

$$\text{So } T = 70 + 230 e^k$$

Now using $T(3) = 200$

$$\Rightarrow 200 = 70 + 230 e^{3k}$$

$$2130 = 230 e^{3k}$$

$$\Rightarrow e^{3t} = \frac{13}{23} \Rightarrow 3t = \ln\left(\frac{13}{23}\right).$$

$$k = \frac{\ln(13/23)}{3} = -0.19018$$

Thus

$$T(t) = 70 + 230 e^{-0.19018 t}$$

Looking at this solution, we get $\lim_{t \rightarrow \infty} T(t) = 70$
 which says that cake will be at room temperature
 after a reasonably long time. It'll take approximately
 Half an hour to reach at room temperature.

$T(t)$	t (min)
75°	20.1
74°	21.3
73°	22.8
72°	24.9
71°	28.6
70.5°	32.3

Series Circuits

The linear differential equation for the current $i(t)$ through a series circuit containing only a resistor and inductor is

$$L \frac{di}{dt} + Ri = E(t) \quad (I)$$

where

L : Inductance

R : resistance

* The current $i(t)$ is called response of the system.

If Voltage drop across capacitor is $q(t)/c$, then DE for the current $i(t)$ through a series circuit containing a resistor and a capacitor is

$$Ri + \frac{q}{c} = E(t).$$

As $i = \frac{dq}{dt}$, so we get

$$R \frac{dq}{dt} + \frac{q}{c} = E(t)$$

Example A 12 volt battery is connected to a series circuit in which inductance is $\frac{1}{2}$ henry, and resistance is 10 ohms. Determine the current i , if initial current is zero.

Solution From (I),

$$\frac{1}{2} \frac{di}{dt} + 10i = 12 \quad i(0) = 0$$

$$\frac{di}{dt} + 20i = 24$$

$$I.F = e^{\int 20 dt} = e^{20t}$$

$$\Rightarrow \frac{d}{dt}(e^{20t}i) = 24e^{20t}$$

Integrating

$$\Rightarrow e^{20t}i = \frac{24}{20} e^{20t} + C$$

$$\Rightarrow i(t) = \frac{6}{5} + Ce^{-20t}$$

using I.C.

$$i(0) = \frac{6}{5} + Ce^0$$

$$0 = \frac{6}{5} + C \Rightarrow C = -6/5$$

So response is

$$i(t) = \frac{6}{5} - \frac{6}{5} e^{-20t}$$

we can write general solution of I is

$$i(t) = \frac{E_0}{R} + Ce^{(R/L)t}$$

MIXTURES

The mixing of two fluids sometimes gives rise to a linear first-order differential equation.

$$\frac{dA}{dt} = (\text{input rate of salt}) - (\text{output rate of salt})$$

$$\frac{dA}{dt} = R_{in} - R_{out} \quad (1)$$

Example

Consider a large tank containing 300 gallons of a brine solution. Salt is entering and leaving the tank; a brine solution was being pumped into the tank at the rate of 3 gal/min ; it mixed with the solution there, and then the mixture was pumped out at the rate of 3 gal/min . The concentration of the salt in the inflow, or solution entering, was 2 lb/gal , so salt was entering the tank at the

rate $R_{in} = 2 \frac{\text{lb}}{\text{gal}} \cdot (3 \frac{\text{gal}}{\text{min}}) = 6 \text{ lb/min}$ and leaving the tank at the rate $R_{out} = \left(\frac{A}{300 \text{ gal}}\right) \left(3 \frac{\text{gal}}{\text{min}}\right) = \frac{A}{100} \text{ lb/min}$. From this data and (1) we get equation

$$\frac{dA}{dt} + \frac{A}{100} = 6 \quad (2)$$

Let us pose the question: If 50 pounds of salt were dissolved initially in the 300 gallons, how much salt is in the tank after a long time?

Solution we have initial value problem

$$\frac{dA}{dt} + \frac{A}{100} = 6, \quad A(0) = 50$$

Here the initial condition $A(0) = 50$ gives the amount of salt
not the initial amount of liquid in the tank.

$$\text{Now I.F.} = e^{\int \frac{1}{100} dt}$$

$$= \int e^{t/100}$$

So Eq. becomes

$$\frac{d}{dt} (A e^{t/100}) = 6 e^{t/100}$$

Integrating

$$A e^{t/100} = 6 \frac{e^{t/100}}{1/100} + C$$

$$A e^{t/100} = 600 e^{t/100} + C$$

$$A = 600 + C e^{-t/100}$$

using I.C.

$$A(0) = 600 + Ce^0 = 50$$

$$C = 500 - 50$$

$$= -550$$

$$A(t) = 600 - 550e^{-t/100}$$

As $t \rightarrow \infty$

$$A(t) = 600$$

Q4: The population of bacteria in a culture grows at a rate proportional to the number of bacteria present at time t . After 3 hours it is observed that 400 bacteria are present. After 10 hours 2000 bacteria are present. What was the initial number of bacteria?

