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# Interaction theory – Photons

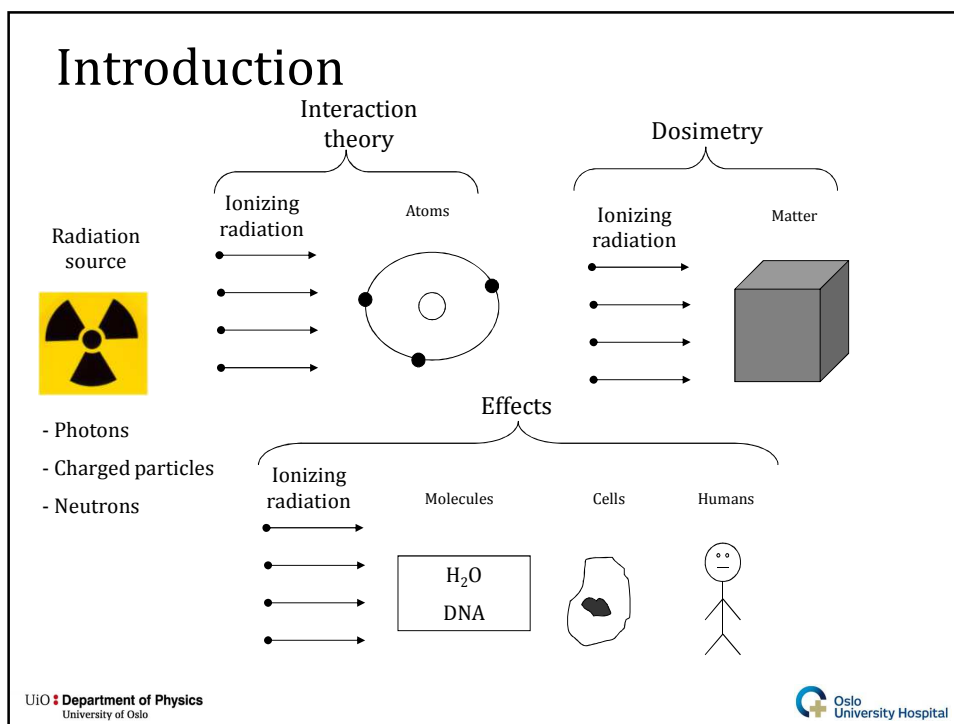
Eirik Malinen



# INTRODUCTION TO RADIOLOGICAL PHYSICS AND RADIATION DOSIMETRY

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## Objectives

- To understand primary effects of ionizing radiation
- How radiation doses are calculated and measured
- To appreciate applications of ionizing radiation

## Contents FYSKJM4710

- Interactions between ionizing radiation and matter
- Radioactive and non-radioactive sources
- Calculations and measurement of absorbed doses (dosimetry)

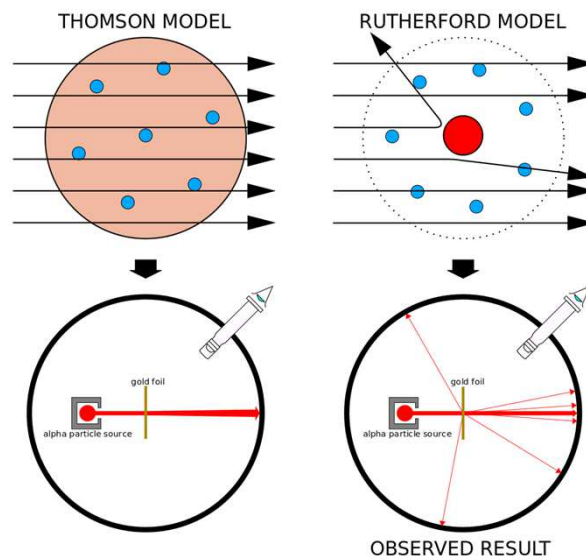
## Relevant issues

- X-ray and CT investigations
- Radiotherapy
- Positron emission tomography
- Radiation protection
- Radiation Biology

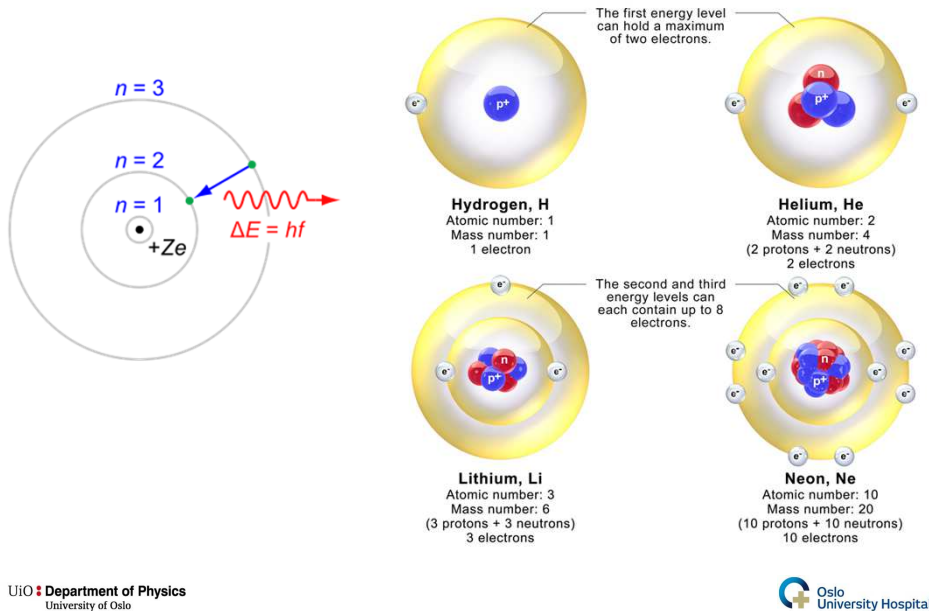


- X-ray contrast: only a matter of differences in density?

