



Cite this: *Chem. Soc. Rev.*, 2018,
47, 8925

Received 3rd September 2018
DOI: 10.1039/c8cs00716k

rsc.li/chem-soc-rev

Recent applications of C–H functionalization in complex natural product synthesis

Dylan J. Abrams,[†] Philip A. Provencher[†] and Erik J. Sorensen  *

In this review, recent examples featuring C–H functionalization in the synthesis of complex natural products are discussed. A focus is given to the way in which C–H functionalization can influence the logical process of retrosynthesis, and the review is organized by the type and method of C–H functionalization.

Introduction

C–H functionalization has long captured the imagination of chemists. The concept has influenced scientists studying problems ranging from feedstock chemical upconversion, understanding the fundamental nature of C–H bonds, to the synthesis of complex organic molecules. In the field of applying C–H functionalization to total synthesis, the key development in the 2010s had been the emergence and exploitation of reliable models to predict selectivities in C–H bond functionalizations. Due to the near-universal preponderance of C–H bonds in organic compounds, a practitioner of organic chemical synthesis must engage in especially careful planning if a C–H functionalization

strategy is to be undertaken. To aid in this synthetic planning, two main factors may govern selectivity in C–H bond activation: innate selectivity¹ or catalyst/reagent control.² Pattern recognition in retrosynthetic analysis enables the utilization of either strategy, and in some syntheses the two methods can be complementary. While most methods for directly transforming unactivated C–H bonds display some degree of regioselectivity, one of the emerging trends in this vibrant subfield of organic chemistry is the quest for enantio- and diastereoselectivity.

C–H functionalization reactions are now possible for the selective conversion of strong, unactivated C–H bonds into C–X, C–O, C–N, and C–C bonds. As patterns emerge from the variety of C–H functionalization methods, they may enter the crucible of total synthesis. Not only are new methods inspiring C–H functionalization strategies in total synthesis, but the complex molecular scaffolds of natural products themselves are motivating

Department of Chemistry, Princeton University, Princeton, NJ 08544, USA.
E-mail: ejs@princeton.edu; Web: <http://www.chemists.princeton.edu/sorensen>
† These authors contributed equally.



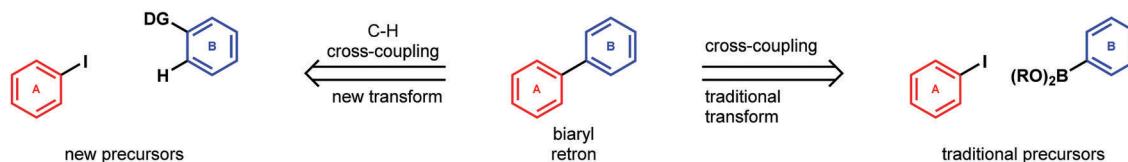
Dylan J. Abrams

Dylan J. Abrams received his Bachelor of Science degree in Chemistry from the University of Rochester in 2015. While there, he engaged in research in the field of natural product synthesis in the lab of Professor Alison Frontier. In 2015, he moved to Princeton University to start his graduate studies with Professor Erik Sorensen. His research interests include studying new C–H functionalization reactions and their applications in the synthesis of natural products.



Philip A. Provencher

Philip A. Provencher completed his BSc (2011) and MSc (2014) at the University of British Columbia under the tutelage of Prof. Jennifer A. Love. At UBC he worked on catalytic C–F bond activation and the development of molecular alkylating agent sensors. He worked in the chemical industry and entrepreneurship before finally heading to Princeton University in 2016, where he currently works on C–H functionalization and total synthesis under the guidance of Prof. Erik J. Sorensen.



Scheme 1 The analogy between traditional transforms and their C–H functionalization variations.

Published on 14 November 2018. Downloaded by University of Birmingham on 2/6/2026 11:13:22 AM.

chemists to develop bold structural transformations to expedite synthesis. In these endeavors, practitioners of target-directed synthesis are free to exploit the steering effects of directing elements and capitalize on differences in the intrinsic reactivities of C–H bonds in their efforts to demonstrate the growing potential of C–H functionalization in complex synthesis. Indeed, the use of several different types of C–H functionalization in a single total synthesis provides a shape to this varied field. This review addresses this burgeoning field and gives an emphasis to recently developed strategies for synthesis that are highly reliant on C–H functionalization methods and the bold ideas that they engender. In our discussions of prime, recent examples of site-selective C–H functionalizations in natural product synthesis, we show how the activation of C–H bonds can simplify and expedite the synthesis of complex molecular scaffolds in relation to traditional methods for synthesis. We also show thoughts about particular C–H bonds in intricate targets have inspired bond disconnections that are distinct from prior syntheses of the same or similar scaffolds.

It can be difficult to envision the use of strategic C–H functionalization in retrosynthesis, and yet it can totally reshape the synthetic patterns of the molecule. As Danishefsky wrote:³

“The most fascinating cases arise when the target itself must actually be modified, mentally, before a... pattern is revealed... there are many levels of opportunity for creativity in deeply disguised patterns”



Erik J. Sorensen

Erik J. Sorensen is the Arthur Allan Patchett Professor in Organic Chemistry in the Department of Chemistry at Princeton University, USA and Perennial Distinguished Guest Professor in the School of Pharmaceutical Science and Technology at Tianjin University, China. He received his PhD degree in Chemistry in 1995 from the University of California, San Diego under the direction of Professor K. C. Nicolaou. From 1995–1997, he was an NSF postdoctoral fellow in the laboratory of Professor Samuel Danishefsky. In 1997, he joined the faculty at The Scripps Research Institute and achieved the rank of Associate Professor in 2001. In 2003, he moved to Princeton University, where he continues to teach and pursue research in organic chemistry. In 2015, he became a Perennial Distinguished Guest Professor at Tianjin University.

And we submit that strategies featuring C–H functionalization can uncover some of the deepest, disguised patterns and seed highly creative approaches for synthesis. As the field of C–H functionalization expands and evolves, so too does our ability to implement these reactions in complex synthetic settings. C–H functionalization specifically changes the relationship between a retrón and its precursor(s), directly influencing the logic of retrosynthesis. While recognition of the new precursors to a retrón are critical to leveraging C–H functionalization in retrosynthesis, so too is recognizing the ways in which a C–H functionalization disconnection can benefit the synthesis as a whole. As Corey wrote:⁴

“Fundamental to the wise choice of transforms is the awareness of the position of each transform on the hierarchical scale of importance with regard to simplifying power and the emphasis on applying those transforms which produce the greatest molecular simplification.”

Of course, this simplification in the literal sense could be in the form of a reduction in molecular complexity. But, it can also be reflected in a much-reduced reliance on activating functional groups and the flexibility this can provide to earlier synthetic steps, or in the introduction of high symmetry elements to a precursor. In this way, C–H functionalization reactions provide not just a new precursor to be matched with traditional retrosynthetic disconnections, but, in a larger sense, a highly simplifying class of transforms that can fundamentally influence the logic of retrosynthesis. Site- and stereoselective C–H functionalizations are emerging into powerful reactions and having a profound impact on efforts to design and execute laboratory syntheses of structurally intricate organic compounds and natural products (Scheme 1).

Owing to the existence of several, excellent reviews on the subject of C–H functionalization in total synthesis,^{5–8} including Baran’s comprehensive, historical account⁹ and Itami’s expansive discussion of the synthesis of both natural products and drug molecules,¹⁰ the scope of this review is restricted to syntheses published in the wake of those two reviews; this treatise addresses the period comprising 2012–2018. However, as this field continues to advance, with ever more synthetic methods to transform C–H bonds, and more accessible and synthetically useful directing groups and catalysts, C–H functionalization in the context of sophisticated undertakings in total synthesis has only grown richer and more inspiring over these years. Our sincere hope is that this review will capture some of the vitality that is so easy to see in recent examples of the power of C–H functionalization in complex synthesis. Like all reviews of this sort, ours is not to be viewed as comprehensive, and we apologize to the readership for unavoidable omissions in our coverage.

The identification and exclusion of which reactions are or are not considered “C–H functionalization” can be difficult at times. Indeed, well studied, traditional reactions occurring at acidic or acidified C–H bonds, such as the reactions of enolate ions or those passing through the Wheland intermediate, are, in fact, transformations of C–H bonds to other functional groups. However, for the purposes of this review, we will define C–H functionalization as those reactions that directly transform otherwise strong and unactivated C–H bonds. This definition thus embraces the diversity of reaction mechanisms by which a C–H functionalization occurs (*e.g.* radical chemistry, non-octet carbenes and nitrenes, metal coordination, *etc.*), as well as creative uses of traditional reactions to transform a C–H bond (*e.g.* directed lithiation reactions).

Finally, because our interest is on the way that C–H functionalization methods can shape retrosynthetic logic, recent syntheses that update or build upon C–H retrosynthetic disconnections that were covered in prior reviews are also excluded. Despite the highly focused nature of our topic, we find that in the last six years, over 50 syntheses featuring bold C–H functionalizations have been published. This contribution will be divided first based on the type of C–H bond being functionalized and the mechanistic/selectivity method by which the reaction takes place. Within the sections herein, we strive to group together synthetic strategies bearing some resemblance to one another (with respect to reaction conditions, type of bond formed, *etc.*) in order to better demonstrate the strategic significance of each reaction type. We begin with reactions controlled by some internal directing element. We then group the reactions by the type of C–H bond hybridization, with functionalization of C(sp²)-H bonds on heteroarenes getting their own category. Following this, we discuss C–H insertion reactions of (typically metal-bound) carbenes and nitrenes. Within this category, we also include non-directed C–H oxygensations. This is not to imply that these reactions proceed by an analogous ‘oxene’ species, but rather because the trend of inherent molecular reactivity of these reactions seems to align itself with carbene and nitrene C–H insertions. Finally, we conclude the review with some creative radical-based reactions and some reactions catalyzed by enzymes.

Directed C(sp³)-H functionalization

The use of directed sp³ C–H functionalization in total synthesis has expanded rapidly since the creative synthesis of rhamnilar by the Sames group featuring a substrate-directed desaturation.¹¹ Organic chemists have benefited from the expansion of new methodologies developed in this arena by groups like Sanford,¹² White,¹³ Yu^{14–16} and many others.¹⁷ While new methods have enabled some spectacular transformations in total synthesis endeavors, the complex architectures of natural products have also inspired the design of new transformations when required by the synthesis. It is atypical, as one might expect, that a C–H functionalization methodology would be initially developed during the course of a total synthesis. This is due to the common (although not universal) necessity of a proximate, covalently

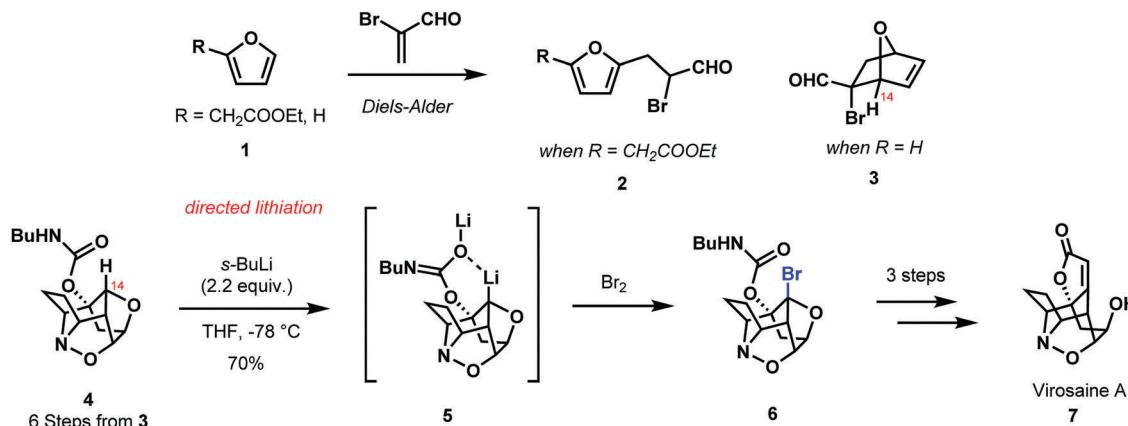
attached directing group, which might not directly correspond to a functionality or functional group precursor found in the natural structure. Because of this, however, it is invariable that total synthesis will proffer unique questions regarding the complex architecture which controls the chemistry. Often, gaining entry into these architectures first necessitates multistep synthesis, at times a non-trivial task. In this way, C–H functionalization lays a strong argument for the pursuit of total synthesis because the questions that the chemistry poses are only answerable in the course of the synthesis.

In particular, the activation of C(sp³)-H bonds is of great interest to the field of natural product synthesis. Unlike sp² hybridized carbons, saturated carbons have no pi-bonding functionality, further reducing the opportunity to functionalize them with more classical reactions. In this sense, the functionalization of C(sp³)-H bonds perhaps most greatly offers the opportunity to discover transformations to install otherwise inaccessible functionalities and access new retrosynthetic disconnections. Additionally, C(sp³)-H bonds offer questions of stereochemistry, both stereospecificity of functionalization (in the case of carbon bearing only one hydrogen) and stereoselectivity (in the case of methylene carbons). The abundance of new syntheses below implementing this type of C–H functionalization reflects these observations.

In their creative synthesis of virosaine A (7), the Gleason group turned to a C–H functionalization strategy when, in the very first step of their synthesis, it was apparent that a functional group would not be tolerated in their planned Diels–Alder reaction (1 to 2). Thus, they redesigned their synthesis to carry through a benign hydrogen atom in place of an unsuitable –CH₂COOEt group with plans to enact a late-stage C–H functionalization to install the requisite butenolide moiety.^{18,19}

The authors described several attempted C–H functionalization strategies, highlighting a key parameter in the field of C–H functionalization: selectivity. Their study of carbene and nitrene C–H insertions all gave either reactivity at undesired C–H bonds or insertions into C–O and N–O bonds. Because many organic compounds are replete with strong, unactivated C–H bonds, attempts to transform just one can be undermined by considerable collateral damage. In the end, this effort was rescued by a pleasingly simple, directed lithiation of the C14-H bond in compound 4 and a subsequent, *in situ* conversion of the chelated organolithium species 5 to the stable polycyclic bromide 6 (Scheme 2). The Gleason synthesis of virosaine A, which required only three additional steps from compound 6, reminds us of the power that still resides in traditional, directed C–H lithiations. The virosaine A synthesis demonstrates how versatile C–H activation chemistry can be and how the strategy can be relied upon in a high stakes total synthesis because of the broad array of conditions that are available. C–H functionalization enables a flexibility in synthesis that is required when planning a feasible approach to complex molecule synthesis.

In their bold synthesis of the scaparvin family of natural products (13, Scheme 3), the Snyder group highlighted the White oxidation in a complex molecular setting.²⁰ Their insight



Scheme 2 Directed lithiation in a synthesis of virosaine A by Gleason and co-workers.

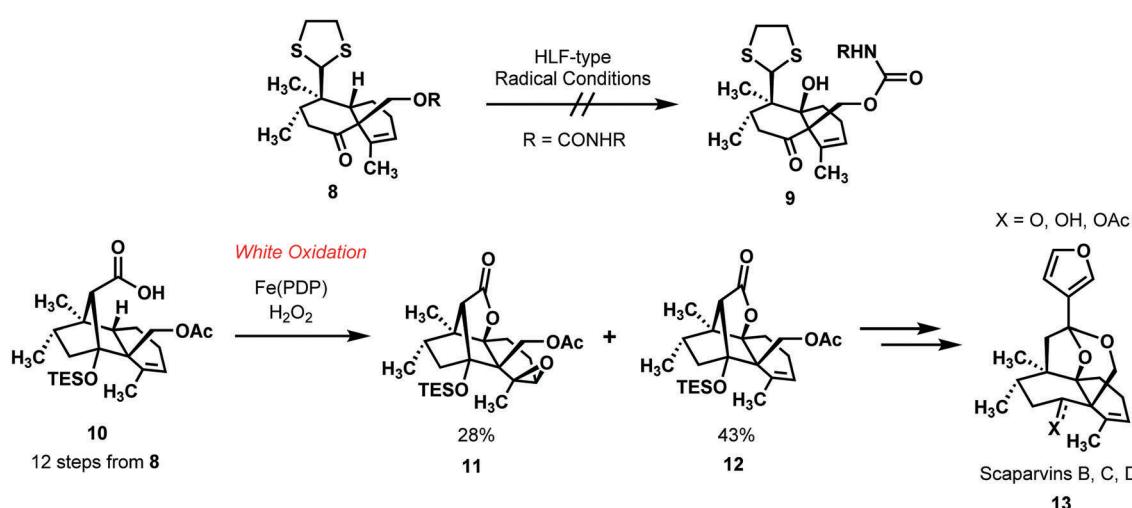
was to make these challenging objectives for synthesis much more tractable by effecting a direct oxidation of the unactivated, tertiary C–H bond. Built into their synthesis was two points of intermediacy at which they could probe the needed C–H oxidation at the position of interest. When the directed C–H oxidation of **8** through Hoffman–Loeffler–Freitag type chemistry failed to achieve the desired oxidation, they reasoned that a directed White oxidation might succeed in the context of the structurally rigid intermediate **10** (Scheme 3). This reasoning was sound, as it was possible to advance compound **10** to compounds **11** and **12** by White's method for forming lactones *via* direct C–H oxidations.¹³ While the alkene in **10** is also epoxidized to some extent in this key step, the desired compound **12** could be isolated in larger relative measure. Some final manipulations furnished the fully functionalized scaparvin framework. This common motif of building flexibility into the synthetic plan is very powerful, and with directed C–H functionalizations this flexibility can be necessary when it is unclear how effective a directing group will be.

In 2016, Brown and Wood disclosed three related approaches to the anti-fungal hippolachnin A, **17**.²¹ Two of these were developed independently in each of their respective research laboratories, and

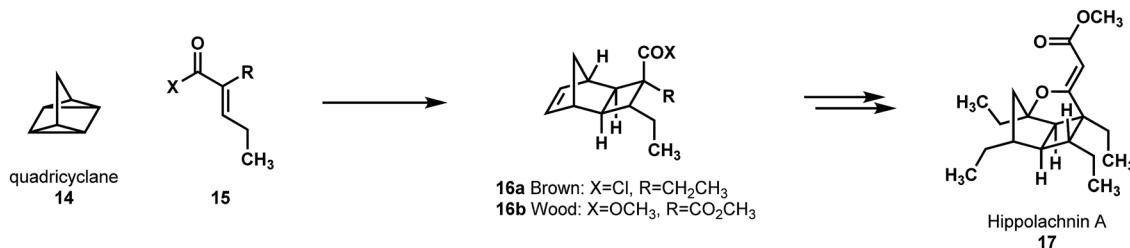
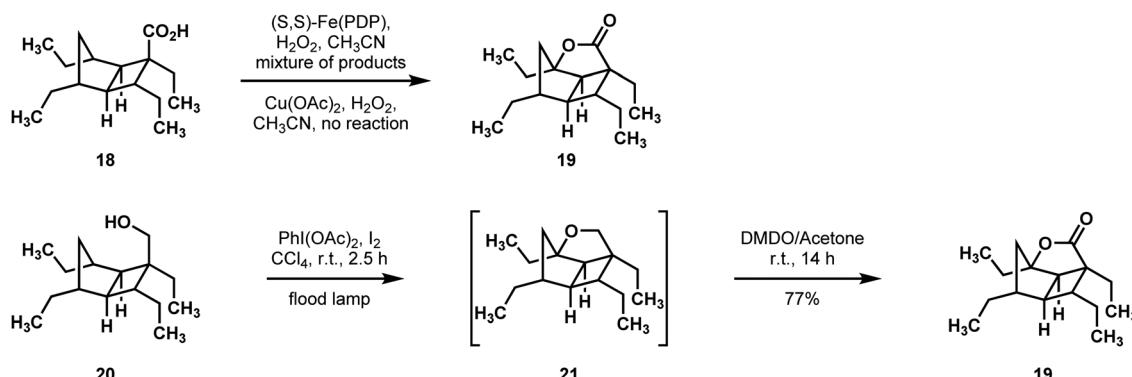
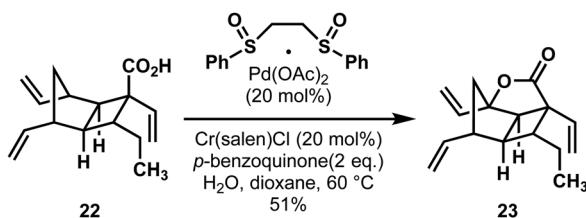
the third was a combination of their two routes. Both routes envisioned forming the cyclobutane core *via* a quadricyclane (**14**) [2π+2σ+2σ] cycloaddition (Scheme 4). To allow the use of this simple, readily available hydrocarbon starting material, both groups designed their syntheses around the installation of the furan ring *via* a late-stage, directed C–H oxidation.

In the Brown route, initial efforts to perform a carboxylic acid-directed C–H oxidation of **18** were unsuccessful (Scheme 5). They report observing either over oxidation products when using White's Fe(PDP) conditions, or quantitative recovery of starting material in the case of a copper reagent. This result prompted a redesign of their oxidation strategy. Simple reduction to alcohol **20** and subsequent exposure to the conditions of Suárez efficiently provided the furan ring, which could be reoxidized with dimethyl dioxirane (DMDO) to provide **19**. Compound **19** could then be elaborated to hippolachnin A (**17**).²²

Interestingly, in the Wood route a similar intermediate was targeted for oxidation: the hexadehydro-analogue to Brown's carboxylic acid intermediate **22** (Scheme 6). By using White's conditions for allylic oxidation,²³ and with the critical addition of superstoichiometric water, the directed oxidation smoothly



Scheme 3 Highlights from the Snyder synthesis of the scaparvin family of natural products.

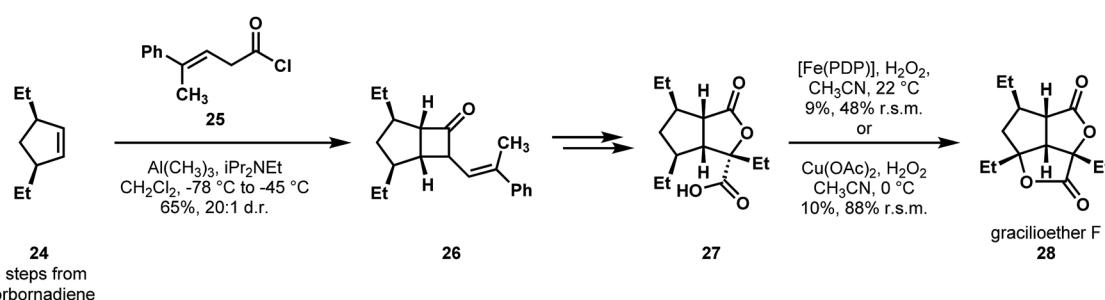
Scheme 4 An approach for synthesizing hippolachnin A (**17**) via the creative and coincident insights of Brown and Wood.Scheme 5 Unsuccessful and successful C–H oxidations in Brown's synthesis of hippolachnin A (**17**).Scheme 6 A carboxyl-directed C–H oxidation in a synthesis of hippolachnin A (**17**) by Wood and coworkers.

provided a similar lactonic intermediate **23**, which could also be elaborated to hippolachnin A.

In both routes, the late-stage C–H oxidation allowed for a dramatic reduction in complexity of the starting materials. Symmetrical quadricyclane could be utilized, and the relative stereochemistry from the resulting cycloaddition could be used to install the necessary functionality.

In 2014, Brown and coworkers described a synthesis of gracilioether F (**28**) reliant on a late-stage C–H oxidation strategy (Scheme 7).²⁴ By deferring the construction of the tertiary lactone for the last step, they could capitalize on latent symmetry and derive the scaffold of the gracilioethers from simple, symmetrical starting materials. The carboxylate moiety would then be used to direct oxidation to break symmetry and form the natural product. Using a *cis*-diethylcyclopentene **24**, derived in three steps from norbornadiene, they developed a Lewis-acid promoted alkenyl ketene–alkene [2+2] cycloaddition, which put most of the carbon skeleton in place (**26**). Several manipulations allowed for the formation of the first lactone with a pendant carboxylic acid (**27**), thus setting the stage for the final C–H oxidation.

They found that traditional oxidants either failed to provide any reactivity, or failed to give selective reactivity, something they attributed to the high steric demand of the concave face tertiary C–H bond. Even White's Fe(PDP) catalyst, which is

Scheme 7 Brown's synthesis of gracilioether F (**28**) from symmetrical cyclopentene (**24**) and featuring a terminal White lactonization.

known to perform carboxyl-directed C–H oxidation,²⁵ only gave 9% yield with significant loss of starting material. They found that catalytic copper acetate with hydrogen peroxide also gave low yields for the desired transformation, but loss of starting material could be avoided by performing the reaction at low temperature. While this last step provided gracilioether F in a modest yield of 10%, the synthesis afforded the natural product in only eight linear steps from the symmetrical norbornadiene.

Maimone's impressive syntheses of *sec*-prezizaane sesquiterpenoids refuse to be placed into a single category in this review. They combine free radical transformations (*i.e.* HLF/Suárez reactions), directed sp³ C–H oxidations and strain-release C–H/C–C oxidations. The syntheses truly highlight the richness and depth of C–H functionalization chemistry.

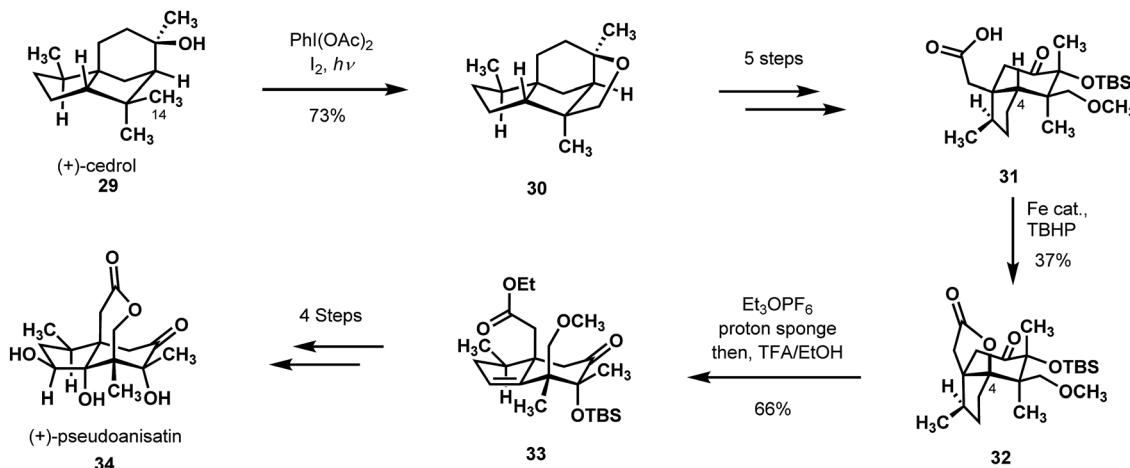
The site-selective oxidations and rearrangements of the feedstock chemical (+)-cedrol (29) by Maimone and coworkers are at the heart of their remarkable syntheses of three *sec*-prezizaane sesquiterpenoids, which originate from the *Illicium* species of plants (Scheme 8).²⁶ The Maimone group creatively utilized the known literature of cedrane C–H oxidation chemistry in combination with skeletal rearrangements to shape their path from the terpene starting material to the targeted natural product family.^{27–29} A Suárez reaction, common to the outset of each synthesis, constructed a strained tetrahydrofuran ring by radical C–H activation of the proximal C-14 methyl group (29 → 30). Also common to the syntheses was the subsequent strain-releasing cleavage of this ring to leave in place oxidation on the desired methyl unit. After this heterocycle was constructed and cleaved, the synthetic strategies towards the *Illicium*-derived sesquiterpenoids diverged.

Specific to their initial synthesis of (+)-pseudoanisatin (34), the researchers performed an α -ketol ring expansion in the wake of an intriguing oxidative α -CH acetoxylation to deliver intermediate 31.³⁰ Intermediate 31 was then poised for a White oxidation of the distal tertiary ring junction C-4 methine.¹³ Uniquely, the ultimate accomplishment of this C–H functionalization was not to construct a γ -lactone, but rather to leave in place an unsaturation after Meerwein alkylation/base elimination of the ester (32 to 33).

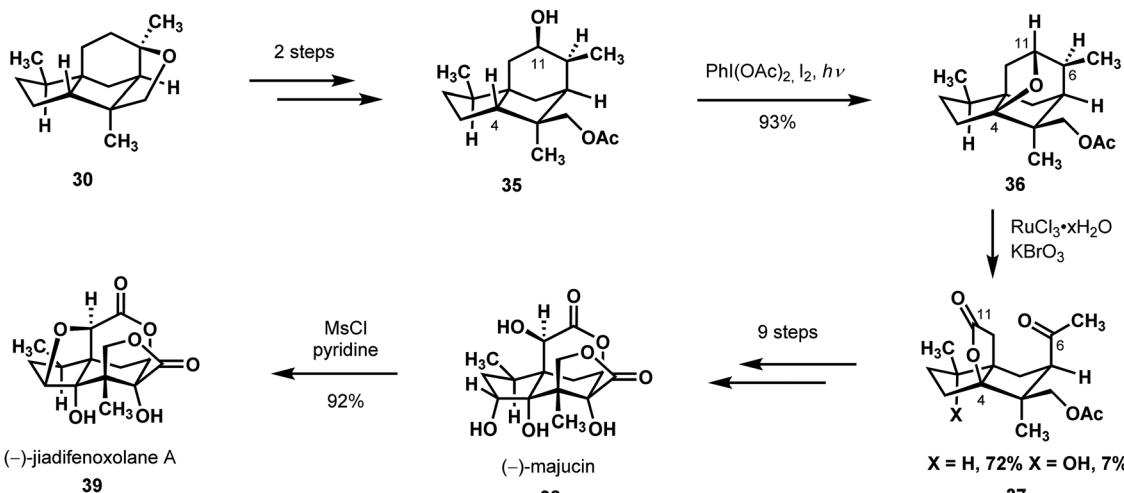
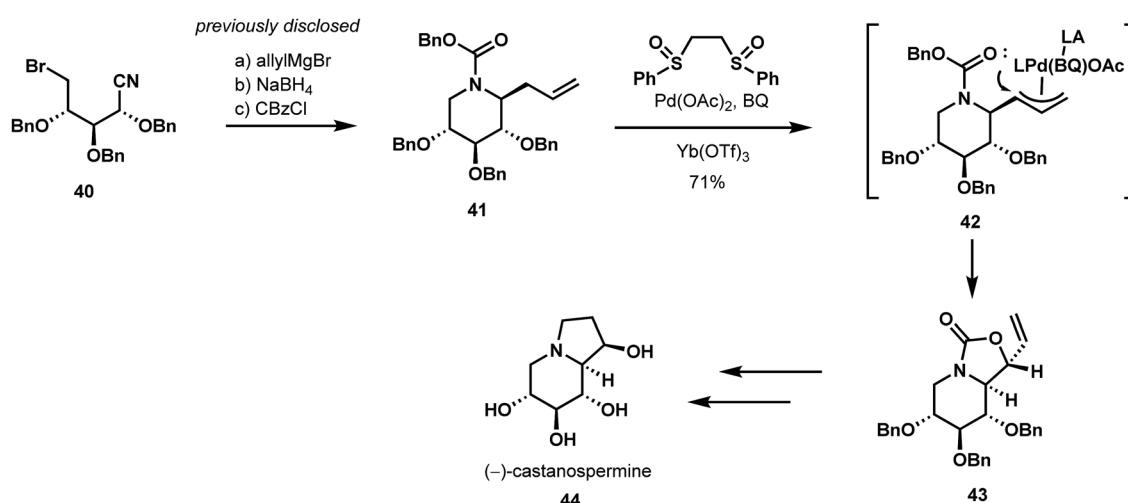
After some manipulation of the residual alkene, (+)-pseudoanisatin (34) was revealed after an expedient 12 steps from (+)-cedrol.

