Introduction

The roaming mechanism for chemical reactions was introduced in reference to the description of the photodissociation of formaldehyde (Zee, Foltz, and Moore 1993; Townsend et al. 2004; Bowman 2006). After excitation of the formaldehyde molecule by a laser pulse, a single CH bond begins to stretch. Rather than proceed directly to dissociation, the hydrogen atom rotates around the molecular fragment in a region of the potential energy surface where it is subject to weak forces. The corresponding motion of the hydrogen atom was termed 'roaming'. At a later stage the roaming hydrogen atom encounters the bound hydrogen atom and undergoes an abstraction reaction. The resulting H₂ molecule then separates from the CO fragment. This reaction is then said to occur by the roaming mechanism.

Since the pioneering formaldehyde studies a similar roaming mechanism has been observed to occur in a number of chemical reactions. Many of these reactions are described in several review articles(Suits 2008; Bowman and Shepler 2011; Bowman and Suits 2011; Bowman 2014; Mauguière et al. 2017).

Essentially all of these roaming reactions have been studied at constant total energy. The goal of this article is to study analogous roaming reaction mechanisms at constant kinetic energy. Specifically, we study Chesnavich's model for an ion-molecule reaction subjected to a Hamiltonian isokinetic thermostat.

Chesnavich's empirical model for the ion-molecule reaction $\mathrm{CH}_4^+ \rightleftharpoons \mathrm{CH}_3^+ + \mathrm{H}$ was introduced in reference . A detailed derivation of this model can be found in reference . Chesnavich's model captures many of the essential aspects of the roaming mechanism. It describes the situation where a hydrogen atom separates from the rigid CH_3^+ core and, instead of dissociating, roams in a region of nearly constant potential only to return to the core. While Chesnavich's model does not accurately describe the intramolecular abstraction and subsequent dissociation, it has nevertheless provided significant insight into the roaming process, see, for example, references .

In the present work we apply a Hamiltonian version of the isokinetic thermostat to Chesnavich's model. This thermostat is not as widely used as the Nosé-Hoover thermostat, and its many variants, but the non-Hamiltonian version of the isokinetic thermostat has been developed and applied to a variety of problems of interest in chemistry by Minary et al. (Minary, Martyna, and Tuckerman 2003a, 2003b). In this thermostat, the particle momenta are subject to a nonholonomic constraint that keeps the kinetic energy constant. The resulting dynamics (if ergodic) generates a canonical distribution in configuration space, where the associated temperature is related to the value of the conserved kinetic energy (Morriss and Dettmann 1998). A Hamiltonian version of the isokinetic thermostat was given by Dettmann (Dettmann and Morriss 1996; Morriss and Dettmann 1998) (see also (Litniewski 1993; Morishita 2003)).

The Hamiltonian isokinetic thermostat dynamics explored in our paper is rela-

tively unfamiliar, although, as noted, there has been some relevant earlier work in the chemical literature (Minary, Martyna, and Tuckerman 2003a, 2003b). The basic theory (Dettmann and Morriss 1996; Morriss and Dettmann 1998) shows that Hamiltonian dynamics in terms of a set of variables related to the physical variables by a noncanonical transformation, together with the use of the exponentiated potential results in a canonical distribution in the physical configuration variables (assuming, as usual, ergodic dynamics). The unfamiliar aspect of the dynamics lies in the imposition of a nonholonomic constraint on the original variables; namely, constant kinetic energy. Again, the general theory shows that the effective temperature is proprtional to the value of the conserved kinetic energy.

It is in this sense that the nonholonomic dynamics explored here for the Chesnavich model is effectively at constant temperature. We do not claim that this dynamics is the same as obtained using a Boltzmann distribution for the physical momenta; nevertheless, one point of our paper is that the phase space structure of the nonholonomic system can in this case be analyzed using methods employed for the microcanonical case, and an analogous roaming mechanism explored.

For our purposes the Hamiltonian formulation of the isokinetic thermostat then has several advantages. There has been considerable development in our understanding of phase space structures governing chemical reaction dynamics in recent years, see, for example, references . This phase space structure approach has been applied to the Hamiltonian isokinetic thermostat formalism (Ezra and Wiggins 2008; Collins, Ezra, and Wiggins 2010). In this paper we continue these developments in our study of the Hamiltonian phase space structure associated with the roaming mechanism as exhibited in Chesnavich's model subjected to a Hamiltonian isokinetic thermostat.

This paper is outlined as follows: In Sec. 2 we discuss the Hamiltonian formulation of the isokinetic thermostat for a general system of the form 'kinetic + potential'. In Sec. 3 we introduce Chesnavich's model for roaming in CH_4^+ , and discuss the dynamical origin of roaming in terms of families of unstable periodic orbits and their associated invariant manifolds. In Sec. 4 we introduce the Hamiltonian for Chesnavich's model subject to an isokinetic thermostat. In the absence of self-retracing orbits for the thermostatted system, we find two families of periodic orbits that are relevant to the roaming phenomenon. Analysis of escape times and Lagrangian descriptors shows that roaming does indeed occur in the thermostatted system, via a mechanism analogous to that found in the Hamiltonian case. Sec. 5 concludes.

