

Module 2: Western Blot

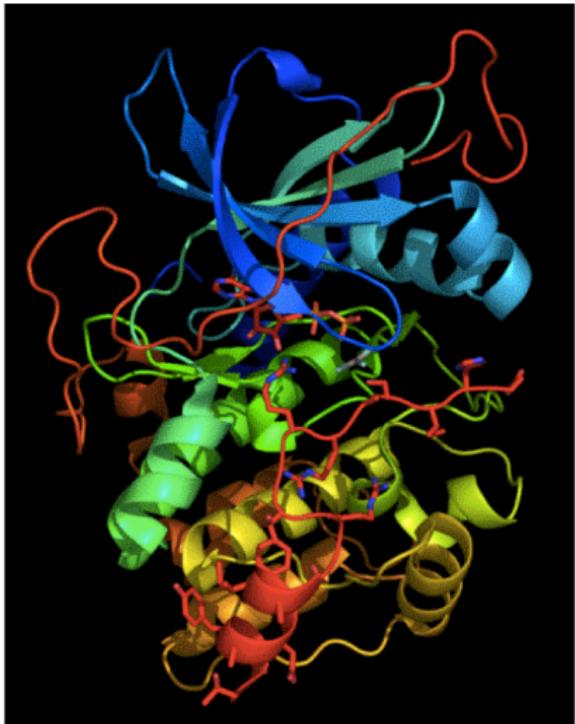
BMES Cell Team

Fall 2020



Outline

- Protein Basics
- What is a Western Blot?
- Western Blot Protocol
 - BCA Assay
 - SDS Gel Electrophoresis
 - Immunoblotting
- Western Blot Video
- Pipetting basics



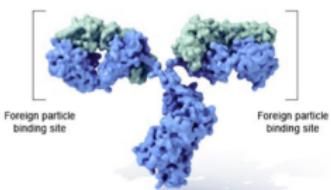
Proteins

- **Definition:** Proteins are macromolecules made of amino acids.

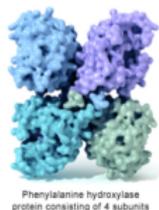
→ Chains of amino acids make up proteins

- Proteins have a wide structural range → large functional range
- Key players in organism's metabolic & regulatory activity

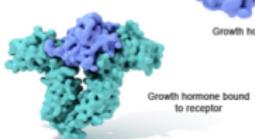
Immunoglobulin G (IgG)



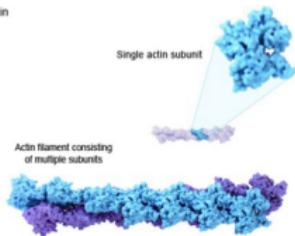
Phenylalanine hydroxylase



Growth hormone



Actin

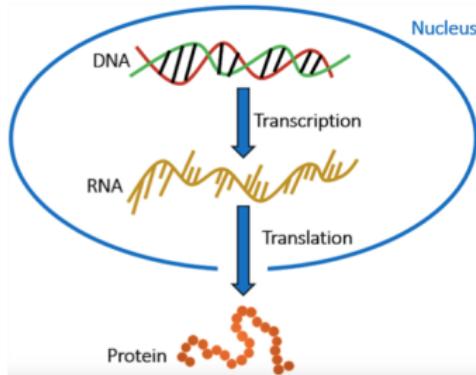


U.S. National Library of Medicine

What is Western Blotting?

- **Definition:** A **Western Blot** separates and identifies target proteins.

- Two stage procedure confirms protein presence and quantifies target
 - Proteins separated by size through gel electrophoresis
 - Target proteins visualized through immunostaining

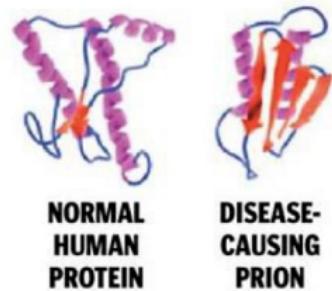


S N O W
D R O P

S - SOUTHERN	- DNA	- D
N - NORTHERN	- RNA	- R
O - OOOOOOOO	- OOOO	- O
W - WESTERN	- PROTEIN	- P

Western Blot Utility

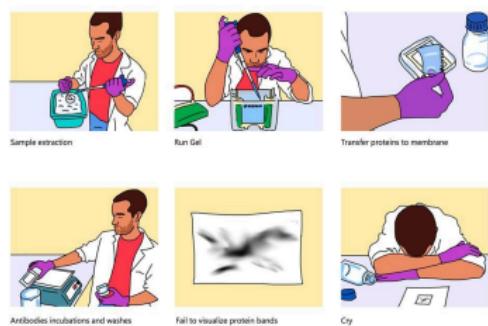
- Understand mechanisms of cell behavior
 - Protein's primary function in organism
- Probe for a specific disease
 - **Viral:**
 - HIV
 - **Bacterial:**
 - Melioidosis
 - **Prion:**
 - Creutzfelt-Jakob disease



