

# Thermodynamics & Statistical Physics

## Chapter 2. Thermodynamical properties of uniform medium

Yuan-Chuan Zou  
zouyc@hust.edu.cn

School of Physics, Huazhong University of Science and Technology

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- Using  $S, p$  as free parameters,
$$dH = \left(\frac{\partial H}{\partial S}\right)_p dS + \left(\frac{\partial H}{\partial p}\right)_S dp,$$
- $\Rightarrow \left(\frac{\partial H}{\partial S}\right)_p = T, \left(\frac{\partial H}{\partial p}\right)_S = V.$
- $\frac{\partial^2 H}{\partial p \partial S} = \frac{\partial^2 H}{\partial S \partial p} \Rightarrow \left(\frac{\partial T}{\partial p}\right)_S = \left(\frac{\partial V}{\partial S}\right)_p.$

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- Using  $T, V$  as free parameters,  
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 $dF = \left(\frac{\partial F}{\partial V}\right)_T dV + \left(\frac{\partial F}{\partial T}\right)_V dT$ .
- $\Rightarrow \left(\frac{\partial F}{\partial V}\right)_T = -p, \left(\frac{\partial F}{\partial T}\right)_V = -S$ .
- $\frac{\partial^2 F}{\partial T \partial V} = \frac{\partial^2 F}{\partial V \partial T} \Rightarrow \left(\frac{\partial p}{\partial T}\right)_V = \left(\frac{\partial S}{\partial V}\right)_T$ .

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- $\Rightarrow \left(\frac{\partial G}{\partial p}\right)_T = V, \left(\frac{\partial G}{\partial T}\right)_p = -S.$
- $\frac{\partial^2 G}{\partial T \partial p} = \frac{\partial^2 G}{\partial p \partial T} \Rightarrow \left(\frac{\partial V}{\partial T}\right)_p = -\left(\frac{\partial S}{\partial p}\right)_T.$

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- $$\begin{aligned}\left(\frac{\partial T}{\partial V}\right)_S &= -\left(\frac{\partial p}{\partial S}\right)_V &= \frac{\partial^2 U}{\partial S \partial V}, \\ \left(\frac{\partial T}{\partial p}\right)_S &= \left(\frac{\partial V}{\partial S}\right)_p &= \frac{\partial^2 H}{\partial S \partial p}, \\ \left(\frac{\partial p}{\partial T}\right)_V &= \left(\frac{\partial S}{\partial V}\right)_T &= -\frac{\partial^2 F}{\partial V \partial T}, \\ \left(\frac{\partial V}{\partial T}\right)_p &= -\left(\frac{\partial S}{\partial p}\right)_T &= \frac{\partial^2 G}{\partial p \partial T}.\end{aligned}$$

## §2.2 Maxwell relations & App 1

- $$\left(\frac{\partial T}{\partial V}\right)_S = -\left(\frac{\partial p}{\partial S}\right)_V = \frac{\partial^2 U}{\partial S \partial V},$$
$$\left(\frac{\partial T}{\partial p}\right)_S = \left(\frac{\partial V}{\partial S}\right)_p = \frac{\partial^2 H}{\partial S \partial p},$$
$$\left(\frac{\partial p}{\partial T}\right)_V = \left(\frac{\partial S}{\partial V}\right)_T = -\frac{\partial^2 F}{\partial V \partial T},$$
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- Applications:

## §2.2 Maxwell relations & App 1

$$\bullet \left(\frac{\partial T}{\partial V}\right)_S = -\left(\frac{\partial p}{\partial S}\right)_V = \frac{\partial^2 U}{\partial S \partial V},$$

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$$\left(\frac{\partial p}{\partial T}\right)_V = \left(\frac{\partial S}{\partial V}\right)_T = -\frac{\partial^2 F}{\partial V \partial T},$$

$$\left(\frac{\partial V}{\partial T}\right)_p = -\left(\frac{\partial S}{\partial p}\right)_T = \frac{\partial^2 G}{\partial p \partial T}.$$

- Applications:

- App 1. Choose  $(T, V)$  as free parameters for  $U$ :

- $$dU = \left(\frac{\partial U}{\partial T}\right)_V dT + \left(\frac{\partial U}{\partial V}\right)_T dV, \dots (1)$$