In a subsequent publication, the Maimone group expanded on their manipulations of (+)-cedrol (29) to deliver two additional members from the *Illicium* sesquiterpenes: (−)-majucin (38) and (−)-jiadifenoxolane A (39) (Scheme 9).³¹ Rather than seek new C–H functionalizations to morph the core of (+)-cedrol into their desired goal, Maimone and coworkers had the insight to leverage strain release to activate otherwise unreactive bonds. From the previously used, strained THF-containing intermediate (30), they performed a similar cleavage by elimination, after which they transposed the parent native oxidation over by one carbon to manipulate the C-4 methine. A Suárez reaction furnished the C-4 oxidized moiety 36 by way of an ether bond formation between C-4 and the C-11-bound oxygen.²⁸ The researchers recognized that the additional strain on the THF ring in 36 would heighten reactivity towards chemical transformations that would ease the molecule's contorted state. Thus, upon exposure to *in situ* generated RuO₄, compound 36 could lower its energy in the course of an oxidative cleavage of the C6–C11 bond to give keto lactone 37 with its [3.3.3] propellane architecture. This interesting oxidative cleavage process, which is founded on prior work by Waegell and Baggaley on the cedrane core,^{27,29} exhibited high selectivity (10 : 1 ratio of oxidation products favoring the desired compound) and efficiency. After a sequence of redox manipulations and an α -ketol rearrangement, they completed their syntheses of (−)-majucin (38) and (−)-jiadifenoxolane A (39) in 14 and 15 steps, respectively.

C–H functionalization has also had an impact on carbohydrate synthesis. Malik and Jarosz have long had an interest in sugar mimics, and, in 2014,³² they disclosed a diastereospecific, allylic White oxidation,³³ which made further use of their previously published methodology for preparing piperidine-containing iminosugar precursors (Scheme 10).³⁴ The researchers explored several reagents to achieve the desired allylic oxidation and found the White catalyst system to be singularly successful at effecting the desired transformation (41 → 43). In the course of



Scheme 8 C–H functionalization in Maimone's synthesis of (+)-pseudoanisatin featuring a Suárez reaction and White oxidation.

Scheme 9 Use of the Suárez reaction and an oxidative C–C cleavage in *Illicium* sesquiterpenes synthesis from (+)-cedrol.

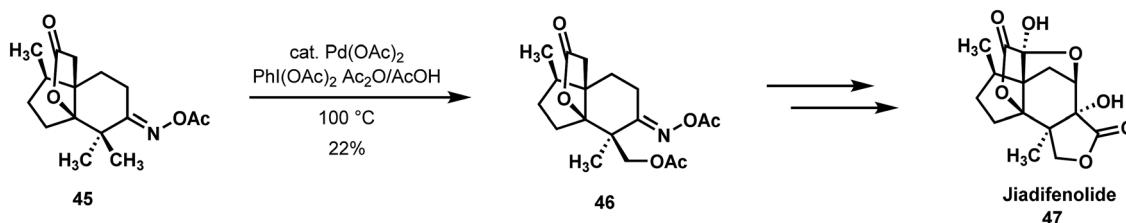
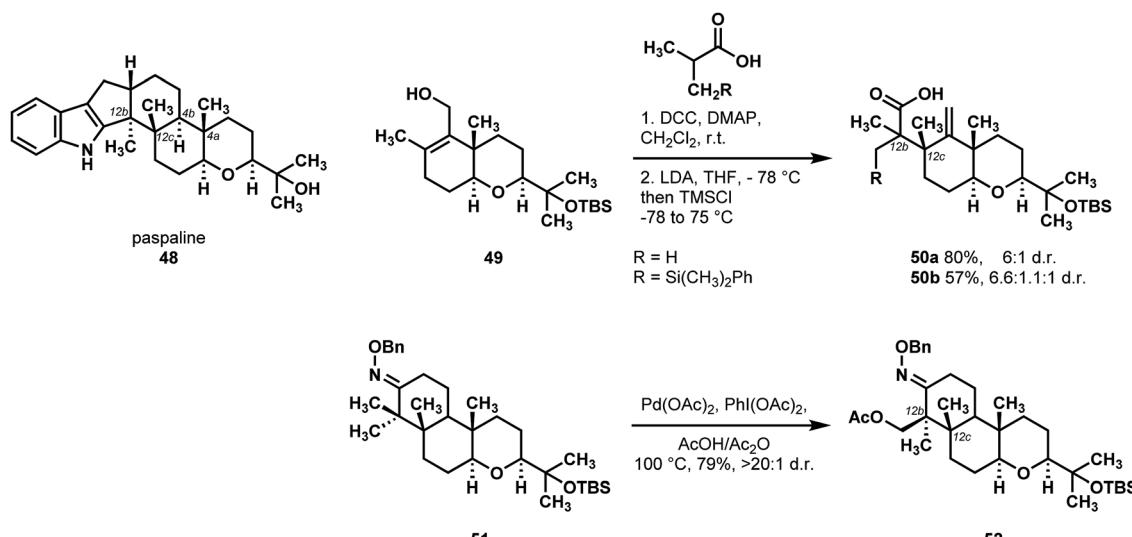
Scheme 10 Diastereoselective allylic C–H oxidation in the synthesis of iminosugar (-)-castanospermine.

this allylic activation, a pendant Cbz carbamate function intramolecularly traps the putative pi-allyl palladium species **42** arising from the catalytic C–H activation. This neighboring-group participation mechanism effectively controls the stereochemical outcome of the key allylic oxidation. Subsequent hydroboration of the vinyl unit, hydrolysis of the carbamate, and a final Mitsunobu cyclization completed their stereospecific synthesis (*–*)-castanospermine (**44**) from D-xylene.

In the Sorensen approach to jiadifenolide (**48**) (Scheme 11), a Sanford oxidation was utilized to differentiate a *gem*-dimethyl unit by means of an *O*-acyloxime directing group.^{12,35} This oxidation allowed for the eventual incorporation of a ring fused γ -lactone ring. Given the prevalence of geminal methyl groups and their oxidized derivatives in natural products and bioactive compounds, symmetry-breaking C–H functionalizations enrich the creative process of retrosynthetic analysis and can reduce the number of steps required to bring a target-directed synthesis to fruition. While the pivotal Sanford oxidation of functionalized

'propellane' **45** afforded a 1:1 mixture of diastereoisomers and only a 22% yield of the desired compound **46**, it was straightforward to introduce the germinal methyl groups at an early stage of the synthesis and carry them through the middle phase of the effort. From the vantage point of **46**, jiadifenolide could be reached without incident.

The Johnson laboratory reported a synthesis of paspaline **48** that utilized C–H functionalization as an enabling reaction (Scheme 12).³⁶ They initially envisioned forming the central rings of paspaline *via* a cationic double cyclization terminating with an indole Friedel–Crafts alkylation. While they were successful in synthesizing the tetrahydropyran ring in good yield and selectivity, they found that they could not append the indole in preparation for the cyclization. Instead, they sought to build the rest of paspaline off of the tetrahydropyran. Their initial assessment of the paspaline family identified the three quaternary centers in the structure (C12b, C12c, and C4a) as challenging elements, and they found in the course of their

Scheme 11 *gem*-Dimethyl differentiation by Sanford oxidation in the Sorensen synthesis of jiadifenolide.Scheme 12 The C–H oxidation strategy to differentiate geminal methyl groups employed by Johnson en route to paspaline (**48**).

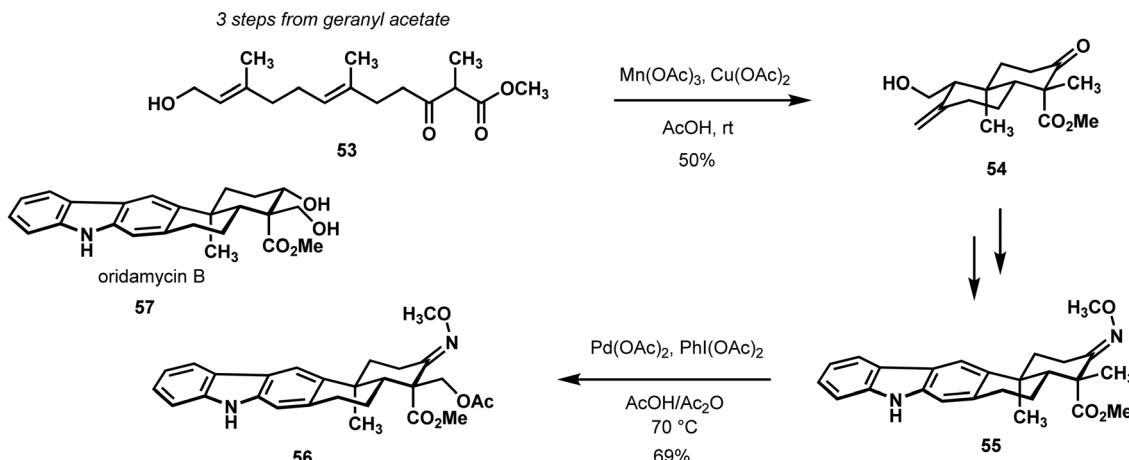
studies that setting the 1,3-*syn*-dixial relationship between two methyl groups (C4a and C12c) was beset with difficulty. They discovered that an Ireland-Claisen rearrangement (**49** → **50a,b**) could set the desired configuration of the methyl-bearing stereocenter at C12c, however only the isobutyrate and the 3-silylsubstituted isobutyrate esters underwent the rearrangement. Unfortunately, the silylmethyl moiety inhibited subsequent steps, which put Johnson and coworkers in a position where they needed to desymmetrize the two methyl groups derived from isobutyric acid to establish the C12b stereocenter. They found that by closing the cyclohexanone and forming the *O*-benzyloxime **51**, they could perform a directed C–H acetoxylation with good diastereoselectivity.³⁷ They attributed this high selectivity to the large degree of coplanarity between the desired equatorial methyl and the oxime directing element. The differentiation of the germinal methyl groups in **51** by C–H acetoxylation afforded **52** and permitted the final ascent to paspaline (**48**).

In two reports published in 2015, total syntheses of the indolosquiterpene oridamycin B (**57**) (Scheme 13) were described, and they both employed late stage oxime-directed Sanford oxidations¹² to install a requisite hydroxyl moiety found in the final product.^{38,39} Both the Li group and Trotta took advantage of a radical cyclization to construct the *trans*-decalin system in a few steps from geranyl acetate (**53** → **54**). Recognition that a directed methyl C–H oxidation of **55** could

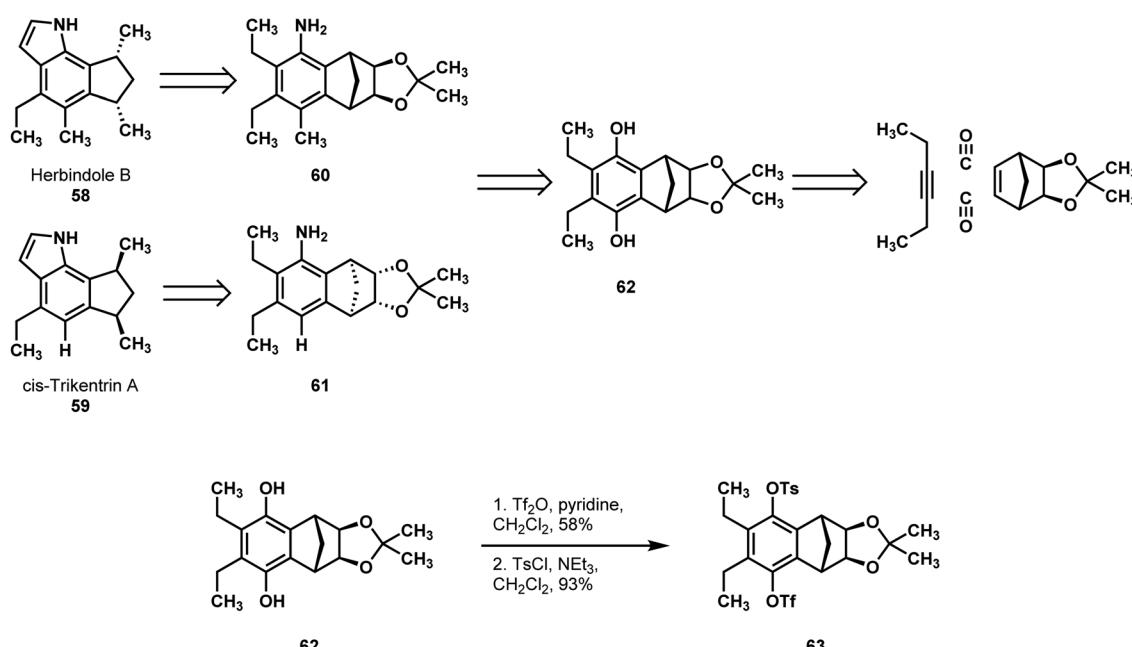
install the final point of oxidation in oridamycin B enabled the Li group to prepare several members of the xiamycin and oridamycin family using a unified strategy. Members of these natural product families have identical skeletal architecture, but they differ in some points of oxygenation and stereochemistry. Thus, a late stage C–H oxidation was a key transformation that allowed oridamycin B to be included in the possible targets under a unified synthetic strategy.

In 2016, Sarpong and Yu reported formal syntheses of the pseudo-enantiomeric natural products herbindole B (**58**) and *cis*-trikentrin A (**59**) (Scheme 14).⁴⁰ At the heart of their retrosynthetic analysis was the perception that an indole nucleus could arise in the course of an intramolecular C–H amination of an *ortho*-ethyl aniline. Thus, they could reduce the two natural products to the same symmetrical (*meso*) hydroquinone **62**, which they synthesized via a unique [2+2+1+1] cycloaddition process. Enantiotopic group-selective and stepwise transformations of the phenolic hydroxyls in **62** provided effective syntheses of compounds **60** and **61** (only racemic route shown).

Following this, group-selective cross coupling on compound **63** yielded the two aniline precursors, **60** and **61**. Rhodium catalysis to effect nitrenoid insertion on the pendant methyl group from the aryl azide (**64**) (Scheme 15) showed less promising yields compared to similar transformations in the literature.⁴¹ Instead, they achieved better results in palladium-catalyzed C–H aminations, founded on Yu's precedent.⁴² Not only did these



Scheme 13 Oridamycin B synthesis by directed C–H acetoxylation.

Scheme 14 Sarpong and Yu's synthesis of a central intermediate toward herbindole B (58) and *cis*-trikentrin A (59).

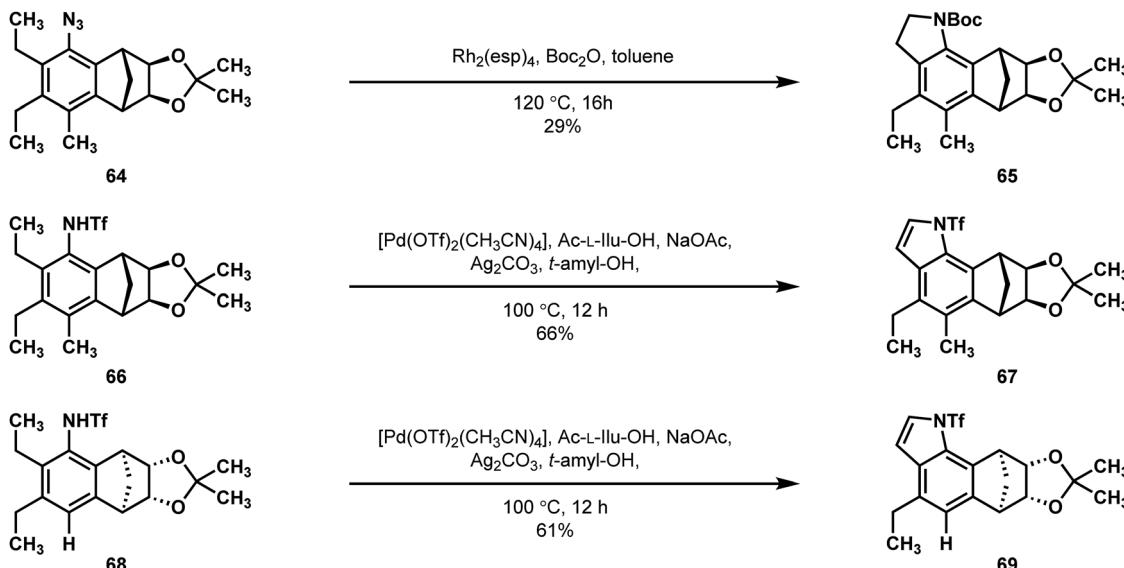
reactions provide the desired C–N bonds in higher yields, but they also induced formation of the indole C–C double bonds directly, providing compounds 67 and 69. The authors suggested that this was likely due to rapid oxidation of the indoline to indole in the reaction vessel. These two intermediates then underwent protecting group conversions to achieve formal syntheses of each natural product.

The syntheses of these related natural products highlight a recurring theme in applications of C–H functionalization to total synthesis: the simplicity and symmetry that results from a ‘C–H retrosynthetic step’. In this case, the authors could derive the left half of the molecule from simple and symmetrical 3-hexyne, and, from this powerful insight, generate the 4-ethylindole core. This reduction in complexity then allowed them to turn to a single synthetic transformation that could

quickly forge the benzene nucleus from readily available starting materials.

In 2015 Baudoin reported a strategy which they successfully used to prepare members of the aeruginosin family (70, Fig. 1), which are naturally occurring serine-protease inhibitors.⁴³ They recognized that a C–H functionalization strategy would offer an expedient route to prepare two of the four amino acid derived subunits, the octahydroindole unit 77 and the hydroxypyphenyllactic acid unit 79. The envisioned methyl C–H functionalization/ring closure would form the 5-membered ring of the octahydroindole unit, and a directed methyl C–H functionalization/cross-coupling would form the hydroxypyphenyllactic acid unit from a derivative of lactic acid.

To form the octahydroindole precursor 74, they subjected a chiral aminoalcohol derivative of L-alanine 72 to allylic bromide



Scheme 15 C–H amination methods explored by Sarpong and Yu in their syntheses of herbindole B (**58**) and *cis*-trikentrin A (**59**).

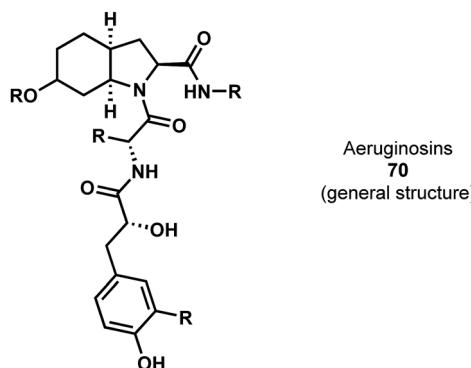
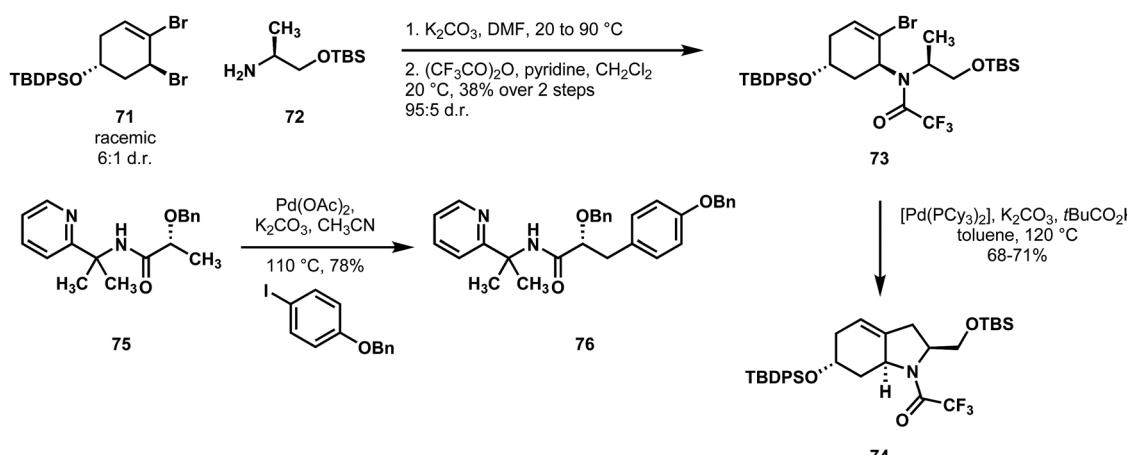


Fig. 1 The general molecular structure of the aeruginosin family of natural products.

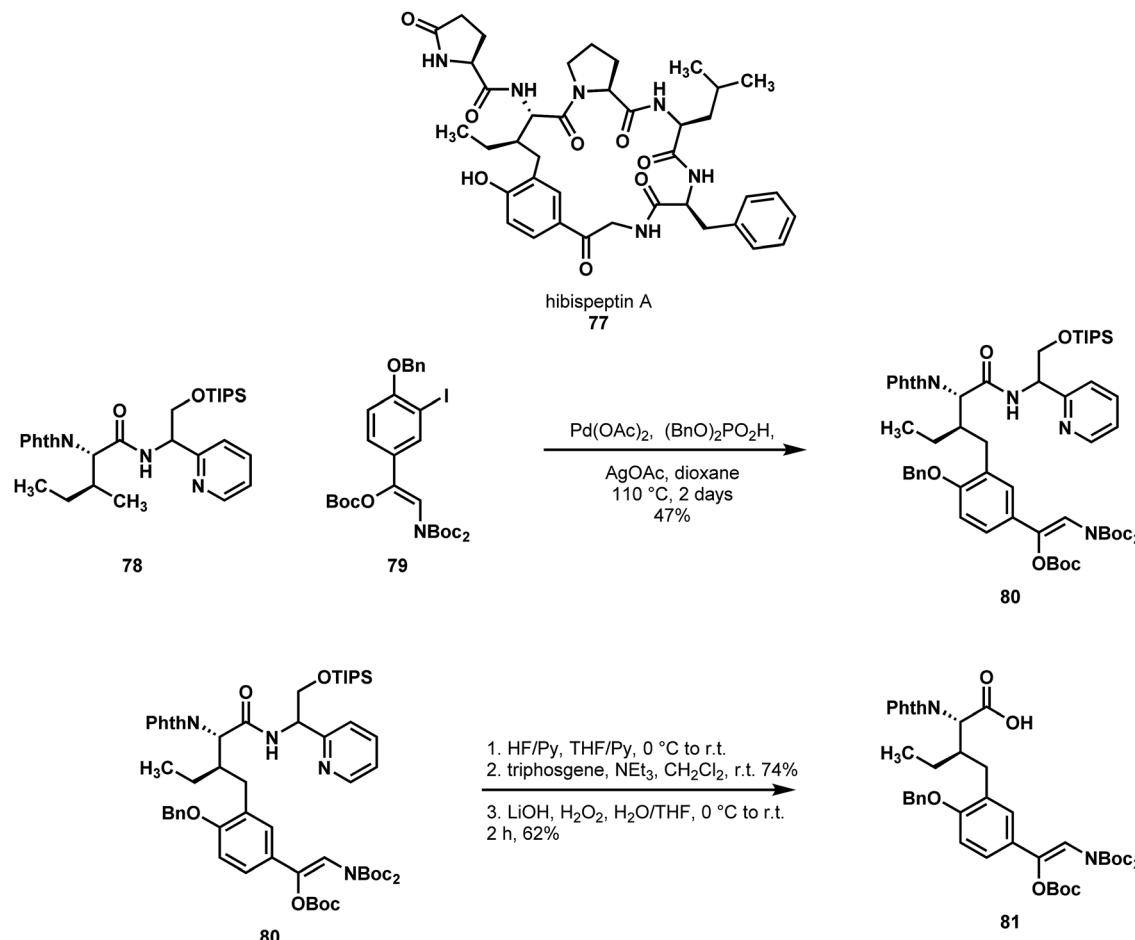
71 under ionizing conditions, presumably proceeding through a diastereoselective trapping of a *meso*-isomeric allyl cation intermediate. Following protection of the amine, a tricyclohexylphosphine

palladium complex catalyzed C(sp³)–H coupling of **73** to close the ring.⁴⁴ To form the other subunit, Baudoine and coworkers synthesized the *O*-benzyl-PIP-amide of lactic acid **75**, and, without any loss of enantiopurity, they were able to cross couple this derivative with several iodoarenes (only one shown) to provide **76**.^{45,46} It is worth noting that despite the sensitivity of lactic acid stereocenters, neither the C–H functionalization nor the directing group removal epimerized the alpha-stereocenter. With these two subunits in hand, Baudoine and coworkers synthesized two members of the aeruginosin family, with the expectation that they would be able to further diversify this family of compounds through analogue synthesis (Scheme 16).

Chen reported the first synthesis of the pseudohexameric cyclic peptide hibispeptin A (**77**) using C–H functionalization (Scheme 17).⁴⁷ Hibispeptin A features a C–C crosslink between an isoleucine and a 4-hydroxy phenacylamine, which they imagined could be established by C–H cross-coupling using isoleucine as a starting material. To this end, they explored a



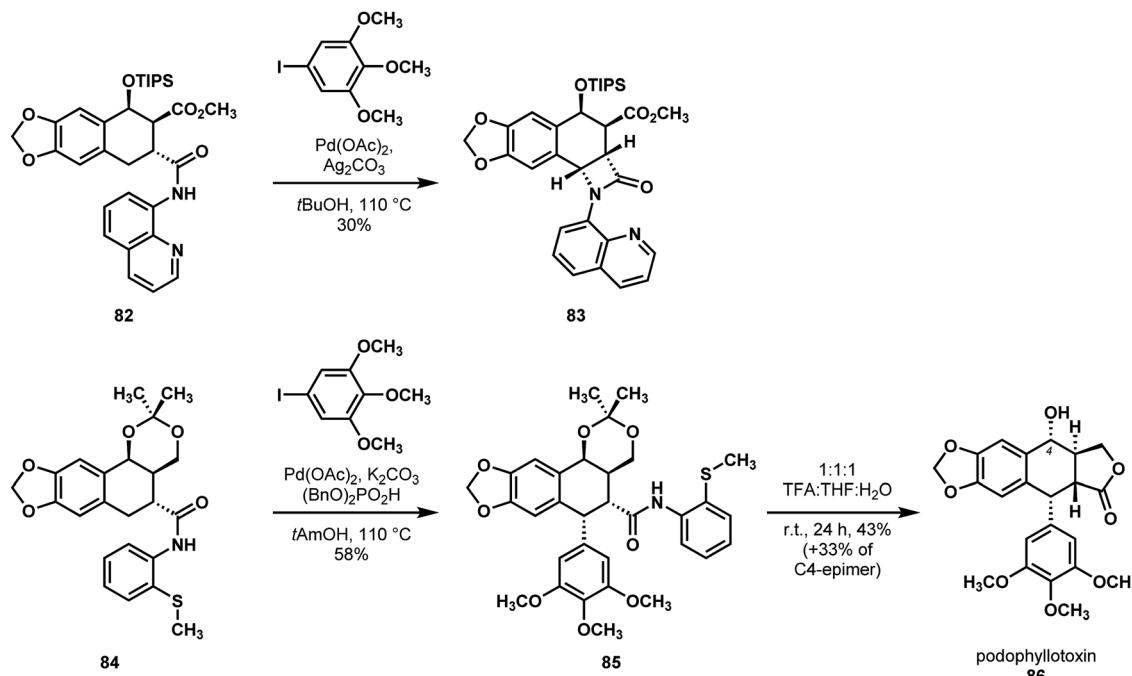
Scheme 16 Two different C(sp³)–H functionalization reactions employed by Baudoine in a synthesis of the aeruginosins (**70**).



Scheme 17 A C–H arylation reaction employed by Chen in a synthesis of hibispeptin A (**77**) and subsequent removal of the directing element.

variety of directing groups to functionalize the γ -methyl of isoleucine derivatives with aryl iodides. In the reaction development stage, the researchers discovered that sterically hindered aryl iodides gave poor conversions. This was problematic due to the hindered nature of the proposed aryl iodide to be used in the total synthesis. Additionally, the pyridyl methylamine (PM) class of directing groups were particularly efficient, but the most effective group, pyridyl ethylamine, was very resistant to the cleavage conditions.⁴⁸ To circumvent this issue, Chen and coworkers designed a new directing group in this class (PR), which they hypothesized could be induced to undergo an *N*-to-*O* acyl shift following desilylation, and that this would aid in cleavage. By installing their PR directing group on isoleucine (**78**, as a mixture of directing group stereoisomers), they found they could readily perform C–H arylation in good yield to provide **80**. Unfortunately, *N*-to-*O* acyl transfer was not as efficient as expected, and the resultant ester was not readily transesterified. However, the directing group could be cleaved by silyl group removal, closure of the oxazolidone using triphosgene, and finally hydrolysis to the free acid **81**. Following this, the remaining 4 amino acids were installed, and macrocyclization afforded hibispeptin A (**77**). Chen's efficient strategy allowed for the direct functionalization of a natural and available amino acid.

In 2014, the Maimone laboratory reported a synthesis of podophyllotoxin (**86**, Scheme 18), a bioactive natural product and precursor to several drug molecules.⁴⁹ They envisioned installing the trimethoxyarene *via* a directed C–H cross-coupling, allowing them to focus on the tetrahydronaphthalene core (**82** or **84**), which they were able to readily synthesize *via* a benzocyclobutane-derived *ortho*-quinonemethane [4+2] cycloaddition. However, their initial attempts to perform the directed C–H arylation surprisingly resulted in C–H amination to provide the beta-lactam **83**.⁴⁶ After examining the ORTEP of the C–H insertion palladium complex, they hypothesized that the cyclohexane conformation was inhibiting the desired cross-coupling with the aryl iodide. Reduction of the methyl ester and acetonide formation rigidified the cyclohexane **84**, and they were able to show by X-ray crystallography that this structural modification significantly altered the conformation of the isolated palladacycle. Compound **84** participated in the desired C–H arylation, and, after some optimization, the researchers were able to obtain acceptable yields of **88**. In a single step, compound **85** was converted to the target molecule. An impressively concise synthesis of podophyllotoxin (**86**) was the outcome of this C–H functionalization dependent strategy, which Maimone and coworkers leveraged to access a variety of otherwise difficult to obtain analogues.



