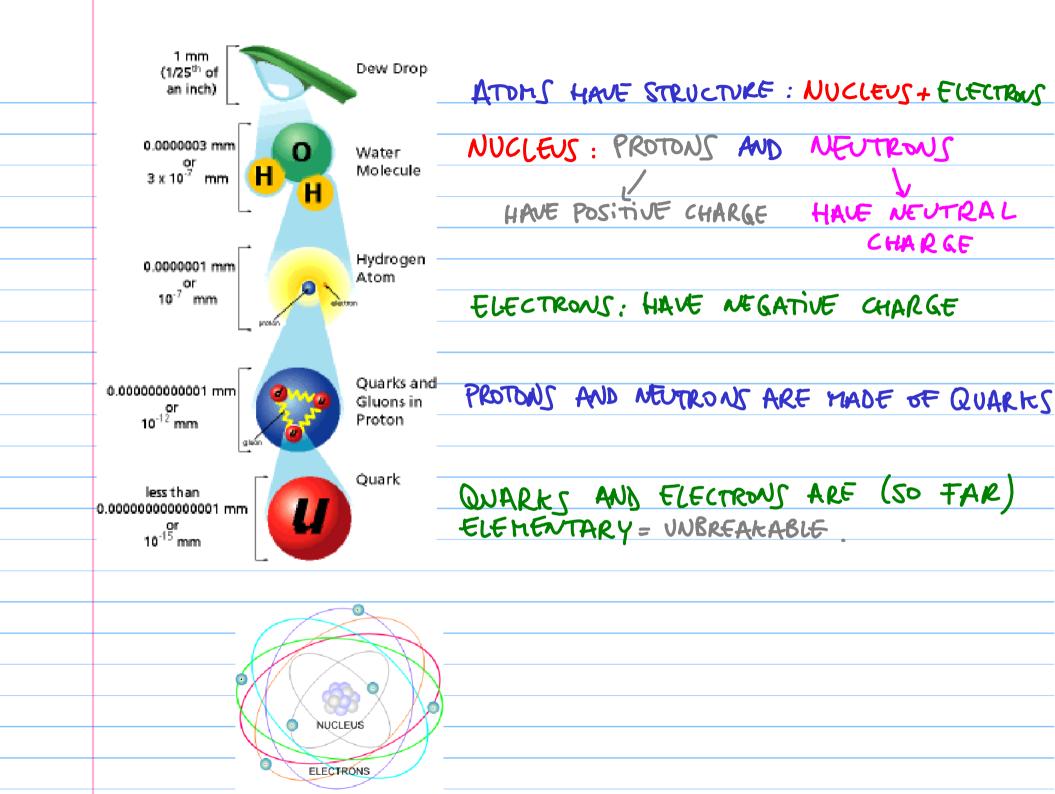
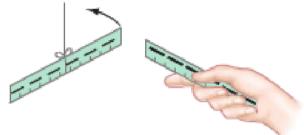
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PHY 122: CHARGES
LECTURE 1
Ch. 21
PHY 121: KINEMATICS: HOW OBJECTS MOVE (trajectory, v, a, t)
          DYNAMICS: WHY OBJECTS HOVE (mass, force, work, E)
           LI CONSERVED QUANTITIES: ENERGY (HINETIC + POTENTIAL)
                                   HOHENTUH
MAIN FORCES YOU HAVE DEALT WITH:
     GRAVITY, TENSION, NORMAL, FRICTION
THESE REALLY DON'T DESCRIBE THE STRUCTURE OF MOST THINGS!
LARGE NUMBER OF PHENOMENA ARE OF ATOMIC ORIGIN
ATOMIC STRUCTURE IS DEFINED BY ELECTROMAGNETISM
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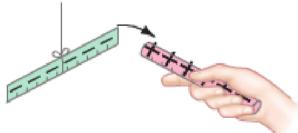
ELECTRICITY



(a) Two charged plastic rulers repel



(b) Two charged glass rods repel



 (c) Charged glass rod attracts charged plastic ruler TWO KINDS OF ELECTRIC CHARGE:

LIKE CHARGES: ++ AND -- REPEL

UNLIKE CHARGES: +- ATTRACT

THE CHARGES THAT WE SEE IN THESE EXPERIMENTS WHE FROM ATOMS: ELECTRONS AND IONS

FRICTION (RUBBING) SEPARATES THEM

ELECTRIC CHARGE

THE ELECTRIC CHARGE IS GONTERVED

HEASURED IN COVLOMBS [C]

FLECTRON HAS A NEGATIVE CHARGE: - e = - 1.6 × 10 °C

PROTON HAS A POSITIVE CHARGE

ATOM IS ELECTRICALLY NEUTRAL

NUCLEUS IS HEAVY, ELECTRON IS LIGHT: USUALLY CHARGE IS TRANSPORTED BY ELECTRONS,

OBJECTS THAT ACQUIRE E BECOME NEGATIVELY CHARGED

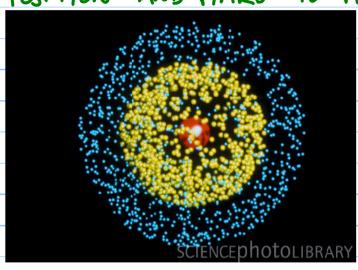
Q = - Ne e

OBJECTS THAT LOST E BECOME PSITIVELY CHARGED

INSULATORS AND CONDUCTORS

IN SOLIDS, ATOMS AND THEIR NUCLEI ARE LOCKED IN THEIR

POSITION AND HARD TO HOUE



INSULATORS: HAVE COMPLETE OR ALMOST

COMPLETE ELECTRON SHELLS

ELECTRONS ARE HARD TO HOVE

AROUND

EXAMPLE: WOOD, PLASTIC

Lithium atom (Z=3, metal) computer representation of 3D structure of orbitals.