The Hamiltonian Isokinetic Thermostat {#sec: HIK}

Thermostatted Hamiltonian

We consider a Hamiltonian system

$$H(q,p) = T(q,p) + \Phi(q), \tag{1}$$

where the position-dependent kinetic energy T(q, p) is a quadratic form of the momenta and $\Phi(q)$ is a potential energy. The aim of the isokinetic thermostat is to constrain T to a constant value $T_0^2 > 0$.

We define the isokinetic Hamiltonian K by

$$\mathcal{K}(q,\pi) = e^{\Phi} T(q,\pi) - e^{-\Phi} T_0^2.$$

Via the non-canonical transformation,

$$\pi = e^{-\Phi} p \tag{2}$$

the level set $\mathcal{K}=0$ corresponds to constant kinetic energy $T=T_0^2$ in the system [eq:Ham]. Note that since T is a quadratic form of the momenta, from [eq:transf] we have

$$T(q, \pi) = T(q, e^{-\Phi}p) = e^{-2\Phi}T(q, p),$$

and hence

$$e^{\Phi}T(q,\pi) - e^{-\Phi}T_0^2 = e^{-\Phi}(T(q,p) - T_0^2).$$

Due to the Hamiltonian structure of the system, $\dot{\mathcal{K}}=0$ along the solutions of Hamilton's equations

$$\begin{split} \dot{q} &= \frac{\partial \mathcal{K}}{\partial \pi} = e^{\Phi} \frac{\partial T(q, \pi)}{\partial \pi}, \\ \dot{\pi} &= -\frac{\partial \mathcal{K}}{\partial q} = -e^{\Phi} \frac{\partial \Phi}{\partial q} T(q, \pi) - e^{\Phi} \frac{\partial T(q, \pi)}{\partial q} - e^{-\Phi} \frac{\partial \Phi}{\partial q} T_0^2. \end{split} \tag{3}$$

Time scaling

In this section, we show that the dynamics defined by equations [eq:eqKham] produce equivalent dynamics regardless of the value of T_0^2 . We prove that a suitable scaling of time and momenta transforms [eq:eqKham] into equations with $T_0 = 1$.

Consider the scaling of time and momenta by

$$s = T_0 t, \quad \Pi = \frac{\pi}{T_0}. \tag{4}$$

From [eq:tpscale] and

$$ds = T_0 dt, (5)$$

it follows that

$$\begin{split} \frac{dq}{ds} &= \frac{dt}{ds} \frac{dq}{dt} = \frac{1}{T_0} \dot{q}, \\ \frac{d\Pi}{ds} &= \frac{dt}{ds} \frac{d\Pi}{d\pi} \frac{d\pi}{dt} = \frac{1}{T_0^2} \dot{\pi}. \end{split} \tag{6}$$

Using [eq:eqKham], [eq:derivs] and

$$T(q,\Pi) = \frac{1}{T_0^2} T(q,\pi),$$
 (7)

we can write the equations of motion in s and Π as

$$\begin{split} \frac{dq}{ds} &= \frac{1}{T_0} e^{\Phi} \frac{\partial T(q,\pi)}{\partial \pi} = T_0 e^{\Phi} \frac{\partial T(q,\Pi)}{\partial \pi} = e^{\Phi} \frac{\partial T(q,\Pi)}{\partial \Pi}, \\ \frac{d\Pi}{ds} &= -e^{\Phi} \frac{\partial \Phi}{\partial q} \frac{1}{T_0^2} T(q,\pi) - e^{\Phi} \frac{1}{T_0^2} \frac{\partial T(q,\pi)}{\partial q} - e^{-\Phi} \frac{\partial \Phi}{\partial q} \frac{1}{T_0^2} T_0^2, \\ &= -e^{\Phi} \frac{\partial \Phi}{\partial q} T(q,\Pi) - e^{\Phi} \frac{\partial T(q,\Pi)}{\partial q} - e^{-\Phi} \frac{\partial \Phi}{\partial q}. \end{split}$$

These equations correspond to the isokinetic Hamiltonian with $T_0 = 1$

$$\mathcal{K}_1(q,\Pi) = e^{\Phi} T(q,\Pi) - e^{-\Phi},$$

which is related to K via

$$\mathcal{K}_1(q,\Pi) = \frac{1}{T_0^2} \mathcal{K}(q,\pi).$$

Clearly the dynamics of \mathcal{K} and \mathcal{K}_1 is equivalent and only differs by scaling by a constant factor.

Chesnavich's Model and Roaming

In this section we introduce Chesnavich's model for the ion-molecule reaction $\mathrm{CH}_4^+ \rightleftharpoons \mathrm{CH}_3^+ + \mathrm{H}$ and recall known results about roaming in this system. The model was introduced by Chesnavich (Chesnavich 1986) to investigate the transition from vibration/librational motion of the H-atom in a deep potential well representing CH_4^+ to nearly-free rotation in a flat and rotationally symmetric region representing the dissociated $\mathrm{CH}_3^+ + \mathrm{H}.$

Chesnavich's Model Hamiltonian

Chesnavich's CH_4^+ model is a Hamiltonian system with 2 degrees of freedom, consisting of a rigid CH_3^+ molecule and a mobile H atom. The system is defined by the Hamiltonian

$$H(r,\theta,p_r,p_\theta) = \frac{1}{2} \frac{p_r^2}{\mu} + \frac{1}{2} p_\theta^2 \left(\frac{1}{\mu r^2} + \frac{1}{I_{CH_3}} \right) + U(r,\theta), \tag{8}$$

where (r, θ, ϕ) are polar coordinates describing the position of the H-atom in a body-fixed frame attached to the CH₃⁺ core (coordinate ϕ is ignorable in this model). The reduced mass of the system $\mu = \frac{m_{CH_3}m_H}{m_{CH_3}+m_H}$, where $m_H = 1.007825$ u and $m_{CH_3} = 3m_H + 12.0$ u, and the moment of inertia of the rigid body CH₃⁺ $I_{CH_3} = 2.373409$ uÅ².

