

物理实验教学中心

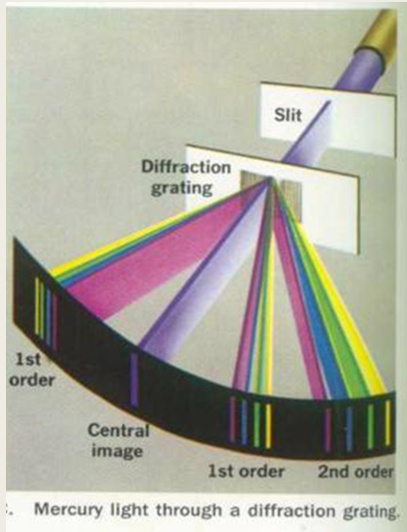
Physics Experiment Center



QUALITATIVE STUDY OF ATOMIC SPECTRA

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2024 Autumn

I. Diffraction grating



diffraction equation:

$$d \sin \theta = k \lambda, k=0, \pm 1, \pm 2, \dots$$

d : diffraction factor

θ : diffraction angle

k : order

λ : wavelength

II. The adjustment of spectrometer

➤ Final vision: see right

➤ **Methods:**

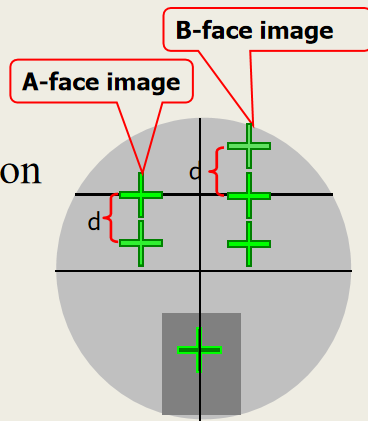
❖ Three adjusting screw button under the loading platform

❖ The telescope elevation adjusting screw

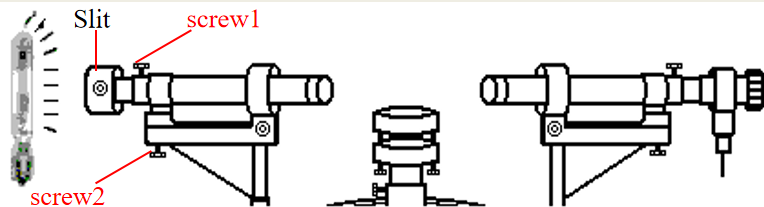
➤ **Steps:**

❖ Coarse adjustment

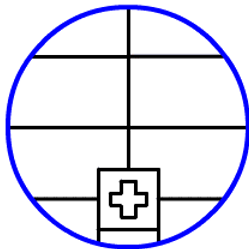
❖ Fine adjustment



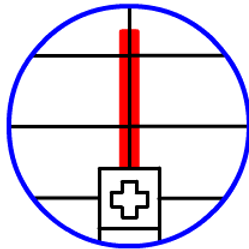
collimator



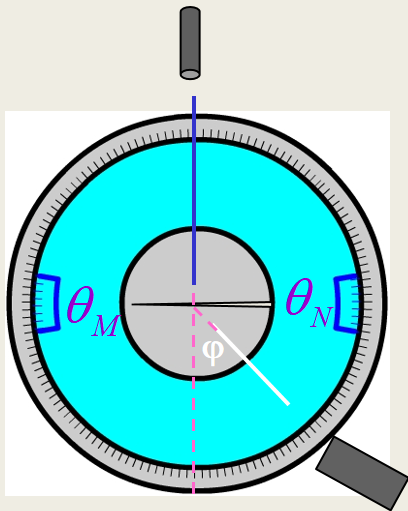
①



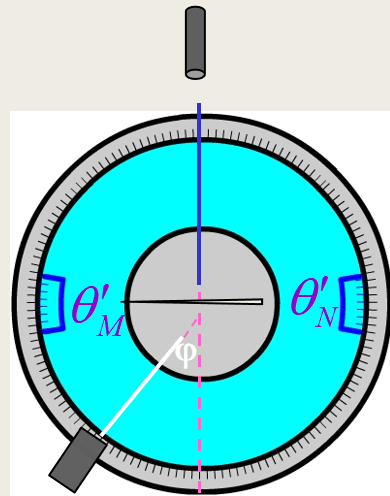
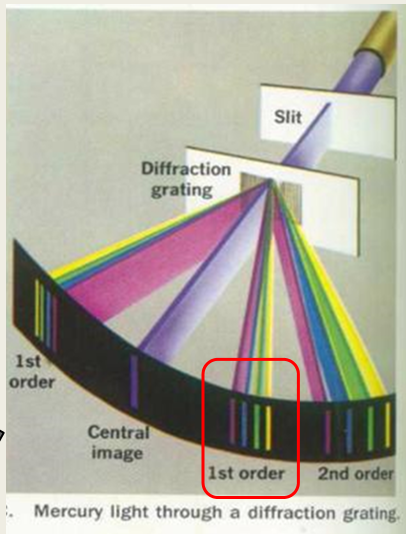
②