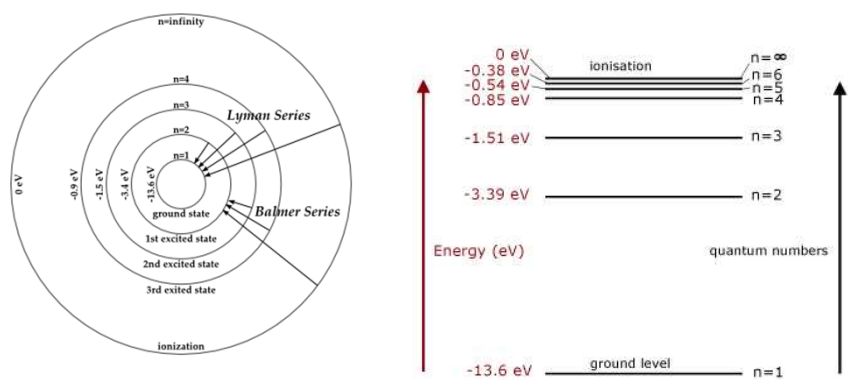
## Basic theory of the atom



# Bohr model

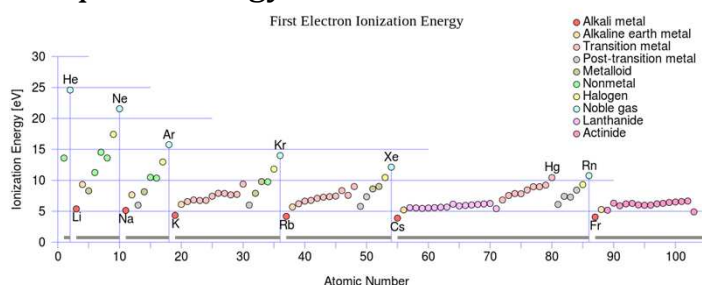


# Hydrogen



## Ionization

- Liberation of electron from atom
- Requires energy transfer  $\sim 4\text{-}25\text{ eV}$



- A lethal whole-body dose of radiation ( $5\text{ J/kg}$ ) results in a temperature increase of  $0.001\text{ }^{\circ}\text{C}$

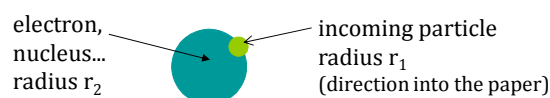
## Ionizing radiation

**Directly ionizing radiation:** Fast charged particles, which deliver their energy to matter directly, through many small Coulomb-force interactions

**Indirectly ionizing radiation:** Photons ( $\gamma$  or X-rays) or neutrons, which transfer their energy to charged particles in the matter

## Cross section 1

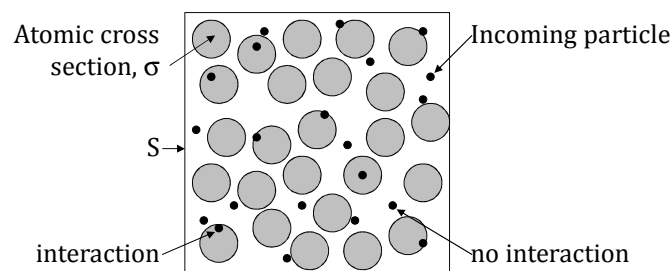
- Cross section  $s$ : “target area”, effective target covering a certain area
- Proportional to the interaction strength between an incoming particle and the target particle
- Consider two discs, one target and one incoming:



- $s$  is the total area:  $\pi(r_1^2 + r_2^2)$

## Cross section 2

- $N$  particles move towards an area  $S$  with  $n$  atoms



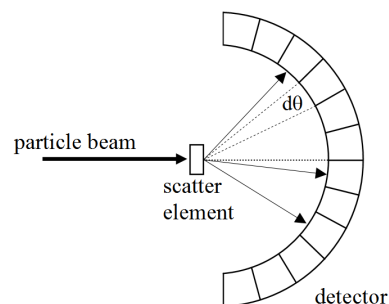
- Probability of interaction:  $p = n\sigma/S$
- Number of interacting particles:  $Np = Nn\sigma/S$

## Cross section 3

- Separate between *electronic* and *atomic* cross section
- The cross section depends on:
  - Type of target (nucleus, electron, ..)
  - Type of and energy of incoming particle (photon, electron...)
- Cross section calculated with quantum mechanics
  - here visualized in a classical window

## Cross section 4

- *Differential cross section* with respect to scattering angle

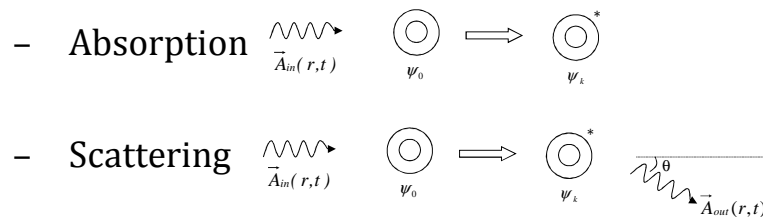


$$\frac{d\sigma}{d\Omega} = \frac{\text{number of particles scattered into } d\Omega}{\text{number of particles per unit area}} \frac{1}{d\Omega}$$



