

A method for redesigning molecular dynamics force field parameterization by use of a Bayesian statistical framework

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I. Objectives

Molecular dynamics (MD) simulation is fast becoming a more useful tool in many scientific studies. However, some limitations remain in the ability of MD force fields to accurately and transferably describe molecular environments. Many popular and currently used force fields were parameterized with fixed functional forms which, often, have poor physical motivation. The chemical intuition of experts is also often required to manually correct parameters, leading to a more suitable product. Additionally, the creation of a transferable method to update existing force fields based on new experimental data is limited due to lack of understanding and lack of consistency in how the original parameterizations were done.

A possible solution to these problems is by recasting the force field parameterization process as a Bayesian inference problem. The objective of this paper is to introduce a framework for using high quality experimental data in order to automatically generate families of MD force fields consistent with the data used. In this paper I will generally describe the overall parameterization framework and my roles in the project thus far. First, collecting and curating large amounts of high quality experimental thermochemical data and, currently, investigating use of the Multistate Bennett Acceptance Ratio (MBAR) as a means to improve parameterization throughput by reducing computational expense while making updates to the posterior distribution of parameter sets consistent with experimental data provided.

II. Significance

A broad variety of research from drug discovery to metallurgy has been greatly impacted by the advent and improvement of MD simulation tools. Observing physical phenomena such as protein folding dynamics and ligand docking at a molecular scale is widely studied using MD tools.^{1,2} Drug discovery and design of new pharmaceutical leads has also been made more efficient.³ The fundamental part in molecular simulation for describing the energetic interactions of a system is referred to as a force field. Hence, the development of force fields which are readily transferable between dissimilar physical systems and are quantitatively accurate is imperative for the use of molecular simulation tools to continue to proliferate.

Transferability of MD force fields, and particularly sets of force field parameters, is an extremely popular topic (and current limitation) in the molecular simulation field.^{4;5;6;7} Transferability of force fields encourages use by providing convenience for scientists with wide arrays of research interests and by making parameter space less complex through generalization by chemical similarity. Inaccurate and poorly parameterized force fields have been shown to grossly misrepresent molecular systems.^{8;9;10}

A few notable attempts, such as GAAMP and ForceBalance, have been made in recent years towards the development of more automated and systematic force field parameterization methods.^{11;12;13;14} Each made important contributions to automated force field parameterization through clever use of objective function optimization, exploiting a variety of fitting data and allowing exploration of functional forms. However, none provided the ability for the computer to automatically and systematically explore choices of fitting data, optimization algorithm and functional forms in order to objectively find families of force fields consistent with fitting data and reward those with the least model complexity. The Bayesian inference scheme described in this paper will provide a workflow for discovering families of force field parameters consistent with experimental data and a variety of functional forms.

Additionally, as I will demonstrate later in my discussion of data mining and curation of the NIST ThermoML database, the chemical diversity in readily available thermochemical databases is lacking. Not only that, but the distribution of data amongst commonly measured properties is heavily skewed towards certain properties. Having learned this since beginning the project, one of the potential uses of the parameterization scheme is to fill in the many gaps in experimental thermochemical data. With updated general descriptions of chemical space and property data on

chemically similar compounds, this parameterization process should provide force fields to accurately simulate property data for which no experimental data exists.

III. Background and related literature (1.5pages \pm 0.5pages)

Molecular dynamics force fields define how to construct the potential energy functions (and thereby the forces) of an atomistic system under study. The potential is constructed such that it is a function of solely the atomic coordinates and a set of parameters associated with the force field. Transferable force fields generally have three major parts:

- 1 The **functional forms** of the potential, i.e. the mathematical equations for the energy equation. A classic example of a non-bonded interaction form is the 12-6 Lennard-Jones (LJ) potential.
- 2 **Atom types** which describe similar chemical environments such that one can assign different atoms (or series of atoms) identical parameters, thereby shrinking the parameter space and helping to avoid overfitting.
- 3 **Parameters** that are associated with one or many atom types which determine the magnitude of the interactions in the system

Rolled into functional forms, **combining rules** are also sometimes considered. **Combining rules** describe how to combine parameters when an interaction contains multiple atom types.