Basic Protocol for a Western Blot

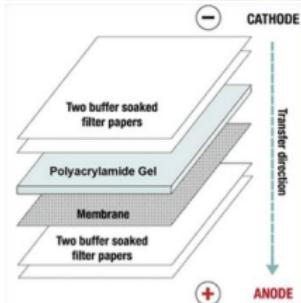
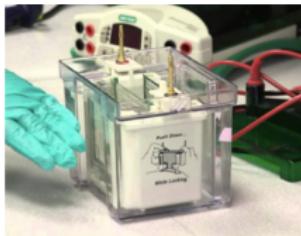
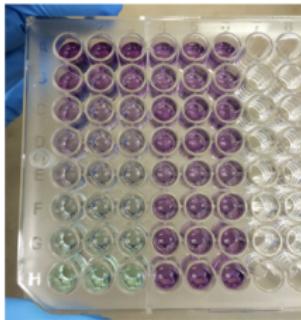
1. Lyse cells and collect proteins
2. BCA Assay
 - Calculate sample protein concentration
3. SDS Gel Electrophoresis
 - Separate proteins by size
4. Transfer proteins to membrane
5. Immunoblot target proteins
 - Antibody Binding
6. Image membrane

Western Blot Protocol



Basic Protocol for a Western Blot

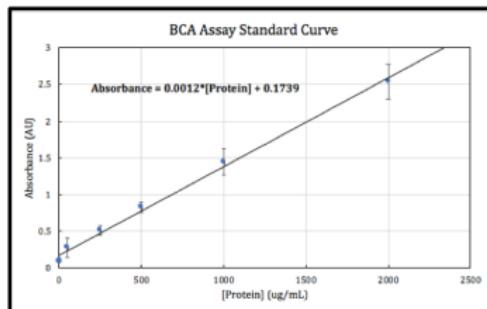
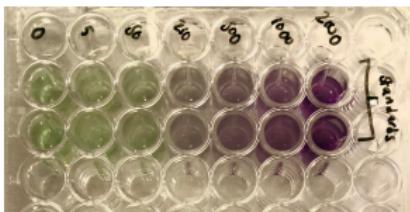
1. Lyse cells and collect proteins
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BCA Assay

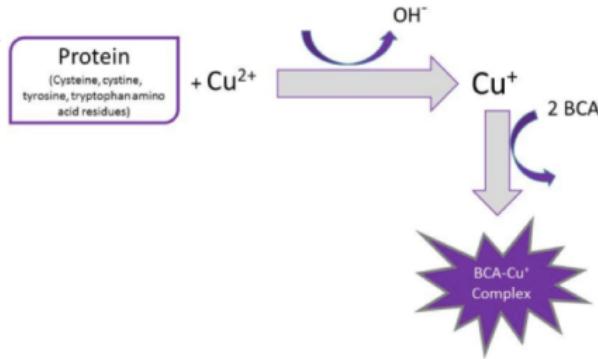
- **Definition:** A bicinchoninic acid assay (**BCA Assay**) uses absorbance readings to determine total protein concentration in a sample.

- First, create a **standard curve**
 - Concentration of protein in each well is known
- Then, add experimental samples
 - Concentration of protein is unknown
- Use standard curve absorbance readings to correlate protein concentration to absorbance
- Use standard curve to calculate experimental protein concentration



How does a BCA Assay work?

- Step 1: Biuret Reaction
 - **Green** Cu²⁺ in BCA reagent binds to sample protein
→ reduction of cuprous Cu¹⁺
- Step 2: BCA and Copper Chelation
 - 2 BCA molecules bind to Cu¹⁺
→ **purple** chelated complex
- Step 3: Measure Absorbance
 - Purple complex absorbs maximally at 562 nm
 - Absorbance \propto purple complexes \propto peptide

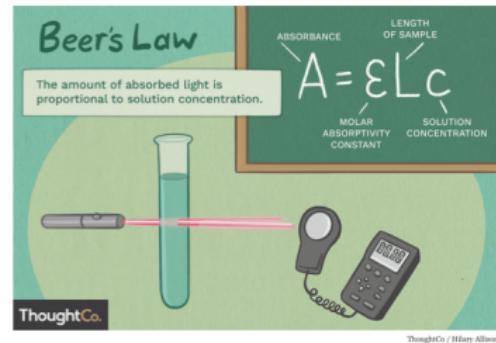


Beer's Law

- **Definition:** Beer's Law relates a sample's absorbance reading to total protein concentration.

$$A = \epsilon \cdot L \cdot C$$

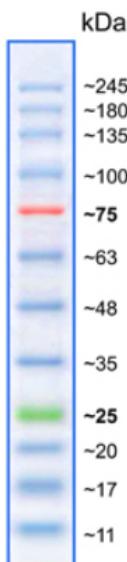
- A = absorbance reading from plate reader
- ϵ = molar absorptivity constant
- L = path length
- C = protein concentration
 - As ϵ and L are constant, there is a linear relationship between absorbance and protein concentration



SDS-PAGE

- **Definition:** Sodium dodecyl sulfate-polyacrylamide gel electrophoresis (**SDS-PAGE**) separates proteins by size.

