

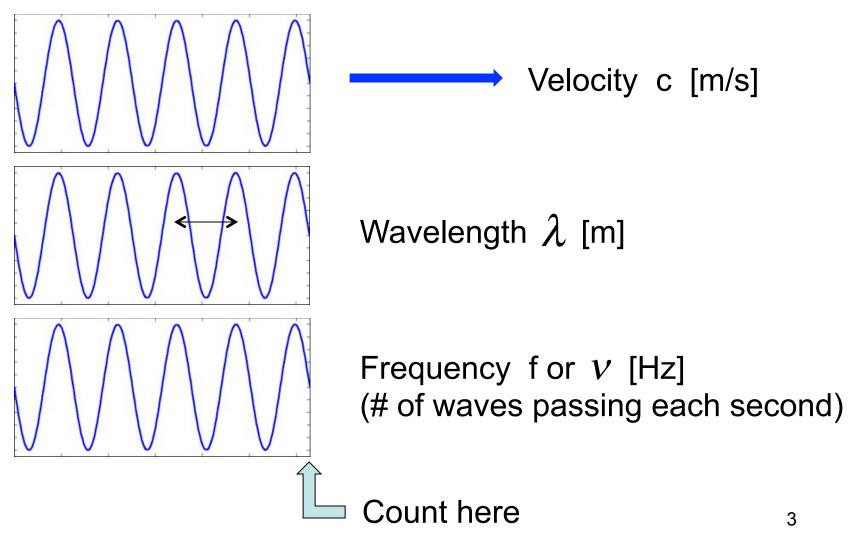
- Light and Distance [Ch 3.1-3.3]
- Temperature [Ch 3.4]
- Doppler Effect [Ch 3.5]
- Angular Measure [Ch 1.3, box p 11]

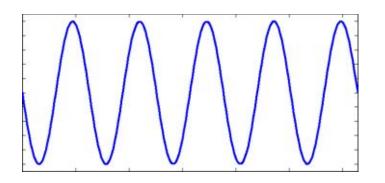
Waves





Three Wave Properties





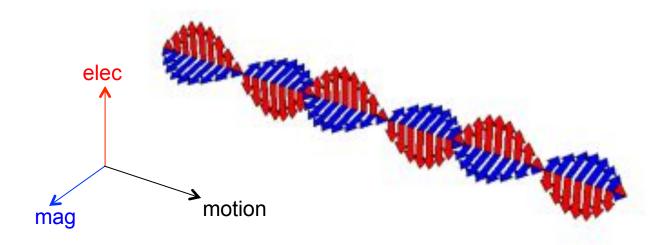
Three Wave Properties

• Not surprisingly, these 3 quantities are related: wavelength = wave speed divided by frequency, or $\lambda = c / f$

 In this course we will mainly use wavelength rather than frequency.

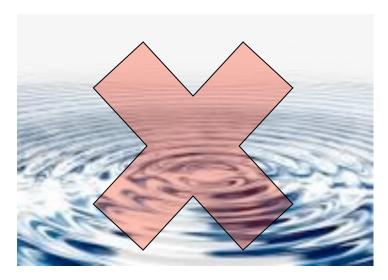
Light

- Light "acts like" a wave.
- The wave is a wave in electric field and magnetic field.



Light

- Water waves, sound waves, ... need a "medium" to propagate. But ...
- Light needs no medium propagates in a vacuum.



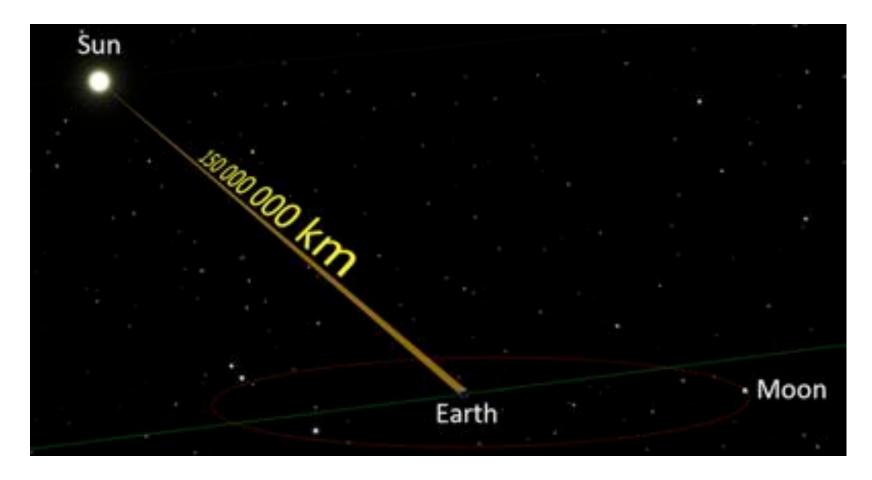
Speed of light, c



- c = 300,000 km per second!
- c = 3 x 10⁸ meters per second (actually 299,792,458 m/s)
- Nothing can travel faster than the speed of light

