

Contents

4 A data assimilating state-space model for algal growth under controlled conditions within a photo-bioreactor	2
4.1 Introduction	2
4.1.1 Coarse	2
4.2 Methods	3
4.2.1 Data Model: Photo-bioreactor setup, experimental design and data collection methods	3
4.2.2 Data model: Data treatment, distributions and measurement error	5
4.2.3 Process model: Carbon chemistry	6
4.2.4 Process model: Gas transfer equilibrium concentrations for O_2 and CO_2	11
4.2.5 Process model: Photosynthesis and respiration	12
4.2.6 Process model: Dilution	12
4.2.7 Process model: Ordinary differential equations	13
4.2.8 Design and setup of data assimilation model with both simulated and experimental data	14
4.2.9 Parameter Model: Priors	15
4.3 Results and Discussion	15
4.3.1 Posteriors with simulated data (photosynthesis and respiration as parameters)	15

4.3.2	Posteriors with simulated data (photosynthesis and respiration modelled as random walks)	19
4.3.3	Posteriors with experimental data (photosynthesis, respiration and respiratory quotients are random walks)	25
4.3.4	Posteriors with experimental data (photosynthesis and respiration are random walks and the respiratory quotients are noisy states)	31
4.3.5	Posteriors with experimental data (estimating obs error) . . .	37
A LiBbi model code		42
Bibliography		56

Chapter 4

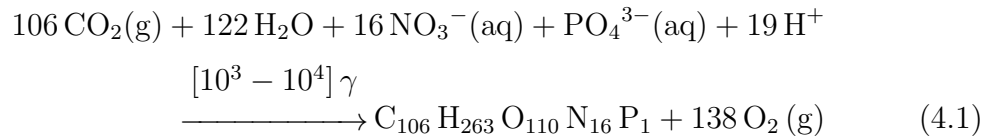
A data assimilating state-space model for algal growth under controlled conditions within a photo-bioreactor

4.1 Introduction

The key goal of biofuels production is the optimisation of biomass productivity in large-scale microalgal culturing systems such as open ponds or closed photobioreactors. Primary goal is maximise the production raw biomass.

4.1.1 Coarse

At the broadest scale / biggest picture the growth of microalgae can be expressed as the following chemical reaction in which C is taken from the atmosphere as CO_2 , along with aqueous nutrients N and P, to produce biomass



This equation captures a number of features of interest. The total biomass is determined by the concentrations of N and P. A typical growth media for microalgae is Guillard's Marine Enriched Seawater (F/2), which has

- **Biomass:** Taking the limiting nutrient to be Nitrogen, the maximum biomass of algae is 200mgL^{-1} , of which about 70mgL^{-1} is Carbon. This is equivalent

species	conc
N	$880\mu M$
P	$36\mu M$
TA	$2300\mu M$

0.12L of pure CO₂ At Standard Temperature and Pressure, which around to 325L of air for each litre of media.

- **Change in alkalinity:** Alkalinity is central concept in seawater chemistry that allows one to calculate how much Carbon can be dissolved in solution, and what form it takes (Dissolved Inorganic Carbon exists as CO₂, HCO₃⁻ and CO₃²⁻). Alkalinity measures the charge imbalance between strongly and weakly dissociating ions in solution. As the charged nutrients are removed from the solution during growth the alkainty increases and more Carbon is able to be dissolved in the media.

Carbon and light availability are two of the most common limiting factors of biomass productivity [?].

4.2 Methods

4.2.1 Data Model: Photo-bioreactor setup, experimental design and data collection methods

All data collection methods for this chapter were carried out by Peter Wood as part of a collection of PhD experiments (Peter Wood 2019 UTS PhD).

Microalgal culture *Nannochloropsis oceanica* (Droop) Green (strain CS-179) obtained from the Australian National Algae Culture Collection was cultured in 200 mL conical flasks; maintained in an incubator (Labec Pty Ltd) at 20°C, under an irra-

diance of $50 \mu\text{mol m}^{-2} \text{s}^{-1}$ of cool-white fluorescent light at a 12 hour light/12 hour dark cycle. Stock cultures were grown in f/2 saltwater medium [6] and diluted 5 days prior to the start of experiments to ensure that *N. oceanica* was in the exponential growth phase and not nutrient deprived. f/2 was sparged prior to stock culture dilutions to maximise carbon and oxygen content.

N. oceanica was cultured in four, 500 mL environmental photo-bioreactors (ePBRs, Phenometrics Inc) with a 10% v/v inoculation of stock culture. Top-side illumination over a path length of 25 cm was provided by a cool-white light LED, whilst temperature was maintained at 27°C using a Peltier heater-cooler connected to a water jacket. In-built thermocouples, calibrated against external temperature sensors attached to the Firesting module (TeX4; PyroScience GmbH), measured every 5 minutes were used to control the Peltier heater-cooler jacket through a feedback loop to an accuracy of $\pm 0.2^\circ\text{C}$. pH was also measured in 5 min intervals by in-built pH electrodes (Van London Inc); controlled by periodic CO_2 (5%) injections using valves in the ePBRs. pH was 3-point calibrated using pH buffer solutions at $\text{pH } 4.00 \pm 0.02$, $\text{pH } 7.00 \pm 0.02$ and $\text{pH } 10.00 \pm 0.02$. PBR mixing was controlled by magnetic stirring bars at 110 rpm. All four ePBRs were aerated with filtered/humidified air through a 1.2 mm needle valve (Terumo Co).

A period of 2 days was allowed for *N. occulata* to acclimate to the ePBRs at an irradiance of $500 \mu\text{mol m}^{-2} \text{s}^{-1}$ and a temperature of 27°C . Following this acclimation period, the PBR was set to the experimental condition of $2,000 \mu\text{mol photons m}^{-2} \text{s}^{-1}$ for another 2 days and a 12 hour light/12 hour dark cycle with a temperature of 27°C . ePBRs were maintained at an optical density (OD) of 0.4 using manual dilutions, creating a semi-batch culturing system. Dilutions occurred once per day (one hour before the light cycle), using aerated f/2 media. The experiment was conducted over a period of 4 days, samples were extracted post and prior dilution, as well as half way through the light cycle. 50 mL was extracted to examine total

alkalinity and dissolved inorganic carbon. Dissolved oxygen (DO) was measured using a 3 mm robust optical probe (OXROB10-OI; PyroScience GmbH) attached to a FireStingO2 logger (PyroScience GmbH). DO measurements were taken every 60 seconds and temperature-corrected using a temperature extension module (TeX4; PyroScience GmbH). DO was two-point calibrated using air-saturated seawater (100% saturation) and sodium sulfate-saturated water (0% saturation). At 2 hour intervals, a solenoid valve (SMC Pneumatics Pty. Ltd.) was used to stop aeration for 10 minutes to allow for observations of net photosynthesis.

Alkalinity and DIC was measured twice a day closely following the Standard Operating Procedures (SOP) outline in [3]. Approximately 30 mL of *N. oceanica* media was titrated against 0.1 M hydrochloric acid on an auto-titrator (800 Dosino; Metrohm AG) [SOP3b Open-cell titration].

[Chris: DIC measurement collection description] The Total Alkalinity and DIC were calculated from the output of the auto-titrator (volume of HCl delivered, pH) by calculating the pH as a function of the volume of the acid delivered (see SOP3a Annexe 1). Bayesian inference on TA, DIC, voffset, fK1, fK2, pHoff, pHslope.

table here for priors

link to code.

[BM: confirm that reference with PW]

4.2.2 Data model: Data treatment, distributions and measurement error

Valve, temperature, light (normalised to 0/1) and dilution rates were used to force the model. Dissolved oxygen, pH, dissolved inorganic carbon and total alkalinity observations for 4 days post acclimation were assimilated. While pH observations were calibrated and corrected, it was visible that O_2 observations were not

completely calibrated and experienced some sensor drift during the experiment.

The data model assigned log normally distributed observation errors for each instrument; $O_{2_{obs}} \sim \text{Log}\mathcal{N}(\log(O_2), \sigma_{O_2})$, $pH_{obs} \sim \text{Log}\mathcal{N}(\log(pH), \sigma_{pH})$, $DIC_{obs} \sim \text{Log}\mathcal{N}(\log(DIC), \sigma_{DIC})$, $TA_{obs} \sim \text{Log}\mathcal{N}(\log(TA), \sigma_{DIC})$, where the standard deviations ($\sigma_{O_2}, \sigma_{pH}, \sigma_{DIC}$) were unknown parameters to be estimated as part of the assimilating model. Dissolved inorganic carbon and total alkalinity measurements were obtained from the same instrument thus the error is shared between these states. Initial observation error priors started at $\sigma_{O_2} \sim \text{Log}\mathcal{N}(\log(0.1), 0.5)$ and then were adjusted during the PMMH tuning phase.

[Chris/SW: should I talk about the thinning out of O_2 and pH obs?] [Chris, if you can be bothered Yes, you might even expand this a bit by showing the a couple of runs with differing density. This issue of how to combine fuse different measurements with huge differences in sampling density is an interesting one)

4.2.3 Process model: Carbon chemistry

[Chris : Carbon in seawater summary] main equations and rate constants role of TA This consists of 4(?) equations with (4?) unknowns

Ideally CO2SYS [7] would have been used to calculate the carbon chemistry of the photo-bioreactor. CO2SYS is a program developed for CO_2 system calculations (CO2SYS) that calculates and returns a detailed state of the carbonate system of oceanographic water samples in seawater and freshwater [7]. A thorough explanation of CO2SYS, and seawater carbon chemistry in general, can be found in Zeebe and Wolf-gladrow [11]. A concise summary of the equations can be found in SOP 3a, Annexe 1 of [3].

Using two of the four measurable carbonate system parameters (total alkalinity, total inorganic CO_2 , pH, and either fugacity $f\text{CO}_2$ or partial pressure of CO_2) to

calculate the other two parameters at a set of input conditions (temperature and pressure). In the model the state variables TA and DIC are given and the pH and pCO₂ are calculated.

To incorporate CO2SYS into LiBbi for solving carbon chemistry on the timescale of the microalgae model, we explicitly defined 3 iterations (Eq. 4.5 - 4.13) of the Newton-Raphson method for finding approximations to roots of real valued functions. The Newton-Raphson method is an iterative process $x_{n+1} = x_n - \frac{f(x_n)}{f'(x_n)}$ considering a function $f(x_n)$, its derivative $f'(x_n)$ and an initial starting value x_0 . The approximate root x_{n+1} converges to the exact solution very quickly if a close initial starting value is picked. To ensure the quick convergence of the Newton-Raphson method, an approximating equation for pH_0 (Eq. 4.4) was obtained by fitting a stepwise regression with interactions to a range of simulated CO2SYS input parameters (temperature: 20-30, salinity: 30-40, DIC: 200-2500, and alkalinity: 1500-3000). A range of initial conditions and parameter values were tested, and each converged with RMSE < 0.01 across pH , HCO_3^- , CO_2 , and CO_3 , DIC, O_2 , TA by the 3rd iteration (Figure 4.1 and Table 4.1).

CO2SYS constants and iterative solution for pH, HCO_3^- , CO_2 , and CO_3

Total Sulfur:

$$\begin{aligned}
 TS &= \frac{0.14}{96.062} * \frac{S}{1.8065} \\
 IS &= 19.924 \frac{S}{(1000 - 1.005S)} \\
 KS_{int} &= -\frac{4276.1}{T_K} + 141.328 - 23.093 \log(T_K) + \left(-\frac{13856.0}{T_K} + 324.57 \right. \\
 &\quad \left. - 47.986 \log(T_K)\right) \sqrt{IS} + \left(\frac{35474}{T_K} - 771.54 + 114.723 \log(T_K)\right) IS \\
 &\quad - \frac{2698}{T_K} IS^{1.5} + \frac{1776}{T_K} IS^2 \\
 KS &= (1 - 0.001005S) e^{(KS_{int})}
 \end{aligned}$$

Fluorine:

$$TF = \frac{\frac{0.000067S}{18.9984}}{1.80655}$$

$$KF = e^{(-(-\frac{874.0}{T_K} - 0.111\sqrt{S} + 9.68))}$$

$$SWS_{2T} = \frac{(1 + \frac{TS}{KS})}{(1 + \frac{TS}{KS} + \frac{TF}{KF})}$$

$$Free_{2T} = 1 + \frac{TS}{KS}$$

H2O dissoc:

$$KW = e^{(148.9802 - \frac{13847.26}{T_K} - 23.6521\log(T_K) + (\frac{118.67}{T_K} - 5.977 + 1.0495\log(T_K))\sqrt{S} - 0.01615S)}$$

Boron:

$$KB = \exp\left(\frac{(-8966.90 - 2890.53\sqrt{S} - 77.942S + 1.728S\sqrt{S} - 0.0996S^2)}{T_K}\right)$$

$$+ 148.0248 + 137.1942\sqrt{S} + 1.62142S$$

$$- (24.4344 + 25.085\sqrt{S} + 0.2474S)\log(T_K) + 0.053105T_K\sqrt{S})$$

$$TB = 0.0004326\frac{S}{35}$$

Choice of carbonate dissociation constants K_1 and K_2 were Mehrbach [8] (refit by Dickson and Millero [2]) with $1.23K_1$ and $0.53K_2$ measured experiment specific adjustments:

$$K_1 = 10^{(-(\frac{3633.86}{T_K} - 61.2172 + 9.6777\log(T_K) - 0.011555S + 0.0001152S^2))} * 1.23 \quad (4.2)$$

$$K_2 = 10^{(-(\frac{471.8}{T_K} + 25.9290 - 3.16967\log(T_K) - 0.01781S + 0.0001122S^2))} * 0.53 \quad (4.3)$$

Approximating equation for the starting value of pH :

$$pH_0 = 12.26 - 0.0030605DIC - 0.043752T - 0.013625S + 0.00011315TA$$

$$+ 1.3463e - 5DIC * T + 5.2215e - 7DIC * TA \quad (4.4)$$

Newton-Raphson iterations:

$$h_n = 10^{-pH_n} \quad (4.5)$$

$$h_{n_{free}} = \frac{h_n}{Free_{2T}} \quad (4.6)$$

$$\begin{aligned} f_n = & (DIC * 1e - 6 * \frac{K_1 h_n + 2K_1 K_2}{h_n^2 + K_1 h_n + K_1 K_2} \\ & - h_{n_{free}} + \frac{KW}{h_n} - TA * 1e - 6 + \frac{TB}{1 + \frac{h_n}{KB}}) * 1e6 \end{aligned} \quad (4.7)$$

$$\begin{aligned} df_n = & (DIC * 1e - 6 * \frac{K_1 + 2K_1 K_2}{h_n^2 + K_1 h_n + K_1 K_2} \\ & - DIC * 1e - 6 * \frac{(K_1 h_n + 2K_1 K_2)}{(h_n^2 + K_1 h_n + K_1 K_2)^2} (2h_n + K_1) \\ & - TB \frac{1}{(1 + \frac{h_n}{KB})^2} / KB \\ & - \frac{KW}{h_n^2} - \frac{1}{Free_{2T}}) * 1e6 * (-\log(10) * 10^{-pH}) \end{aligned} \quad (4.8)$$

$$pH_{n+1} = pH_n - \frac{f_n}{df_n} \quad (4.9)$$

$$H_{n+1} = 10^{-pH_{n+1}} \quad (4.10)$$

$$CO_{2n+1} = \frac{H_{n+1}^2 DIC}{H_{n+1}^2 + K_1 H_{n+1} + K_1 K_2} \quad (4.11)$$

$$HCO_{3n+1} = \frac{H_{n+1} K_1 DIC}{H_{n+1}^2 + K_1 H_{n+1} + K_1 K_2} \quad (4.12)$$

$$CO_{3n+1} = \frac{K_1 K_2 DIC}{H_{n+1}^2 + K_1 H_{n+1} + K_1 K_2} \quad (4.13)$$

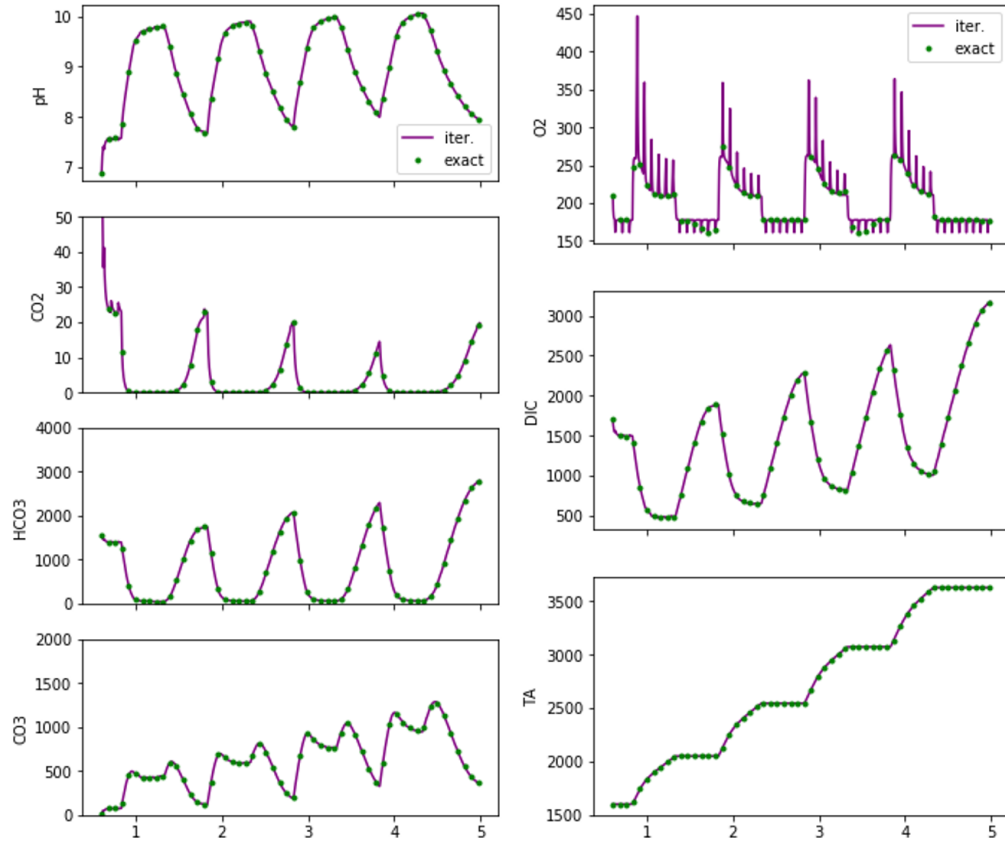


Figure 4.1 : Iterative (3rd iteration) vs exact solution for carbon chemistry CO_2 , HCO_3 , CO_3 , pH and state variables O_2 , DIC , and TA .