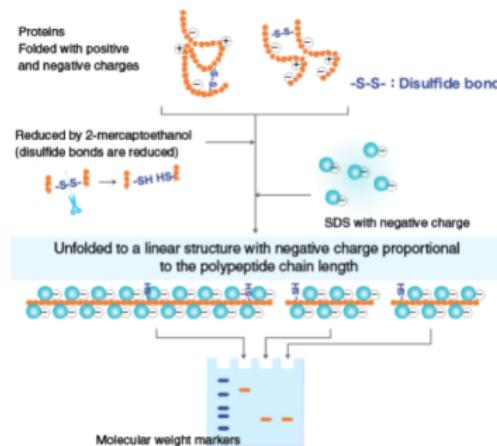
- First, load the protein ladder at the two ends of the well
 - Set of standards that allow us to estimate protein size
 - Dalton (Da) = atomic mass unit
 - $kDa = 1000 \text{ Da}$
- Then, load your sample into the central wells
- Run the gel and use the ladder to estimate protein size



How does SDS-PAGE work?

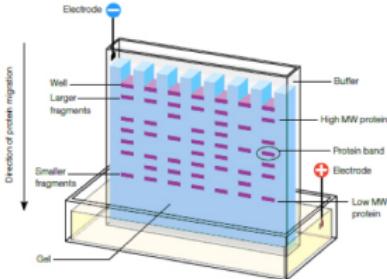
Key Reagents

- Sodium Dodecyl Sulfate (SDS)
 - Anionic detergent that binds to protein side chains → disruption noncovalent bonds → protein denaturation
 - Coats denatured protein in uniform negative charge → charge of protein \propto length of protein
- Beta-mercaptopethanol (β ME)
 - Reduces disulfide bonds in protein → disruption covalent bonds → protein linearization



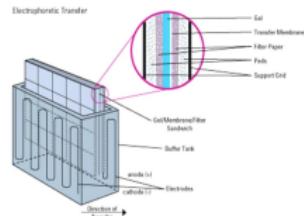
How does SDS-PAGE work?

- Polyacrylamide (PA)
 - Water soluble polymer
 - 3D networks of polyacrylamide → porous gel
 - Smaller proteins can travel faster through the porous gel
- Gel Electrophoresis (GE)
 - When placed in an electric field, the negatively charged proteins will migrate toward the positive electrode
 - Since smaller proteins can travel faster through a porous gel, loading protein samples into a gel then creating an electric field around the gel **separates proteins by size**



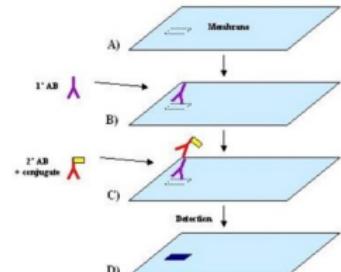
Protein Transfer

- Need to transfer proteins from gel to nitrocellulose membrane
 - Antibodies cannot bind to proteins when they are on the gel
 - Must transfer proteins onto a nitrocellulose membrane and retain the gel electrophoresis size sorting
- Transfer proteins using **electroblotting**
 - Align the gel and the membrane
 - Use an electric current to pull negatively charged proteins toward a positively charged anode and onto the membrane

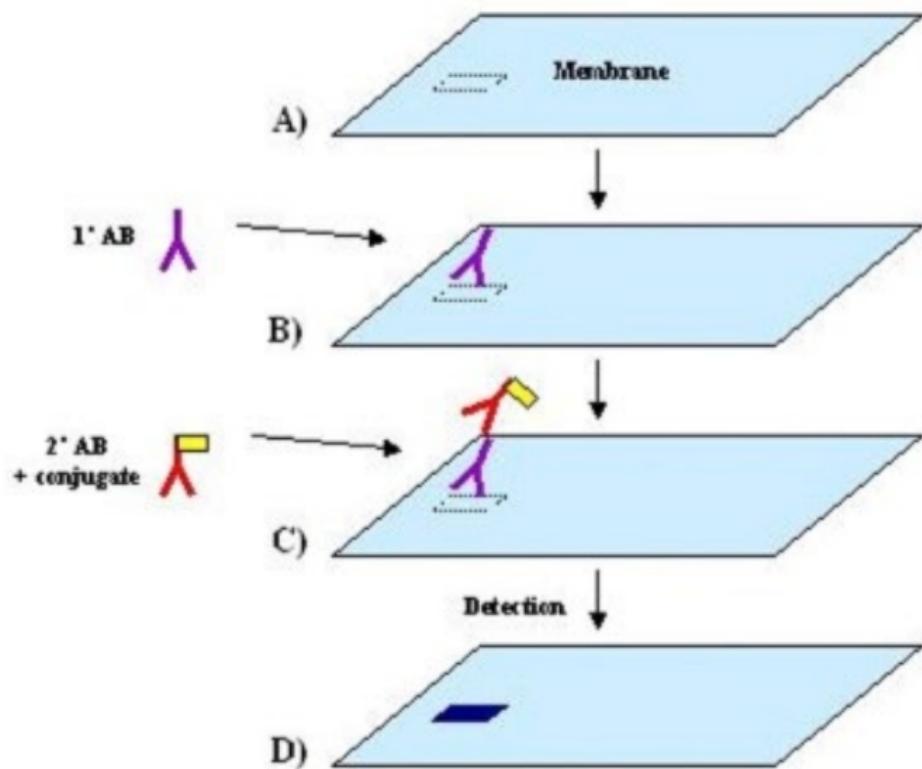


Immunoblotting (What Makes it a Western Blot!)

