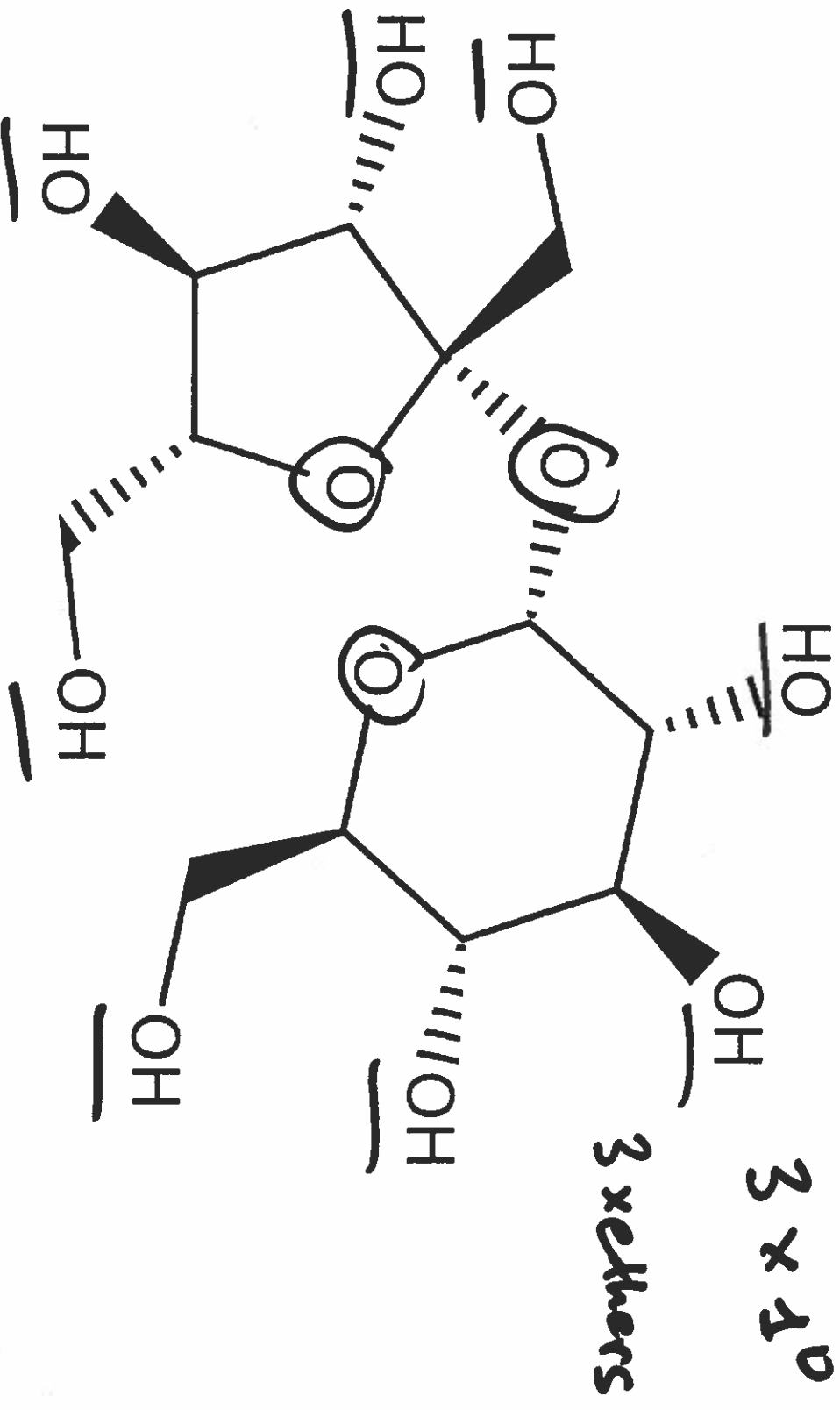


Comments on midterm 1

- Chapters 7 and 8*
 - *some of the material overlaps with Chapter 9.6 (synthesis of alcohols and ethers) and you may find relevant problems in the midterm
- Excludes double elimination (Chapter 8.10)
- What should (must?) you do first when preparing for midterm 1?
 - focus on everything S_N1 , S_N2 , $E1$, $E2$ (including synthesis)!!!

