

Thermodynamics & Statistical Physics

Chapter 6. The most probable distribution of nearly independent particles

Yuan-Chuan Zou
zouyc@hust.edu.cn

School of Physics, Huazhong University of Science and Technology

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Table of contents

- ① §6. Most probable distribution of nearly independent particles
 - 6.1 Classical description of particle's movement
 - 6.2 Quantum description of particle's movement
 - 6.3 Description of the system's microscopic arrangement
 - 6.4 The principle of equal a priori probabilities
 - 6.5 Distribution and microscopic state
 - 6.6 Boltzmann distribution
 - 6.7 Bose distribution and Fermi distribution
 - 6.8 Relations between Boltzmann, Bose and Fermi distributions

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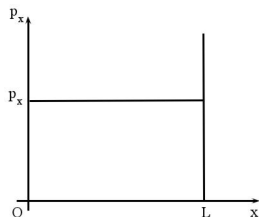
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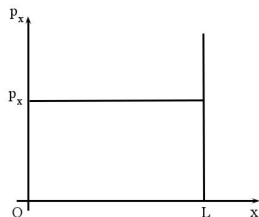
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- $p_x = \frac{h}{L}n_x$, there is one state for each n_x .
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E.g., electrons, $s = 1/2$, degeneracy is $2s + 1 = 2$.

Table of contents

- ① §6. Most probable distribution of nearly independent particles
 - 6.1 Classical description of particle's movement
 - 6.2 Quantum description of particle's movement
 - 6.3 Description of the system's microscopic arrangement
 - 6.4 The principle of equal a priori probabilities
 - 6.5 Distribution and microscopic state
 - 6.6 Boltzmann distribution
 - 6.7 Bose distribution and Fermi distribution
 - 6.8 Relations between Boltzmann, Bose and Fermi distributions

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- Aim: find the number of microscopic arrangement.

Description of the system's microscopic arrangement

Classical particle system

State 1	State 2	State 3
A B		
	A B	
		A B
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Fermion system

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Table of contents

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6.4 The principle of equal a priori probabilities

- Given an isolated system in equilibrium, it is found with equal probability in each of its accessible micro-states.

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- $\{a_l\}$ represents one kind of distribution (contains multiple micro-state), obeys: $\sum a_l = N, \sum a_l \varepsilon_l = E$.
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Distribution and microscopic state for Boltzmann system

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4. Total number of micro-states for one distribution $\{a_l\}$:

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- In the classical limit, both F.D. and B.E. reach to the same number of micro-states $\frac{\Omega_{\text{M.B.}}}{N!}.$

Classical statistics

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energy	$\varepsilon_1, \varepsilon_2, \dots, \varepsilon_l, \dots$
phase volume	$\Delta w_1, \Delta w_2, \dots, \Delta w_l, \dots$
degeneracy	$\frac{\Delta w_1}{h_0^r}, \frac{\Delta w_2}{h_0^r}, \dots, \frac{\Delta w_l}{h_0^r}, \dots$
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- Number of micro-states: $\Omega_{\text{cl}} = \frac{N!}{\prod a_l!} \prod \left(\frac{\Delta w_l}{h_0^r} \right)^{a_l}$.

Table of contents

- ① §6. Most probable distribution of nearly independent particles
 - 6.1 Classical description of particle's movement
 - 6.2 Quantum description of particle's movement
 - 6.3 Description of the system's microscopic arrangement
 - 6.4 The principle of equal a priori probabilities
 - 6.5 Distribution and microscopic state
 - **6.6 Boltzmann distribution**
 - 6.7 Bose distribution and Fermi distribution
 - 6.8 Relations between Boltzmann, Bose and Fermi distributions

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 $\ln \frac{a_1}{\omega_1} + \alpha + \beta \varepsilon_1 = 0$,
 $\ln \frac{a_2}{\omega_2} + \alpha + \beta \varepsilon_2 = 0$.

Boltzmann distribution

- $$\sum_{l=1} [\ln \frac{a_l}{\omega_l} + \alpha + \beta \varepsilon_l] \delta a_l$$

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- This $\{a_l\}$ is the most probable distribution.

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Table of contents

- ① §6. Most probable distribution of nearly independent particles
 - 6.1 Classical description of particle's movement
 - 6.2 Quantum description of particle's movement
 - 6.3 Description of the system's microscopic arrangement
 - 6.4 The principle of equal a priori probabilities
 - 6.5 Distribution and microscopic state
 - 6.6 Boltzmann distribution
 - 6.7 Bose distribution and Fermi distribution
 - 6.8 Relations between Boltzmann, Bose and Fermi distributions

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- Similarly, for any individual quantum state s , the
average number of particles: $f_s = \frac{1}{e^{\alpha + \beta \varepsilon_s} \pm 1}$,
and $\sum f_s = N, \quad \sum \varepsilon_s f_s = E.$

Table of contents

- ① §6. Most probable distribution of nearly independent particles
 - 6.1 Classical description of particle's movement
 - 6.2 Quantum description of particle's movement
 - 6.3 Description of the system's microscopic arrangement
 - 6.4 The principle of equal a priori probabilities
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 - 6.7 Bose distribution and Fermi distribution
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 $\therefore \Omega_B \simeq \Omega_F \simeq \frac{\Omega_M}{N!}.$

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