Name		
1 TUILL		

Period \_\_\_\_\_

## **Worksheet on Lewis Dots #1**

**Part I:** Draw the electron dot structure for each of the following atoms. 1. K 6. Be

2. P

7. S

3. I

8. Cs

4. Ca

9. Xe

5. Ga

10. Bi

**Part II:** Draw the Lewis Dot structures for the following molecules: 1.  $I_2$  6.  $CH_2I_2$ 

2. CCl<sub>4</sub>

7. NH<sub>3</sub>

3. HCl

8. PCl<sub>3</sub>

4. H<sub>2</sub>O

9. NF<sub>3</sub>

5. SiCl<sub>4</sub>

10.  $C_2H_6$