First "determined" by Ole Rømer 1656

Light distance



Light takes 8 min 19 sec to travel from the sun to the earth. Or, sun is 8.3 light minutes in distance.

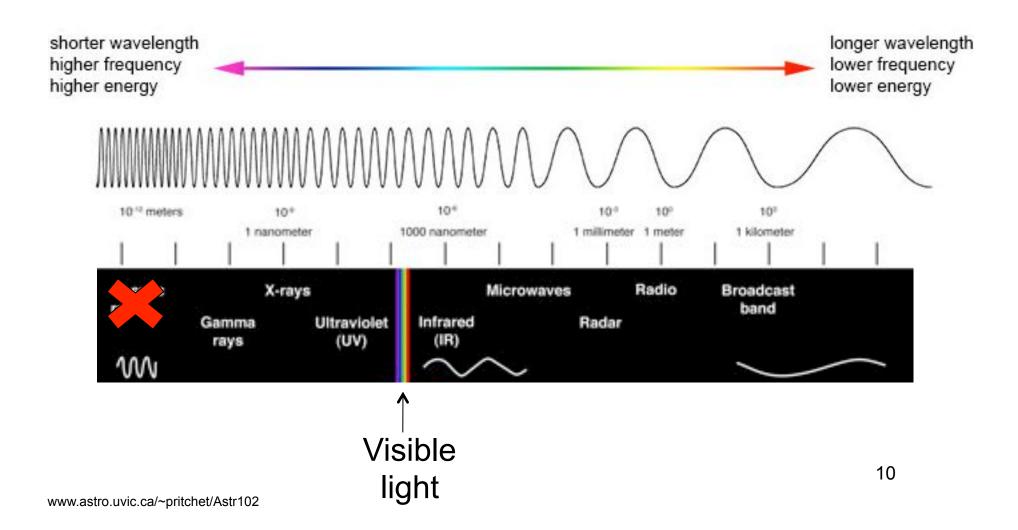
Light distance

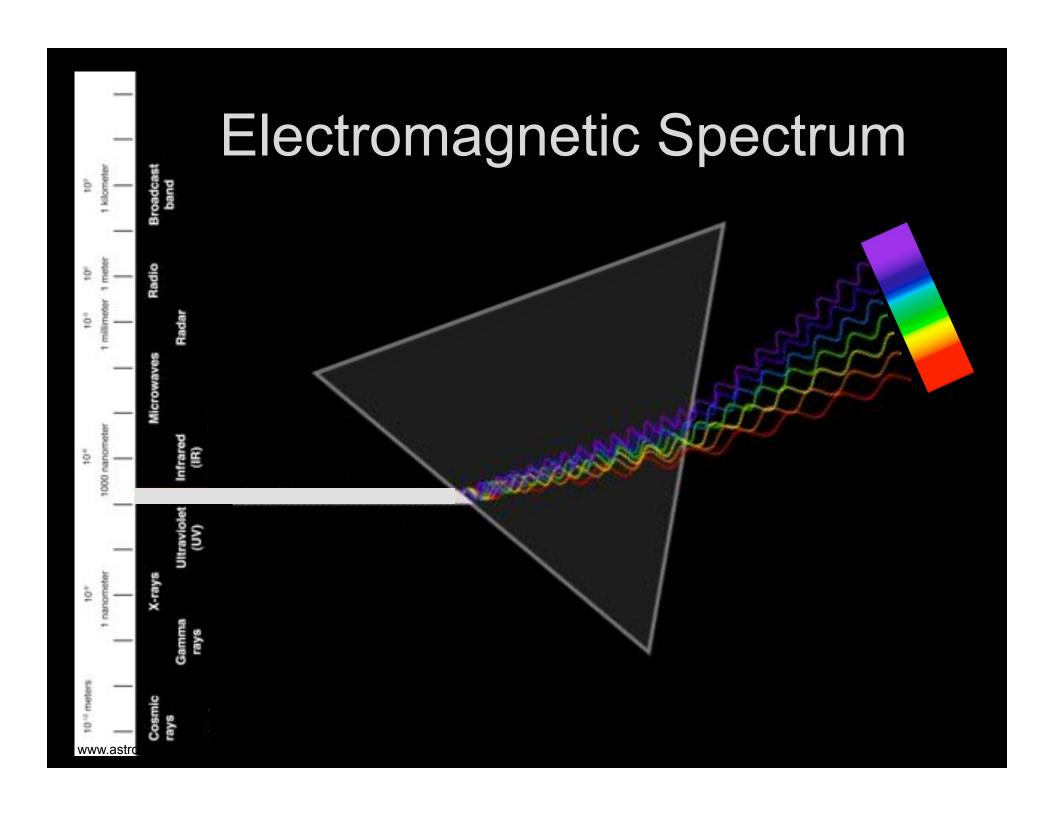
- One light year = distance travelled by light in one year = about 10¹³ km
 - Nearest star is 4.2 light years distant