Variable	Iter. 1	Iter. 2	Iter. 3	Iter. 4	Iter. 5
pH	0.036092734	0.002355758	1.41E-05	6.93E-06	6.93E-06
CO_2	2.109401968	0.145719349	0.001222812	0.000866728	0.000866727
HCO_3	19.81869214	1.21021115	0.008016765	0.001025002	0.001025139
CO_3	20.89660704	1.307061652	0.00867642	0.001102278	0.001102434
DIC	16.78775711	0.958511825	0.005229318	0.002305054	0.002333411
O_2	0.308389964	0.016044284	4.18E-05	6.89E-05	7.59E-05
TA	2.607767674	0.160897272	0.000688102	0.001257725	0.001218981

Table 4.1 : RMSE for 5 iterations of the Newton-raphson carbon chemistry iterative solution.

4.2.4 Process model: Gas transfer equilibrium concentrations for O₂ and CO₂

The equilibrium concentration for CO₂ solubility in water CO_{2H} (μmol/L) is calculated using Henry's law,

$$CO_{2H} = K0_{CO2} * fCO2 * 1.0220 * 1e6 \quad (4.14)$$

where fCO₂ (atm) is the fugacity or approximately the partial pressure of CO₂, 1.0220 is the density of seawater (kg/L) at salinity 34 ppt and temperature 27°C [9] [4]. K0_{CO2} (mol/kg_{soln}/atm) is the solubility of gas in seawater [BM: ask Chris: solubility of gas? is this right] and is calculated from the fitted van't Hoff equation and the logarithmic Setchenow salinity dependence [10],

$$K0_{CO2} = e^{(-60.2409 + 93.4517 \frac{100}{T_K} + 23.3585 * \log(\frac{T_K}{100}) + S(0.023517 - 0.023656 \frac{T_K}{100} + 0.0047036 (\frac{T_K}{100})^2))} \quad (4.15)$$

where T_K is the temperature (K) and S is salinity (ppt). Similarly the equilibrium concentration for O₂ solubility in water O_{2H} is calculated using Henry's law,

$$O_{2H} = K0_{O2} * fO2 * 1.0220 * 1e - 6 \quad (4.16)$$

where fO₂ (atm) is the fugacity or approximately the partial pressure of O₂, 1.0220 is the density of seawater (kg/L) at salinity 34 ppt and temperature 27°C [9] [4], and K0_{O2} (mol/kg_{soln}/atm) is the solubility of oxygen in seawater with an adjusted salinity dependence [1],

$$K0_{O2} = \frac{e^{(-1282.8704 + \frac{36619.96}{T_K} + 223.1396 \log(T_K) - 0.354707 T_K + S(5.957e - 3 - \frac{3.7353}{T_K}) + 3.68e - 6 S^2)}}{0.2094e - 6} \quad (4.17)$$

where T_K is the temperature (K) and S is salinity (ppt). The equilibrium concentrations for O₂ and CO₂ are modelled together with the gas turning on and off during the experiment, as

$$Q^{air} kLa_{O_2}^{air} (O_{2H} - O_2) \quad (4.18)$$

$$Q^{air} kLa_{CO_2}^{air} (CO_{2H} - CO_2) \quad (4.19)$$

where Q^{air} is the gas state (1= on, 0= off), $kLa_{O_2}^{air}$ and $kLa_{CO_2}^{air}$ are the mass transfer coefficients for air (d^{-1}), and 0.893 is the ratio between measured O_2 and CO_2 mass transfer constants [5].

4.2.5 Process model: Photosynthesis and respiration

Net photosynthesis

$$\frac{\partial DIC}{\partial t} = -(P_1 I \frac{HCO_3^-}{K_m + HCO_3^-} - R_1) \quad (4.20)$$

$$\frac{\partial O_2}{\partial t} = \frac{1}{(RQ_d I + RQ_n(1 - I))} (P_1 I \frac{HCO_3^-}{K_m + HCO_3^-} - R_1) \quad (4.21)$$

$$\frac{\partial TA}{\partial t} = R_R (P_1 I \frac{HCO_3^-}{K_m + HCO_3^-}) \quad (4.22)$$

$$\frac{\partial C_p}{\partial t} = (P_1 I \frac{HCO_3^-}{K_m + HCO_3^-} - R_1) \quad (4.23)$$

RQ_d and RQ_n are the day and night respiratory quotients, the ratio of CO_2 produced and O_2 consumed by a cell.

The michaelis menton term $\frac{HCO_3^-}{K_m + HCO_3^-}$ represents the photosynthetically active carbon that the microalgae use for photosynthesis.

4.2.6 Process model: Dilution

$$\frac{Q^M}{V} (DIC^M - DIC)$$

$$\frac{Q^M}{V} (O_2^M - O_2)$$

$$\frac{Q^M}{V} (TA^M - TA)$$

$$\frac{Q^M}{V} (-C_p)$$

4.2.7 Process model: Ordinary differential equations

A summary of the ode's that make up the process model:

$$\begin{aligned}
 \text{Rate} \quad \frac{\partial DIC}{\partial t} &= \text{flux into cells} \quad -\left(P_1 I \frac{HCO_3^-}{K_m + HCO_3^-} - R_1\right) + \hat{Q}^{air} kLa_{CO_2}^{air} (CO_2^{air} - CO_2) + \text{gas transfer} \quad + \frac{Q^M}{V} (DIC^M - DIC) \\
 \frac{\partial O_2}{\partial t} &= \frac{(P_1 I \frac{HCO_3^-}{K_m + HCO_3^-} - R_1)}{(RQ_d I + RQ_n(1 - I))} + \hat{Q}^{air} kLa_{O_2}^{air} (O_2^{air} - O_2) + \frac{Q^M}{V} (O_2^M - O_2) \\
 \frac{\partial TA}{\partial t} &= R_R \left(P_1 I \frac{HCO_3^-}{K_m + HCO_3^-} - R_1\right) + \frac{Q^M}{V} (TA^M - TA) \\
 \frac{\partial C_p}{\partial t} &= (P_1 I \frac{HCO_3^-}{K_m + HCO_3^-} - R_1) + \frac{Q^M}{V} (-C_p)
 \end{aligned} \tag{4.24}$$

	Symbol	Description	Prior / Value	Unit
Initial conditions	DIC^0	Dissolved inorganic carbon	$\text{Log}\mathcal{N}(\log(1300), 0.2)$	$\mu\text{M L}^{-1}$
	O_2^0	Oxygen	$\text{Log}\mathcal{N}(\log(225), 0.2)$	$\mu\text{M L}^{-1}$
	TA^0	Total alkalinity	$\text{Log}\mathcal{N}(\log(1750), 0.1)$	$\mu\text{M L}^{-1}$
	C_p^0		$\text{Log}\mathcal{N}(\log(300), 0.2)$	$\mu\text{M L}^{-1}$
	pH^0	-	$\text{Log}\mathcal{N}(\log(8.5), 0.2)$	$\log_{10}(\text{-mol/L H+})$
	CO_2^0	Carbon dioxide	$\text{Log}\mathcal{N}(\log(5), 0.4)$	$\mu\text{M L}^{-1}$
	HCO_3^{-0}	Bicarbonate	$\text{Log}\mathcal{N}(\log(1500), 0.3)$	$\mu\text{M L}^{-1}$
	CO_3^{2-0}	Carbonate	$\text{Log}\mathcal{N}(\log(100), 0.4)$	$\mu\text{M L}^{-1}$
Flux into cells	P_1	Maximum photosynthesis rate	*	$\mu\text{M L}^{-1} \text{ hour}^{-1}$
	R_1	Respiration rate	*	$\mu\text{M L}^{-1} \text{ hour}^{-1}$
	K_m	Carbon restriction	*	$\mu\text{M L}^{-1}$
	RQ_d	Daytime respiratory quotient	*	-
	RQ_n	Night respiratory quotient	*	-
	R_R	Redfield ratio	*	-
	I	Light indicator	forcing (0/1)	-

Gas transfer terms	\hat{Q}^{air}	Indicator for flow in air line	forcing (0/1)	-
	$x_{CO_2}^{air}$	Mole fraction of CO ₂ atmosphere	400	ppm
	CO_{2H}	Equilibrium CO ₂ concentration	Eq. 4.14	$\mu\text{M L}^{-1}$
	CO_2^{air}	Sat CO ₂ conc with atmosphere	$x_{CO_2}^{air} CO_{2H}$	
	$kLa_{CO_2}^{air}$	Mass transfer coefficient for CO ₂	$0.893kLa_{O_2}^{air}$	day^{-1}
	$x_{O_2}^{air}$	Mole fraction of O ₂ atmosphere	0.2094	atm
	O_{2H}	Equilibrium O ₂ concentration	Eq. 4.16	$\mu\text{M L}^{-1}$
	O_2^{air}	Sat O ₂ conc with atmosphere	$x_{O_2}^{air} O_{2H}$	
	τ	half-life of $kLa_{O_2}^{air}$	range(2-20)	min^{-1}
	$kLa_{O_2}^{air}$	Mass transfer coefficient for O ₂	$\ln(2) * 24 * 60 / \tau$	day^{-1}
Dilution terms	Q^M	Dilution rate	forcing	ml day^{-1}
	V	Volume of the reactor	500	ml
	DIC^M	Media dissolved inorganic carbon	1724.20	$\mu\text{M L}^{-1}$
	O_2^M	Media oxygen concentration	226.65	$\mu\text{M L}^{-1}$
	TA^M	Media total alkalinity	1797.90	$\mu\text{M L}^{-1}$

Table 4.2 : State variable, parameter and forcing definitions with units, and their assignments: either fixed values, priors on initial condition or priors on parameters (*) defined later in Section 4.2.9.

4.2.8 Design and setup of data assimilation model with both simulated and experimental data

A simulated dataset was created by running the process model described in Section 4.2.7 with a fixed set of parameters ($P_1 = 200$, $R_1 = 30$, $kLa_{O_2}^{air} = 200$, $K_m = 150$, $RQ_d = 0.85$, $RQ_n = 0.95$, $R_R = 0.075$) and initial conditions ($O_2^0 = 225$, $DIC^0 = 1250$, $TA^0 = 1750$).

The first set of results assimilated the simulated observations and attempted to

recover the true value parameters.

[Chris: noise on simulated dataset? sigmas?]

4.2.9 Parameter Model: Priors

[BM: I've put these in at the beginning of each results section for now and described how each run is different in the **In this run:** parts, so that they're clear when we're discussing the results. Will move it here after our meeting.]

4.3 Results and Discussion

4.3.1 Posteriors with simulated data (photosynthesis and respiration as parameters)

State posteriors are visualised by plotting the median and shading 95% credible intervals, while parameter priors and posteriors are displayed by histograms.

In this run:

P_1 , R_1 , kLA_{O_2} , K_m , R_R , RQ_d , and RQ_n and all treated as parameters constant through time but unknown. The data model assigned log normally distributed observation errors for each instrument;

$$O_{2_{obs}} \sim \text{Log}\mathcal{N}(\log(O_2), \sigma_{O_2})$$

$$pH_{obs} \sim \text{Log}\mathcal{N}(\log(pH), \sigma_{pH})$$

$$DIC_{obs} \sim \text{Log}\mathcal{N}(\log(DIC), \sigma_{DIC})$$

$$TA_{obs} \sim \text{Log}\mathcal{N}(\log(TA), \sigma_{DIC})$$

Parameter	Prior	Proposal
P_1	$\text{Log}\mathcal{N}(\log(250.0), 0.8)$	$\text{Log}\mathcal{N}(\log(P_1), 0.08)$
R_1	$\text{Log}\mathcal{N}(\log(20.0), 0.8)$	$\text{Log}\mathcal{N}(\log(R_1), 0.08)$
kLA_{O_2}	$\text{Log}\mathcal{N}(\log(200.0), 0.3)$	$\text{Log}\mathcal{N}(\log(kLA_{O_2}), 0.03)$
K_m	$\text{Log}\mathcal{N}(\log(200.0), 0.6)$	$\text{Log}\mathcal{N}(\log(K_m), 0.06)$
R_R	$\text{Uniform}(0, 0.2)$	$\text{Trun}\mathcal{N}(R_R, 0.01, \text{lower} = 0, \text{upper} = 0.2)$
RQ_d	$\text{Uniform}(0.6, 1)$	$\text{Trun}\mathcal{N}(RQ_d, 0.005, \text{lower} = 0.6, \text{upper} = 1.0)$
RQ_n	$\text{Uniform}(0.6, 1)$	$\text{Trun}\mathcal{N}(RQ_n, 0.005, \text{lower} = 0.6, \text{upper} = 1.0)$
σ_{O_2}	0.3	*
σ_{pH}	0.3	*
σ_{DIC}	0.3	*

Table 4.3 : Table of Parameters, their priors and proposal distributions. * indicates the parameter was held fixed.

Parameter	Quantiles (25%, 75%)	Quantiles (5%, 95%)	True value
P_1	(237.2167, 302.0565)	(200.8971, 365.1751)	200
R_1	(17.5370, 32.3501)	(10.6537, 45.3096)	30
$kLA_{O_2}^{air}$	(177.1383, 222.9641)	(148.2838, 267.5934)	200
K_m	(182.6643, 362.6714)	(118.4125, 643.8518)	150
R_R	(0.0504, 0.1041)	(0.0196, 0.1458)	0.075
RQ_d	(0.7529, 0.8718)	(0.6894, 0.9666)	0.85
RQ_n	(0.7469, 0.9338)	(0.6400, 0.9920)	0.95

Table 4.4 : True values for parameters used to create the simulated observations, and posterior (25%, 75%), (5%, 95%) quantiles for parameters after assimilating observations.

Results:

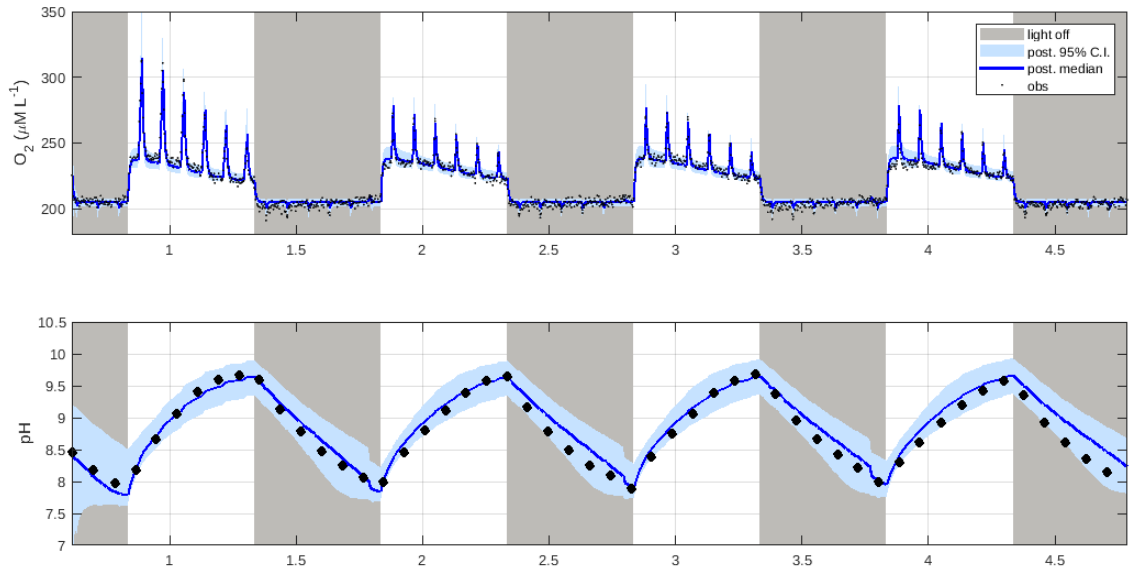


Figure 4.2 : Posterior medians (solid blue line), 95% credible intervals (shaded blue), and simulated observations (black) for O_2 and pH across 4 days.

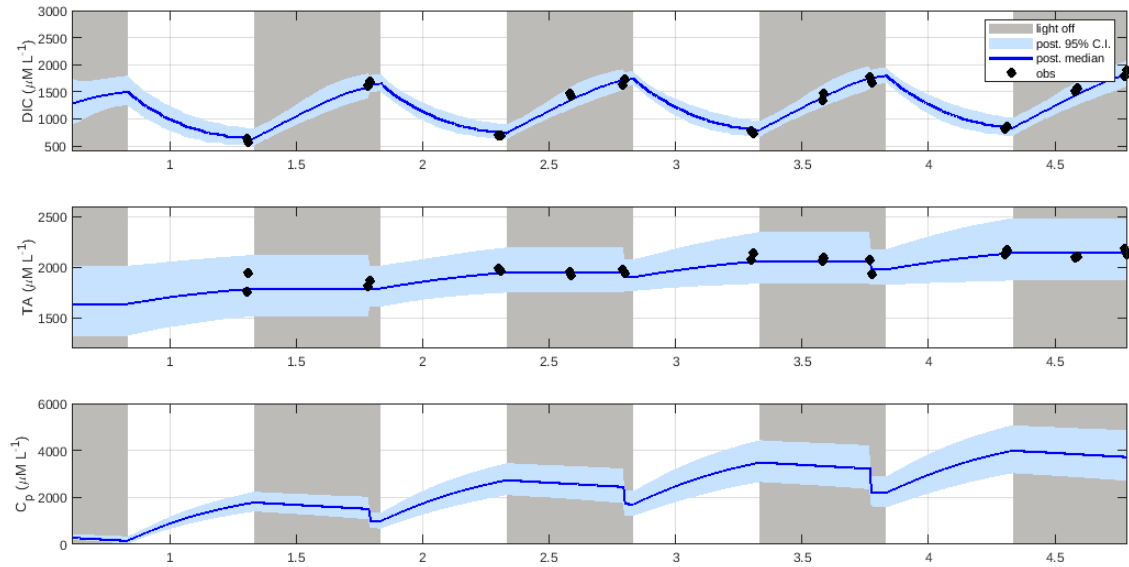


Figure 4.3 : Posterior medians (solid blue line), 95% credible intervals (shaded blue), and simulated observations (black) for DIC , TA and C_p across 4 days.