The potential $U(r, \theta)$ consists of a radial long range potential U_{CH} and a short range potential U_{coup} that models the short range anisotropy of the rigid CH₃⁺ body.

$$U(r,\theta) = U_{CH}(r) + U_{coup}(r,\theta). \tag{9}$$

It is characterised by two deep wells corresponding to bound CH_4^+ , two areas of high potential and a flat area beyond them as shown in Fig. [fig:pot].

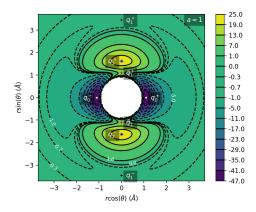


Figure 1: Contour plot of Chesnavich's potential energy surface U for a=1. Dashed lines correspond to U<0, solid lines correspond to $U\geq 0$. Contours correspond to values of potential shown on the colorbar right, with some values indicated in the plot.

The long range potential is

$$U_{CH}(r) = \frac{D_e}{c_1 - 6} \left(2(3 - c_2)e^{c_1(1 - x)} - (4c_2 - c_1c_2 + c_1)x^{-6} - (c_1 - 6)c_2x^{-4} \right),$$
(10)

with parameter values as used in the original work (Chesnavich 1986): $x = \frac{r}{r_e}$, dissociation energy $D_e = 47$ kcal/mol, equilibrium bond length $r_e = 1.1$ Å, $c_1 = 7.37$ and $c_2 = 1.61$. The short range hindered rotor potential U_{coup} is defined by

$$U_{coup}(r,\theta) = \frac{U_e e^{-a(r-r_e)^2}}{2} (1 - \cos 2\theta),$$
 (11)

where $U_e=55$ kcal/mol is the equilibrium barrier height. The distance at which transition occurs from rotation to vibration is determined by the parameter a (in Å $^{-2}$). Various values of a have been considered in previous works, namely a=1 (Chesnavich 1986; Mauguière, Collins, Ezra, Farantos, et al. 2014a, 2014b; Krajňák and Waalkens 2018), a=4 (Chesnavich 1986; Mauguière, Collins, Ezra, Farantos, et al. 2014a) and a range of values $0.7 \le a \le 8$. (Krajňák and Wiggins 2018)

The CH₃⁺ is a symmetric top in Chesnavich's model. Although the range of coordinate θ is $0 \le \theta \le \pi$, in the planar (zero overall angular momentum) version of the model the range of θ is extended to $0 \le \theta \le 2\pi$, and the potential has four-fold symmetry

$$U(r,\theta) = U(r,-\theta) = U(r,\pi-\theta) = U(r,\pi+\theta). \tag{12}$$

The potential admits four pairs of equilibrium points pairwise related by symmetry [eq:sym], as listed in Tab. [table:equil] and shown in Fig. [fig:pot].

Table 1: Equilibrium points of the potential $U(r, \theta)$.

| Energy (kcal mol^{-1}) | r (Å) | θ (radians) | Significance | Label |
|---|----------------------------|--------------------------------|---|---|
| $ \begin{array}{r} -47 \\ -0.63 \\ 8 \\ 22.27 \end{array} $ | 1.1 3.45 1.1 1.63 | $0 \\ \pi/2 \\ \pi/2 \\ \pi/2$ | potential well isomerisation saddle isomerisation saddle local maximum | $ \begin{array}{c} q_0^+ \\ q_1^+ \\ \widetilde{q}_1^+ \\ q_2^+ \end{array} $ |

Roaming in Chesnavich's Model

As mentioned in Section 1, we are interested in the roaming mechanism wherein the hydrogen atom separates from the CH_3^+ core only to return to the vicinity of the core before dissociating. Here we review the dynamical definition of roaming as introduced in (Mauguière, Collins, Ezra, Farantos, et al. 2014a), which is based on periodic orbits as invariant structures having dynamical relevance.

In the relevant energy interval $0 \le E \le 5$, there are three important families of periodic orbits (Mauguière, Collins, Ezra, Farantos, et al. 2014b). At any given fixed energy $0 \le E \le 5$, there are three pairs of important periodic orbits present in this system, pairwise related by symmetry [eq:sym], as shown in Fig.

[fig:orbitse5]. Therefore we refer to a continuum of periodic orbits parametrised by energy as a family of periodic orbits. We will refer to these families as the inner (Γ^i) , middle (Γ^a) and outer (Γ^o) periodic orbits. It is important to note that none of the orbits is directly related to a saddle point of the system.