- Alternate: one parsec = 3.26 light years
 - Nearest star is 1.4 pc distant

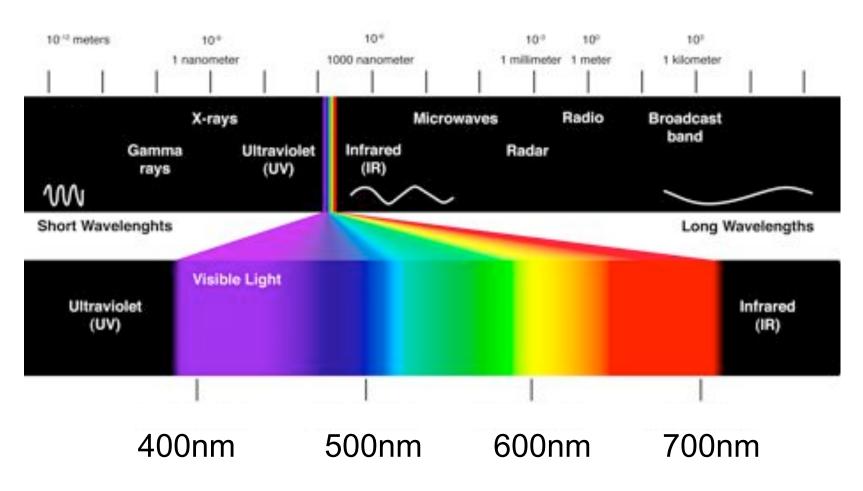
The Electromagnetic Spectrum

"The Spectrum of Light"





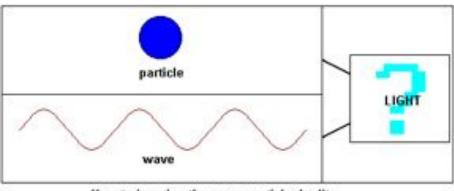
Electromagnetic Spectrum



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Nature of Light

- Light acts both as a wave and a particle!
- A particle of light is called a "photon"



How to imagine the wave-particle duality.