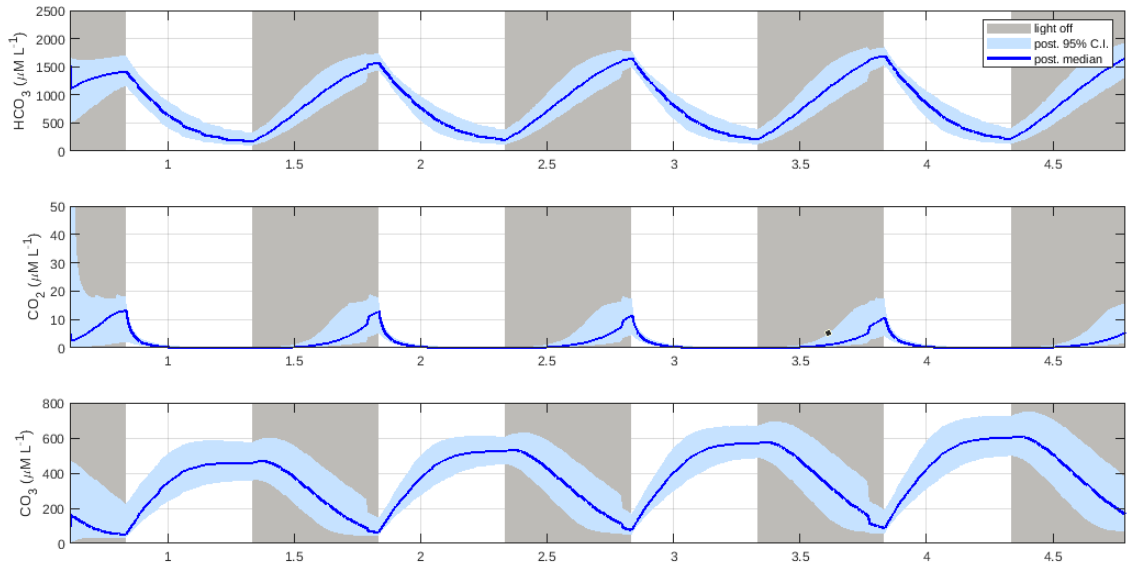


Figure 4.4 : Posterior medians (solid blue line) and 95% credible intervals (shaded blue) for HCO_3 , CO_2 and CO_3 across 4 days.

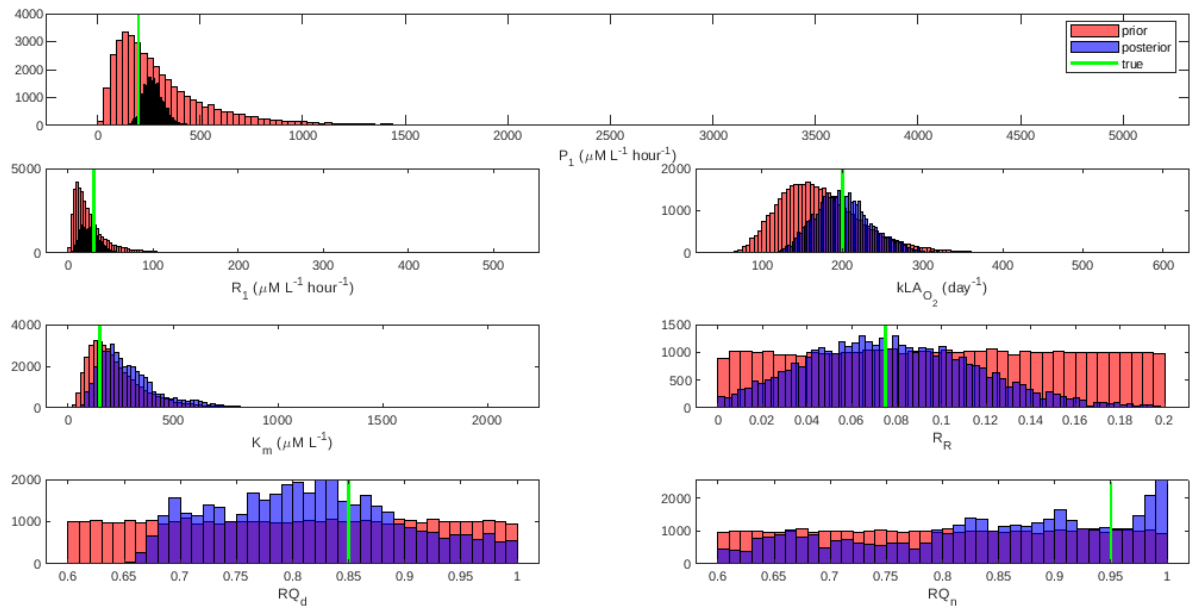


Figure 4.5 : Priors (pink), posteriors (purple) and true values (green) for model parameters.

After 50,000 samples the true values of parameters R_1 , $kLA_{O_2}^{air}$, R_R , and RQ_d were recovered to the 25th and 75th percentile (Table 4.4). While the true values of K_m and RQ_n did not lie within the 25th and 75th quantile, they were captured by the 5th and 95th percentiles (Table 4.4). P_1 was the only parameter whose true value lied on the cusp of the 5th percentile. Parameters P_1 , R_1 , $kLA_{O_2}^{air}$, and R_R mixed well, while parameters K_m , RQ_d , and RQ_n did not mix as well, some autocorrelation between samples was present (Figure A.3).

Log-likelihood stopped rising (Figure A.2). Acceptance rate was XX.

Observed state variables O_2 , DIC , TA , and observed pH posteriors were in excellent agreement with observations, with all observations fitting within the 95% credible interval posteriors (Figure 4.2, Figure 4.3).

Unobserved state variable C_p .. (Figure 4.3).

Carbon chemistry variables HCO_3 , CO_2 , and CO_3 ... (Figure 4.4).

4.3.2 Posteriors with simulated data (photosynthesis and respiration modelled as random walks)

State posteriors are visualised by plotting the median and shading 95% credible intervals, while parameter priors and posteriors are displayed by histograms.

In this run:

kLA_{O_2} , K_m , R_R , RQ_d , and RQ_n and all treated as parameters constant through time but unknown. Photosynthesis (P_1) and respiration (R_1) are both modelled as random walks, by taking P and R , previously constant parameters, and replacing them by $P_1(t)$ and $R_1(t)$. Here, we take $P_1(t)$ and $R_1(t)$ to be such that

$$P_1(t + \Delta t) = P(t) + r_P$$

$$R_1(t + \Delta t) = R(t) + r_R$$

where $r_P \sim N(0, \sigma_{r_P})$, $r_R \sim N(0, \sigma_{r_R})$, and Δt is the length of discrete time-step. For the purpose of the Bayesian analysis here, σ_{r_P} and σ_{r_R} are treated as parameters

to be inferred.

The data model assigned log normally distributed observation errors for each instrument;

$$O_{2_{obs}} \sim \text{Log}\mathcal{N}(\log(O_2), \sigma_{O_2})$$

$$pH_{obs} \sim \text{Log}\mathcal{N}(\log(pH), \sigma_{pH})$$

$$DIC_{obs} \sim \text{Log}\mathcal{N}(\log(DIC), \sigma_{DIC})$$

$$TA_{obs} \sim \text{Log}\mathcal{N}(\log(TA), \sigma_{DIC})$$

Parameter	Prior	Proposal
kLA_{O_2}	$\text{Log}\mathcal{N}(\log(200.0), 0.3)$	$\text{Log}\mathcal{N}(\log(kLA_{O_2}), 0.03)$
K_m	$\text{Log}\mathcal{N}(\log(200.0), 0.6)$	$\text{Log}\mathcal{N}(\log(K_m), 0.06)$
R_R	$\text{Uniform}(0, 0.2)$	$\text{Trun}\mathcal{N}(R_R, 0.01, \text{lower} = 0, \text{upper} = 0.2)$
RQ_d	$\text{Uniform}(0.6, 1)$	$\text{Trun}\mathcal{N}(RQ_d, 0.005, \text{lower} = 0.6, \text{upper} = 1.0)$
RQ_n	$\text{Uniform}(0.6, 1)$	$\text{Trun}\mathcal{N}(RQ_n, 0.005, \text{lower} = 0.6, \text{upper} = 1.0)$
σ_{r_P}	$\mathcal{N}(0.01, 0.001)$	$\mathcal{N}(\sigma_{r_P}, 0.0001)$
σ_{r_R}	$\mathcal{N}(0.01, 0.001)$	$\mathcal{N}(\sigma_{r_R}, 0.0001)$
σ_{O_2}	0.3	*
σ_{pH}	0.3	*
σ_{DIC}	0.3	*

Table 4.5 : Table of Parameters, their priors and proposal distributions. * indicates the parameter was held fixed.

Parameter	Quantiles (25%, 75%)	Quantiles (5%, 95%)	True value
$kLA_{O_2}^{air}$	(170.9854, 216.0578)	(145.6899, 253.4652)	200
K_m	(187.3586, 386.4103)	(93.9670, 641.1432)	150
R_R	(0.0521, 0.1075)	(0.0192, 0.1544)	0.075
RQ_d	(0.7906, 0.8965)	(0.6833, 0.9711)	0.85

RQ_n	(0.6654, 0.8334)	(0.6164, 0.9739)	0.95
--------	------------------	------------------	------

Table 4.6 : True values for parameters used to create the simulated observations, and posterior (25%, 75%), (5%, 95%) quantiles for parameters after assimilating observations.

Results:

After 50,000 samples the true values of parameters $kLA_{O_2}^{air}$, R_R , and RQ_d were recovered to the 25th and 75th percentile (Table 4.6). While the true values of K_m and RQ_n did not lie within the 25th and 75th quantile, they were captured by the 5th and 95th percentiles (Table 4.6). Parameters σ_{rP} , σ_{rR} , $kLA_{O_2}^{air}$, and R_R mixed well, while parameters K_m , RQ_d , and RQ_n did not mix as well, some autocorrelation between samples was present (Figure 4.11). The true values of P_1 (200) and R_1 (30) lie within the posterior 95% credible intervals (Figure 4.10).

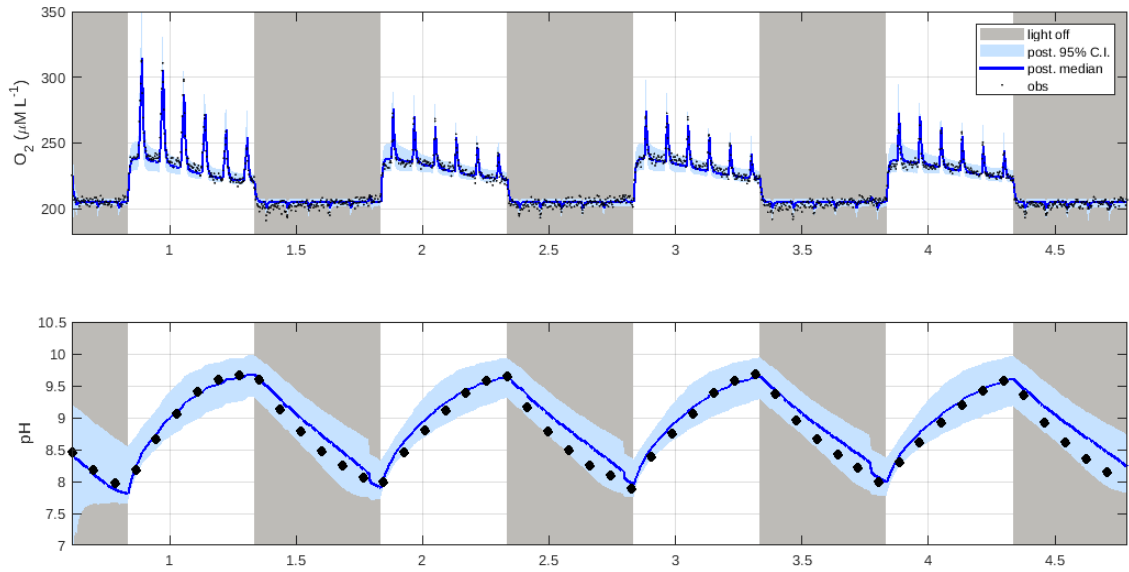


Figure 4.6 : Posterior medians (solid blue line), 95% credible intervals (shaded blue), and simulated observations (black) for O_2 and pH across 4 days.

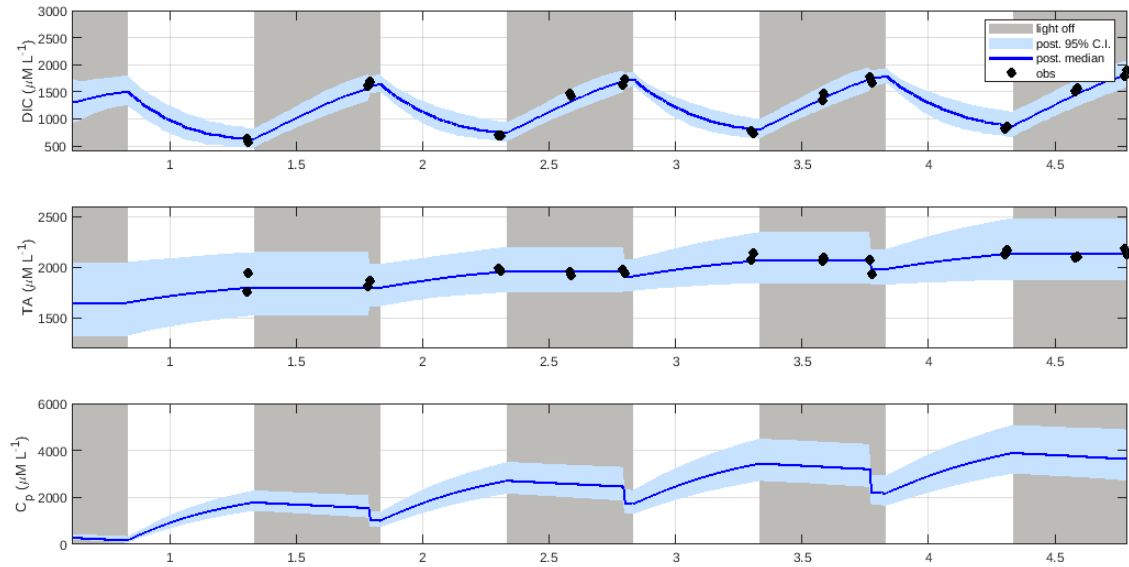


Figure 4.7 : Posterior medians (solid blue line), 95% credible intervals (shaded blue), and simulated observations (black) for DIC , TA and C_p across 4 days.

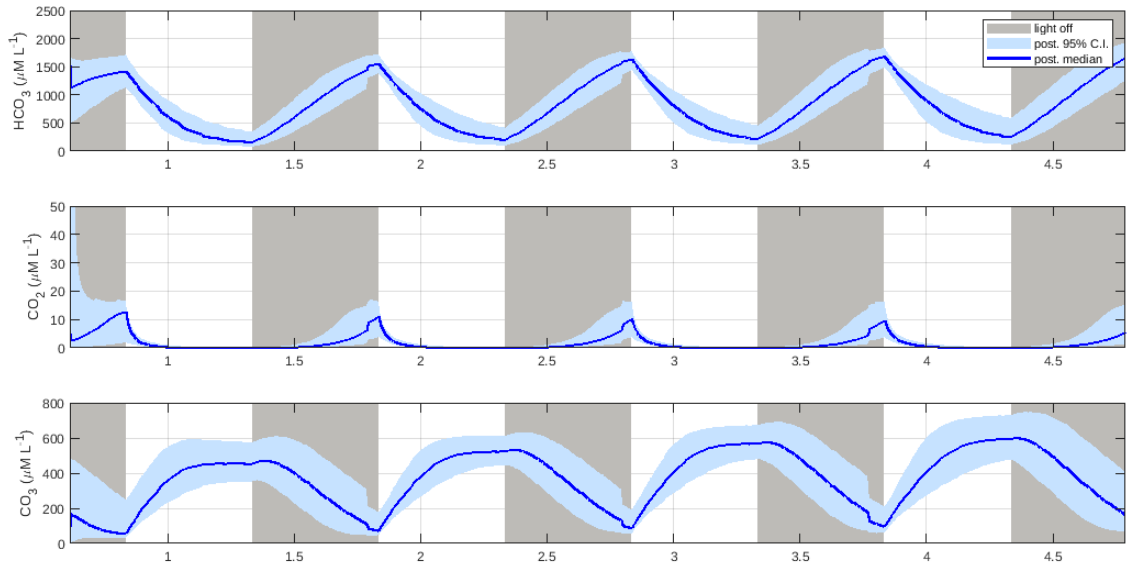


Figure 4.8 : Posterior medians (solid blue line) and 95% credible intervals (shaded blue) for HCO_3 , CO_2 and CO_3 across 4 days.

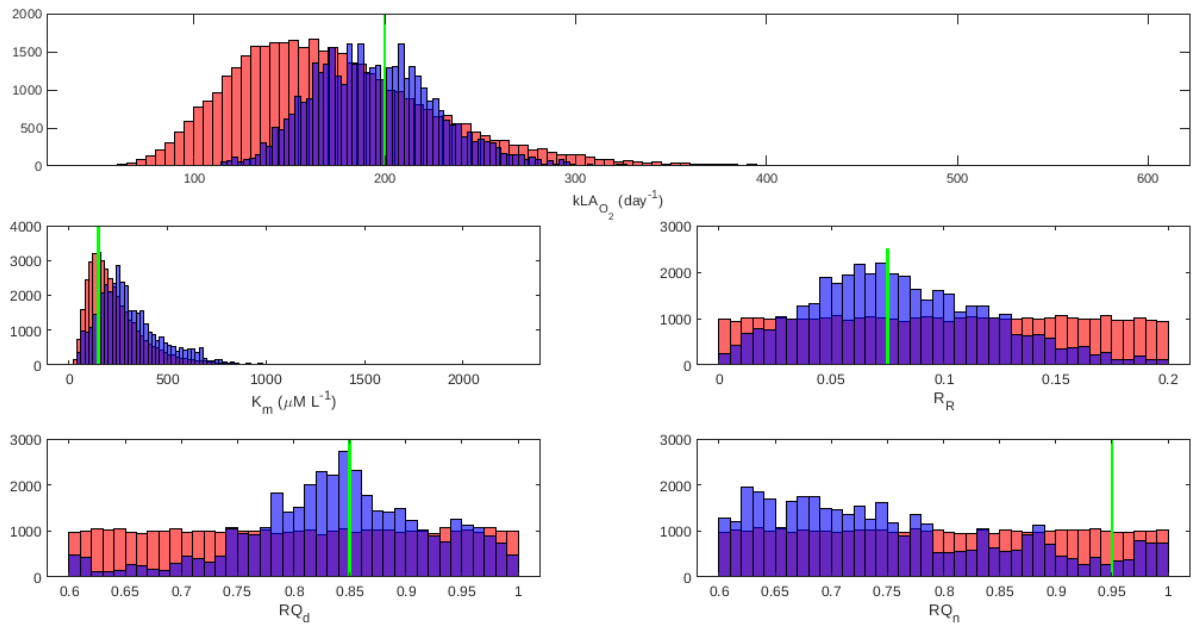


Figure 4.9 : Priors (pink), posteriors (purple) and true values (green) for model parameters.