Let population of bacteria in culture is x . Then we have

$$\frac{dx}{dt} = kx \quad (1)$$

with conditions

$$x(3) = 400, \quad x(10) = 2000$$

Solving Eq. (1)

$$\begin{aligned} \frac{dx}{x} &= kdt \\ \ln x &= kt + c_1 \\ \ln x &= \ln e^{kt+c_1} \\ x &= ce^{kt} \end{aligned} \quad (2)$$

using initial condition, $x(3) = 400$

$$\Rightarrow 400 = ce^{3k}$$

$$c = \frac{400}{e^{3k}}$$

using in (2)

$$\Rightarrow x = \frac{400}{e^{3k}} e^{kt} = 400 e^{(t-3)k}$$

Now using 2nd initial condition

$$x(10) = 400 e^{(10-3)k}$$

$$2000 = 400 e^{7k} \Rightarrow e^{7k} = 5$$

$$\Rightarrow 7k = \ln 5$$

$$k = \frac{\ln 5}{7} = 0.23$$

So we have

$$x(t) = 400 e^{0.23(t-3)}$$

To find the initial number of bacteria, put $t = 0$

$$\Rightarrow x(0) = 400 e^{0.23(-3)} = 201$$

Q9: When a vertical beam of light passes through a transparent medium, the rate at which its intensity I decreases is proportional to $I(t)$, where t represents the thickness of the medium (in feet). In clear seawater, the intensity 3 feet below the surface is 25% of the initial intensity I_0 of the incident beam. What is the intensity of the beam 15 feet below the surface?

As intensity of light decreases, proportional to $I(t)$

So we have

$$\frac{dI}{dt} = -KI \quad (1)$$

with conditions

$$I(t=0) = I_0 \quad (2)$$

$$I(t=3) = 25\% I_0 = \frac{25}{100} I_0 = 0.25 I_0 \quad (3)$$

We have to find $I(t=15)$

Solving Eq (1)

$$\Rightarrow \frac{dI}{I} = -K dt$$

$$\Rightarrow \ln I = -Kt + \ln C$$

$$\ln I = \ln e^{-kt} + \ln C$$

$$\Rightarrow I(t) = Ce^{-kt}$$

Using (2)

$$I(0) = Ce^0 \Rightarrow I_0 = C$$

$$\Rightarrow I(t) = I_0 e^{-kt} \quad (4)$$

Now using $I(3) = 0.25 I_0$

$$\Rightarrow 0.25 I_0 = I_0 e^{-3K} \Rightarrow e^{-3K} = 0.25$$

$$-3K = \ln(0.25) \Rightarrow K = -\frac{\ln(0.25)}{3} \\ = 0.462$$

Using in (4)

$$\Rightarrow I(t) = I_0 e^{-0.462t}$$

Now

$$I(15) = I_0 e^{-0.462 \times 15} = 0.00098 I_0 \text{ W/ft.}$$

Q11: Archaeologists used pieces of burned wood, or charcoal, found at the site to date prehistoric paintings and drawings on walls and ceilings of a cave in Lascaux, France. Determine the approximate age of a piece of burned wood, if it was found that 85.5% of the C-14 found in living trees of the same type had decayed.

We have

$$\frac{dA}{dt} = KA$$

$$A(0) = A_0 \quad \text{and} \quad A\left(\frac{1}{2}\right) = A(5600) = \frac{A_0}{2}$$

$$\text{So} \quad A = A_0 e^{-\frac{\ln 2}{5600} t}$$

As 85.5% of A is decayed, we have $A = 14.5\%$.

At $A = 0.145 A_0$

$$= 0.145 A_0 = A_0 e^{-\frac{\ln 2}{5600} t}$$

$$-\frac{\ln 2}{5600} t = \ln(0.145)$$

$$t = -\frac{5600 \ln(0.145)}{\ln 2}$$

$$= 15963 \text{ years}$$

Q16: Two large containers A and B of the same size are filled with different fluids. The fluids in containers A and B are maintained at 0°C and 100°C , respectively. A small metal bar, whose initial temperature is 100°C , is lowered into container A. After 1 minute the temperature of the bar is 90°C . After 2 minutes the bar is removed and instantly transferred to the other container. After 1 minute in container B the temperature of the bar rises 10° . How long, measured from the start of the entire process, will it take the bar to reach 99.9°C ?