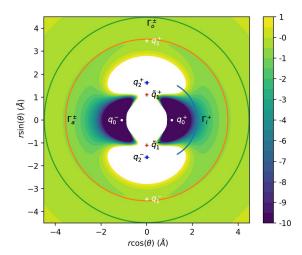


Figure 2: Configuration space projections of inner (Γ^i) , middle (Γ^a) and outer (Γ^o) periodic orbits for E=5.

Their significance is as follows:

- Γ^i : Delimits the potential wells that correspond to CH_4^+ isomers. The orbits oscillate around the axes $\theta=0$ and $\theta=\pi$.
- Γ^a : Two rotational orbits with opposite orientation one clockwise, one counter-clockwise that are crucial for defining roaming.
- Γ^o: Centrifugal barrier delimiting the region of dissociated states. Two rotational orbits with opposite orientation - one clockwise, one counterclockwise.

Roaming was shown (Krajňák and Waalkens 2018) to disappear for $E \geq 2.5$. In the energy interval 0 < E < 2.5, all of the above-mentioned periodic orbits are unstable. Define the dividing surfaces using DSⁱ, DS^a and DS^o as the set of all phase space points $(r, \theta, p_r, p_\theta)$ that satisfy

$$H(r, \theta, p_r, p_\theta) = E,$$

and their configuration space projections (r, θ) coincide with the configuration space projections of Γ^i , Γ^a and Γ^o respectively. Due to the instability of the orbits, the resulting dividing surfaces do not admit local recrossing.

A roaming trajectory is then defined as a trajectory that crosses DS^a an odd number of times between leaving the potential well and dissociating. Trajectories

that return to their region of origin cross DS^a an even number of times - isomerising trajectories returning to either of the potential wells, nonreactive trajectories return to dissociated states.

As explained in reference, the oscillatory nature of Γ^i implies that DS^i consists of two spheres, while the rotational nature of Γ^a and Γ^o means that DS^a and DS^o are tori(Mauguière et al. 2016). Each sphere can be divided using the corresponding periodic orbit into two hemispheres and each torus can be divided using both corresponding periodic orbits into two annuli(Mauguière et al. 2016). All hemispheres and annuli are surfaces of unidirectional flux, for example all trajectories leaving the potential well cross the same (outward) hemisphere of DS^i , while all trajectories entering the potential well cross the other (inward) hemisphere of DS^i .

Roaming can hereby be reformulated as a transport problem in phase space. Every trajectory leaving the well must cross the outward hemisphere of DS^i and every trajectory that dissociates must cross the outward annulus of DS^o . Dissociation of a CH_4^+ molecule is therefore equivalent to the transport of trajectories from the outward hemisphere of DS^i to the outward annulus of DS^o . Roaming involves crossing the inward annulus of DS^a , because between two crossings of the outward annulus trajectories must cross the inward annulus and vice versa.

Transport of trajectories in the neighbourhood of an unstable periodic orbit (or NHIM in general) is governed by invariant manifolds of this orbit (Wiggins et al. 2001; Uzer et al. 2002; Waalkens, Schubert, and Wiggins 2008; Wiggins 2016). It was shown(Krajňák and Waalkens 2018; Krajňák and Wiggins 2018) that the roaming phenomenon involves a heteroclinic intersection of the invariant manifolds of Γ^i and Γ^o . The condition $H(r, \theta = 0, p_r > 0, p_\theta = 0)$ defines an invariant subsystem that consists of precisely one dissociating trajectory for every fixed E > 0. Therefore if the invariant manifolds of Γ^i and Γ^o do not intersect, the former are contained in interior of the latter and each trajectory leaving the potential well dissociates immediately. An intersection assures that some trajectories leaving the well do not dissociate immediately and return to DS^a as illustrated in Fig. [fig:DSa]. This allows for roaming and isomerisation.

Isokinetic Chesnavich Model

Hamiltonian and periodic orbits for thermostatted Chesnavich model

The isokinetic Hamiltonian $\mathcal K$ for Chesnavich's model is defined as follows:

$$\mathcal{K}(r, \pi_r, \theta, \pi_\theta) = \frac{1}{2} e^U \left(\frac{\pi_r^2}{\mu} + \pi_\theta^2 \left(\frac{1}{\mu r^2} + \frac{1}{I_{CH_3}} \right) \right) - \frac{1}{2} e^{-U}, \tag{13}$$

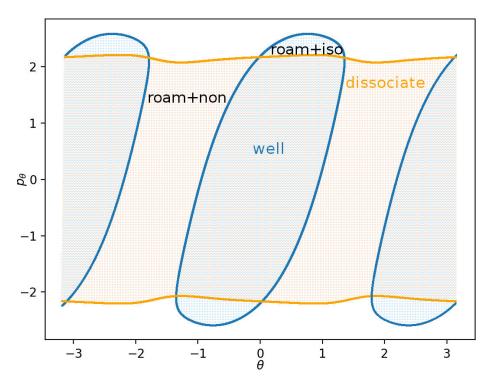


Figure 3: Intersection of invariant manifolds of Γ^i (blue) and Γ^o (orange) with the outward annulus of DS^a for E=1. Trajectories that just left the potential well are shown in blue, immediately dissociating trajectories in orange. Roaming and isomerising trajectories in the blue area just left the well and do not dissociate immediately, while roaming and dissociating trajectories in the orange are dissociate immediately but did not just leave the potential wells.

