

# Recap from last time

“Any substance in equilibrium at a temperature  $T$  has an average energy of  $\frac{k_B T}{2}$  for each **degree of freedom**”

**A degree of freedom** is defined as any term in the expression for the energy of the system which contains a squared position or velocity component...

For our ideal gas in 1D, we have  $E(v_x) = \frac{1}{2} m v_x^2 = \frac{1}{2} k_B T$   $C_V = \frac{1}{2} R$

In 3D:  $E(v_x, v_y, v_z) = \frac{1}{2} m v_x^2 + \frac{1}{2} m v_y^2 + \frac{1}{2} m v_z^2 = \frac{3}{2} k_B T$   $C_V = \frac{3}{2} R$

No gravity ( $mgh$ ) term because the position term is linear

# Recap from last time

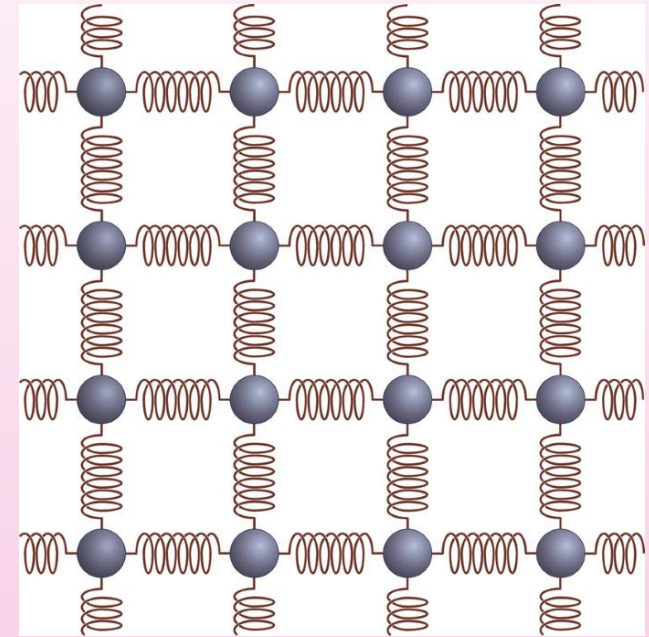
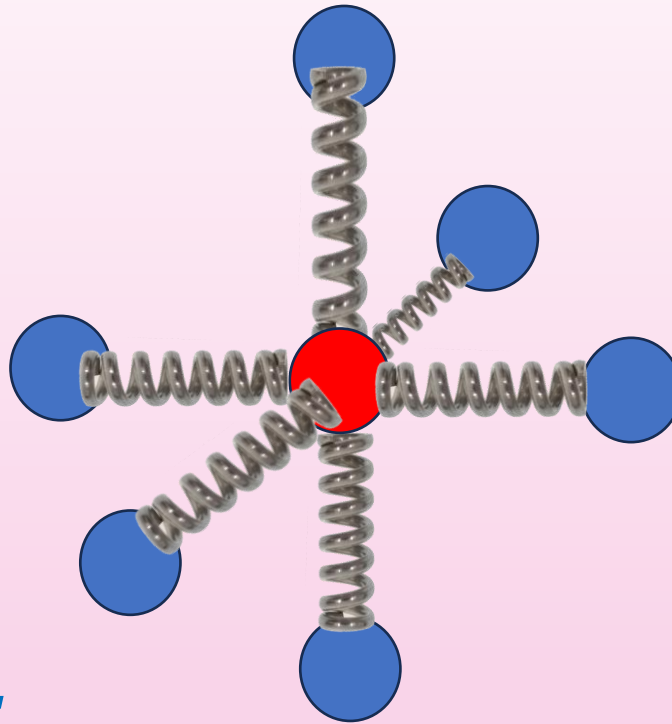
Each **atom** (mass  $m$ ) is fixed, connected to its neighbours by interatomic forces (which we can treat like springs with spring constant  $K$ )

Each spring has **vibrational energy**

$$E = \frac{1}{2}mv_x^2 + \frac{1}{2}Kx^2 = k_B T$$

Each atom connected to 6 springs (but each spring is shared by 2 atoms...)