A small metal bar of Temperature 100°C is lowered into container A with temperature 0°C & then it is lowered into B with temperature 100°C

By Newton's Law of cooling

$$\frac{dT}{dt} = K(T - T_m)$$

Conditions

$$T = T_0 \text{ for container A at } t=0 = 100^\circ\text{C}$$

$$T = (at t=1) = 90^\circ$$

$$T = T_0 \text{ for container B at } (t=2 \text{ in container A}).$$

$$T = (T_0 + 1^\circ) \text{ for container B at } (t=1 \text{ in container A})$$

Goal

we have to find the time when temperature of metal bar is 99.9°C .

For container A. ($T_m = 0^\circ\text{C}$)

$$\frac{dT}{dt} = K_A (T - 0)$$

$$\frac{dT}{T} = K_A dt$$

$$\Rightarrow \ln T = K_A t + \ln C$$

$$T(t) = C e^{K_A t}$$

$$\text{At } t=0, T = 100$$

$$\Rightarrow C = 100$$

$$T(t) = 100 e^{K_A t}$$

$$\text{At } t=1, T=90^\circ$$

$$\Rightarrow 90 = 100 e^{K_A}$$

$$\Rightarrow e^{K_A} = \frac{90}{100}$$

$$K_A = \ln\left(\frac{90}{100}\right) = -0.105$$

Temperature of bar after $t=2$

$$T(t) = 100 e^{-0.105t}$$

$$T(2) = 100 e^{-0.105 \times 2}$$

$$= 81^\circ C$$

Now this is initial temperature of metal bar when it is transferred to B.

For container B

$$T_m = 100^\circ C$$

$$\frac{dT}{dt} = K_B (T - 100)$$

$$\Rightarrow \frac{dT}{T-100} = K_B t \Rightarrow T - 100 = C e^{K_B t}$$

$$\Rightarrow T = 100 + C e^{K_B t}$$

$$\text{At } t=0, T = 81^\circ$$

$$\Rightarrow 81 = 100 + C e^0 \Rightarrow C = -19$$

$$\Rightarrow T = -19 e^{K_B t} + 100$$

$$\text{Now at } t=1, T = 81 + 10 = 91$$

$$\Rightarrow 91 = -19 e^{K_B} + 100$$

$$\Rightarrow e^{K_B} = 9/19 \Rightarrow K_B = \ln(9/19) = -0.747$$

$$\text{So } T(t) = -19 e^{-0.747t} + 100$$

$$\text{Finding } t \text{ when } T = 99.9$$

$$\Rightarrow 99.9 = -19 e^{-0.747t} + 100$$

$$-19 e^{-0.747t} = -0.1.$$

$$e^{-0.747t} = \frac{1}{190}$$

$$t = \frac{\ln(1/190)}{-0.747}$$

$$t = 7.02 \text{ minutes.}$$

Now total time at which metal bar reaches 99.9°C

$$\begin{aligned} \Rightarrow \text{Total time} &= \text{Time in container A} + \text{Time} \\ &\quad \text{in container B} \\ &= 2 + 7.02 \\ &= 9.02 \end{aligned}$$

Q23: A large tank is filled to capacity with 500 gallons of pure water. Brine containing 2 pounds of salt per gallon is pumped into the tank at a rate of 5 gal/min. The well mixed solution is pumped out at the same rate. Find the number $A(t)$ of pounds of salt in the tank at time t .

Solution Initially, we have 500 gallons of pure water with no salt.

Input flow rate F_{in} of brine containing 2 pounds/gallon of salt is same as output flow rate F_{out} of well mixed fluid

$$\frac{dA}{dt} = \text{Input rate of salt} - \text{output rate of salt}$$

$$= F_{in} C_{in} - F_{out} C_{out}$$

A is amount of salt

C is concentration of salt

As $F_{in} = F_{out} = 5 \text{ gal/min}$, $C_m = 2 \text{ lb/gallon}$.

$$C_{out} = \frac{\text{Amount of salt}}{\text{Volume of tank}} = \frac{A}{500} \text{ lb/gallon}$$

So

$$\frac{dA}{dt} = 5 \times 2 - 5 \times \frac{A}{500}$$

$$= 10 - \frac{A}{100}$$

$$\int \frac{dA}{10 - A/100} = \int dt$$

$$-100 \int \frac{-dA}{1000 - A} = \int dt$$

$$-100 \ln |1000 - A| = t + C_1$$

$$\ln |1000 - A| = -\frac{t}{100} + C_2$$

$$1000 - A = \frac{-t}{100} + C_2$$

$$1000 - A = e^{C_2} e^{-t/100}$$

$$A(t) = 1000 - C e^{-t/100}$$

Now using the initial condition that initially there is no salt i.e. At $t=0$, $A=0$

$$\Rightarrow 0 = 1000 - C e^0$$

$$\Rightarrow C = 1000$$

$$\Rightarrow A(t) = 1000 - 1000 e^{-t/100}$$

Q32: A 200-volt electromotive force is applied to an RC series circuit in which the resistance is 1000 ohms and the capacitance is 5×10^{-6} farad. Find the charge $q(t)$ on the capacitor if $i(0) = 0.4$. Determine the charge and current at $t = 0.005$ s. Determine the charge as $t \rightarrow \infty$.

