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We left off lant stime having derived SAHA nphe = (2TT Me KT) exp (-13.6eV)

We first demand $N_p = N_e$ to the Value ionization place where $N_H = N_p = N_e$, then

The np = (2TT Me KT) 2 exp()

 $n_p = 2.4 \times 10^{21} \text{cm}^{-3} T_4^{3/2} \exp\left(-\frac{15.76}{T_4}\right)$ Set $n_p = 10^{16}$ so we get

This gives, for np=1016

 $+12.38 = \frac{-3}{2} ln Ty + \frac{15.76}{Ty}$

T4# 15.76 = 12.38 + 3/nTy

Ty = 12,38+3/10Ty =7 T= 12,400. (KT=1.1eV)

So the 1/2 ionization point happens about photospheric for typical photospheric Note: It ends up that Saha collisional ionitations me protospheres I. Also remember that recombination occurs in the early univen N=3×104, Nue = 5×1000 + T= 4000K, Z=1400.

The atoms begin to "verlap"

or

fourly me so me connot

fourly me sahar.

Sho into more detail

on this

Lecture 14 Htmospheres + Spectrul Types & Lines: living on the Main sequence han Teff nould like Ar talk today about how an observer finde these things out quantitiely. Kemember that h= KT mpg << R so that are plane parallel To discurred simb-dark lains all more about that here. The tot mosphere has two $\frac{M}{R^2} = 2.7 \times 10^4 \frac{M}{R^2}$ in 91 and F=0sB Teff server to define F=1/4 $T_{eff} = \left(\frac{F}{\sigma_{sB}}\right)^{1/4}$ If there was no lines etc etc
+ from you would soo a BB Tc = Teff.

For a photosphere we take. DB == [KSdr= Ky => 4p=3R Pph = gy = g = 2nkT nph = gmp
30Th kBT = (10 cm3) (Kos) (104k) (2.7×104) Now, for the upper main segmence I showed that the R= MODIS so g = GM = 15 = 2.7x104/MU)/2 masses. K& Kes and so he can stay low.

the ionization temp for Hydrogen is about 104 Ko The Tisof the Main Sequence go as:

Sp Type		M	
03	52000	120 60 23	
08 BO	35,800	23	
BA	22,000 15,400 9 520	3	
FO	7,200	1.65	
MO	3850 3850	0.79	
////5	3240		

Jonization States

Now, most of the understanding

Let 1 of the presences or abscenes of

Let lines in the atmospher Come

from considering 3 types it

Metallic-Like (low FIP) Li, Nu, Mg, Al \subseteq 5e Y Middles: H, C, N, D, ... 10 < E, <.

Noble: He+ Ne > 20eV

He & Na

24.6

S.14

elute the g.s. population

of jours

Na+e- - Na+&

$$\frac{n_{+}n_{e}}{n_{N}} = \left(\frac{2\pi m_{e}kT}{h^{2}}\right)^{3/2} exp\left[\frac{-E_{i}}{k_{B}T}\right]$$

$$\frac{N+}{NN} = \frac{2.4 \times 10^{15} - \frac{3}{2}}{he} = \frac{1-E;}{kasT}$$

Now, let's compare Sodium to H, in which come

$$\frac{n_{Na}^{+}}{n_{Na}} = \exp\left[-\frac{(E_{Na} - E_{H})}{K_{B}T}\right] \frac{n_{p}}{n_{H}} \approx 10^{7} \frac{n_{p}}{n_{H}}$$

at T=6000 K. Whereas

$$\frac{N(He^{\dagger})}{N(He)} = \exp\left(-\frac{(E_{He}-E_{H})}{K_{B}T}\right)\frac{N_{P}}{h_{H}} = 6\times10^{-10}\frac{N_{P}}{N_{H}}$$
for $T = 6000K$,

so we really expect that the Alkali stoms ionize long before the Alkali Hope we even relatively rue, but can account four nearly cull e's which we free For example for the sun, he = 1013 cm-3, T=6000 k, then:

n(Nu) = 5.4×103

So Not and all other alkalis will be at least singly ionized in the sun wherear H+ He are nentral.