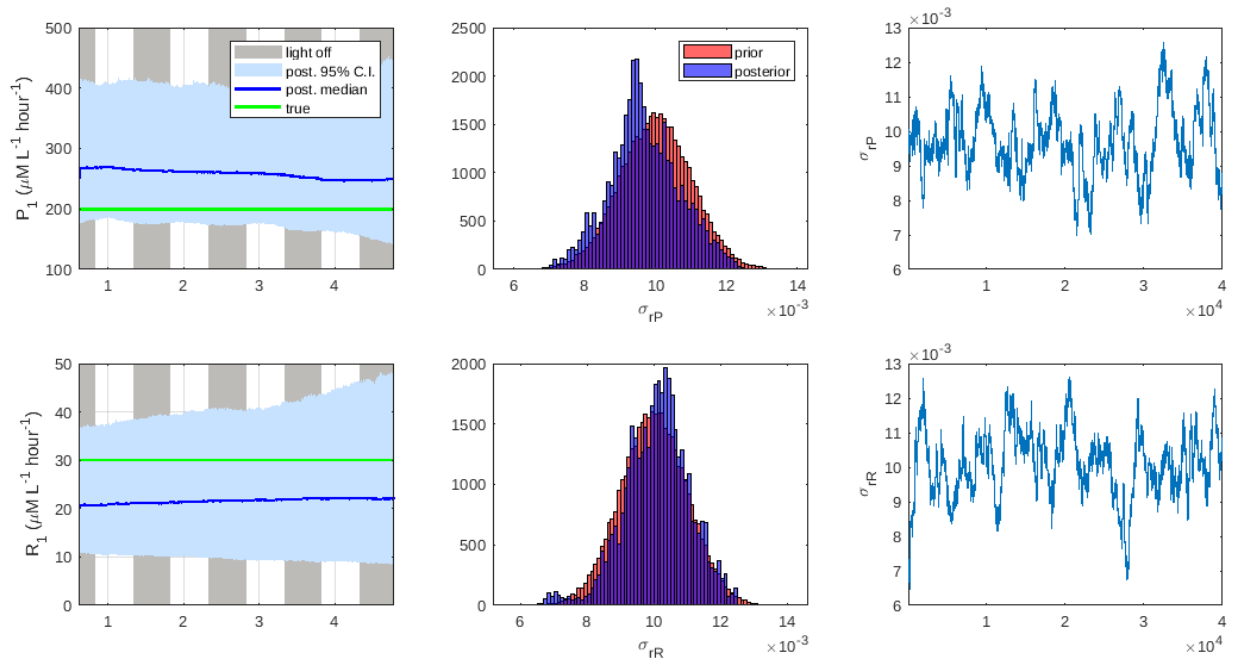


Figure 4.10 : Random walk posteriors P_1 and R_1 medians (solid blue), 95% credible intervals (shaded blue), and true values (green). σ_{rP} and σ_{rR} priors (pink), posteriors (purple), true values (green) and traces.

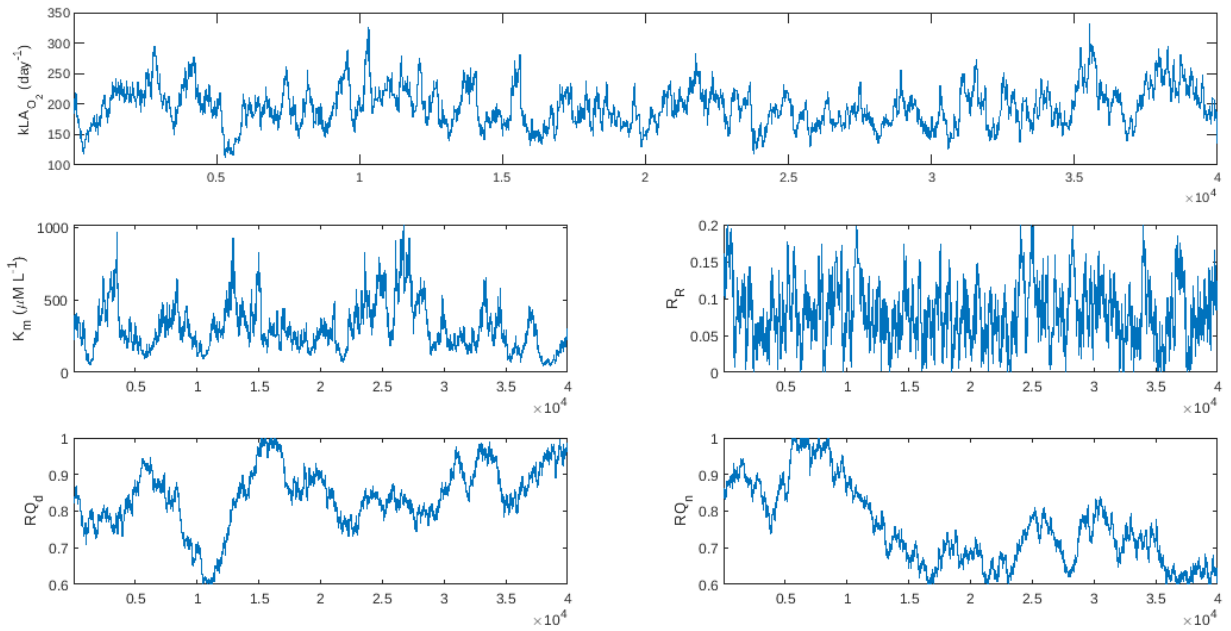


Figure 4.11 : Traces for model parameters.

4.3.3 Posteriors with experimental data (photosynthesis, respiration and respiratory quotients are random walks)

State posteriors are visualised by plotting the median and shading 95% credible intervals, while parameter priors and posteriors are displayed by histograms.

In this run:

Photosynthesis (P_1) and respiration (R_1) are both modelled as random walks, by taking P and R , previously constant parameters, and replacing them by $P_1(t)$ and $R_1(t)$. Here, we take $P_1(t)$ and $R_1(t)$ to be such that

$$P_1(t + \Delta t) = P(t) + r_P$$

$$R_1(t + \Delta t) = R(t) + r_R$$

where $r_P \sim N(0, \sigma_{r_P})$, $r_R \sim N(0, \sigma_{r_R})$, and Δt is the length of discrete time-step. For the purpose of the Bayesian analysis here, σ_{r_P} and σ_{r_R} are treated as parameters to be inferred. The respiratory quotients RQ_d and RQ_n were also treated as random walks with rP and rR as wiener processes. kLA_{O_2} , K_m , R_R , σ_{rP} , σ_{rR} are all treated as parameters constant through time but unknown. The data model assigned log normally distributed observation errors for each instrument;

$$O_{2_{obs}} \sim \text{Log}\mathcal{N}(\log(O_2), \sigma_{O_2})$$

$$pH_{obs} \sim \text{Log}\mathcal{N}(\log(pH), \sigma_{pH})$$

$$DIC_{obs} \sim \text{Log}\mathcal{N}(\log(DIC), \sigma_{DIC})$$

$$TA_{obs} \sim \text{Log}\mathcal{N}(\log(TA), \sigma_{DIC})$$

Parameter	Prior	Proposal
kLA_{O_2}	$\text{Log}\mathcal{N}(\log(200.0), 0.3)$	$\text{Log}\mathcal{N}(\log(kLA_{O_2}), 0.03)$
K_m	$\text{Log}\mathcal{N}(\log(200.0), 0.6)$	$\text{Log}\mathcal{N}(\log(K_m), 0.06)$
R_R	$\text{Uniform}(0, 0.2)$	$\text{Trun}\mathcal{N}(R_R, 0.005, \text{lower} = 0, \text{upper} = 0.2)$
σ_{r_P}	$\mathcal{N}(0.02, 0.002)$	$\mathcal{N}(\sigma_{r_P}, 0.0002)$
σ_{r_R}	$\mathcal{N}(0.01, 0.001)$	$\mathcal{N}(\sigma_{r_R}, 0.0001)$
σ_{O_2}	0.3	*
σ_{pH}	0.3	*
σ_{DIC}	0.5	*

Table 4.7 : Table of Parameters, their priors and proposal distributions. * indicates the parameter was held fixed.

Parameter	Quantiles (25%, 75%)	Quantiles (5%, 95%)
$kLA_{O_2}^{air}$	(139.5979, 170.8102)	(120.9171, 204.5474)
K_m	(168.1931, 378.8001)	(104.2598, 599.6881)
R_R	(0.0815, 0.1512)	(0.0284, 0.1844)
σ_{r_P}	(0.0178, 0.0202)	(0.0159, 0.0222)
σ_{r_R}	(0.0094, 0.0113)	(0.0081, 0.0121)

Table 4.8 : Posterior (25%, 75%), (5%, 95%) quantiles for parameters after assimilating observations.

Results:

40,000 samples run with 1024 particles (40,000 was all that could complete in the maximum time allocation on the hpc cluster). 10,000 discarded as burn-in, acceptance rate was XX.

Observed state variables DIC and TA posteriors perform well with all observations lying within the 95% credible intervals (Figure 4.13). O_2 posteriors tracked the

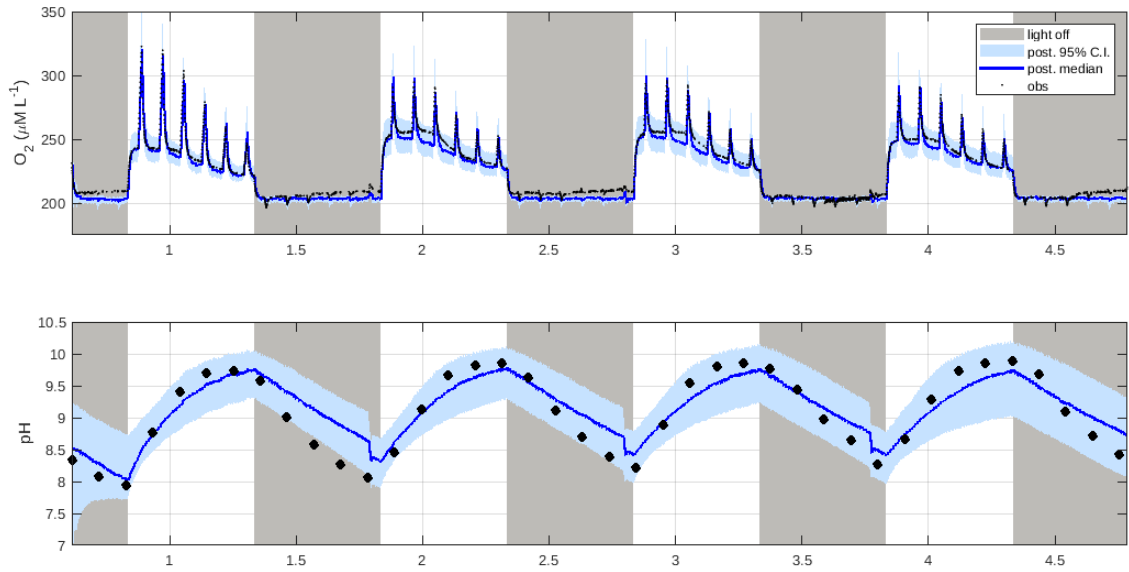


Figure 4.12 : Posterior medians (solid blue line), 95% credible intervals (shaded blue), and simulated observations (black) for O_2 and pH across 4 days.

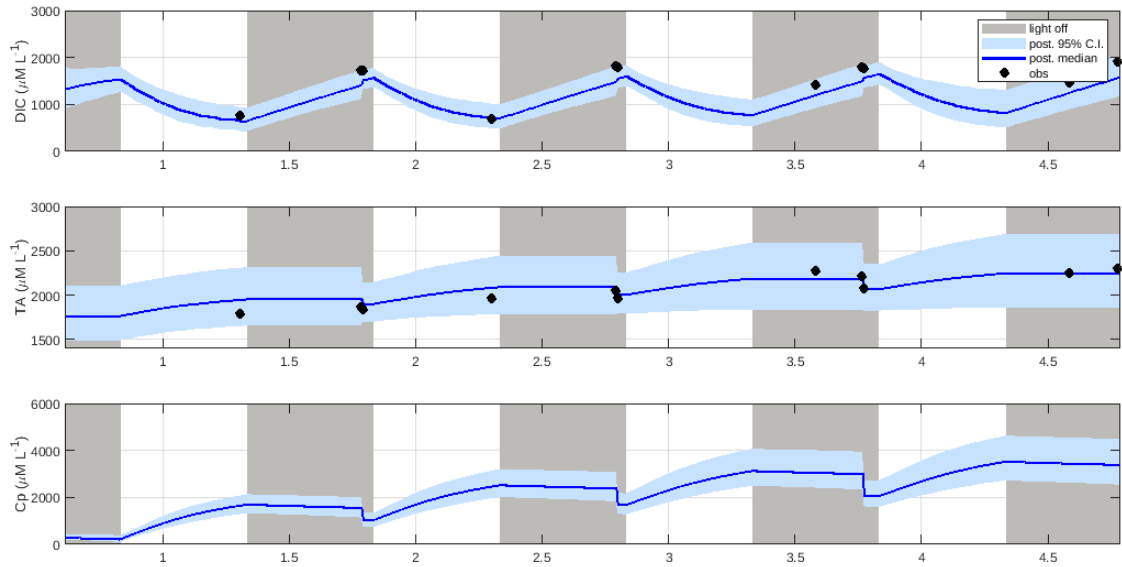


Figure 4.13 : Posterior medians (solid blue line), 95% credible intervals (shaded blue), and simulated observations (black) for DIC , TA and C_p across 4 days.

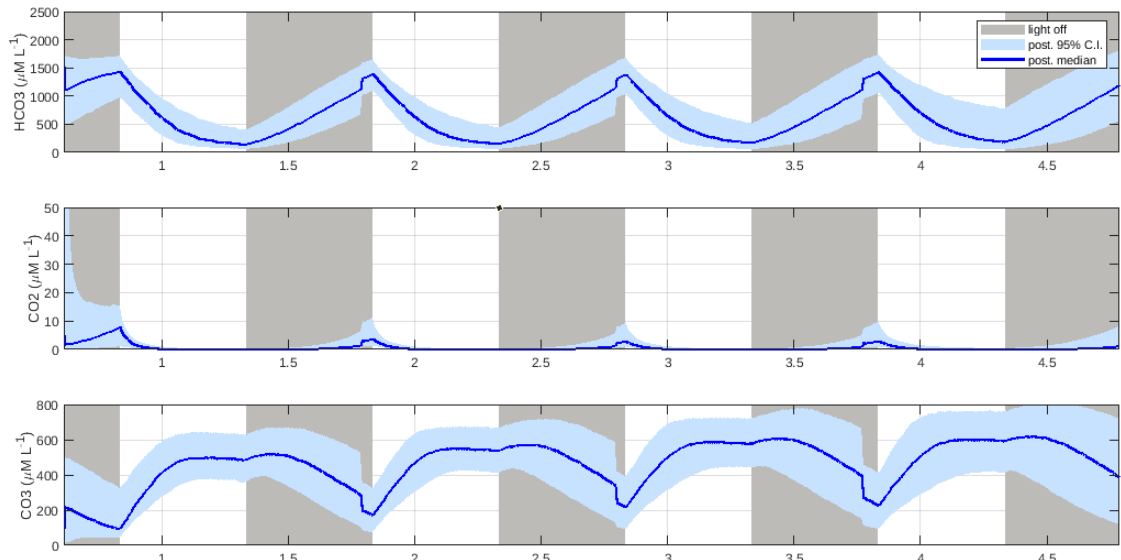


Figure 4.14 : Posterior medians (solid blue line) and 95% credible intervals (shaded blue) for HCO_3^- , CO_2 and CO_3 across 4 days.

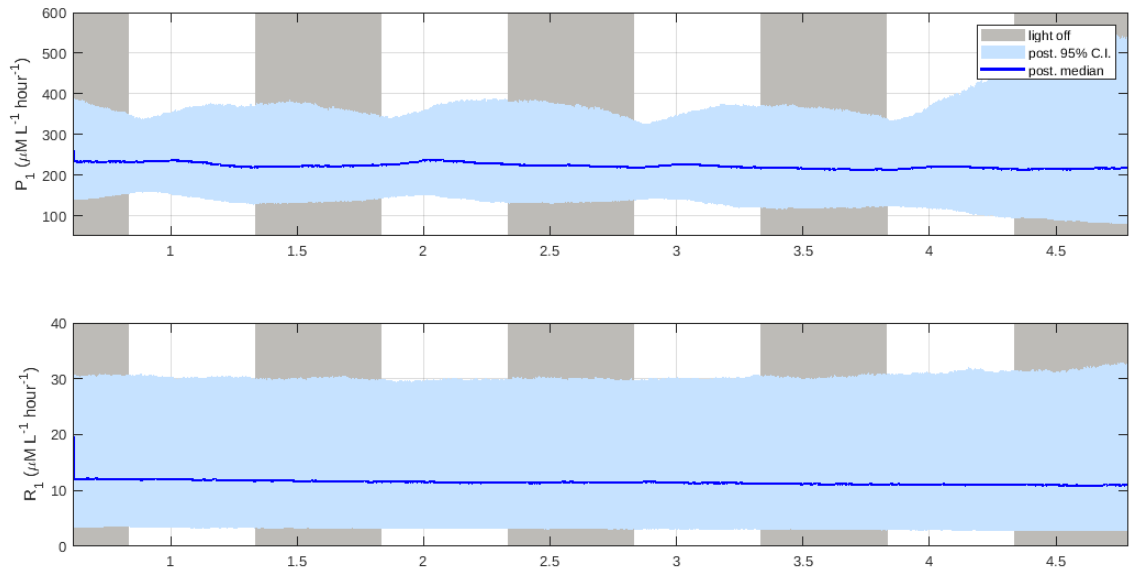


Figure 4.15 : Posterior medians (solid blue line) and 95% credible intervals (shaded blue) for photosynthesis P_1 and respiration R_1 .