Measurement of the first-order diffraction angles

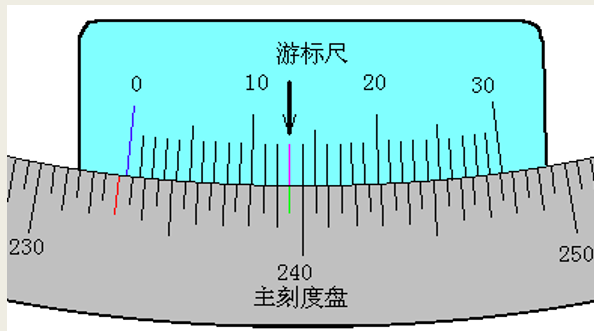


Right side



Left side

Read the angle



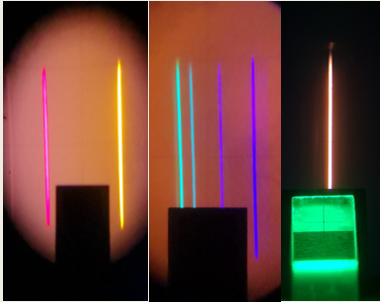
233°13'

IV. Experiment contents

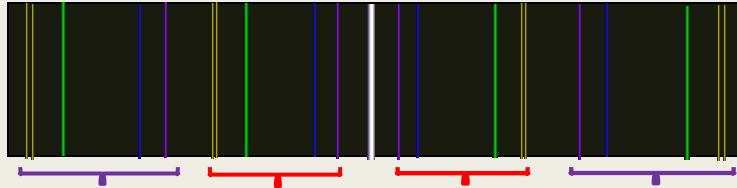
1. Measure the first-order diffraction angle of helium, determine the diffraction factor d according to the known helium spectral wavelength. Note that $d \cdot \sin \phi = k \cdot \lambda (k=1)$

Color	Wavelength (nm)	Color	Wavelength (nm)	Color	Wavelength (nm)
Red(dark)	706.52	Green(light)	504.77	Blue	471.31
Red	667.82	Green	501.57	Purple	447.15
Yellow	587.56	Cyan	492.19	Purple (dark)	438.79

2. Measure the first-order diffraction angle of mercury lamp, find out the corresponding wavelength(λ) of mercury according to the diffraction formula $d \cdot \sin \phi = k \cdot \lambda (k=1)$.



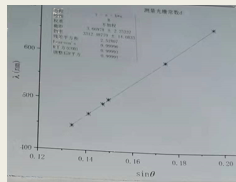
Spectrum of helium



Spectrum of mercury

V. Original data

Table 1. Helium lamp data



$$d \cdot \sin \varphi = k \cdot \lambda, (k=1)$$

Color	$\lambda(\text{nm})$	θ_M	θ_N	θ'_M	θ'_N	$\varphi = \frac{1}{4}(\theta'_M - \theta_M + \theta'_N - \theta_N)$	$\sin \varphi$
Red	667.82	264°42'	84°46'	241°34'	61°39'		
Yellow	587.56	263°24'	83°28'	243°4'	63°8'		
Green	501.57	261°49'	81°53'	244°34'	64°38'		
Cyan	492.19	261°41'	81°45'	244°44'	64°48'		
Blue	471.31	261°21'	81°25'	245°7'	65°10'		
Purple	447.15	260°56'	80°59'	245°31'	65°35'		

Plot $\sin \varphi - \lambda$ curve for Helium.

The **slope** of the curve is d , $d = (\lambda_{\text{Red}} - \lambda_{\text{Purple}}) / (\sin \varphi_{\text{Red}} - \sin \varphi_{\text{Purple}})$

Table 2. Mercury lamp data

Color	θ_M	θ_N	θ'_M	θ'_N	$\varphi = \frac{1}{4} (\theta'_M - \theta_M + \theta'_N - \theta_N)$	$\sin \varphi$	$\lambda(nm)$
Orange	272°35'	92°33'	252°37'	72°34'			
Yellow	272°33'	92°31'	252°38'	72°36'			
Green	272°3'	92°2'	253°11'	73°9'			
Purple	270°6'	90°4'	255°6'	75°4'			

Using $d \cdot \sin \varphi = k \cdot \lambda$, ($k=1$) to calculate λ for each color.

Note that d is obtained from page 11.

Homework

- Please complete the report to describe this experiment, plot $\sin\phi-\lambda$ curve for Helium, complete the tables.
- DL: September 26, 2024.

- Useful link(s):
 1. <https://github.com/bliseu/phylab>

END