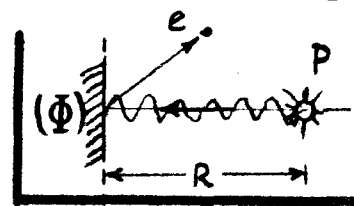


506 Problems

48(2)

- ④ A lightbulb, radiating total power $P=10\text{W}$, is placed at distance $R=1\text{m}$ from a large metal plate. The metal has a work function $\Phi=3\text{eV}$. An electron in the metal can absorb the incident radiation over an area adjacent to it -- suppose that area is a circle of radius $r \sim$ several atomic radii (say 5). Treat this problem classically.



- (A) Calculate the time Δt required for the electron to absorb enough energy to be ejected from the surface. It is observed that $\Delta t < 1\text{ns}$. Comment.
- (B) With all else the same, what lightbulb power is needed to eject the electron in 1ns ? Compare this with the output of a typical commercial power plant.

- ⑤ (A) Show that a photon cannot transfer all of its energy to a free electron.
- (B) Suppose a photon, incident on a free electron, has energy $E_i = Nmc^2$, where m is the electron rest mass and N is a numerical factor. What value of N is required for the photon to transfer 99% of its energy? What is λ for this photon?

- ⑥ Apply the Correspondence Principle to radiation from a transition $n \rightarrow n-1$ in a hydrogenlike Bohr atom. The average power radiated is: $P_n = \frac{1}{\tau_n} \Delta E(n \rightarrow n-1)$, $\Delta E =$ energy emitted, and $\tau_n =$ "lifetime" of state n . As $n \rightarrow$ large, P_n must become the classical Larmor radiation rate: $P_n = (2e^2/3c^3) |\ddot{a}_n|^2$, for a charge e undergoing (centripetal) acceleration \ddot{a}_n . Assume circular orbits, and define the dimensionless constant: $\alpha = e^2/\hbar c \approx 1/137 \dots$ α is the "fine-structure" const.

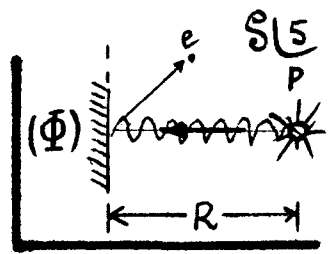
- (A) Let $\Gamma_n = 1/\tau_n$ be the "transition probability / unit time" for $n \rightarrow n-1$. Show that: $\Gamma_n = \Gamma_1/n^5$, where: $\Gamma_1 = \frac{2}{3} Z^4 \alpha^5 \cdot (mc^2/\hbar)$.
- (B) For hydrogen ($Z=1$), compare the result in part (A) for the n^{th} state lifetime $\tau_n = 1/\Gamma_n$ with known values: $\tau(n=2 \rightarrow 1) = 1.6\text{ns}$, $\tau(n=4 \rightarrow 3) = 73\text{ns}$, and $\tau(n=6 \rightarrow 5) = 610\text{ns}$. Comment on the trend of the comparisons.

§ 506 Solutions

④ Time & power scales for the classical photo-electric effect.

(A) The radiation incident on the plate delivers a flux of size:

$$\rightarrow \frac{\text{energy/time}}{\text{area}} = P/4\pi R^2. \quad (1)$$



If a given electron gathers energy from an area = circle of radius na , a = atomic radius and n = numerical factor, it absorbs energy at a rate...

$$\rightarrow \frac{\text{energy}}{\text{time}} = (P/4\pi R^2) \cdot \pi(na)^2 = P \cdot (na/2R)^2. \quad (2)$$

In order to be ejected in time Δt , the electron must gather an energy exceeding the work function barrier, i.e.

$$\rightarrow \left(\frac{\text{energy}}{\text{time}}\right) \Delta t = P \Delta t \cdot (na/2R)^2 \geq \Phi \Rightarrow \Delta t \geq \frac{\Phi}{P} \cdot (2R/na)^2. \quad (3)$$

For the given numbers...

$$\left\{ \begin{array}{l} \Phi = 3\text{eV} = 4.81 \times 10^{-19} \text{ J}, \quad P = 10\text{W} = 10 \text{ J/sec} \\ R = 1\text{m} = 100\text{cm}; \quad a = 0.53 \times 10^{-8} \text{ cm (Bohr radius)} \quad \& \quad n = 5 \dots \\ \underline{\underline{\Delta t = \frac{4.81 \times 10^{-19}}{10} (2 \times 100 / 5 \times 0.53 \times 10^{-8})^2 = 2.74 \text{ sec}}} \end{array} \right. \quad (4)$$

The observed $\Delta t < 1\text{ns}$. Clearly the classical picture of the electron continuously absorbing the radiant energy is monumentally WRONG. Enter Einstein, center stage: the "absorption" is actually a collision.