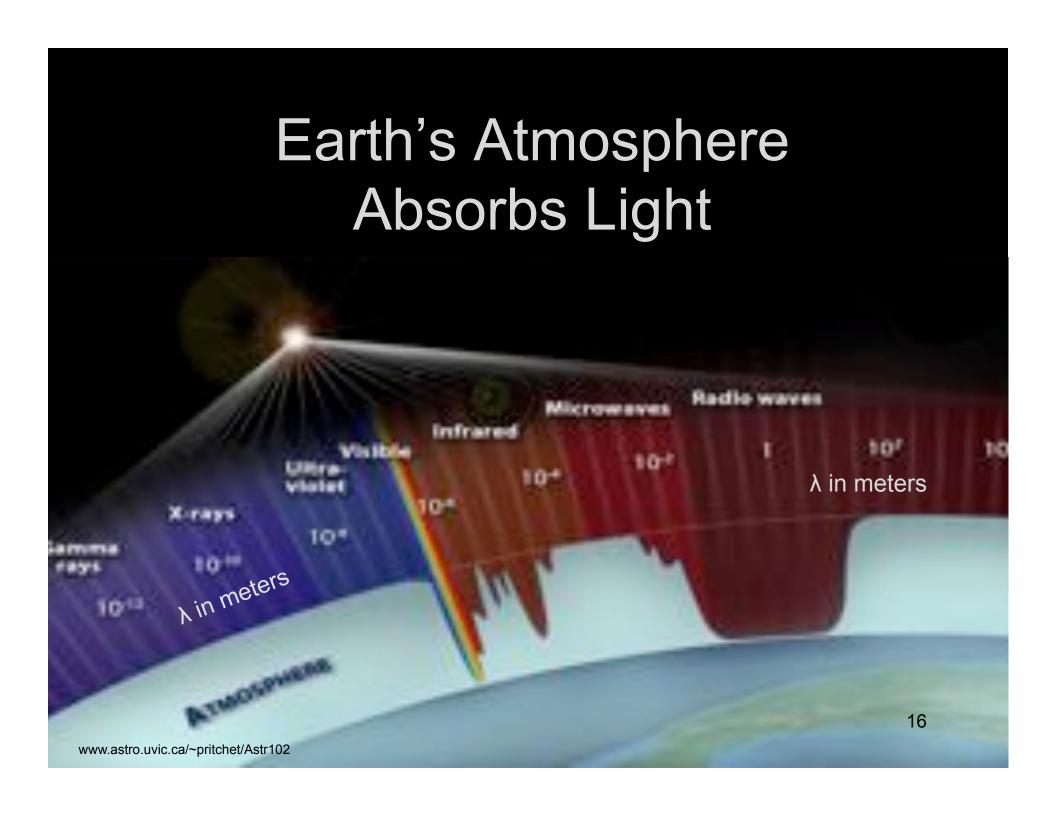
Nature of Light



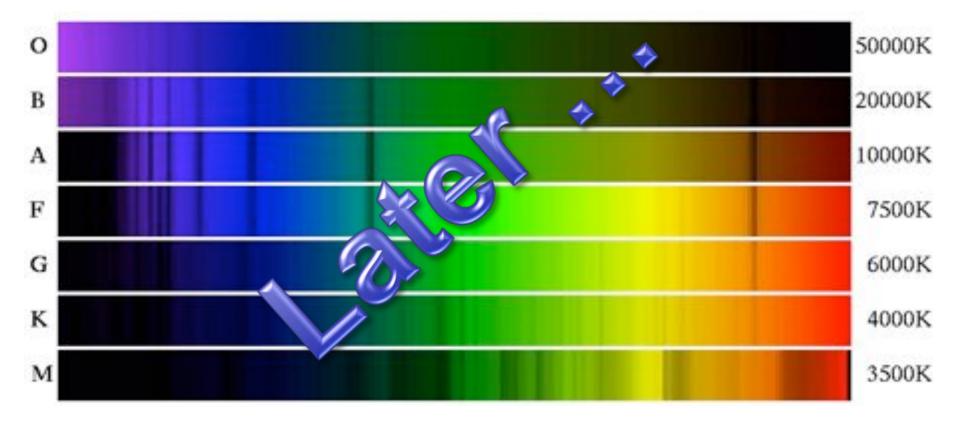
"Once and for all I want to know what I'm paying for. When the electric company tells me whether light is a wave or a particle I'll write my cheque."

Nature of Light

- Photons carry energy E=hf, where h is Planck's constant, 6.6 x 10⁻³⁴ J s
- Smaller wavelength means higher energy