Scheme 18 Successful and unsuccessful attempts at a key C–H arylation in Maimone's synthesis of podophyllotoxin (**86**).

Baran comments on how syntheses of substituted cyclobutanes are becoming increasingly reliant on C–H functionalization strategies.⁵⁰ This can likely be traced to the difficulty of their formation by other means. A tetrasubstituted cyclobutane can be imagined to arise in the course of an intermolecular pairing of two substituted alkenes by a photochemical [2+2] cycloaddition. However, in a practical sense, this is largely only a useful disconnection if the cyclobutane is symmetrical. With cyclobutanes bearing four different substituent groups, chaotic product mixtures can arise by homo [2+2]- and hetero [2+2]-dimerizations, unselective regiochemical outcomes (*i.e.* head-to-tail or head-to-head couplings), in addition to a complete lack of diastereoselectivity in the establishment of relative stereochemical relationships. The [2+2] photocycloaddition process is biogenetically relevant and a key element of successful syntheses of differentially substituted cyclobutanes; however, the method is by no means a panacea for these challenging carbocycles. C–H functionalization reactions, which add complexity to simpler cyclobutane starting materials, provide powerful options in efforts to synthesize substituted cyclobutanes with high margins of regio- and stereocontrol.⁵⁰

The Baran laboratory had achieved success in the area of C–H functionalization of cyclobutanes in their prior synthesis of piperarborene B (**88**, Fig. 2).¹⁰ In the wake of this effort, they undertook a synthesis of the analogous, yet more challenging target, pipercyclobutanamide A (**87**).⁵¹ Since the cyclobutane core of compound **87** possesses four different substituent groups, Baran and coworkers could not simply transfer the same retrosynthetic logic underlying their earlier synthesis of **88**. Instead, they found it necessary to explore the methods by which a styrenyl moiety could be installed through C–H functionalization. Unlike

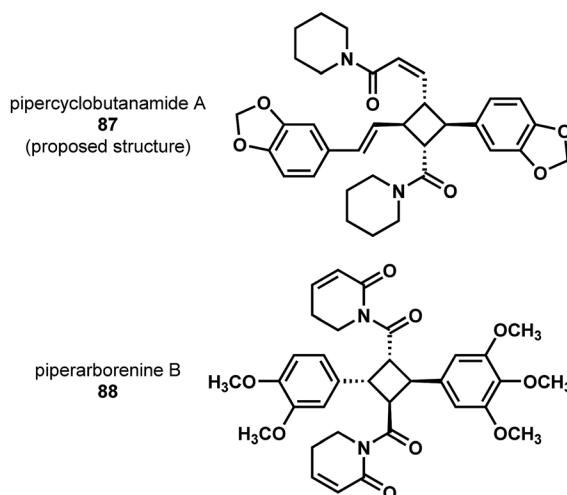
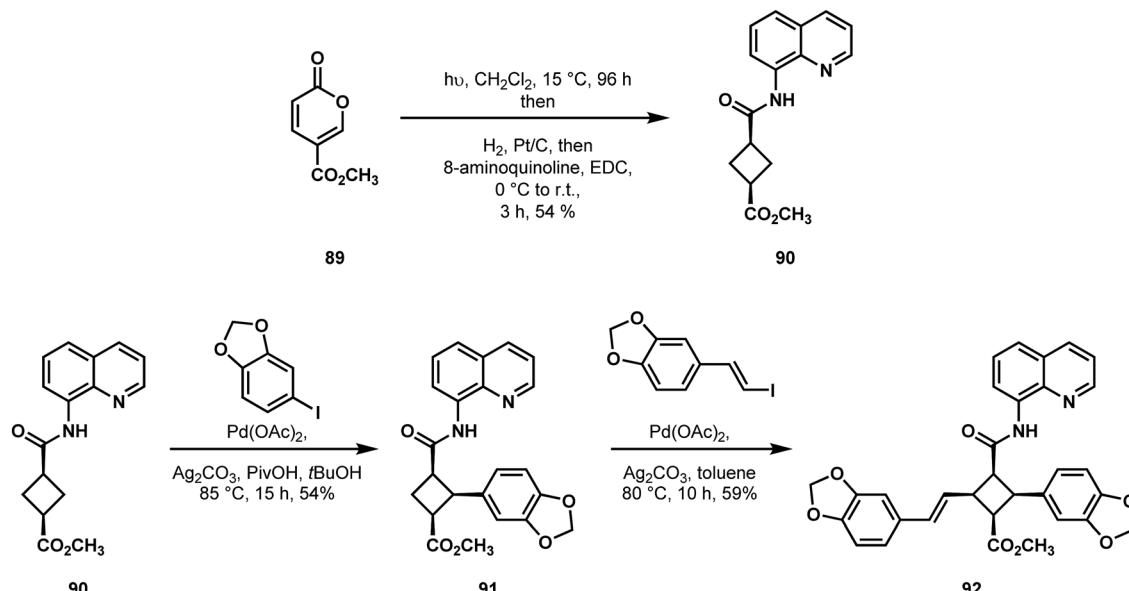


Fig. 2 The molecular structure of pipercyclobutanamide (**87**, proposed) and piperarborene B (**88**).

the case of piperarborene B (**88**), where amide epimerization needed to precede the second C–H cross-coupling event, cyclobutane **90** underwent rapid two-fold C–H vinylation, something the authors attributed to the reduced steric bulk of the vinyl coupling partner relative to aryl. With this knowledge in hand, they first performed a directed C–H arylation on their cyclobutane derivative (**90** → **91**, Scheme 19), followed by a directed C–H olefination on compound **91** to yield the all-*cis* functionalized cyclobutane **92**. Stepwise transformations of the carbonyl functionality in **92**, including the pivotal epimerization to the all-*trans* cyclobutane yielded the proposed structure of the natural product. It was at this point that these investigators

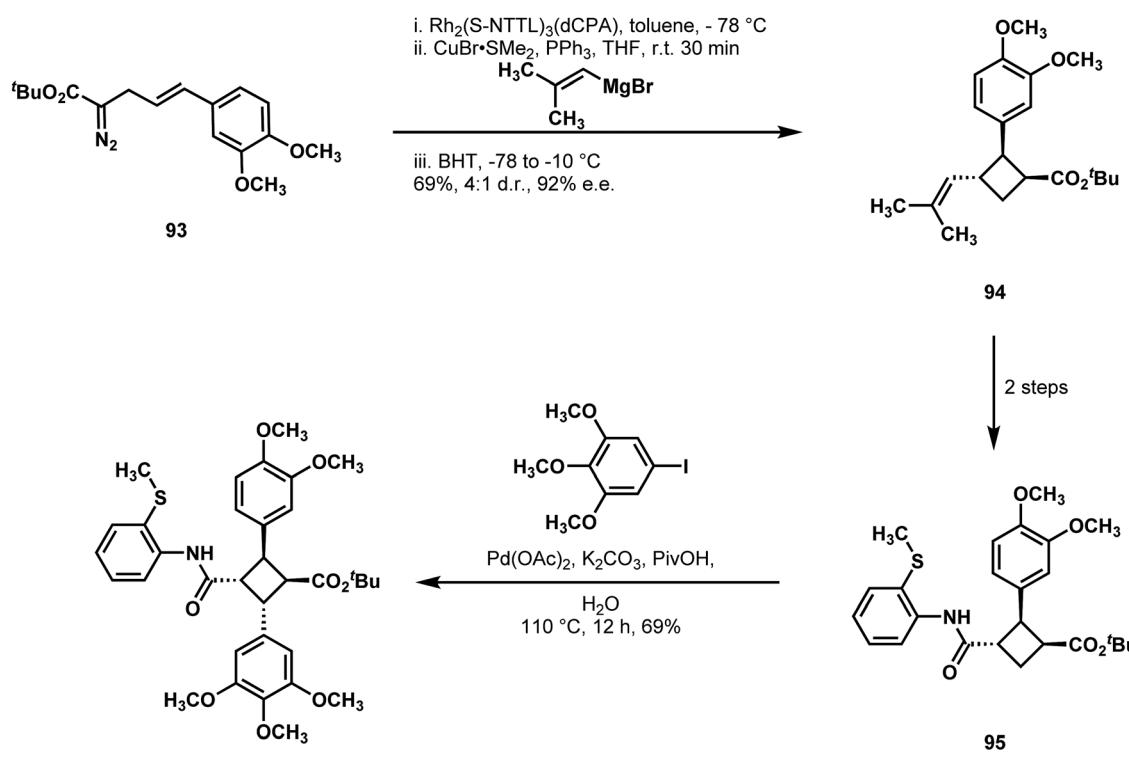


Scheme 19 Baran's synthesis of the proposed structure of pipercyclobutanamide A (**87**) featuring two sequential C–H cross-coupling reactions.

exposed the proposed structure of pipercyclobutanamide A (**87**) to be incorrect and in need of revision.

Fox and coworkers also targeted piperarborene B (**88**) with an approach featuring a C–H arylation, although their effort sought an enantioselective entry to the natural product family (Scheme 20).⁵² They recognized the simplifying power of Baran's C–H

disconnections, and thus sought a creative strategy for achieving an enantioselective construction of a key cyclobutane building block. On styrenyl diazoester **93**, they effected an intramolecular cyclopropanation to form an activated [1.1.0]bicyclobutane, which promptly underwent nucleophilic ring opening to afford tri-substituted cyclobutane **94** with good enantiocontrol and a



Scheme 20 An asymmetric synthesis of piperarborene B (**88**) reported by Fox.

useful pattern of relative stereochemistry.⁵³ Stepwise transformations of the alkene substituent in **94** introduced the same directing element that had served Baran's synthesis of **88** and set the stage for the key, *cis*-selective C–H arylation of **95**. This transformation afforded compound **96**, from which piperaborenine B (**88**) could be reached in two additional steps.

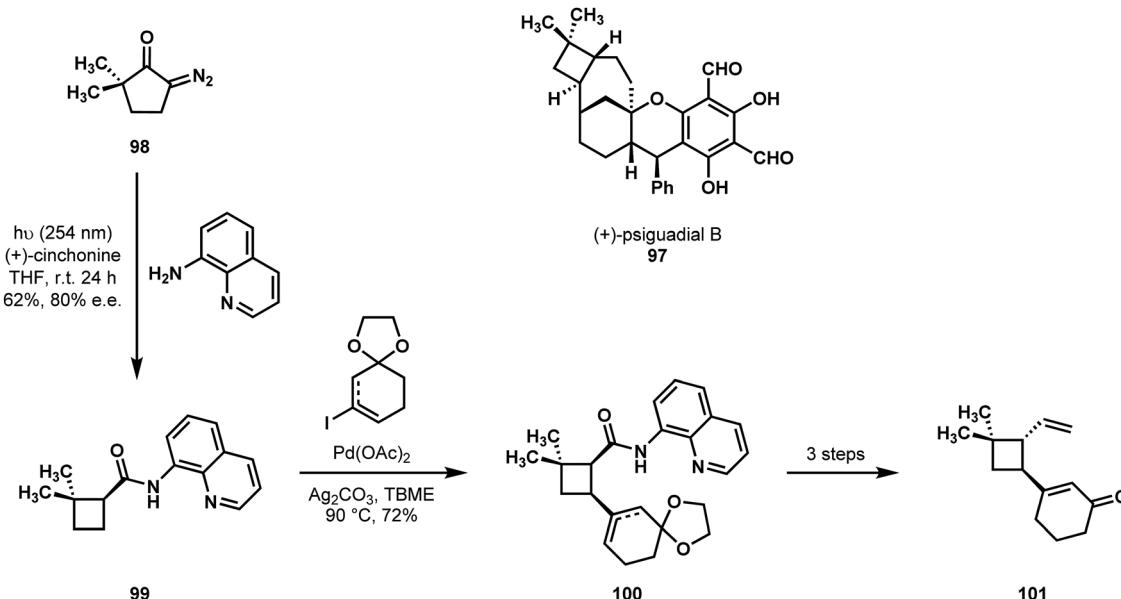
In 2016, Reisman reported a synthesis of psiguadial B (**97**) featuring C–H functionalization logic.⁵⁴ They considered the cyclobutane moiety *trans*-fused to the [4.3.1]bicyclodecane as a central synthetic challenge, and their effort began by establishing the absolute and relative stereochemistry of the cyclobutane structural element. Their achievement offers the creative idea that the cyclobutane frame and the directing group for an eventual C–H cross-coupling could be introduced at once in the course of a Wolff rearrangement/catalytic asymmetric ketene addition. In what is believed to be the first example of a tandem Wolff rearrangement/catalytic asymmetric ketene addition, the irradiation of a solution of compound **98**, 8-aminoquinoline, and 20 mol% of (+)-cinchonine resulted in the formation of trisubstituted acylcyclobutane **99** in 62% yield and with an enantiomer ratio of 90:10. By performing a directed C–H cross-coupling on compound **99**, they generated **100**, and following selective epimerization of the amide α -stereocenter they could gain access to *trans*-1,1,2,3-tetrasubstituted cyclobutane **101** with the appropriate relative stereochemistry, which set the remainder of their synthesis in motion (Scheme 21).

Pu and coworkers reported the isolation and synthesis of scopariusicide A (**102**) from *I. scoparius* (Scheme 22).⁵⁵ They hypothesized that the biogenesis of this natural product might involve a head-to-head [2+2] cycloaddition. In the course of testing this idea, Pu and coworkers explored the cycloaddition between the precursor lactone **103**, itself synthesized in one

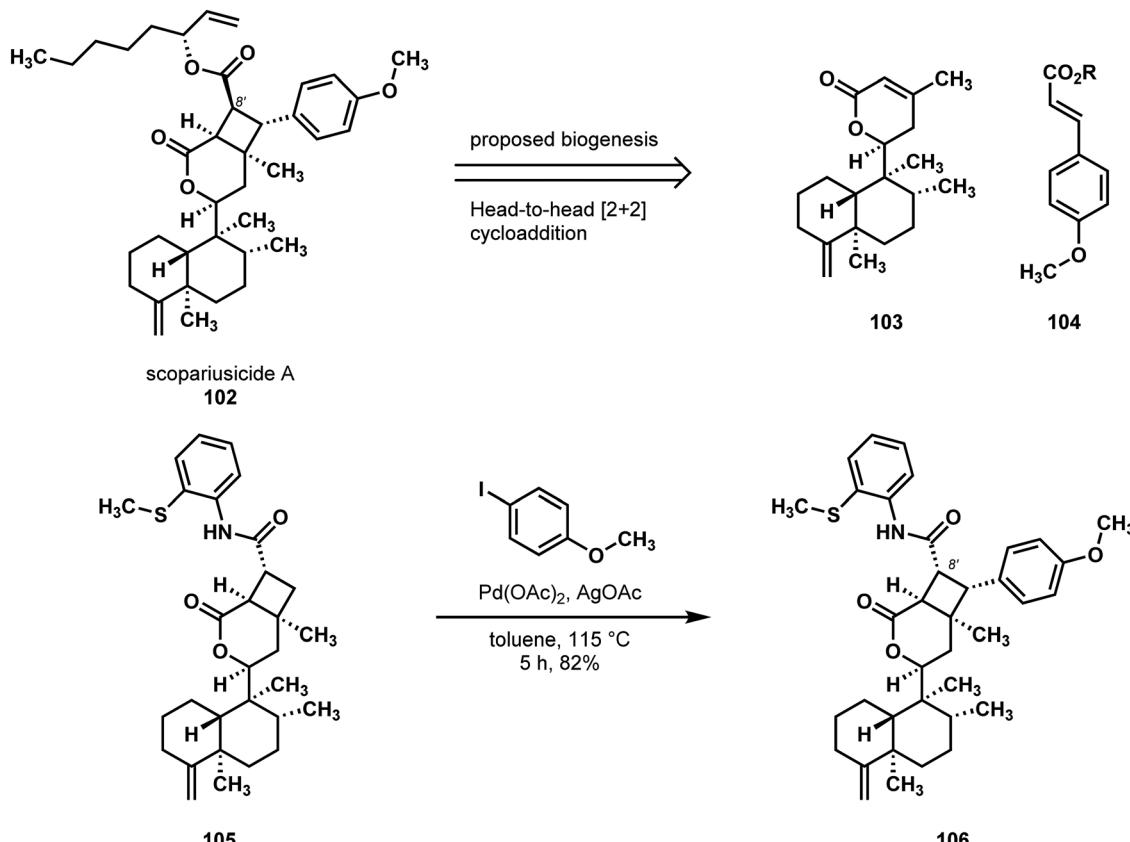
step from a readily available isolate of *I. scoparius*, and methyl 4-methoxycinnamate **104**. However, they observed no product formation, which they attributed to deleterious, non-radiative relaxation pathways. Evidence for these proposed pathways involved re-isolation of starting material, which showed *cis*–*trans* isomerization of **104**, and endo-/exo-cyclic olefin isomerization of **103**.

Still seeking access to the natural product, Pu and coworkers imagined that if a head-to-head cycloaddition were possible with a suitable acrylate, perhaps then a directed C–H functionalization could install the remaining arene ring. Notably, since the scopariusicide (**102**) conceals a *trans*-cinnamate moiety, the planned cycloaddition would need to give the "wrong" stereocenter at C8' to ensure the desired stereochemical outcome in the C–H arylation step. As matters transpired, the [2+2] cycloaddition of α,β -unsaturated lactone **103** with methyl acrylate was efficient and displayed high head-to-head selectivity, as well as high selectivity for the unnatural configuration at C8'. The straightforward conversion of this cycloadduct to compound **105** enabled the key, directed C–H arylation and construction of tetrasubstituted cyclobutane **106**. In the wake of a late-stage, thermodynamically controlled epimerization at C8', a hydrolytic cleavage reaction could remove the amide directing element that was essential for the C–H functionalization step.

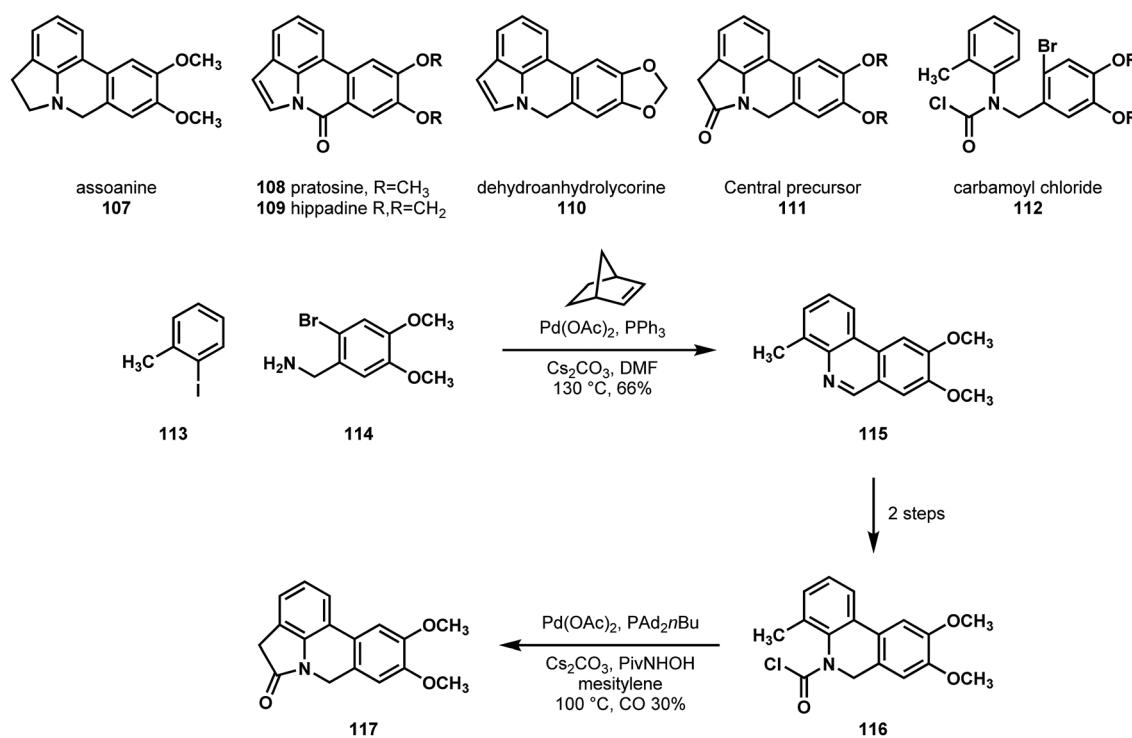
Takemoto reported a route to the pyrrolophenanthridine alkaloids **107**–**110** (Scheme 23).⁵⁶ These alkaloids had been the subject of many synthetic endeavors, and in fact had already been explored in the context of C–H functionalization. The central biaryl linkage had been studied in stoichiometric,⁵⁷ catalytic,⁵⁸ and metal-free⁵⁹ intramolecular C–H functionalization cross-coupling reactions. Takemoto envisioned an even bolder approach to a central precursor **111** which could give rise to assoanine, pratosine, hippadine, and dehydroanhydrolycorine.



Scheme 21 Reisman's entry into an asymmetric synthesis of (+)-psiguadial B (**97**), featuring asymmetric ketene capture of a Wolff rearrangement and subsequent C–H functionalization.



Scheme 22 The hypothesized biogenesis of scopariuside A (**102**) and a C–H arylation strategy employed by Pu in the synthesis of **102**.



Scheme 23 The molecular structure of assoanine (**107**), pratosine (**108**), and hippadine (**109**), and Takemoto's synthetic strategy featuring a C–H acylation.

They imagined that both the biaryl bond and the lactam could arise from a two-fold C–H functionalization of an *N,N*-difunctionalized carbamoyl chloride **112**. They set about to explore this strategy, with the halogen atom either on the benzylic fragment or the aryl ring (not shown). However, they found that several other reaction pathways were competitive at the temperature necessary to effect C(sp³)–H functionalization: a variety of side products arose from oxidative addition into both C–X bonds.

Undaunted, Takemoto and coworkers sought instead to perform the two functionalizations stepwise. Using the benzylic amine **114** and *o*-iodotoluene **113**, they performed a Catellani reaction, forging the entire phenanthridine core **115** in a single step.⁶⁰ Although this very bond formation had been explored in previous syntheses, those were largely intramolecular. Following this, they formed the carbamoyl chloride **116** and effected methyl C–H functionalization to complete the central intermediate **117**.⁶¹ Intermediate **117** could then be converted into assoanine and pratosine in a few steps. This strategy also enabled access to hippadine and dehydroanhydrolycorine.

Directed C(sp²)–H functionalization of benzene derivatives

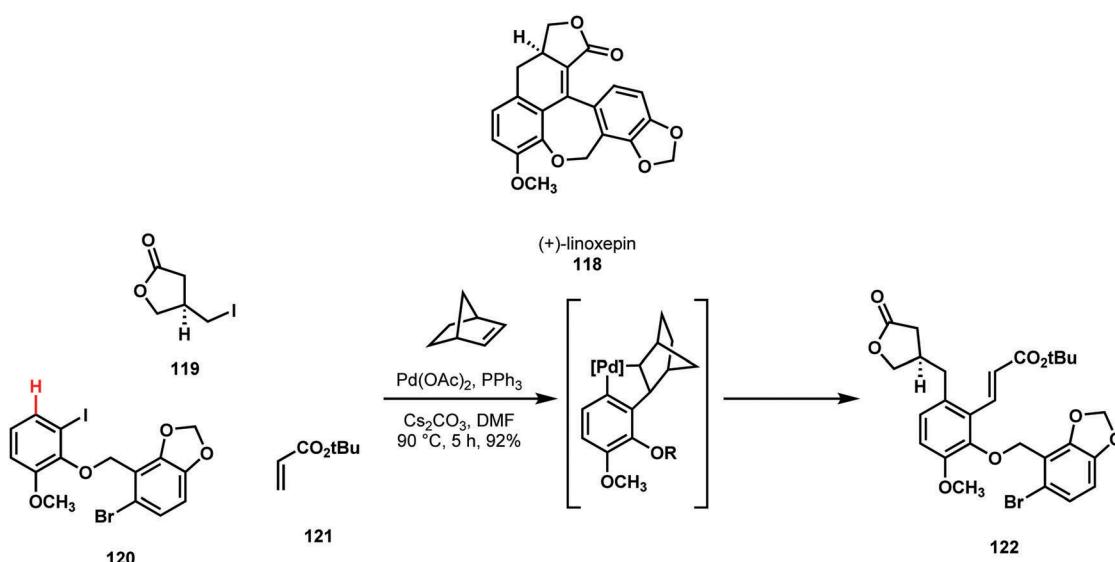
Functionalization of C(sp²)–H bonds is among the oldest examples of metal-directed C–H activation.⁶² This, combined with the ubiquity of aromatic systems in drug molecules, has contributed to the interest in new reactions to functionalize the C–H bonds of benzene rings. While planar aromatic systems do not offer questions of chirality the way C(sp³)–H bonds do, there do remain some examples of chiral information being introduced via C(sp²)–H functionalization. A key example is found in the atropisomerism/axial chirality of biaryl systems, but instances of desymmetrization arise as well.

With respect to the application of these reactions to total synthesis, key questions center on the directing group: are the

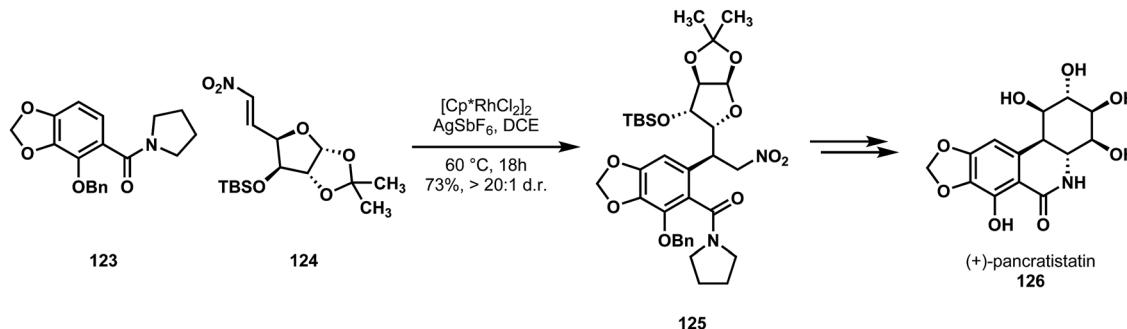
atoms of the directing element in the final structure? How easy is it to append/remove these species from the intermediate? And what level of regioselectivity and reactivity do these elements provide? These questions have inspired new explorations in synthesis and methodology, and expanded the reactivity pathways of various benzenoid structures by the use of directing elements of varying distance and atomic makeup. The examples below have been organized by the identity of the directing group in an attempt to delineate the strategic significance of directing group-choice in reference to the targeted natural product structure.

Lautens reported the first enantioselective total synthesis of linoxepin (**118**), along with a synthesis of its double bond regiosomer which they dubbed iso-linoxepin (Scheme 24).^{63,64} Their strategy was founded on the idea that the tetrasubstituted arene could be formed *via* a Catellani reaction, allowing one of the four substituted positions to be converted retrosynthetically to C–H (shown in red). Following this domino transformation, they envisioned that manipulation of the three carbon-containing constituents would allow them to form the remaining rings. However, over the course of their studies, they found that closing the lactonic ring structure was particularly challenging.⁶⁴ They explored a variety of alkyl and allyl halides, together with various olefin acceptors. In the end, their optimized synthesis comprised a Catellani reaction between the trisubstituted iodoarene **120**, *t*-butylacrylate **121**, and a known chiral primary alkyl iodide **119**. From **122**, two-fold ring closure in three steps provided linoxepin. The Lautens C–H functionalization strategy towards linoxepin enabled exploration of several different late-stage cyclization strategies, which highlights the versatility imparted by C–H functionalization retrosynthesis. Although only one strategy was successful in synthesizing the natural product, the tolerance and scalability of the Catellani reaction supported broad studies of linoxepin synthesis.

In 2017, Ellman and Potter reported the most rapid synthesis of pancratistatin (**126**) to date.⁶⁵ Using a reaction previously developed in their lab to forge a key sp²–sp³ bond, the Ellman group



Scheme 24 The creative use of a Catellani reaction in Lauten's synthesis of (+)-lineoxepin (**118**).



Scheme 25 Ellman and Potter's creative use of C–H functionalization in a synthesis of (+)-pancratistatin (126).

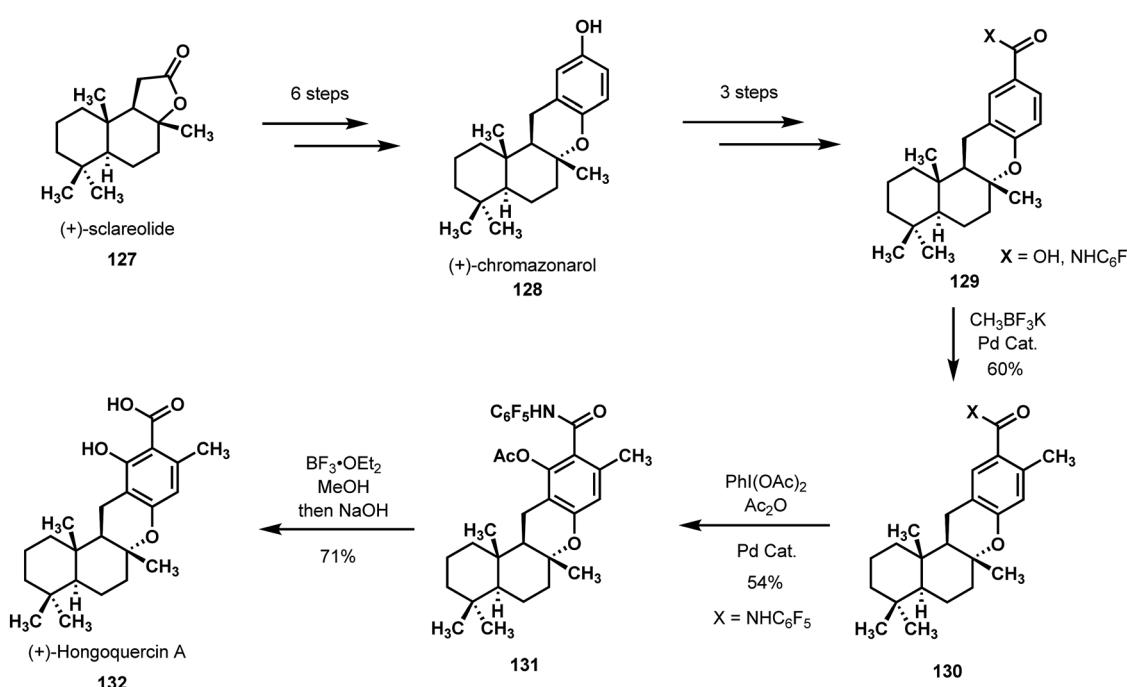
synthesized the natural product through an amide-directed C–H activation followed by nitroalkene 1,4-addition.⁶⁶ A traditional variant of this strategy had already been used by Sato, *via* the cupration of a 3,4,5-substituted aryl bromide and conjugate addition to a nitroalkene. Ellman noted that this strategy required several subsequent manipulations to install the carbonyl C1 unit and form the lactam.⁶⁷ They envisioned that their strategy, which would already have the amide in place, could rapidly forge the central carbon–carbon bond, and then use more efficient methods to reduce the nitro group and lactamize (Scheme 25).

