### **Title Page**

Include the following information in the order given:

- descriptive title
- course # and section #
- author's name
- partner's name
- date(s) performed

#### Abstract

- · identify the sample or reaction being analyzed
- state what is being determined
- identify the analytical method(s) used
- summarize the results. For example: "The average mass loss was determined to be 54.0%, with a standard deviation of 5.2%".

#### **Procedure**

• Sufficient detail and clarity for someone else to repeat the experiment and results.

#### Data & Results

- **Equation 1** Show the balanced chemical equation for the reaction between hydrochloric acid (HCl) and calcium carbonate (CaCO<sub>3</sub>).
- Table 1a Mass data
- Table 1b % Mass Loss Results

### Discussion

- Discuss the experimental % mass loss. Does the average +/- one standard deviation agree with the theoretical % mass loss of 43.97%?
- Discuss sources of error and specific ways to reduce those errors. Avoid talking about "human" errors.

# **Calculations**

Experimental % mass loss

# **Overall Format**

• Same as before