- **Definition:** Immunoblotting uses antibodies to identify proteins.
- Antibodies are proteins in the immune system that target specific antigens
- Primary Antibody: binds to target protein
 - Loading Control: Actin (constitutively expressed in all cells)
- Secondary antibody: binds to primary antibody and amplifies the signal
 - Primary and secondary antibodies must be from a different species than the target protein
 - If not, will have non-specific binding

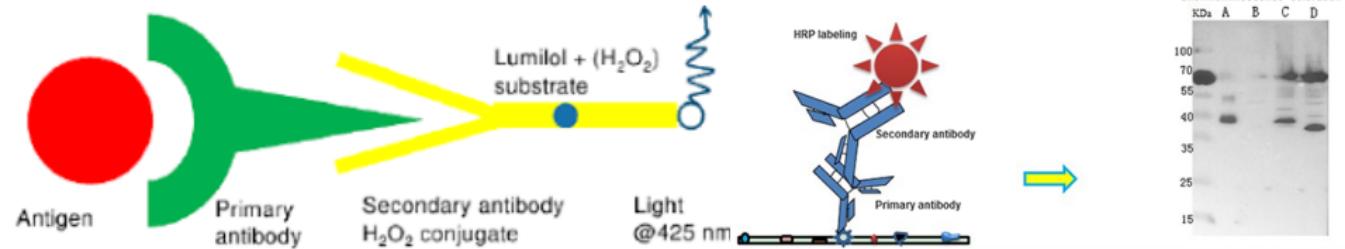


How does Immunoblotting work?



Imaging

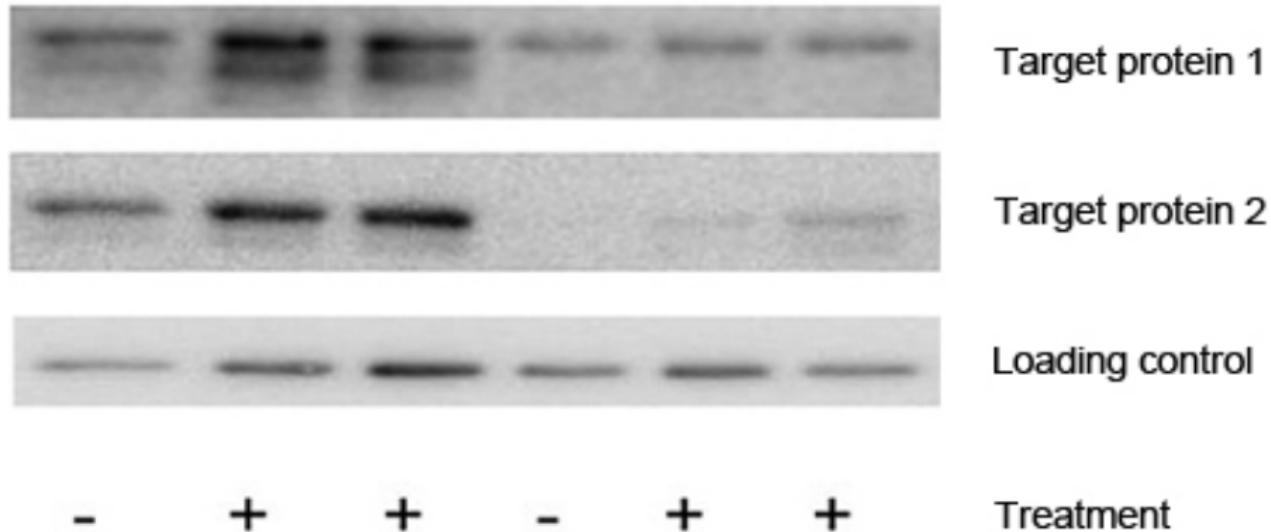
- Chemiluminescence
 - Chemical reaction between ECL substrate and HRP enzyme in secondary antibody
 - Releases energy as light
 - One of the easiest ways to understand internal cell behavior and controls



Interpreting a Western Blot

- Loading control band:
 - should be the same in all samples
 - If loading control is not the same, result is invalid
- Band position on gel:
 - Different sized proteins show up at different heights
 - Larger proteins show up closer to the original well position
- Band intensity:
 - The darker the band, the more protein is present

Interpreting a Western Blot



- Loading control band is the same in all trials
- Greater target protein intensity in wells 2 and 3

Western Blot Video



 Cell Signaling
TECHNOLOGY®

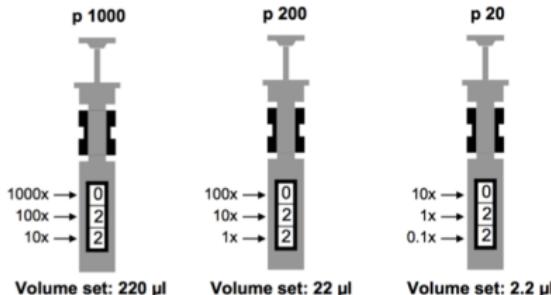

<https://youtu.be/yUstng0npaY>

Micropipettes

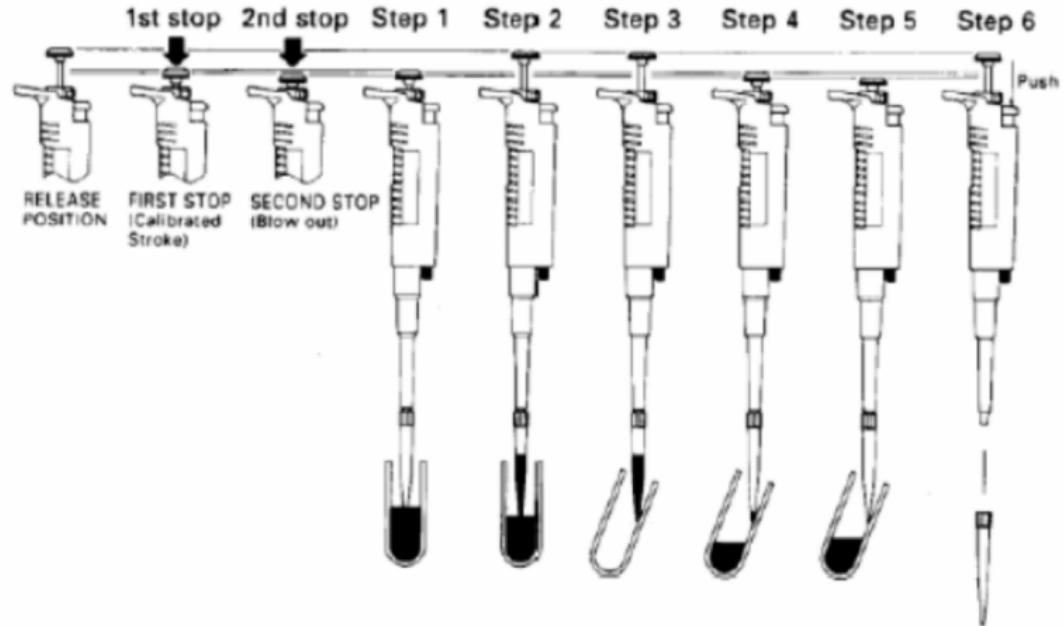
- **Definition:** A **micropipette** is a laboratory instrument used to measure small volumes (on the order of microliters).

- Micropipette sizes
 - P20: 2 – 20 μL
 - P200: 20-200 μL
 - P1000: 100-1000 μL
- Be mindful of which size you are using

Micropipettor: Reading the Volume



How to use a Micropipette



<https://www.youtube.com/watch?v=TMFeV9h6zEA>

Module 2 Worksheet

BMES Cell Team

Fall 2020



Overview of Module 2

1. Name one protein function:

2. What is the purpose of a Western blot?

3. What is the purpose of a BCA Assay? Which metal is present in BCA reagent?

4. What is the purpose of SDS-PAGE?

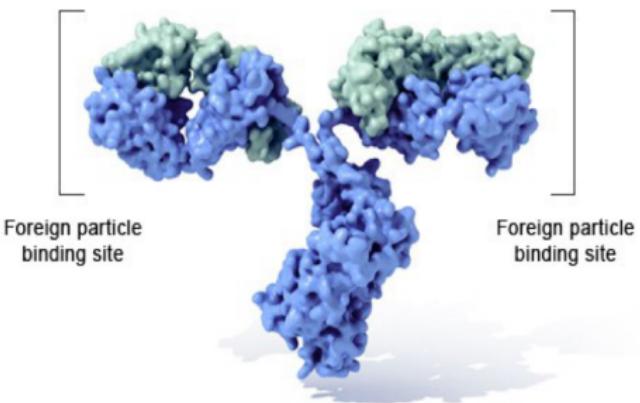
5. Why do we need to transfer proteins from a gel to a membrane?

Overview of Module 2

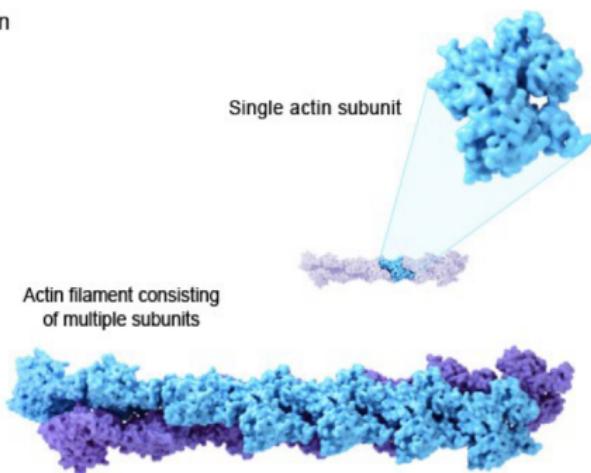
1. Name one protein function:

antibody, structural component, messenger, enzyme, regulatory

Immunoglobulin G (IgG)



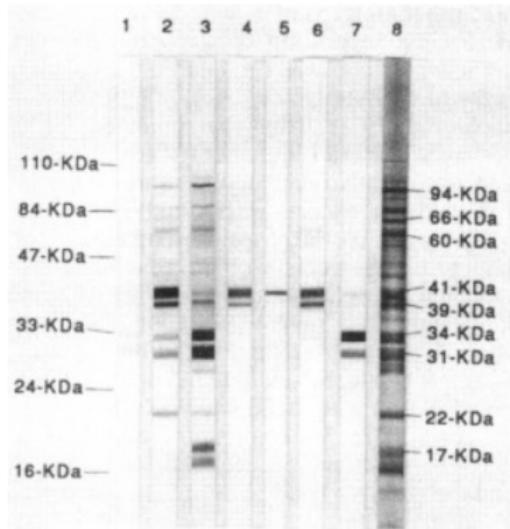
Actin



Overview of Module 2

2. What is the purpose of a Western blot?

separate and identify target protein to better understand the protein's role and how protein mutations can result in disease state



You know what goes great
with the Corona virus?