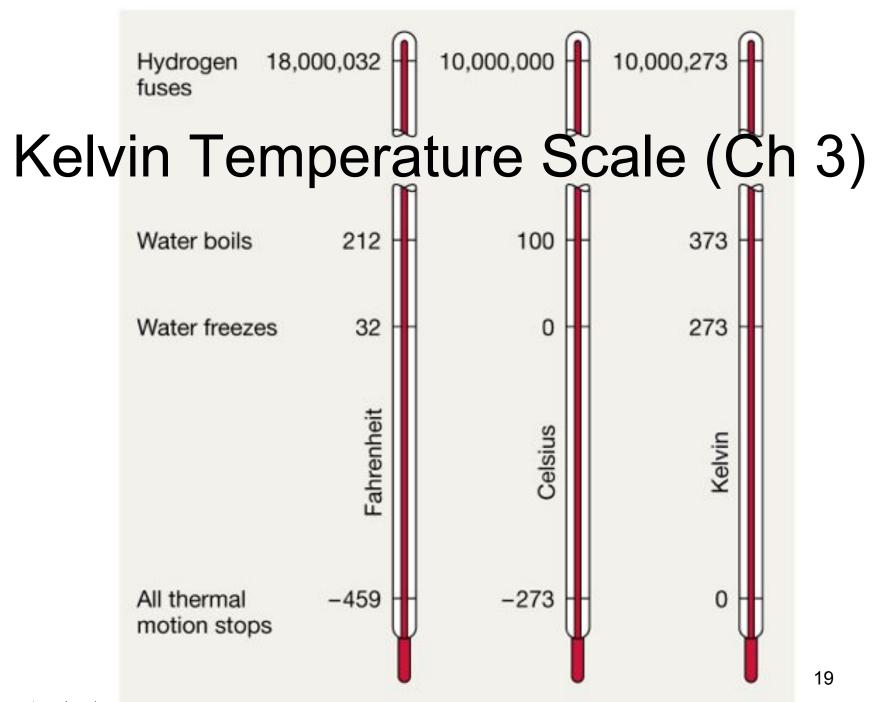


Two more things about light



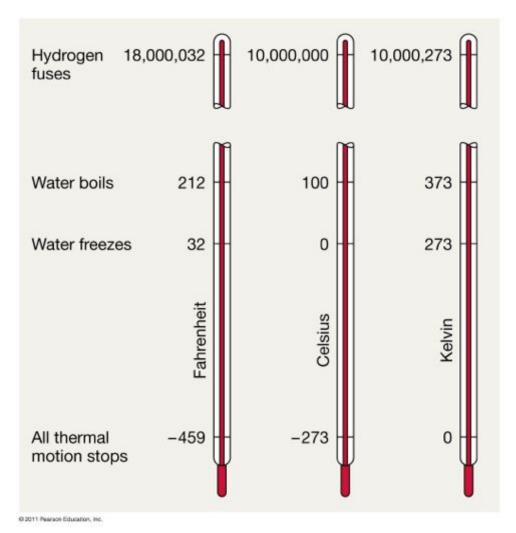
- 1. black body radiation hot=blue, cool=red
- 2. atomic energy levels dark lines

Dark lines in spectrum Hydrogen are due to absorption by particular elements Wavelength tells you which element way of deriving Magnesium abundances of elements in stars Can also have bright lines!



Kelvin Temperature Scale (Ch 3)

- The hotter a body, the faster its atoms are moving!
- All thermal motion ceases at 0 K.
 "Absolute Zero".
- Water freezes at 273 K and boils at 373 K



The Doppler Effect

http://www.youtube.com/watch?v=8WgSQIRymwE



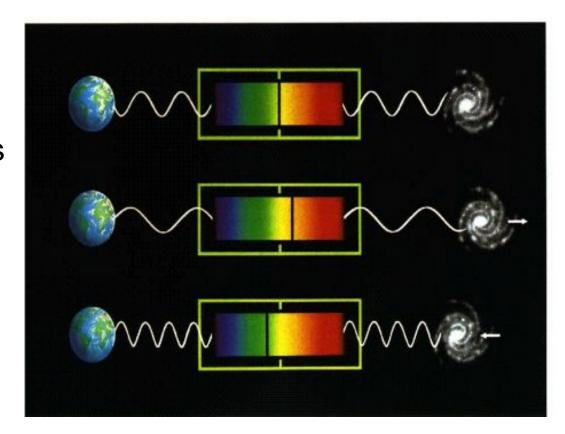
http://www.youtube.com/watch?v=GsCaSX6E-6U



Doppler Effect [Ch 3.5]

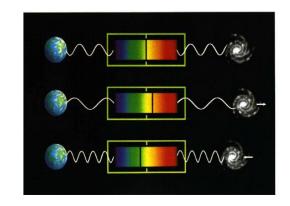
Christian Doppler 1835

- Applies to any wave
- Doppler proposed the effect to explain colour differences in binary stars [not correct]
- Buys Ballot tested the Doppler effect for sound waves in 1845 by using a group of musicians playing a calibrated note on a train in the Utrecht-Amsterdam line.





Doppler Effect



$$\frac{\Delta\lambda}{\lambda} = \frac{v}{c}$$

v= velocity [- means approach]
c= speed of wave [300,000km/s
 for light]

$$\frac{\text{change in wavelength}}{\text{true wavelength}} = \frac{\text{recession velocity}}{\text{wave speed}}$$

 If velocity is away from us, change in wavelength is positive, or to the red.