Red: nucleus

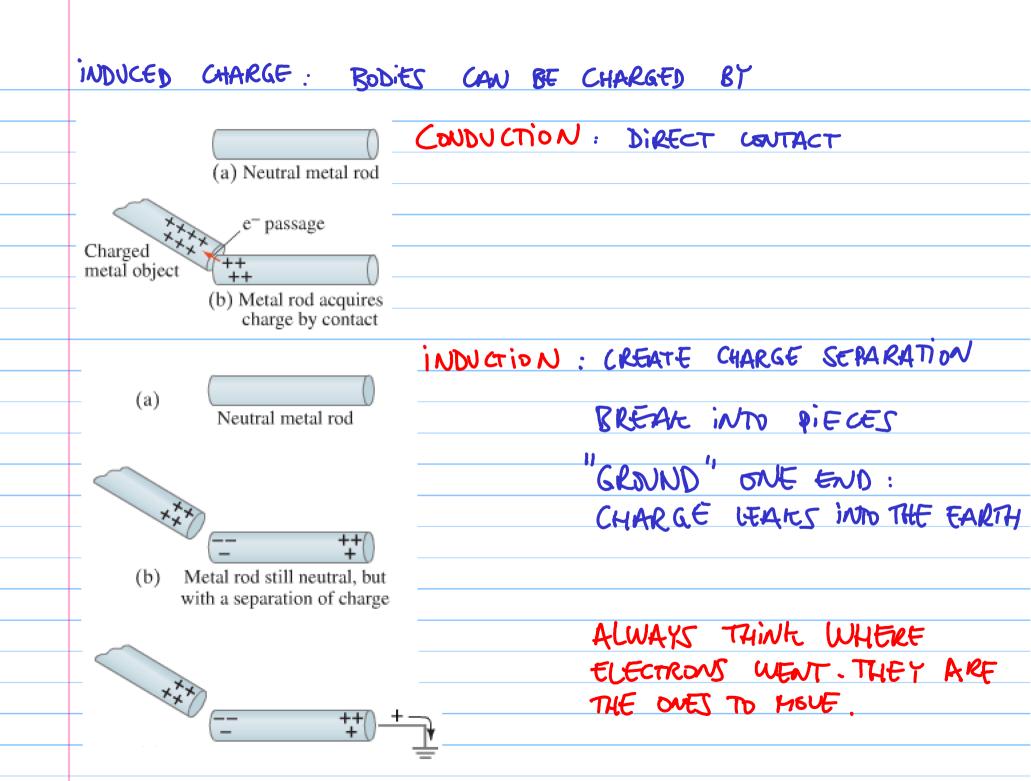
Yellow: 1s orbital (2 electrons)

Blue: 2s orbital (1 electron)

CONDUCTORS: HAVE ONE OR TWO ELECTRUS

"FREE" FLECTRONS ARE EASY
TO TIOUE AROUND OBJECT.

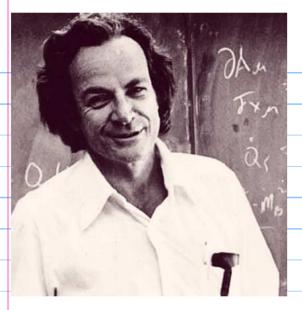
EXAMPLE: METALS



Quiz 1:
1.1 Two electrically neutral materials are rubbed together.
One acquires a net positive charge. The other must:
A. have lost electrons
B. have gained electrons
C. have lost protons
D. have gained protons

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EXTRA:



Richard Feynman (Nobel laureate, and legendary educator) in the Feynman Lectures:

If you were standing at arm's length from someone and each of you had one percent more electrons than protons, the repelling force would be incredible. How great? Enough to lift the Empire State building? No! To lift Mount Everest? No! The repulsion would be enough to lift a 'weight' equal to that of the entire earth!

SHOULD WE BELIEVE FEYNMAN?

HOW MANY ELECTRONS IN A PERSON?

ASSUME CHEMICAL COMPOSITION OF A PERSON IS ONLY WATER (SIMPLIFICATION). MOLECULAR WEIGHT OF WATER = 18.

WHAT IS THE NUMBER OF ELECTRONS/GRAM IN A PERSON?

6x 10 23 molecules/mole x 10 e/molecule = 3.3x 10 e/g
18 g/mole

ASSUME 80 kg FOR PERSON: 3.3×10 e/g ×80 kg = 2.6×10 e

HOW MUCH CHARGE IS 1% OF ELECTRONS IN A PERSON? 190 x 2,6x10 e x 1.6x10 C/e = 4.2×107 C

WHAT IS THE FORCE BETWEEN TWO REOPLE AT ARMS LENGTH IF THEY EACH HAD 1% EXCESS ELECTRONS?

$$F = k \frac{Q_1 Q_2}{\Gamma^2} = 9 \times 10^9 Nm^2 \left(\frac{4.2 \times 10^7 C}{0.35 m}\right)^2 = \frac{2.8 \times 10^{25} N}{10^{10} N}$$

WHAT IS THE WEIGHT OF THE EARTH? W = Mg $W = G_{2} lo \quad kg \times 9.8 \, m/s^{2} = 5.9 \times 10^{25} \, N$

SO, YES! FEYNDAN WAS CORRECT!

GRAVITY IS EXTREMELY WEAR // IT INVOLVES HUGE MASSES.

ELECTRICITY IS 1036 TIMES STRONGER THAN GRAVITY