where $U = U(r, \theta)$ is Chesnavich's potential energy and

$$\pi_r = e^{-U} p_r, \quad \pi_\theta = e^{-U} p_\theta. \tag{14}$$

The level set $\mathcal{K} = 0$ corresponds to the surface of constant kinetic energy

$$\frac{1}{2}\frac{p_r^2}{\mu} + \frac{1}{2}p_\theta^2 \left(\frac{1}{\mu r^2} + \frac{1}{I_{CH_3}}\right) = \frac{1}{2},\tag{15}$$

in system [eq:chesHam]. Equations of motion in the isokinetic system are

$$\dot{r} = \frac{\partial \mathcal{K}}{\partial \pi_r} = e^U \frac{\pi_r}{\mu},$$

$$\dot{\pi}_r = -\frac{\partial \mathcal{K}}{\partial r} = -\frac{1}{2} e^U \frac{\partial U}{\partial r} \left(\frac{\pi_r^2}{\mu} + \pi_\theta^2 \left(\frac{1}{\mu r^2} + \frac{1}{I_{CH_3}} \right) \right) + \frac{1}{2} e^U \frac{2}{\mu r^3} \pi_\theta^2 - \frac{1}{2} e^{-U} \frac{\partial U}{\partial r},$$

$$\dot{\theta} = \frac{\partial \mathcal{K}}{\partial \pi_\theta} = e^U \pi_\theta \left(\frac{1}{\mu r^2} + \frac{1}{I_{CH_3}} \right),$$

$$\dot{\pi}_\theta = -\frac{\partial \mathcal{K}}{\partial \theta} = -\frac{1}{2} e^U \frac{\partial U}{\partial \theta} \left(\frac{\pi_r^2}{\mu} + \pi_\theta^2 \left(\frac{1}{\mu r^2} + \frac{1}{I_{CH_3}} \right) \right) - \frac{1}{2} e^{-U} \frac{\partial U}{\partial \theta}.$$
(16)

To achieve greater numerical precision, it is preferable to integrate the equations of motion in $(r, p_r, \theta, p_\theta)$ coordinates instead of $(r, \pi_r, \theta, \pi_\theta)$. Equations [eq:chesKHameq] transform using [eq:isotransf] to

$$\dot{r} = \frac{p_r}{\mu},
\dot{p}_r = p_r \left(\frac{\partial U}{\partial r} \dot{r} + \frac{\partial U}{\partial \theta} \dot{\theta} \right) + \frac{1}{\mu r^3} p_{\theta}^2 - \frac{\partial U}{\partial r},
\dot{\theta} = p_{\theta} \left(\frac{1}{\mu r^2} + \frac{1}{I_{CH_3}} \right),
\dot{p}_{\theta} = p_{\theta} \left(\frac{\partial U}{\partial r} \dot{r} + \frac{\partial U}{\partial \theta} \dot{\theta} \right) - \frac{\partial U}{\partial \theta},$$
(17)

where we used the isokinetic constraint [eq:constantT] equivalent to $\mathcal{K} = 0$.

The potential $-\frac{1}{2}e^{-U}$ has the same critical points and characteristics as U, but the wells are considerably deeper and have steeper walls. In contrast to the microcanonical case, the isokinetic model only possesses two periodic orbits with period 2π and due to constant nonzero kinetic energy does not admit self-retracing orbits (also referred to as brake orbits) such as Γ^i introduced in Section 3.2. One of the periodic orbits delimits the potential wells, see Fig. [fig:disI]; we therefore refer to it as the inner orbit.

The outer orbit, beyond which trajectories do not return and corresponds to the dissociated state of the molecule, is associated with a centrifugal barrier. It is

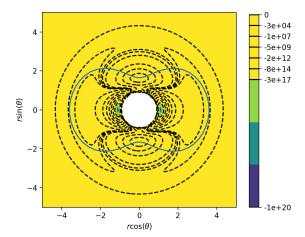


Figure 4: Inner periodic orbit on the potential energy surface $-\frac{1}{2}e^{-U}$.

rotationally symmetric and has a radius $r \approx 13.43$. Its existence can be proven using a similar argument as in the original system (Krajňák and Waalkens 2018): suppose r is sufficiently large so that U is effectively independent of θ . Denote r_{po} the solution of

$$\frac{1}{\mu r_{po}^3} p_{\theta}^2 - \frac{\partial U}{\partial r} = 0.$$

Then the equations [eq:chesKHameqreduced] admit a rotationally symmetric periodic orbit with $\dot{\theta} = const$, provided

$$\dot{r} = 0,$$

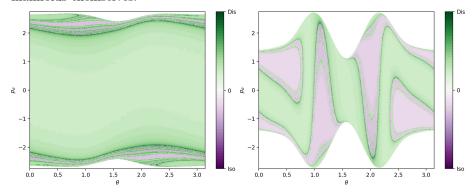
$$\dot{p}_r = 0.$$

This is satisfied by the initial condition $r=r_{po}, p_r=0$ and p_{θ} given implicitly by $\mathcal{K}=0$ for any θ . The existence of r_{po} is guaranteed for the potential U and any other potential with leading order term $-cr^{-(2+\varepsilon)}$ for large r, with c>0 and $\varepsilon>0$.