$$E = 3k_B T \Rightarrow c_V = 3R$$



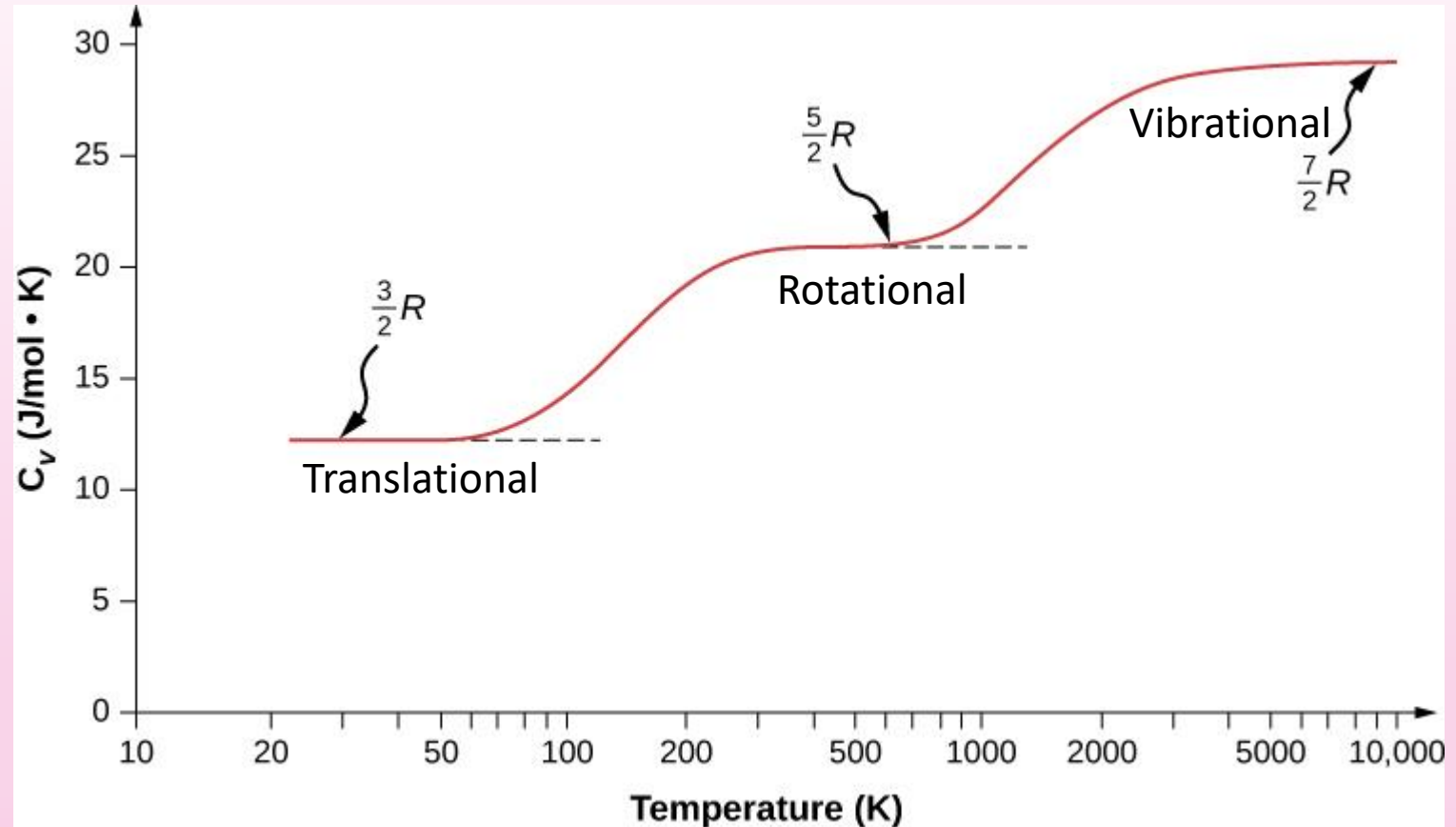
[https://media.springernature.com/lw685/springer-static/image/chp%3A10.1007%2F978-3-031-18286-0\\_4/MediaObjects/318291\\_2\\_En\\_4\\_Fig5\\_HTML.png](https://media.springernature.com/lw685/springer-static/image/chp%3A10.1007%2F978-3-031-18286-0_4/MediaObjects/318291_2_En_4_Fig5_HTML.png)

# Recap from last time

Degrees of freedom can be considered fully “excited” if

$$k_B T \gg E_1 - E_0$$

If  $k_B T \ll E_1 - E_0$ , the degree of freedom is said to be “frozen out”



