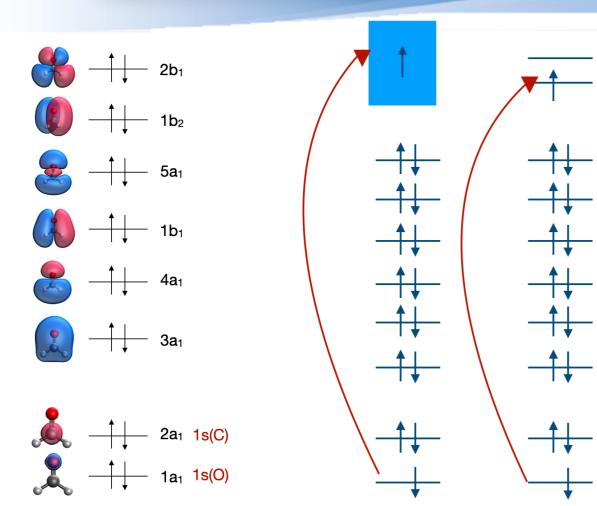
O-CHEM A QUANTUM LEAP INTO THE FUTURE OF CHEMISTRY

CVS-EOM-IP and CVS-EOM-EE calculations of formaldehyde (CH₂O)

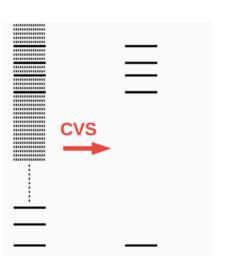


- X-ray spectroscopies are based on transitions involving core electrons
- These states can be computed using CVS-EOM-CC methods (IP and EE)

Core ionization (XPS) Core excitation (XAS)

 Transitions between core-level states and valence states can also be computed (for XES)

- Core-level states are high in energy
- Core-level states are metastable wrt electron loss
- Core-valence separation (CVS) projects the continuum out and removes low-lying states



$$|\Psi_I>=\hat{R}_Ie^{\hat{T}}|\Phi_0>$$

$$\bar{H}\hat{R}_I=E_n\hat{R}_I;\;\bar{H}=e^{-\hat{T}}He^{\hat{T}}$$

$$\hat{R}_{CVS-IP}=\sum_Ir_I\hat{a}_I+\frac{1}{2}\sum_{a,I,J}r_{IJ}^a\hat{a}_a^{\dagger}\hat{a}_J\hat{a}_I+\sum_{a,I,J}r_{IJ}^a\hat{a}_a^{\dagger}\hat{a}_J\hat{a}_I\sum_{i,j,k,\ldots}^{I,J,K,\ldots}\text{-occupied core spin-orbitals}\\ \hat{R}_{CVS-EE}=r_0+\sum_Ir_I^a\hat{a}_a^{\dagger}\hat{a}_I+\frac{1}{4}\sum_{a,I,J}r_{IJ}^{ab}\hat{a}_a^{\dagger}\hat{a}_J^{\dagger}\hat{a}_I+\frac{1}{4}\sum_{a,I,J}r_{IJ}^{ab}\hat{a}_a^{\dagger}\hat{a}_J^{\dagger}\hat{a}_I+\frac{1}{4}\sum_{a,I,J}r_{IJ}^{ab}\hat{a}_a^{\dagger}\hat{a}_J^{\dagger}\hat{a}_I+\frac{1}{4}\sum_{a,I,J}r_{IJ}^{ab}\hat{a}_a^{\dagger}\hat{a}_J^{\dagger}\hat{a}_I$$



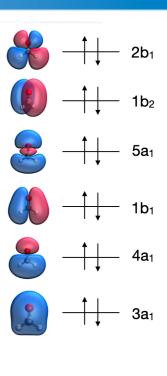
CVS-EOM-IP/EE calculation of formaldehyde (CH₂O)