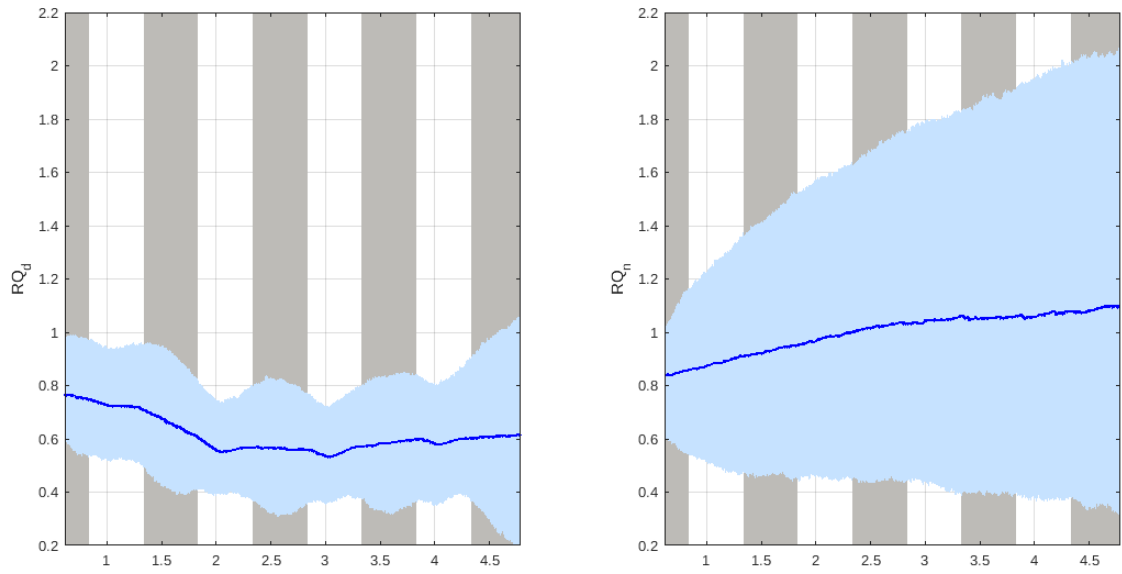


Figure 4.16 : Posterior medians (solid blue line) and 95% credible intervals (shaded blue) for RQ_d and RQ_n .

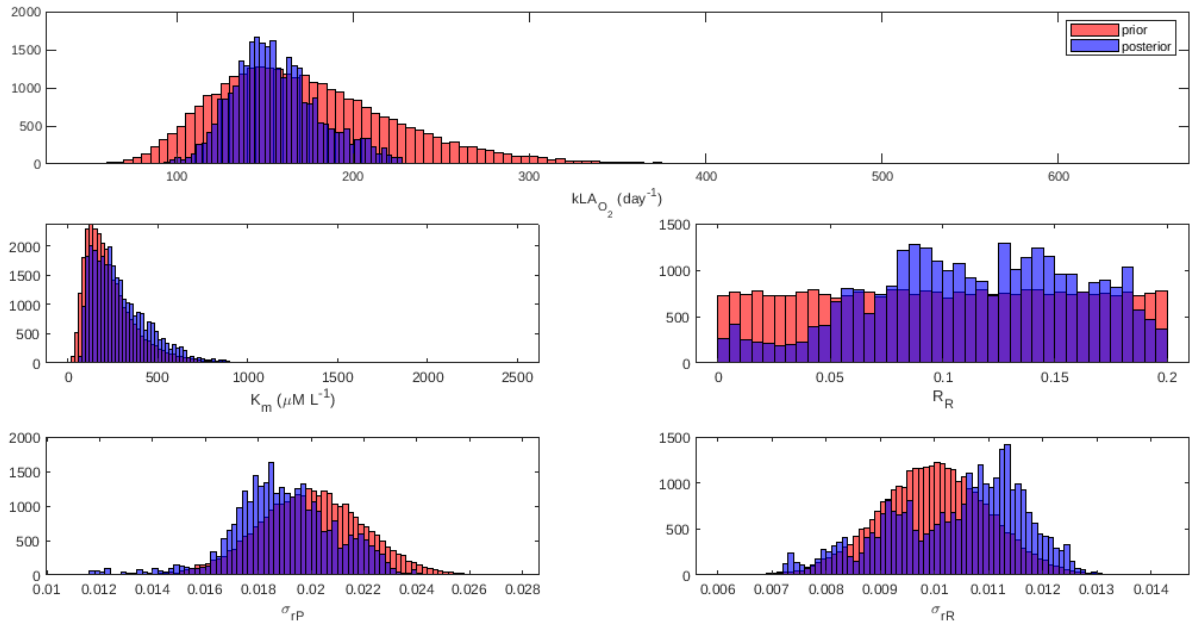


Figure 4.17 : Priors (pink) and posteriors (purple) for model parameters.

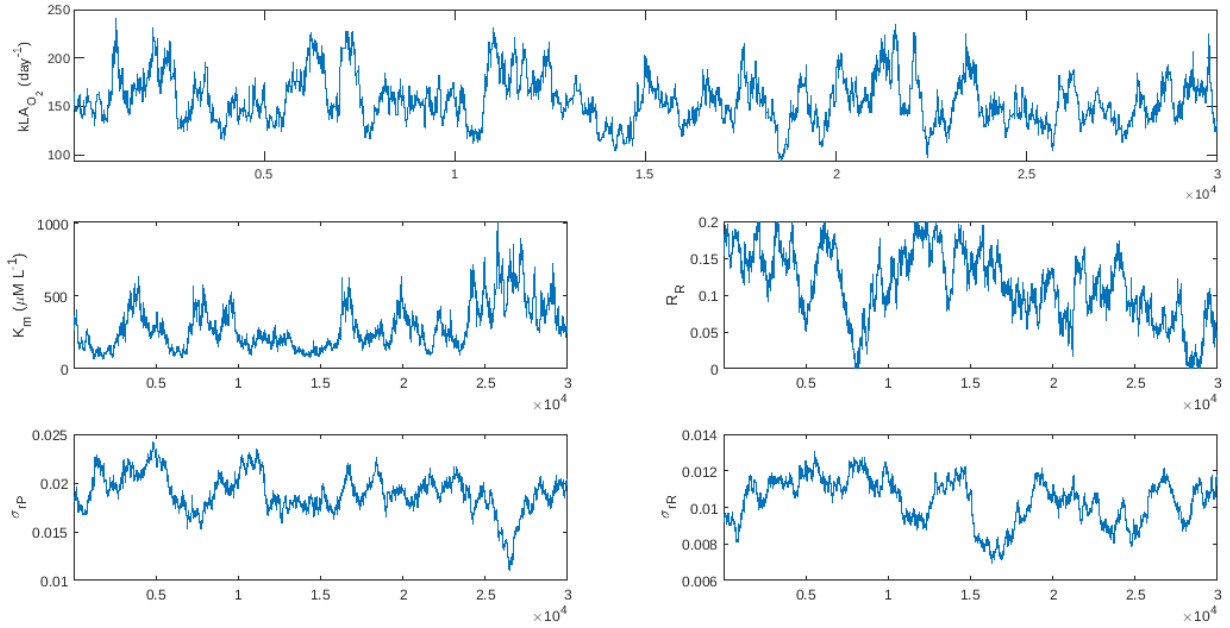


Figure 4.18 : Traces for model parameters.

observations well while the light was on, with all observations falling inside tightly constrained 95% credible intervals. During times when there was no light, most days the posteriors would fit the observations well to start and then potentially there was a sensor drift causing increasing observations that the model was not accounting for (Figure 4.12). pH captured most observations within the 95% credible intervals except for 2 points on the 1st day while the light was off (Figure 4.12).

$P_1 R_1$ (Figure 4.15).

The day-time respiratory quotient RQ_d centred around 0.75 during the first day, dropped to 0.6 during the 2nd, 3rd and 4th days. The night-time respiratory quotient RQ_n centred around 0.85 and slowly rose to 1.1 by the end of the experiment. [Values >1 are not really realistic but this can be a compensation for not having an offset on the O_2 obs]

4.3.4 Posteriors with experimental data (photosynthesis and respiration are random walks and the respiratory quotients are noisy states)

State posteriors are visualised by plotting the median and shading 95% credible intervals, while parameter priors and posteriors are displayed by histograms.

In this run:

Photosynthesis (P_1) and respiration (R_1) are both modelled as random walks, by taking P and R , previously constant parameters, and replacing them by $P_1(t)$ and $R_1(t)$. Here, we take $P_1(t)$ and $R_1(t)$ to be such that

$$P_1(t + \Delta t) = P(t) + r_P$$

$$R_1(t + \Delta t) = R(t) + r_R$$

where $r_P \sim N(0, \sigma_{r_P})$, $r_R \sim N(0, \sigma_{r_R})$, and Δt is the length of discrete time-step. For the purpose of the Bayesian analysis here, σ_{r_P} and σ_{r_R} are treated as parameters to be inferred. The respiratory quotients (RQ_d and RQ_n) are treated as normally distributed noisy states, truncated between 0.6 and 1, where the mean and standard deviation are unknown parameters to be estimated.

$$RQ_d \sim \text{truncated}\mathcal{N}(\mu_{RQ_d}, \sigma_{RQ_d}, \text{lower} = 0.6, \text{upper} = 1.0)$$

$$RQ_n \sim \text{truncated}\mathcal{N}(\mu_{RQ_n}, \sigma_{RQ_n}, \text{lower} = 0.6, \text{upper} = 1.0)$$

kLA_{O_2} , K_m , R_R , σ_{r_P} , σ_{r_R} , μ_{RQ_d} , σ_{RQ_d} , μ_{RQ_n} , σ_{RQ_n} are all treated as parameters constant through time but unknown.

Parameter	Prior	Proposal
kLA_{O_2}	$\text{Log}\mathcal{N}(\log(200.0), 0.3)$	$\text{Log}\mathcal{N}(\log(kLA_{O_2}), 0.03)$
K_m	$\text{Log}\mathcal{N}(\log(200.0), 0.6)$	$\text{Log}\mathcal{N}(\log(K_m), 0.06)$
R_R	$\text{Uniform}(0, 0.2)$	$\text{Trun}\mathcal{N}(R_R, 0.01, \text{lower} = 0, \text{upper} = 0.2)$
μ_{RQ_d}	$\text{Uniform}(0.6, 1)$	$\mathcal{N}(\mu_{RQ_d}, 0.01)$
μ_{RQ_n}	$\text{Uniform}(0.6, 1)$	$\mathcal{N}(\mu_{RQ_n}, 0.01)$
σ_{RQ_d}	$\text{Uniform}(0, 0.5)$	$\mathcal{N}(\sigma_{RQ_d}, 0.01)$
σ_{RQ_n}	$\text{Uniform}(0, 0.5)$	$\mathcal{N}(\sigma_{RQ_n}, 0.01)$
σ_{r_P}	$\mathcal{N}(0.01, 0.001)$	$\mathcal{N}(\sigma_{r_P}, 0.0001)$
σ_{r_R}	$\mathcal{N}(0.01, 0.001)$	$\mathcal{N}(\sigma_{r_R}, 0.0001)$
σ_{O_2}	0.4	*
σ_{pH}	0.4	*
σ_{DIC}	0.4	*

Table 4.9 : Table of Parameters, their priors and proposal distributions. * indicates the parameter was held fixed.

Results:

10,000 samples run (no burn-in was discarded) with 1024 particles.

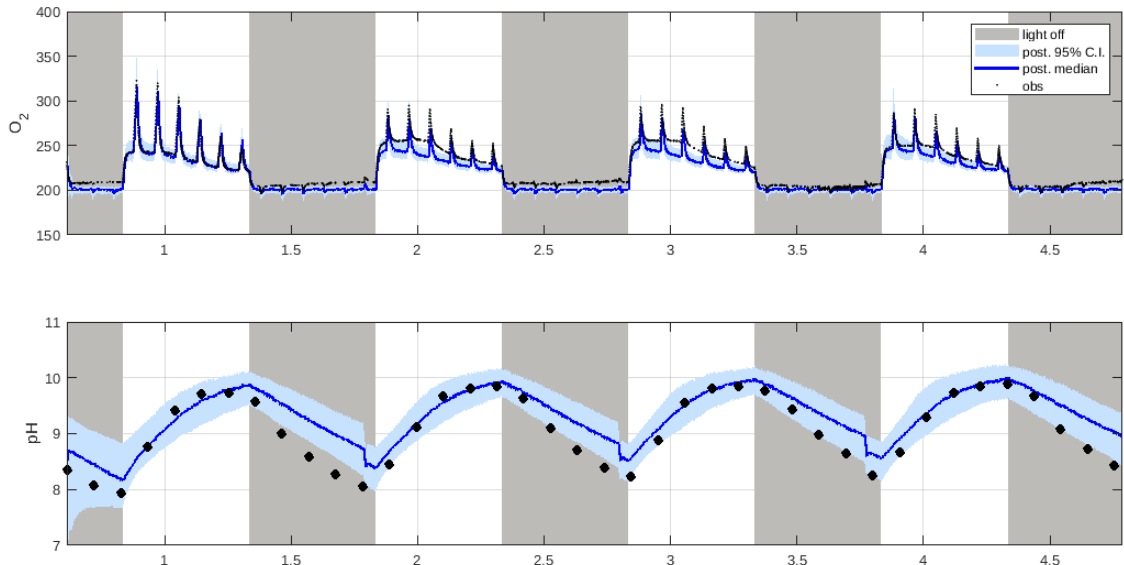


Figure 4.19 : Posterior medians (solid blue line), 95% credible intervals (shaded blue), and simulated observations (black) for O_2 and pH across 4 days.

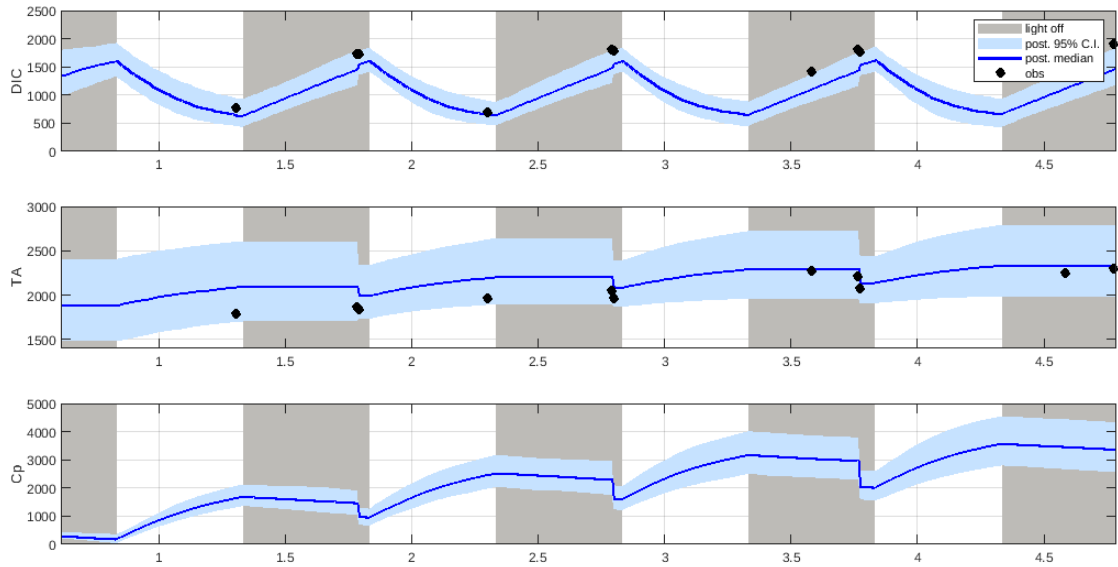


Figure 4.20 : Posterior medians (solid blue line), 95% credible intervals (shaded blue), and simulated observations (black) for DIC , TA and C_p across 4 days.

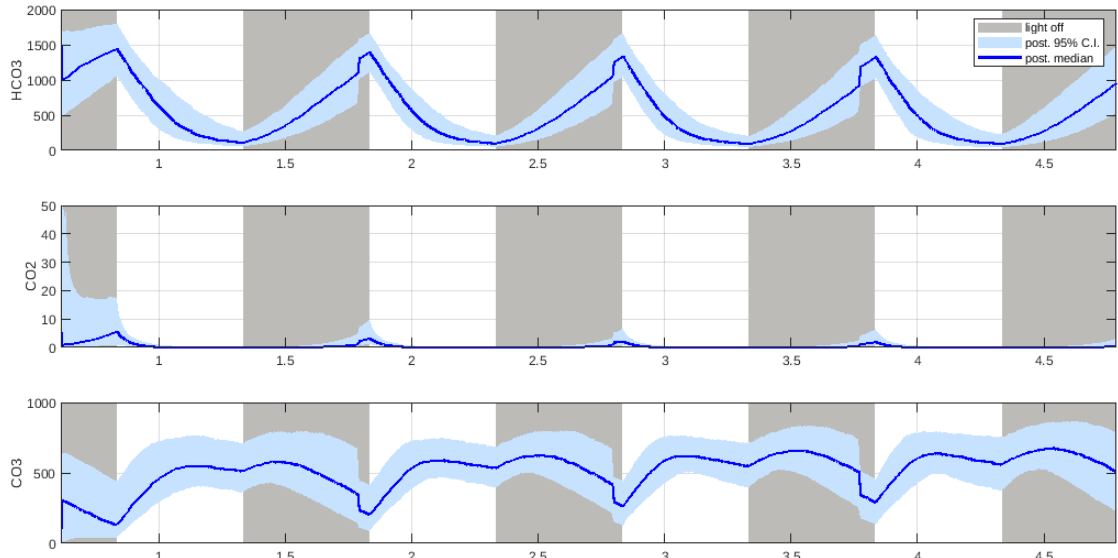


Figure 4.21 : Posterior medians (solid blue line) and 95% credible intervals (shaded blue) for HCO_3 , CO_2 and CO_3 across 4 days.

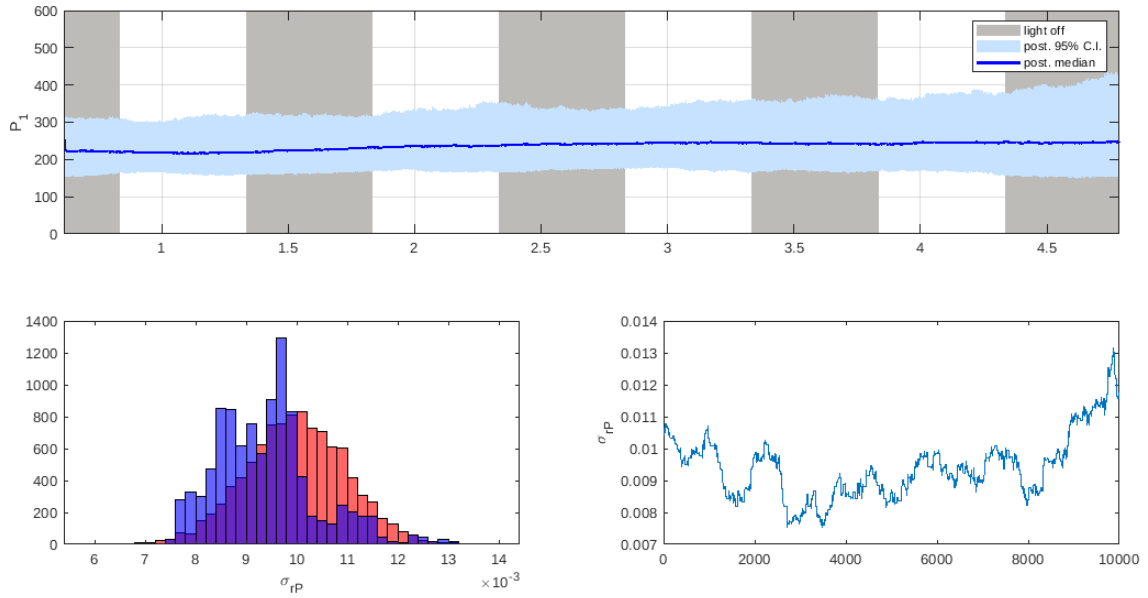


Figure 4.22 : Posterior medians (solid blue line) and 95% credible intervals (shaded blue) for photosynthesis P_1 .

