

$$\sum \left\{ \begin{array}{c} \text{charge of} \\ \text{conservative positive ions} \end{array} \right\} - \sum \left\{ \begin{array}{c} \text{charge of} \\ \text{conservative negative ions} \end{array} \right\} = [\text{HCO}_3^-] + 2[\text{CO}_3^{2-}] + [\text{OH}^-] + [\text{B(OH)}_4^-] + [\text{H}_3\text{SiO}_4^-] + [\text{HS}^-] + [\text{H}_2\text{PO}_4^-] + 2[\text{HPO}_4^{2-}] - [\text{H}^+]$$