They found that with proper protecting groups in place, their 2,3,4-substituted benzamide 123 underwent C–H activation and cleanly produced the conjugate addition product 125 with good regio- and diastereoselectivity. This intermediate was converted to (+)-pancratistatin (126) in 10 overall steps (longest linear), down from the previous shortest route of 13 steps. Their synthesis highlights the complementary nature of C–H functionalization methods in relation to traditional routes, and the

opportunities they offer with respect to functional group tolerance.

C–H functionalization chemistry can serve a powerful role in both target-based and diversity-oriented synthesis strategies. The Baran and Yu laboratories collaborated in 2013 in the design and execution of a synthesis of (+)-hongoquercin (132) starting from a related natural product (+)-chromazonarol (128) through successive applications of directed C(sp^2)–H alkylation and oxidation (Scheme 26).⁶⁸ These newly developed reactions built on the foundation of a larger effort in the Baran lab: the leveraging of the substituted *trans*-decalin structure of commercially available (+)-sclareolide (127) in the development of divergent syntheses of meroterpenoids.⁶⁹ From this body of work, their efficient six-step synthesis furnished several grams of (+)-chromazonarol.

Their development of a Pd(II)-catalyzed selective C–H methylation of the 6-position on *meta*-substituted benzoic acids required ligand acceleration to increase the C–H functionalization



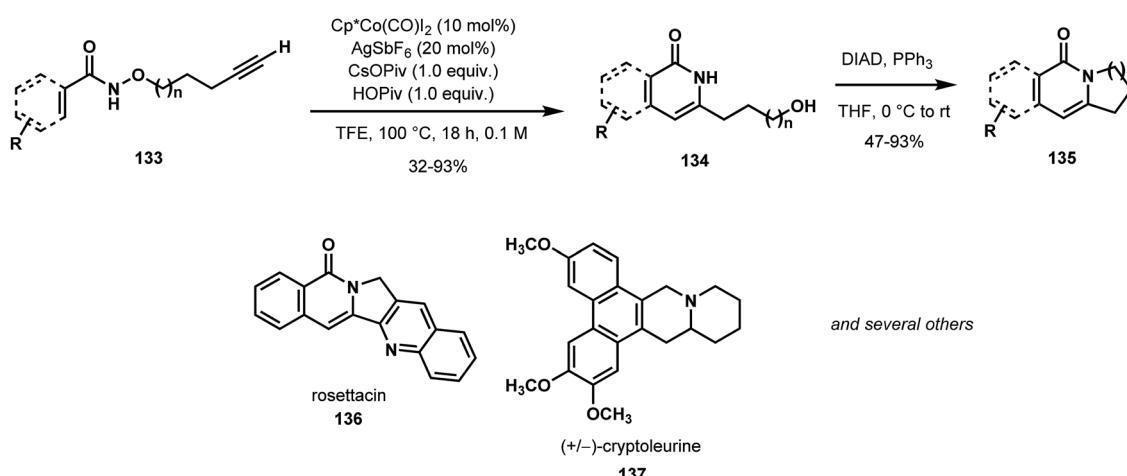
Scheme 26 Highlights in Baran and Yu's synthesis of (+)-hongoquercin A.

conversion. Thus, (+)-chromazonarol (**128**) was converted to the benzoic acid derivative **129** in two steps, which underwent efficient, directed C–H methylation to furnish **130**. In a subsequent step, a C–H hydroxylation was achieved on the remaining C(sp²)–H bond *ortho* to a carboxyl function. However, the carboxyl functional group was not a sufficiently strong directing group, and the yield for the desired hydroxylation was poor. The researchers examined different directing groups and reaction conditions: when the carboxylic acid group was converted to the perfluoroaryl-amide **130** and the reaction was performed in a slightly basic environment, C–H acetoxylation was achieved in moderate yields (**130** → **131**). Global deprotection of the amide delivered (+)-hongoquercin A (**132**) in six steps from (+)-chromazonarol (**128**).

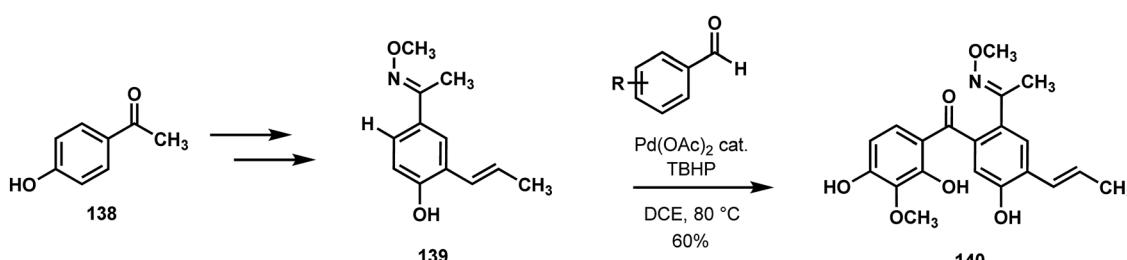
The Glorius group used their cobalt-catalyzed C–H functionalization/annulation method to develop rapid syntheses of indolizidine- and quinolizidine-containing natural products. The method prepared 3-substituted pyridones and isoquinolones (**134**) from alkyne-tethered, unsaturated hydroxamic esters **133** (Scheme 27).⁷⁰ The reaction strategically furnished the heterocycles with free NH and terminal OH bonds in position for a second annulation by the Mitsunobu method. In only 2–4 steps from the C–H functionalization products, the researchers were able to prepare several alkaloid natural products belonging to the aromathecin, protoberberine, and tylophora alkaloid families (**136**, **137**, etc.).

The Kim/Jung/Pyo synthesis of penchinone A (**140**) and analogues was notable due to the employment of an oxime-directed C(sp²)–H acylation of substituted acetophenone **139** (Scheme 28).⁷¹ Palladium catalyzed C–H activation enabled the use of the symmetrical starting material 4-hydroxyacetophenone (**138**). Acyl surrogates were used in the form of substituted benzaldehydes, and thus an array of analogues was prepared and subjected to biological testing. This achievement demonstrates how controlled C–H functionalizations can rapidly simplify the problem of synthesizing highly substituted arenes that are often present in natural products and bioactive molecules.

(+)-Steganone (**142**, Fig. 3) and (+)-isoschizandrin (**141**) are members of a family of highly substituted dibenzocyclooctadiene natural products, featuring axial chirality along their biaryl bond. Traditional strategies toward setting this element of stereochemistry involve either resolution of the rapidly interconverting chiral axis already bearing the 2,2'-functionality⁷² or the use of existing chirality elsewhere in the molecule followed by traditional cross-coupling to form the biaryl system with diastereoinduction.^{73,74} Two syntheses have been disclosed toward these natural products that take a different approach.^{75,76} In both cases, the racemic (and likely rapidly interconverting) biaryl system is subjected to C–H functionalization to form one of the necessary C–C bonds in the cyclooctadiene core, while also generating stable axial chirality. Interestingly, although both routes approach the axial chirality in similar ways, they retrosynthetically break different aryl–carbon



Scheme 27 Glorius's cobalt catalyzed C–H functionalization method for the synthesis of pyridones and isoquinolones and its use in the synthesis of alkaloids.



Scheme 28 Rapid access to highly substituted arenes by directed C–H acylation with palladium catalysis. Application to the synthesis of penchinone A.

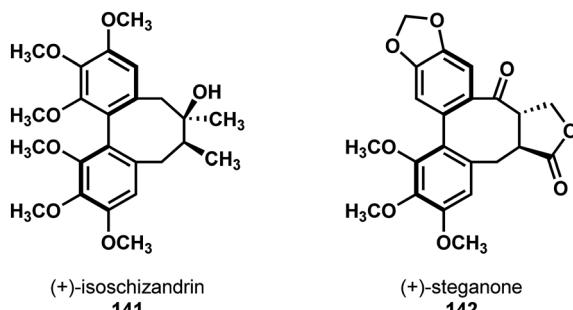


Fig. 3 The molecular structure of (+)-isoschizandrin (141) and (+)-steganone (142).

bonds, based on their respective directing groups. This strategic similarity of diverging strategies highlights some of the flexibility provided by the ever-expanding field of directed C–H functionalization.

Colobert and Wencel-Delord previously utilized chiral sulfoxides as directing elements for C–H functionalizations to form axially chiral biaryls, which they utilized in a formal synthesis of (+)-steganone (142, Fig. 3).^{75,77} Their initial thoughts were to perform a directed methylation followed by benzylic oxidation; however, they found that neither methyl electrophiles nor nucleophiles provided the desired methylation. Instead, they turned to an oxidative Heck reaction, believing that the resulting olefin could be transformed to a benzyl alcohol by a sequence of oxidative cleavage and carbonyl reduction reactions. Exposing chiral sulfoxide 143, as a mixture of interconverting axial diastereoisomers, to their oxidative Heck conditions gave the olefinated product 144 with good diastereoselectivity (Scheme 29). The sulfoxide itself was formed via aryl anion attack on a chiral sulfinate, followed by palladium-catalyzed biaryl formation. From compound 144, a known intermediate 145 on the way to steganone could be synthesized in 6 steps.

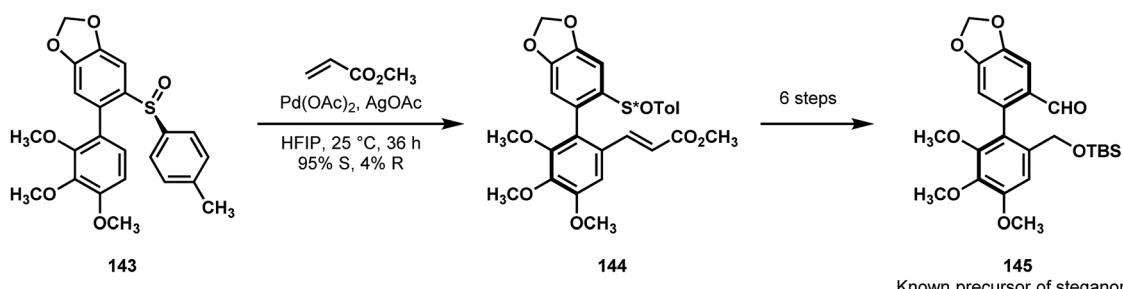
In 2018 Shi and coworkers reported formal syntheses of (+)-isoschizandrin (141, Fig. 3) and (+)-steganone.⁷⁶ On the foundation of their previous, atropselective C–H olefination, they envisioned that a conceptually-similar, aldehyde-directed vinylation could give them rapid access to functionalized dibenzocyclooctadiene frames from racemic precursors.⁷⁸ After developing optimized, general conditions, they applied their new C–H alkynylation method to racemic biaryl precursors 146 and 149 (Scheme 30). These reactions afforded the desired

products 147 and 150, respectively, in good enantioselectivity, and, in the case of steganone, good regioselectivity. In 5–6 steps, both intermediates were advanced to known precursors of each natural product (148 and 151), completing their formal syntheses.

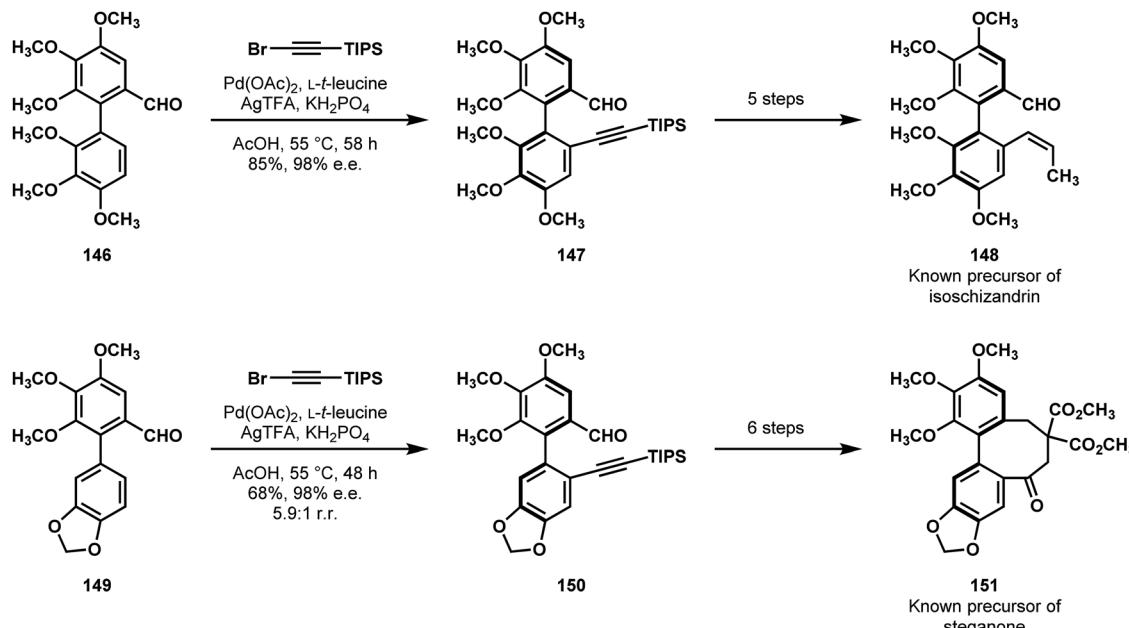
In 2017, Shen, Li and Zhang reported the first synthesis of delavatine A (152) (Scheme 31), the stereochemistry of which had been assigned by one of their groups during its isolation with some uncertainty.⁷⁹ As such, their synthesis sought to not just access the target molecule, but also its stereoisomers to confirm the structural assignment, as well as to further explore its bioactivity. Strategically, they reported that they were discouraged from taking an approach that would functionalize an isoquinoline core, over concerns regarding the regio- and stereoselectivity of subsequent manipulations. Instead, they were drawn to a C–H functionalization approach, wherein the five-membered ring would arise from a 1-indanone derivative, and the nitrogen-containing ring would be annulated on during the synthesis.

Starting with 5-methoxy-1-indanone, 153, they were able to rapidly access a triflamide-bearing tetrasubstituted olefin 154. Here, their synthesis diverged into two approaches, one of which was simply enantioselective, while the other which featured a kinetic resolution, giving access to both isomers. Extensive optimization of a model substrate demonstrated that tetrasubstituted indene derivatives could undergo highly selective hydrogenation with a chiral BIBOP ligand and rhodium. From here, slight adjustment of Yu's conditions for directed olefination provided the desired C7 olefin 156 (Scheme 32).⁸⁰ By taking the racemic reduction product, a mono-protected amino acid ligand gave high selectivity for the kinetic resolution (also inspired by Yu's conditions⁸¹), which they subsequently used to confirm the stereochemical configuration of the natural product.

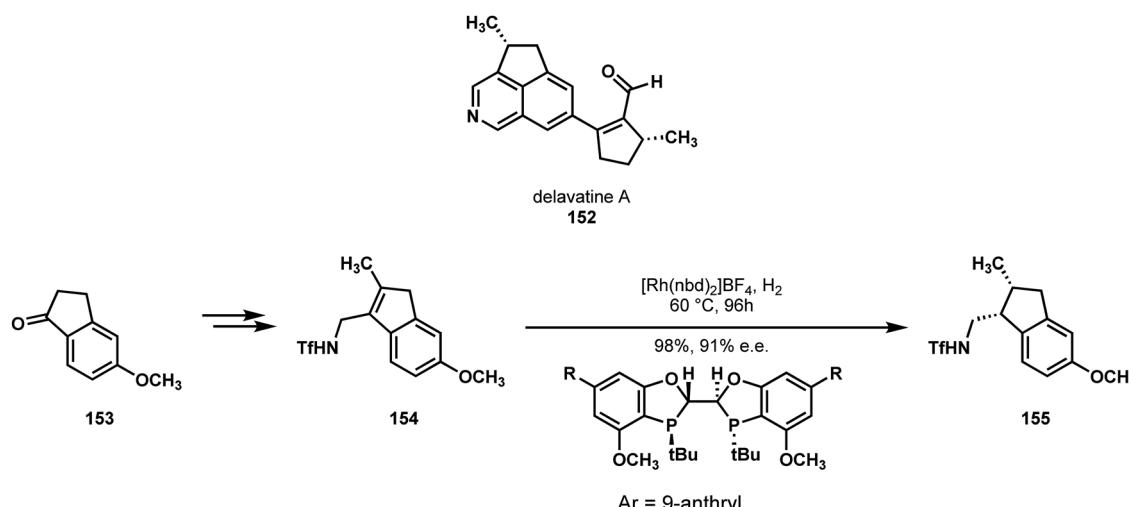
From here, Shen, Li and Zhang were able to synthesize the isoquinoline fragment and cross couple it to a chiral pool-derived cyclopentene fragment. Their directing group for the C–H functionalization, the triflamide, fit their synthesis well, as it served as both the nitrogen source for annulation of the isoquinoline, and permitted a late-stage elimination to aromatize the ring system. In this case, C–H functionalization served multiple roles in their synthesis. It allowed them to utilize simple starting materials, obviating issues they predicted with a more traditional approach. It also served as a stereo-determining kinetic resolution, which served the



Scheme 29 C–H arylation in a synthesis of (+)-steganone (142) by Colobert and Wencel-Delord.



Scheme 30 Symmetry breaking C–H alkynylations featured in Shi's formal syntheses of (+)-isoschizandrin (**141**) and (+)-steganone (**142**).



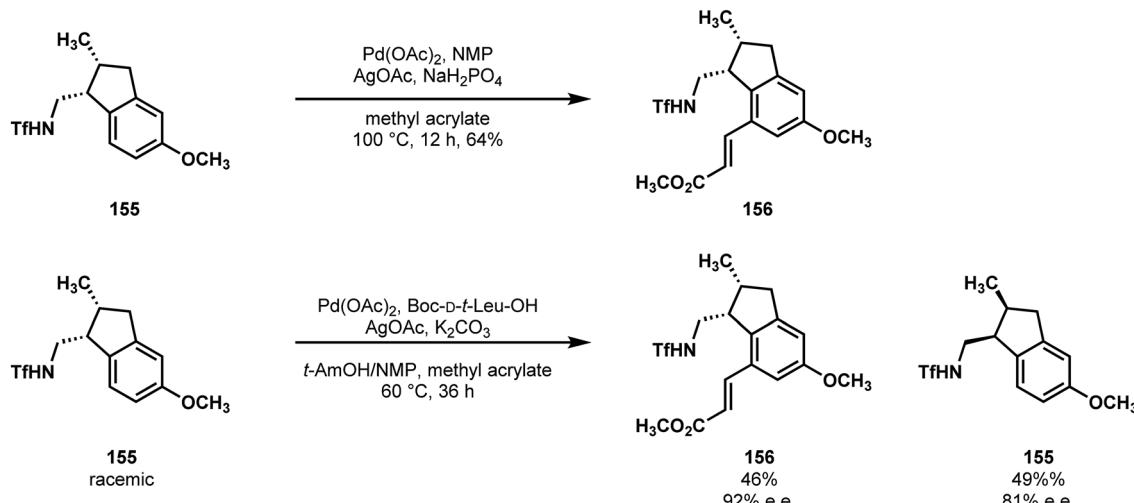
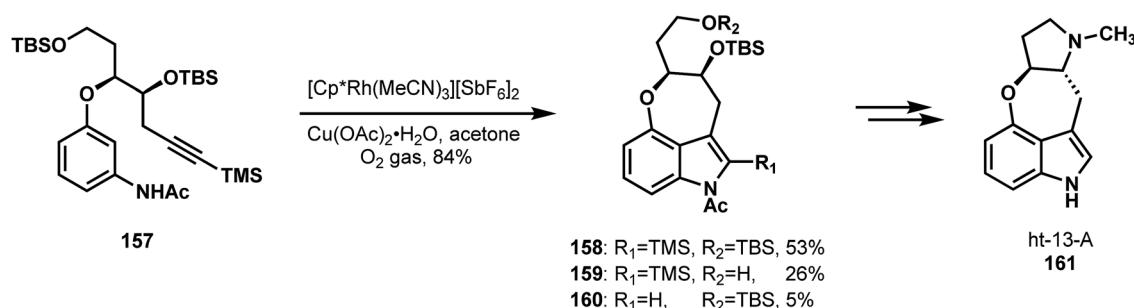
Scheme 31 The molecular structure of delavatine A (**152**) and an asymmetric reduction used to define a challenging stereocenter in Shi, Li, and Zhang's synthesis.

authors purposes of fully elucidating the natural product's configuration.

In 2016, Jia and coworkers reported a stereospecific synthesis of the indole alkaloid ht-13-A (**161**, Scheme 33).⁸² Their retrosynthetic analysis centered on the intramolecular, directed C–H indole annulation that they developed based on Fagnou's intermolecular variant, followed by late stage formation of the pyrrole unit *via* double alkylation.^{83,84} This strategy and their insights underlying it allowed them to back up to simple 3-nitrophenol and a chiral derivative of (*R*)-malic acid as precursors to most of the carbon skeleton. These fragments were combined *via* an invertive Mitsunobu coupling; this step, in combination with some straightforward functional group manipulations, set all of the

carbon atoms in place. The pivotal C–H annulation process on compound **157** provided the desired 3,4-fused indole core in good yield as a mixture containing partial desilylation products (**158–160**). Nevertheless, these products could be taken together to the natural product in only three additional steps.

In total, Jia and coworkers accomplished a 10-step synthesis of ht-13-A (**161**), which could be performed on gram scale. By deferring the pyrrolidine annulation to the very end, their approach permits straightforward syntheses of a variety of *N*-alkylated analogues. Perhaps most impressive, however, is how the key C–H functionalization step allows for the use of simple and accessible starting materials. One might imagine that traditional cross-coupling methods would provide a similar route from the

Scheme 32 Directed C–H olefination reactions used in Shi, Li, and Zhang's synthesis of delavatine A (**152**).

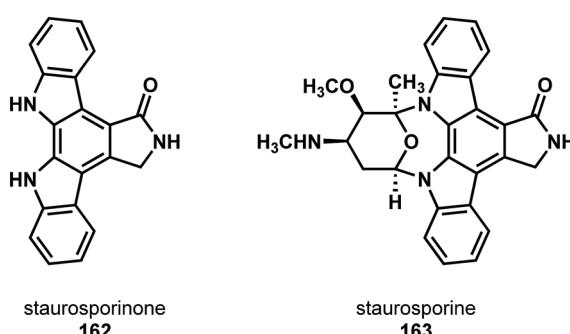
Scheme 33 Creative C–H annulation in a synthesis of ht-13-A reported by Jia.

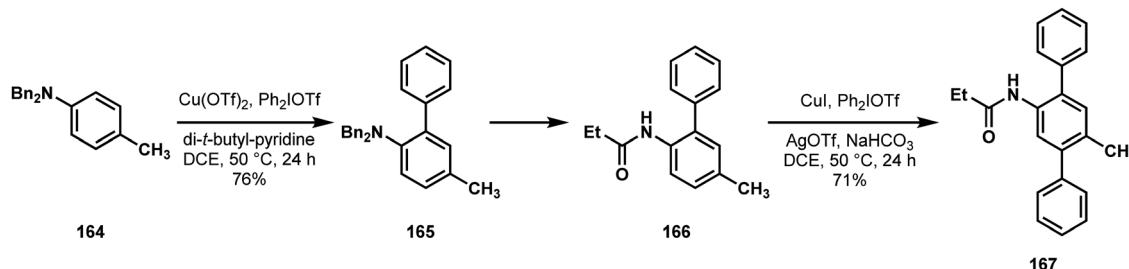
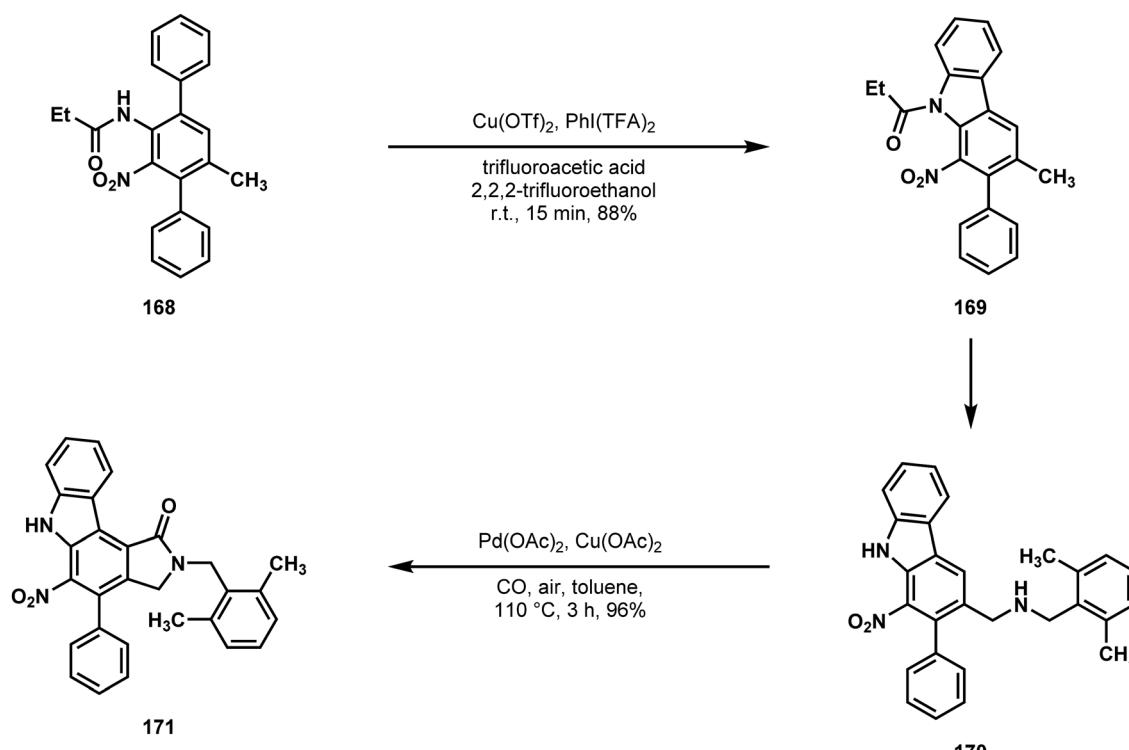
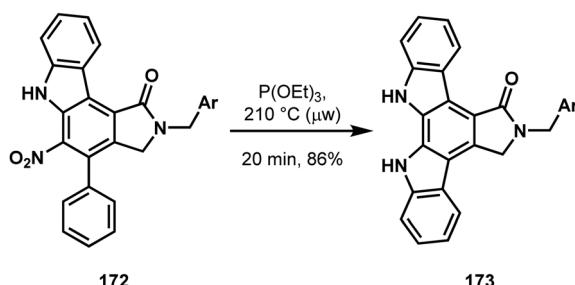
2-halogenated species (indeed, such a starting material was used in the Söderberg synthesis of the related ht-13-B⁸⁵); however, by perceiving the C–H bond as a reactive ‘functional group’, the synthesis could have a pleasingly simple beginning.

In 2016 Gaunt and coworkers reported a synthesis of the indolocarbazole natural product staurosporinone (K-252c) (**163**, Fig. 4).⁸⁶ Previous reports toward this class of natural products tended to feature the joining of two indoles or their equivalent

to form the indolocarbazole core. The authors noted the difficulty this strategy could impart to subsequent steps which would necessitate differentiating the indole nitrogens (see staurosporine **163**). Instead, they imagined forming the two indoles on a central benzene core. The benefit to this type of strategy is that sequential indole formation would allow for analogue synthesis with selective functionalization at either end of the indolocarbazole.

Beginning with *N,N*-dibenzyl-*p*-toluidine **164**, Gaunt and coworkers first installed a phenyl at the *ortho* C–H relative to the aniline nitrogen to give **165** (Scheme 34).⁸⁷ Functional group interconversion to **166** allowed for the installation of another phenyl group *meta* to the same nitrogen (**166** → **167**), setting all of the carbon atoms of the indolocarbazole *via* two C–H functionalizations.⁸⁸ After nitration, the first C–H carbazole synthesis was performed (**168** → **169**, Scheme 35), followed by benzylic oxidation and amine formation.⁸⁹ This set the stage for lactamization onto the benzene core of **170** using carbon monoxide as a C1 equivalent. Judicious choice of arene prevented competing C–H insertion into the amine protecting group, providing **171** in good yield.⁹⁰ The remaining carbazole synthesis was completed *via* reductive nitrene insertion onto the remaining phenyl of compound **172** (Scheme 36), leaving a simple deprotection of **173** as the final operation on their path to staurosporinone (**162**).⁹¹

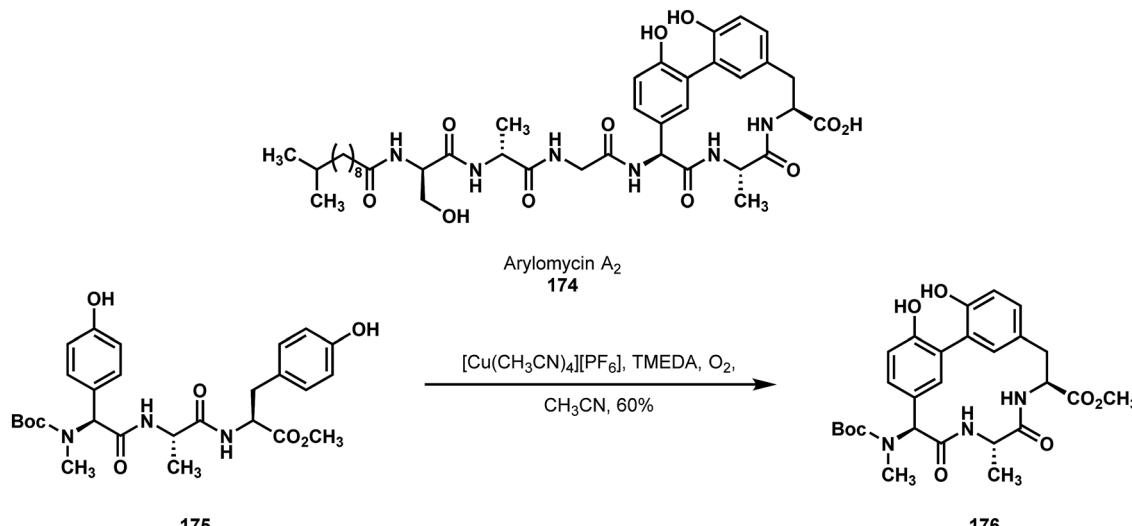
Fig. 4 The molecular structure of staurosporinone (**162**) and staurosporine (**163**).