There are severe limitations in current methods for force field parameterization. Until very recently, force fields have primarily been made manually, guided by experimental and quantum chemical simulation data as well as the intuition of expert computational chemists.^{15;16;17;18;19} Some functional forms used in modern force fields, like the 12-6 LJ potential, have poor physical basis. While the attractive term of the LJ potential has physical basis on the true behavior of dispersion forces, the repulsive term loosely approximates Pauli repulsion and is used for computational convenience. Despite attempts at improvement, many of the functional forms and parameters of popular force fields remain mostly unchanged due to the lack of clear, systematic methods for updating them.²⁰

Parameterization methods have slowly become more sophisticated over the last decade and a half with advances in computational power and to accommodate modeling increasingly more complex systems. Many early force fields were parameterized manually for narrow classes of molecules with large redundant parameter spaces.²¹ Force fields like AMBER *parm94* showed intuitive departure by shrinking parameter space with clever atom typing defined by expert computational chemists.²² The parameterization of GAFF used a semi-automated genetic algorithm approach to select parameters.¹⁹ Even more sophisticated optimization approaches such as least-squares optimization of an objective function have been utilized in the creation of the TIP4P-Ew water model²³ and in the ForceBalance parameterization scheme^{12;13;14}. Even with these more sophisticated optimization schemes there are still issues in needing for the user to assign weights to different kinds of data (i.e. different properties) when they are included in the same objective function. Molecular systems aren't necessarily uniquely defined by a single parameter set. There are possibilities of multiple optima in parameter space (i.e. different sets of parameters that all are consistent with data used during parameterization) and least-squares optimization does not discriminate the global optima from the other possibilities.

Bayesian inference provides a robust statistical framework for force field parameterization. It has been shown that bayesian approaches can be applied to a wide variety of data driven sciences. It's been used for balancing data to help minimize influence of oversampled populations and generate more robust predictive models²⁴ to recalibrating initial force estimates in coarse grained MD models to target atomistic MD and experimental data²⁵. Baye's theorem clearly provides a framework for the problem at hand thusly:

$$P(\theta|D) \propto P(D|\theta) P(\theta) \quad (1)$$

In **Equation (1)**, consider a model M (including functional forms and atom types) with some unknown set of parameters which produced data D . θ is a choice of parameters consistent with data D . What **equation (1)** states is that the probability of θ given D (the *posterior*) can be determined from the probability of observing D given θ (the *likelihood function*) and the probability of θ (the *prior*). The *prior* is imposed by physical constraint or by the previous round of inference. Note that in iterative bayesian inference, the posterior of the previous round becomes the prior in the new iteration. This bayesian inference produces not just a single parameter set, but an entire posterior distribution of parameters given data. This is advantageous given that many different parameter

sets can be consistent with the data used and the distribution of these consistent sets of parameters can inform what new data could help narrow the distribution and improve the parameter estimates.

