



FILL IN THE BLANKS

1. WATER AND ITS TREATMENT

1. Convert 10 ppm of hardness of water in to Degree Clark (Cl°) _____.
2. Which Phosphate is used for treating of too alkaline water _____.
3. Temporary hardness is caused by _____.
4. Cation Exchange resins are regenerated by using _____.
5. Hardness of water is expressed in equivalents of _____.
6. The chemical name of Calgon is _____.
7. Temporary hardness of water can be removed by _____.
8. Anion exchange resins are regenerated by using _____.
9. _____ is considered to be the naturally distilled water.
10. A semi permeable membrane allows the flow of _____.

2. BATTERY CHEMISTRY & CORROSION

1. In Li-Ion cell _____ is used as cathode.
2. In Zn-Air battery _____ is used as anode.
3. In Methanol-Oxygen fuel cell _____ is used as anode.
4. In Solid Oxide fuel cell _____ gas is passed through the cathode.
5. Solar cell is also known as _____.
6. The rate of corrosion _____ with the increasing temperature.
7. Impressed current cathodic protection, converted the corroding metal from _____.
8. The example of Sacrificial anodic metal is _____.
9. In wet corrosion Absorption oxygen takes place in _____ medium.
10. In waterline corrosion the anode will be _____ oxygenated part.

MULTIPLE CHOICE QUESTIONS

1. WATER AND ITS TREATMENT

1. What is the pH range maintained in EDTA method? []
a. 4-5 b. 6-7 c. 9-10 d. 12-14
2. The colour of stable complex in EDTA method is []
a. Blue b. Wine red
c. Colour less d. Purple
3. Which of the chemical are act as Coagulants help in the settling of []
a. Alum b. $\text{Al}_2(\text{SO}_4)_3$
c. Both (a) & (b) d. Mg SO_4
4. Desalination of water is. []
a. Removal of NaCl b. Removal of suspended
c. Removal of gases d. Removal of metals
5. Which of the following will act as disinfectant? []
a. Cl_2 b. O_2
c. F_2 d. H_2
6. EDTA method of determining hardness of water can be used to determine. []
a. All types of hardness b. Temporary hardness only
c. Permanent hardness only d. alkaline hardness only
7. Coagulants help in the settling of. []
a. Suspended impurities only b. Fine suspended matter only
c. Colloidal particles only d. both (b) & (c)
8. Reverse osmosis process is also called as. []
a. Super filtration b. Hyper filtration
c. Both (a) & (b) d. Hypo filtration
9. Disinfection by ozone is due to liberation of []
a. Oxygen b. Nascent oxygen
c. Molecular oxygen d. Oxide
10. Potable water treatment doesn't involve []
a. Demineralization b. sedimentation
c. filtration d. disinfection

2. BATTERY CHEMISTRY & CORROSION

1. A battery is a device that can operate []

a. as electrolytic cell	b. as voltaic cell
c. both as electrolytic cell and voltaic cell	d. none of the above
2. The cell whose reaction is reversible is called []

a. Fuel cell	b. Primary cell
c. Secondary cell	d. All the above
3. Fuel cells are used as auxiliary energy source in []

b. Trains	b. Aeroplanes
c. Space vehicles	d. Automobile engines
4. The Li-Ion battery is a []

a. Fuel cell	b. Primary cell
c. Secondary cell	d. All the above
5. What is the electrolyte used in Zn-air battery. []

a. KOH	b. NaOH	c. LiOH	d. $\text{Mg}(\text{OH})_2$
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6. Corrosion is a process of []

a. Oxidation	b. Reduction	c. Electrolysis	d. ozonolysis
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7. The chemical composition of rust is []

c. $\text{Fe}_2\text{O}_3 \cdot \text{H}_2\text{O}$	b. $\text{Fe}_3\text{O}_4 \cdot x\text{H}_2\text{O}$
c. $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$	d. $\text{Fe}_3\text{O}_4 \cdot \text{H}_2\text{O}$
8. In waterline corrosion the maximum amount of corrosion takes place []

a. Along a line just above the level of the water meniscus
b. Along a line at the level of the water meniscus
c. Along a line just below the level of the water meniscus
d. At the bottom of the vessel.
9. The rate of corrosion of iron in atmosphere depends upon: []

a. The humidity of the atmosphere	b. The degree of pollution of the atmosphere
c. The frequency of rain fall	d. All of these factors
10. When a buried pipeline is protected from corrosion by connecting to Mg block, it is called []

a. Impressed voltage protection	b. Sacrificial cathodic protection
c. Sacrificial anodic protection	d. Any of these