

Quantum Mechanics 1 – Solution 5

- a) The wavelength of the X-rays is obtained from $E = \frac{hc}{\lambda}$.

$$\lambda = \frac{hc}{E} = \frac{4.14 \times 10^{-15} \times 3 \times 10^8}{3.55 \times 10^3} = 3.50 \times 10^{-10} \text{ m} = 350 \text{ pm}.$$

Note that I have used the value of Planck's constant in units of eV s, therefore the energy of the X-ray is given in eV. [1 mark]

To find the distance between the crystal planes, use the Bragg equation and, for the first interference maximum, set $n = 1$:

$$2d \sin \theta = n\lambda$$

[1 mark]

$$d = \frac{n\lambda}{2 \sin \theta} = \frac{1 \times 350}{2 \times \sin(18^\circ)} = 566 \text{ pm}.$$

[2 marks]

- b) To calculate the other angles, use

$$\sin \theta = \frac{n\lambda}{2d} = n \times \frac{350 \text{ pm}}{2 \times 566 \text{ pm}} = n \times 0.309$$

[1 mark]

Hence:

$n = 2$ gives $\sin \theta = 0.618$ and so $\theta = 0.667$ radians or 38.2°

$n = 3$ gives $\sin \theta = 0.928$ and so $\theta = 1.19$ radians or 68.1°

Values of $n > 3$ give values of $\sin \theta$ greater than unity. Hence there may only be two further interference maxima. Condition is that $\sin \theta \leq 1$.

[3 marks]

- c) Using the same formula, we now require that $\sin \theta \leq 1$ for $n = 2$. This then gives the condition

$$\sin \theta = \frac{n\lambda}{2d} \leq 1$$

$$\frac{2\lambda}{2d} = \frac{\lambda}{d} \leq 1$$

$$\lambda \leq 566 \text{ pm}$$

[2 marks]