Scheme 34 Two C–H arylation reactions used by Gaunt in a synthesis of staurosporinone (**163**).Scheme 35 C–H amination and C–H acylation reactions used by Gaunt in a synthesis of staurosporinone (**163**).Scheme 36 The final C–H amination reaction in Gaunt's synthesis of staurosporinone (**163**).

This multiple C–H functionalization strategy allowed Gaunt and coworkers to rapidly synthesize the complex staurosporinone structure from simple *p*-toluidine. Recognition of this simple, symmetrical molecule embedded into the core of

this family of natural products was key to the conception of this strategy, which is highly reliant on the functionalization of C–H bonds.

In 2018, Baran and Romesberg reported a synthetic route to a host of arylomycin analogues, **174** (Scheme 37).⁹² Arylomycins, a family of “latent” antibiotics, had typically been synthesized by Suzuki–Miyaura coupling to forge the biaryl linkage.⁹³ However, this route had been plagued by high palladium loadings, poor yields, and the need to perform functional/protecting group manipulations. Baran and Romesberg instead sought to form the biaryl linkage *via* direct C–H/C–H coupling. To this end, they screened a variety of copper catalysts, ligands, and oxidants, and found optimized conditions to take a simple Boc- and OMe-protected tripeptide **175** to the arylomycin macrocyclic core **176** in good yield. From here, they were able to append a variety of functionalities to the N-terminus to test for improved antibiotic activity.



Scheme 37 A direct cross-coupling developed by Baran and Romesberg to obviate issues with an analogous, traditional cross-coupling in a synthesis of arylomycin A₂.

C(sp²)–H functionalization of heteroarenes

Heteroarenes represent an important class of scaffolds for synthesis of diverse bioactive molecules. Like simple arenes, substituent effects can play a role in the regioselectivity of the classical reactions of heteroarenes; however, the identity and location of a heteroatom can also be of great significance. Because this can also play a role in C–H functionalization reactions, the C(sp²)–H functionalization of heteroarenes merits its own category. Although many of these examples possess a directing group, in others there is guidance based on the inherent reactivity of the heteroarene itself. Recognizing this interplay can be critical to effective retrosynthetic planning for functionalized heteroaromatic targets.

In 2015, two different syntheses of members of the dictyodendrin family (**177** to **179**, Fig. 5), both of which were highly reliant on C–H functionalization, were published in quick succession.^{94,95} Both were driven by recognition of a simple

heterocycle embedded into the core of the dictyodendrins. The first, which recognized the central pyrrole unit as a simple and symmetrical starting material, was reported jointly by the groups of Yamaguchi, Itami, and Davies.⁹⁴ The second, which recognized the central indole unit as a starting point, was reported by Gaunt.⁹⁵ In both syntheses, the central intermediate, decorated with a variety of C–H bonds, serves as a platform to study the use of multiple, successive C–H functionalization reactions.

But, as one might imagine, in cases where multiple C–H functionalization strategies are used to modify such simple building blocks, questions regarding the order of operations come to the fore. When there is little or no concern for highly reactive elements, like those which guide cross couplings or traditional ionic chemistry, the possibilities can become overwhelming. These syntheses exhibit the power that a logical and experimental assessment of various possible routes can contribute to a synthesis.

Yamaguchi, Itami, and Davies explored the dictyodendrin skeleton with the vision that the multiring system would be

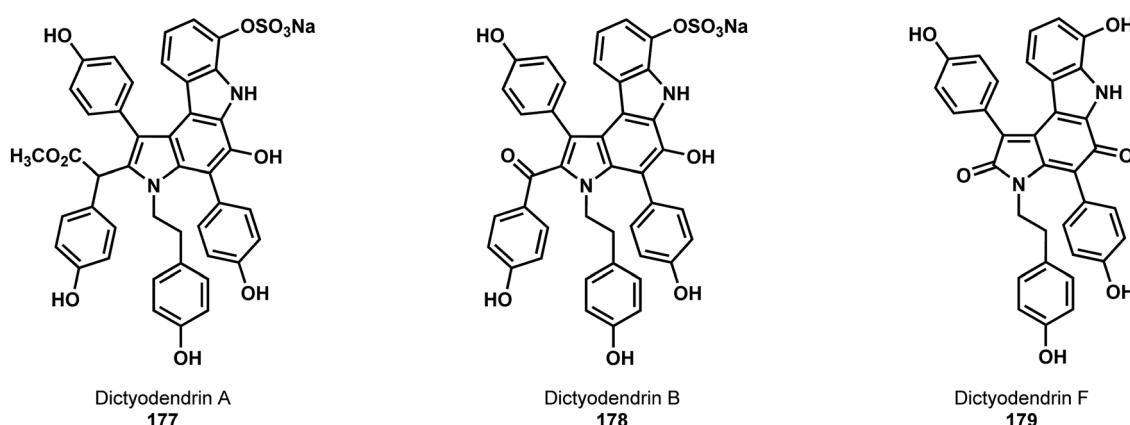
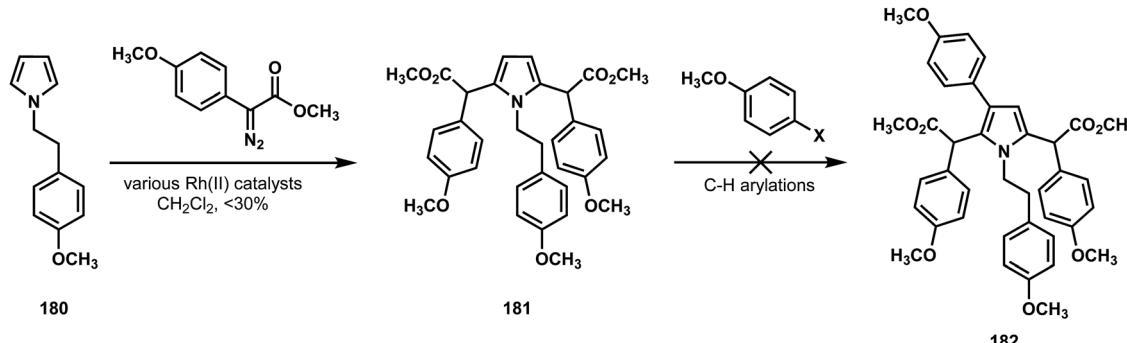


Fig. 5 The molecular structure of dictyodendrin A (**177**), B (**178**), and F (**179**).

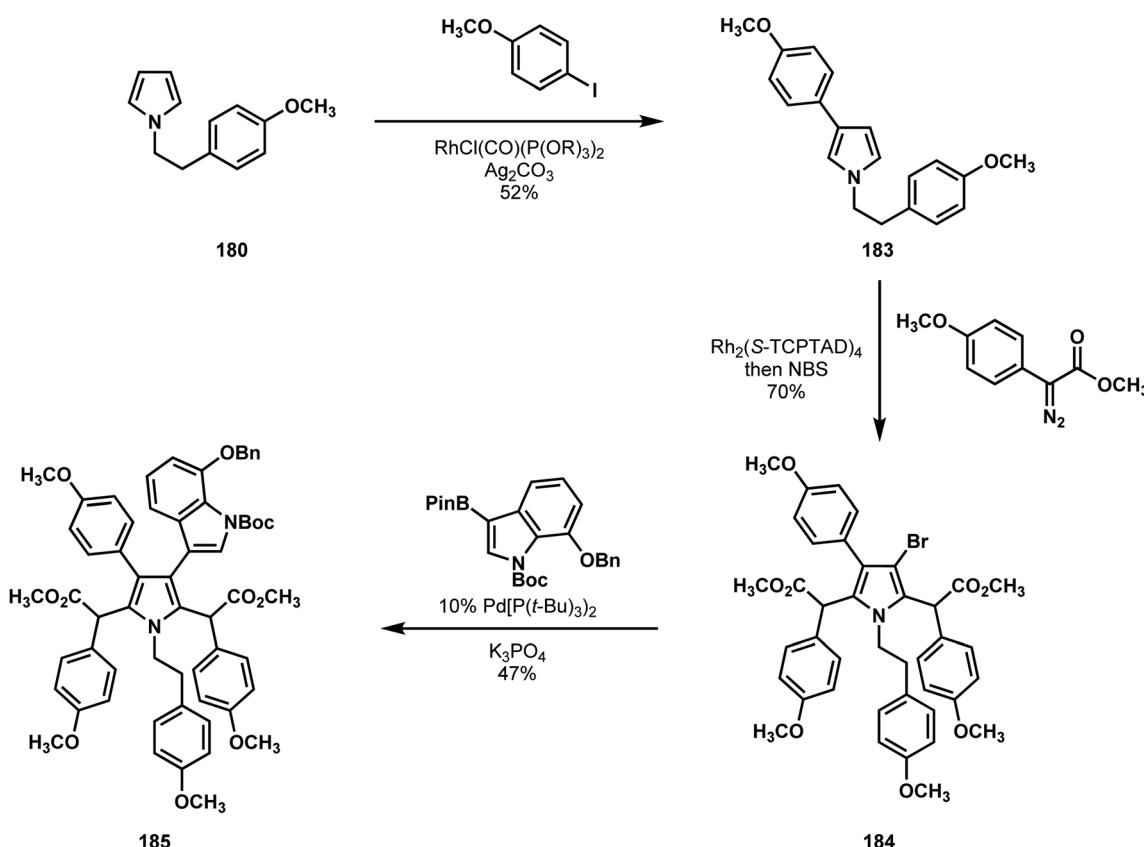


Scheme 38 The initial, unsuccessful strategy explored by Yamaguchi, Itami, and Davies toward dictyodendrin A (**177**) and F (**179**).

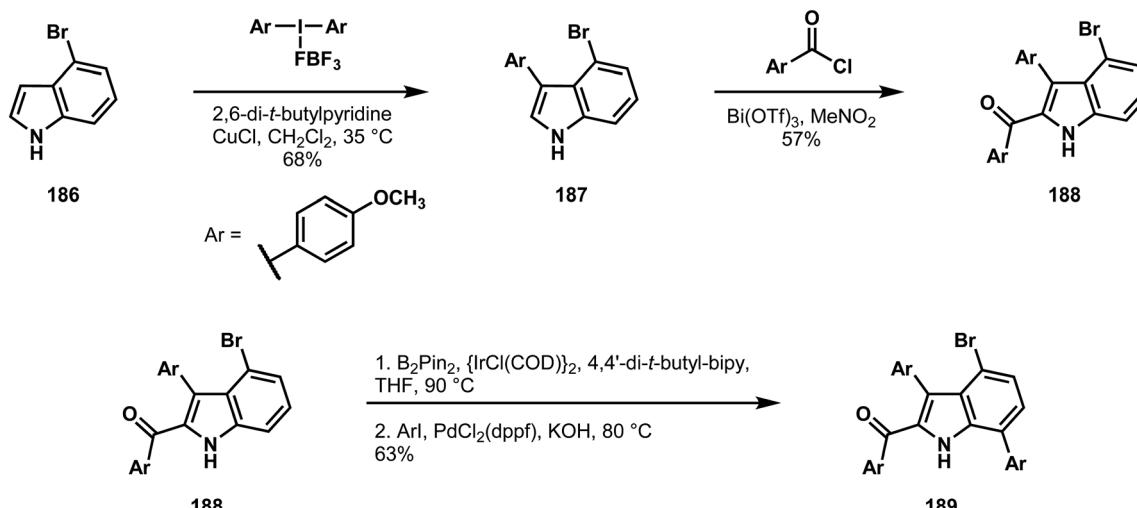
derived from a pentasubstituted pyrrole.⁹⁴ From there, they envisioned functionalizing three of the five positions of a pyrrole nucleus through metal-catalyzed C–H functionalizations. However, the exact ordering of these reactions was unclear to the authors at the outset. Indeed, their attempts to install the functionalities at positions 2 and 5 *via* a two-fold dirhodium-catalyzed C–H insertion of compound **180** gave only low yields of the desired product **181** (Scheme 38). In addition, compound **181** could not be advanced to the 2,3,5 functionalized intermediate **182**. Instead, they found that, by first installing the arene in the 3-position to give **183** (Scheme 39), they could, with catalyst optimization, perform the three successive

C–H functionalizations.⁹⁶ An unprecedented, two-fold pyrrole dirhodium C–H insertion on substrate **186** was followed by bromination of the 3-position (**183** → **184**), and a Suzuki–Miyaura coupling with the indole-containing boronate shown to afford pentasubstituted pyrrole **185**. From this central intermediate, 6π-electrocyclization formed the remaining arene ring, and functional group interconversions yielded both dictyodendrin A (**177**) and F (**179**).

The Gaunt laboratory approached the dictyodendrins from the vantage point of a simple indole intermediate.⁹⁵ Beginning with 4-bromoindole **186** (Scheme 40), they could capitalize on the high, intrinsic reactivities at positions 3 and 2 in successive



Scheme 39 The C–H functionalization strategy that successfully provided Yamaguchi, Itami, and Davies access to dictyodendrin A (**177**) and F (**179**) *via* a pentasubstituted pyrrole nucleus.



Scheme 40 Three C–H functionalization reactions around the indole nucleus employed by Gaunt in a synthesis of dictyodendrin B (**178**).

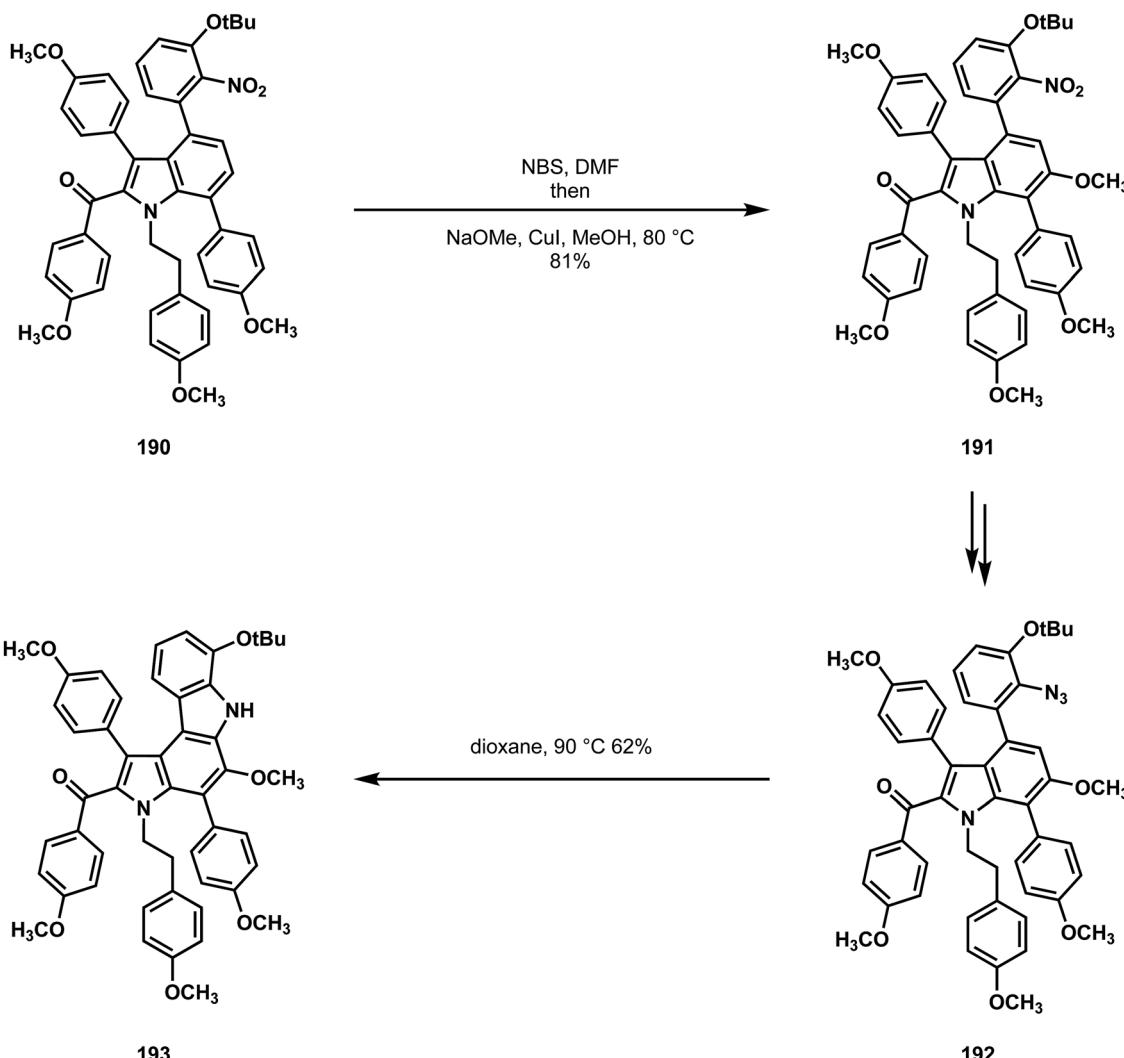
carbon–carbon bond forming reactions to give compounds **187** and **188**, respectively.⁹⁷ From here, the directing capacity of the N–H bond was exploited to borylate the 7 position, which enabled a traditional cross-coupling to provide **189**.⁹⁸ Sequential bond formations at the indole nitrogen and the bromine-bearing carbon at position 4 gave access to compound **190** (Scheme 41). In the context of **190**, C6–H bromination, which was found to be critically reliant on the nitro group of the C4 arene, set the stage for a copper(i)-mediated Ullman coupling under the conditions shown to yield the C6 ether **191**. Finally, the last C–H functionalization to form the carbazole was addressed. Efforts to directly convert the nitro group or a protected amine derivative to the carbazole failed, resulting in explorations of the readily available azide **192**. However, metal-catalyzed conditions also failed to yield the desired product. From here, Gaunt and coworkers turned to thermal decomposition, as had been used in a previous synthesis of the dictyodendrins.⁹⁹ While this was successful, because of scalability concerns around the thermal decomposition of azides, they demonstrated that flow chemistry could achieve a large scale synthesis of the protected dictyodendrin species **193**. This intermediate could then be deprotected and sulfonated, yielding dictyodendrin B (**178**). Impressively, the entire synthesis was performed on multigram scale, yielding a gram of the natural product as of the time of publication. In total, of the six carbons on the indole frame, all of which are functionalized in the natural product, five began with C–H bonds, and underwent a variety of transformations under the control of either the intrinsic reactivity of the indole nucleus or proximity-dependent directivity effects.

The seemingly simple structure of frutinone A (**198**) (Scheme 42) has beset several efforts to achieve its direct and efficient synthesis.^{100,101} Modular syntheses of frutinone A and site-specifically modified variants are desirable due to the potentially useful biological activity of the natural product. The Hong group designed and executed a straightforward three-step synthesis of frutinone A featuring a palladium catalyzed C–H activation/carbonylation reaction to forge the fused coumarin heterocycle.¹⁰² In the development of this transformation, the

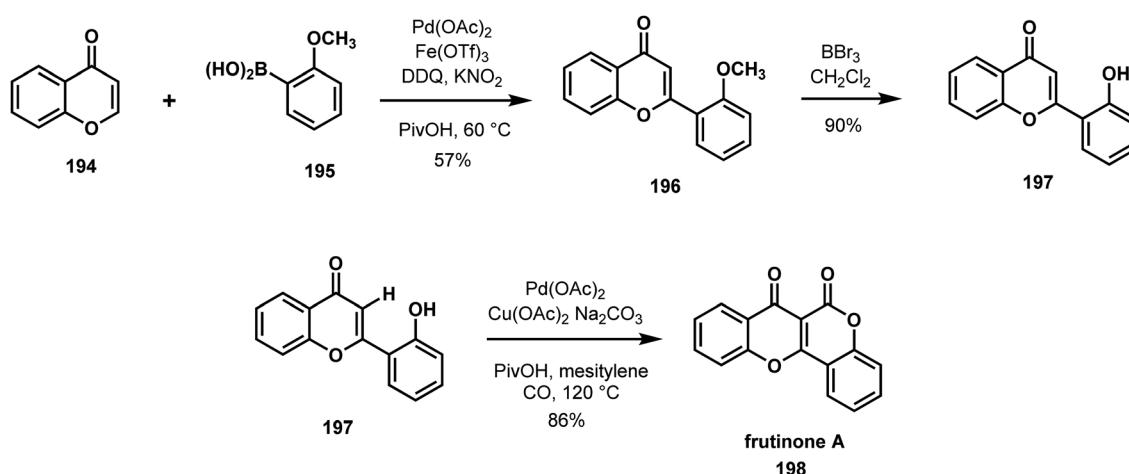
researchers evaluated several oxidants, solvents, and reaction conditions and found that Cu(OAc)₂ and a nonpolar solvent like mesitylene were important to the efficient construction of **198**. The success of this carbonylative C–H functionalization reaction allowed the Hong group to leverage their previously reported method for synthesizing 2-arylchromones **196**, a convenient intermediate in their short synthesis of frutinone A (**198**).

In 2016, Movassaghi and coworkers reported a synthesis of haplocidine (**199**) and haplocine (**200**) (Scheme 43) that was reliant on a C–H functionalization strategy.¹⁰³ Since their group had previously worked on the pentacyclic aspidosperma alkaloids, Movassaghi sought a unified approach to the hexacyclic hemiaminal aspidosperma alkaloids [e.g. fendleridine (**201**)], including those possessing A-ring oxidation (e.g. compounds **199** and **200**). To this end, they envisioned a strategy involving a late-stage C–H oxidation, directed by the indoline nitrogen, to unite their synthesis of fendleridine with those of haplocidine and haplocine. Following their synthesis of fendleridine (**201**) and the derived *N*-acyl and *N*-propionyl variants **202** and **203**, respectively, they explored the feasibility of directed arene hydroxylations. However, they found that the tertiary amine likely competed with amide coordination to palladium, thereby preventing the desired C–H activation. Instead, they devised a strategy where, in a single pot, they would open the hemiaminal to form the iminium and ester, perform the arene C–H oxidation, and then hydrolyze the ester to reclose the hemiaminal (Scheme 43). This strategy was spectacularly successful and afforded haplocidine (**199**) and haplocine (**200**) in good yields. These attractive, divergent syntheses from a common alkaloid precursor creatively localize the palladium catalyst to ensure that the desired arene C–H bond is the site at which the pivotal, late-stage oxidation occurs. It is easy to imagine how this strategy could permit analogous, directed C–H functionalizations in efforts to diversify the structures of these biologically active alkaloids.

In 2015, Baran's laboratory described a synthesis of the 8-membered endoperoxides verruculogen (**205**) and fumitremorgin A (**204**) (Scheme 44).¹⁰⁴ Their retrosynthetic design was



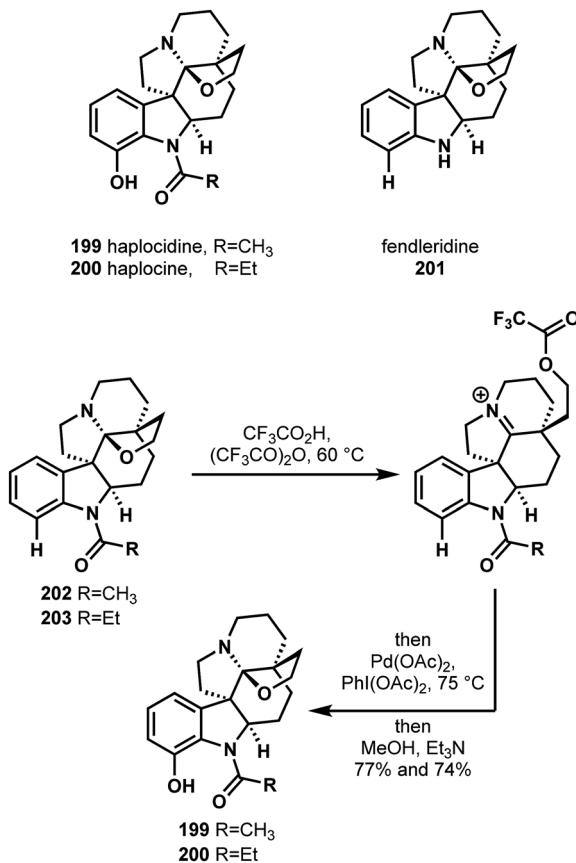
Scheme 41 The final steps of a synthesis of dictyodendrin B (**178**) reported by Gaunt, featuring site-specific C–H halogenation and amination.



Scheme 42 Frutinone A synthesis by carbonylative C–H functionalization.

based on the idea that the left-most 6,5-ring system could be traced back to the chiral and readily available L-tryptophan.

To their surprise, however, they found that the literature methods for directly installing C6 oxidation on L-tryptophan was lacking.



Scheme 43 The C–H hydroxylation strategy reported by Movassagh in their unified synthesis of haplocidine (**199**), haplocine (**200**), and fenderidine (**201**).

Thus, they explored methods for achieving a borylation at C6 using iridium and found that the 3-substituted indole system had a great propensity for reacting at the undesired C5 position. Extensive screening allowed them to elucidate optimal conditions for C6 borylation of the 1,3-substituted indole derivative **206**, which they carried on in a single pot reaction to the C6 methoxy group found in compound **207**, a substance that could be advanced to natural products **205** and **204** in 9–10 additional steps.

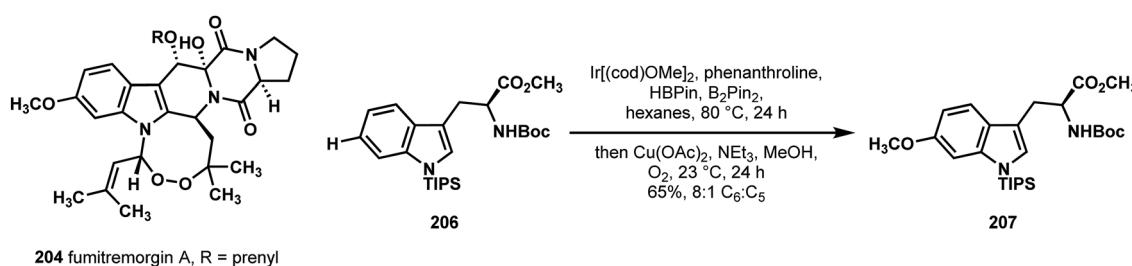
Banwell compared and contrasted traditional and C–H functionalization approaches to indoles in an effort that highlighted the advantages of modern C–H functionalization in both speed and strategy.¹⁰⁵ Using traditional conditions, Banwell

synthesized the C7-brominated indole **212** required for amaryllidaceae alkaloid synthesis in the three-step sequence shown in Scheme 44. By contrast, indole **208** was transformed to the C7 boronated indole **209** in a single step *via* Hartwig's catalytic C–H borylation method.¹⁰⁶ Compound **209**, primed for a carbon–carbon bond forming cross-coupling, was advanced to three members of the amaryllidaceae alkaloids **213** (Scheme 45).

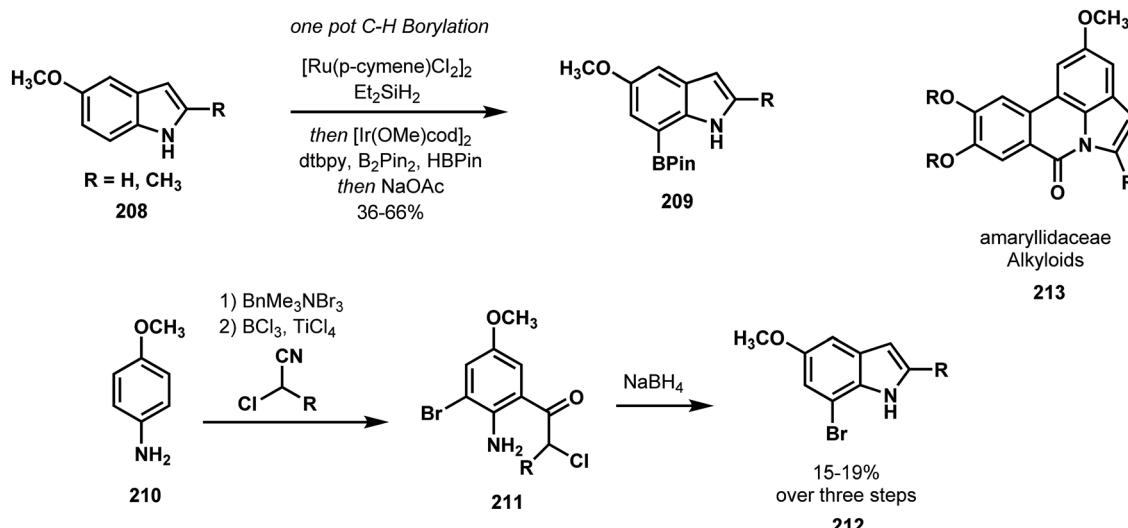
There has been significant interest in the synthesis of the complex indole alkaloid (+)-actinophyllic acid (**216**) (Scheme 46), a scarce natural product discovered in the leaves of *Alstonia actinophyllia*, which is a potent inhibitor of carboxypeptidase U. One of the main challenges faced by chemists working on this molecule has been the introduction or construction of the indole moiety in the unusual azabicyclo[3.3.2]decanene architecture. In the case of direct functionalization of the 2-position on indole moieties to forge the C2–C16 bond, these approaches tend to require either the installation/deinstallation of directing groups for metallation¹⁰⁷ or some type of prefunctionalization¹⁰⁸ towards C–C bond formation. In the formal synthesis by Qin, the use of photoredox catalysis/C–H functionalization¹⁰⁹ to directly construct the C2–C16 bond in an intermolecular fashion enabled the convenient introduction of the requisite indole from simple skatole (**214** → **215**).¹¹⁰ Oxidation of the indole-bearing methyl group afforded an opportunity to close the 8-membered ring with annulation of the remainder of the ring framework by an intramolecular 1,3-dipolar nitrone/alkene cycloaddition; these processes put in place all of the carbons needed to complete a formal synthesis of this complex natural product.

C–H functionalization *via* carbenes, nitrenes, and oxos

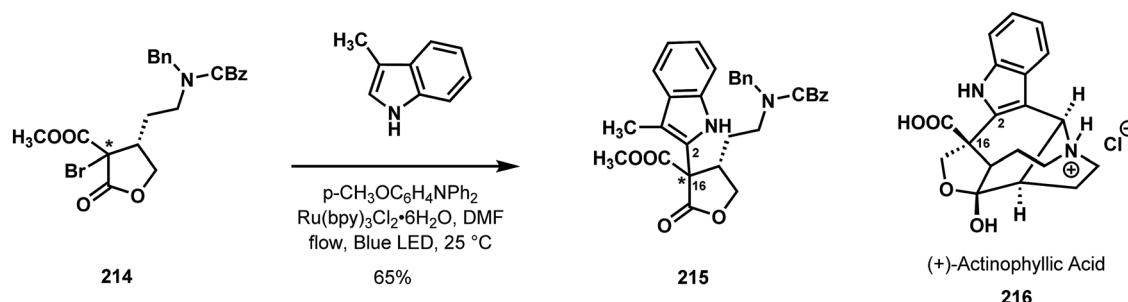
Directed C–H functionalization can offer powerful retrosynthetic transformations, with high confidence in regioselectivity and even stereochemistry. However, installation and removal of a directing group has the potential to be laborious and prohibitive if there is no other strategic utility of the functional group. Instead, C–H functionalization reactions that selectively target positions of sufficient reactivity, can be particularly powerful. Achieving regio- and stereoselectivity is still a considerable challenge in efforts to exploit the intrinsic reactivities of C–H bonds. Work in the past ten years has yielded an expanded menu of catalysts that display exquisite selectivities in C–H functionalization reactions^{1,111–113}



Scheme 44 C6 selective C–H borylation of protected L-tryptophan developed by Baran in a synthesis of fumitremorgin A (**204**) and verruculogen (**205**).