Lyme Disease

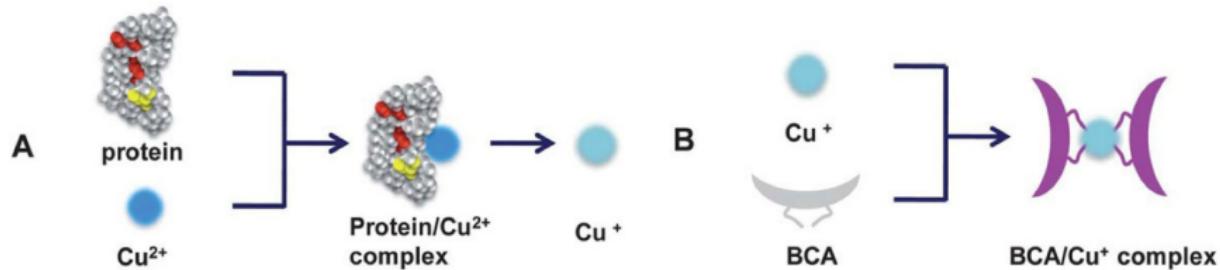


Overview of Module 2

3. What is the purpose of a BCA Assay? Which metal is present in BCA reagent?

A BCA Assay allows us to calculate sample protein concentration using absorbance readings.

The copper ion (Cu^{2+}) is reduced during the Biuret Reaction in a BCA Assay

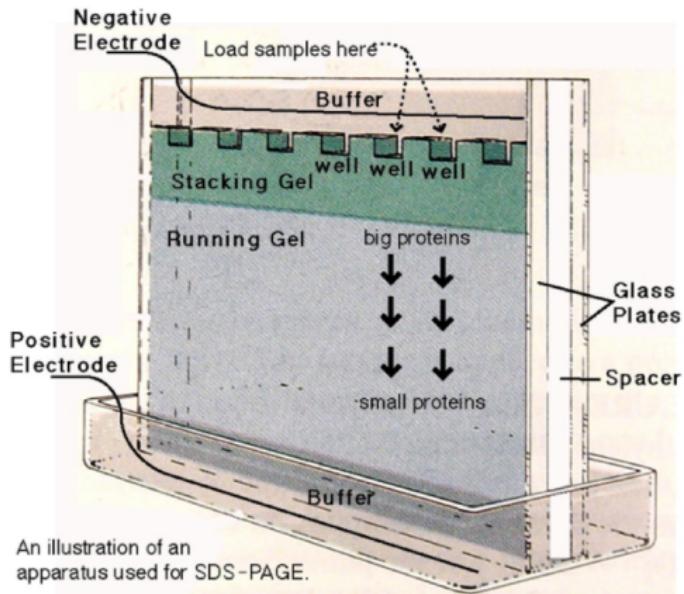


Overview of Module 2

4. What is the purpose of SDS-PAGE?

Gel electrophoresis separates protein fragments by size.

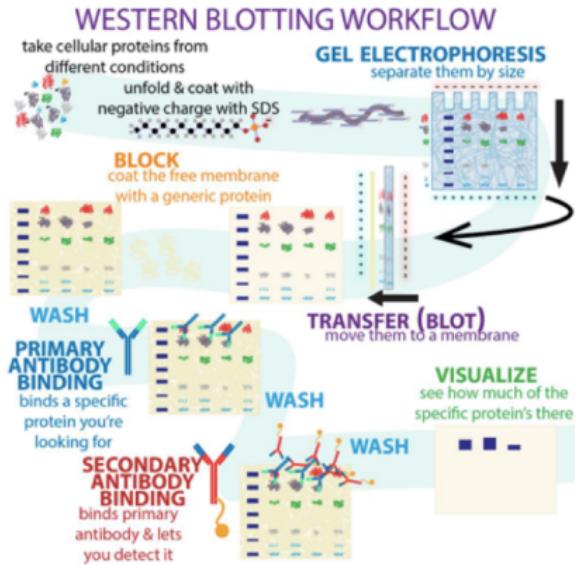
1D SDS-PAGE



Overview of Module 2

5. Why do we need to transfer proteins from a gel to a membrane?

The antibodies utilized in immunoblotting cannot bind to the protein when the protein is on the gel.



Practice Preparing a BCA Assay

1. Calculate serial dilution volumes.

We will start with a 20 ug/mL BSA solution.

- Need 25 uL of standard for each well x 3 replicates = 75 uL per standard
- Always make a little extra: _____ uL per standard

BSA Concentration (ug/mL)	BSA Solution Added (uL)	Water Added (uL)	Total Volume (uL)	Final Volume (uL)
2000 ug/mL	160 uL of 2000 ug/mL	0	160	160 - 80 = 80
1000 ug/mL	80 uL of 2000 ug/mL	80	160	160 - _____ = _____
500 ug/mL	_____ uL of 1000 ug/mL	_____	160	160 - _____ = _____
250 ug/mL	_____ uL of 500 ug/mL	_____	160	160 - _____ = _____
50 ug/mL	_____ uL of 250 ug/mL	_____	160	160 - _____ = _____
5 ug/mL	_____ uL of 50 ug/mL	_____	160	160 - _____ = _____
0 ug/mL	0	160	160	160

Practice Preparing a BCA Assay

1. Calculate serial dilution volumes.

We will start with a 20 ug/mL BSA solution.

