CAMBRIDGE INTERNATIONAL EXAMINATIONS

Cambridge International General Certificate of Secondary Education

MARK SCHEME for the October/November 2014 series

0653 COMBINED SCIENCE

0653/31

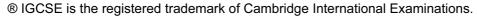
Paper 3 (Extended Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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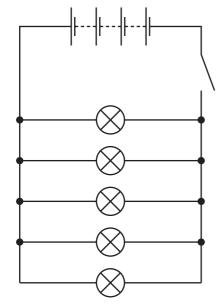
Cambridge is publishing the mark schemes for the October/November 2014 series for most Cambridge IGCSE[®], Cambridge International A and AS Level components and some Cambridge O Level components.





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1 (a)



symbols all correct;

circuit connected correctly (allow ±1 cell or lamp);

[2]

[1]

(b) (i)
$$5 \times 0.5 = 2.5$$
 (A);

(ii) (R =) V/I (or words); = $6 / 2.5 = 2.4 (\Omega)$;

[2]

(c) series: all bulbs go out AND parallel: rest of bulbs stay alight;

[Total: 6]

[1]

2 (a) BC;

(BC)DA;

[2]

(allow 1 mark if both **B** and **A** are correctly located)

(b) (i) catalyst;

[1]

(ii) increases rate / frequency of collision of particles;

increases speed of reaction / increases surface area (of catalyst);

[2]

(iii) (petroleum) jelly (diesel) oil

(refinery) gas

in order;

[1]

(iv) (petroleum) jelly

(diesel) oil (refinery) gas

in order;

[1]

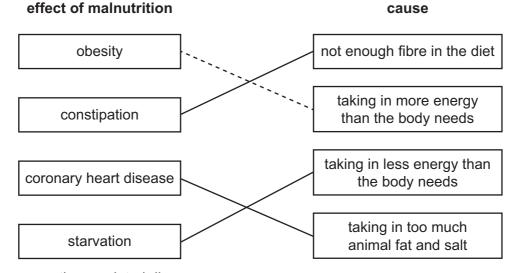
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(v) the higher the boiling point the longer / larger the molecules; reference to greater / stronger intermolecular forces; (allow reference to intermolecular bonds)

[2]

[Total: 9]

3 (a) (i)



correctly completed diagram ;; (3 correct = 2 marks, 2 or 1 correct = 1 mark) [2]

(ii) example of fruit or vegetable containing fibre; provides bulk to propel food through the intestines;

- [2]
- (iii) any food rich in carbohydrate or fat / carbohydrate or fat (no mark) reference to reducing energy intake / avoiding the carbohydrate or fat content of the stated food;

[1]

- (b) (i) more females than males / fewer males than females took exercise; more normal weight than obese / fewer obese than normal weight took exercise; [2]
 - (ii) reference to small sample size;
 reference to the lack of information about variables that should be controlled (if the study were to be extended);
 reference to the need for information gathered over a longer time period; [max 2]

[Total: 9]

_						
4	(a)	(i)	initial between 8 and 14 to 7 (final);	[1]		
		(ii)	purple / blue to green ;	[1]		
	(b)	(i)	KC <i>l</i> ; H ₂ O ;	[2]		
		(ii)	repeat without indicator / use pH meter / use indicator paper ; using same volume(s) of solution(s) ;			
			evaporate (the water from the neutral mixture) / heat (the solution) then cool;	[3]		
	(c)	reference to the involvement of ions / ionic compound / particles with opposite charge the idea of strong forces / bonds between particles that must be broken / ions must be separated;				
			eaking bonds / separating ions) requires a large amount of energy;	[max 2]		
				[Total: 9]		
5	(a)	gre	abel line to green area and Y label line to white area ; en area containing chlorophyll / chloroplasts only in cell X / ite area does not contain chlorophyll /chloroplasts shown in cell Y ;	[2]		
	(b)	(i)	black or shaded in area matching green area of leaf and indicated as black;	[1]		
	(2)		chlorophyll / chloroplasts traps <u>light</u> energy ;	[.]		
			for photosynthesis; which makes (glucose / sugar which leads to) starch;	[3]		
	(c)		d <u>denatures</u> enzyme ;			
			longer optimum pH / owtte ; anges shape of enzyme / active site / substrate no longer fits active site ;	[3]		
				[Total: 9]		