Scheme 45 C-H borylation of indoles in the synthesis of amaryllidaceae alkaloids.



Scheme 46 Photoredox C-H functionalization of skatole in the synthesis of (+)-actinophyllic acid.

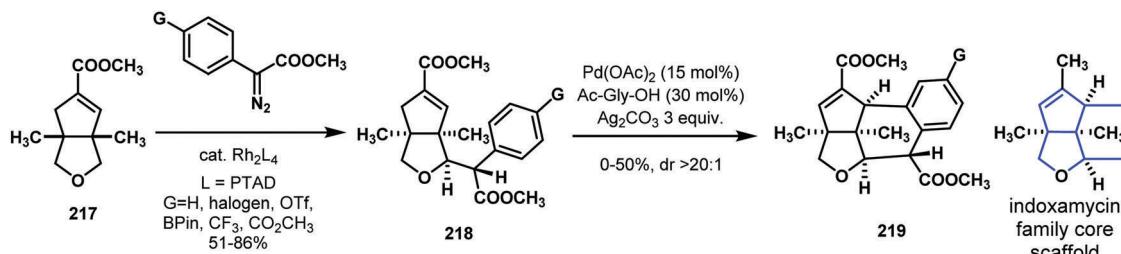
and has enabled their use in total synthesis as reliable reactions. Below, we address recent examples of the tendency of metal carbenoid and nitrenoid species to insert into so-called “hydridic” C-H bonds, which are now well-established bond forming processes in synthesis, and also include some examples of C-H bond oxidations mediated by electrophilic oxygen and metal-oxo reagents; these latter reactions display similar patterns of selectivity in relation to carbenoid and nitrenoid species.

Carbenes

The Sorensen group has long studied the use of C-H bond activation in total synthesis. In addition to C-H oxidations like the Sanford one highlighted in jiadifenolide, the group is also interested in C-C bond forming events coupled with C-H activations. One achievement in that field was the synthesis of a benzo-fused core of the indoxamycin family, which was achieved by sequential dirhodium catalyzed carbene C-H insertion and an ester directed C-H palladation/Heck cyclization.¹¹⁴ This synthesis developed through a unique collaboration between the Sorensen, Davies and Yu lab. The previous syntheses of the indoxamycin framework by Carreira and Ding began with the six-membered ring in place and was followed by annulation of the peripheral 5,5 cycles to complete the *cis*-peri-fused 6,5,5 tricyclic

skeleton (Scheme 47, shown in blue). The C-H functionalization approach by the Sorensen group reoriented the synthetic strategy and enabled the use of a pseudosymmetric enoate starting material 217 which could be readily prepared through a Pauson-Khand reaction. Not only did the high symmetry of the starting material expedite the synthesis for study of the ensuing C-H functionalizations, it also had a strong influence on the stereochemical outcomes of the subsequent skeletal construction. For most of the reactions, the diastereomeric ratio was excellent, mostly 20 : 1.

The first C-H functionalization consisted of diastereoselective C-H insertion of a donor-acceptor carbene. A small library of dirhodium catalysts was screened for reactivity and selectivity in the desired intermolecular insertion of the metallocarbene into the C-H bond adjacent to the THF oxygen to furnish a series of molecules 219. Use of the PTAD ligand gave good yields and selectivity for bond formation on the convex side of the bicyclo[3.3.0]octane. In the next C-H functionalization, after some difficulty in reaction development, an unprecedented ester directed C-H palladation/Heck cyclization forged the six-membered ring in moderate yields with excellent diastereoselectivities (218 → 219). Indeed, some low-level conformational analysis suggested that approach of the palladated moiety was impossible from the other side of the ring. In this



Scheme 47 The Sorenson synthesis of benzo-fused indoxamycin core featuring sequential metallocarbene C–H insertion and ester directed C–H palladation/Heck cyclization.

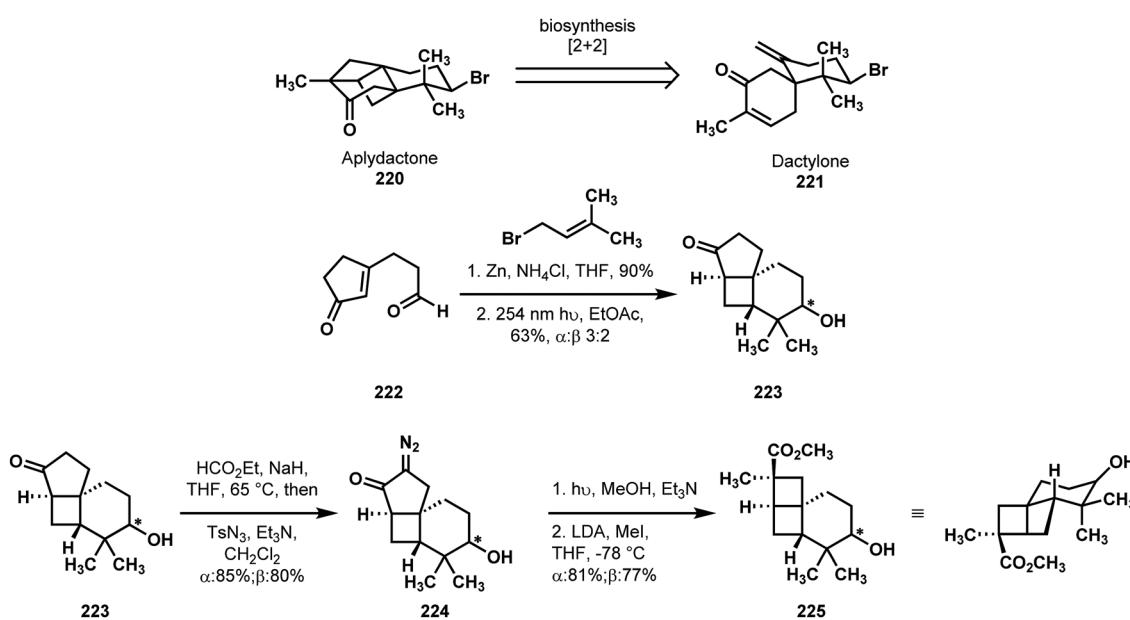
approach to the core of indoxamycin, a C–H functionalization strategy allowed for the use of a highly symmetric and easily synthesized starting material for the study of key reactions, and the fused 5,5 bicyclic nature of this material also exerted fine control on the stereochemical outcomes of C–H functionalization reactions in a complex steric environment.

In 2017, Zhang and coworkers reported a synthesis of the halogenated terpene aplydactone (220) (Scheme 48).¹¹⁵ The complex molecule features a rigid core structure, including a [2]-ladderane embedded into the *trans*-decalin skeleton. The proposed biosynthetic pathway of this motif involves a photochemical [2+2] cycloaddition of the bis-alkene dactylone (221) to form the aptly named aplydactone.

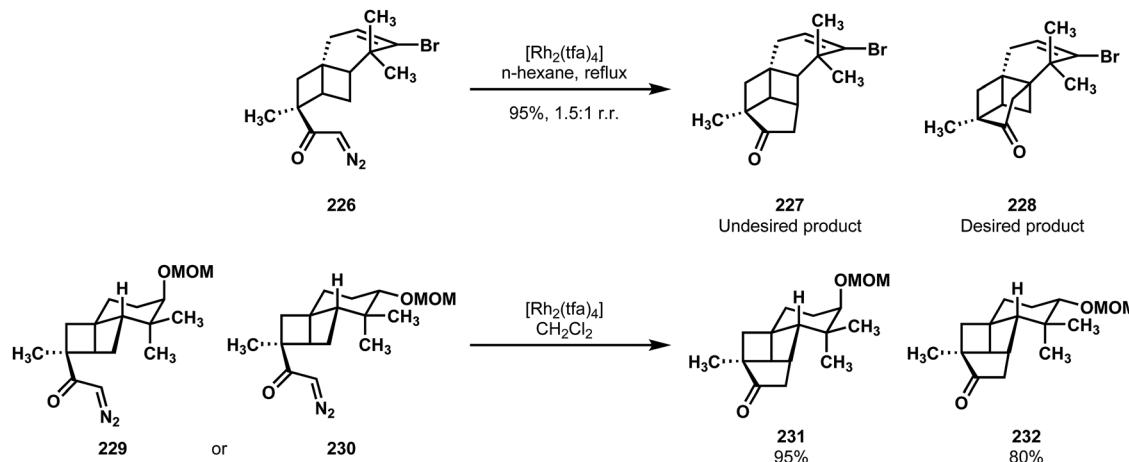
The Zhang group envisioned forming the cyclohexanone portion from an unusual 6-membered ring-forming metallocarbenoid C–H insertion. They imagined that this would be a difficult transformation, with competition between the desired insertion and potentially more sterically accessible positions which would form 4- or 5-membered rings. Drawing on the biosynthetic hypothesis, and in line with other syntheses that appeared in the year prior to their disclosure,^{116–118} Zhang and

coworkers forged the [2]-ladderane, or [2.2.0]-bicyclohexane, fused to the cyclohexane ring *via* an intramolecular [2+2] cycloaddition to form 223 (as a mixture of α and β diastereoisomers at the carbinol carbon), followed by a Wolff rearrangement of compound 224 to provide 225. After exchanging functional groups and appending the diazoketone, the late stage C–H insertion was evaluated.

Computational studies led the Zhang group to propose that an unsaturation in the 6-membered ring backbone would increase the preference for an insertion leading to a 6-membered ring over the more accessible 5-membered ring product. Looking solely at the backbone geometry, they noted that in the chair conformer, the necessary C–H bond points away from the carbon that would bear the diazo ketone, while in the twist-boat it points into the ladderane. In the unsaturated system, the boat would be the expected geometry, so they targeted such an intermediate. Indeed, they found that the vinyl bromide could be optimized to give the desired product 228 as a mixture with the undesired 1,5-insertion product (226 \rightarrow 227 + 228, Scheme 49), but that even the optimal insertion catalyst for reaction of 226 with either diastereoisomer of the saturated ether (229 or 230) failed



Scheme 48 The molecular structure of aplydactone (220) and its proposed biosynthesis from dactylone (221), as well as Zhang's creative entry into the molecular architecture.

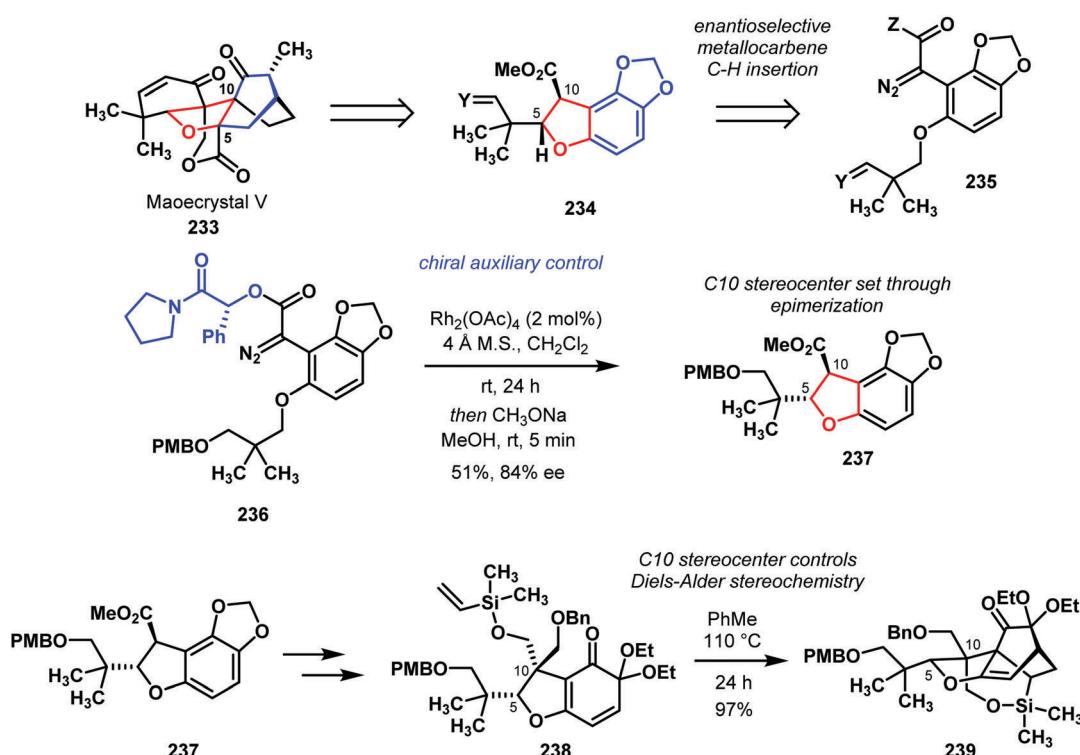


Scheme 49 Successful and unsuccessful attempts to forge the architecture of aplydactone (**220**) reported by Zhang, highlighting the critical influence of molecular geometry.

to provide the desired regioselectivity. Following this key step, reduction of the olefin provided the natural product in good yield.

Zhang's synthesis showcased an unusual implementation of dirhodium C–H insertion in a complex setting. By deferring a key bond formation to a late stage in the synthesis, they could construct the core architecture efficiently *via* reliable, traditional methods. Furthermore, while their synthesis was guided by computational methods, they demonstrate that geometrical assessments, which are a critical component of a chemist's analysis of selectivity, can still map well onto C–H functionalization in total synthesis.

As an objective for synthesis, maoecrystal V (**233**, Scheme 50) has captured the imaginations of some of the most creative synthetic chemists, and their collective efforts define a saga of success, despair, and fascinating chemistry. The natural product has an intricate architecture with variegated ring structures and fusions: the bicyclo[2.2.2]octan-2-one abuts a central strained tetrahydrofuran ring, which itself is fused to δ -valerolactone and cyclohexenone moieties. In 2014, the molecule finally succumbed to the onslaught of synthetic campaigns in the first enantioselective synthesis performed by way of a collaboration between the Zakarian and Davies groups.¹¹⁹



Scheme 50 The first reported enantioselective synthesis of maoecrystal V using a diastereoselective metallocarbene C–H insertion.

This enantioselective synthesis was built on a strategy that came to fruition in Zakarian's earlier, racemic synthesis of the natural product.¹²⁰ As part of their meticulous retrosynthetic analysis, the strained oxolane ring (highlighted in red in 233) was identified as the most difficult ring to construct: the contorted cycle is strung on all carbons with stereocenters, including three, contiguous, fully substituted stereogenic centers. Difficulties in the late stage *de novo* synthesis of this central five-membered heterocycle has curtailed many planned efforts to synthesize this prized target.^{121–129}

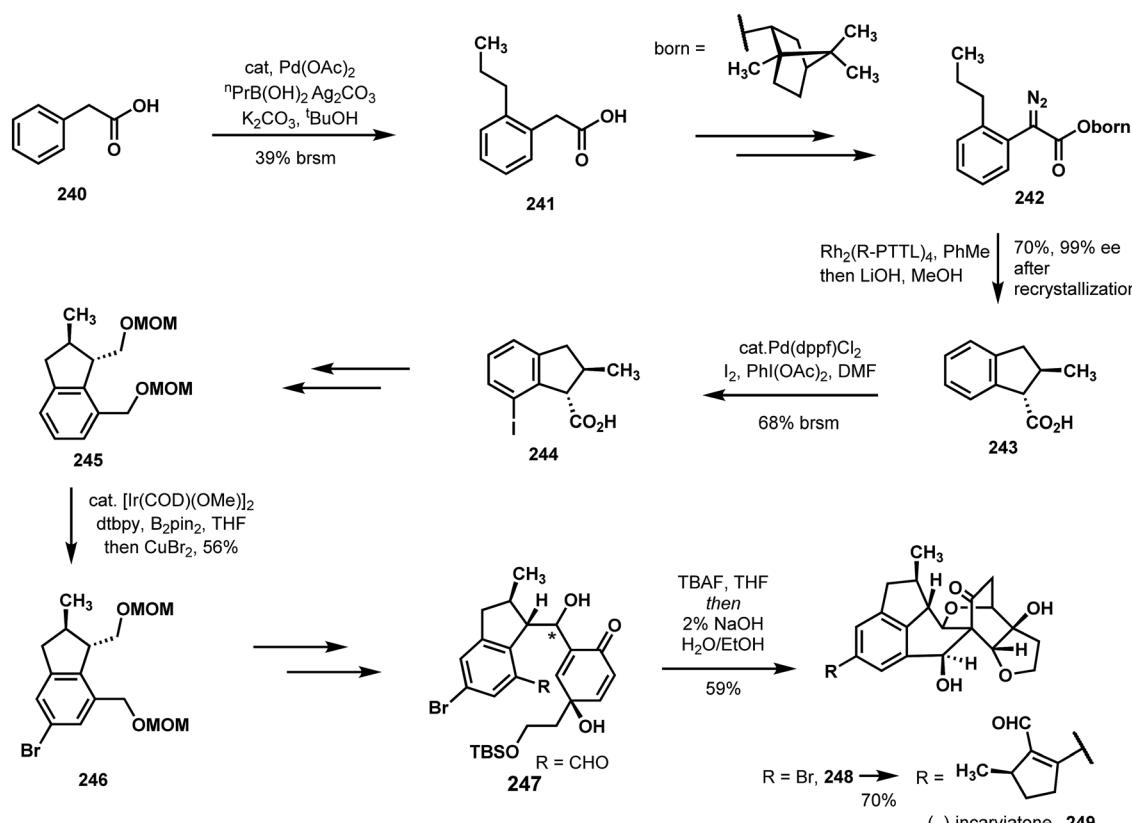
Their synthesis would address the challenge of forming the oxolane ring at an early stage of the effort. To achieve this key objective, they would turn to a stereoselective, rhodium carbene C–H insertion to establish the C5–C10 bond in a compound of type 235. They reasoned that if a compound of type 234 could be formed enantioselectively, then the rest of the synthesis could rely on diastereoselective reactions. Their earlier efforts demonstrated relative stereochemical control in a diastereoselective, racemic synthesis of dihydrobenzofuran 234 with an achiral dirhodium catalyst.

After significant screening of catalysts (both chiral and achiral), substrates 235 (variable substitution at Y and Z), solvents, and various reaction conditions, optimized conditions as shown in Scheme 49 furnished the desired benzohydrofuran 237 with acceptable yield and enantiomeric excess. Surprisingly, the use of chiral catalysts gave poor enantiomeric excess without the chiral auxiliary to influence the reaction outcome. When the chiral

auxiliary was in place, achiral Rh₂(OAc)₄ provided the greatest level of selectivity. Most important to this reaction was the stereochemistry at C5, from which the configuration at C10 could be established in a simple, base-mediated epimerization. As the synthesis proceeded, the importance of the C10 stereochemistry increased because the differentiated hydroxy methylene units bound to C10 controlled the important skeleton-forming Diels–Alder reaction through a tether (238 → 239). This C10 tether not only controlled the approach of the dienophile, but also provided the needed reactivity, since no cycloaddition was observed when the cycloaddition partners reacted in an intermolecular fashion. The two final carbocycles found in maoecrystal V were formed though an unprecedented 6-*exo* radical cyclization (lactone formation) and a ring-closing metathesis (cyclohexenone formation).

Strategically, an early metallocarbene C–H insertion reaction formed the challenging oxolane ring and set the important stereochemistry at C5; in turn, the [Rh₂]-catalyzed transformation became the enantiodetermining step of the completed enantioselective synthesis of maoecrystal V.

In 2015, the Lei group reported a synthesis of the potent monoamine oxidase inhibitor (−)-incarviatone (249, Scheme 51) featuring an array of C–H functionalization reactions¹³⁰ A scalable synthesis of this complex species would solve the currently low availability of the complex molecule (4.1 mg per 17 kg dried plants). To start the synthesis, the Lei group utilized the directing power of the carboxylic acid group on phenylacetic acid (240) to directly install a butyl group in the *ortho* position using a method



Scheme 51 A broad range of C–H functionalization chemistry used in the Lei synthesis of (−)-incarviatone.

developed by the Yu group.¹³¹ Subsequent attempts to perform an asymmetric, rhodium catalyzed 5-membered-ring closing C–H insertion were stymied by low enantioselectivity. However, when a chiral auxiliary was installed in combination with a matched chiral catalyst, the reaction proceeded smoothly and with high diastereoselectivity ($242 \rightarrow 243$).^{132–136} The authors noted that this was the first reported asymmetric catalytic synthesis of indane *via* intramolecular C–H insertion.

The Lei group demonstrated that the first step in their synthesis was achievable from either carboxylic acid **240** by C–H functionalization or from methyl (2-bromophenyl)acetate by traditional cross coupling chemistry. Despite the reduced efficiency of the bold C–H functionalization step, it is worth noting that the relative cost of **240** (\$0.04 per g) compared to the traditional cross-coupling precursor (>\$1 per g) made this direction enticing for a synthesis highly focused on availability of the final product. Analogous to the first C–H activation step, Lei again exploits a carboxylic acid to direct a C(sp²)–H iodination ($243 \rightarrow 244$), based on work by the Yu group.¹⁶ In addition to the versatility that the Yu C–H activation chemistry provides in this context, it also allows for protecting group free reactions.

Amazingly, the Lei group continued to utilize the power of C–H functionalization later in the synthesis. When bis-MOM ether **245** was subjected to Hartwig's C–H borylation conditions, the iridium-catalyzed C–H borylation took place *meta* to both substituents on the benzene ring. Cupric bromide (CuBr₂) was added in the same pot to convert the boronic ester to the aryl bromide **246**. After elaboration and installation of the main skeletal carbons, a remarkable, one-pot biomimetic cascade cyclization process was triggered by desilylation and aqueous basic conditions to yield aryl bromide **248**. This classical reaction complemented the power of C–H functionalization chemistry well: the stereocenters set earlier by rhodium-catalyzed carbene C–H insertion served to effectively control the stereochemical outcome of this cyclization. Suzuki cross-coupling of the aryl bromide with the appropriate cyclopentenyl boronic ester furnished (–)-incarviatone in 70% yield.

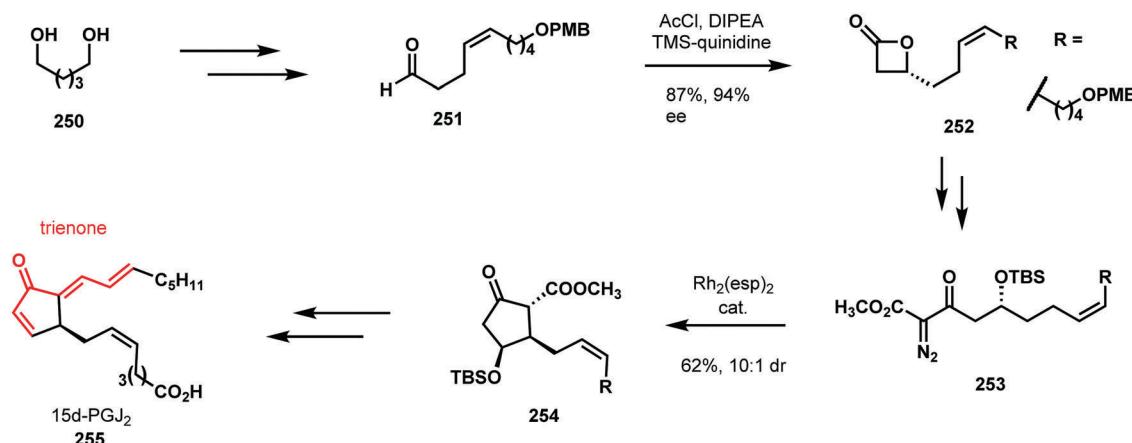
C–H functionalization has also contributed to the rich field of prostaglandin total synthesis. In 2015, the Carreira group

utilized a dirhodium-catalyzed carbene C–H insertion to prepare the central cyclopentenone carbocycle of the structurally and biologically unique prostaglandin 15d-PGJ₂ (Scheme 52).¹³⁷ This strategy permitted the construction of the central cyclic enone from a linear substrate **253**, eschewing the common approach wherein a preformed five-membered ring is modified. Their synthesis of **253** featured Nelson's enantioselective [2+2] ketene cycloaddition (**251** → **252**), which also determined the stereochemical development of the synthesis.¹³⁸ The β-stereocenter set in this fashion worked in concert with a Rh₂(esp)₂ catalyzed carbene C–H insertion to deliver the substituted cyclopentanone **254** in a highly diastereoselective manner. Strategically, the β-TBS ether in **254** was utilized to control the diastereoselectivity of the C–H insertion cyclization, and, after this effort was concluded, the β-TBS ether was eliminated to unveil the endocyclic enone moiety through β-elimination.

The diastereoselective, intramolecular C–H insertion of diazo keto ester **253** also highlights the chemoselectivity of this cyclization technique: metallocarbenes can react with nearby alkenes by way of cyclopropanation or with proximal heteroatoms, but, in this case, the five-membered ring formation outcompeted either of these possibilities. And finally, the reaction displayed regioselectivity as well; even in the presence of activated ethereal and allylic C–H bonds, homoallylic C–H insertion occurs preferentially relative to cyclobutanone or cyclohexanone formation. After cyclization, the Carreira group carried the C–H insertion product forward to complete the synthesis of 15d-PGJ₂ and several analogues. The strategy that guided this notable achievement in synthesis was based on previous work by the Carreira group to prepare the structurally related epoxyisoprostanes.¹³⁹

Nitrenes

The Movassagh group published several syntheses of members of the oligocyclotryptamine family and related compounds (256–258, Fig. 6).^{140–142} Their key insight has been the formation of diazene linkages between either the C3a–C3a' carbons, or C3a–C7' carbons, followed by light-induced loss of molecular



Scheme 52 Highlights of Carreira's total synthesis of prostaglandin 15d-PGJ₂.

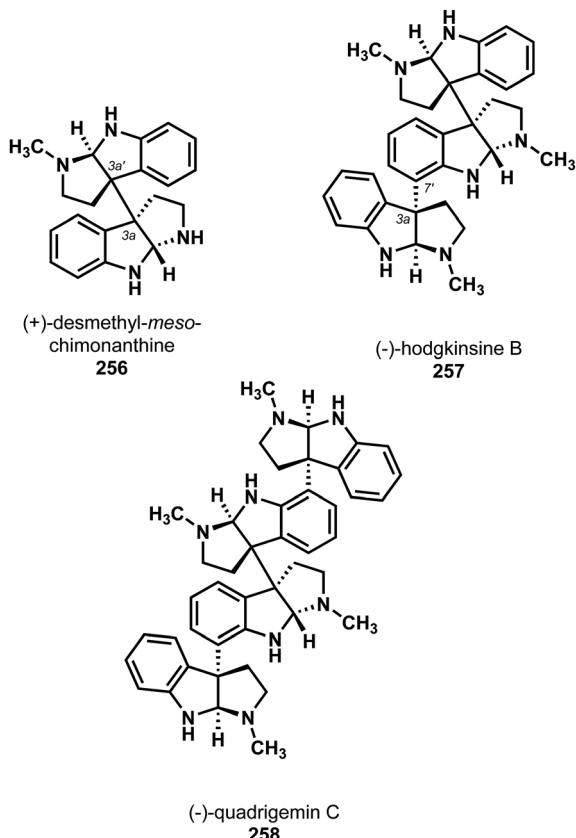


Fig. 6 The molecular structure of (+)-desmethyl-meso-chimonanthine (256), (−)-hodgkinsine B (257), and (−)-quadrigemin C (258).

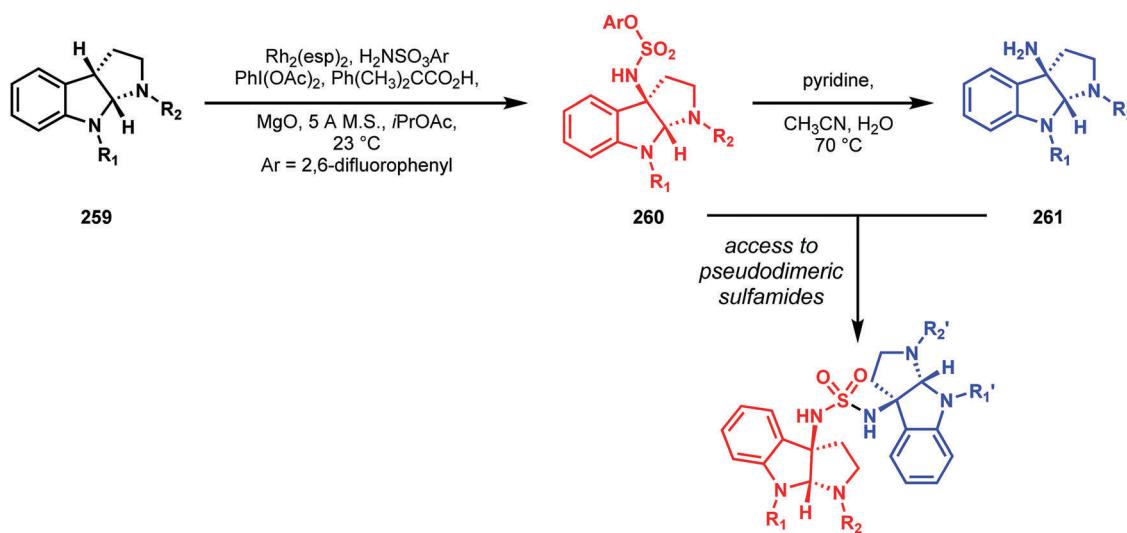
nitrogen and radical recombination to form the quaternary stereocenter(s) in high enantiomeric excess. C–H functionalization has served to introduce key nitrogen atoms in many of these synthetic efforts.