Core-ionized states at carbon edge for CH₂O

- Import structure (<u>formaldehyde.xyz</u>) to IQmol
- Freeze 2 core orbitals (N_FROZEN_CORE = 2)
- Request one CVS-IP state (of A1 symmetry)

```
$rem
METHOD = EOM-CCSD
BASIS = 6-31g
CVS_IP_STATES = [1,0,0,0]
N_FROZEN_CORE = 2
CC_DO_DYSON = 1
GUI = 2
MEM_TOTAL = 4000
$end
```

!Specify the method !Specify the basis for calculation !Requesting CVS-IP states in A₁ symmetry !N_FROZEN_CORE = n will freeze n cores. !To obtain Dyson orbitals



1a₁ 1s(O)

- To compute oxygen edge, use N_FROZEN_CORE = 1
- To compute Dyson orbitals, use CC_DO_DYSON = 1
- To compute core-excited states (for XAS) use keywords CVS_EE_STATES = [1,0,0,0]
- Use CC_TRANS_PROP = 1 for transition properties (XAS intensities).



CVS-EOM-IP/EE calculation of formaldehyde (CH₂O)

CVS-EOM-IP state	Energy (eV)	Dyson orbital
1/A ₁ C-edge	297.51	
1/A ₁ O-edge	544.16	

CVS-EOM-EE state	Energy (eV)	Oscillator strength
1/A ₁ C-edge	295.12	0.015
1/A ₁ O-edge	543.37	0.005

- The excitation energy in IP and EE calculation gives ionization (XPS) and excitation (XAS) energies, respectively
- The oscillator strength, transition dipole moment are printed out in transition properties section
- Note the huge differences in core-ionization, and excitation energies computed at O-edge and C-edge of CH₂O

```
Total energy = -103.10187509 a.u. Excitation energy = 297.5084 eV.
R1^2 = 0.8061 R2 CCVV^2 = 0.0003 R2 COVV^2 = 0.1936 Res^2 = 1.68e^{-0}
Conv-d = yes
Amplitude
             Transitions between orbitals
-0.8978
              2 (A1) A
                                               infty
              Summary of significant orbitals:
              Number Type
                                        Irrep
                                                    Energy
                                        2 (A1)
                                                   -11.3549
                       Occ Alpha
CVS-EOMEE transition 1/A1
Total energy = -103.18974191 a.u. Excitation energy = 295.1175 eV.
R0^2 = 0.0000 R1^2 = 0.8313 R2_{CCVV^2} = 0.0004 R2_{COVV^2} = 0.0842 Res^2
Conv-d = yes
Amplitude
            Transitions between orbitals
            2 (A1) A
-0.6370
            2 (A1) B
 0.0868
        2 (A1) A
                                   -> 10 (A1) A
            2 (A1) B
                                       10 (A1) B
            Summary of significant orbitals:
             Number Type
                                    Irrep
                                              Energy
                    Occ Alpha
                                   2 (A1)
                                             -11.3549
                    Occ Beta
                                   2 (A1)
                                             -11.3549
                    Vir Alpha
                                   6 (A1)
                                               0.2380
                                              1.1323
                    Vir Alpha
                                  10 (A1)
                    Vir Beta
                                   6 (A1)
                                              0.2380
                                  10 (A1)
                                              1.1323
```

Start computing the transition properties

```
State A: ccsd: 0/A1
State B: cvs_eomee_ccsd/rhfref/singlets: 1/A1
Energy GAP = 10.8454 a.u. = 295.1176 eV

Transition dipole moment (a.u.):
    A->B: 0.045179 (X 0.000000, Y 0.000000, Z -0.045179)
    B->A: 0.045092 (X 0.000000, Y 0.000000, Z -0.045092)
Oscillator strength (a.u.): 0.014730

Transition angular momentum against gauge origin (a.u.):
    A->B: (X 0.000000i, Y 0.000000i, Z 0.000000i)
    B->A: (X 0.000000i, Y 0.000000i, Z 0.000000i)
Norm of one-particle transition density matrix:
    A->B: 0.904540; B->A: 0.889668
||gamma^AB||*||gamma^BA||: 0.804740
```