(B) If $\Delta t = 2.74\text{sec}$ is to be decreased to 1ns , then -- per Eq.(3) -- the lightbulb power P must be increased by a factor $2.74/10^{-9}$. This gives a required power...

$$\rightarrow \text{for } \Delta t = 1\text{ns} : \underline{\underline{P = 2.74 \times 10^9 \times 10\text{W} = 27,400 \text{ MW}}}. \quad (5)$$

A typical commercial power plant puts out $\sim 100\text{-}300\text{MW}$. So the power required in Eq.(5) is equivalent to the output of ~ 100 such plants.

506 Solutions

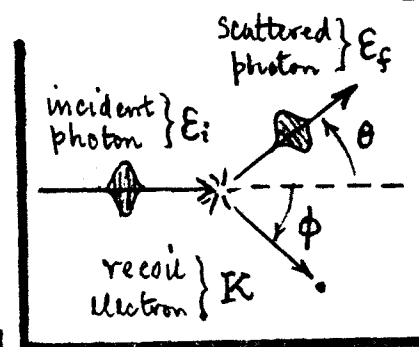
S16

⑤ Most energetic photon → electron collision.

(A) Ref. CLASS NOTES, p. Intro. 12, Eqs. (26) - (27). There,

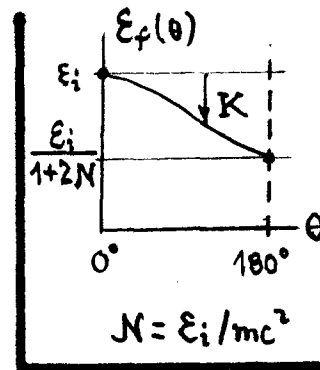
for a photon → free electron collision, we have written energy & momentum conservation eqns, and have found the final photon frequency. Since energy $E = h\nu$ for the photon, Eq. (27) is:

$$\underline{E_f = E_i / [1 + (E_i/mc^2)(1 - \cos\theta)]}, \quad \theta = \text{scattering \& for photon.} \quad (1)$$



The photon transfers a recoil (kinetic) energy K to the e , of an amount ...

$$\rightarrow K = E_i - E_f = E_i \left[\frac{N(1 - \cos\theta)}{1 + N(1 - \cos\theta)} \right] < E_i, \quad N = \frac{E_i}{mc^2}. \quad (2)$$



K increases monotonically with photon scattering θ , reaching its max. value @ $\theta = 180^\circ$ (photon backscattering), whence

$$\rightarrow K_{\text{MAX}} = K(\theta = 180^\circ) = E_i \left[\frac{2N}{1 + 2N} \right] < E_i. \quad (3)$$

For any finite photon energy N , $K_{\text{MAX}} < E_i$, and -- with this max. energy transfer -- the photon still has energy $E_f = E_i / (1 + 2N)$. So, indeed, the photon cannot transfer all its energy to the free e ... that transaction would violate conservation of momentum.

(B) If the transfer $E_i \rightarrow K$ is to be 99% complete, then the least value of N necessary is -- from Eq. (3)...

$$\rightarrow K_{\text{MAX}} = E_i \left[\frac{2N}{1 + 2N} \right] = 0.99 E_i \Rightarrow \underline{N = \frac{1}{2} \left(\frac{1}{0.99} - 1 \right)} \Big|_{\epsilon=0.01} = \underline{49.5}. \quad (4)$$

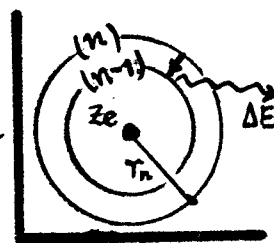
This is a very robust photon, with incident energy $E_i = Nmc^2 = 25.3 \text{ MeV}$. Its wavelength is...

$$\underline{\underline{\lambda = hc/E_i = \frac{1}{N}(h/mc) = \frac{1}{49.5} \times 2.43 \times 10^{-10} \text{ cm} = 4.91 \times 10^{-4} \text{ \AA}}} \quad (5)$$

⑥ II-atom transition rates per the Correspondence Principle.