The Movassaghi group's initial disclosures of this method involved formation of the diazene linkage from a mixed

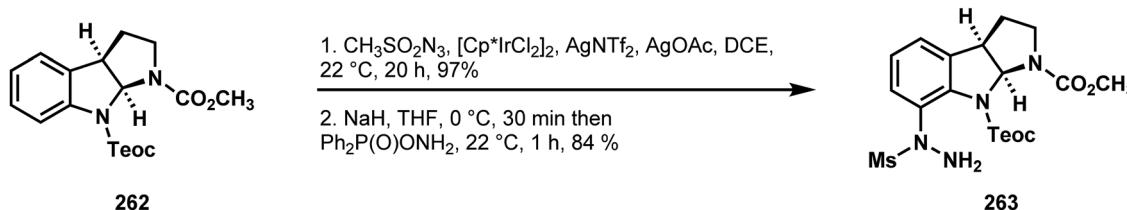
sulfamide, with subsequent oxidative formation of the diazene. However, their early route necessitated forming a sulfamoyl chloride such that the mixed sulfamide could be accessed.¹⁴³ In their study toward calycanthidine, *meso*-chimonanthine, and desmethyl-*meso*-chimonanthine (256), they described the use of dirhodium catalyzed C–H amination, which directly provided the aryl sulfamate esters of type 260.¹⁴⁴ This sulfamate ester could then be hydrolyzed to give the simple amine of type 261. Interestingly, by reacting an aryl sulfamate ester 260 of one scaffold with the free amine 261 of another, they could form a variety of mixed sulfamides, providing rapid access to various mixed diazenes to explore the family of cyclotryptamine dimers. While this strategy provided rapid access to the C3a–C3a' bond formation in various subsequent syntheses of members of the oligocyclotryptamine family, the ability to rapidly assemble regio- and stereochemical isomers in particular allowed the Movassaghi group to assign the absolute stereochemistry of desmethyl-*meso*-chimonanthine (256) (Scheme 53).¹⁴⁰

In subsequent studies, Movassaghi and coworkers explored the trimer and tetramer landscape, which possess sp^2 – sp^3 linkages as well (see 257 and 258). After developing a method by which to trap benzylic cations with aryl hydrazines to give immediate access to aryl alkyl diazenes, they explored methods by which to form the requisite aryl hydrazine. They found that iridium catalyzed amination¹⁴⁵ (262 → 263, step 1, Scheme 54), followed by oxidation of the mesylaniline to the mesylarylhydrazine gave them rapid access to the C7–C3' diazene linkages, which were utilized in syntheses of multiple natural products, including (−)-hodgkinsine B (257) and (−)-quadrigemin C (258), shown above.¹⁴²

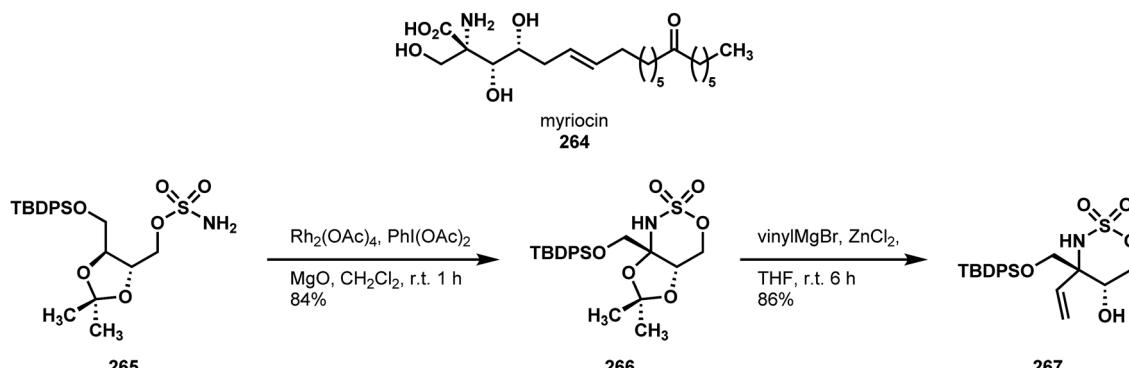
In their exploration of the oligocyclotryptamines, the Movassaghi laboratory demonstrated highly concise routes to their desired intermediates by way of C–H functionalization reactions. They obviated the need to pass through reactive species such as sulfamoyl chlorides and provided ready access to chemical scaffolds that could have been difficult to obtain otherwise.



Scheme 53 The benzylic C–H amination strategy employed by Movassaghi in their syntheses of 256–258 and related compounds.



Scheme 54 The directed C–H amination strategy employed by Movassaghi to forge the C3a–C7' linkage in (−)-hodgkinsine B (**257**) and (−)-quadrigemin C (**258**) and related compounds.



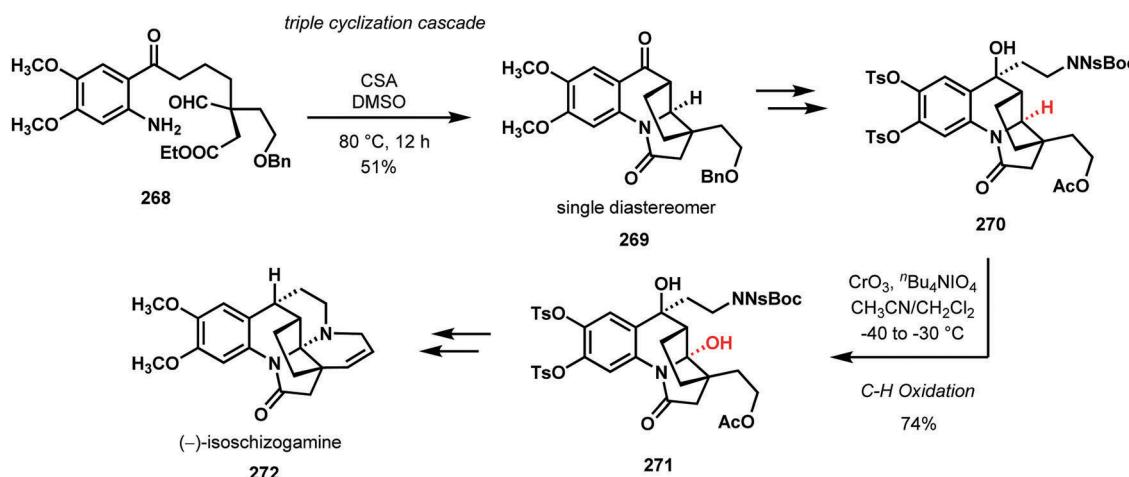
Scheme 55. A C–H amination strategy developed by Yakura for a synthesis of myriocin (**264**)

The Yakura laboratory reported a synthesis of myriocin **264** (Scheme 55) featuring a site-selective C–H amination to set a fully substituted nitrogen-bearing stereocenter.¹⁴⁶ Yakura and coworkers imagined that myriocin could be formed *via* olefin cross-metathesis, dividing the molecule into two halves, and set out to selectively synthesize the stereocenter-containing half of the molecule. They imagined that the α,α -disubstituted amino acid could be synthesized *via* a directed nitrene C–H insertion followed by aminal opening and attack on the iminium. Although Du Bois first reported this strategy and used it in their synthesis of saxitoxin,¹⁴⁷ Yakura and coworkers state that it had not been used to synthesize fully substituted stereocenters prior to their work. After synthesizing the requisite

sulfonamide **265** from a chiral derivative of diethyl tartarate, they exposed it to a dirhodium catalyst and (diacetoxymethyl)benzene and obtained the stereospecific C-H insertion product **266** in good yield. Vinylation of the latent iminium ion then occurred with good stereoselectivity, which they propose to proceed *via* coordination to the vinylzinc reagent by the free alcohol in analogy to Du Bois's work. Compound **267** was subsequently advanced to myriocin (**264**).

Oxos

The Tokuyama group from Tohoku University completed a pleasing synthesis of (*-*)-isoschizogamine (272, Scheme 56) by employing a late stage C–H oxidation of a carbon adjacent



Scheme 56 C–H oxidation in the synthesis of (–)-isoschizogamine

to an amide nitrogen.¹⁴⁸ This seemingly simple oxidation enabled a triple cyclization cascade (aldol condensation/aza-conjugate addition/lactamization, **268** → **269**) to construct the majority of the natural product's multicyclic framework. The group studied several conditions for the C–H oxidation on a small model system and discovered that ruthenium-based catalysts, previously used for the oxidation of amines, were unsuccessful. Chromium based oxidants gave some reactivity, and, after some additional screening, the Fuch's protocol of CrO₃ and ⁷Bu₄IO₄ was discovered to be the most effective for the desired transformation (**270** → **271**). An interesting comment made by Tokuyama coworkers was the need for the arene ring to be deactivated by tosylation of the phenolic oxygens for a selective oxidation to occur; this observation highlights some of the unexpected problems that can arise in efforts to achieve site-selective C–H functionalizations in complex architectures replete with different types of C–H bonds. The ultimately successful hydroxylation of the highlighted C–H bond in **270** permitted a Lewis acid-assisted cyclic aminal formation to complete five of the six rings in the product; a subsequent ring-closing metathesis completed the synthesis of (−)-isoschizogamine (**272**).

In 2016, the Baran laboratory described their impressive synthesis of phorbol (**273**, Scheme 57).¹⁴⁹ Their retrosynthetic design was centered on a bioinspired cyclase phase/oxidase phase strategy. By doing so, they could utilize an intermediate they had synthesized in their pursuit of ingenol,¹⁵⁰ which is believed to be related to phorbol in a biosynthetic relationship as well. Starting with their known intermediate **274**, it was necessary to install oxidation at C12 and C13, which is absent in the ingenol structure. To this end, they devised a strategy where C–H oxidation would initiate a series of steps to install the vicinal diol substitution. Trifluoromethylmethyl dioxirane showed high selectivity for C12 methylene oxidation over tertiary C–H bonds and other methylenes (**276** → **277**, Scheme 57). From here they could rupture the cyclopropane, effectively extending the

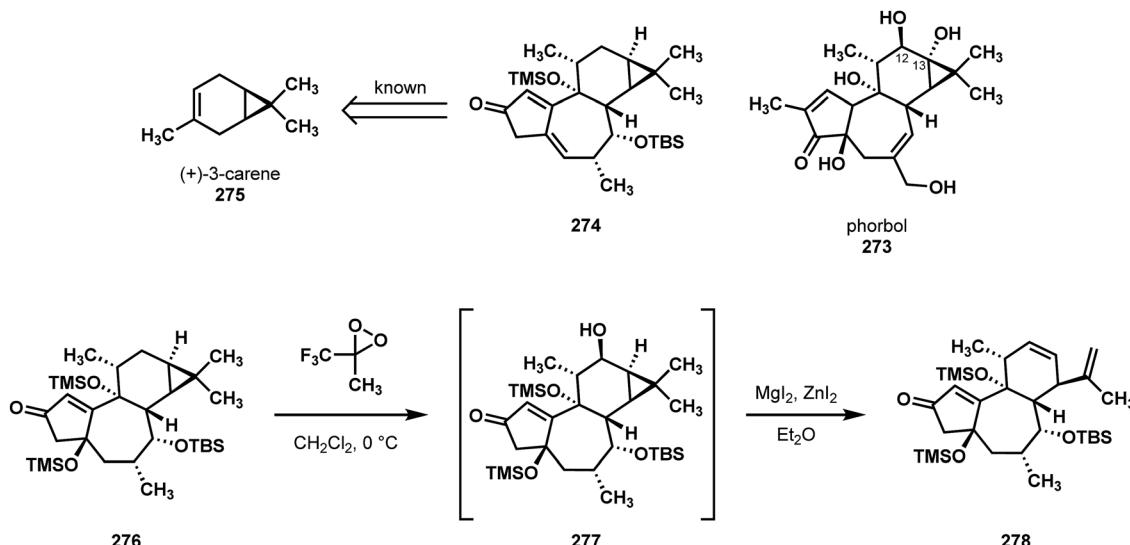
functionality to C13 in the form of a double bond (**278**). This allowed for the installation of the requisite oxidation, followed by reformation of the cyclopropane, and installation of the remaining oxidation pattern in a more traditional manner.

C–H functionalization *via* radicals

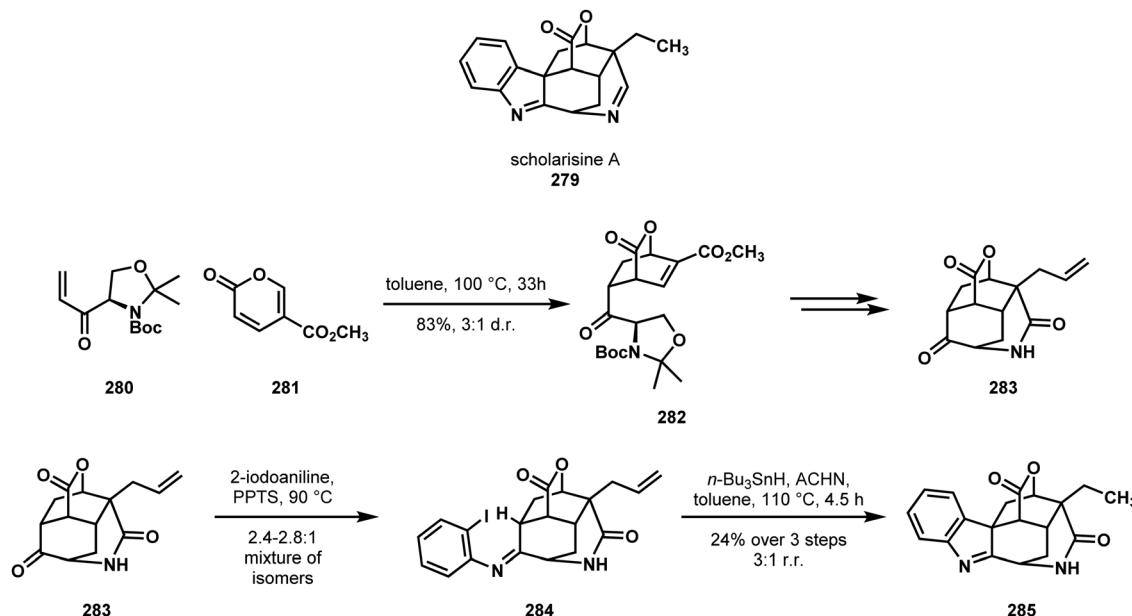
As discussed in Baran's earlier review, many of the original C–H functionalization reactions used in classical syntheses were based on free radical reactions.⁹ Hoffman–Löffler–Freytag and Barton nitrite ester reactions, in the right structural and geometrical contexts, were able to rapidly deliver structural complexity that far outpaced many of the available forms of reactivity. Decades later, reactions proceeding through radical intermediates are still of great utility. While these reactions often do not benefit from metal catalysts that can enforce a high degree of regio- or stereoselectivity, understanding the mechanistic constraints on hydrogen atom transfer (HAT) can allow the disciplined chemist to invoke these reactive intermediates in the creative process of retrosynthetic analysis.

In 2013, Snyder reported a synthesis of the indole alkaloid scholarisine A (**279**, Scheme 58).¹⁵¹ A key retrosynthetic disconnection was based on the idea that the cyclic ketimine of the target could arise *via* an unprecedented, intramolecular C–H arylation of a non-enolizable *N*-aryl ketimine (e.g. **284** → **285**). Snyder and coworkers report that the oxabicyclo[2.2.2]octane **282** was readily formed *via* an *endo*-diastereoselective Diels–Alder reaction of **280** with pyrone derivative **281**. The core cage structure **283** was then smoothly synthesized *via* a radical cyclization and trapping with an allyl equivalent.

This central intermediate was then condensed with 2-iodoaniline to give a mixture of imine geometrical isomers **284** (only one shown for clarity). Exposure of the geometrically isomeric imines to tributyltin radical effected sequential single-electron dehalogenation, 1,5-hydrogen atom abstraction, and a final radical arene cyclization



Scheme 57 A C–H oxidation that allowed Baran to access phorbol (**273**) using intermediate **274** from their synthesis of ingenol.



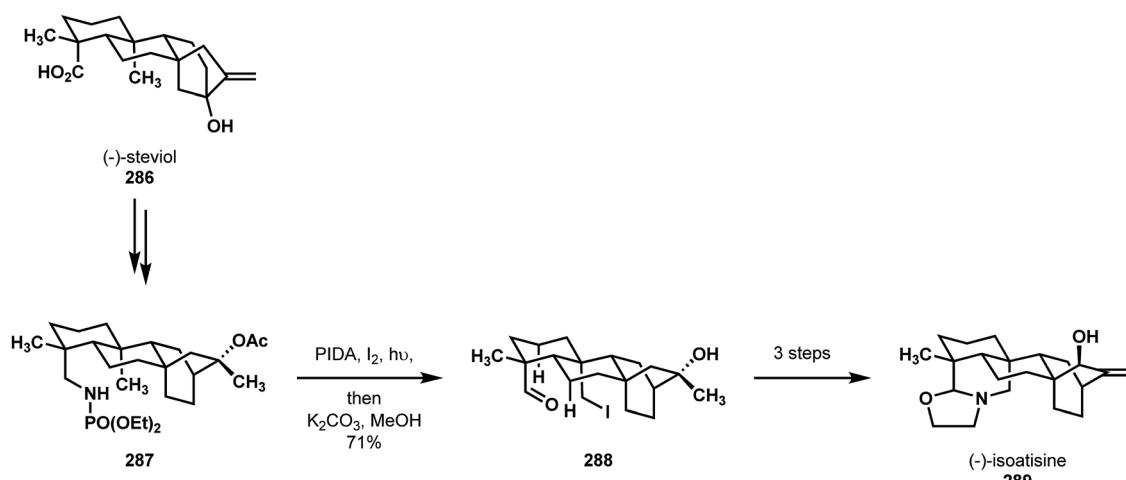
Scheme 58 A site-selective radical arene cyclization reaction used by Snyder in a synthesis of scholarisine A (279).

to give 285, a compound bearing the full polycyclic skeleton of the natural product. From 285, a straightforward functional group conversion completed their synthesis of scholarisine A (279). Of note, despite the similarity of the two tertiary C–H bonds flanking the imine, the cyclization gives 3:1 regioselectivity in favor of the desired regioisomer. Additionally, although this series of transformations exhibited reduced efficiency, the authors report that all attempts to pre-install the arene prior to this step were unsuccessful. Indeed, this example displays not just the power of C–H functionalization in relation to other methods, but also the appeal of carrying a simple hydrogen atom through much of the synthesis.

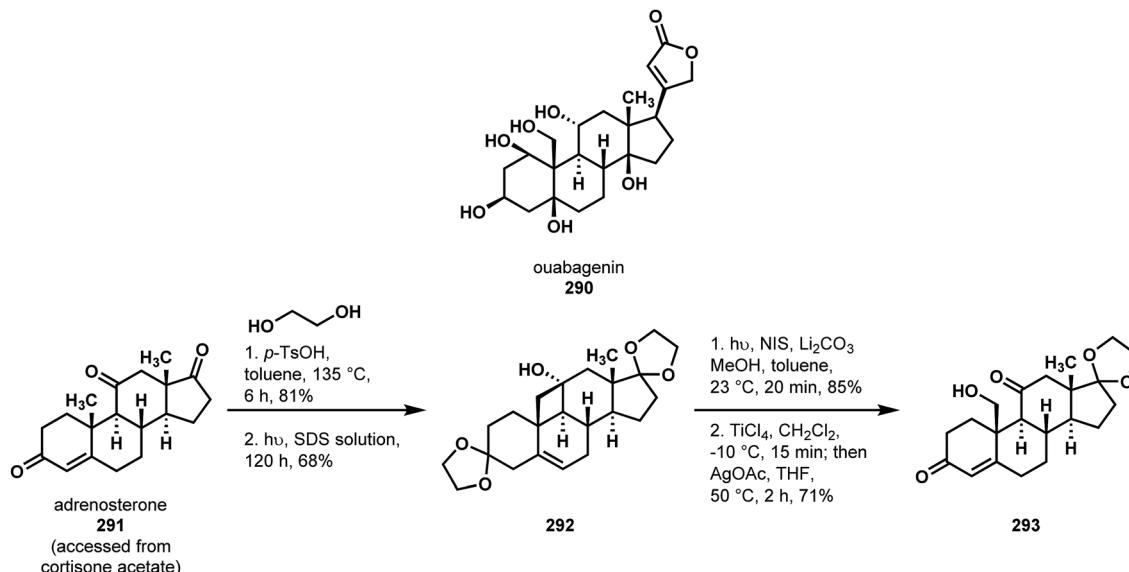
In 2014, Baran reported efforts toward the *ent*-atisane diterpenes and their relatives, including a synthesis of (–)-isoatisine (289, Scheme 59).¹⁵² Their retrosynthetic design features a divergent strategy wherein a central intermediate derived from

steviol 286 (itself accessible from the commercial sweetener stevioside) would provide access to the hetisine, hetidine, and atisine skeletons through several C–H functionalization steps. Toward (–)-isoatisine (289), they envisioned forming the C–N bond through Suárez/HLF-type conditions followed by eventual displacement of the halide by the appended nitrogen atom. From (–)-steviol (286), they formed the phosphoramidate 287, which they found to be uniquely successful in the selective 1,7-hydrogen atom transfer reaction to provide 288. The reaction displayed regioselectivity over the 1,6-HAT at the nearby methylene positions (shown in figure). Following this, they could form (–)-isoatisine in three additional steps.

The Baran laboratory also reported a semisynthesis of ouabagenin (290) based on a “redox relay” strategy, beginning with the accessible steroid adrenosterone (291, Scheme 60).¹⁵³



Scheme 59 Baran's creative entry into the (–)-isoatisine (289) skeleton from (–)-steviol (286) via a radical C–H halogenation.



Scheme 60 Two-step C–H hydroxylation in a synthesis of ouabagenin (**290**) reported by Baran.

Their retrosynthetic design featured a relay of oxidation around the molecule, using the three initial carbonyl groups to establish the six carbinols in the natural product. Their first such bond formation began with C–H oxidation of one of the two angular methyl groups. Using Norrish type II chemistry, they found that they could selectively forge the C19 cyclobutanol **292**, which, despite unfavorable relative bond strengths, could then be fragmented to form the carbon–iodine bond which was subsequently hydrolyzed to **293**. This hydroxymethyl group was then instrumental in their redox relay strategy later in the synthesis to introduce the A-ring hydroxyl groups in a diastereoselective fashion.

C–H functionalization via biocatalysis

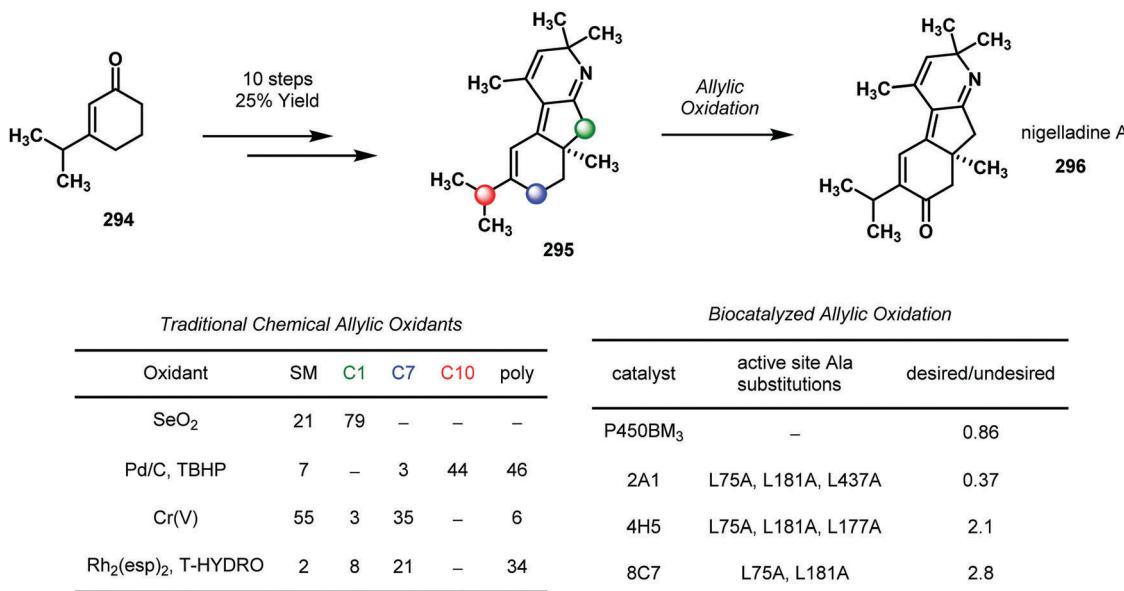
In target directed synthesis, the use of enzymes as reaction catalysts has not been widespread. However, the field of biocatalysis is constantly developing catalysts with new modes of reactivity and for substrates that look less like the native biomolecules and more like the average synthetic intermediate. Enzyme pockets offer high control over the orientation of the substrate with respect to the reactive species, directly mapping onto regiochemistry and stereochemistry. As C–H oxidation is already a critical reaction performed endogenously by large classes of enzymes, the use of biocatalysis to effect new C–H functionalization reactions has experienced close examination. While only a few examples are featured below, we anticipate that the use of enzymes to rapidly generate complexity in total synthesis endeavors will be the subject of much study in the coming years.

The Stoltz group utilized their expertise in setting quaternary stereocenters to formulate an efficient synthesis of nigelladine A (**296**, Scheme 61).¹⁵⁴ They expected to use a late stage allylic oxidation of **295** to complete the molecule by directly converting

the C7 methylene to a ketone; however, traditional synthetic methods failed to provide selectivity in the oxidation. With several, activated allylic C–H bonds, compound **295** is a challenging context in which to achieve a selective C–H oxidation event, and, in fact, they observed oxidation at C1, C7, C10, as well as polyhydroxylation. A subsequent collaboration with the Arnold group revealed a new strategy for achieving the desired site-selectivity in the key C–H oxidation step: through a P450-mediated enzymatic C–H oxidation, they believed that appropriate modification of the active site would allow for the enzyme to accept the unnatural substrate and induce oxidation at the desired site in **295**. Typically, optimization of P450 enzymes for biocatalytic transformations of unnatural substrates requires a large amount of substrate and extensive screening.^{155–157} In the case of a late-stage intermediate in a total synthesis effort, with limited material, biocatalysis would not be the planned reagent. Remarkably, the teams of Arnold and Stoltz were able to utilize a focused screen of P450 variants to rapidly close in on the desired reactivity. This novel transformation yielded 160 mg of the desired C7 allylic alcohol, which was subsequently advanced to nigelladine A (**296**) in 21% yield (43% brsm) through an oxidation with Dess–Martin periodinane (DMP).

This collaborative synthesis of nigelladine A impacts the broader field of organic synthesis by providing a promising strategy for gaining control over the regiochemical outcome of a challenging C–H allylic oxidation. The risk associated with a planned strategy of late-stage C–H oxidation can be reduced by the possible use of biocatalysis. With the increasing use of biocatalysis in synthesis has come greater knowledge that may enable further use of the tool without the need for the extensive screens of directed evolution.

In 2017, Renata and coworkers reported a formal synthesis of manzacidin C (**297**, Fig. 7) using a two-fold C–H functionalization strategy.¹⁵⁸ The authors had been studying the use of iron- and α -ketoglutarate-dependent enzymes, specifically those



Scheme 61 Selective allylic C–H oxidation through biocatalysis by Stoltz and Arnold: synthesis of nigelladine A.

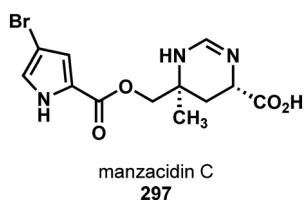


Fig. 7 The molecular structure of manzacidin C (297).

which act as amino acid hydroxylases, to form non-canonical amino acids. They found that GriE, one such enzyme that acts as a leucine 5-hydroxylase, demonstrated a wide substrate scope. With this in mind, they devised a strategy to a known precursor of manzacidin C *via* geminal difunctionalization of L-leucine (298, Scheme 62). First, inspired by the work of Britton, they found that leucine would undergo a photocatalytic C–H azidation reaction, forming 299.¹⁵⁹ They then exposed 299 to the biocatalytic conditions, where it underwent hydroxylation to full conversion. Reduction of the azide, followed by Boc-protection-induced lactonization gave known intermediate 301, completing the 5-step formal synthesis.

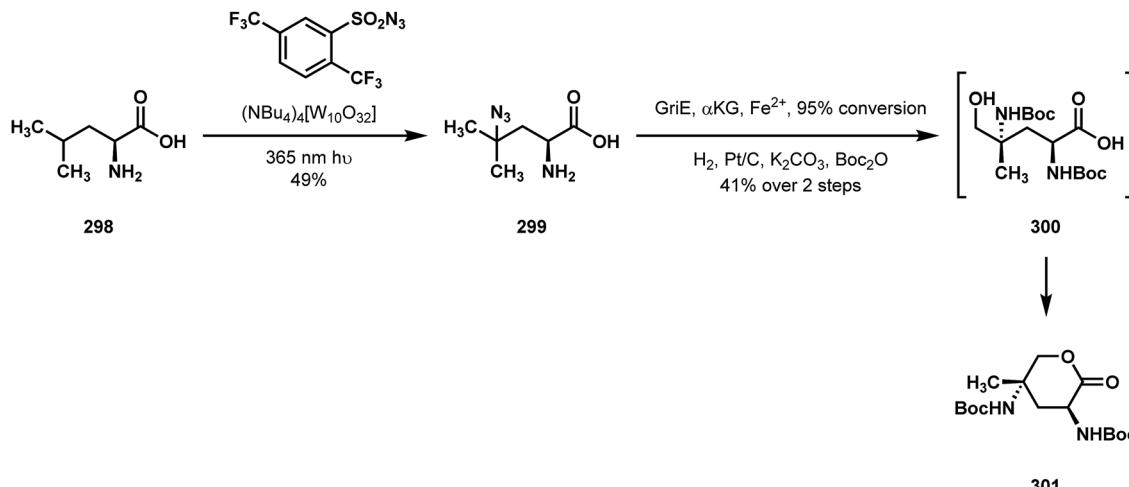
Renata and coworkers comment on the limitations of known methods to functionalize canonical amino acids, specifically that not all positions along the side chain can be functionalized with high selectivity. They found that their enzymatic conditions provided complimentary regioselectivity to existing amino acid functionalizing reactions. The high stereoselectivity and diverse substrate scope of their new reaction allowed them to utilize this 2-fold functionalization strategy.