## §2.2 Maxwell relations & App 1

$$\bullet \left( \frac{\partial T}{\partial V} \right)_S = - \left( \frac{\partial p}{\partial S} \right)_V = \frac{\partial^2 U}{\partial S \partial V},$$

$$\left( \frac{\partial T}{\partial p} \right)_S = \left( \frac{\partial V}{\partial S} \right)_p = \frac{\partial^2 H}{\partial S \partial p},$$

$$\left( \frac{\partial p}{\partial T} \right)_V = \left( \frac{\partial S}{\partial V} \right)_T = - \frac{\partial^2 F}{\partial V \partial T},$$

$$\left( \frac{\partial V}{\partial T} \right)_p = - \left( \frac{\partial S}{\partial p} \right)_T = \frac{\partial^2 G}{\partial p \partial T}.$$

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- $$dU = \left( \frac{\partial U}{\partial T} \right)_V dT + \left( \frac{\partial U}{\partial V} \right)_T dV, \dots (1)$$

- and  $dU = TdS - pdV$

## §2.2 Maxwell relations & App 1

- $$\left(\frac{\partial T}{\partial V}\right)_S = -\left(\frac{\partial p}{\partial S}\right)_V = \frac{\partial^2 U}{\partial S \partial V},$$

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$$\left(\frac{\partial V}{\partial T}\right)_p = -\left(\frac{\partial S}{\partial p}\right)_T = \frac{\partial^2 G}{\partial p \partial T}.$$

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- $$dU = \left(\frac{\partial U}{\partial T}\right)_V dT + \left(\frac{\partial U}{\partial V}\right)_T dV, \dots (1)$$

- and  $dU = TdS - pdV$ 

$$= T \left[ \left(\frac{\partial S}{\partial T}\right)_V dT + \left(\frac{\partial S}{\partial V}\right)_T dV \right] - pdV$$

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- $$\left(\frac{\partial T}{\partial V}\right)_S = -\left(\frac{\partial p}{\partial S}\right)_V = \frac{\partial^2 U}{\partial S \partial V},$$

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- Applications:

- App 1. Choose  $(T, V)$  as free parameters for  $U$ :

- $$dU = \left(\frac{\partial U}{\partial T}\right)_V dT + \left(\frac{\partial U}{\partial V}\right)_T dV, \dots (1)$$

- and  $dU = TdS - pdV$

$$= T \left[ \left(\frac{\partial S}{\partial T}\right)_V dT + \left(\frac{\partial S}{\partial V}\right)_T dV \right] - pdV$$

$$= T \left(\frac{\partial S}{\partial T}\right)_V dT + \left[ T \left(\frac{\partial S}{\partial V}\right)_T - p \right] dV. \dots (2)$$

# Maxwell relations App 1

- Comparing the two expressions above,
- $\Rightarrow C_V = \left(\frac{\partial U}{\partial T}\right)_V = T \left(\frac{\partial S}{\partial T}\right)_V$
- and  $\left(\frac{\partial U}{\partial V}\right)_T = T \left(\frac{\partial S}{\partial V}\right)_T - p,$

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- $\xrightarrow{\left(\frac{\partial p}{\partial T}\right)_V = \left(\frac{\partial S}{\partial V}\right)_T} \left(\frac{\partial U}{\partial V}\right)_T = T \left(\frac{\partial p}{\partial T}\right)_V - p,$   
where  $(\partial p/\partial T)_V$  can be taken from the EOS.

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- $\xrightarrow{\left(\frac{\partial p}{\partial T}\right)_V = \left(\frac{\partial S}{\partial V}\right)_T} \left(\frac{\partial U}{\partial V}\right)_T = T \left(\frac{\partial p}{\partial T}\right)_V - p,$   
 where  $(\partial p/\partial T)_V$  can be taken from the EOS.
- For ideal gas ( $pV_m = RT$ ),
- $\left(\frac{\partial U_m}{\partial V_m}\right)_T = T \left(\frac{\partial(RT/V_m)}{\partial T}\right)_{V_m} - p$

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- For ideal gas ( $pV_m = RT$ ),
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 $= T \frac{R}{V_m} - p = p - p = 0, \text{ Joule's law: } U = U(T).$

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- Comparing the two expressions above,

- $\Rightarrow C_V = \left(\frac{\partial U}{\partial T}\right)_V = T \left(\frac{\partial S}{\partial T}\right)_V$

- and  $\left(\frac{\partial U}{\partial V}\right)_T = T \left(\frac{\partial S}{\partial V}\right)_T - p,$

- $\xrightarrow{\left(\frac{\partial p}{\partial T}\right)_V = \left(\frac{\partial S}{\partial V}\right)_T} \left(\frac{\partial U}{\partial V}\right)_T = T \left(\frac{\partial p}{\partial T}\right)_V - p,$

where  $(\partial p/\partial T)_V$  can be taken from the EOS.