The differential eq. for charge is

$$R \frac{dq}{dt} + \frac{q}{C} = E(t)$$

we have to obtain charge $q(t)$ on capacitor, when $i(0) = 0.4$.

$$\text{As } R = 1000 \Omega \quad C = 5 \times 10^{-6}, \quad E = 200$$

$$\Rightarrow 1000 \frac{dq}{dt} + \frac{q}{5 \times 10^{-6}} = 200$$

$$\Rightarrow \frac{dq}{dt} + 200q = 0.2$$

$$\frac{dq}{dt} = 0.2 - 200q$$

Solving DE

$$\Rightarrow -\frac{1}{200} \int \frac{-200 dq}{0.2 - 200q} = \int dt$$

$$-\frac{1}{200} \ln |0.2 - 200q| = t + \ln c$$

$$\ln |0.2 - 200q| = -200t + \ln C.$$

$$0.2 - 200q = Ce^{-200t}$$

$$200q = 0.2 - Ce^{-200t}$$

$$q = \frac{1}{1000} - Ke^{-200t}$$

As $i(t) = \frac{dq}{dt}$

$$i(t) = 200Ke^{-200t}$$

As $i(0) = 0.4$

$$\Rightarrow 0.4 = 200Ke^0 \Rightarrow K = \frac{1}{500}$$

So

$$q(t) = \frac{1}{1000} - \frac{1}{500} e^{-200t}$$

& $i(t) = \frac{2}{5} e^{-200t}$

To find charge after $t = 0.005$

$$q(0.005) = \frac{1}{1000} - \frac{1}{500} e^{-200 \times 0.005}$$

$$= 2.64 \times 10^{-4} \text{ C}$$

To find current $i(t)$ at $t = 0.005$

$$i(0.005) = \frac{2}{5} e^{-200 \times 0.005}$$

$$= 14.72 \times 10^{-2} \text{ Amp}$$

To find charge as $t \rightarrow \infty$

$$q = \frac{1}{1000} - \frac{1}{500} e^{-200t \rightarrow \infty}$$

$$= \frac{1}{1000}$$

Q42: A model that describes the population of a fishery in which harvesting takes place at a constant rate is given by

$$\frac{dP}{dt} = kP - h,$$

where k and h are positive constants.

- Solve the DE subject to $P(0) = P_0$.
- Describe the behavior of the population $P(t)$ for increasing time in the three cases $P_0 > \frac{h}{k}$, $P_0 = \frac{h}{k}$ and $0 < P_0 < \frac{h}{k}$
- Use the results from part (b) to determine whether the fish population will ever go extinct in finite time, that is, whether there exists a time $T > 0$ such that $P(T) = 0$. If the population goes extinct, then find T .

Solution (a)

Solving

$$\frac{dP}{dt} = KP - h$$

$$\Rightarrow \int \frac{dP}{KP-h} = \int dt$$

$$\frac{1}{K} \int \frac{K dP}{KP-h} = \int dt$$

$$\frac{1}{K} \ln|KP-h| = t + \ln C$$

$$\ln|KP-h| = kt + \ln C$$

$$KP-h = Ce^{kt}$$

$$KP = h + Ce^{kt}$$

$$P = \frac{h}{K} + Ce^{kt}$$

using I.C. $P(0) = P_0$

$$\Rightarrow P_0 = \frac{h}{K} + c \Rightarrow c = P_0 - \frac{h}{K}$$

So

$$P(t) = \frac{h}{K} + \left(P_0 - \frac{h}{K}\right)e^{kt} \quad (A)$$

(b) For $P_0 > h/k$

R.H.S of Eq (A) is positive, so population of the fishery keep on increasing.

$$\text{For } P_0 = \frac{h}{K}$$

R.H.S of (A) becomes $\frac{h}{K}$, so population is constant.

$$\text{For } 0 < P_0 < h/K.$$

R.H.S of (A) is -ve, so population of fishery decreases.

(C) Fish population will go extinct for $0 < P_0 < h/K$.

To find the time where population become extinct let we have

$$P_0 - \frac{h}{K} = P'$$

using $P=0$ & $P_0 - \frac{h}{K} = P'$ in (A)

$$\Rightarrow 0 = -P' e^{kt_c} + \frac{h}{K}$$

$$P' e^{kt_c} = \frac{h}{K} \Rightarrow e^{kt_c} = \frac{h}{kp'}$$

$$kt_c = \ln\left(\frac{h}{kp'}\right)$$

$$t_c = \frac{1}{k} \ln\left(\frac{h}{kp'}\right)$$