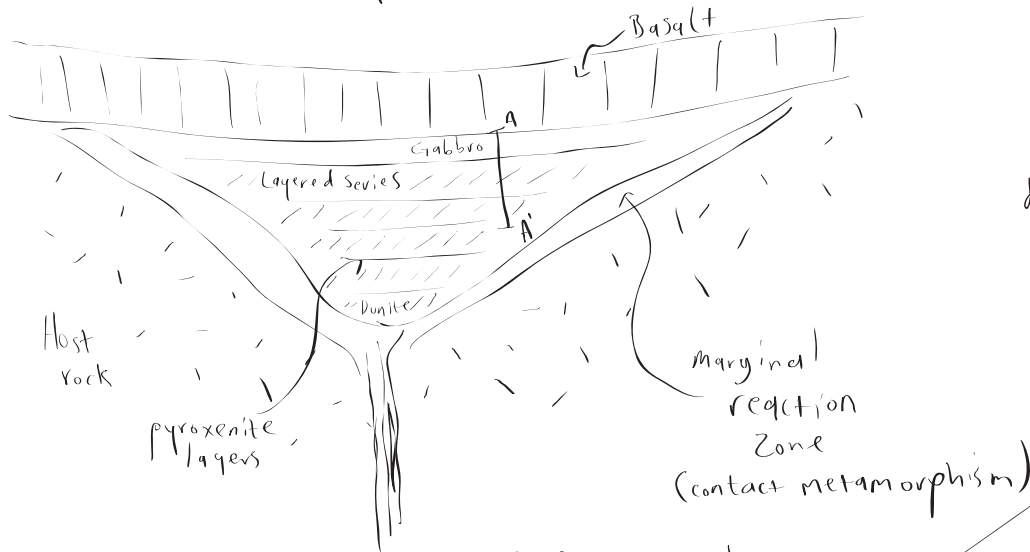


fractional crystallization

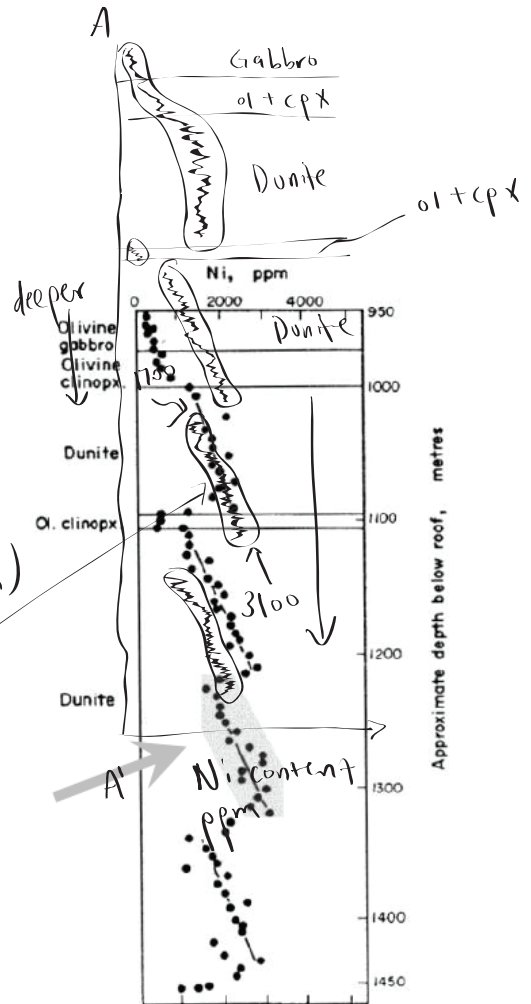
$$\frac{C_M}{C_{M_0}} = F^{D-1} \quad \text{or} \quad \frac{C_S}{C_{M_0}} = D F^{D-1}$$

Example of crystal cumulates
Muskox layered intrusion



Assume $D_{Ni}^{Bulk} = 15$

If fractional crystallization,
how much of the melt solidified?
(What is F?)



Variation in the Ni content of rocks of part of the Muskox intrusion, Northwest Territories, Canada. (After Irvine and Smith, 1967)
Henderson, Inorganic Chemistry, 1984

$$D F^{(D-1)} = \frac{C_S}{C_{M_0}} \leftarrow \text{also } C_{L_0}, D = \frac{C_{S_0}}{C_{L_0}}$$

$$D F^{(D-1)} = \frac{C_S}{\left(\frac{C_{S_0}}{D}\right)}$$

$$15 \cdot F^{(D-1)} = \frac{C_S}{C_{S_0}} \leftarrow \begin{matrix} 1700 \\ 3100 \end{matrix}$$

$$F^{14} = \frac{1700}{3100}$$

$$F = \left(\frac{1700}{3100}\right)^{1/14} = 0.958$$

or ~4% melt solidifies