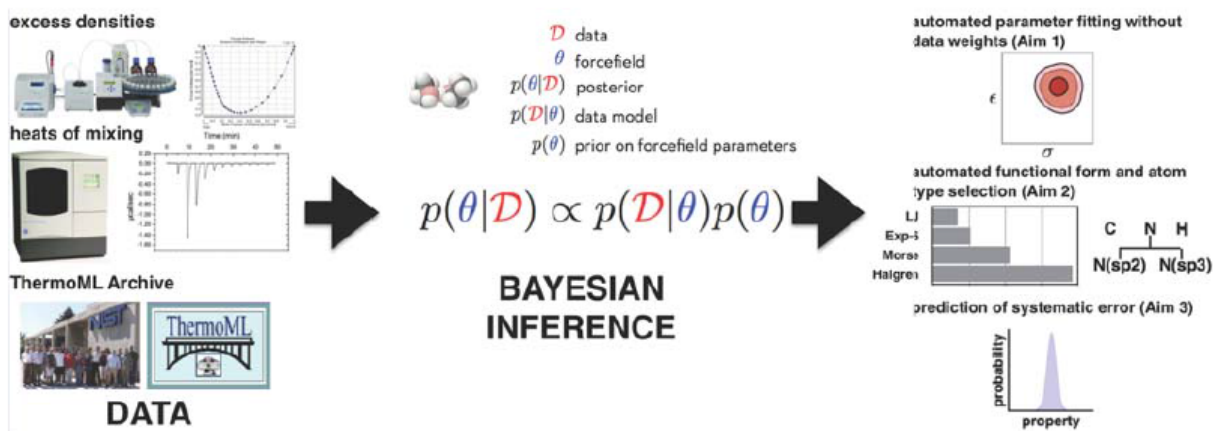


FIG. 1: A schematic overview of the bayesian inference workflow where force field parameters are inferred given experimental data and a model

A. Methods (1.5pages ± 0.5pages)

- Data Mining and curation from ThermoML
 - ThermoPyL
 - Filtering algorithm(s)
 - Some stats on total database
 - Some of the different tests/sets we searched for to check size of chemical environment
 - AlkEthOH
- Reweighting workflow to determine length of jumps allowable in parameter space
 - Purpose of reweighting workflow
 - Brief overview of MBAR maybe
 - Criteria for safe jump and motivation behind that
 - Simulations ran

All the stats from the data mining shit

Potential figure from results of jump tests

B. Progress (1.5pages ± 0.5pages)

C. Research plan (0.5pages)

References

- [1] G. Jayachandran, V. Vishal, and V. S. Pande, "Using massively parallel simulation and Markovian models to study protein folding: Examining the dynamics of the villin headpiece," *J Chem Phys*, vol. 124, no. 16, p. 164902, Apr. 2006. [Online]. Available: <http://scitation.aip.org/content/aip/journal/jcp/124/16/10.1063/1.2186317>
- [2] K. A. Beauchamp, D. L. Ensign, R. Das, and V. S. Pande, "Quantitative comparison of villin headpiece subdomain simulations and triplet-triplet energy transfer experiments," *Proc. Natl. Acad. Sci. U.S.A.*, vol. 108, no. 31, pp. 12 734–12 739, Aug. 2011.