Lets find the transition temps

Lor a number of ne's.

13.6cV ne (cm-3) TNu THE TH 3090 14,500 8000 3508 9082 16,500 19000 4057 10500 4810 22570 2450

 $1 = \frac{2.4 \times 10^{8}}{n_{13}} \exp\left(\frac{-E_{i}}{k_{\delta T}}\right) \implies k_{\delta T} = \frac{E_{i}}{19.3 - \ln n_{13}}$

 $T = \frac{1.16 \times 10^4 \ (Eilev)}{19.2 - 10 M.2}$

How: Balmer Series

Now, staros > 10' k will have

ionizid H completed & show no H

lines.

 $\frac{3}{2} \frac{1/3^{2} = -1.51}{-13.6eV} = -3.4eV$ $\frac{1}{2} = -3.4eV$ $\frac{hc}{2} = -hc$ $\frac{hc}{2} = -13.6eV$ $\frac{hc}{2} = -1.81eV$

Lyman Series (m > 1) Lyman Balmer Panchen 12 1215 1025 6562 4 977 4861 4340

3646

So to sec any of the Balmer liner we must have an appereciable # of H in the n=2 state

 $= \frac{N_{n=2}}{N_{n=1}} = \exp\left(\frac{-10.20 \, \text{eV}}{kT}\right) = 10^{-5} \, \text{wf } T = 10^4$

8 203

which is large enough to give an appreciable absorption line, as

Oline-center >> OTh , and actually $\int 0 \approx T \chi^2 = 10^{-16} \text{ cm}^2, \text{ which is}$ $\int A = 10^{-8} \Rightarrow \text{ OTh } = 0 \text{ are long ga}$ this ratio is satisfied, which is about $exp\left(-\frac{10.20eV}{KT}\right) \xrightarrow{OTh} \xrightarrow{Th} > 1$ we get Balance lines. So when most of H is neutral (TZ104), then we just want

exp (-10,20eV) > OTh = 6×109

horder So A -> F stars have compicuous H-Balmer liner. Hotter stars have all Hionized and, once T>15,000 or 50. He is singly ionized and one seen liner from that

O-B Type: Absolutely all H is completely lovized thelium is either neutral or partially ionized At love B, state getting H Bulme.

A: 11,000=7500: H + ionized My are provalent. Few metal lines.

note that there exists a way of converting the strength of absorption lines to abundances.

THE SPECTRAL TYPES

One of the earliest results of observational astronomy was the realization that there existed a correlation between properties of the stellar surface, such as its surface temperature, and the strength of specific absorption lines as seen in the spectrum of the star. In 1863 the Jesuit astronomer Angelo Secchi classified stars into four groups according to the prominent absorption lines in their spectra. An empirical classification scheme was subsequently developed in which stars were sorted into seven principal spectral types, each type being characterized by a certain range of surface temperature and the appearance of characteristic absorption lines. The detailed understanding of the correlation between surface temperatures and prominent absorption lines rests almost entirely in various applications of the Boltzmann and Saha formulas.

The principal classification groups of stars are traditionally labeled by the letters O, B, A, F, G, K, and M. The significances of the actual letters of the alphabet chosen to represent the spectral classes are mostly historical and are reminiscent of the analogous importance of the letters chosen to represent the various angular-momentum states in quantum mechanics. Each spectral class corresponds to a certain range in surface temperatures. Each of these major divisions or classes is further subdivided into 10 groups. For instance, the spectral type B is further subdivided into 10 subclasses labeled B0, B1, B2, . . . , B9, in order of decreasing temperature. A rough correlation of the surface temperatures of each spectral type with the prominent absorption lines appearing in the spectra of those stars is as follows:

Class O: Temperatures of 25,000°K and up. Lines of ionized helium are prominent. From the discussion of the Saha equation it is apparent that lines of ionized helium will appear only in such an extremely hot gas. Other atoms in

high degrees of ionization are observed. Class B: 25,000 to 11,000°K. The lines of hydrogen and neutral helium are conspicuous at class B0. Ionized oxygen and ionized carbon become strong at Neutral helium lines are strongest at class B5. Hydrogen lines become progressively stronger in the higher-numbered subdivisions of this class. By hydrogen lines we mean, of course, the Balmer series of hydrogen lines appearing in absorption. The intensity of such lines will, among other things, be proportional to the fraction of hydrogen atoms existing in the first excited state of hydrogen in thermal equilibrium. Thus, the strength of hydrogen lines is primarily determined by combined application of the Boltzmann and Saha formulas. Class A: 11,000 to 7500°K. At class A0 hydrogen and ionized magnesium lines are strongest, whereas the helium and ionized oxygen lines have disappeared. Hydrogen lines weaken continuously in the higher subdivisions of this class, whereas ionized metals (Fe, Ti, Ca, etc.) strengthen. The hydrogen lines will A PHYSICAL

continue t becomes le atoms in t because th In the hot nance line metals, he cools. Class F: strongest particular subdivisio Class G: strong, w cium are v to the cla-Class K: metals be continue t Class M: M stars is are strong

> The mi ever, som in a rang spectral li which par

Class S: terized by bands. bands. abundanc has been Classes h K and M carbide t Class W with bri oxygen, bon line lines and rapidly

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continue to weaken as we progress toward cooler stars, because the temperature becomes less and less sufficient to maintain a significant fraction of hydrogen atoms in the first excited state. The lines of ionized metals are growing stronger because the relative number of metals in low degrees of ionization is increasing. In the hotter stars, the higher degree of ionization of the metals produces resonance lines that lie in the ultraviolet. The resonance lines of the slightly ionized metals, however, lie in the visible, and these grow stronger as the temperature

cools.

Class F: 7500 to 6000°K. Class F0 is rich in lines of the ionized metals, the strongest being the H and K lines of singly ionized calcium. Metallic lines, particularly iron, strengthen and hydrogen lines weaken in the higher-numbered subdivisions of this class.

Class G: 6000 to 5000°K. In this class the lines of the neutral metals become strong, whereas the hydrogen lines continue to weaken. Lines of ionized calcium are very strong. Molecular bands of CN and CH appear. The sun belongs to the class G2.

Class K: 5000 to 3500°K. In general, molecular bands and lines of neutral metals become much stronger, whereas the lines of hydrogen and ionized metals continue to weaken. At K5 the lines of TiO are weakly visible.

Class M: 3500 to 2200°K. The characteristic feature of the spectrum of class M stars is the appearance of complex molecular oxide bands, of which TiO bands are strongest.

The majority of stars fall into one of these spectral classes. There are, however, some exceptions to this classification scheme: some stars have temperatures in a range parallel to one of the existing classes but show strikingly different spectral lines. Therefore, some additional spectral classes have been established which parallel the above classes in temperatures:

Class S: A low-temperature class parallel to class M. This class is still characterized by molecular oxide bands, but the most prominent feature is the ZrO bands. The elements Zr, Y, Ba, La, and Sr give strong atomic lines and oxide bands. Lines of neutral technetium are usually seen. It is believed that these abundances are enhanced because of nucleosynthesis within the interior which has been mixed to the surface.

Classes R and N (or Class C): Parallel in temperature to the ordinary classes K and M. The spectrum is characterized not by oxide bands but by molecular carbide bands, such as those of CN, C₂, and CH.

Class W: Extremely high-temperature type O objects, called Wolf-Rayet stars, with bright, broad emission lines of ionized helium and highly ionized carbon, oxygen, and nitrogen. Two sequences exist: (1) the WC stars have strong carbon lines and weak nitrogen lines, and (2) the WN stars have strong nitrogen lines and weak carbon lines. These stars are generally found to be emitting gas rapidly in space. The progression of spectral properties through the sequences

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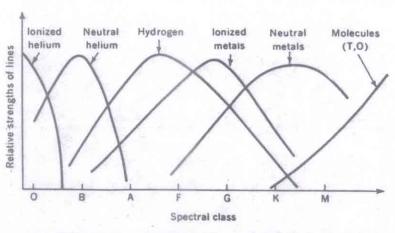


Fig. 1-11 The progression of selected spectral properties through the sequence of spectral classes. (G. Abell, "Exploration of the Universe," Holt, Rinehart and Winston, Inc., New York, 1964.)

of spectral classes and the correlation of color with surface temperature and spectral type are shown in Fig. 1-11.