## Photon interactions

- Photon represented by a plane wave  $\vec{A}_{in}(r,t) \sim e^{i(\vec{p}_{in} \cdot \vec{r} - \omega_{in} t)}$  in quantum mechanical calculations
- In principle, two different processes:



- Scattering: coherent (elastic) og incoherent (inelastic)

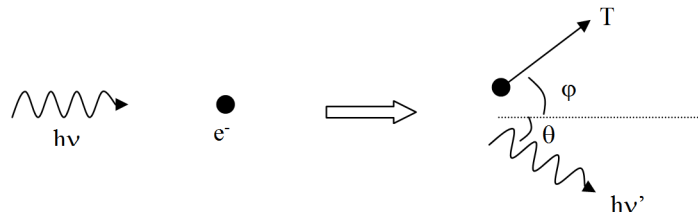
## Coherent (Rayleigh) scattering

- Scattering without loss of energy:  $h\nu = h\nu'$
- Photon is absorbed by atom, thereby emitted at a small deflection angle
- Depends on atomic structure and photon energy
- Atomic cross section:

$$\sigma_R \propto \left( \frac{Z}{h\nu} \right)^2$$

## Incoherent (Compton) scattering

- Scattering with loss of energy:  $h\nu' < h\nu$
- Photon-electron scattering; electron may be assumed free (i.e. unbound)



- Thomson scattering: low energy limit,  $h\nu \rightarrow 0$

## Compton scattering – kinematics

- Conservation of energy and momentum:

$$h\nu = h\nu' + T$$

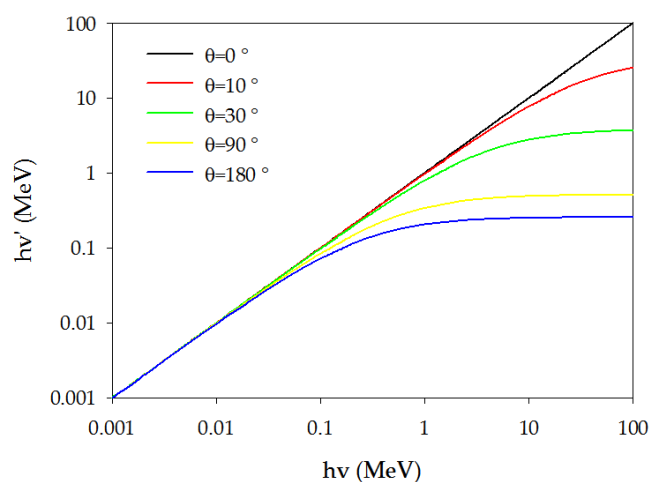
$$\frac{h\nu}{c} = \frac{h\nu'}{c} \cos \theta + p \cos \varphi, \quad \frac{h\nu'}{c} \sin \theta = p \sin \varphi$$

$$(pc)^2 = T^2 + 2Tm_e c^2$$

→

$$h\nu' = \frac{h\nu}{1 + \frac{h\nu}{m_e c^2} (1 - \cos \theta)}, \quad \cot \varphi = \left( 1 + \frac{h\nu}{m_e c^2} \right) \tan \left( \frac{\theta}{2} \right)$$

## Compton scattering – kinematics



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## Compton scattering – example

- An X-ray unit is to be installed, with the beam direction towards the ground. Employees in the floor above the unit are worried. Maximum X-ray energy is 250 keV. What is the maximum energy of the backscattered photons?

$$\theta = 180^\circ \Rightarrow h\nu' = \frac{h\nu}{1 + \frac{h\nu}{m_e c^2} (1 - \cos \theta)} = \frac{h\nu}{1 + \frac{2h\nu}{m_e c^2}}$$

$$h\nu = 250 \text{ keV} \Rightarrow h\nu' = \frac{250}{1 + \frac{2 \times 250}{511}} = \underline{\underline{126 \text{ keV}}}$$

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## Compton scattering – cross section 1

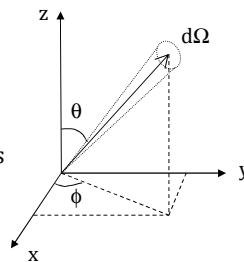
- Klein and Nishina derived the cross section for Compton scattering, assuming free electron
- Differential cross section:

$$\left( \frac{d\sigma}{d\Omega} \right) = \frac{r_0^2}{2} \left( \frac{\nu'}{\nu} \right)^2 \left( \frac{\nu'}{\nu} + \frac{\nu}{\nu'} - \sin^2 \theta \right)$$

