

# Lipids: Not Just Fat

## © Chapter Outline

The chapter outline provides you with an organizational guide to the topics and ideas presented in this chapter of the text.

#### What Are Lipids?

### Fatty Acids Are Key Building Blocks

Chain Length Saturation

Cis Versus Trans

Nonessential and Essential Fatty

Acids

#### **Triglycerides**

Triglyceride Structure Triglyceride Functions Triglycerides in Food

#### **Phospholipids**

Phospholipid Structure Phospholipid Functions Phospholipids in Food

#### Sterols

Cholesterol Functions Cholesterol Synthesis Sterols in Food

#### Lipid Digestion and Absorption

Digestion of Triglycerides and Phospholipids Lipid Absorption Digestion and Absorption of Sterols

#### Lipids in the Body

Chylomicrons
Very-Low-Density Lipoprotein
Intermediate-Density Lipoprotein
Low-Density Lipoprotein
High-Density Lipoprotein

#### Lipids in the Diet

Recommended Intakes
Essential Fatty Acid Requirements
Omega-6 and Omega-3 Balance
Current Dietary Intakes
Role of Fat Replacers

#### Lipids and Health

Obesity
Heart Disease
Dietary and Lifestyle Factors for Reducing
Atherosclerosis Risk
Metabolic Syndrome
Putting It All Together
Cancer

# Key Terms

Define the following terms.

1.	adipocytes
2.	squalene
3.	diglyceride
4.	subcutaneous fat
5.	cholesterol
6.	visceral fat
7.	metabolic syndrome
8.	hypercholesterolemia
9.	chain length
10.	hydrogenation
11.	lanugo
	phytosterols
13,	omega-3 fatty acid
14.	micelles
	adipose tissue
16.	eicosanoids
17.	very-low-density lipoproteins (VLDL)
18.	omega-6 fatty acid
19.	monounsaturated fatty acid
20.	atherosclerosis
21.	intermediate-density lipoproteins (IDL)
22.	sterols
	oxidation
24.	phosphate group
25.	polyunsaturated fatty acid
26.	obesity

# 

is an essential omega-3 fatty acid that contains		
carbon atoms and 3 carbon-carbon double bonds.		
term for diseases in which abnormal cells divide without control is		
ese cells can invade nearby tissues and can spread through the bloodstream and lymphatic system to		
her parts of the body.		
is an unsaturated fatty acid with a bent carbon		
ain. Most naturally occurring unsaturated fatty acids are this type.		
is the backbone of mono-, di-, and		
glycerides; alone, it is a thick, smooth liquid.		
are fatty acids that the body needs but cannot		
nthesize and must obtain from the diet.		
is a fat replacer made from a sucrose		
ckbone with six to eight fatty acids attached. The fatty acid arrangement prevents breakdown by the		
gestive enzyme lipase, so the fatty acids are not absorbed. It can withstand heat and is stable at frying		
mperatures. Its trade name is Olean.		
are compounds containing a long hydrocarbon		
nain with a carboxyl group (-COOH) at one end and a methyl group (-CH <sub>3</sub> ) at the other end.		
A fatty acid in which the carbon chain contains one or more double bonds is an		
are compounds that consist of a glycerol		
olecule bonded to two fatty acid molecules, and a phosphate group with a nitrogen-containing		
omponent. They have both water-soluble and fat-soluble regions, which make them good emulsifiers.		
the blood lipoproteins that contain high levels of protein and low levels of triglycerides are called		
Synthesized primarily in the liver and small		
testine, these pick up cholesterol released from dying cells and other sources and transfer it to other		
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12.	is a protein released by the body			
•	in response to acute injury, infection, or other inflammatory stimuli. It is associated with future			
	cardiovascular events.			
13.	A molecule of glycerol combined with one fatty acid is a			
14.	A general term for all disorders affecting the heart and blood vessels is			
15.	are the cholesterol-rich lipoproteins that result			
	from the breakdown and removal of triglycerides from intermediate-density lipoprotein. Sometimes			
	called "bad cholesterol."			
16.	is a substance that consists of an LDL part plus			
	a protein (apoprotein a) whose exact function is currently unknown.			
17.	A fatty acid completely filled by hydrogen, with all carbons in the chain linked by single bonds, is a			
18.	is an essential omega-6 fatty acid that contains			
	18 carbon atoms and 2 carbon-carbon double bonds (18:2); it is a thin liquid at room temperature.			
19.	are compounds that imitate the functional and			
	sensory properties of fats, but contain less available energy than fats.			
20.	is a nitrogen-containing compound that is part			
	of phosphatidylcholine, a phospholipid. It also is part of the neurotransmitter acetylcholine. The body			
	can synthesize it from the amino acid methionine.			
21.	In the body, is a phospholipid with the			
	nitrogen-containing component choline. In foods, it is a blend of phospholipids with different nitrogen-			
	containing components.			
22.	A is an unsaturated fatty acid with a straighter			
	chain than a cis fatty acid, usually as a result of hydrogenation; it is more solid than a cis fatty acid.			
23.	A is a large lipoprotein formed in intestinal			
	cells following the absorption of dietary fats. It has a central core of triglycerides and cholesterol			
	surrounded by phospholipids and proteins.			
	are fatty acids that your body can make when			