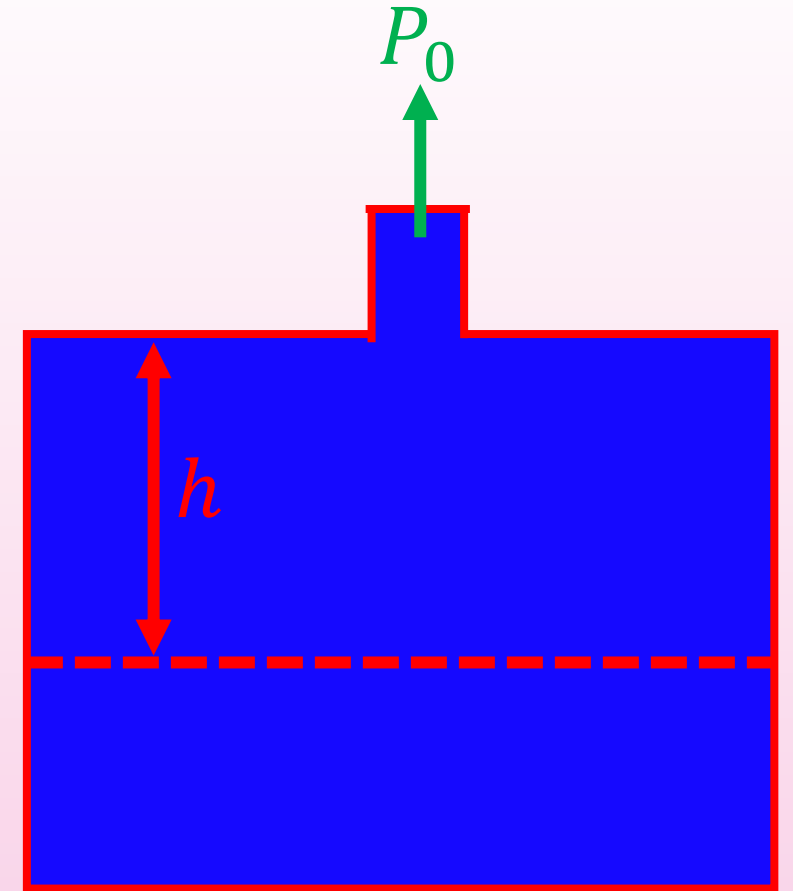
# Fluid dynamics

Say I have no applied force, and the pressure at the surface of the liquid is given by  $P_0$ ... what is the pressure,  $P$ , at some arbitrary depth  $h$ ?

$$P = P_0 + \rho gh$$

Needs units of pressure, so force/area  
 $[\text{kg}][\text{m}]^{-3} [\text{m}][\text{s}]^{-2} [\text{m}] \rightarrow [\text{kg}][\text{m}][\text{s}]^{-2}/[\text{m}]^2$

Only true for a fluid of constant density  $\rho$ !

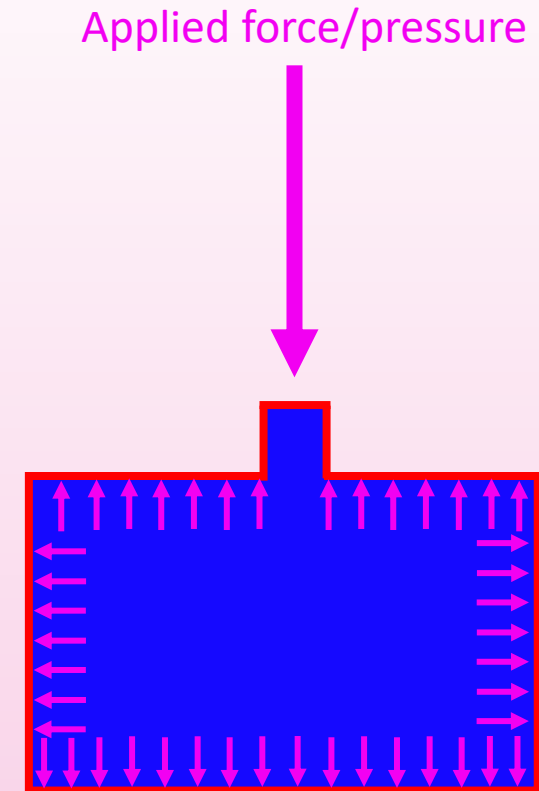


# Fluid dynamics

If I apply a force to the liquid shown in this container here, what will happen?

