

Chris Braissant

# **VPM - Variable Permeability Model**

Ban's Diving Resort

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# Introduction

## Introduction

MultiDeco  
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MultiDeco - Chris

Diver Dive log Save Print Config Calc Bailout Lost gas - or + Turn p About Exit

Configure dive parameters - Chris

Surface interval ☐ 5 day ☐

☒ Open ☐ SCR

Bottom mix & travel  
Depth, time, O<sub>2</sub>/He

☐ 43, 15, 21  
☒ 40, 26, 21  
☒ 40, 16, 21

Deco  
O<sub>2</sub>/He

☒ 50  
☐ 80  
☐ 100  
☐ 21.2

Depths

☐ Feet  
☒ Meters

RMV's

☐ Cu. ft.  
☒ Litres

Stop times

☒ 1 min  
☐ 30 sec  
☐ 10 sec  
☐ 1 sec

Water

☒ Salt  
☐ Fresh

CCR setpt

☒ Bar  
☐ Atm

Conditions

Deco Steps: ☐ 6 m deep ☐ 30 sec deep

Stop Size:  3  3  4.5

Max depth for 100% O<sub>2</sub>:  6

Deco mix depths: Max ppO<sub>2</sub> for deco mix...% O<sub>2</sub>  
up to 28 28...45 45...99

1.4  1.5  1.60

RMV: Bottom  20 Deco  20

SCR: ☐ fiO<sub>2</sub> drop/factor adjust

Decompression Model

☒ VPM-B  
☐ VPM-B/E  
☐ VPM-B/S  
☐ ZHL16-B + GF  
☐ ZHL16-C + GF

Conservatism  + 2

Elevation data

Dive elevation  0 m.

Hours dive elev.  0  0 Travel time hr

Ascended from  0  0 Average elev. last 2 weeks.

Depth meter calibration ☒ Sea level ☐ Altitude

Descent rate

0-300, 18

Ascent rate

300-0, 9

Extended stops  
Extra stop time for deco mix changes:

2  7..30m  2  30+

☐ Accelerate stops  
Accelerate the last 2 stops for these mixes (O<sub>2</sub> deco req'd):

☐ up to 28 / 25..80  
☐ 28..45 / 25..70  
☐ 50..75 / 25..50

☐ Add to stop  
☐ All deco mix changes  
☐ O<sub>2</sub> window affect

Add a level or mix

Vol

2363  
54.8  
37.1  
33.2  
73.0  
139  
181  
225

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### Multi-deco

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V-Planner - Chris

Surface interval ☐ 5 day ☐ 48 hr ☐ 24 hr 000 days 00:00 hh:mm

Open ☐ SCR ☐ CCR

Bottom mix & travel Depth, time, O<sub>2</sub>/He

Deco gas O<sub>2</sub>/He

Configure dive parameters - Chris

Depths

☒ Feet

☐ Meters

RMV's

☒ Cu. ft.

☐ Litres

Stop times

☒ 1 min

☐ 30 sec

☐ 10 sec

☐ 1 sec

Water

☒ Salt

☐ Fresh

CCR setpt

☒ Bar

☐ Atm

Conditions

Deco Steps: ☐ 20 ft deep ☐ 30 sec deep

Stop Size: 10

Max depth for 100% O<sub>2</sub>: 20

Max ppO<sub>2</sub> for deco mix... % O<sub>2</sub> up to 28 28...45 45...99

1.40 1.50 1.60

RMV: Bottom 0.7 Deco 0.6

SCR: ☐ fiO<sub>2</sub> drop/factor adjust

Extended stops

Extra stop time for deco mix changes:

2 21..100 2 100+

☐ Add to stop

☐ All deco mix changes

☐ O<sub>2</sub> window affect

Accelerate stops

Accelerate the last 2 stops for these mixes (O<sub>2</sub> deco req'd):

☐ up to 28 / 25..80

☐ 28..45 / 25..70

☐ 50..75 / 25..50

Conservatis

☐ Nominal

☐ + One

☒ + Two

☐ + Three

☐ + Four

☐ + Five

VPM Model

☒ VPM-B ☐ VPM-B/E ☐ VPM-B+CFS

Elevation data

Dive elevation 0

Hours dive elev. 0

Ascended from 0

Travel time (hr) 0

Average elev. last 2 weeks. 0

Depth meter calibration ☒ Sea level ☐ Altitude

OK Cancel Advanced

\* IBCD and tank capacity checks and warnings.