Problem 1-23: What is the spectral type of a normal star having a maximum in its continuous spectrum at H_{α} ?

Ans: K3.

Problem 1-26: What is the spectral type of a star for which the number of hydrogen atoms in the first excited state exceeds the number in the second excited state by the ratio of 4:1?

Problem 1-25: Absorption lines of singly ionized helium cannot be seen if the temperature is too low, because the helium is mostly neutral and what little ionized helium there is lies in the ground level and cannot make visible absorption lines. On the other hand, the lines cannot be seen if the temperature is too high, because all the helium becomes doubly ionized. The helium absorption lines arising from a state in level n of singly ionized helium will be strongest when the largest possible fraction of the helium lies in that level, i.e., when $He_n^+/(He + He^+ + He^{++})$, the fraction of He in state n of He^+ , is a maximum. Show that this condition occurs when the quantity

$$\begin{split} G(\text{He}^+) \, \exp \left[\left(1 \, - \, \frac{1}{n^2} \right) \frac{\chi^+}{kT} \right] + G(\text{He}) \, \frac{n h^2}{2 (2 \pi m k T)^{\frac{3}{2}}} \exp \left\{ \frac{\chi^0 + [(n^2 \, - \, 1)/n^2] \chi^+}{kT} \right\} \\ + \frac{2 (2 \pi m k T)^{\frac{3}{2}}}{n h^2} \exp - \frac{\chi^+}{n^2 k T} \end{split}$$

is a minimum. The ionization potential of neutral helium is $\chi^o=24.58$ eV, and χ^+ is the ionization potential of the He⁺. Calculate for n=4 and electron densities in the range $\log n_*=19$ (the answer is not very sensitive to n_*) that temperature yielding the largest fraction of He⁺ ions in the n=4 level (assume G=1). You may also notice that most of the helium is doubly ionized at this temperature. What is the spectral type?

One of the synthesized l of a star sho the star for usually migr reactions in into somethi presented. lines of those Zr. Y. Ba, abundant in in the interior usual course ance of the of an unstab The nuclear such that th dance. The examined in

Another and N star because oxy after the for bands. In process or a oxygen; the formation o lution and those conta own lifetim

1-3 MASS

The masses when the or measure th center of m nents can a by a simple masses of t

$$\frac{M_1+M}{M_0+M_e}$$

where A i

Prissure Ionization The Souha Equation goes bird when the density is high enough so that the e-OBOAR Atom. Envelope

F 7500-6000 : More Met Ns on things cool off. GKM: Neutrals start appearing. One Portele Hion/ For very cool stones it was found that the is implied H= bound state of one p+2e-e- barely bound with 0.75eV, so unlikely except at low To This is crucial, as at low T, one must identify a process to make p's, in this care e+H=8+H Now, it ends up that the metals, make the free e- at some density, thus we get Me +MH= MH-

 $\frac{n(H)}{n(H)} = \frac{n_Q}{n_e} e^{\times p} \left(\frac{-0.75eV}{KT} \right)$

or just $\frac{h(H)}{h(H)} = \frac{ne}{nQ} \exp\left(\frac{8700}{T}\right)$ Take ne=1013 cm-3, then = 2.4x101013 exp (8714/T)
2.4x1021 (T/104K)3/2 = 10-8 exp (8700) Sun naut e-from metals! 4000 2000 3000 5000 So, we find that at 3000 = 10' of His in H which is about the peak.

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So, below 3000 k, there is a bit of a contentrophe, as most things are neutral. What we find begin to form and modify the oparity, dramatically.

Now, that is about all we will do for the photosphere. The next question is more for the state of the interior of the state of the interior of the star, where ionization willoccur. For marrive stare, nearly tell elements are ionized at the photosphere, however around the photosphere, however around the set in the sun, for example, the H is newtral at the sur face and ionizer in the interior. There zone have in them rather abrupt changes in the convertion quite often occurs.

Even deepen in the story of and and me innert thorny of out pressure ionization.