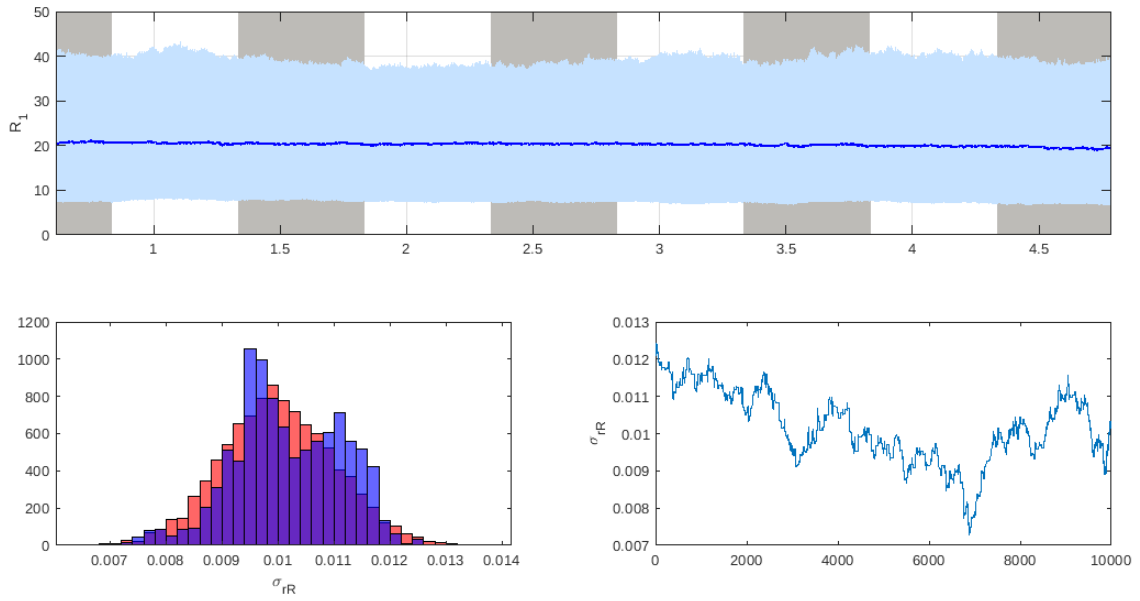


Figure 4.23 : Posterior medians (solid blue line) and 95% credible intervals (shaded blue) for respiration R_1 .

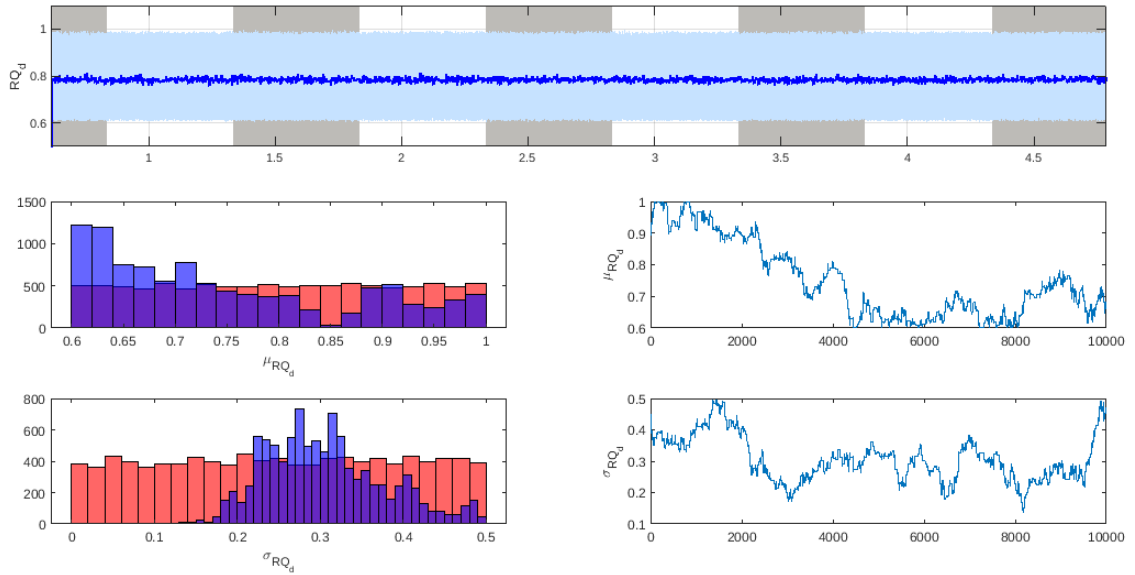


Figure 4.24 : Posterior medians (solid blue line) and 95% credible intervals (shaded blue) for RQ_d .

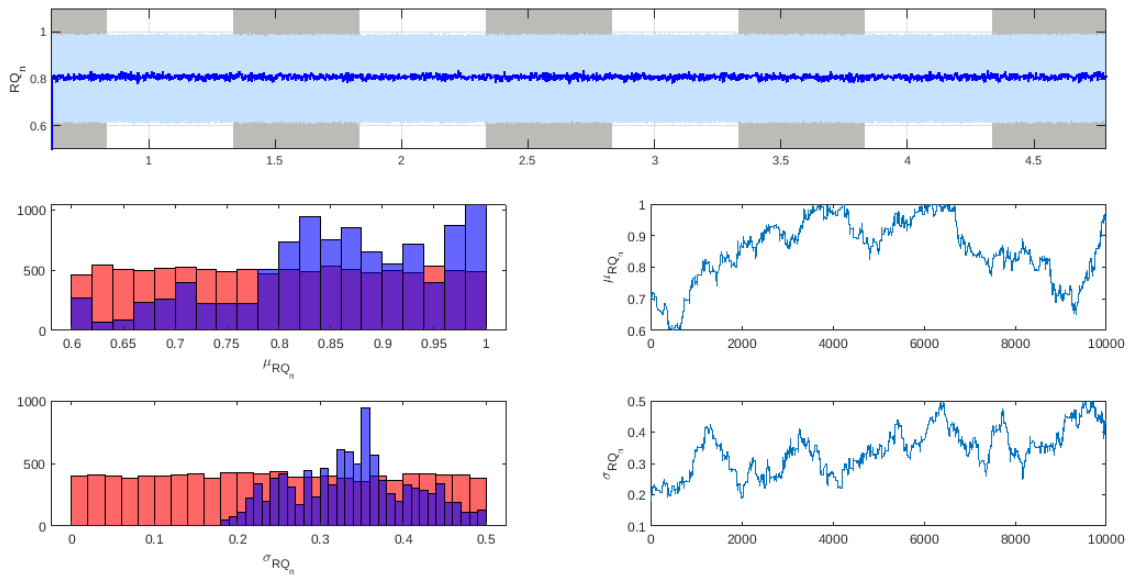


Figure 4.25 : Posterior medians (solid blue line) and 95% credible intervals (shaded blue) for RQ_n .

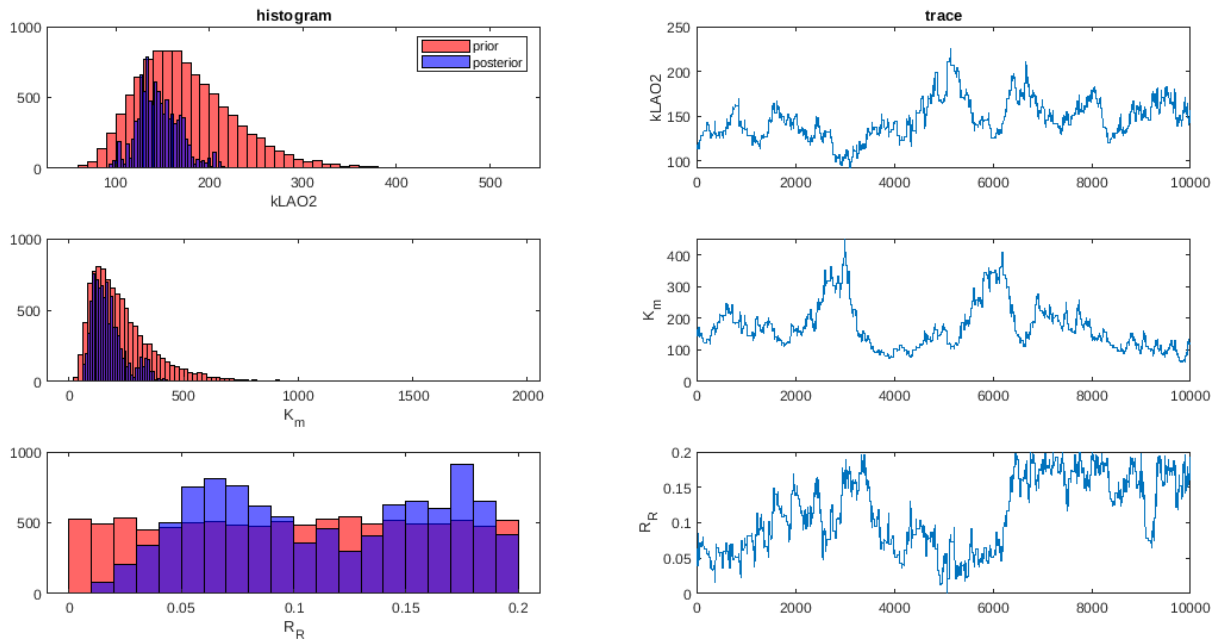


Figure 4.26 : Priors (pink) and posteriors (purple) for model parameters.

4.3.5 Posteriors with experimental data (estimating obs error)

State posteriors are visualised by plotting the median and shading 95% credible intervals, while parameter priors and posteriors are displayed by histograms.

In this run:

P_1 and R_1 are random walks with rP and rR as wiener processes. kLA_{O_2} , K_m , R_R , RQ_d , RQ_n , σ_{O_2} , σ_{pH} , and σ_{DIC} are all treated as parameters constant through time but unknown. The data model in this run is:

$$O2_{obs} \sim normal(O_2, \sigma_{O_2})$$

$$pH_{obs} \sim normal(pH, \sigma_{pH})$$

$$DIC_{obs} \sim normal(DIC, \sigma_{DIC})$$

$$TA_{obs} \sim normal(TA, \sigma_{DIC})$$

Parameter	Prior	Proposal
kLA_{O_2}	$\text{Log}\mathcal{N}(\log(200.0), 0.3)$	$\text{Log}\mathcal{N}(\log(kLA_{O_2}), 0.03)$
K_m	$\text{Log}\mathcal{N}(\log(200.0), 0.6)$	$\text{Log}\mathcal{N}(\log(K_m), 0.06)$
R_R	$\text{Uniform}(0, 0.2)$	$\text{Trun}\mathcal{N}(R_R, 0.01, \text{lower} = 0, \text{upper} = 0.2)$
RQ_d	$\text{Uniform}(0.6, 1)$	$\text{Trun}\mathcal{N}(RQ_d, 0.01, \text{lower} = 0.6, \text{upper} = 1.0)$
RQ_n	$\text{Uniform}(0.6, 1)$	$\text{Trun}\mathcal{N}(RQ_n, 0.01, \text{lower} = 0.6, \text{upper} = 1.0)$
σ_{O_2}	$\text{Uniform}(0, 20)$	$\text{Trun}\mathcal{N}(\sigma_{O_2}, 2, \text{lower} = 0, \text{upper} = 20)$
σ_{pH}	$\text{Uniform}(0, 1)$	$\text{Trun}\mathcal{N}(\sigma_{pH}, 0.1, \text{lower} = 0, \text{upper} = 1)$
σ_{DIC}	$\text{Uniform}(0, 150)$	$\text{Trun}\mathcal{N}(\sigma_{DIC}, 15, \text{lower} = 0, \text{upper} = 150)$

Table 4.10 : Table of Parameters, their priors and proposal distributions. * indicates the parameter was held fixed.

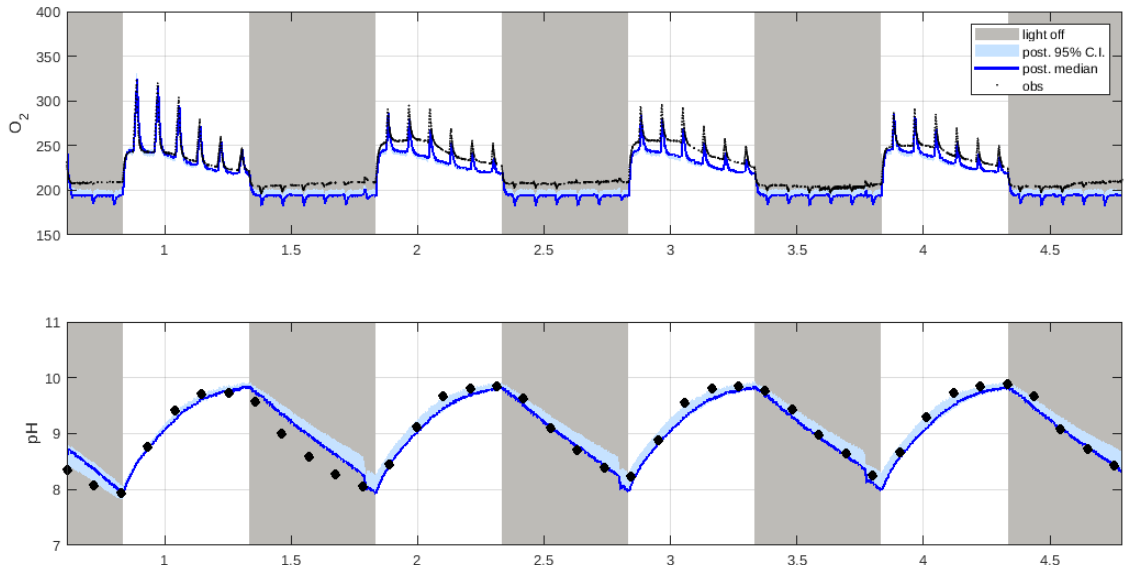


Figure 4.27 : Posterior medians (solid blue line), 95% credible intervals (shaded blue), and simulated observations (black) for O_2 and pH across 4 days.

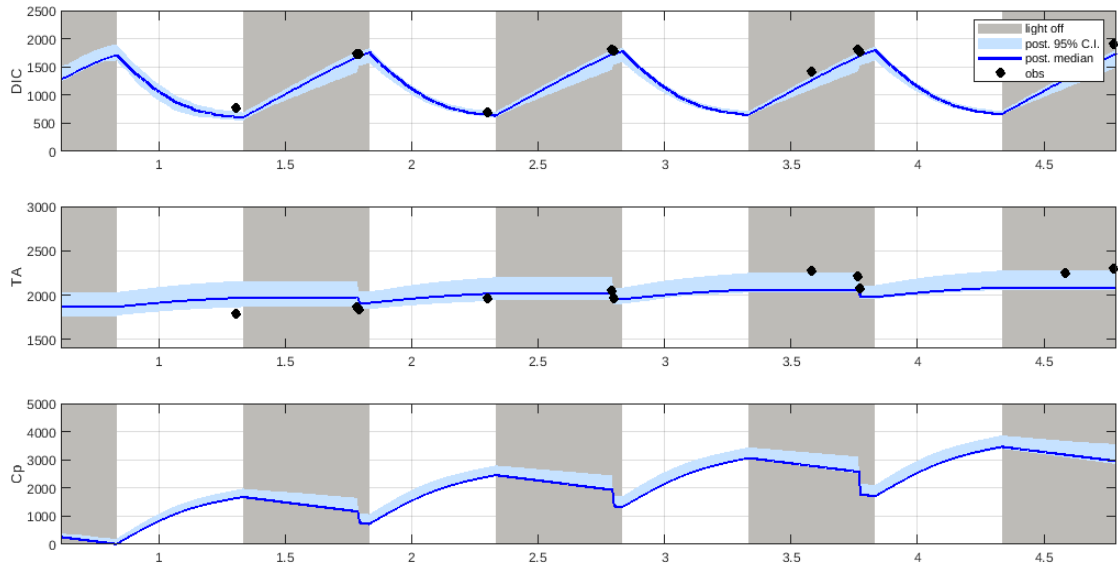


Figure 4.28 : Posterior medians (solid blue line), 95% credible intervals (shaded blue), and simulated observations (black) for DIC , TA and C_p across 4 days.

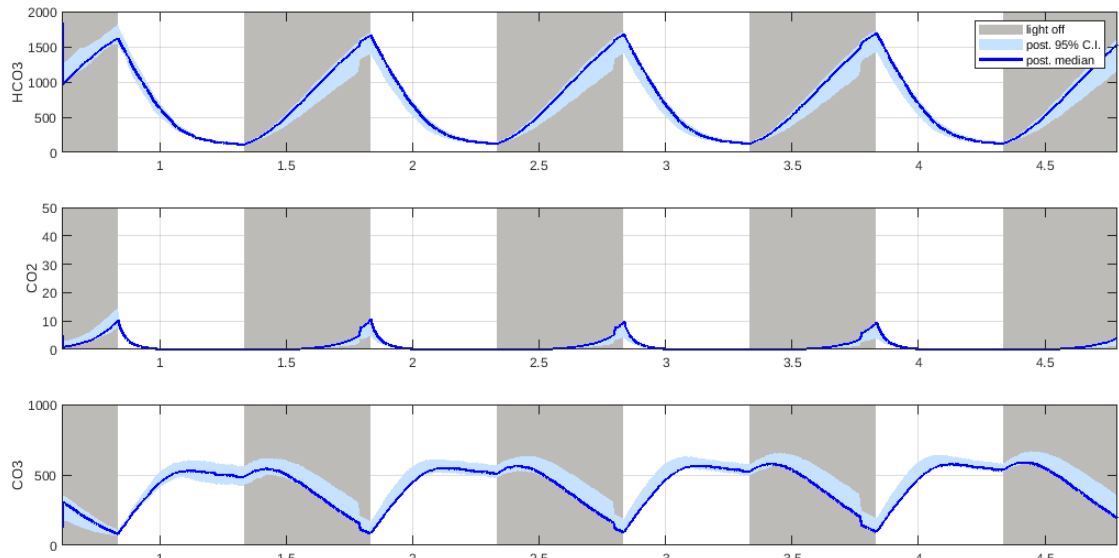


Figure 4.29 : Posterior medians (solid blue line) and 95% credible intervals (shaded blue) for HCO_3 , CO_2 and CO_3 across 4 days.