$$d\Omega = \sin \theta d\theta d\phi$$

$r_0$ : classical electron radius

incoming photon along z-axis



## Compton scattering – cross section 2

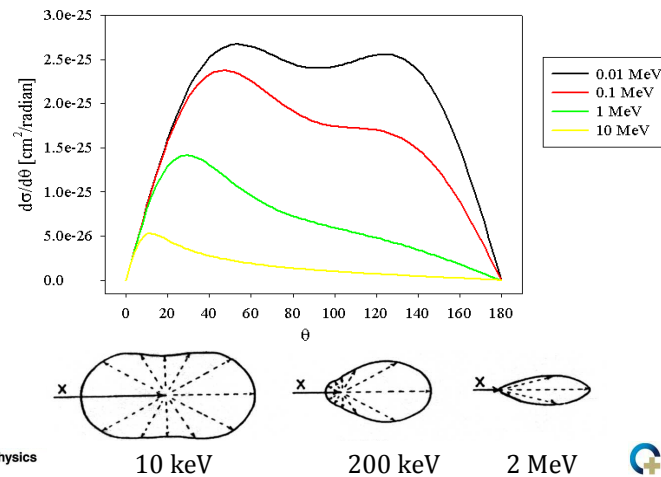
- Cylinder symmetry results in:
- $$\left( \frac{d\sigma}{d\theta} \right) = \pi r_0^2 \left( \frac{\nu'}{\nu} \right)^2 \left( \frac{\nu'}{\nu} + \frac{\nu}{\nu'} - \sin^2 \theta \right) \sin \theta$$
- $\sim$  probability of finding a scattered photon in the interval  $[\theta, \theta+d\theta]$
  - Total electronic cross section:

$$\sigma_e = \int_0^\pi \pi r_0^2 \left( \frac{\nu'}{\nu} \right)^2 \left( \frac{\nu'}{\nu} + \frac{\nu}{\nu'} - \sin^2 \theta \right) \sin \theta d\theta$$

- Atomic cross section:  $\sigma_a = Z \sigma_e$

## Compton scattering – cross section 3

- Scattered photons are more forwardly directed with increasing photon energy:



## Compton scattering – cross section 3

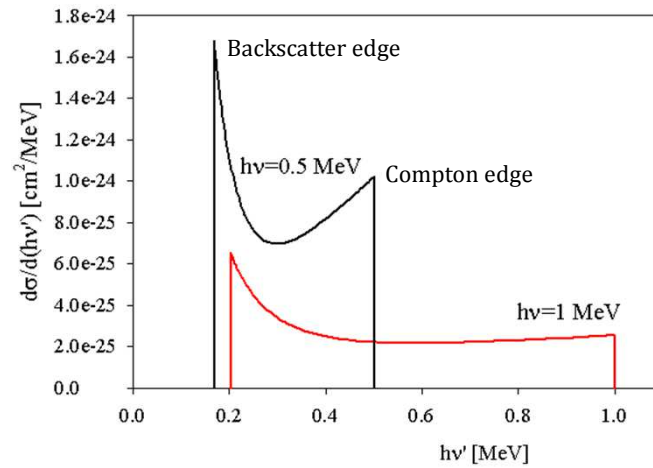
- Cross section may be modified with respect to energy:

$$\frac{d\sigma}{d(h\nu')} = \frac{d\sigma}{d\Omega} \frac{d\Omega}{d(h\nu')} = \frac{d\sigma}{d\Omega} 2\pi \sin\theta \frac{d\theta}{d(h\nu')}$$

$$h\nu' = \frac{h\nu}{1 + \frac{h\nu}{m_e c^2} (1 - \cos\theta)}$$

$$\Rightarrow \frac{d\sigma}{d(h\nu')} = \frac{\pi r_0^2 m_e c^2}{(h\nu)^2} \left[ \frac{h\nu'}{h\nu} + \frac{h\nu}{h\nu'} - 1 + \left( 1 - \left( \frac{h\nu}{h\nu'} - 1 \right) \frac{m_e c^2}{h\nu} \right)^2 \right]$$

## Compton scattering – cross section 4

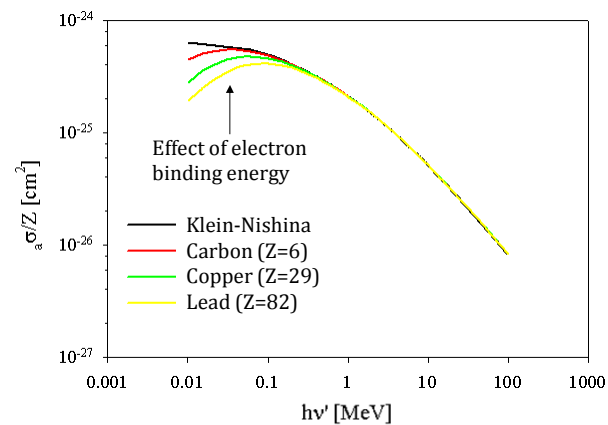


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## Compton scattering – cross section 5

- Correct atomic cross section:



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## Compton scattering – transferred energy 1

- The energy transferred to an electron in a Compton process:

$$T = h\nu - h\nu'$$

- The cross section for energy transfer:

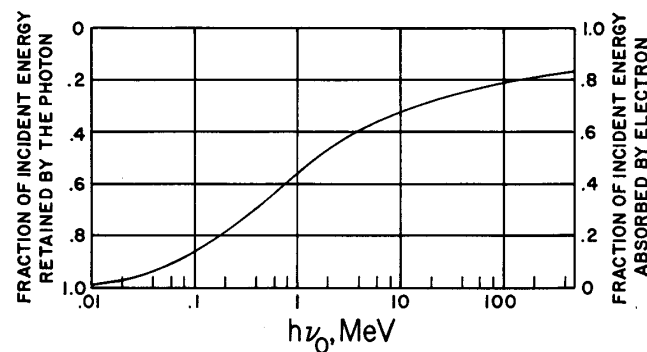
$$\frac{d\sigma_{tr}}{d\Omega} = \frac{d\sigma}{d\Omega} \frac{T}{h\nu} = \frac{d\sigma}{d\Omega} \frac{h\nu - h\nu'}{h\nu}$$

