

## Formula Sheet

### The first law of thermodynamics

$$dU = \delta Q - p dV$$

$$\Delta U = Q_{\text{in}} + W_{\text{on}}$$

### Ideal gas equation of state

$$pV = nRT$$

### Ideal gas adiabatic process

$$pV^\gamma = \text{constant}, \text{ where } \gamma = C_p/C_v.$$

### Heat Transfer

$$\text{Rate of heat flow by conduction} \quad \dot{Q} = -\kappa A \frac{\partial T}{\partial x}$$

$$\text{Stefan-Boltzmann} \quad \dot{Q} = \sigma e A T^4$$

### Linear coefficient of thermal expansion, $\alpha$

$$\ell(T) = \ell(T_0)[1 + \alpha(T - T_0)].$$

### Gamma function and Stirling's approximation

$$\Gamma(N + 1) = N! = \int_0^\infty dx x^N e^{-x} \quad \text{and} \quad N! \approx \left(\frac{N}{e}\right)^N \Leftrightarrow \ln N! \approx N(\ln N - 1).$$

### Gaussian integral

$$\int_{-\infty}^{\infty} e^{-\alpha x^2} dx = \sqrt{\frac{\pi}{\alpha}}.$$

### Normalised one-dimensional Maxwell-Boltzmann distribution

$$p_{1d}(v) = \sqrt{\frac{m}{2\pi k_B T}} \exp\left(-\frac{mv^2}{2k_B T}\right).$$