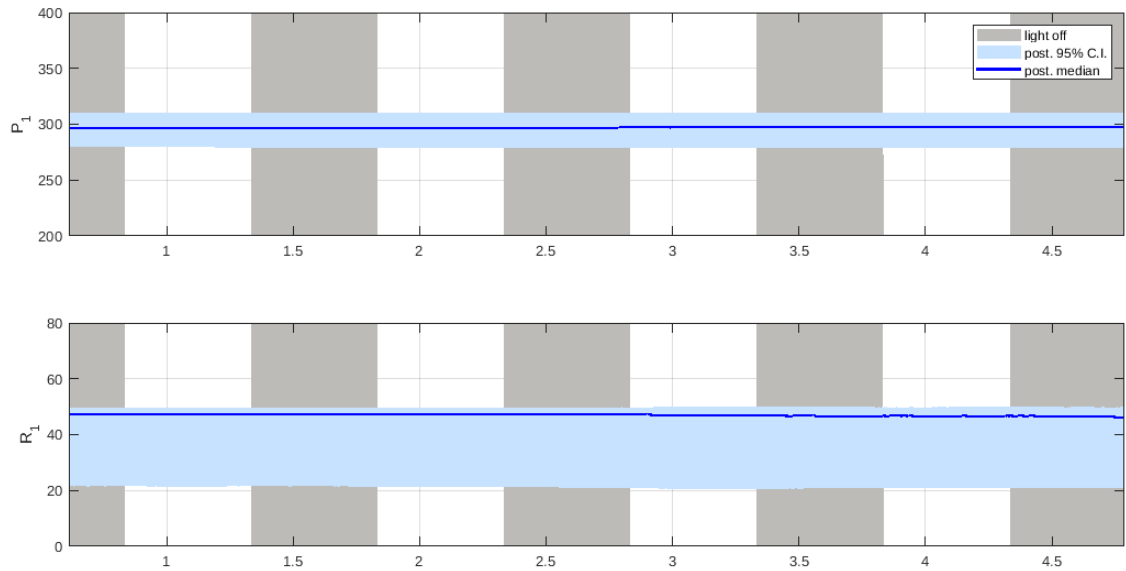


Figure 4.30 : Posterior medians (solid blue line) and 95% credible intervals (shaded blue) for photosynthesis P_1 .

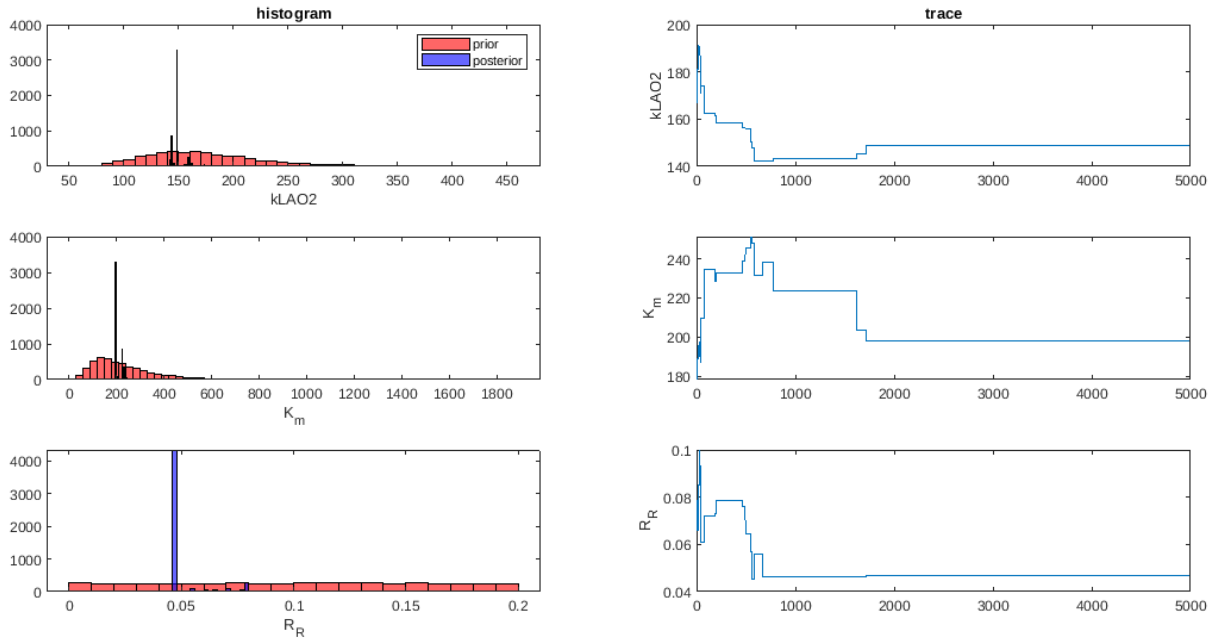


Figure 4.31 : Priors (pink) and posteriors (purple) for model parameters.

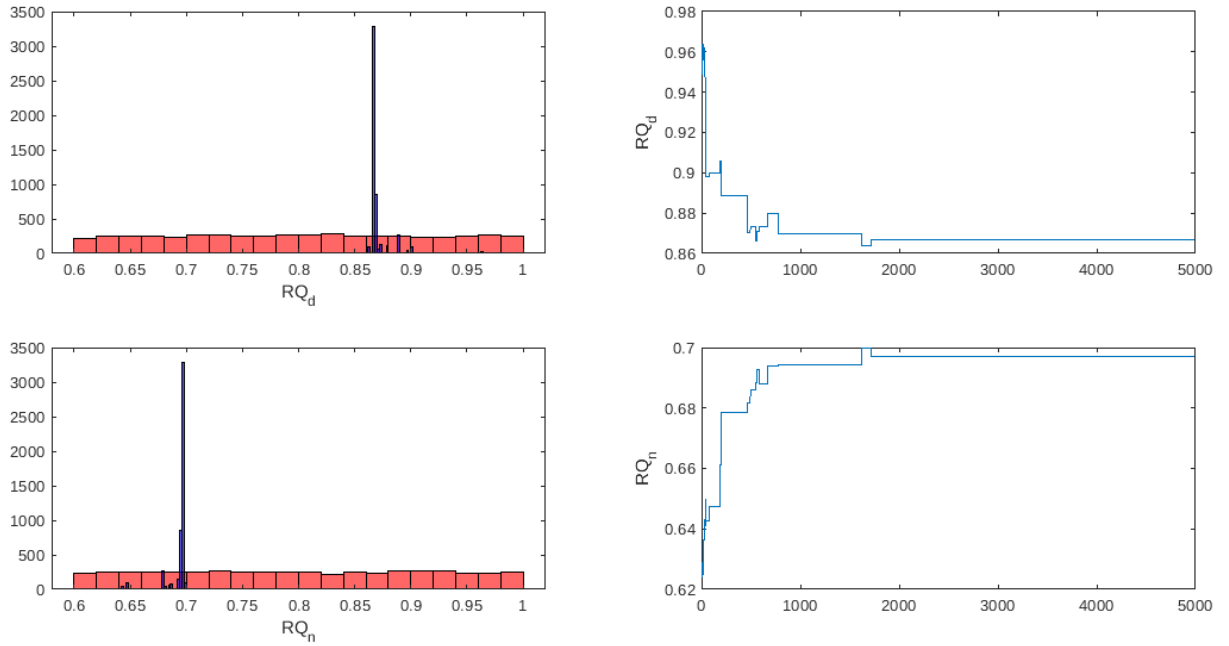


Figure 4.32 : Priors (pink) and posteriors (purple) for model parameters.

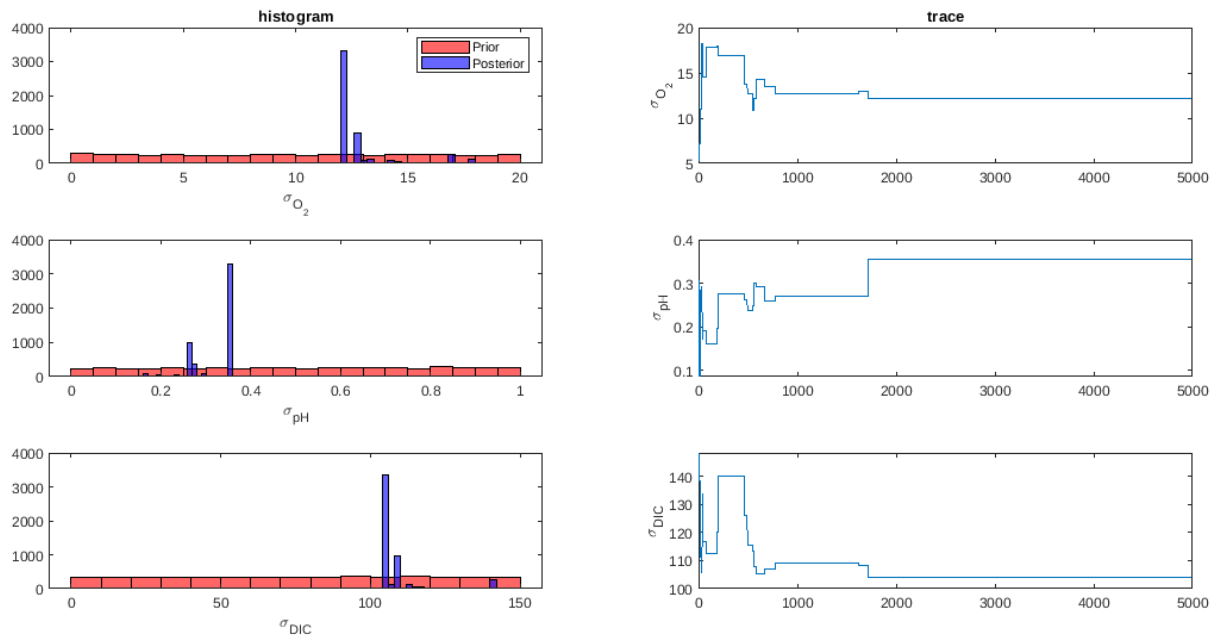


Figure 4.33 : Priors (pink) and posteriors (purple) for model parameters.

Appendix A

LiBbi model code

LiBbi model file: micro_iterative.bi

```

1 model micro_iterative {
2
3   const F02          = 0.2094
4   const FC02         = 397e-6
5   const S            = 34.0
6   const V            = 500.0           // volume of the
      reactor
7   const DIC_M        = 1724.20         // calculated with
      CO2SYS[ DIC_M = 1724.20, Alk = 1797.90, T = 27, S = 34]
8   const O_2_M        = 226.65
9   const alk_M         = 1797.90
10  const tau           = 6.0
11  const kLA02_m       = log(2.0)*24.0*60.0/tau
12
13  param kLA02
14  param Km
15  param RR
16  param RQ_d
17  param RQ_n
18  param sigma_O_2
19  param sigma_pH
20  param sigma_DIC

```

```

21 param offset_0_2
22
23 input I           // light intensity
24 input T           // temperature (C)
25 input gas         // gas on/off
26 input dil         // dilution rate
27
28 state DIC // state variables
29 state O_2
30 state pH
31 state Cp
32 state mich_ment
33 state O2H_pr
34 state CO2H_pr
35 state R
36 state R1
37 state P
38 state P1
39 state alk
40 state CO2
41 state HCO3
42 state CO3
43 state O_2H
44 state CO2H
45 state h_3
46 state h_free_3
47
48 noise r_R
49 noise r_P

```

```

50
51 /* random walk parameter */
52 param sigma_r_R
53 param sigma_r_P
54
55 obs 02_obs
56 obs pH_obs
57 obs DIC_obs
58 obs alk_obs
59
60 sub parameter { /* prior distribution over parameters */
61 Km      ~ log_normal(log(100.0), 0.5)
62 kLA02   ~ log_normal(log(kLA02_m), 0.3)
63 RR      ~ uniform(0.0001, 0.2)
64 RQ_d    ~ uniform(0.66, 1.0)
65 RQ_n    ~ uniform(0.66, 1.0)
66
67 sigma_0_2 ~ log_normal(log(0.03), 0.5)
68 sigma_pH  ~ log_normal(log(0.03), 0.5)
69 sigma_DIC ~ log_normal(log(0.03), 0.5)
70
71 offset_0_2 ~ normal(0, 2.0)
72
73 sigma_r_R      ~ normal(0.01, 0.001)
74 sigma_r_P      ~ normal(0.05, 0.01)
75 }
76
77 const prop_std = 0.1;
78 sub proposal_parameter {

```

```

79 Km      ~ log_normal(log(Km), 0.5*prop_std)
80 kLA02    ~ log_normal(log(kLA02), 0.3*prop_std)
81 RR      ~ truncated_normal(RR, 0.2*prop_std, lower = 0.0001,
      upper = 0.2)
82 RQ_d     ~ truncated_normal(RQ_d, 0.2*prop_std, lower = 0.66,
      upper = 1.0)
83 RQ_n     ~ truncated_normal(RQ_n, 0.2*prop_std, lower = 0.66,
      upper = 1.0)
84
85
86 sigma_0_2 ~ log_normal(log(sigma_0_2), 0.5*prop_std)
87 sigma_pH  ~ log_normal(log(sigma_pH), 0.5*prop_std)
88 sigma_DIC ~ log_normal(log(sigma_DIC), 0.5*prop_std)
89
90 offset_0_2 ~ normal(offset_0_2, 2.0*prop_std)
91
92 sigma_r_R ~ normal(sigma_r_R, 0.001*prop_std)
93 sigma_r_P ~ normal(sigma_r_P, 0.01*prop_std)
94 }
95
96 sub initial {/* prior distribution over initial conditions,
      given parameters */
97 // specify the initial condition model
98 R      ~ normal(log(20.0), 0.4)
99 R1     ~ log_normal(log(20.0), 0.4)
100 P      ~ normal(log(200.0), 0.4)
101 P1     ~ log_normal(log(200.0), 0.4)
102
103 Cp     ~ log_normal(log(300.0), 0.2)

```

```

104 alk      ~ log_normal(log(1750.0), 0.1)
105 DIC      ~ log_normal(log(1300.0), 0.2)
106 O_2      ~ log_normal(log(225.0), 0.2)
107 pH       ~ log_normal(log(8.5), 0.2)
108 CO2      ~ log_normal(log(3.0), 0.4)
109 HCO3     ~ log_normal(log(1000.0), 0.3)
110 CO3      ~ log_normal(log(300.0), 0.4)
111 O_2H     ~ log_normal(log(200.0), 0.2)
112 CO2H     ~ log_normal(log(10.0), 0.2)
113 }
114
115
116 //sub transition(delta = 0.0023) { // obs are in days ie
      delta=1.0 for daily solving. delta=0.00069 for solving
      every minute, 0.0014 for every 2 mins, 0.0021 for 3 mins,
      0.0028 for 4mins, delta=0.000011574 for solving every
      second
117 sub transition(delta = 0.0021) {
118
119 /* processes */
120
121 inline TK      = T + 273.15           // temp in kelvin
122 inline K0_CO2 = exp(-60.2409 + 93.4517*(100.0/TK) + 23.3585*
      log(TK/100.0)+ S*(0.023517 - 0.023656*(TK/100) +
      0.0047036*(TK/100.0)*(TK/100.0)))
123 CO2H          <- K0_CO2*FCO2*1.0220*1e6
124
125 inline K0_O2   = (exp(-1282.8704 + 36619.96/TK + 223.1396*log
      (TK) -0.354707*TK + S*(5.957e-3 -3.7353/TK) + 3.68e-6*S*S)

```

```

    )/(0.2094e-06)

126 O_2H          <- K0_02*F02*1.0220*1e-6
127
128 inline PAC      = HC03          //PAC=photosynthetically
    active carbon. if the phyto are just using CO2 to
    photosynthesise then PAC=CO2
129 inline mm       = PAC/(Km + PAC)
130
131 // CO2SYS iterative solution
132 // set up all the constants
133
134 inline logTK     = log(TK)
135 inline S2        = S*S
136 inline sqrtS     = sqrt(S)
137
138 // total sulphur
139
140 inline TS        = (0.14/96.062)*(S/1.80655)
141 inline IS        = 19.924*S/(1000.0 - 1.005*S)
142
143 inline KS_int    = -4276.1/TK + 141.328 - 23.093*logTK +
    (-13856.0/TK + 324.57 - 47.986*logTK)*sqrt(IS) + (
    35474.0/TK - 771.54 + 114.723*logTK)*IS - 2698.0/TK*IS
    **1.5 + 1776.0/TK*IS**2
144 inline KS       = exp(KS_int)*(1 - 0.001005*S)
145
146 // Fluorine
147
148 inline TF        = 0.000067*S/18.9984/1.80655

```



```

149 inline KF          = exp(-(-874.0/TK - 0.111*sqrtS + 9.68))
150 inline SWS_2_T      = (1.0 + TS/KS)/(1.0 + TS/KS + TF/KF)
151 inline Free_2_T     = 1.0 + TS/KS
152
153 // H2O dissoci
154
155 inline KW = exp(148.9802 - 13847.26/TK - 23.6521*logTK +
    (118.67/TK - 5.977 + 1.0495*logTK)*sqrtS - 0.01615*S)
156
157 // Boron
158
159 inline KB = exp((-8966.90 - 2890.53*sqrtS - 77.942*S + 1.728*
    S*sqrtS - 0.0996*S2)/TK + 148.0248 + 137.1942*sqrtS +
    1.62142*S - (24.4344 + 25.085*sqrtS + 0.2474*S)*logTK +
    0.053105*sqrtS*TK)
160 inline TB = 0.0004326*S/35.0
161
162 // Carbon eq constants
163
164 inline K1 = 10**(-(3633.86/TK - 61.2172 + 9.6777 *logTK -
    0.011555*S + 0.0001152*S**2))*1.23 //1.23 experiment
    specific and measured
165 inline K2 = 10**(-( 471.8/TK + 25.9290 - 3.16967*logTK -
    0.01781*S + 0.0001122*S**2))*0.53 //0.53 experiment
    specific and measured
166
167 // end all the constants
168
169 // intial guess at the pH (use the approximating equation)