- [3] M. De Vivo, M. Masetti, G. Bottegoni, and A. Cavalli, "Role of Molecular Dynamics and Related Methods in Drug Discovery," *J. Med. Chem.*, vol. 59, no. 9, pp. 4035–4061, May 2016. [Online]. Available: <http://dx.doi.org/10.1021/acs.jmedchem.5b01684>
- [4] N. A. Vellore, J. A. Yancey, G. Collier, R. A. Latour, and S. J. Stuart, "Assessment of the Transferability of a Protein Force Field for the Simulation of Peptide-Surface Interactions," *Langmuir*, vol. 26, no. 10, pp. 7396–7404, May 2010. [Online]. Available: <http://dx.doi.org/10.1021/la904415d>
- [5] D. A. Puleo and R. Bizios, *Biological Interactions on Materials Surfaces: Understanding and Controlling Protein, Cell, and Tissue Responses*. Springer Science & Business Media, Jun. 2009.
- [6] F. Sato, S. Hojo, and H. Sun, "On the Transferability of Force Field Parameters With an ab Initio Force Field Developed for Sulfonamides," *J. Phys. Chem. A*, vol. 107, no. 2, pp. 248–257, Jan. 2003. [Online]. Available: <http://dx.doi.org/10.1021/jp026612i>
- [7] A. Martin-Calvo, J. J. Gutiérrez-Sevillano, J. B. Parra, C. O. Ania, and S. Calero, "Transferable force fields for adsorption of small gases in zeolites," *Phys Chem Chem Phys*, vol. 17, no. 37, pp. 24 048–24 055, Oct. 2015.
- [8] O. F. Lange, D. van der Spoel, and B. L. de Groot, "Scrutinizing Molecular Mechanics Force Fields on the Submicrosecond Timescale with NMR Data," *Biophys J*, vol. 99, no. 2, pp. 647–655, Jul. 2010. [Online]. Available: <http://www.ncbi.nlm.nih.gov/pmc/articles/PMC2905107/>
- [9] F. Martín-García, E. Papaleo, P. Gomez-Puertas, W. Boomsma, and K. Lindorff-Larsen, "Comparing Molecular Dynamics Force Fields in the Essential Subspace," *PLoS One*, vol. 10, no. 3, Mar. 2015. [Online]. Available: <http://www.ncbi.nlm.nih.gov/pmc/articles/PMC4374674/>
- [10] K. Vanommeslaeghe, M. Yang, and A. D. MacKerell, "Robustness in the fitting of molecular mechanics parameters," *J. Comput. Chem.*, vol. 36, no. 14, pp. 1083–1101, May 2015. [Online]. Available: <http://onlinelibrary.wiley.com/doi/10.1002/jcc.23897/abstract>
- [11] L. Huang and B. Roux, "Automated Force Field Parameterization for Nonpolarizable and Polarizable Atomic Models Based on Ab Initio Target Data," *J. Chem. Theory Comput.*, vol. 9, no. 8, pp. 3543–3556, Aug. 2013. [Online]. Available: <http://dx.doi.org/10.1021/ct4003477>
- [12] L.-P. Wang, T. J. Martinez, and V. S. Pande, "Building Force Fields: An Automatic, Systematic, and Reproducible Approach," *J. Phys. Chem. Lett.*, vol. 5, no. 11, pp. 1885–1891, Jun. 2014. [Online]. Available: <http://dx.doi.org/10.1021/jz500737m>
- [13] L.-P. Wang, T. Head-Gordon, J. W. Ponder, P. Ren, J. D. Chodera, P. K. Eastman, T. J. Martinez, and V. S. Pande, "Systematic Improvement of a Classical Molecular Model of Water," *J. Phys. Chem. B*, vol. 117, no. 34, pp. 9956–9972, Aug. 2013. [Online]. Available: <http://dx.doi.org/10.1021/jp403802c>
- [14] L.-P. Wang, J. Chen, and T. Van Voorhis, "Systematic Parametrization of Polarizable Force Fields from Quantum Chemistry Data," *J. Chem. Theory Comput.*, vol. 9, no. 1, pp. 452–460, Jan. 2013. [Online]. Available: <http://dx.doi.org/10.1021/ct300826t>
- [15] B. R. Brooks, R. E. Bruccoleri, B. D. Olafson, D. J. States, S. Swaminathan, and M. Karplus, "CHARMM: A program for macromolecular energy, minimization, and dynamics calculations," *J. Comput. Chem.*, vol. 4, no. 2, pp. 187–217, Jun. 1983. [Online]. Available: <http://onlinelibrary.wiley.com/doi/10.1002/jcc.540040211/abstract>
- [16] A. D. MacKerell, D. Bashford, M. Bellott, R. L. Dunbrack, J. D. Evanseck, M. J. Field, S. Fischer, J. Gao, H. Guo, S. Ha, D. Joseph-McCarthy, L. Kuchnir, K. Kuczera, F. T. K. Lau, C. Mattos, S. Michnick, T. Ngo, D. T. Nguyen, B. Prodhom, W. E. Reiher, B. Roux, M. Schlenkrich, J. C. Smith, R. Stote, J. Straub, M. Watanabe, J. WiÅkiewicz-Kuczera, D. Yin, and M. Karplus, "All-Atom Empirical Potential for Molecular Modeling and Dynamics Studies of Proteins," *J. Phys. Chem. B*, vol. 102, no. 18, pp. 3586–3616, Apr. 1998. [Online]. Available: <http://dx.doi.org/10.1021/jp973084f>