- Mean energy transferred:

$$\bar{T} = \frac{\int T \frac{d\sigma}{d\Omega} d\Omega}{\int \frac{d\sigma}{d\Omega} d\Omega} = \frac{\int \frac{h\nu - h\nu'}{h\nu} \frac{d\sigma}{d\Omega} d\Omega}{\sigma} = \frac{\sigma_{tr}}{\sigma} \times h\nu$$

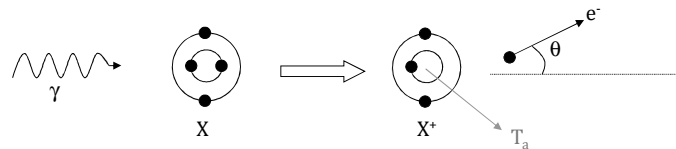
## Compton scattering – transferred energy 2

- The fraction of incident energy transferred:

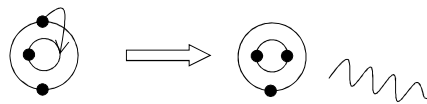


## Photoelectric effect 1

- Photon is absorbed by atom/molecule; the result is an excitation or ionization



- Atom may deexcite and emit characteristic radiation:



## Photoelectric effect 2

- In the kinematics, the binding energy of the ejected electron should be taken into account:

$$T = h\nu - E_b - T_a \approx h\nu - E_b$$

- Assuming  $E_b=0$ , the atomic cross section is:

$$\frac{d\tau}{d_e W} = 2\sqrt{2} r_0^2 \alpha^4 Z^5 \left( \frac{m_e c^2}{h\nu} \right)^{7/2} \sin^2 \theta \left( 1 + 4 \sqrt{\frac{2h\nu}{m_e c^2}} \cos \theta \right)$$

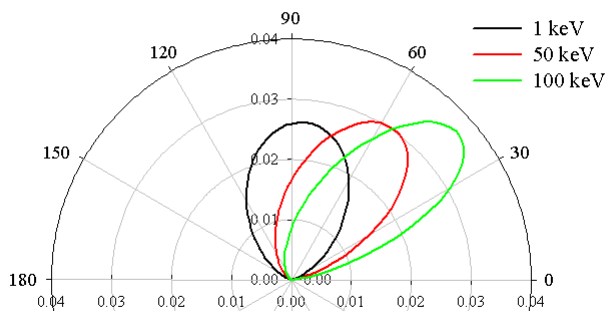
$\alpha$ : The fine-structure constant

Solid angle  $d\Omega$  gives the direction of the ejected electron



## Photoelectric effect 2

Photoelectric cross section  $(d\sigma/d\theta)/s$



## Characteristic radiation

- Energy of characteristic radiation depends on electronic structure and transition probabilities
- "K- and L-shell" vacancies  $\leftrightarrow h\nu_K$  and  $h\nu_L$
- Isotropic emission
- Fraction of photoelectric interactions:  
 $P_K [h\nu > (E_b)_K]$  and  $P_L [(E_b)_L < h\nu < (E_b)_K]$
- Probability for emission:  $Y_K$  og  $Y_L$  (fluorescence yield)
- Energy emitted from the atom:  
 $P_K Y_K h\nu_K + (1 - P_K) P_L Y_L h\nu_L$

## Auger effect

- Energy release by ejection of loosely bound electron
- Energy of emitted electron equal to deexcitation energy
- Low Z: Auger dominates
- High Z: characteristic radiation dominates

## Photoelectric cross section

- General formula:
- $$\tau \propto \frac{Z^n}{(h\nu)^m}, \quad 4 < n < 5, \quad 1 < m < 3$$
- Fraction of energy transferred to photoelectron:

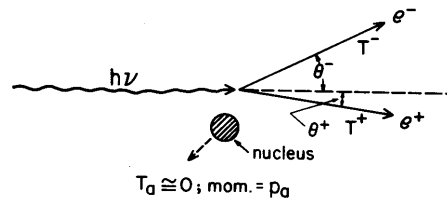
$$\frac{T}{h\nu} = \frac{h\nu - E_b}{h\nu}$$

- However: don't forget Auger electron(s)
- Cross section for energy transfer to photoelectron:

$$\tau_{tr} = \tau \frac{(h\nu - P_K Y_K h\nu_K - (1 - P_K) P_L Y_L h\nu_L)}{h\nu}$$

## Pair production 1

- Photon absorption in the nuclear electromagnetic field where an electron-positron pair is created



- Triplet production: in the electromagnetic field of an electron

## Pair production 2

- Conservation of energy:

$$h\nu = 2m_e c^2 + T^+ + T^-$$

- Average kinetic energy after absorption:

$$\bar{T} = \frac{h\nu - 2m_e c^2}{2}$$

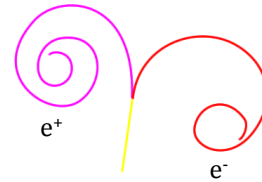
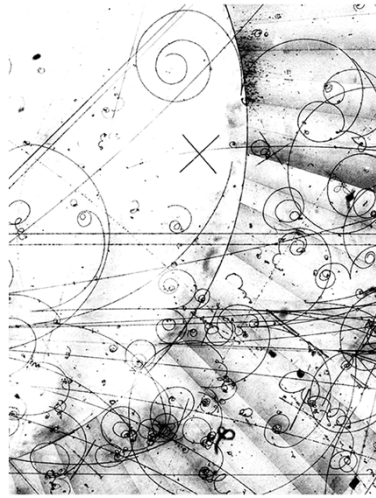
- Estimated electron/positron scattering angle:

$$\bar{\theta} \approx \frac{m_e c^2}{\bar{T}}$$

- Total cross section:

$$\kappa \approx \alpha r_0^2 Z^2 \bar{P}$$

## Discovery of pair production



⊙ Magnetic field

## Triplet production

- In the electromagnetic field from an electron, an electron-positron pair is created