```

```

170
171 inline pH_init = 12.26 -0.0030605*DIC -0.043752*T -0.013625*S
      + 0.00011315*alk + 1.3463e-05*DIC*T + 5.2215e-07*DIC*alk
172
173 // iteration 1
174
175 inline h_1      = 10.0**(-pH_init)
176 inline h_free_1 = h_1/Free_2_T
177 inline f0_1     = (DIC*1e-6*(K1*h_1 + 2.0*K1*K2)/(h_1*h_1 +
      K1*h_1 + K1*K2) - h_free_1 + KW/h_1 - alk*1e-6 + TB/(1.0 +
      h_1/KB))*1e6
178 inline df0_1    = (DIC*1e-6*(K1 + 2.0*K1*K2)/(h_1**2.0 + K1*
      h_1 + K1*K2) - DIC*1e-6*(K1*h_1 + 2.0*K1*K2)/(h_1**2.0 +
      K1*h_1 + K1*K2)**2.0*(2.0*h_1 + K1) - TB*1.0/(1.0 + h_1/KB
      )**2.0/KB - KW/h_1**2.0 - 1.0/Free_2_T)*1e6*(-log(10.0)
      *10.0**(-pH_init))
179 inline pH_1     = pH_init - f0_1/df0_1
180
181 // iteration 2
182
183 inline h_2      = 10.0**(-pH_1)
184 inline h_free_2 = h_2/Free_2_T
185 inline f0_2     = (DIC*1e-6*(K1*h_2 + 2.0*K1*K2)/(h_2*h_2 +
      K1*h_2 + K1*K2) - h_free_2 + KW/h_2 - alk*1e-6 + TB/(1.0 +
      h_2/KB))*1e6
186 inline df0_2    = (DIC*1e-6*(K1 + 2.0*K1*K2)/(h_2**2.0 + K1*
      h_2 + K1*K2) - DIC*1e-6*(K1*h_2 + 2.0*K1*K2)/(h_2**2.0 +
      K1*h_2 + K1*K2)**2.0*(2.0*h_2 + K1) - TB*1.0/(1.0 + h_2/KB
      )**2.0/KB - KW/h_2**2.0 - 1.0/Free_2_T)*1e6*(-log(10.0)

```

```

      *10.0**(-pH_1))
187 inline pH_2      = pH_1 - f0_2/df0_2
188
189 // iteration 3
190
191 h_3                <- 10.0**(-pH_2)
192 h_free_3           <- h_3/Free_2_T
193 inline f0_3        = (DIC*1e-6*(K1*h_3 + 2.0*K1*K2)/(h_3*h_3 +
      K1*h_3 + K1*K2) - h_free_3 + KW/h_3 - alk*1e-6 + TB/(1.0 +
      h_3/KB))*1e6
194 inline df0_3       = (DIC*1e-6*(K1 + 2.0*K1*K2)/(h_3**2.0 + K1*
      h_3 + K1*K2) - DIC*1e-6*(K1*h_3 + 2.0*K1*K2)/(h_3**2.0 +
      K1*h_3 + K1*K2)**2.0*(2.0*h_3 + K1) - TB*1.0/(1.0 + h_3/KB
      )**2.0/KB - KW/h_3**2.0 - 1.0/Free_2_T)*1e6*(-log(10.0)
      *10.0**(-pH_2))
195 pH                <- pH_2 - f0_3/df0_3
196
197 // iteration 4
198
199 //      inline h_4      = 10.0**(-pH_3)
200 //      inline h_free_4 = h_4/Free_2_T
201 //      inline f0_4      = (DIC*1e-6*(K1*h_4 + 2.0*K1*K2)/(h_4
      *h_4 + K1*h_4 + K1*K2) - h_free_4 + KW/h_4 - alk*1e-6 + TB
      /(1.0 + h_4/KB))*1e6
202 //      inline df0_4     = (DIC*1e-6*(K1 + 2.0*K1*K2)/(h_4
      **2.0 + K1*h_4 + K1*K2) - DIC*1e-6*(K1*h_4 + 2.0*K1*K2)/(
      h_4**2.0 + K1*h_4 + K1*K2)**2.0*(2.0*h_4 + K1) - TB
      *1.0/(1.0 + h_4/KB)**2.0/KB - KW/h_4**2.0 - 1.0/Free_2_T)
      *1e6*(-log(10.0)*10.0**(-pH_3))

```

```

203 //      inline pH_4      = pH_3 - f0_4/df0_4
204
205 //      pH                <- pH_4
206
207 // calculate the final concentrations
208
209 inline H      = 10.0**(-pH)
210 inline H2     = H*H
211 inline denom  = (H2 + K1*H + K1*K2)
212 CO2          <- DIC*H2/denom
213 HCO3         <- DIC*H*K1/denom
214 CO3          <- DIC*K1*K2/denom
215
216 // end CO2SYS iterative solution
217
218
219 /* R and P as random walks */
220
221 r_R      ~ normal(0.0, sigma_r_R)
222 R        <- R + r_R
223 R1       <- exp(R)
224
225 r_P      ~ normal(0.0, sigma_r_P)
226 P        <- P + r_P
227 P1       <- exp(P)
228
229 ode(h = 0.1, atoler = 1.0e-6, rtoler = 1.0e-6, alg = 'RK4(3)
      '){

```

```

230 dDIC/dt = -P1*24.0*I*mm + R1*24.0
          + gas*0.893*kLA02*(C02H - C02)          + dil/V*(
          DIC_M - DIC)
231 dO_2/dt = (P1*24.0*I*mm - R1*24.0)/(RQ_d*I + RQ_n*(1.0-I))
          + gas*kLA02*(O_2H - O_2)          + dil/V*(O_2_M
          - O_2) + offset_O_2
232 dalk/dt = RR*P1*24.0*I*mm
          + dil/V*(
          alk_M - alk)
233 dCp/dt = (P1*24.0*I*mm - R1*24.0)
          + dil/V*(
          Cp)
234
235 }
236
237 mich_ment <- mm
238 O2H_pr    <- O_2H
239 C02H_pr   <- C02H
240
241 }
242
243
244 sub observation {
245
246 O2_obs ~ log_normal(log(O_2), sigma_O_2)
247 pH_obs ~ log_normal(log(pH), sigma_pH)
248 DIC_obs ~ log_normal(log(DIC), sigma_DIC)
249 alk_obs ~ log_normal(log(alk), sigma_DIC)
250 }

```

251 }

LiBbi prior sampling file: prior.conf

```
1 --target prior
2 --model-file micro_iterative.bi
3 --nsamples 500
4 --start-time 0.61304
5 --end-time 4.7866
6 --noutputs 6049
7 --input-file data/input_all_2018_normalised.nc
8 --output-file results/prior_micro_iterative.nc
```

LiBbi posterior sampling file: posterior.conf

```
1 --target posterior
2 --model-file micro_iterative.bi
3 --input-file data/input_all_2018_normalised.nc
4 --obs-file data/obs_all_2018.nc
5 --nsamples 500
6 --nparticles 1024
7 --start-time 0.61304
8 --end-time 4.7866
9 --noutputs 6049
10 --output-file results/posterior_micro_iterative.nc
11 --with-transform-initial-to-param
```

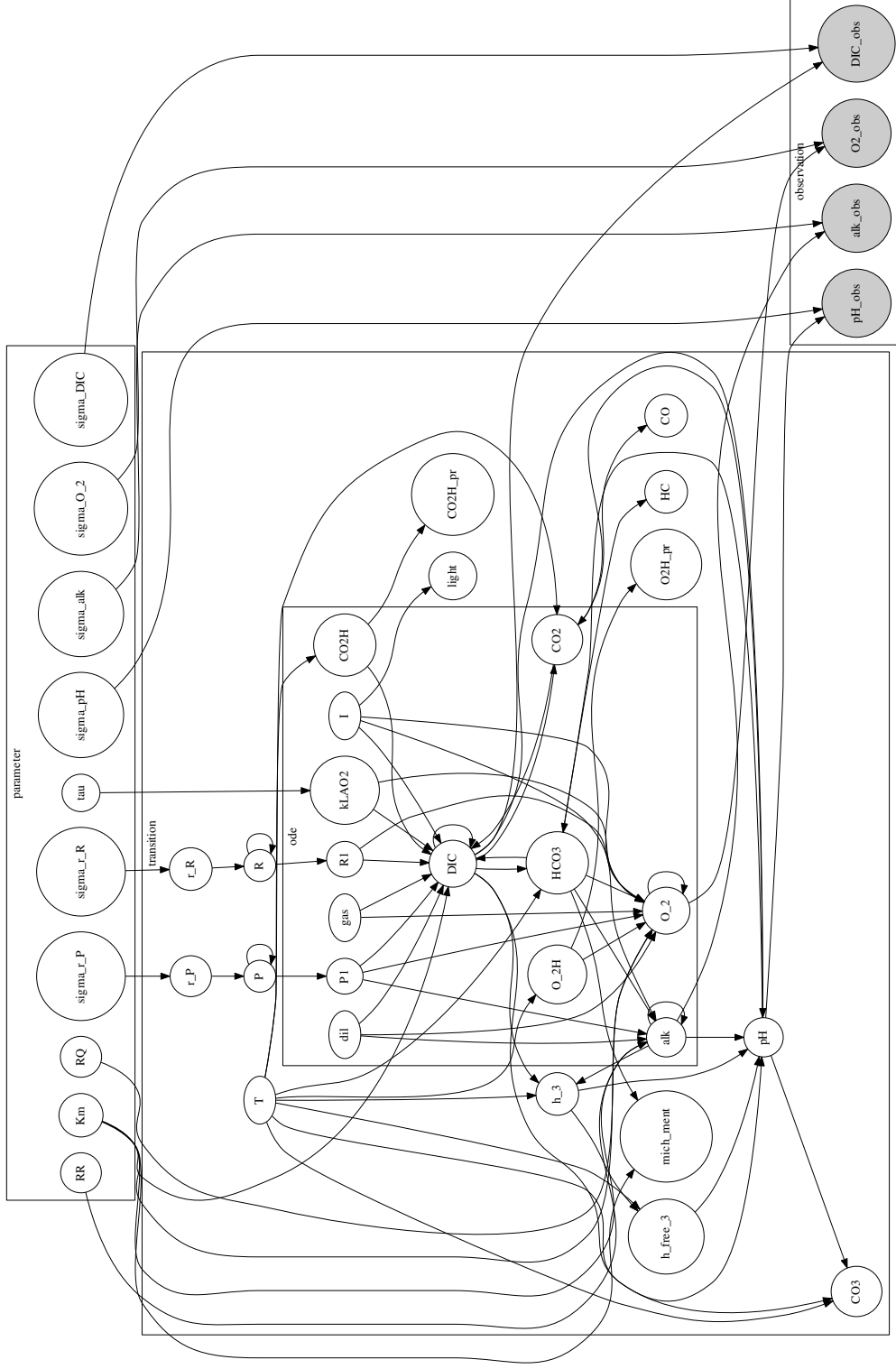


Figure A.1 : Directed Acyclic Graph of the LiBbi model file `micro_iterative.bi`

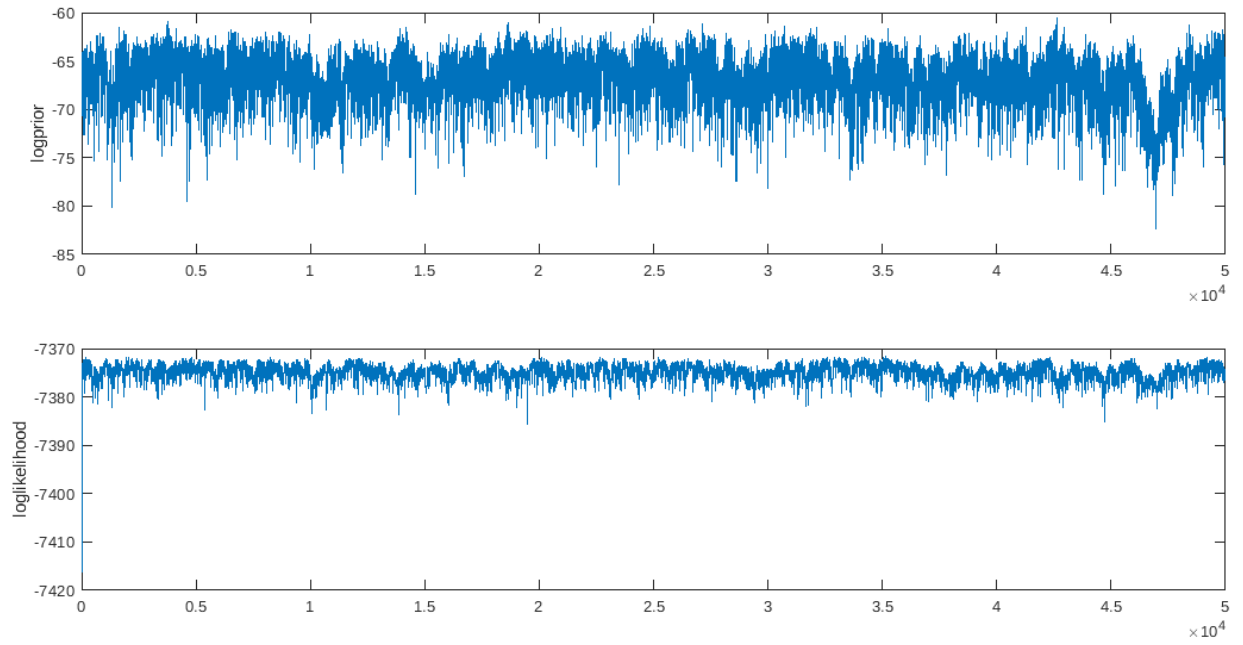


Figure A.2 : Log-prior and log-likelihood for the simulated data experiment.

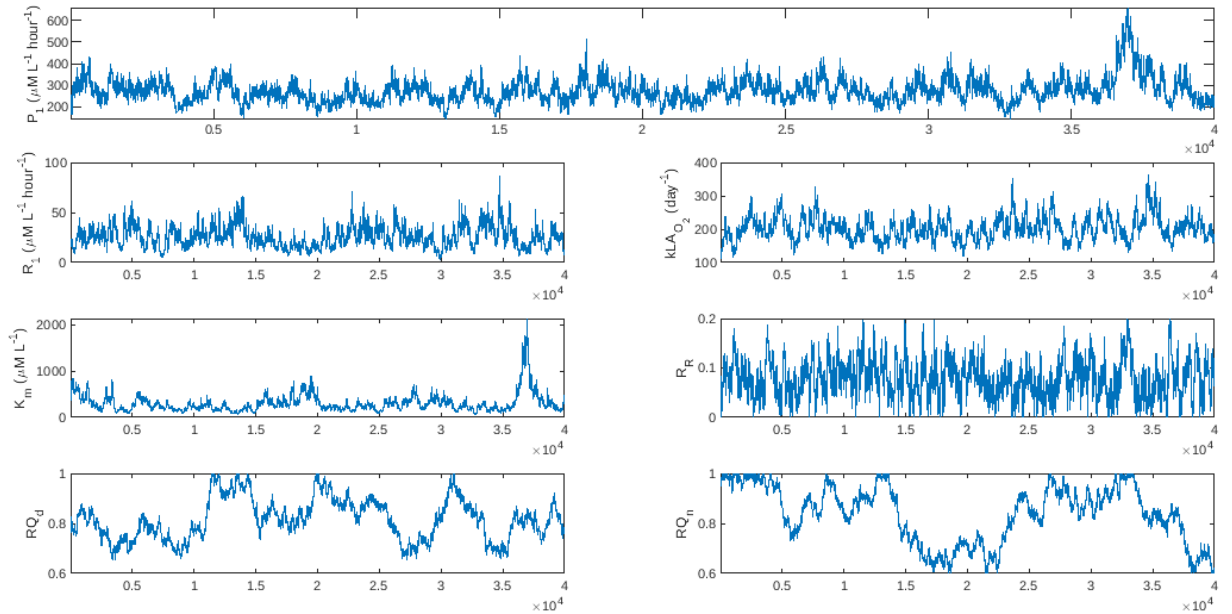


Figure A.3 : Parameter posterior traces for the simulated data experiment.

Bibliography

- [1] Rubin Battino, Timothy R Rettich, and Toshihiro Tominaga. The solubility of oxygen and ozone in liquids. *Journal of physical and chemical reference data*, 12(2):163–178, 1983.
- [2] AG Dickson and FJ Millero. A comparison of the equilibrium constants for the dissociation of carbonic acid in seawater media. *Deep Sea Research Part A. Oceanographic Research Papers*, 34(10):1733–1743, 1987.
- [3] Andrew Gilmore Dickson, Christopher L Sabine, and James Robert Christian. *Guide to best practices for ocean CO₂ measurements*. Number PICES Special Publication 3. North Pacific Marine Science Organization, 2007.
- [4] Arnold E Greensberg, LS Clesceri, Andrew D Eaton, and MAH Franson. Standard methods for the examination of water and wastewater. *American Public Health Asociation, Whashington, DC*, 1992.
- [5] E Molina Grima, JA Sánchez Pérez, F Ía Garc Camacho, and A Robles Medina. Gas-liquid transfer of atmospheric co₂ in microalgal cultures. *Journal of Chemical Technology & Biotechnology*, 56(4):329–337, 1993.
- [6] Robert RL Guillard and John H Ryther. Studies of marine planktonic diatoms: I. cyclotella nana hustedt, and detonula confervacea (cleve) gran. *Canadian journal of microbiology*, 8(2):229–239, 1962.
- [7] Ernie Lewis, Doug Wallace, and Linda J Allison. Program developed for co {sub 2} system calculations. Technical report, Brookhaven National Lab., Dept. of Applied Science, Upton, NY (United States , 1998.

- [8] Carl Mehrbach, CH Culberson, JE Hawley, and RM Pytkowicz. Measurement of the apparent dissociation constants of carbonic acid in seawater at atmospheric pressure¹. *Limnology and Oceanography*, 18(6):897–907, 1973.
- [9] Niels Ramsing and Jens Gundersen. Seawater and gases. *Limnol. Oceanogr*, 37:1307–1312, 2011.
- [10] R.F Weiss. Carbon dioxide in water and seawater: the solubility of a non-ideal gas. *Marine chemistry*, 2(3):203–215, 1974.
- [11] Richard E Zeebe and Dieter Wolf-Gladrow. *CO₂ in seawater: equilibrium, kinetics, isotopes*. Number 65. Gulf Professional Publishing, 2001.