- [17] G. C. Soo, F. K. Cartledge, R. J. Unwalla, and S. Profeta, "Development of a molecular mechanics (MM2) force field for $\dot{\text{I}}\dot{\text{S}}$ -chlorosilanes," *Tetrahedron*, vol. 46, no. 24, pp. 8005–8018, Jan. 1990. [Online]. Available: <http://www.sciencedirect.com/science/article/pii/S0040402001814575>
- [18] T. A. Halgren, "Merck molecular force field. I. Basis, form, scope, parameterization, and performance of MMFF94," *J. Comput. Chem.*, vol. 17, no. 5-6, pp. 490–519, Apr. 1996. [Online]. Available: [http://onlinelibrary.wiley.com/doi/10.1002/\(SICI\)1096-987X\(199604\)17:5/6<490::AID-JCC1>3.0.CO;2-P/abstract](http://onlinelibrary.wiley.com/doi/10.1002/(SICI)1096-987X(199604)17:5/6<490::AID-JCC1>3.0.CO;2-P/abstract)
- [19] J. Wang, R. M. Wolf, J. W. Caldwell, P. A. Kollman, and D. A. Case, "Development and testing of a general amber force field," *J. Comput. Chem.*, vol. 25, no. 9, pp. 1157–1174, Jul. 2004. [Online]. Available: <http://onlinelibrary.wiley.com/doi/10.1002/jcc.20035/abstract>
- [20] J. W. Ponder and D. A. Case, "Force fields for protein simulations," *Adv. Protein Chem.*, vol. 66, pp. 27–85, 2003.
- [21] N. L. Allinger, M. T. Tribble, M. A. Miller, and D. H. Wertz, "Conformational analysis. LXIX. Improved force field for the calculation of the structures and energies of hydrocarbons," *J. Am. Chem. Soc.*, vol. 93, no. 7, pp. 1637–1648, Apr. 1971. [Online]. Available: <http://dx.doi.org/10.1021/ja00736a012>
- [22] W. D. Cornell, P. Cieplak, C. I. Bayly, I. R. Gould, K. M. Merz, D. M. Ferguson, D. C. Spellmeyer, T. Fox, J. W. Caldwell, and P. A. Kollman, "A Second Generation Force Field for the Simulation of Proteins, Nucleic Acids, and Organic Molecules," *J. Am. Chem. Soc.*, vol. 117, no. 19, pp. 5179–5197, May 1995. [Online]. Available: <http://dx.doi.org/10.1021/ja00124a002>
- [23] H. W. Horn, W. C. Swope, J. W. Pitera, J. D. Madura, T. J. Dick, G. L. Hura, and T. Head-Gordon, "Development of an improved four-site water model for biomolecular simulations: TIP4p-Ew," *The Journal of Chemical Physics*, vol. 120, no. 20, pp. 9665–9678, May 2004. [Online]. Available: <http://scitation.aip.org/content/aip/journal/jcp/120/20/10.1063/1.1683075>
- [24] K. Klein, S. Hennig, and S. K. Paul, "A Bayesian Modelling Approach with Balancing Informative Prior for Analysing Imbalanced Data," *PLOS ONE*, vol. 11, no. 4, p. e0152700, Apr. 2016. [Online]. Available: <http://journals.plos.org/plosone/article?id=10.1371/journal.pone.0152700>
- [25] P. N. Patrone, T. W. Rosch, and F. R. P. Jr, "Bayesian calibration of coarse-grained forces: Efficiently addressing transferability," *The Journal of Chemical Physics*, vol. 144, no. 15, p. 154101, Apr. 2016. [Online]. Available: <http://scitation.aip.org/content/aip/journal/jcp/144/15/10.1063/1.4945380>