25.	A is a complex that transports lipids in the			
	lymph and blood. It consists of a central core of triglycerides and cholesterol surrounded by a shell			
	composed of proteins, cholesterol, and phospholipids. The various types differ in size, composition, and			
	density.			
0	Fill-in-the-Blank Summaries			
Fatt	y Acids Are Key Building Blocks			
Lipi	ds are a broad range of molecules that dissolve easily in organic solvents, but are much less soluble in			
	They are generally hydrophobic and lipophilic. The three			
	main types of lipids are triglycerides, phospholipids, and sterols.			
	are the largest category of lipids and are stored in the body in			
	are major building blocks of cell membranes.			
	They keep fats suspended in watery fluids. The most famous sterol, cholesterol, is manufactured in			
	the body and is a precursor to the synthesis of sex hormones,, and			
	vitamin D. Lipids share many of the same functional properties and transport mechanisms.			
	are common components of triglycerides and			
	phospholipids. There are many types of these substances, which are basically chains of			
	atoms with a carboxyl group on one end and a methyl group at the other end.			
	Short-chain fatty acids have less than carbons; medium			
	chains have 6 to 10; and long chains have or more. The			
	water-soluble property of shorter fatty acids affects their absorption.			
A tr	iglyceride is made of fatty acids attached to a glycerol molecule.			
	Triglycerides are esters that form when a hydrogen and an oxygen from the carboxyl group combine			
	with a hydrogen atom from the alcohol's hydroxyl group. A molecule of water is produced, which makes			
	this a reaction.			
Lip	ids in the Body			
Lip	oprotein carriers transport lipids through the bloodstream. Chylomicrons are formed in the			
	They eventually reach the bloodstream through the			
	in the neck. As they travel through the bloodstream, they gradually give up			
	triglycerides to capillary walls. Lipoprotein lipase breaks them down. Afterhours			
	little is left of the chylomicron except cholesterol-rich remnants.			

Elevated levels of		(also referred to as "bad	cholesterol") in the blood increase the risk				
	of artery and heart disease.						
	, (also referred to as "good cholesterol") appear to protect against atherosclerosis.						
	They are scavenger lipoproteins	, picking up excess	released by dying cells and				
	arterial plaques. These plaques are created by LDL degrading over a long period of time.						
Lipi	ds in the <b>Die</b> t						
The Key Recommendations of the Dietary Guidelines for Americans, 2010, with respect to lipid intake, include:							
	Consume less than	of calories from	m saturated fatty acids by replacing them				
	with monounsaturated and polyunsaturated fatty acids.						
	• Consume less than 300 mg per day of						
	<ul> <li>Keep trans fatty acid consumption as low as possible by limiting foods that contain synthetic sources</li> </ul>						
•	of trans fats, such as	, and by limiti	ng other solid fats.				
Whe	When protein is used as part of a fat substitute, the product cannot be used in cooking because high						
temperatures cause denaturation. Many products use carbohydrates as fat replacements and bind							
		to further dilute calorie	es. Olestra is a very controversial fat				
	substitute. Because olestra is no	t absorbed, it may cause sym	ptoms of fat malabsorption such as				