\* Accelerated He deco stops option.

\* Favorite (standard) gas mix lists.

\* Dive profile graphs, Excel data sheets for custom built graphs.

\* Improved extended stops (O<sub>2</sub> window) time options.

\* More deep stop time interval options.

\* Elevation diving adjustments to match your depth meter.

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What is that VPM-B and why is it important to know it?

- Popular decompression model
- Alternative to Buhlmann ZHL16
- Way to double check your schedule
- Understand your dive profile

## Understand:

- Buhlmann dissolved model
- Bubbles behaviour
  - Diffusion with tissue compartment
  - Laplace pressure
  - Grow or shrink
  - Critical radius
- Principle of VPM model
  - How the model is made
  - Variable permeability
  - Bubble crushing
  - Algorithm
  - Variants

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## Topics:

- Dissolved Gas Model (Buhlmann ZHL16)
- Bubble behaviour
- VPM model



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# Presentation

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**Dissolved Gas Model**

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# Dissolved Gas Model

## Buhlmann ZHL16

- Gas dissolved in tissue
- 16 tissue compartments
- Compartments on-gas and off-gas at different rates (Half-Time)
- Track supersaturation in the tissue compartments
- Decompression stop when a compartment reaches the M-Value
- Pressure in a tissue compartment ("tissue tension") vary for each compartment

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# Bubble Behaviour

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A bubble will grow if the pressure pushing out is greater than the pressure pushing in

A bubble will grow if the pressure pushing out is greater than the pressure pushing in

Pressure pushing in:

- Ambient Pressure
- Surface tension (Laplace pressure)

A bubble will grow if the pressure pushing out is greater than the pressure pushing in

Pressure pushing in:

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Pressure pushing out:

- Bubble pressure

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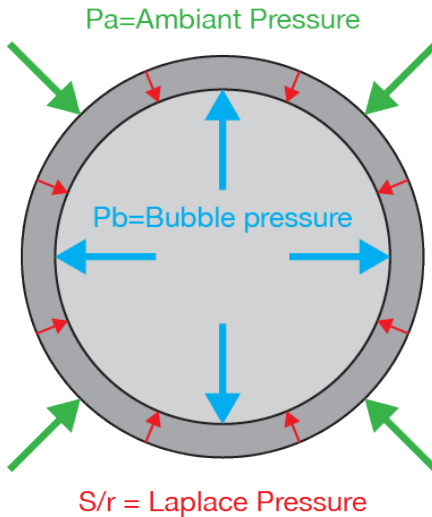
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## Comportment at saturation:

Equilibrium of pressure: (Stable bubble)

$$P_{bubble} = P_{ambient} + P_{Laplace}$$

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## Comportment at saturation:

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Tissue tension around the bubble:

$$P_{tissue} = \textit{Partial pressure of inert gas}$$

## Comportment at saturation:

Equilibrium of pressure: (Stable bubble)

$$P_{bubble} = P_{ambient} + P_{Laplace}$$

Tissue tension around the bubble:

$$P_{tissue} = \textit{Partial pressure of inert gas}$$

If the pressure in the bubble is greater than the tissue tension, gas will diffuse out.