(A) $P_n = \Gamma_n \Delta E(n \rightarrow n-1) \rightarrow \frac{2}{3}(e^2/c^3)|a_n|^2$, so we want to calculate

$$\left[\Gamma_n = \frac{2}{3}(e^2/c^3)|a_n|^2 / \Delta E(n \rightarrow n-1) \right] \quad \begin{array}{l} \checkmark a_n = \text{accel}^n \text{ in } n^{\text{th}} \text{ orbit,} \\ \Delta E = n \rightarrow n-1 \text{ transition energy.} \end{array} \quad (1)$$



Assume circular orbits of radii r_n . Bohr model gives [Eq. (16), p. Duality 7]:

$$\left[\begin{array}{l} \text{orbit} \\ \text{energy} \end{array} \right] E_n = -\frac{1}{2}(Z\alpha)^2 mc^2/n^2 \Rightarrow \underline{\Delta E(n \rightarrow n-1)} = E_n - E_{n-1} \underset{n \gg 1}{\approx} \frac{(Z\alpha)^2 mc^2}{n^3}; \quad (2)$$

$$\left[\begin{array}{l} \text{orbit} \\ \text{radius} \end{array} \right] r_n = \frac{n^2 a_0}{Z}, \quad \checkmark a_0 = \frac{\hbar^2}{me^2} = \text{Bohr radius. NOTE: } \alpha = e^2/\hbar c. \quad (3)$$

The electron's centripetal accelⁿ a_n is found from: $ma_n = -\frac{Ze^2}{r_n^2}$, i.e.

$$\rightarrow a_n = -\frac{1}{m} Ze^2/r_n^2 = -Z^3 e^6 m/n^4 \hbar^4, \quad \checkmark a_n = -Z^3 e^2 \alpha^2 mc^2/n^4 \hbar^2. \quad (4)$$

$$\text{so } \text{Larmor power: } P_n = \frac{2}{3}(e^2/c^3)|a_n|^2 = \frac{2}{3} Z^6 \alpha^2 (mc^2)^2/n^8 \hbar. \quad (5)$$

Then Eq. (1) yields...

$$\rightarrow \Gamma_n = P_n[\text{Eq. (5)}] / \Delta E[\text{Eq. (2)}] = \frac{2}{3} Z^4 \alpha^5 mc^2/n^5 \hbar,$$

$$\checkmark \Gamma_n = \Gamma_1/n^5, \quad \checkmark \Gamma_1 = \frac{2}{3} Z^4 \alpha^5 mc^2/\hbar. \quad \underline{\text{QED}} \quad (6)$$

(B) The lifetime for $n \rightarrow n-1$ is: $\tau_n = 1/\Gamma_n = n^5 \tau_1$, where: (7)

$$\tau_1 = 1/\Gamma_1 = \frac{1}{Z^4} \cdot \left(\frac{3}{2} \hbar / \alpha^5 mc^2 \right) \checkmark \alpha = e^2/\hbar c \approx 1/137, \quad \hbar/mc^2 = 1.29 \times 10^{-21} \text{ sec.}$$

$$\text{so numerically: } \underline{\tau_1 = \frac{1}{Z^4} \cdot 0.0932 \text{ ns.}} \quad (8)$$

For $Z=1$, we calculate the #s at right by use of $\tau_n = n^5 \tau_1$ in Eq. (7). The agreement with known values improves as n increases; this is in accord with

TRANSITION	THIS CALCN $\tau_n, \text{ ns}$	KNOWN $\tau_n, \text{ ns}$	ratio
$n=2 \rightarrow 1$	3.0	1.6	1.88
$n=4 \rightarrow 3$	95	73	1.30
$n=6 \rightarrow 5$	725	610	1.19

the Correspondence Principle feature that the classical result becomes more accurate as the Quantum system becomes larger.