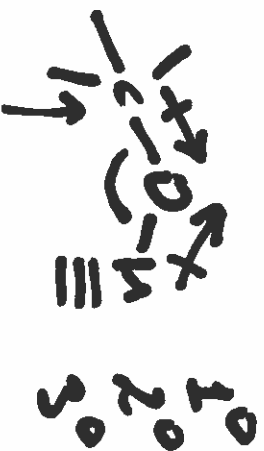
Sucrose (sugar)

5×2^0



Alcohols, ethers, epoxides

Alcohols, enols, phenols

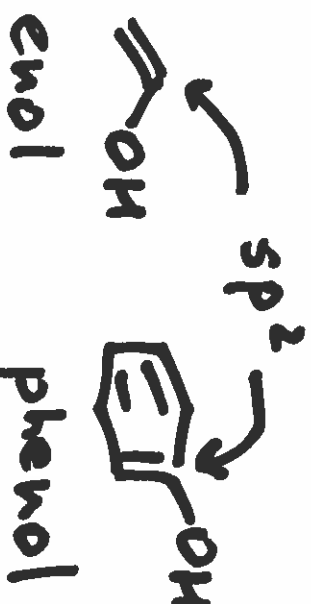
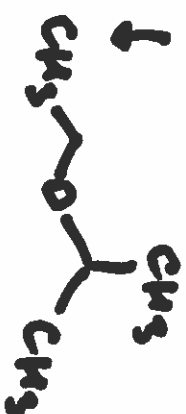


sp³ hybridized

Dialkyl ethers, symmetrical and unsymmetrical



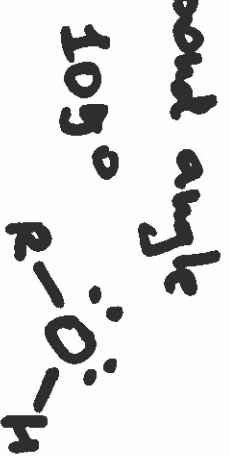
bond angle 111°



phenol



H-bonding!



Epoxides



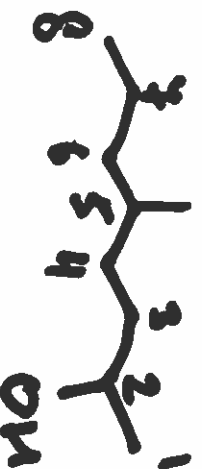
special case of

cyclic ethers

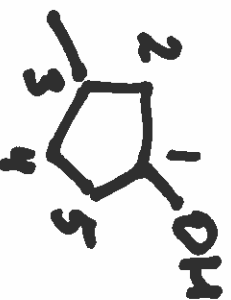
bond angle 60° → strain (angle strain)

Alcohols: nomenclature

IUPAC, common names, cyclic alcohols, diols, triols



- octane (8c)
- octan-2-ol
- 5-methyl octan-2-ol



- 3-methylcyclopentanol

no numbering
for cyclic
compounds



tert-butyl alcohol



sec-butyl alcohol

Alcohols: nomenclature

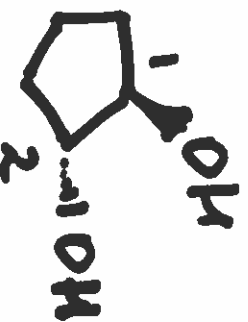
IUPAC, common names, cyclic alcohols, diols, ~~trials~~

diols or glycols → 2xOH groups



ethane-1,2-diol

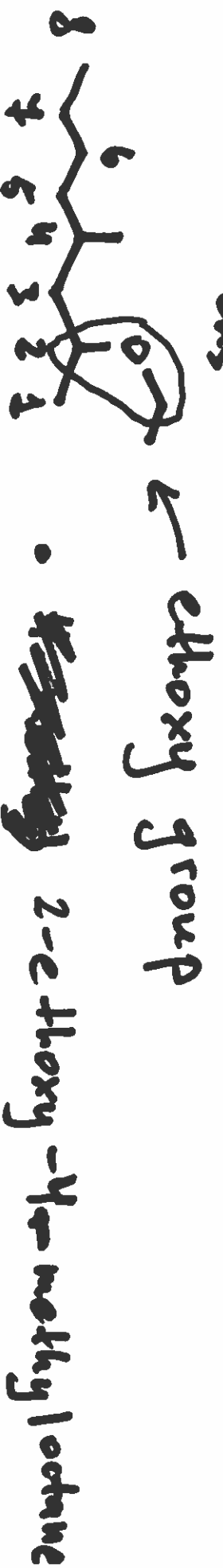
ethylene glycol



trans-cyclopentane-1,2-diol

Ethers: nomenclature

Simple ethers (dialkyl, alkyl alkyl), complex ethers, cyclic ethers



tetrahydrofuran



tetrahydropyran

Epoxides: nomenclature

Epoxylkanes, oxiranes, alkene oxides



1,2-epoxycyclohexane



cis-3,4-epoxyhexane

• alkyl chain \rightarrow hexane



oxirane



• 2,2-dimethyloxirane

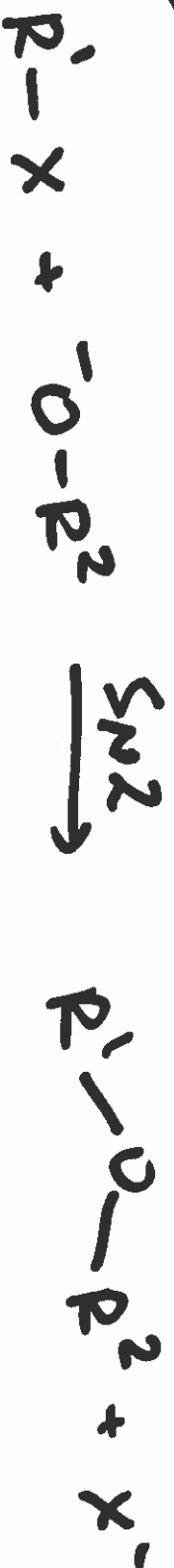
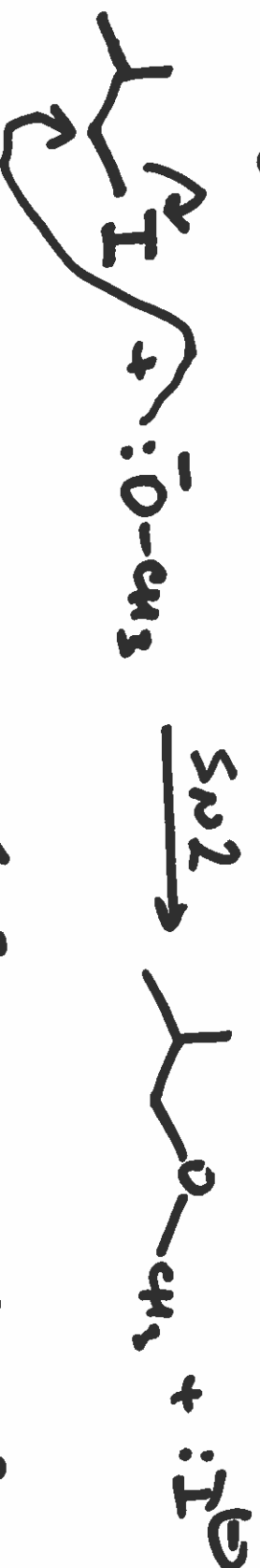
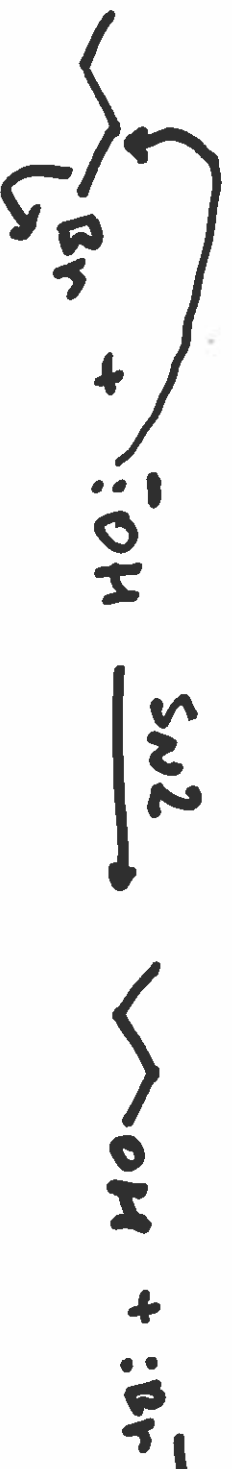