- For ideal gas ( $pV_m = RT$ ),

- $\left(\frac{\partial U_m}{\partial V_m}\right)_T = T \left(\frac{\partial(RT/V_m)}{\partial T}\right)_{V_m} - p$   
 $= T \frac{R}{V_m} - p = p - p = 0$ , Joule's law:  $U = U(T)$ .

- For van der Waals gas  $\left((p + \frac{a}{V_m^2})(V_m - b) = RT\right),$

- $\left(\frac{\partial U_m}{\partial V_m}\right)_T = T \left(\frac{\partial p}{\partial T}\right)_{V_m} - p = T \frac{R}{V_m - b} - p = \frac{a}{V_m^2}.$



## Maxwell relations App 2

- App 2: Choose  $(T, p)$  as free parameters for  $H$ ,
- $$dH = \left(\frac{\partial H}{\partial T}\right)_p dT + \left(\frac{\partial H}{\partial p}\right)_T dp,$$

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- and 
$$dH = TdS + Vdp$$
$$= T \left[ \left(\frac{\partial S}{\partial T}\right)_p dT + \left(\frac{\partial S}{\partial p}\right)_T dp \right] + Vdp$$

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- and 
$$\begin{aligned} dH &= TdS + Vdp \\ &= T \left[ \left(\frac{\partial S}{\partial T}\right)_p dT + \left(\frac{\partial S}{\partial p}\right)_T dp \right] + Vdp \\ &= T \left(\frac{\partial S}{\partial T}\right)_p dT + \left[ T \left(\frac{\partial S}{\partial p}\right)_T + V \right] dp. \end{aligned}$$

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- App 2: Choose  $(T, p)$  as free parameters for  $H$ ,
- $$dH = \left(\frac{\partial H}{\partial T}\right)_p dT + \left(\frac{\partial H}{\partial p}\right)_T dp,$$
- and 
$$\begin{aligned}dH &= TdS + Vdp \\&= T \left[ \left(\frac{\partial S}{\partial T}\right)_p dT + \left(\frac{\partial S}{\partial p}\right)_T dp \right] + Vdp \\&= T \left(\frac{\partial S}{\partial T}\right)_p dT + \left[ T \left(\frac{\partial S}{\partial p}\right)_T + V \right] dp.\end{aligned}$$
- $$\therefore C_p = \left(\frac{\partial H}{\partial T}\right)_p = T \left(\frac{\partial S}{\partial T}\right)_p,$$
$$\text{and } \left(\frac{\partial H}{\partial p}\right)_T = T \left(\frac{\partial S}{\partial p}\right)_T + V.$$

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- App 2: Choose  $(T, p)$  as free parameters for  $H$ ,

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- and  $dH = TdS + Vdp$   

$$= T \left[ \left(\frac{\partial S}{\partial T}\right)_p dT + \left(\frac{\partial S}{\partial p}\right)_T dp \right] + Vdp$$
  

$$= T \left(\frac{\partial S}{\partial T}\right)_p dT + \left[ T \left(\frac{\partial S}{\partial p}\right)_T + V \right] dp.$$

- $\therefore C_p = \left(\frac{\partial H}{\partial T}\right)_p = T \left(\frac{\partial S}{\partial T}\right)_p,$   
 and  $\left(\frac{\partial H}{\partial p}\right)_T = T \left(\frac{\partial S}{\partial p}\right)_T + V.$

- $\xrightarrow{\left(\frac{\partial S}{\partial p}\right)_T = -\left(\frac{\partial V}{\partial T}\right)_p} \left(\frac{\partial H}{\partial p}\right)_T = -T \left(\frac{\partial V}{\partial T}\right)_p + V,$

where  $\left(\frac{\partial V}{\partial T}\right)_p$  can be taken from the EOS.

# Maxwell relations Apps 1 & 2

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 $C_p - C_V = T \cdot \beta p \cdot \alpha V = \alpha \beta p V T$ , or  $= \frac{\alpha^2 V T}{\kappa_T}$ .

