

# Thermodynamics

learning note For reading translation

我真的不懂忧郁



# Thermodynamics

learning note For reading translation

by

我真的不懂忧郁

Student Name	Student Number
First Surname	1234567

Instructor:	I. Surname
Teaching Assistant:	I. Surname
Project Duration:	Month, Year - Month, Year
Faculty:	Faculty of Aerospace Engineering, Delft

Cover: Canadarm 2 Robotic Arm Grapples SpaceX Dragon by NASA under  
CC BY-NC 2.0 (Modified)

Style: TU Delft Report Style, with modifications by Daan Zwaneveld

# Preface

*A preface...*

我真的不懂忧郁  
*Delft, October 2024*

# Summary

*A summary...*

# 目录

<b>Preface</b>	<b>i</b>
<b>Summary</b>	<b>ii</b>
<b>Nomenclature</b>	<b>iv</b>
<b>1 王竹溪, chapter1</b>	<b>1</b>
<b>References</b>	<b>3</b>
<b>A Source Code Example</b>	<b>4</b>
<b>B Task Division Example</b>	<b>5</b>

# Nomenclature

*If a nomenclature is required, a simple template can be found below for convenience. Feel free to use, adapt or completely remove.*

## Abbreviations

Abbreviation	Definition
ISA	International Standard Atmosphere
...	

## Symbols

Symbol	Definition	Unit
$V$	Velocity	[m/s]
...		
$\rho$	Density	[kg/m <sup>3</sup> ]
...		

# Chapter 1

王竹溪. chapter1

**Question 1** (p24,2):  $f, g, h$  都是二独立变量  $x, y$  的函数<sup>1</sup>

- (1)  $(\partial f / \partial g)_h = 1 / (\partial g / \partial f)_h$ ;
- (2)  $(\partial f / \partial g)_x = (\partial f / \partial y) / (\partial g / \partial y)$ ;
- (3)  $(\partial y / \partial x)_f = -(\partial f / \partial x) / (\partial f / \partial y)$ ;
- (4)  $(\partial f / \partial g)_h (\partial g / \partial h)_f (\partial h / \partial f)_g = -1$ ;
- (5)  $(\partial f / \partial x)_g = (\partial f / \partial x) + (\partial f / \partial y)(\partial y / \partial x)$ ;

**proof:** .

(1) 根据隐函数求导链式法则

$$\frac{\partial f}{\partial g} \cdot \frac{\partial g}{\partial f} = \frac{\partial f}{\partial f} = 1 \quad (1.1)$$

(2) 根据隐函数求导链式法则和反函数定理

$$\frac{\partial f}{\partial g} = \frac{\partial f}{\partial y} \cdot \frac{\partial y}{\partial g} = \frac{\partial f}{\partial y} / \frac{\partial g}{\partial y} \quad (1.2)$$

(3)

(4)

(5)

**Question 2** (p24,3):  $f, g, h, k$  都是二独立变量  $x, y$  的函数, 且

$$\frac{\partial(f, g)}{\partial(x, y)} = \begin{vmatrix} \frac{\partial f}{\partial x} & \frac{\partial f}{\partial y} \\ \frac{\partial g}{\partial x} & \frac{\partial g}{\partial y} \end{vmatrix} = \frac{\partial f}{\partial x} \frac{\partial g}{\partial y} - \frac{\partial f}{\partial y} \frac{\partial g}{\partial x} \quad (1.3)$$

证明

---

<sup>1</sup> $(\frac{\partial f}{\partial g})_h$  为保持  $h$  不变求  $f$  对  $g$  的偏导

- (1)  $\partial(f, g)/\partial(h, k) = [\partial(f, g)/\partial(x, y)]/[\partial(h, k)/\partial(x, y)];$
- (2)  $\partial(f, g)/\partial(x, y) = 1/(\partial(x, y)/\partial(f, g));$
- (3)  $(\partial f/\partial g)_h = \partial(f, h)/\partial(g, h);$
- (4)  $(\partial f/\partial g)_h = (\partial(f, h)/\partial(x, y))/(\partial(g, h)/\partial(x, y));$
- (5)  $(\partial f/\partial x)_g = (\partial(f, g)/\partial(x, y))/(\partial g/\partial y)$

**proof:**

上述结论可以推广到任意  $n$  个独立变量的函数上。

**Question 3** (p25,6): 证明理想气体的膨胀系数、压强系数以及压缩系数各为  $\alpha = \beta = 1/T, \kappa = 1/p$

**proof:**

**Question 4** (p25,7): 证明任何一个二独立变量  $\partial, p$  的物体, 其物态方程可以通过实验观测的膨胀系数  $\alpha$  和压缩系数  $\kappa$  根据下列积分得到

$$\ln V = \int (\alpha d\partial - \kappa dp) \quad (1.4)$$

应用这个公式到理想气体, 把积分求出, 选  $T$  为温标, 并假设  $\alpha = 1/T, \kappa = 1/p$

**proof:**

**Question 5** (p25,8): 假如某一测温物质的定压温度计的温标等于定容温度计的温标, 证明这一物质的物态方程为

$$\partial = \alpha(p + a)(V + b) + c \quad (1.5)$$

其中  $\alpha, a, b, c$  都是常数,  $\partial$  为这一物质的定压温度计和定容温度计所测量的共同的温度,

**proof:**



# References

- [1] I. Surname, I. Surname, and I. Surname. “The Title of the Article”. In: *The Title of the Journal* 1.2 (2000), pp. 123–456.

# Chapter A

## Source Code Example

*Adding source code to your report/thesis is supported with the package listings. An example can be found below. Files can be added using `\lstinputlisting[language=<language>]{<filename>}`.*

```
1 """
2 ISA Calculator: import the function, specify the height and it will return a
3 list in the following format: [Temperature,Density,Pressure,Speed of Sound].
4 Note that there is no check to see if the maximum altitude is reached.
5 """
6
7 import math
8 g0 = 9.80665
9 R = 287.0
10 layer1 = [0, 288.15, 101325.0]
11 alt = [0,11000,20000,32000,47000,51000,71000,86000]
12 a = [-.0065,0,.0010,.0028,0,-.0028,-.0020]
13
14 def atmosphere(h):
15     for i in range(0,len(alt)-1):
16         if h >= alt[i]:
17             layer0 = layer1[:]
18             layer1[0] = min(h,alt[i+1])
19             if a[i] != 0:
20                 layer1[1] = layer0[1] + a[i]*(layer1[0]-layer0[0])
21                 layer1[2] = layer0[2] * (layer1[1]/layer0[1])**(-g0/(a[i]*R))
22             else:
23                 layer1[2] = layer0[2]*math.exp((-g0/(R*layer1[1]))*(layer1[0]-layer0[0]))
24     return [layer1[1],layer1[2]/(R*layer1[1]),layer1[2],math.sqrt(1.4*R*layer1[1])]
```

# Chapter B

## Task Division Example

*If a task division is required, a simple template can be found below for convenience. Feel free to use, adapt or completely remove.*

表 B.1: Distribution of the workload

Task	Student Name(s)
Summary	
Chapter 1 Introduction	
Chapter 2	
Chapter 3	
Chapter *	
Chapter * Conclusion	
Editors	
CAD and Figures	
Document Design and Layout	