- Energy conservation:

$$h\nu = 2m_e c^2 + T^+ + T_1^- + T_2^-$$

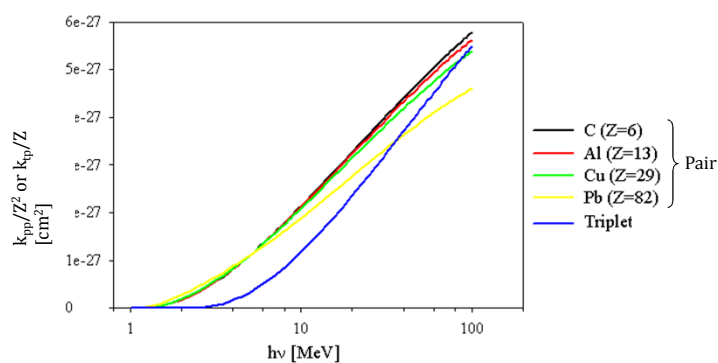
- Average kinetic energy:

$$\bar{T} = \frac{h\nu - 2m_e c^2}{3}$$

- Primary electron is also given energy
- Threshold:  $4m_0c^2$

## Pair- and triplet production

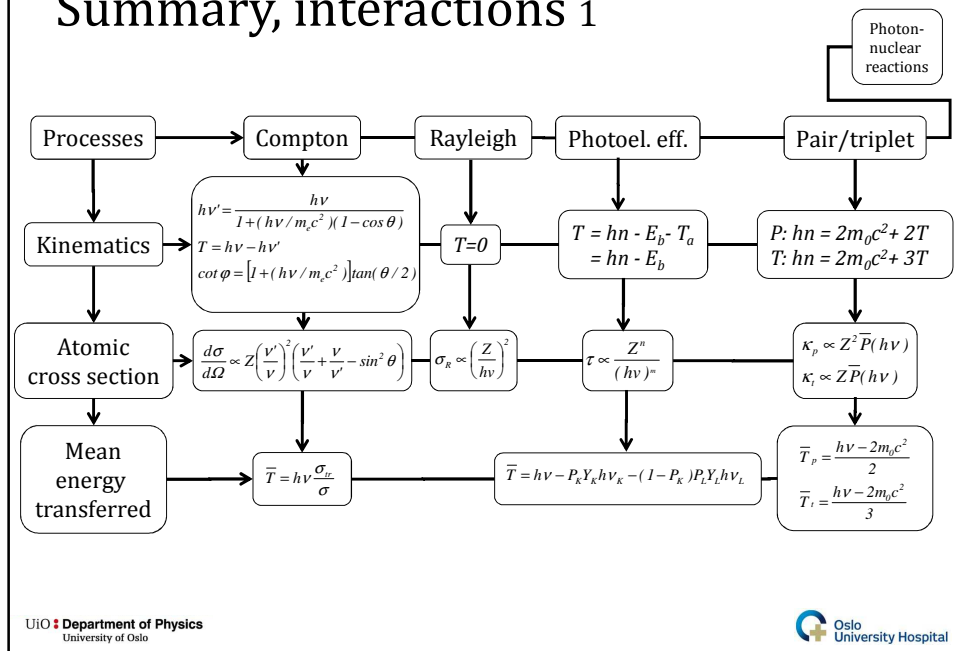
- Pair production dominates:



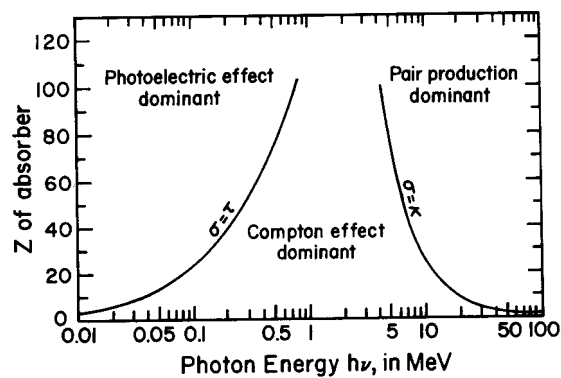
## Photonuclear reactions

- Photon (energy above a few MeV) excites a nucleus
- Proton or neutron is emitted
- $(\gamma, n)$  interactions may have consequences for radiation protection
- Example: Tungsten W  $(\gamma, n)$

## Summary, interactions 1

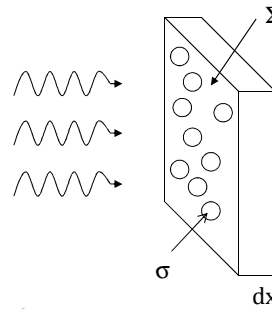


## Summary, interactions 2



## Attenuation coefficients 1

- $n_V$  atoms per volume =  $\rho(N_A/A)$
- Number of atoms:  
 $n = n_V V = n_V \Sigma dx$
- Interaction probability  
 $p = n\sigma / \Sigma = n_V \sigma dx$
- Probability per unit length:  
 $\mu = p/dx = n_V \sigma = \rho(N_A/A)\sigma$   
 $\mu$ : linear attenuation coefficient



## Attenuation coefficients 2

- $N_A$ : Avogadro's constant;  $6.022 \times 10^{23} \text{ mole}^{-1}$
- $A$ : number of grams per mole
- $N_A/A$ : number of atoms per gram
- $N_A Z/A$ : number of electrons per gram
- Number of atoms per volume:  $r(N_A/A)$
- Etc.

