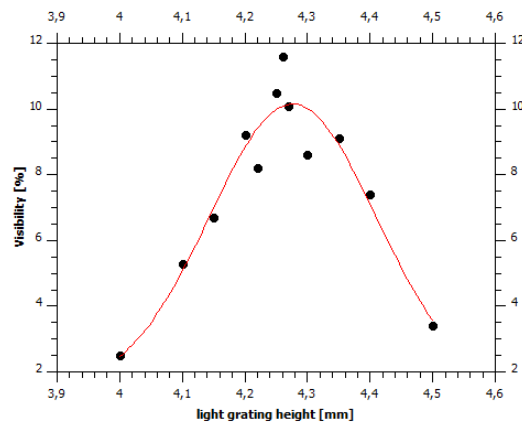


Figure 3: Interference as function of the light grating height

In the end, the visibility was measured as function of the laser power to get as much visibility as possible.

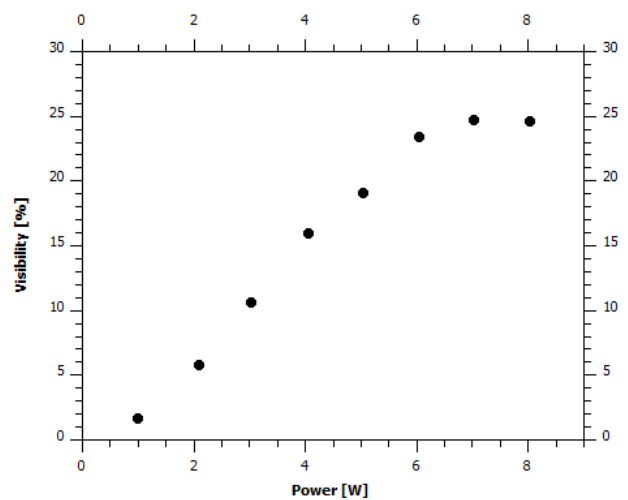


Figure 4: Interference as function of Laserpower

For the calculation of the visibility, a software program written in matlab was used. Alternatively, one can calculate it oneself by fitting a sine curve over the obtained data. For this reason, the data with the maximum visibility was used. The following figure shows the plot. The calculated visibility is  $24,30 \pm 0,03$  which is the same as the calculated visibility by the software program.

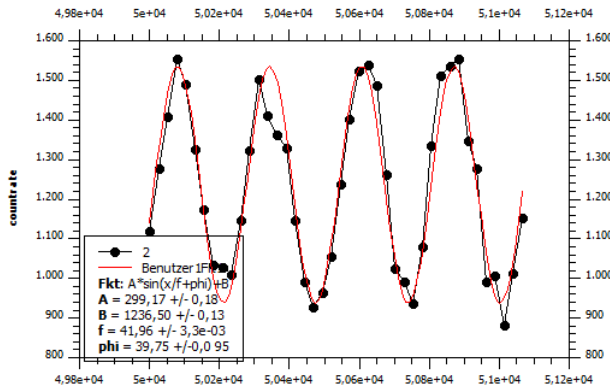


Figure 5: Calculation of visibility

## 5 Discussion

Since it was worked on a real experiment which is currently in use, great caution had to be taken. Therefore, a simulation prior to the actual day in the laboratory had to be done. This simulation was very useful and a good preparation. It also helped to save time at the day of the experiment. Another positive effect of using a real experiment is that this helped to decrease the possibility of systematic errors done by false alignment for instance. The experiment was a success, since we could show interference of large molecules. Thus, we could show the existence of matter waves. The remarkable thing about matter waves is, that classical

physics does not provide an explanation for them. Although, the maximum visibility of approximately 24% could have been better. For this experiment, it is important that the table on which it is standing, is free of vibration. For this reason, it is possible to levitate the hole table on air. Regrettably one of the pillars broke down the week before our experiment. This could have led to a decrease in the visibility. Moreover, the alignment could have been better to maximise the count rates.

## 6 Appendix

<i>Lightgratingheight</i> ( $\pm 0.1$ )[mm]	<i>Visibility</i> $\pm 0.7$ [%]
4.00	3
4.50	3
4.40	7
4.35	9
4.30	9
4.25	11
4.20	9
4.15	7
4.10	5
4.22	8
4.27	10
4.26	12

Figure 6: Light grating height optimization

<i>Laserpower</i> ( $\pm 0.01$ )[W]	<i>Visibility</i> $\pm 0.9$ [%]
0.99	2
2.08	6
3.01	11
4.05	16
5.02	19
6.02	24
7.02	25
8.02	25

Figure 7: Laserpower optimization

## 7 Resources

- Simulation of the Experiment and background information <http://interactive.quantumnano.at>

Sandra Eibenberger: Diplomarbeit, Kapitza-Dirac-Talbot-Lau Interferometrie mit komplexen Molekülen, Wien 2010  
Leitfaden Praktikum Quantenoptik