Steam Tables Process Overview

Fundamentals of Engineering Thermodynamics

8th Edition by Moran, Shapiro, Boettner, and Bailey, 2014

Given Properties	Test to Determine Region of Vapor Dome	Vapor Dome Diagram	Where/How to Obtain Properties
1. p 2. T	Look up p in Table A-3/3E: a. If $T < T_{\text{sat}}$, Compressed liquid b. If $T = T_{\text{sat}}$, Two-phase, liquid-vapor mixture c. If $T > T_{\text{sat}}$, Superheated vapor d. If $T > T_{\text{c}}$, Superheated vapor	T T T T T T T T T T	a. Table A-5/5E or Table A-2/2E (approximate as saturated liquid) b. <i>T</i> and <i>p</i> are not independent; need another property c.&d. Table A-4/4E
1. p 2. T	Look up T in Table A-2/2E: a. If $p > p_{sat}$, Compressed liquid b. It $p = p_{sat}$, Two-phase, liquid- vapor mixture c. If $p < p_{sat}$, Superheated vapor	$p > p_{\text{sat}}$ $p = p_{\text{sat}}$ $p < p_{\text{sat}}$	a. Table A-5/5E or Table A-2/2E (approximate as saturated liquid at <i>T</i>) b. <i>T</i> and <i>p</i> are not independent; need another property c. Table A-4/4E
1. T 2. v, u, h, or s	Look up T in Table A-2/2E: a. If $v < v_f$, Compressed liquid b. If $v_f < v < v_g$, Two-phase, liquid-vapor mixture c. If $v > v_g$, Superheated vapor Apply the same procedure if u, h , or s is given.	$T \downarrow a \qquad b \qquad c \qquad v_{\text{g}} \qquad v \qquad v_{\text{g}} \qquad$	a. Table A-5/5E or Table A-2/2E (approximate as saturated liquid) b. Table A-2/2E and quality (<i>x</i>) calculations below c. Table A-4/4E
1. p 2. v, u, h, or s	Look up p in Table A-3/3E: a. If $v < v_f$, Compressed liquid b. If $v_f < v < v_g$, Two-phase, liquid-vapor mixture c. If $v > v_g$, Superheated vapor Apply the same procedure if $v_f < v_g < v_g$, and $v_f < v_g < v_g$, Superheated vapor	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	a. Table A-5/5E or Table A-2/2E (approximate as saturated liquid) b. Table A-3/3E and quality (x) calculations below c. Table A-4/4E
1. <i>T</i> 2. <i>x</i>	b. Two-phase, liquid vapor mixture		b. Table A-2/2E and quality (x) calculations below
1. p 2. x	b. Two-phase, liquid vapor mixture		b. Table A-3/3E and quality (x) calculations below

Quality Calculations (Eqs. 3.2, 3.6, 3.7, and 6.4)

$x = \frac{v - v_{\rm f}}{v_{\rm g} - v_{\rm f}}$	$x = \frac{u - u_{\rm f}}{u_{\rm g} - u_{\rm f}}$	$x = \frac{h - h_{\rm f}}{h_{\rm g} - h_{\rm f}} = \frac{h - h_{\rm f}}{h_{\rm fg}}$	$x = \frac{s - s_{f}}{s_{g} - s_{f}}$
$v = v_{\rm f} + x(v_{\rm g} - v_{\rm f})$	$u = u_{\rm f} + x(u_{\rm g} - u_{\rm f})$	$h = h_{\rm f} + x(h_{\rm g} - h_{\rm f}) = h_{\rm f} + xh_{\rm fg}$	$s = s_{\rm f} + x(s_{\rm g} - s_{\rm f})$