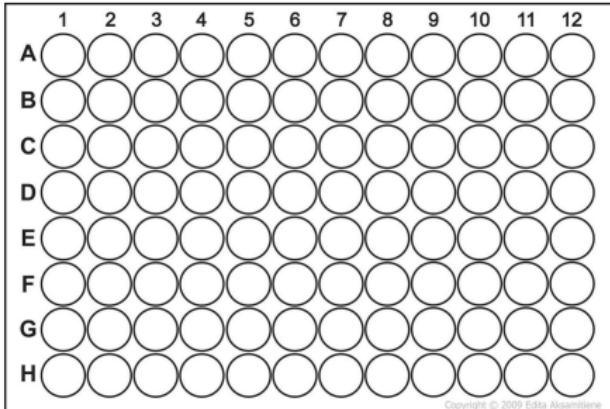
- Need 25 uL of standard for each well x 3 replicates = 75 uL per standard
- Always make a little extra: 80 uL per standard

BSA Concentration (ug/mL)	BSA Solution Added (uL)	Water Added (uL)	Total Volume (uL)	Final Volume (uL)
2000 ug/mL	160 uL of 2000 ug/mL	0	160	160 - 80 = 80
1000 ug/mL	80 uL of 2000 ug/mL	80	160	160 - <u>80</u> = <u>80</u>
500 ug/mL	<u>80</u> uL of 1000 ug/mL	<u>80</u>	160	160 - <u>80</u> = <u>80</u>
250 ug/mL	<u>80</u> uL of 500 ug/mL	<u>80</u>	160	160 - <u>80</u> = <u>80</u>
50 ug/mL	<u>32</u> uL of 250 ug/mL	<u>128</u>	160	160 - <u>16</u> = <u>144</u>
5 ug/mL	<u>16</u> uL of 50 ug/mL	<u>144</u>	160	160 - <u>0</u> = <u>160</u>
0 ug/mL	0	160	160	160

Practice Preparing a BCA Assay

2. Design layout for 96 well plate:

- a. Number of standards _____ x Number of repeats _____ = _____ wells
- b. Number of samples 2 x Number of repeats 2 = 4 wells
- c. Total number of wells = Standard wells _____ + Sample wells 4 = _____ wells
- d. Label the well plate below with standard and sample placement



Practice Preparing a BCA Assay

2. Design layout for 96 well plate:

- Number of standards 7 x Number of repeats 3 = 21 wells
- Number of samples 2 x Number of repeats 2 = 4 wells
- Total number of wells = Standard wells 21 + Sample wells = 4 = 25 wells
- Label the well plate below with standard and sample placement:

	Standards			samples								
	1	2	3	4	5	6	7	8	9	10	11	12
A	2000 μg/ml	2000 μg/ml	2000 μg/ml									
B	1000 μg/ml	1000 μg/ml	1000 μg/ml			PC9	PC9					
C	500 μg/ml	500 μg/ml	500 μg/ml									
D	250 μg/ml	250 μg/ml	250 μg/ml			PC9	PC9	PC9				
E	50 μg/ml	50 μg/ml	50 μg/ml									
F	10 μg/ml	10 μg/ml	10 μg/ml									
G	0 μg/ml	0 μg/ml	0 μg/ml									
H												

Practice Preparing a BCA Assay

3. Prepare reaction buffer:

- a. Total number of wells _____ x 200 uL/well = _____ uL reaction buffer
- b. Always make a little extra: _____ uL
- c. Combine reagents A, B, and C in ratio 25:24:1 to make reaction buffer
 - i. Reagent A = (total volume of buffer _____ uL)(25/50) = _____ uL
 - ii. Reagent B = (total volume of buffer _____ uL)(24/50) = _____ uL
 - iii. Reagent C = (total volume of buffer _____ uL)(1/50) = _____ uL

Practice Preparing a BCA Assay

3. Prepare reaction buffer:

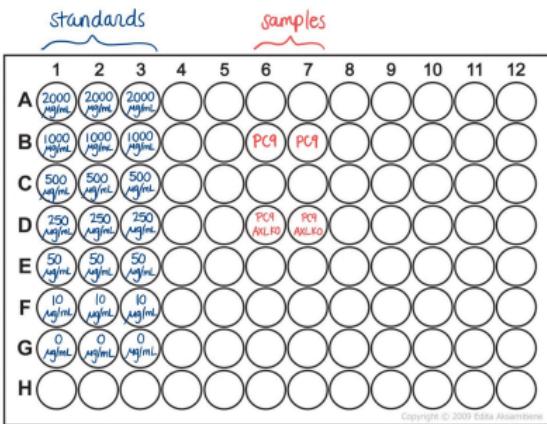
- a. Total number of wells 25 x 200 uL/well = 5000 uL reaction buffer
- b. Always make a little extra: 5500 uL
- c. Combine reagents A, B, and C in ratio 25:24:1 to make reaction buffer
 - i. Reagent A = (total volume of buffer 5500 uL)(25/50) = 2750 uL
 - ii. Reagent B = (total volume of buffer 5500 uL)(24/50) = 2640 uL
 - iii. Reagent C = (total volume of buffer 5500 uL)(1/50) = 110 uL