Pressure in the liquid will increase, so applied force to the walls of the container increases!

Pascal's law: If pressure is applied to an enclosed fluid, the pressure is applied equally (and undiminished) to every part of the fluid and the walls of enclosure.



# Hydraulic presses

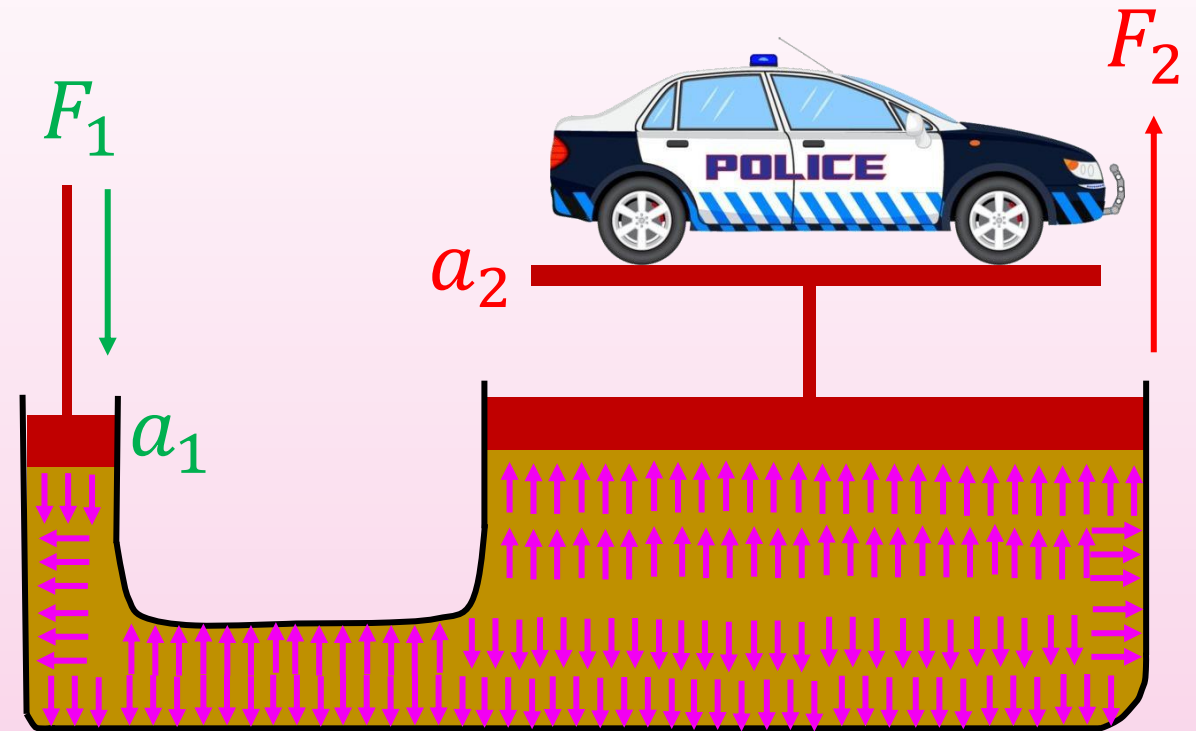
If a force  $F_1$  is applied to a liquid through some relatively small area,  $a_1$ , then the pressure determined by these quantities must be applied at all points to the liquid

$$P = F_1 / a_1$$

Therefore

$$P = F_2 / a_2$$

$$F_2 = \frac{F_1 a_2}{a_1}$$



Q: incompressible fluid, with cross-sectional area  $a_1$  of  $0.25 \text{ m}^2$ . On the other side, the cross sectional area is  $2.5 \text{ m}^2$ . What force is required to lift a 1 tonne car?