Both these orbits are unstable, with the largest eigenvalue of the inner orbit under the return map being of the order 10^{21} . This large instability poses a serious challenge to calculation of its invariant manifolds that guide trajectories in phase space.

Classification of trajectories and roaming in the isokinetic Chesnavich model

In this section we investigate dynamics in phase space and show the presence of roaming in the thermostatted Chesnavich model. Due to the strength of instability of the inner periodic orbit, we visualise phase space structures using escape times and Lagrangian descriptors(Jiménez Madrid and Mancho 2009; Lopesino et al. 2015) on surfaces of section rather than calculating invariant manifolds themselves.



The two most natural surfaces of section for this system are: (i) $\theta = 0$, $\dot{\theta} > 0$ and (ii) the outward annulus of the inner dividing surface (DS). The latter surface of section is constructed as follows: let the configuration space projection of the inner periodic orbit be parametrised by

$$r = \bar{r}(\theta). \tag{18}$$

Then all points in phase space satisfying [eq:disI] form a surface with coordinates θ and the canonically conjugate momentum(Arnold 1978)

$$p_{\sigma} = p_{\theta} + \bar{r}'(\theta)p_r$$
.

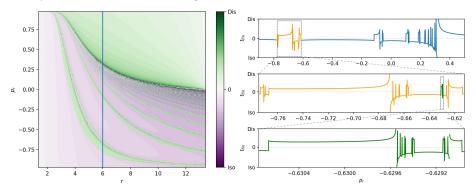
We remark that this surface is, similarly to DS^a and DS^o in Sec. 3.2, a torus. On this torus, the value of p_r^2 is given implicitly by θ , p_{σ} and [eq:constantT] and the sign of p_r is chosen so that

$$\dot{r} > \bar{r}'(\theta),$$

hence the outward direction on the inner DS.

The escape time is defined as the time a trajectory takes to escape the interaction region and reach a dissociated state beyond the outer orbit or either of the wells. Escape time plots obtained for initial conditions on the inner DS in the isokinetic model and on DS^a in the Hamiltonian Chesnavich model are shown in Fig. [fig:DSescape]. The dynamics on DS^a is explained in Sec. 3.2. We can see that while the surfaces share a toroidal geometry, the distributions of escape times of trajectories on them are quite dissimilar. Most of the trajectories in the microcanonical case escape in the same manner as the prototypical dissociation trajectory $\theta = p_{\sigma} = 0$ and more complicated escape dynamics are located near the edges of the surface of section. On the other hand, escape times for the isokinetic model reflect the effects of the constant kinetic energy constraint

note the nearly uniform dissociation around the local maximum at $\theta = \pi/2$ and nearly uniform isomerisation regions around $\theta = 1$ and $\theta = 2$.



The boundary between dissociating and isomerising trajectories, as well as between direct dissociation and more complicated dynamics, consists of invariant manifolds. Fig. [fig:thetaescape] shows escape times on the surface of section $\theta = 0$, $\dot{\theta} > 0$. Note the singularity in the plot separating a homogeneous area of relatively fast dissociation for large values of p_r from more complicated dynamics for lower values of p_r . The singularity can also be seen in the line plot showing escape times on r = 6, $\theta = 0$, $\dot{\theta} > 0$.

As it is the case in the microcanonical system, directly dissociating trajectories are guided by the stable invariant manifold of the outer periodic orbit across the phase space bottleneck associated with the outer periodic orbit. Directly dissociating trajectories are located above the stable invariant manifold in Fig. [fig:thetaescape]. All trajectories below the manifold correspond to roaming, isomerizing or nonreactive trajectories. As argued in Sec. 3.2, if there are trajectories on the inner DS that do not dissociate, the invariant manifolds of the inner and outer orbits must intersect. Fig. [fig:DSescape] shows regions of isomerization, therefore the manifolds intersect and all dissociating trajectories in the fractal structures of complicated dynamics on the inner DS correspond to roaming trajectories.

The bands of isomerizing and dissociating trajectories form a fractal structure and can be classified by the number of rotations in θ before escaping from the interaction region. These too are separated by invariant manifolds on the inner and outer periodic orbits. The fractal structure from Fig. [fig:thetaescape] can be understood using the configuration space projections of representative trajectories shown in Fig. [fig:timetraj]. In general, trajectories exhibit two dynamical 'motifs': they can perform a rotation at large r or interact with the areas of high potential near the local maxima at small r. Interaction with areas of high potential leads to escape out of the interaction region, capture in one of the potential wells, rotation in the same direction or rotation in the opposite direction. Trajectories are grouped by the sequence of these motifs in the fractal structure, for example trajectories that rotation by $\pi/2$ and then dissociate or

