

Lecture 1 – Atomic Structure

Lecture 2 – The Ultraviolet Catastrophe

Lecture 3 – Particle Nature of Light

Lecture 4 – Atomic Energy Levels and Spectra

Lecture 5 – X-ray Production and Diffraction

Lecture 6 – X-ray Spectra

Lecture 7 – Matter Waves

Lecture 8 – Wave-Particle Duality

Lecture 9 – Wave functions for Quantum Particles

Lecture 10 – A Quantum Mechanical Wave Equation

Lecture 11 – Applications of Schrödinger's Equation



Recap of Lecture 6

- Spectra produced by X-ray sources
 - Brehmsstralung
 - \circ Discrete lines (K_{α} and friends)
- Absorption exponential inside a material plus absorption edges when we hit high enough energy to ionise electrons from a shell

In this lecture

- Particles acting as waves de Broglie wavelength
- Proof Davisson & Germer experiment

de Broglie Wavelength

Louis de Broglie (1892-1987)

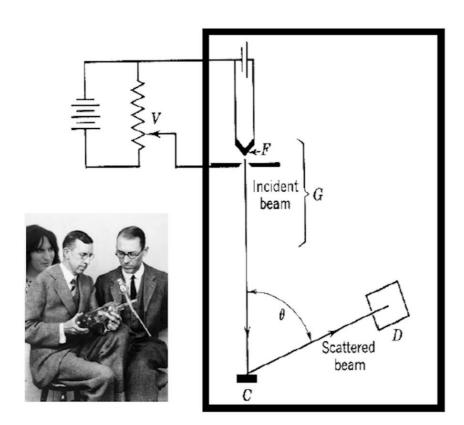


In 1924, Louis de Broglie argued that classical particles should also have wave-like properties – in his PhD thesis. He was awarded a Nobel Prize for his hypothesis in 1929

Davisson & Germer experiment

In 1925 Elsasser predicted that electrons would be diffracted by crystalline materials

- This was observed experimentally in 1927 by Davisson & Germer, and independently by G.P. Thomson
- de Broglie's hypothesis appeared in his PhD thesis in 1924. He was awarded a Nobel Prize for his hypothesis in 1929
- Davisson and Thomson shared a Nobel Prize for their experimental work in 1937



Davisson & Germer experiment - Results

NATURE

Letters to the Editor.

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The Scattering of Electrons by a Single Crystal of Nickel.

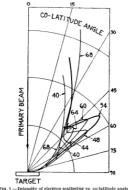
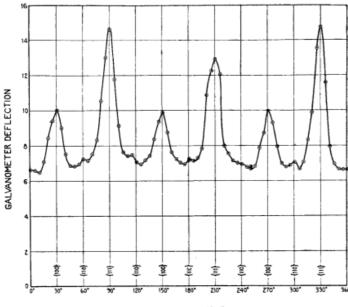


Fig. 1.—Intensity of electron scattering vs. co-latitude angle for various bombarding voltages—azimuth-{111}-330°.

target. There are six such azimuths, and any one of these will be referred to as a [10]-azimuth. If follows from considerations of symmetry that if the intensity of scattering exhibits a dependence upon azimuth as we pass from a [100]-azimuth to the next adjacent (111]-azimuth (60°), the same dependence must be exhibited in the reverse order as we continue on through 60° to the next following [100]-azimuth. Dependence on azimuth must be an ever function of the continue on through 60° to the next following [100]-azimuth. Dependence on azimuth must be an ever function of the continue of the target to a highest value in co-latitude 20°, the limit of observations. If bombarding potential in azimuth, a variation in the intensity of scattering increases continuously and regularly from zero in the plane of the target to a highest value in co-latitude 20°, the limit of observations. If bombarding potentials in general this variation is slight, amounting in some cases to not more than a few per cent. of the average intensity. This is the nature of the scattering for bombarding potentials in the range from 15 voils to Co-latitude curve for azimuth-[111]. This hump develops rapidly with increasing voltage into a strong spur, at the same time moving slowly upward toward the incident bearn. It attains a maximum intensity in co-latitude 50° for a bombarding potential of 54 volts, then decreases in intensity, and disappears of the continuence of the

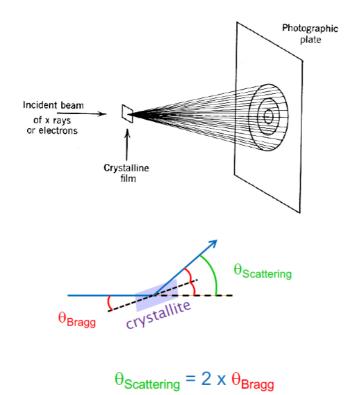
Diffraction peaks from a crystal, just like we found from X-rays:



AZIMUTH ANGLE

Fig. 2.—Intensity of electron scattering vs. azimuth angle—54 volts, co-latitude 50°.

GP Thomson experiment – powder electron diffraction



G.P. Thomson was the son of J.J. Thomson – his father discovered the electron, and he established its wave behaviour – they both won Nobel Prizes (in 1906 and 1937)

Can use the same target to diffract both X-rays and electrons:

