

Simulation Methods in Physics I

Worksheet 3: Molecular Dynamics 2 and Observables

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1 Command line parameters

For the simulation different command line parameters were defined. In the following code blocks you can see them. They will be explained during the report.

Code block 1: Commands for ljsim.py

script: src/ljsim.py

```
9 # command line arguments
10 parser = argparse.ArgumentParser()
11 parser.add_argument("--cont", type=double, help="continue
    calculation with for cont further time")
12 parser.add_argument("--time", type=double, help="How long
    do you want to run the simulation? | default time=10s")
13 parser.add_argument("--tstat", type=double, help="Uses a
    thermostat with a given temperature")
14 parser.add_argument("--warm", type=double, help="Use the
    warming up | pass force")
15 parser.add_argument("--ctstat", type=double, help="
    continue for the simulation with tstat")
16 parser.add_argument("--cwarm", type=double, help="continue
    for the simulation with warm")
17
18 args = parser.parse_args()
```

Code block 2: Commands for ljanalyze.py

script: src/ljanalyze.py

```
8 # command line arguments
9 parser = argparse.ArgumentParser()
10 parser.add_argument("--M", type=int, default=10, help="
    window size | default: M=10")
11 parser.add_argument("--datafile", type=str, default='../
    dat/ljsim.dat', help="datafilename | default: '../dat/
    ljsim.dat'")
12 parser.add_argument("--teq", type=double, help="
    equilibration time [s] | calculates averages after
    equilibration time")
13 parser.add_argument("--tlim", type=double, nargs='+', help
    ="type in the time limits you want to plot")
14
15 args = parser.parse_args()
```

2 Restart the program where left it

First we want to change the program in a way, that it is able to restart the simulation where it ended last time. The first thing that we have to do is to store the necessary information about the end state of the system, the position x and velocity v of each particle, into the file `ljsim.dat`. This happens in code block 3. Furthermore there are the new variables T_s and P_s . They represent the temperature and pressure of the system over time and will be explained in the next chapter.

Code block 3: Storing data

script: `src/ljsim.py`

```

208 # write out simulation data
209 print("Writing simulation data to {}".format(datafilename
      ))
210 datafile = open(datafilename, 'w')
211 pickle.dump([ts, Es, Ts, Ps, x, v], datafile)
212 datafile.close()

```

Now this data must be read by `ljsim.py`. Therefore we define a command line parameter `--cont` which accepts as argument the time for which the simulation should be continued (code block 1). If the parameter is not used the simulation will start new. Furthermore there is another argument `--time` which takes the simulation time for a new simulation. The code in code block 4 does exactly this.

Code block 4: Continue simulation

script: `src/ljsim.py`

```

65 # Import previous data
66 if args.cont:
67     # open datafile
68     datafile = open(datafilename, 'r')
69     ts, Es, Ts, Ps, x, v = pickle.load(datafile)
70     datafile.close()
71
72     # length of run
73     t = ts[-1]
74     tmax = t+args.cont
75     step = 0
76
77 else:
78     # length of run
79     if args.time:
80         tmax = args.time
81     else:
82         tmax = 10.0

```

3 Calculating Temperature and Pressure

3.1 Expanding Energy Calculation

First the energy calculation should be changed in order to also get the kinetic and potential energy. This is done by changing the `compute_energy()` function call and the parameter `Es` in `ljsim.py` (code block 5).

Code block 5: $E_{\text{kin}} + E_{\text{pot}}$

script: `src/ljsim.py`

```

179     if step % measurement_stride == 0:
180         E_pot, E_kin, E_tot = compute_energy(x, v)
181         T = 2*E_kin/(3*N)
182         P = compute_pressure(E_kin, x)
183         print("t={}: \n\tE_pot={}\n\tE_kin={}\n\tE_tot={}\n\tT={}\n\tP={}".format(t, E_pot, E_kin, E_tot,
184                                                     T, P))
185         ts.append(t)
186         Es.append([E_pot, E_kin, E_tot])
187         Ts.append(T)
188         Ps.append(P)

```

3.2 Calculating the Temperature

The temperature of the system can be calculated from the kinetic Energy E_{kin} like in code block 5 shown. This is a single scalar calculation and therefore can be done in python.

3.3 Calculating Pressure

Derivation of the pressure

The virial of a system is defined as:

$$G = \sum_i^N \mathbf{p}_i \cdot \mathbf{r}_i \quad (1)$$

In the case

$$0 = \left\langle \frac{dG}{dt} \right\rangle = \left\langle \sum_i^N \frac{\mathbf{p}_i^2}{m_i} \right\rangle + \sum_i^N \langle \mathbf{F}_i \cdot \mathbf{r}_i \rangle \quad (2)$$

the following equation can be derived:

$$-\sum_i^N \langle \mathbf{F}_i \cdot \mathbf{r}_i \rangle = 2 \left\langle \sum_i^N \frac{\mathbf{p}_i^2}{2m_i} \right\rangle = 2 \langle E_{\text{kin}} \rangle = 3Nk_B T \stackrel{\text{id. gas}}{=} 3PV \quad (3)$$

For an pair interaction (like in the simulation) let the forces \mathbf{f}_{ij} and the vectors \mathbf{r}_{ij} be defined as follows:

$$\mathbf{r}_{ij} = \mathbf{r}_j - \mathbf{r}_i \quad (4)$$

$$\mathbf{f}_{ij} = f_{ij}(\mathbf{r}_{ij}^2) \cdot \frac{\mathbf{r}_{ij}}{|\mathbf{r}_{ij}|} \quad (5)$$

With this knowldege we can derive the pressure of the system:

$$P = P_{\text{id. gas}} + P_{\text{interaction}} \quad (6)$$

$$= \frac{Nk_B T}{V} + \frac{1}{3V} \sum_i^N \langle \mathbf{F}_{i,\text{interaction}} \cdot \mathbf{r}_i \rangle \quad (7)$$

$$= \frac{1}{3V} \left[\left\langle \sum_i^N \frac{\mathbf{p}_i^2}{2m_i} \right\rangle + \left\langle \sum_{i,j \neq i}^N -\mathbf{f}_{ij} \cdot \mathbf{r}_i \right\rangle \right] \quad (8)$$

$$= \frac{1}{3V} \left[\left\langle \sum_i^N \frac{\mathbf{p}_i^2}{2m_i} \right\rangle + \left\langle \sum_{i,j > i}^N \mathbf{f}_{ij} \cdot \mathbf{r}_j - \mathbf{f}_{ij} \cdot \mathbf{r}_i \right\rangle \right] \quad (9)$$

$$= \frac{1}{3V} \left[\left\langle \sum_i^N \frac{\mathbf{p}_i^2}{2m_i} \right\rangle + \left\langle \sum_{i,j > i}^N \mathbf{f}_{ij} \cdot \mathbf{r}_{ij} \right\rangle \right] \quad (10)$$

$$= \frac{1}{3V} \left[2 \langle E_{\text{kin}} \rangle + \left\langle \sum_{i,j > i}^N \mathbf{f}_{ij} \cdot \mathbf{r}_{ij} \right\rangle \right] \quad (11)$$

Pressure in Cython

Beacaus of the vectorial calculations for each particle the calculation has to be done in C if it should be fast. Therefor the function `c_compute_pressure()` is written in `c_lj.cpp`.

Code block 6: $E_{\text{kin}} + E_{\text{pot}}$

script: `src/c_lj.cpp`

```

253 double c_compute_pressure(double E_kin, double* x) {
254     double rij[3];
255     double fij[3];
256     double interaction = 0.0;
257
258     // for (int i = 1; i < N; i++)
259     //     for (int j = 0; j < i; j++) {
260
261         // add up fij * rij
262     vector<int>::iterator it = verlet_list.begin();
263     vector<int>::iterator end = verlet_list.end();
264     while (it != end) {
```

```

265     int i = *it;
266     ++it;
267     int j = *it;
268     ++it;
269
270     if (i!=j){
271         minimum_image(x, i, j, rij);
272         compute_lj_force(rij,fij);
273
274         for (int k = 0; k < 3; k++)
275             interaction += fij[k]*rij[k];
276     }
277 }
278 return (2.*E_kin+interaction)/(3*L*L*L);
279 }

```

4 Molecular Dynamics at a Desired Temperature

For the *velocity rescaling* thermostat we can derive the rescaling-factor f_{re} from equation (12).

$$\frac{3}{2}k_B T_0 = \frac{E_{\text{kin},0}}{N} \quad (12)$$

$$= \frac{1}{N} \sum_{i=1}^N \frac{(f_{\text{re}} \cdot \mathbf{v}^{(i)})^2}{2m} \quad (13)$$

$$= f_{\text{re}}^2 \frac{E_{\text{kin}}}{N} \quad (14)$$

$$= f_{\text{re}}^2 \frac{3}{2} k_B T \quad (15)$$

$$f_{\text{re}} = \sqrt{\frac{T_0}{T}} \quad (16)$$