Mark Scheme

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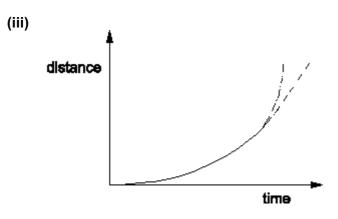
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Syllabus 0653 Paper 31

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6 (a) (i) P and R; [1]

(ii) R;
(R) is the weight; [2]



[1] (ignore whether curve becomes linear or continues to curve)

(b) (i) gravitational / potential energy and kinetic energy; (both required) [1]

(ii) (rest of energy transferred to) heat / sound; [1]

(c) (i) $(720 \times 1000) \div 3600 / 200 \text{ (m/s)}$; [1] $(OR\ 200 \times 3600/1000 = 720 \text{ km/h})$

(ii) (KE =) ½ m v²; = ½ × 200 000 × 200 × 200 = 4 000 000 000(J); (allow ecf from (c)(i)) (allow answers in kJ or MJ provided unit is stated)

7 (a) (i) starch digested to glucose / sugar;

(glucose / sugar) absorbed and taken to cells (of sheep);
(glucose broken down by) respiration;
(respiration produces) carbon dioxide /
carbon dioxide breathed out into the air;
(allow reference to respiration and exhalation in the wild cat)
(allow correct reference to the formation and release of methane)

[Total: 9]

(ii) decomposers;

release carbon / carbon dioxide (into the air); by respiration; [max 3]

feed on dead / decaying organisms / feed on waste from organisms;

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	(b)		increases level of carbon dioxide / carbon monoxide; reduces oxygen level; increases sulfur dioxide level; carbon dioxide:						[n	nax 2]
		(,	(increases) g e.g. changed	lobal warn						
			sulfur dioxide causes acid r e.g. chemical reference to l	ain / desc I weatherir	ng of structu	ires / damage		aquatic organis	ms ;	
			(allow other v				···9 ,		[n	nax 1]
									[Tot	tal: 9]
8	(a)	(i)	number of vib	orations / v	vaves per s	econd / unit o	f time ;			[1]
	,				·		,			
		(ii)	highest freq	uency				lowe	st freque	ency
			(gamma radiation)	X-rays	ultra- violet	(visible light)	infra-red	(microwaves)	(radio waves	-
			all three corre and in correc (allow 1 mark	t positions	•	amed and loc	ated)			[2]
	(b)	(i)	move further apart / increase distance between them; decrease / weaken / get less; quicker / more rapidly / faster / further apart;				[3]			
		(ii)	infra-red radiation (from Sun warms water); (energy from sun) absorbed by water (molecules); which move faster / gain kinetic energy.; forces between molecules are weakened / broken; (molecules) evaporate / leave the (liquid) surface / turn to gas / vapour; [magestate of the content of the				nax 2]			
	(c)	(i)	sound is a lor sound needs space is a va	medium to	o travel thro	ough ;			[n	nax 2]
		(ii)	8 minutes / th		•	avel at the sar	•			[4]

[Total: 11]

[1]

electromagnetic waves travel at the same speed through space / vacuum ;

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9	(a)	(i)	exothermic;		[1]
		(ii)	$\underline{\text{chemical}} \; (\text{potential}) \; \rightarrow \; \text{thermal / heat / kinetic} \; ;$		[1]
		(iii)	aluminium (gains oxygen and) is oxidised; iron (oxide) (loses oxygen and) is reduced; (allow correct references to electron gain by iron and electron loss	from alumir	[2] nium)
		(iv)	iron will not react with / reduce aluminium oxide; iron is lower in the reactivity series / less reactive than aluminium;		[2]
	(b)	(i)	cations / aluminium ions migrate / move / are attracted to the catho / negative electrode; electrons flow on to ions / ions gain electrons; the idea that the ions are discharged as the result of electron gain;		[max 2]
		(ii)	oxygen;		[1]

[Total: 9]