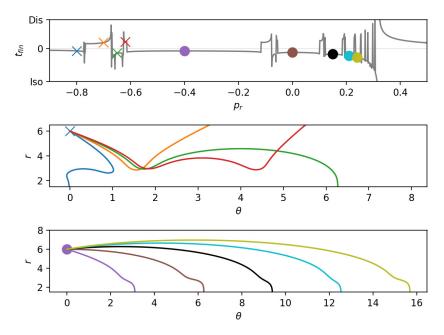
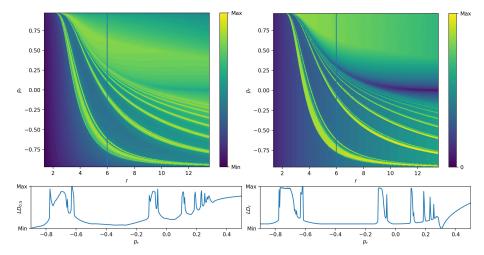


Figure 5: Representative trajectories for different classes along $\theta=0,\,\dot{\theta}>0,\,r=6.$

isomerise (see second panel in Fig. [fig:timetraj]) are close to each other and well separated from trajectories that rotate by π . Each class could be denoted by a sequence of integers denoting the number of rotations (see third panel in Fig. [fig:timetraj]) between interactions with the areas of high potential.

In this way we can find trajectories that perform any possible combination of rotations in the flat area of the interaction region and return any given number of times to the areas of high potential. In other words, regardless of the lack of self-retracing orbits in the interaction region, the scale of complicated dynamics exhibited by the isokinetic system is the same as in the microcanonical system.



It is important to note that the same results can be obtained using Lagrangian descriptors (Jiménez Madrid and Mancho 2009; Lopesino et al. 2015) as shown in Fig. [fig:thetald]. We present the plot for the forward Lagrangian descriptor

$$\int_{0}^{40} \left(|\dot{r}|^{\frac{1}{2}} + |\dot{p_r}|^{\frac{1}{2}} + |\dot{\theta}|^{\frac{1}{2}} + |\dot{p_{\theta}}|^{\frac{1}{2}} \right) dt, \tag{19}$$

with a cut-off at r=1.1 for numerical reasons. As opposed to escape times, this Lagrangian descriptor is well defined on invariant manifolds and attains a local minimum on them. Note that the invariant manifolds are visible for an integration time less than six times the period of the inner periodic orbit. The invariant manifolds are even more pronounced for the radial gain Lagrangian descriptor

$$\int_{0}^{40} |\dot{r}| dt. \tag{20}$$

Conclusions and Outlook

In this paper we have studied the nonholonomic dynamics of a Hamiltonian system under the constraint of constant kinetic energy enforced by a Hamiltonian isokinetic thermostat. The thermostatted dynamics, if ergodic, generates a canonical (constant temperature) distribution in the system configuration space. Changing the characteristic temperature is equivalent to time scaling.

We further investigated the roaming mechanism in Chesnavich's model for an ion-molecule reaction subject to an isokinetic thermostat. Imposing the nonholonomic constraint does not prevent the system from exhibiting roaming dynamics, where the origin of the roaming mechanism turns out to be analogous to that found in the Hamiltonian case.

The nonexistence of so-called "brake orbits" in the isokinetic case (periodic orbits with points of zero velocity) leads to differences in the detailed phase space structure as compared to the microcanonical case, but the qualitative description of the roaming mechanism as a result of trapping in a region of phase space demarcated by invariant objects remains unchanged.

Arnold, V. I. 1978. *Mathematical Methods of Classical Mechanics*. New York: Springer-Verlag.

Bowman, J. M. 2006. "Skirting the Transition State, a New Paradigm in Reaction Rate Theory." Proc.~Nat.~Acad.~Sci.~103~(44): 16061-2.~https://doi.org/10.1073/pnas.0607810103.

——. 2014. "Roaming." Mol. Phys. 112 (19): 2516–28. https://doi.org/10. 1080/00268976.2014.897395.

Bowman, J. M., and B. C. Shepler. 2011. "Roaming Radicals." *Annu. Rev. Phys. Chem.* 62: 531–53. https://doi.org/10.1146/annurev-physchem-032210-103518.

Bowman, J. M., and A. G. Suits. 2011. "Roaming reaction: The third way." *Phys. Today* 64 (11). AIP: 33. https://doi.org/10.1063/PT.3.1330.

Chesnavich, W. J. 1986. "Multiple Transition States in Unimolecular Reactions." J. Chem. Phys. 84 (5): 2615–9. https://doi.org/10.1063/1.450331.

Collins, Peter, Gregory S Ezra, and S. Wiggins. 2010. "Phase Space Structure and Dynamics for the Hamiltonian Isokinetic Thermostat." *J. Chem. Phys.* 133 (1). AIP: 014105.

Dettmann, C. P., and G. P. Morriss. 1996. "Hamiltonian formulation of the Gaussian isokinetic thermostat." *Phys. Rev. E* 54: 2495–2500. https://doi.org/10.1103/PhysRevE.54.2495.

Ezra, Gregory S, and S. Wiggins. 2008. "Impenetrable Barriers in Phase Space for Deterministic Thermostats." *Journal of Physics A: Mathematical and Theoretical* 42 (4). IOP Publishing: 042001.

Jiménez Madrid, J. A., and A. M. Mancho. 2009. "Distinguished Trajectories in Time Dependent Vector Fields." *Chaos* 19: 013111. https://doi.org/10.1063/1. 3056050.

Krajňák, V., and H. Waalkens. 2018. "The Phase Space Geometry Underlying Roaming Reaction Dynamics." *J. Math. Chem.* Springer, 1–38. https://doi.org/10.1007/s10910-018-0895-4.

