

1.

By the symmetry of the system we can see that

$$A + B = N \Rightarrow B = N - A$$

2.

$$\frac{dP_n}{dt} = P_{n-1} \left(\underbrace{k_1 (N-n+1)}_{\text{Reac. a)}} + \underbrace{\frac{k_2}{V} (n-1) (N-n+1)}_{\substack{\text{Reac. c)}} \\ B = N-(n-1)}} + \underbrace{\frac{k_3}{V^2} (n-1) (n-2) (N-n+1)}_{\substack{\text{Reac. e)}} \\ A \quad (A-1) \quad B}} \right) + P_{n+1} \left(\underbrace{k_1 (n+1)}_{\substack{\text{Reac. b)}} \\ A}} + \underbrace{\frac{k_2}{V} (n+1) (N-n-1)}_{\substack{\text{Reac. d)}} \\ A \quad B = N-(n+1)}} + \underbrace{\frac{k_3}{V^2} (n+1) (N-n-1) (N-n-2)}_{\substack{\text{Reac. f)}} \\ A \quad B \quad B-1}} \right)$$

term 2

$$- P_n \left(k_1 N + 2 \frac{k_2}{V} n (N-n) + \frac{k_3}{V^2} n (N-n) (N-2) \right)$$

term 3

Reactive reactions as follow:



$$P_n(t+dt) = P_n(t) \overset{*1}{- \text{prob. Something happens}} + P_{n+1}(t) \overset{*2}{+ \text{propensity func.}} + P_{n-1}(t) \overset{*3}{+ \text{propens. func.}}$$

*1 we are in the correct state and nothing happens.

*2 we have n+1 molecules of A, so we loose one. (Reac. b), d), f))

*3 we have n-1 molecules of A, so we get one. (Reac. a), c), e))