ethylene

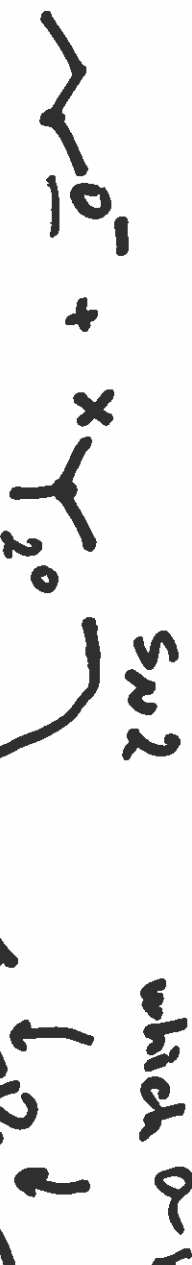
ethylene oxide

Alcohols, ethers, epoxides: synthesis

S_N2 reactions, Williamson ether synthesis

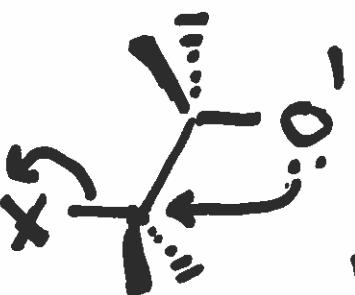


which O-bond to make?



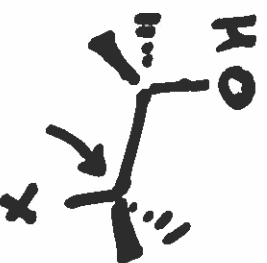
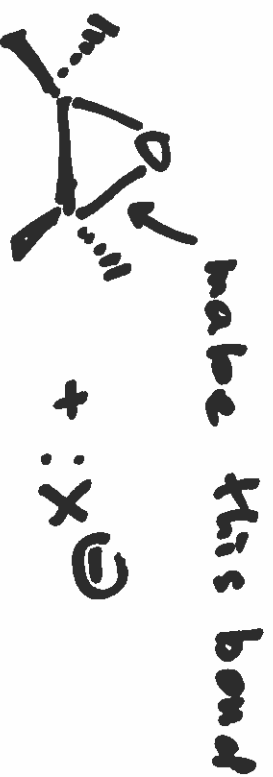
Alcohols, ethers, epoxides: synthesis

Intramolecular nucleophilic substitution reactions and synthesis of epoxides



intramolecular

Similar to S_N2



halohydrin

Alcohols, ethers, epoxides: reactions

Converting hydroxy group into a good leaving group and reactions of alcohols



good Nu

weak Nu

good leaving

bad leaving group

group