# Jacobian determinant



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  - 2.1 Complete differential of  $U, H, F, G$
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  - 2.5 Characteristic functions
  - 2.6 Thermodynamics of thermal radiation
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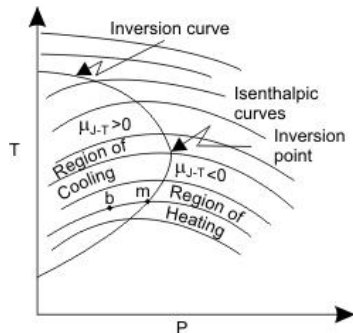
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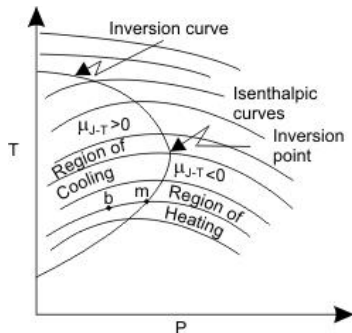
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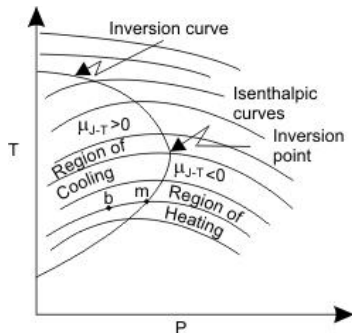
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- Low  $T$ , attractive,  $B < 0$ ,  $\mu > 0$ , cooling; higher  $T$ ,  $B > 0$ , it is possible  $\mu < 0$ , heating.

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- Expansion ( $V$  increasing) makes  $T$  dropping. ...

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  - 2.1 Complete differential of  $U, H, F, G$
  - 2.2 Maxwell relations
  - 2.3 Throttling process and adiabatic expansion
  - 2.4 Determine the basic thermodynamical functions
  - 2.5 Characteristic functions
  - 2.6 Thermodynamics of thermal radiation
  - 2.7 Thermodynamics of magnetic medium

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 where  $\varphi(T) = \frac{H_{m,0}}{RT} - \int \frac{1}{RT^2} dT \int C_{p,m} dT - \frac{S_{m,0}}{R}.$

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- ① Chpt 2. Thermodynamical properties of uniform medium
  - 2.1 Complete differential of  $U, H, F, G$
  - 2.2 Maxwell relations
  - 2.3 Throttling process and adiabatic expansion
  - 2.4 Determine the basic thermodynamical functions
  - 2.5 Characteristic functions
  - 2.6 Thermodynamics of thermal radiation
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$\sigma(T)$  (EOS) is only the quantity needed to be measured.

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- ① Chpt 2. Thermodynamical properties of uniform medium
  - 2.1 Complete differential of  $U, H, F, G$
  - 2.2 Maxwell relations
  - 2.3 Throttling process and adiabatic expansion
  - 2.4 Determine the basic thermodynamical functions
  - 2.5 Characteristic functions
  - 2.6 Thermodynamics of thermal radiation
  - 2.7 Thermodynamics of magnetic medium



## §2.6 Thermodynamics of thermal radiation

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Properties: uniform, isotropic,  $u_\nu(T)$ . (*prove*)

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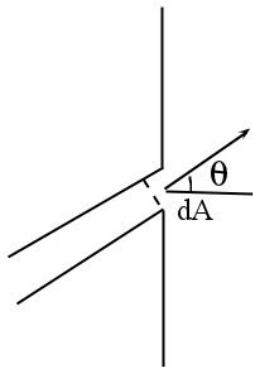
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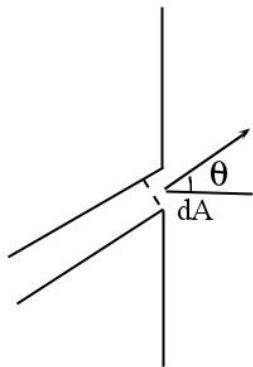
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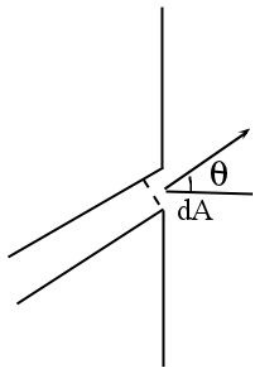


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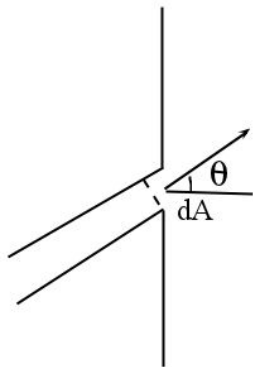
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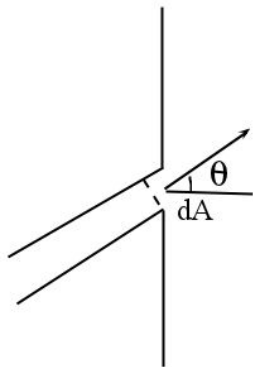
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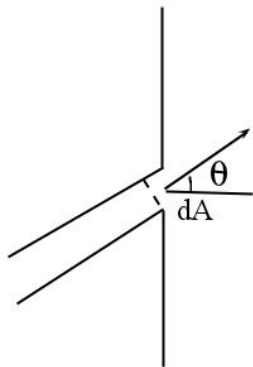