Krajňák, V., and S. Wiggins. 2018. "Influence of Mass and Potential Energy Surface Geometry on Roaming in Chesnavich's Ch 4+ Model." *J. Chem. Phys.* 149 (9). AIP Publishing: 094109. https://doi.org/10.1063/1.5044532.

- Litniewski, M. 1993. "Molecular dynamics method for simulating constant temperature-volume and constant temperature-pressure systems." *J. Phys. Chem.* 97: 3842–8. https://doi.org/10.1021/j100117a035.
- Lopesino, C., F. Balibrea, S. Wiggins, and A. M. Mancho. 2015. "Lagrangian Descriptors for Two Dimensional, Area Preserving, Autonomous and Nonautonomous Maps." *Communications in Nonlinear Science and Numerical Simulation* 27 (1): 40–51. https://doi.org/https://doi.org/10.1016/j.cnsns.2015.02.022.
- Mauguière, F. A. L., P. Collins, G. S. Ezra, S. C. Farantos, and S. Wiggins. 2014a. "Roaming Dynamics in Ion-Molecule Reactions: Phase Space Reaction Pathways and Geometrical Interpretation." *J. Chem. Phys.* 140 (13). AIP Publishing: 134112. https://doi.org/10.1063/1.4870060.
- Mauguière, F. A. L., P. Collins, G. S. Ezra, S. C Farantos, and S. Wiggins. 2014b. "Multiple Transition States and Roaming in Ion–Molecule Reactions: A Phase Space Perspective." *Chem. Phys. Lett.* 592. Elsevier: 282–87. https://doi.org/10.1016/j.cplett.2013.12.051.
- Mauguière, F. A. L., P. Collins, Z. C. Kramer, B. K. Carpenter, G. S. Ezra, S. C. Farantos, and S. Wiggins. 2016. "Phase space barriers and dividing surfaces in the absence of critical points of the potential energy: Application to roaming in ozone." *J. Chem. Phys.* 144: 054107. https://doi.org/10.1063/1.4940798.
- ———. 2017. "Roaming: A Phase Space Perspective." *Annu. Rev. Phys. Chem.* 68 (1). Annual Reviews 4139 El Camino Way, PO Box 10139, Palo Alto, California 94303-0139, USA. https://doi.org/10.1146/annurev-physchem-052516-050613.
- Minary, P., G. J. Martyna, and M. E. Tuckerman. 2003a. "Algorithms and novel applications based on the isokinetic ensemble. I. Biophysical and path integral molecular dynamics." *J. Chem. Phys.* 118: 2510–26.
- ———. 2003b. "Algorithms and novel applications based on the isokinetic ensemble. II. Ab initio molecular dynamics." *J. Chem. Phys.* 118: 2527–38.
- Morishita, T. 2003. "Generalized coupling to a heat bath: Extension of the Gaussian isokinetic dynamics and effect of time scaling." *J. Chem. Phys.* 119: 7075–82. https://doi.org/10.1063/1.1605731.
- Morriss, G. P., and C. P. Dettmann. 1998. "Thermostats: Analysis and application." *Chaos* 8: 321–36. https://doi.org/10.1063/1.166314.
- Suits, A. G. 2008. "Roaming Atoms and Radicals: A New Mechanism in Molecular Dissociation." Acc. Chem. Res. 41 (7): 873–81. https://doi.org/10.1021/ar8000734.
- Townsend, D., S.A. Lahankar, S. K. Lee, S. D. Chambreau, A. G. Suits, X. Zhang, J. Rheinecker, L. B. Harding, and J. M. Bowman. 2004. "The roaming atom: Straying from the reaction path in formaldehyde decomposition." *Science*

306 (5699). American Association for the Advancement of Science: 1158-61. https://doi.org/10.1126/science.1104386.

Uzer, T., C. Jaffé, J. Palacián, P. Yanguas, and S. Wiggins. 2002. "The Geometry of Reaction Dynamics." *Nonlinearity* 15 (4). IOP Publishing: 957. https://doi.org/10.1088/0951-7715/15/4/301.

Waalkens, H., R. Schubert, and S. Wiggins. 2008. "Wigner's Dynamical Transition State Theory in Phase Space: Classical and Quantum." *Nonlinearity* 21 (1): R1–R118. https://doi.org/10.1088/0951-7715/21/1/R01.

Wiggins, S. 2016. "The Role of Normally Hyperbolic Invariant Manifolds (Nhims) in the Context of the Phase Space Setting for Chemical Reaction Dynamics." *Regul. Chaotic Dyn.* 21 (6). Springer: 621–38. https://doi.org/10.1134/S1560354716060034.

Wiggins, S., L. Wiesenfeld, C. Jaffe, and T. Uzer. 2001. "Impenetrable Barriers in Phase-Space." *Phys. Rev. Lett.* 86(24): 5478–81. https://doi.org/10.1103/PhysRevLett.86.5478.

Zee, R. D. van, M. F. Foltz, and C. B. Moore. 1993. "Evidence for a second molecular channel in the fragmentation of formaldehyde." *J. Chem. Phys.* 99 (3). AIP: 1664–73. https://doi.org/10.1063/1.465335.