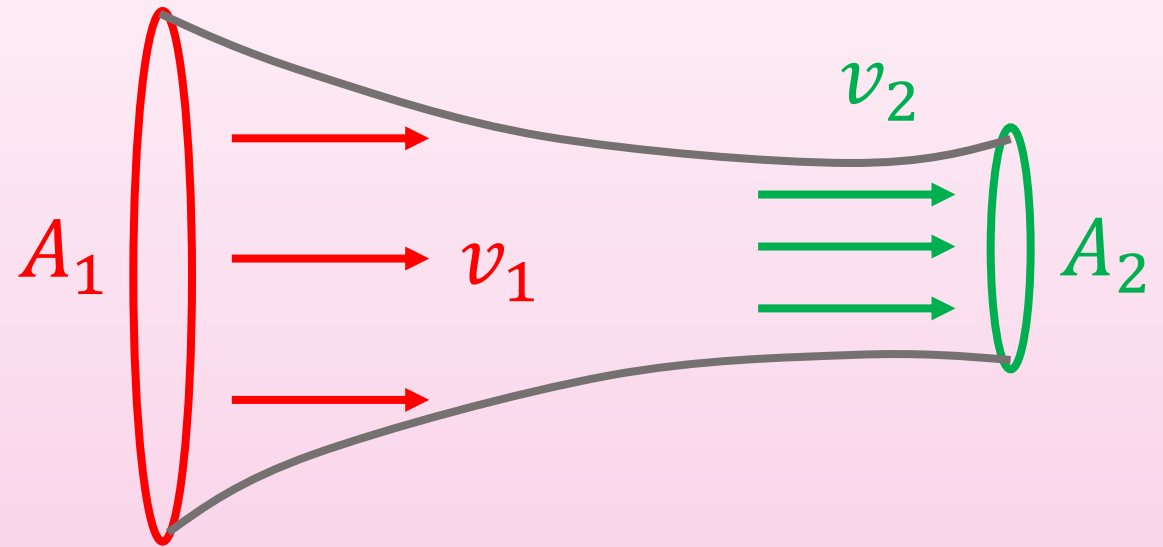
A: 980 N (100 kg of mass equivalent)

# Fluid flowing through a pipe

In the steady state (unchanged with respect to time), the amount of water that passes a given point should always be the same at any time...

Speed of the flow depends on the area of the pipe, such that

$$A_1 v_1 = A_2 v_2$$



For a change in speed, we require an acceleration and hence a force  
This force comes from the neighbouring water, but how does this affect pressure?

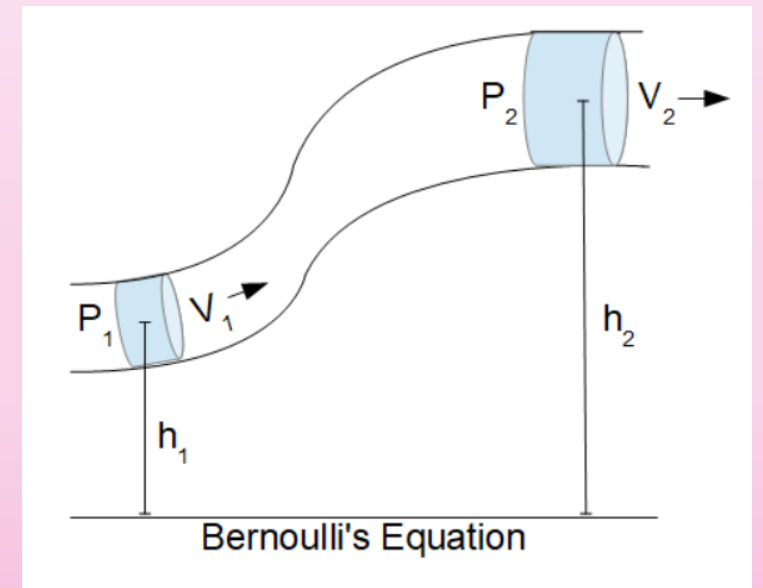
# Bernoulli's equation

Before, we had  $P = P_0 + \rho gh$

To include the velocity, we need the pressure from the kinetic energy  
(think back to our gravitational pressure term)

$$\text{Constant} = P + \rho gh + \frac{1}{2} \rho v^2$$

↑ Static pressure    ↑ Hydrostatic pressure    ↑ Dynamic pressure





# Bernoulli's equation example

$$\text{Constant} = P + \rho gh + \frac{1}{2}\rho v^2$$

Water enters the pipe at a speed of  $1.5 \text{ ms}^{-1}$  and pressure  $4 \times 10^5 \text{ Pa}$  through a pipe of diameter 2 cm. It then travels through a pipe of diameter 1 cm to a tap in a bathroom 5 metres above.

What speed does the water flow out when the tap is opened?  
What is the pressure at the tap?

