

## **Cambridge International Examinations**

Cambridge International General Certificate of Secondary Education

## **CO-ORDINATED SCIENCES**

0654/61

Paper 6 Alternative to Practical

May/June 2016

MARK SCHEME
Maximum Mark: 60

## **Published**

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Pa	age 2		Syllabus	Paper		
		Cambridge IGCSE – May/June 2016	0654	61		
1	(a)	time/minutes; volume/cm³;		[2]		
	(b)	6.8; 0.5;		[2]		
	(c)	both axes labelled at least one with units; linear scale covering >1/2 paper; at least 4 plots correct ± half square;				
		best fit line ;		[4]		
	(d)	increases amount of juice produced/more juice per unit time;		[1]		
	(e)	keeps volume in each beaker constant/show that the water of enzyme solution does not have an effect/no effect without enzyme;				
				[Total: 10]		
2	(a)	(i) 124;		[1]		
		(ii) C is $2.00 \mathrm{mol}\mathrm{dm}^{-3}$ D is $0.50 \mathrm{mol}\mathrm{dm}^{-3}$ E is $1.00 \mathrm{mol}\mathrm{dm}^{-3}$ ;;		[2]		
		one correct = 1 mark, three correct = 2 marks				
	(b)	(b) add marble chip/add UI/add Mg; (marble chips or magnesium) count bubbles/collect gas/measure volume of gas; in a certain time; OR				
		(for marble chips) time ; for limewater to go milky ; OR				
		add NaOH from measuring cylinder/burette; until UI just green;				
		the more bubbles or gas the more concentrated/the shorter the time (followed) limewater) the more concentrated/the more NaOH the more concentrated				
		equal volumes of the acids (in test–tubes);		[5]		
	(c)	(acidified) silver nitrate/AgNO <sub>3</sub> AND white ppt.;		[1]		
	(d)	too long for magnesium to disappear/reaction too slow;		[1]		
				[Total: 10]		

P	age :	3	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – May/June 2016	0654	61
3	(a)	p =	29.5 cm;		[1]
	(b)		alues correct (e.c.f. <i>p</i> ) 5 ecf, (21.8), <b>19.1, 16.4, 13.6</b> ;		
		-	alues correct 5 ecf, (18.2), <b>15.9, 13.6, 11.4</b> ;		[2]
	(c)	(i)	suitable choice of scales $\geqslant \frac{1}{2}$ the grid (can plot the 5 points) used a minimum 4 plots correct to $\frac{1}{2}$ small square on easy to read scale; good best fit straight line judgement;	AND linear ;	[3]
		(ii)	indication on graph of how the data were obtained AND more than calculation correct;	half the line	; [2]
	(d)	m c	correct to 2/3 significant figures ;		[1]
	(e)	An	y one from:		
		cer	iculty in obtaining balance ; itre of mass of rule not at the 50.0 cm mark ; d not uniform ;		
			iculty in placing the centre of load over the mark on the rule ;		[1]
					[Total: 10]
4	(a)		ced in the dark ; east 24 hours ;		[2]
	(b)	(i)	potassium hydroxide/sodium hydroxide/soda lime;		[1]
		(ii)	any in the same state as (i) that does not absorb $CO_2$ ;		[1]
	(c)	(i)	iodine solution; boiling/hot water; hot alcohol/ethanol; rinse with water; (safety) water bath/not naked flame;		[max 4]
		(ii)	G is blue-black AND F is brown/orange; (because) G can photosynthesise and F cannot (photosynthesise)	• ,	
			OR		
			F is brown no photosynthesis ; G is blue-black can photosynthesise ;		[2]
			•		Total: 10]
					-

Mark Scheme

Syllabus

Paper

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Page 4		4	Mark Scheme		Paper
			Cambridge IGCSE – May/June 2016	0654	61
5	(a)	(i)	limewater; white ppt.;		[2]
		(ii)	diagram showing filter funnel and paper ; two relevant labels ;		[2]
		(iii)	blue ppt. AND blue ppt.; (deep) blue solution; blue ppt.;		[3]
	(b)	cop	per carbonate / CuCO <sub>3</sub> ;		[1]
	(c)		e of (acidified aqueous) barium chloride/barium nitrate ; te ppt. etc. ;		[2]
					[Total: 10]
6	(a)	(i)	112;		[1]
		(ii)	correct symbol for ammeter and voltmeter; ammeter in series and voltmeter in parallel; correct symbols for lamp and switch in series; workable circuit (no short circuits, no gaps);		[4]
		(iii)	54 and 21 ; 33 (ecf) ;		[2]
		(iv)	112 (ecf) $\times$ 33 (ecf) $\times$ 4.2/1000 = 15.5/16;		[1]
	(b) air/surroundings; wires/leads/(heater) casing/circuit;				
			P e.g. heat transferred to: beaker/used in evaporation ;		[max 2]
					[Total: 10]