The bubble would then shrink!

$$P_{bubble} > P_{tissue} \rightarrow \textit{Bubble shrinks}$$

## Exemple:

A diver breathing air at 40m:

$$P_{ambient} = 5bar$$

$$P_{Laplace} = 0.5bar$$

$$P_{bubble} = P_{ambient} + P_{Laplace} = 5.5bar$$

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Pressure in the tissue compartment:

$$P_{tissue} = FN_2 * P_{ambient} = 0.79 * 5 = 3.95bar$$

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Pressure in the tissue compartment:

$$P_{tissue} = FN_2 * P_{ambient} = 0.79 * 5 = 3.95bar$$

The bubble pressure is greater than the tissue tension, the bubble would then diffuse out and shrink

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## **Ascent:**

The ambient pressure will change quickly and the tissue tension will be delayed (depending on the compartment half-time).

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The ambient pressure will change quickly and the tissue tension will be delayed (depending on the compartment half-time).

The pressure in the bubble would drop below the tissue tension.

The gas will diffuse in and the bubble will grow.

Hence the decompression stop to allow the tissue compartment and the bubble size to adjust

## Exemple:

Same diver as the before, ascending quickly to 20m:

$$P_{ambient} = 3bar$$

$$P_{Laplace} = 0.5bar$$

$$P_{bubble} = 3.5bar$$

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Pressure in the tissue compartment (don't have time to offgas):

$$P_{tissue} = 3.95bar$$

## Example:

Same diver as the before, ascending quickly to 20m:

$$P_{ambient} = 3bar$$

$$P_{Laplace} = 0.5bar$$

$$P_{bubble} = 3.5bar$$

Pressure in the tissue compartment (don't have time to offgas):

$$P_{tissue} = 3.95bar$$

The bubble pressure is lower than the tissue tension, the bubble would then on-gas and grow

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## Bubble radius:

Another factor will affect the growth of a bubble: its radius

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## Bubble radius:

Another factor will affect the growth of a bubble: its radius

The Laplace pressure is directly linked to the bubble radius. As a bubble grows, its "skin" becomes thinner, and the Laplace pressure decreases

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The bubble pressure will also decrease, and the bubble will grow faster!



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The Laplace pressure is directly linked to the bubble radius. As a bubble grows, its "skin" becomes thinner, and the Laplace pressure decreases

The bubble pressure will also decrease, and the bubble will grow faster!

Large bubbles grow, small bubbles shrink

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## **Critical radius:**

Maximum radius a bubble can have to maintain an adequate Laplace pressure

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## **Critical radius:**

Maximum radius a bubble can have to maintain an adequate Laplace pressure

Bubbles models try to control the size of the bubbles

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## **Critical radius:**

Maximum radius a bubble can have to maintain an adequate Laplace pressure

Bubbles models try to control the size of the bubbles

Under a critical radius, bubble will shrink as the diver ascent

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# VPM: Variable Permeability Model

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## **VPM: Variable Permeability Model**

- Same idea as Buhlmann dissolved gas model
- 16 compartment tracking dissolved inert gases
- Add a single bubble to each compartment
- 16 bubbles in those compartments
- Dissolved gas and free gases (bubbles) interacting

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## Specific properties:

- Specific size indicated by its radius
- Varying permeability
- Defined skin
- Elasticity and memory
- Gas pressure inside the bubble

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## Variable Permeability:

- "Skin" made of tiny balls (molecules)
- Dissolved gases from the tissue compartment diffuse through the skin (around the tiny balls)
- When the ambient pressure increase, the bubble is compressed and the skin becomes impermeable
- Further crushing becomes more difficult
- Hence its name: Variable Permeability Model



## Bubble crushing:

During the descent:

- Ambient pressure increase
- Compartments on-gas following Buhlmann model
- Tissue tension in the compartment increase
- Bubbles are crushed
- Bubbles in slow compartment crush faster than in fast compartment

## Bubble crushing:

During the descent:

- Ambient pressure increase
- Compartments on-gas following Buhlmann model
- Tissue tension in the compartment increase
- Bubbles are crushed
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Remark:

The bubble's size is linked to difference of pressure between the ambient pressure and the tissue tension

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## VPM Algorithm:

- Buhlmann track dissolved gases in the tissues compartment (tissue tension)
- VPM track the bubbles' radius
- VPM calculates bubbles' pressures during the ascent
- Bubbles grow only when pressure of the bubble exceed the tissue tension (VPM vs Buhlmann)
- Decompression stop before a bubble starts to grow

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## Original VPM algorithm

- Too conservative
- "no-bubble-growth model"

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- Too conservative
- "no-bubble-growth model"

## Modifications made:

- Body can handle a certain amount of free gas
- DCS over a critical volume
- Allow bubbles from a certain size (critical radius) to grow
- "not-too-much-bubble-growth model"

## Original VPM algorithm

- Too conservative
- "no-bubble-growth model"

## Modifications made:

- Body can handle a certain amount of free gas
- DCS over a critical volume
- Allow bubbles from a certain size (critical radius) to grow
- "not-too-much-bubble-growth model"

## Bubble population:

- Numerous small bubble, few large bubbles

Latest modifications:

VPM-B:

- Incorporates Boyle's law to the bubbles behaviour
- Note bubbles' size at the first stop
- Increase the radius of each bubbles for next stop according to Boyle's law
- Tissue tension allowed for the next stop is reduced due to the larger bubble radius
- Add conservatism to the profile

VPM-B/E:

- Add conservatism for extreme dives

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# Summary



## Key Points:

- Buhlmann dissolved model
- Bubbles behaviour
  - Diffusion with tissue compartment
  - Laplace pressure
  - Grow or shrink
  - Critical radius
- Principle of VPM model
  - How the model is made
  - Variable permeability
  - Bubble crushing
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What is that VPM-B and why is it important to know it?

- Popular decompression model
- Alternative to Buhlmann ZHL16
- Way to double check your schedule
- Understand your dive profile

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## Question 1:

How many tissue compartment does VPM use for it's Buhlmann model?

- (a) 8
- (b) 12
- (c) 16
- (d) 24

**Question 1:**

How many tissue compartment does VPM use for it's Buhlmann model?

- (a) 8
- (b) 12
- (c) 16
- (d) 24

**Answer:(c)**

The model in that presentation is the Buhlmann ZHL16

**Question 2:**

To what kind of model does the Buhlmann ZHL belongs to?

- (a) Dissolved Model
- (b) Bubble Model
- (c) Reduced Gradient Model
- (d) Variable Permeability Model

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To what kind of model does the Buhlmann ZHL belongs to?

- (a) Dissolved Model
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- (c) Reduced Gradient Model
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**Answer:**(a)

**Question 3:**

Why would a bubble grow

- (a) Pressure pushing in greater than pressure pushing out
- (b) Tissue tension greater than bubble pressure
- (c) Skin becomes impermeable
- (d) Ambient pressure increase

**Question 3:**

Why would a bubble grow

- (a) Pressure pushing in greater than pressure pushing out
- (b) Tissue tension greater than bubble pressure
- (c) Skin becomes impermeable
- (d) Ambient pressure increase

**Answer:**(b )



**Question 4:**

Why does a bubble becomes impermeable?

- (a) High pressure squeezes molecules against each other
- (b) The thickness of the skin is too small
- (c) Tissue tension exceed the M-Value
- (d) Bubble pressure in null

**Question 4:**

Why does a bubble becomes impermeable?

- (a) High pressure squeezes molecules against each other
- (b) The thickness of the skin is too small
- (c) Tissue tension exceed the M-Value
- (d) Bubble pressure in null

**Answer:**(a)

**Question 5:**

What is the main reason of DCS according to VPM model?

- (a) Tissue tension exceeding the M-Value
- (b) Amount of free gas in the body
- (c) Supersaturation of the tissue compartment
- (d) The size of the bubble in the slow compartment

**Question 5:**

What is the main reason of DCS according to VPM model?

- (a) Tissue tension exceeding the M-Value
- (b) Amount of free gas in the body
- (c) Supersaturation of the tissue compartment
- (d) The size of the bubble in the slow compartment

**Answer:**(b)