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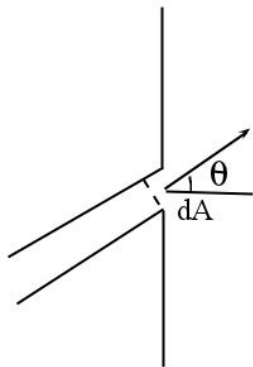
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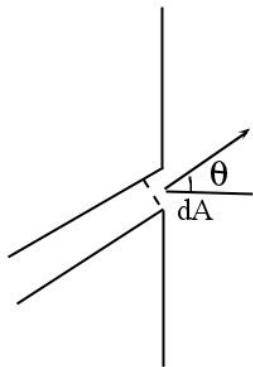
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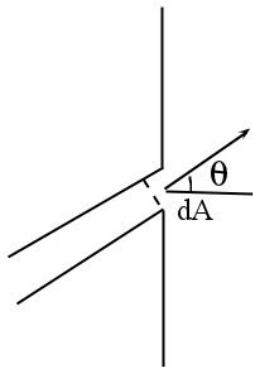
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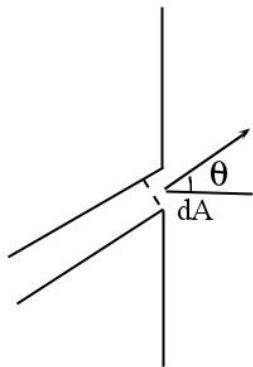
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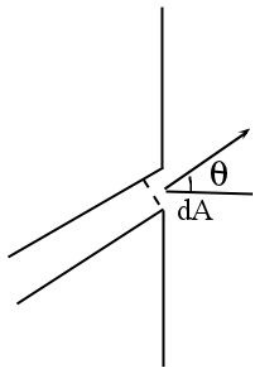
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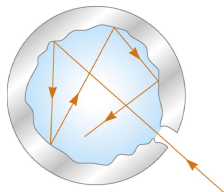
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Example of blackbody:  
tiny window of the cavity in  
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- ① Chpt 2. Thermodynamical properties of uniform medium
  - 2.1 Complete differential of  $U, H, F, G$
  - 2.2 Maxwell relations
  - 2.3 Throttling process and adiabatic expansion
  - 2.4 Determine the basic thermodynamical functions
  - 2.5 Characteristic functions
  - 2.6 Thermodynamics of thermal radiation
  - 2.7 Thermodynamics of magnetic medium

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$$V = \left( \frac{\partial G}{\partial p} \right)_{T, \mathcal{H}}, \quad -\mu_0 m = \left( \frac{\partial G}{\partial \mathcal{H}} \right)_{T, p}, \quad \frac{\partial^2 G}{\partial \mathcal{H} \partial p} = \frac{\partial^2 G}{\partial p \partial \mathcal{H}},$$

$$\Rightarrow \left( \frac{\partial V}{\partial \mathcal{H}} \right)_{T, p} = -\mu_0 \left( \frac{\partial m}{\partial p} \right)_{T, \mathcal{H}}.$$

- 3°. In non-uniform magnetic field, potential energy change:

# Thermodynamics of magnetic medium: general

- 2°. Considering the volume change:

$$\delta W = \mu_0 \mathcal{H} dm - p dV.$$

- $dU = T dS - p dV + \mu_0 \mathcal{H} dm,$

$$G = U - TS + pV - \mu_0 \mathcal{H} m,$$

$$dG = -S dT + V dp - \mu_0 m d\mathcal{H},$$

$$V = \left( \frac{\partial G}{\partial p} \right)_{T, \mathcal{H}}, \quad -\mu_0 m = \left( \frac{\partial G}{\partial \mathcal{H}} \right)_{T, p}, \quad \frac{\partial^2 G}{\partial \mathcal{H} \partial p} = \frac{\partial^2 G}{\partial p \partial \mathcal{H}},$$

$$\Rightarrow \left( \frac{\partial V}{\partial \mathcal{H}} \right)_{T, p} = -\mu_0 \left( \frac{\partial m}{\partial p} \right)_{T, \mathcal{H}}.$$

- 3°. In non-uniform magnetic field, potential energy change:

$$\delta W = -\mu_0 m d\mathcal{H}.$$

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