Practice Making a Standard Curve

→ Consider: You run this plate ...
and get the below absorbance readings

→ Create a standard curve

→ Use the standard curve to determine
the sample concentration of protein



1	2	3	4	5	6	7
A	2.799	2.482	2.343	0.047	0.049	0.048
B	1.644	1.313	1.368	0.048	0.253	0.246
C	0.904	0.806	0.774	0.049	0.047	0.049
D	0.566	0.529	0.452	0.047	0.305	0.26
E	0.206	0.201	0.438	0.048	0.048	0.047
F	0.106	0.119	0.101	0.047	0.048	0.048
G	0.097	0.096	0.096	0.048	0.048	0.048

Practice Making a Standard Curve

Step 1: Determine the average and standard deviation of replicates

1	2	3	4	5	6	7
A	2.799	2.482	2.343	0.047	0.049	0.048
B	1.644	1.313	1.368	0.048	0.253	0.246
C	0.904	0.806	0.774	0.049	0.047	0.049
D	0.566	0.529	0.452	0.047	0.305	0.26
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G	0.097	0.096	0.096	0.048	0.048	0.048



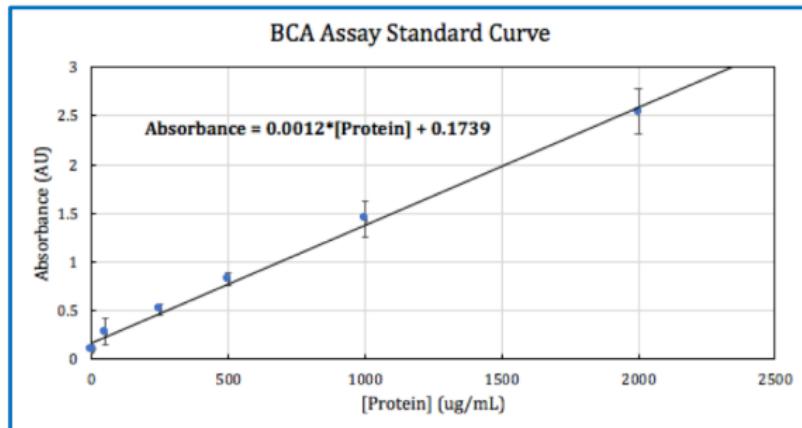
[Protein] (ug/mL)	AVG Absorbance (AU)	STD Absorbance (AU)
2000	2.541333333	0.233718492
1000	1.441666667	0.17737061
500	0.828	0.067734777
250	0.515666667	0.058157831
50	0.281666667	0.135411718
5	0.108666667	0.009291573
0	0.096333333	0.00057735

Practice Making a Standard Curve

Step 2: Plot the average and standard deviation of replicates

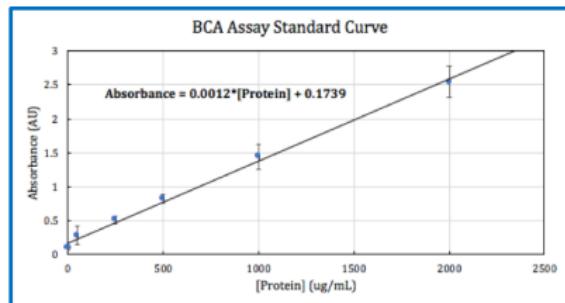
[Protein] (ug/mL)	AVG Absorbance (AU)	STD Absorbance (AU)
2000	2.541333333	0.233718492
1000	1.441666667	0.17737061
500	0.828	0.067734777
250	0.515666667	0.058157831
50	0.281666667	0.135411718
5	0.108666667	0.009291573
0	0.096333333	0.00057735

- Display equation of trendline
→ Note y-intercept



Practice Making a Standard Curve

Step 3: Use standard curve equation and absorbance data to calculate sample protein concentration



1	2	3	4	5	6	7
A	2.799	2.482	2.343	0.047	0.049	0.049
B	1.644	1.313	1.368	0.048	0.253	0.246
C	0.904	0.806	0.774	0.049	0.047	0.049
D	0.566	0.529	0.452	0.047	0.305	0.26
E	0.206	0.201	0.438	0.048	0.048	0.047
F	0.106	0.119	0.101	0.047	0.048	0.048
G	0.097	0.096	0.096	0.048	0.048	0.048



$$\text{Absorbance} = 0.0012 * [\text{Protein}] + 0.1739 \rightarrow [\text{Protein}] = \frac{\text{Absorbance} - 0.1739}{0.0012}$$

Sample	Absorbance 1	Absorbance 2
PC9	0.253	0.246
PC9 AXO KO	0.305	0.26

Sample	[Protein] 1	[Protein] 2
PC9	65.916666667	60.083333333
PC9 AXO KO	109.25	71.75

Sample	AVG [Protein] (ug/mL)	STD [Protein] (ug/mL)
PC9	63	4.124789557
PC9 AXO KO	90.5	26.51650429