## Attenuation coefficients 3

- Total *mass* attenuation coefficient:

$$\frac{\mu}{\rho} = \frac{\tau}{\rho} + \frac{\sigma}{\rho} + \frac{\kappa}{\rho} + \frac{\sigma_R}{\rho}$$

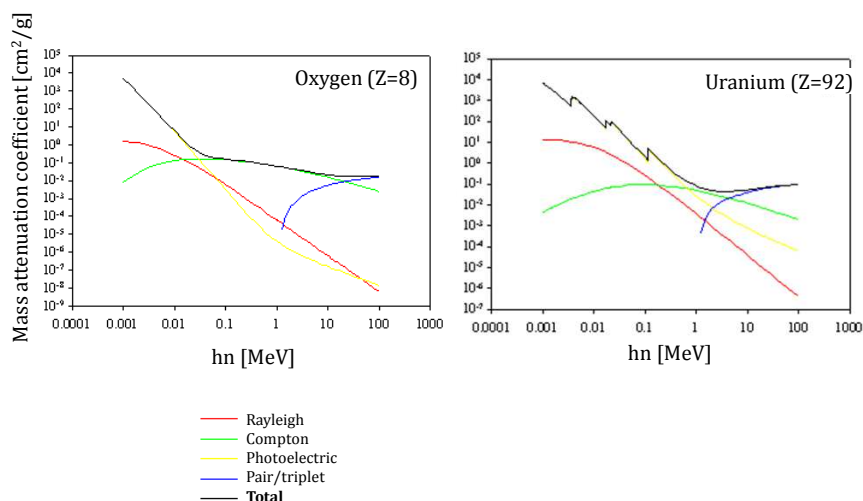
- Coefficient for energy transfer:

$$\frac{\mu_{tr}}{\rho} = \frac{\mu}{\rho} \frac{\bar{T}}{h\nu}$$

- Braggs rule for mixture of atoms:

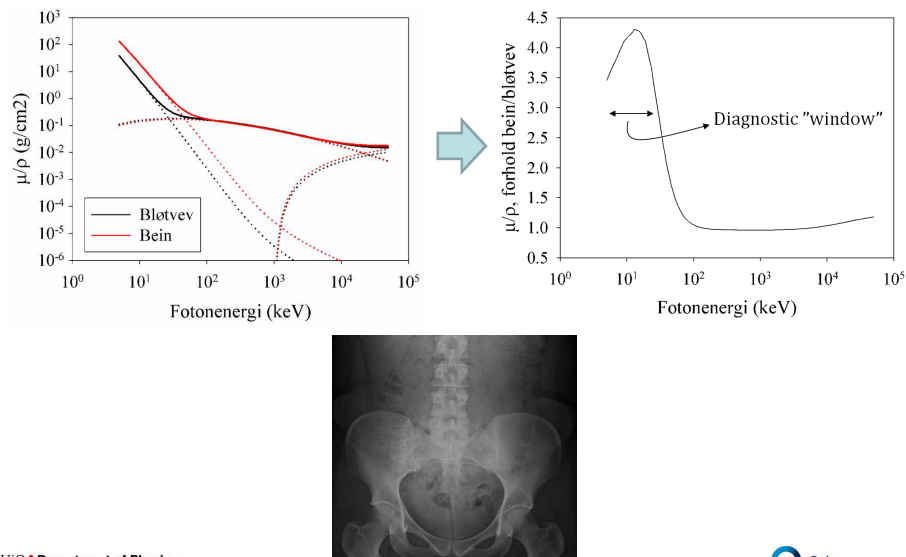
$$\left(\frac{\mu}{\rho}\right)_{mix} = \sum_{i=1}^n f_i \left(\frac{\mu}{\rho}\right)_i, \quad f_i = \frac{m_i}{\sum_{i=1}^n m_i}$$

## Attenuation coefficients 4





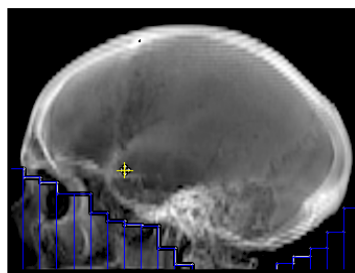
## Attenuation coefficients 5



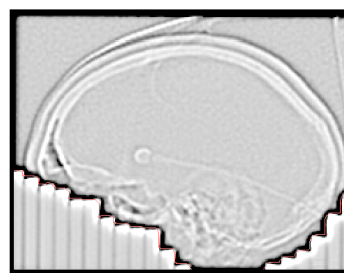
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## X-ray images at different energies



Conventional X-rays (120 kV)



Linear accelerator (5 MV)

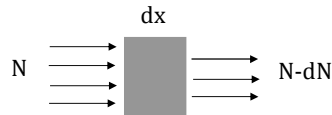


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## Attenuation 1

- Beam with  $N$  photons impinge absorber with thickness  $dx$ :