Measuring Velocities with the Doppler Effect

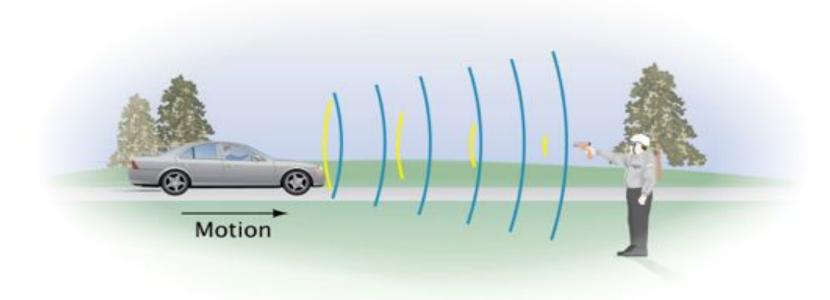
Example: For a speed of 30 km/s, the Doppler shift is given by

$$\frac{\text{change in wavelength}}{\text{true wavelength}} = \frac{\text{recession velocity}}{\text{wave speed}}$$

$$=\frac{30 \text{ km/s}}{300,000 \text{ km/s}} = 0.01 \text{ percent.}$$

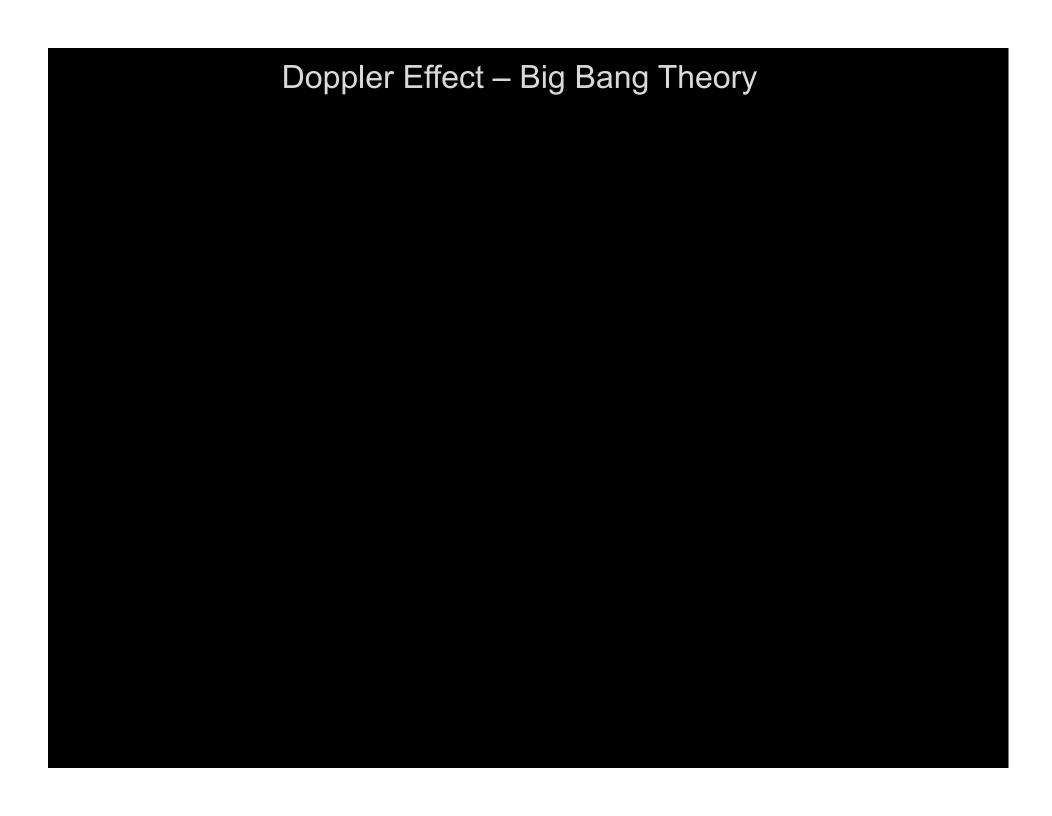
Measuring Velocities with the Doppler Effect

This may seem small, but it is easily detectable with a radar gun!



For the musicians (don't have to know)

- Car travels at 100 km/hr (assume)
- Pitch of trumpet is 440 Hz (A) (assume)
- Speed of sound is 1200 km/hr
- Fractional change in wavelength or frequency is then v/c=0.08=8% which is about 1 semitone – which is what was observed!





Angular measure [AT Ch 1]

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1 arcminute = 60 arcseconds
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1 degree = 60 arcminutes = 3600 arcseconds

360 degrees = a full circle

$$360^{\circ}$$
 = circle