In 2018, the Renata laboratory also reported a synthesis of the non-ribosomal peptide natural product tambromycin, (302, Scheme 63).¹⁶⁰ The molecule consists of four non-canonical, highly modified amino acids: 2 equivalents of 2-methyl serine, a

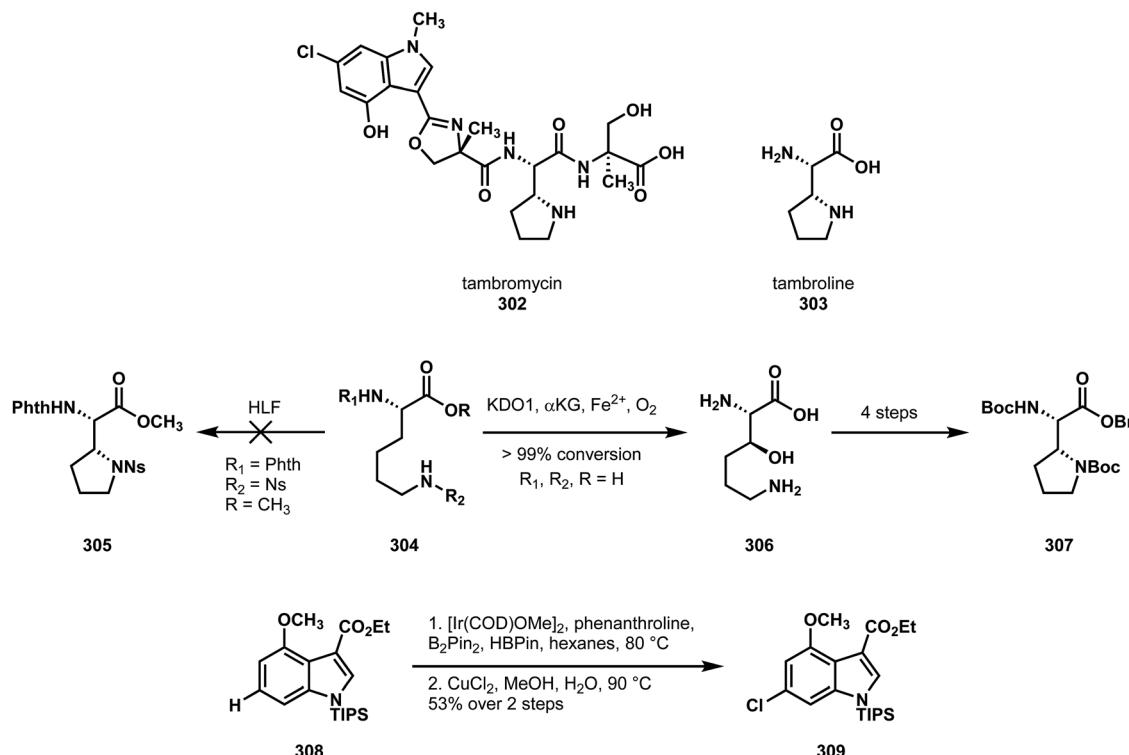
3,4,6-substituted indole, and 2-amino-2-(pyrrolidin-2-yl)acetic acid, also known as tambroline. Biosynthetically, tambroline (303) is thought to arise from the dehydrogenation of serine, and subsequent intramolecular 5-*exo-trig* cyclization to forge the pyrrolidine. However, the enzymes that perform this synthetic function are currently unknown. Renata and coworkers envisioned forging both the functionalized indole and tambroline moieties *via* C–H functionalization disconnections, leading them back to readily available materials.

Renata and coworkers sought to produce tambromycin (302) on large scale, and so the existing solutions, which either featured wasteful and dangerous reagents or were not enantio- or diastereoselective, were unsuited for their purposes. Initially, they sought a chemical route to tambroline, however, they found that a differentially protected lysine, when exposed to Hofmann–Löffler–Freitag conditions, failed to give cyclization (304 → 305). Instead, they studied the 3-hydroxylation of lysine *via* an iron- and α-ketoglutarate dependent enzyme, the lysine hydroxylase KDO1. They found that, upon optimization of expression level, a gram-scale conversion of lysine (304) to 3-hydroxylysine (306) could be achieved with less than a liter of cell lysate, with perfect regio- and stereoselectivity. Their attempts to then employ a separate biocatalytic transformation to dehydrate and cyclize 3-hydroxylysine to tambroline, as in the proposed biosynthesis, failed however. Instead, they enlisted a four-step chemical synthesis, which yielded protected tambroline (306) in good yield on gram scale.

Subsequently, they sought to synthesize the 3,4,6-substituted indole, 309, using known methods for 4-functionalization of indoles using thallium; however, these methods were unsuited for scale up, so they turned to C-6 functionalization. They found that C6 borylation of accessible 3,4-substituted indole derivative 308 provided access to the necessary fragment, based on the method developed



Scheme 62 A biocatalytic C–H oxidation in Renata's route to mazacidin C (297).



Scheme 63 Biocatalytic C–H oxidation and metal-catalyzed C–H halogenation in Renata's synthesis of tambromycin (302).

by Baran toward fumitremorgin.¹⁰⁴ After coupling the indole and tambroline fragments with 2 equivalents of 2-methylserine, they were able to deliver the natural product tambromycin (302). This same indole borylation strategy was subsequently used by Thomson and coworkers in their synthesis of tambromycin.¹⁶¹

In Renata's synthesis, C–H functionalization serves to provide rapid and selective access to intermediates in a scalable process with full control over stereochemical outcome. This achievement offers a creative and complementary blending of

enzyme-mediated biocatalytic and transition metal-mediated transformations in a challenging undertaking in synthesis.

Conclusion

New methods of C–H activation are rapidly being adopted by the total synthesis community to furnish efficient syntheses of complex architectures. With this chemistry comes a new logic to disconnections in retrosynthesis. This “new logic” comprises a pattern

recognition that is pushing the creative limits of synthetic chemistry. In this review we have sought to demonstrate the tenets of this pattern recognition through the analysis of total syntheses that have successfully applied C–H functionalization in preparations of natural products.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

We would like to thank Princeton University and the National Science Foundation CCI Center for Selective C–H Functionalization (CHE-1700982) for sponsoring and inspiring our studies in C–H functionalization.

References

- H. M. L. Davies and D. Morton, *Chem. Soc. Rev.*, 2011, **40**, 1857–1869.
- I. J. S. Fairlamb and A. R. Kapdi, in *Strategies for Palladium-Catalyzed Non-Directed and Directed C–H Bond Functionalization*, ed. A. R. Kapdi and D. Maiti, Elsevier, 1st edn, 2017, pp. 9–48.
- R. M. Wilson and S. J. Danishefsky, *J. Org. Chem.*, 2007, **72**, 4293–4305.
- E. J. Corey and X.-M. Cheng, *The Logic of Chemical Synthesis*, John Wiley & Sons, Inc., New York, 1995.
- Y. Qiu and S. Gao, *Nat. Prod. Rep.*, 2016, **33**, 562–581.
- T. Yakura and H. Nambu, *Tetrahedron Lett.*, 2018, **59**, 188–202.
- P. B. Brady and V. Bhat, *Eur. J. Org. Chem.*, 2017, 5179–5190.
- L. McMurray, F. O'Hara and M. J. Gaunt, *Chem. Soc. Rev.*, 2011, **40**, 1885–1898.
- W. R. Gutekunst and P. S. Baran, *Chem. Soc. Rev.*, 2011, **40**, 1976–1991.
- J. Yamaguchi, A. D. Yamaguchi and K. Itami, *Angew. Chem., Int. Ed.*, 2012, **51**, 8960–9009.
- J. A. Johnson and D. Sames, *J. Am. Chem. Soc.*, 2000, **122**, 6321–6322.
- L. V. Desai, K. L. Hull and M. S. Sanford, *J. Am. Chem. Soc.*, 2004, **126**, 9542–9543.
- M. S. Chen and M. C. White, *Science*, 2007, **318**, 783–787.
- R. Y. Zhu, L. Y. Liu and J. Q. Yu, *J. Am. Chem. Soc.*, 2017, **139**, 12394–12397.
- J. He, Q. Shao, Q. Wu and J. Q. Yu, *J. Am. Chem. Soc.*, 2017, **139**, 3344–3347.
- T. S. Mei, D. H. Wang and J. Q. Yu, *Org. Lett.*, 2010, **12**, 3140–3143.
- J. He, M. Wasa, K. S. L. Chan, Q. Shao and J.-Q. Yu, *Chem. Rev.*, 2017, **117**, 8754–8786.
- J. M. E. Hughes and J. L. Gleason, *Angew. Chem., Int. Ed.*, 2017, **56**, 10830–10834.
- J. M. E. Hughes and J. L. Gleason, *Tetrahedron*, 2017, **74**, 759–768.
- Q. Ye, P. Qu and S. A. Snyder, *J. Am. Chem. Soc.*, 2017, **139**, 18428–18431.
- M. E. McCallum, C. M. Rasik, J. L. Wood and M. K. Brown, *J. Am. Chem. Soc.*, 2016, **138**, 2437–2442.
- J. I. Concepción, C. G. Francisco, R. Hernández, J. A. Salazar and E. Suárez, *Tetrahedron Lett.*, 1984, **25**, 1953–1956.
- S. E. Ammann, G. T. Rice and M. C. White, *J. Am. Chem. Soc.*, 2014, **136**, 10834–10837.
- C. M. Rasik and M. K. Brown, *Angew. Chem., Int. Ed.*, 2014, **53**, 14522–14526.
- M. A. Bigi, S. A. Reed and M. C. White, *J. Am. Chem. Soc.*, 2012, **134**, 9721–9726.
- K. Hung, M. L. Condakes, T. Morikawa and T. J. Maimone, *J. Am. Chem. Soc.*, 2016, **138**, 16616–16619.
- A. Tenaglia, E. Terranova and B. Waegell, *J. Org. Chem.*, 1992, **57**, 5523–5528.
- P. Brun and B. Waegell, *Tetrahedron*, 1976, **32**, 1137–1145.
- K. H. Baggaley, H. Erdtman and T. Norin, *Tetrahedron*, 1968, **24**, 3399–3405.
- H. Miyake, A. Nishimura, M. Yago and M. Sasaki, *Chem. Lett.*, 2007, **36**, 332–333.
- M. L. Condakes, K. Hung, S. J. Harwood and T. J. Maimone, *J. Am. Chem. Soc.*, 2017, **139**, 17783–17786.
- M. Malik, G. Witkowski and S. Jarosz, *Org. Lett.*, 2014, **16**, 3816–3819.
- K. J. Fraunhofer, N. Prabagaran, L. E. Sirois and M. C. White, *J. Am. Chem. Soc.*, 2006, **128**, 9032–9033.
- M. Malik, G. Witkowski, M. Ceborska and S. Jarosz, *Org. Lett.*, 2013, **15**, 6214–6217.
- D. A. Siler, J. D. Mighion and E. J. Sorensen, *Angew. Chem., Int. Ed.*, 2014, **53**, 5332–5335.
- R. J. Sharpe and J. S. Johnson, *J. Org. Chem.*, 2015, **80**, 9740–9766.
- L. V. Desai, K. L. Hull and M. S. Sanford, *J. Am. Chem. Soc.*, 2004, **126**, 9542–9543.
- Z. Meng, H. Yu, L. Li, W. Tao, H. Chen, M. Wan, P. Yang, D. J. Edmonds, J. Zhong and A. Li, *Nat. Commun.*, 2015, **6**, 4–11.
- A. H. Trotta, *J. Org. Chem.*, 2017, **82**, 13500–13516.
- R. A. Leal, C. Bischof, Y. V. Lee, S. Sawano, C. C. McAtee, L. N. Latimer, Z. N. Russ, J. E. Dueber, J. Q. Yu and R. Sarpong, *Angew. Chem., Int. Ed.*, 2016, **55**, 11824–11828.
- Q. Nguyen, K. Sun and T. G. Driver, *J. Am. Chem. Soc.*, 2012, **134**, 7262–7265.
- K. S. L. Chan, M. Wasa, L. Chu, B. N. Laforteza, M. Miura and J.-Q. Yu, *Nat. Chem.*, 2014, **6**, 146–150.
- D. Dailler, G. Danoun and O. Baudoin, *Angew. Chem., Int. Ed.*, 2015, **54**, 4919–4922.
- J. Sofack-Kreutzer, N. Martin, A. Renaudat, R. Jazzaar and O. Baudoin, *Angew. Chem., Int. Ed.*, 2012, **51**, 10399–10402.
- Q. Zhang, K. Chen, W. Rao, Y. Zhang, F.-J. Chen and B.-F. Shi, *Angew. Chem., Int. Ed.*, 2013, **52**, 13588–13592.
- D. Shabashov and O. Daugulis, *J. Am. Chem. Soc.*, 2010, **132**, 3965–3972.

- 47 G. He, S.-Y. Zhang, W. A. Nack, R. Pearson, J. Rabb-Lynch and G. Chen, *Org. Lett.*, 2014, **16**, 6488–6491.
- 48 G. He, S.-Y. Zhang, W. A. Nack, Q. Li and G. Chen, *Angew. Chem., Int. Ed.*, 2013, **52**, 11124–11128.
- 49 C. P. Ting and T. J. Maimone, *Angew. Chem., Int. Ed.*, 2014, **53**, 3115–3119.
- 50 W. R. Gutekunst and P. S. Baran, *J. Org. Chem.*, 2014, **79**, 2430–2452.
- 51 W. R. Gutekunst, R. Gianatassio and P. S. Baran, *Angew. Chem., Int. Ed.*, 2012, **51**, 7507–7510.
- 52 R. A. Panish, S. R. Chintala and J. M. Fox, *Angew. Chem., Int. Ed.*, 2016, **55**, 4983–4987.
- 53 R. Panish, S. R. Chintala, D. T. Boruta, Y. Fang, M. T. Taylor and J. M. Fox, *J. Am. Chem. Soc.*, 2013, **135**, 9283–9286.
- 54 L. M. Chapman, J. C. Beck, L. Wu and S. E. Reisman, *J. Am. Chem. Soc.*, 2016, **138**, 9803–9806.
- 55 M. Zhou, X. R. Li, J. W. Tang, Y. Liu, X. N. Li, B. Wu, H. B. Qin, X. Du, L. M. Li, W. G. Wang, J. X. Pu and H. D. Sun, *Org. Lett.*, 2015, **17**, 6062–6065.
- 56 C. Tsukano, N. Muto, I. Enkhtaivan and Y. Takemoto, *Chem. – Asian J.*, 2014, **9**, 2628–2634.
- 57 D. S. C. Black, P. A. Keller and N. Kumar, *Tetrahedron Lett.*, 1989, **30**, 5807–5808.
- 58 Y. Miki, H. Shirokoshi and K. Matsushita, *Tetrahedron Lett.*, 1999, **40**, 4347–4348.
- 59 S. De, S. Ghosh, S. Bhunia, J. A. Sheikh and A. Bisai, *Org. Lett.*, 2012, **14**, 4466–4469.
- 60 G. Maestri, M.-H. Larraufie, E. Derat, C. Ollivier, L. Fensterbank, E. Lacôte and M. Malacria, *Org. Lett.*, 2010, **12**, 5692–5695.
- 61 C. Tsukano, M. Okuno and Y. Takemoto, *Angew. Chem., Int. Ed.*, 2012, **51**, 2763–2766.
- 62 R. A. Gossage, G. van Koten, W. D. Jones, F. Kakiuchi, S. Murai, A. Sen, M. Murakami, Y. Ito, M. Sugino, Y.-S. Lin, A. Yamamoto, V. V. Grushin, H. Alper, M. Hidai, Y. Mizobe and T. G. Richmond, *Activation of Unreactive Bonds and Organic Synthesis*, Springer Berlin Heidelberg, Berlin, Heidelberg, 1999, vol. 3.
- 63 H. Weinstabl, M. Suhartono, Z. Qureshi and M. Lautens, *Angew. Chem., Int. Ed.*, 2013, **52**, 5305–5308.
- 64 Z. Qureshi, H. Weinstabl, M. Suhartono, H. Liu, P. Thesmar and M. Lautens, *Eur. J. Org. Chem.*, 2014, 4053–4069.
- 65 T. J. Potter and J. A. Ellman, *Org. Lett.*, 2017, **19**, 2985–2988.
- 66 T. J. Potter, D. N. Kamber, B. Q. Mercado and J. A. Ellman, *ACS Catal.*, 2017, **7**, 150–153.
- 67 S. Akai, M. Kojima, S. Yamauchi, T. Kohji, Y. Nakamura and K. Sato, *Asian J. Org. Chem.*, 2013, **2**, 299–302.
- 68 B. R. Rosen, L. R. Simke, P. S. Thuy-Boun, D. D. Dixon, J. Q. Yu and P. S. Baran, *Angew. Chem., Int. Ed.*, 2013, **52**, 7317–7320.
- 69 D. D. Dixon, J. W. Lockner, Q. Zhou and P. S. Baran, *J. Am. Chem. Soc.*, 2012, **134**, 8432–8435.
- 70 A. Lerchen, T. Knecht, M. Koy, C. G. Daniliuc and F. Glorius, *Chem. – Eur. J.*, 2017, **23**, 12149–12152.
- 71 Y. Oh, Y. J. Jang, M. Jeon, H. S. Kim, J. H. Kwak, K. H. Chung, S. Pyo, Y. H. Jung and I. S. Kim, *J. Org. Chem.*, 2017, **82**, 11566–11572.
- 72 H. Abe, S. Takeda, T. Fujita, K. Nishioka, Y. Takeuchi and T. Harayama, *Tetrahedron Lett.*, 2004, **45**, 2327–2329.
- 73 M. Uemura, A. Daimon and Y. Hayashi, *J. Chem. Soc., Chem. Commun.*, 1995, **0**, 1943–1944.
- 74 B. Yalcouye, S. Choppin, A. Panossian, F. R. Leroux and F. Colobert, *Eur. J. Org. Chem.*, 2014, 6285–6294.
- 75 Q. Dherbassy, J. Wencel-Delord and F. Colobert, *Tetrahedron*, 2016, **72**, 5238–5245.
- 76 G. Liao, Q.-J. Yao, Z.-Z. Zhang, Y.-J. Wu, D.-Y. Huang and B.-F. Shi, *Angew. Chem., Int. Ed.*, 2018, **57**, 3661–3665.
- 77 C. K. Hazra, Q. Dherbassy, J. Wencel-Delord and F. Colobert, *Angew. Chem., Int. Ed.*, 2014, **53**, 13871–13875.
- 78 Q.-J. Yao, S. Zhang, B.-B. Zhan and B.-F. Shi, *Angew. Chem., Int. Ed.*, 2017, **56**, 6617–6621.
- 79 Z. Zhang, J. Wang, J. Li, F. Yang, G. Liu, W. Tang, W. He, J.-J. Fu, Y.-H. Shen, A. Li and W.-D. Zhang, *J. Am. Chem. Soc.*, 2017, **139**, 5558–5567.
- 80 J.-J. Li, T.-S. Mei and J.-Q. Yu, *Angew. Chem., Int. Ed.*, 2008, **47**, 6452–6455.
- 81 L. Chu, K.-J. Xiao and J.-Q. Yu, *Science*, 2014, **346**, 451–455.
- 82 P. Tao, Z. Chen and Y. Jia, *Chem. Commun.*, 2016, **52**, 11300–11303.
- 83 P. Tao and Y. Jia, *Chem. Commun.*, 2014, **50**, 7367–7370.
- 84 D. R. Stuart, M. Bertrand-Laperle, K. M. N. Burgess and K. Fagnou, *J. Am. Chem. Soc.*, 2008, **130**, 16474–16475.
- 85 Y. Zhang, J. W. Hubbard, N. G. Akhmedov, J. L. Petersen and B. C. G. Söderberg, *J. Org. Chem.*, 2015, **80**, 4783–4790.
- 86 J. C. Fox, R. E. Gilligan, A. K. Pitts, H. R. Bennett and M. J. Gaunt, *Chem. Sci.*, 2016, **7**, 2706–2710.
- 87 C.-L. Ciana, R. J. Phipps, J. R. Brandt, F.-M. Meyer and M. J. Gaunt, *Angew. Chem., Int. Ed.*, 2011, **50**, 458–462.
- 88 R. J. Phipps and M. J. Gaunt, *Science*, 2009, **323**, 1593–1597.
- 89 S. H. Cho, J. Yoon and S. Chang, *J. Am. Chem. Soc.*, 2011, **133**, 5996–6005.
- 90 K. Orito, A. Horibata, T. Nakamura, H. Ushito, H. Nagasaki, M. Yuguchi, S. Yamashita and M. Tokuda, *J. Am. Chem. Soc.*, 2004, **126**, 14342–14343.
- 91 J. I. G. Cadogan, M. Cameron-Wood, R. K. Mackie and R. J. G. Searle, *J. Chem. Soc.*, 1965, **2078**, 4831.
- 92 D. S. Peters, F. E. Romesberg and P. S. Baran, *J. Am. Chem. Soc.*, 2018, **140**, 2072–2075.
- 93 T. C. Roberts, P. A. Smith, R. T. Cirz and F. E. Romesberg, *J. Am. Chem. Soc.*, 2007, **129**, 15830–15838.
- 94 A. D. Yamaguchi, K. M. Chepiga, J. Yamaguchi, K. Itami and H. M. L. Davies, *J. Am. Chem. Soc.*, 2015, **137**, 644–647.
- 95 A. K. Pitts, F. O'Hara, R. H. Snell and M. J. Gaunt, *Angew. Chem., Int. Ed.*, 2015, **54**, 5451–5455.
- 96 K. Ueda, K. Amaike, R. M. Maceiczyk, K. Itami and J. Yamaguchi, *J. Am. Chem. Soc.*, 2014, **136**, 13226–13232.
- 97 R. J. Phipps, N. P. Grimster and M. J. Gaunt, *J. Am. Chem. Soc.*, 2008, **130**, 8172–8174.
- 98 S. Paul, G. A. Chotana, D. Holmes, R. C. Reichle, R. E. Maleczka and M. R. Smith, *J. Am. Chem. Soc.*, 2006, **128**, 15552–15553.

- 99 K. Okano, H. Fujiwara, T. Noji, T. Fukuyama and H. Tokuyama, *Angew. Chem., Int. Ed.*, 2010, **49**, 5925–5929.
- 100 M. Yoshida, K. Saito, Y. Fujino and T. Doi, *Tetrahedron*, 2014, **70**, 3452–3458.
- 101 K. Lei, D.-W. Sun, Y.-Y. Tao and X.-H. Xu, *Aust. J. Chem.*, 2015, **69**, 98–106.
- 102 Y. Shin, C. Yoo, Y. Moon, Y. Lee and S. Hong, *Chem. – Asian J.*, 2015, **10**, 878–881.
- 103 K. L. White and M. Movassaghi, *J. Am. Chem. Soc.*, 2016, **138**, 11383–11389.
- 104 Y. Feng, D. Holte, J. Zoller, S. Umemiya, L. R. Simke and P. S. Baran, *J. Am. Chem. Soc.*, 2015, **137**, 10160–10163.
- 105 H. S. Kim, M. G. Banwell and A. C. Willis, *J. Org. Chem.*, 2013, **78**, 5103–5109.
- 106 D. W. Robbins, T. A. Boebel and J. F. Hartwig, *J. Am. Chem. Soc.*, 2010, **132**, 4068–4069.
- 107 B. A. Granger, I. T. Jewett, J. D. Butler, B. Hua, C. E. Knezevic, E. I. Parkinson, P. J. Hergenrother and S. F. Martin, *J. Am. Chem. Soc.*, 2013, **135**, 12984–12986.
- 108 L. Cai, K. Zhang and O. Kwon, *J. Am. Chem. Soc.*, 2016, **138**, 3298–3301.
- 109 L. Furst, B. S. Matsuura, J. M. R. Narayanan, J. W. Tucker and C. R. J. Stephenson, *Org. Lett.*, 2010, **12**, 3104–3107.
- 110 F. Xue, H. Lu, L. He, W. Li, D. Zhang, X. Y. Liu and Y. Qin, *J. Org. Chem.*, 2018, **83**, 754–764.
- 111 J. Du Bois, *Org. Process Res. Dev.*, 2011, **15**, 758–762.
- 112 J. L. Jeffrey and R. Sarpong, *Chem. Sci.*, 2013, **4**, 4092.
- 113 F. Collet, C. Lescot and P. Dauban, *Chem. Soc. Rev.*, 2011, **40**, 1926–1936.
- 114 T. A. Bedell, G. A. B. Hone, D. Valette, J.-Q. Yu, H. M. L. Davies and E. J. Sorensen, *Angew. Chem., Int. Ed.*, 2016, **55**, 8270–8274.
- 115 C. Liu, R. Chen, Y. Shen, Z. Liang, Y. Hua and Y. Zhang, *Angew. Chem., Int. Ed.*, 2017, **56**, 8187–8190.
- 116 R. Meier and D. Trauner, *Angew. Chem., Int. Ed.*, 2016, **55**, 11251–11255.
- 117 A. J. Burckle, V. H. Vasilev and N. Z. Burns, *Angew. Chem., Int. Ed.*, 2016, **55**, 11476–11479.
- 118 M. Shen, M. Kretschmer, Z. G. Brill and S. A. Snyder, *Org. Lett.*, 2016, **18**, 5018–5021.
- 119 P. Lu, A. Mailyan, Z. Gu, D. M. Guptill, H. Wang, H. M. L. Davies and A. Zakarian, *J. Am. Chem. Soc.*, 2014, **136**, 17738–17749.
- 120 P. Lu, Z. Gu and A. Zakarian, *J. Am. Chem. Soc.*, 2013, **135**, 14552–14555.
- 121 J. Gong, G. Lin, C. Li and Z. Yang, *Org. Lett.*, 2009, **11**, 4770–4773.
- 122 F. Peng, M. Yu and S. J. Danishefsky, *Tetrahedron Lett.*, 2009, **50**, 6586–6587.
- 123 K. C. Nicolaou, L. Dong, L. Deng, A. C. Talbot and D. Y.-K. Chen, *Chem. Commun.*, 2010, **46**, 70–72.
- 124 V. Singh, P. Bhalerao and S. M. Mobin, *Tetrahedron Lett.*, 2010, **51**, 3337–3339.
- 125 I. Baitinger, P. Mayer and D. Trauner, *Org. Lett.*, 2010, **12**, 5656–5659.
- 126 F. Peng and S. J. Danishefsky, *Tetrahedron Lett.*, 2011, **52**, 2104–2106.
- 127 K. E. Lazarski, B. Akpinar and R. J. Thomson, *Tetrahedron Lett.*, 2013, **54**, 635–637.
- 128 P. Carberry, D. R. Viernes, L. B. Choi, M. W. Fegley and J. D. Chisholm, *Tetrahedron Lett.*, 2013, **54**, 1734–1737.
- 129 Y. Zhang, J. Gong and Z. Yang, *Chem. Rec.*, 2014, **14**, 606–622.
- 130 B. Hong, C. Li, Z. Wang, J. Chen, H. Li and X. Lei, *J. Am. Chem. Soc.*, 2015, **137**, 11946–11949.
- 131 P. S. Thuy-Boun, G. Villa, D. Dang, P. Richardson, S. Su and J. Q. Yu, *J. Am. Chem. Soc.*, 2013, **135**, 17508–17513.
- 132 D. H. Wang and J. Q. Yu, *J. Am. Chem. Soc.*, 2011, **133**, 5767–5769.
- 133 H. M. L. Davies and J. R. Manning, *Nature*, 2008, **451**, 417–424.
- 134 H. M. L. Davies, M. V. A. Grazini and E. Aouad, *Org. Lett.*, 2001, **3**, 1475–1477.
- 135 H. Saito, H. Oishi, S. Kitagaki, S. Nakamura, M. Anada and S. Hashimoto, *Org. Lett.*, 2002, **4**, 3887–3890.
- 136 C. Soldi, K. N. Lamb, R. A. Squitieri, M. González-López, M. J. Di Maso and J. T. Shaw, *J. Am. Chem. Soc.*, 2014, **136**, 15142–15145.
- 137 J. Egger, S. Fischer, P. Bretscher, S. Freigang, M. Kopf and E. M. Carreira, *Org. Lett.*, 2015, **17**, 4340–4343.
- 138 C. Zhu, X. Shen and S. G. Nelson, *J. Am. Chem. Soc.*, 2004, **126**, 5352–5353.
- 139 J. Egger, P. Bretscher, S. Freigang, M. Kopf and E. M. Carreira, *Angew. Chem., Int. Ed.*, 2013, **52**, 5382–5385.
- 140 S. P. Lathrop and M. Movassaghi, *Chem. Sci.*, 2014, **5**, 333–340.
- 141 S. P. Lathrop, M. Pompeo, W.-T. T. Chang and M. Movassaghi, *J. Am. Chem. Soc.*, 2016, **138**, 7763–7769.
- 142 P. Lindovska and M. Movassaghi, *J. Am. Chem. Soc.*, 2017, **139**, 17590–17596.
- 143 M. Movassaghi, O. K. Ahmad and S. P. Lathrop, *J. Am. Chem. Soc.*, 2011, **133**, 13002–13005.
- 144 J. L. Roizen, D. N. Zalatan and J. Du Bois, *Angew. Chem., Int. Ed.*, 2013, **52**, 11343–11346.
- 145 L. Xu, L. Tan and D. Ma, *J. Org. Chem.*, 2016, **81**, 10476–10483.
- 146 H. Nambu, N. Noda, W. Niu, T. Fujiwara and T. Yakura, *Asian J. Org. Chem.*, 2015, **4**, 1246–1249.
- 147 J. J. Fleming and J. Du Bois, *J. Am. Chem. Soc.*, 2006, **128**, 3926–3927.
- 148 A. Takada, H. Fujiwara, K. Sugimoto, H. Ueda and H. Tokuyama, *Chem. – Eur. J.*, 2015, **21**, 16400–16403.
- 149 S. Kawamura, H. Chu, J. Felding and P. S. Baran, *Nature*, 2016, **532**, 90–93.
- 150 L. Jorgensen, S. J. McKerrall, C. A. Kuttruff, F. Ungeheuer, J. Felding and P. S. Baran, *Science*, 2013, **341**, 878–882.
- 151 M. W. Smith and S. A. Snyder, *J. Am. Chem. Soc.*, 2013, **135**, 12964–12967.
- 152 E. C. Cherney, J. M. Lopchuk, J. C. Green and P. S. Baran, *J. Am. Chem. Soc.*, 2014, **136**, 12592–12595.
- 153 H. Renata, Q. Zhou and P. S. Baran, *Science*, 2013, **339**, 59–63.

- 154 S. A. Loskot, D. K. Romney, F. H. Arnold and B. M. Stoltz, *J. Am. Chem. Soc.*, 2017, **139**, 10196–10199.
- 155 S. Kille, F. E. Zilly, J. P. Acevedo and M. T. Reetz, *Nat. Chem.*, 2011, **3**, 738–743.
- 156 A. Rentmeister, F. H. Arnold and R. Fasan, *Nat. Chem. Biol.*, 2009, **5**, 26–28.
- 157 J. C. Lewis, S. Bastian, C. S. Bennett, Y. Fu, Y. Mitsuda, M. M. Chen, W. A. Greenberg, C.-H. Wong and F. H. Arnold, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 16550–16555.
- 158 C. R. Zwick and H. Renata, *J. Am. Chem. Soc.*, 2018, **140**, 1165–1169.
- 159 M. B. Nodwell, H. Yang, M. Čolović, Z. Yuan, H. Merkens, R. E. Martin, F. Bénard, P. Schaffer and R. Britton, *J. Am. Chem. Soc.*, 2017, **139**, 3595–3598.
- 160 X. Zhang, E. King-Smith and H. Renata, *Angew. Chem., Int. Ed.*, 2018, **57**, 5037–5041.
- 161 G. P. Miley, J. C. Rote, R. B. Silverman, N. L. Kelleher and R. J. Thomson, *Org. Lett.*, 2018, **20**, 2369–2373.