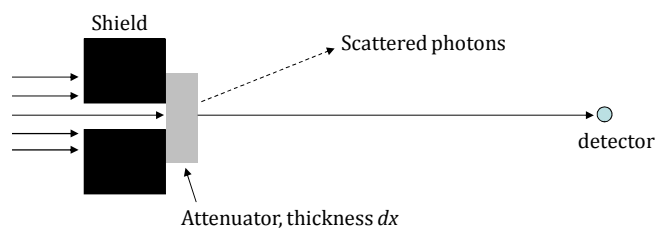
- Probability for interaction:  $\mu dx$
- Number of photons interacting:  $N\mu dx$

$$dN = N\mu dx \quad \Rightarrow \quad \int \frac{dN}{N} = \int \mu dx$$

$$\Rightarrow \underline{\underline{N = N_0 e^{-\mu x}}}$$

## Attenuation 2

- Note that  $\mu$  is the *interaction probability* per unit length – *not* the *absorption* probability
- $e^{-\mu x}$  corresponds to a narrow beam measurement geometry:



## Attenuation 3

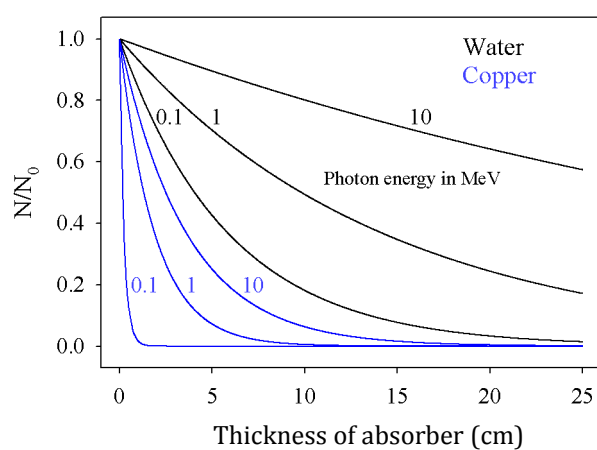
- 'Probability' for photon not interacting:  $e^{-\mu x}$
- Normalized probability

$$p_{ni} = Ce^{-\mu x}, \int_0^{\infty} p_{ni} dx = 1, \Rightarrow p_{ni} = \mu e^{-\mu x}$$

- Mean free path:

$$\langle x \rangle = \int_0^{\infty} x p_{ni} dx = \int_0^{\infty} x \mu e^{-\mu x} dx = \frac{1}{\mu}$$

## Attenuation 4



## Attenuation 4

- 'Probability' for photon not interacting:  $\sim e^{-\mu x}$
- Normalized probability

$$p_{ni} = Ce^{-\mu x}, \int_0^{\infty} p_{ni} dx = 1, \Rightarrow p_{ni} = \mu e^{-\mu x}$$

- Mean free path:

$$\langle x \rangle = \int_0^{\infty} x p_{ni} dx = \int_0^{\infty} x \mu e^{-\mu x} dx = \frac{1}{\mu}$$

- A distance of 3 MFP reduces the beam intensity to 5%

## Attenuation - example

- 2 MeV photons

$$\text{Pb:} \quad \mu = 0,516 \text{ cm}^{-1}$$

$$\text{H}_2\text{O:} \quad \mu = 0,049 \text{ cm}^{-1}$$

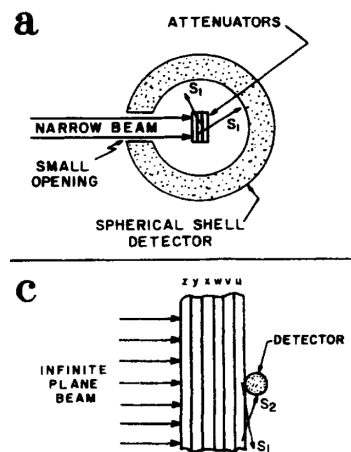
$$e^{-\mu_{\text{H}_2\text{O}} x_{\text{H}_2\text{O}}} = e^{-\mu_{\text{Pb}} x_{\text{Pb}}}$$

$$\Rightarrow \frac{x_{\text{H}_2\text{O}}}{x_{\text{Pb}}} = \frac{\mu_{\text{Pb}}}{\mu_{\text{H}_2\text{O}}}$$

- 10 times as much water necessary

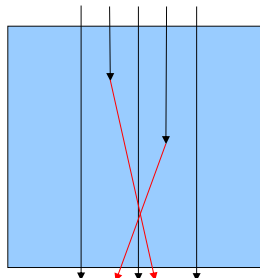
## Broad beam attenuation

- Broad-beam geometry: every scattered or secondary uncharged particle strikes the detector



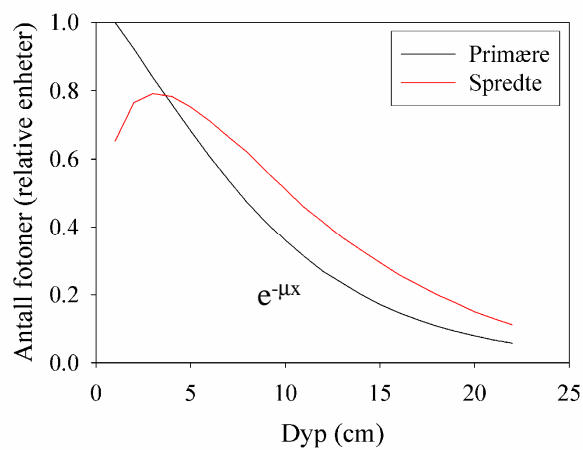
## Scattered photons

- $e^{-\mu x}$  : number of primary photons at a given depth
- What about the scattered photons?



- Monte Carlo simulations

## Primary and scattered photons, 100 keV



## Primary and scattered photons, 1 MeV

