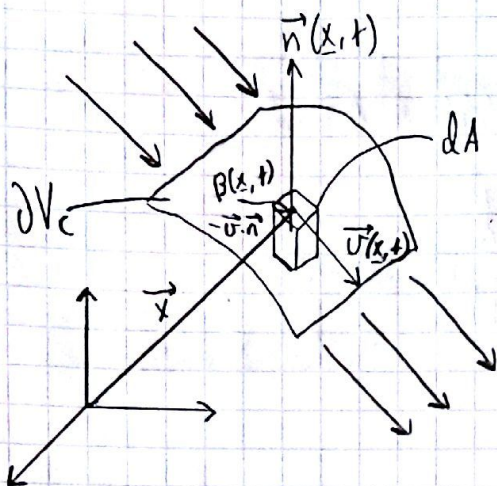


21.04.16



Instromning av β per tidenshet

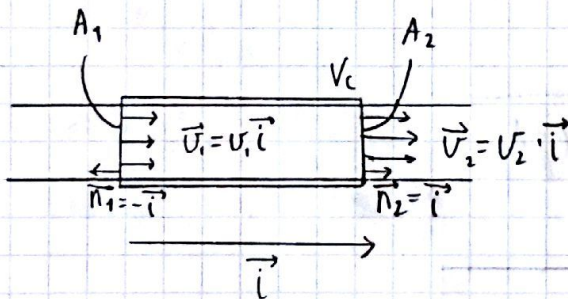
$$-\rho(x, t)\beta(x, t)\vec{u}(x, t) \cdot \vec{n}(x, t) dA$$

Gjennom hele ∂V_c

$$-\iint \rho(x, t)\beta(x, t)\vec{u}(x, t) \cdot \vec{n}(x, t) dA$$

$$= \iint_{\partial V_c} \rho\beta\vec{u} \cdot \vec{n} dA$$

Eks. Massstrom i rør



$$-\iint_{\partial V_c} \rho\vec{u} \cdot \vec{n} dA = -\iint_{A_1} \rho_1 u_1 \vec{n}_1 dA - \iint_{A_2} \rho_2 u_2 \vec{n}_2 dA$$

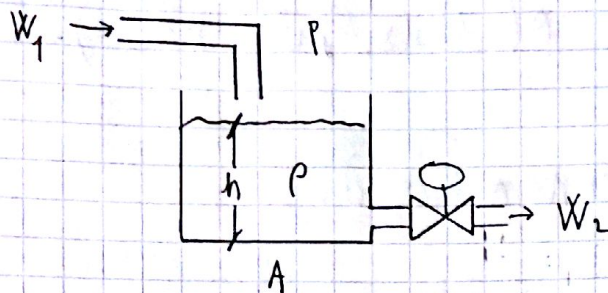
$$= -\rho_1(-u_1)A_1 - \rho_2 u_2 A_2 = \rho_1 u_1 A_1 - \rho_2 u_2 A_2 \quad | \quad q_i = u_i A$$

$$= \rho_1 q_1 - \rho_2 q_2 \quad | \quad W_i = \rho_i \cdot q_i$$

$$= W_1 - W_2$$

$$\boxed{\frac{d}{dt} m = W_1 - W_2}$$

Exs. Tank



mass balance

$$\frac{d}{dt} m = w_1 - w_2$$

kap. 4

$$q_2 = C_d A_v \sqrt{\frac{2}{\rho} \Delta P}$$

$$\Delta P = p + \rho g h - p = \rho g h$$

$$m = \rho g h$$

$$\frac{d}{dt} (\rho g h) = w_1 - \rho C_d A_v \sqrt{\frac{2}{\rho} g h}$$

$$\boxed{\frac{d}{dt} h = \frac{1}{\rho A} w_1 - \frac{1}{A} C_d A_v \sqrt{2 g h}}$$

Flerkomponent-system

$k = 1, \dots, n$ komponenter

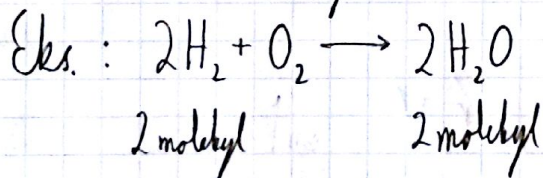
Tetthet: ρ_k

dannelserate: r_k $[r_k] = \frac{kg}{m^3 s}$

Balanslov, komponent k :

$$\frac{d}{dt} m_k = \frac{d}{dt} \iiint_{V_c} \rho_k dV = - \iint_{\partial V_c} \rho_k \vec{v} \cdot \vec{n} dA + \iiint_{V_c} r_k dV$$

Mer vanlig: kjemiske reaksjoner
1 molekyl



mer praktisk med molekyl-balanser
eller enn mass-balanser.

molekyl måles i mol, symbol n_k

$$n_k = \frac{m_k}{M_k}, M_k: \text{molekylvekt} \left[\frac{kg}{mol} \right]$$

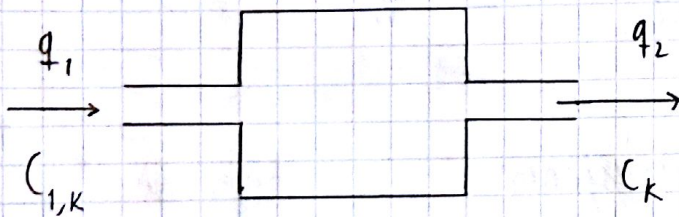
Definer $c_k = \frac{n_k}{V} = \frac{\frac{m_k}{M_k}}{V} = \frac{\frac{\rho_k V}{M_k}}{V} = \frac{\rho_k}{M_k}$

$$\hat{r}_k = \frac{r_k}{M_k}$$

"Mol balance":

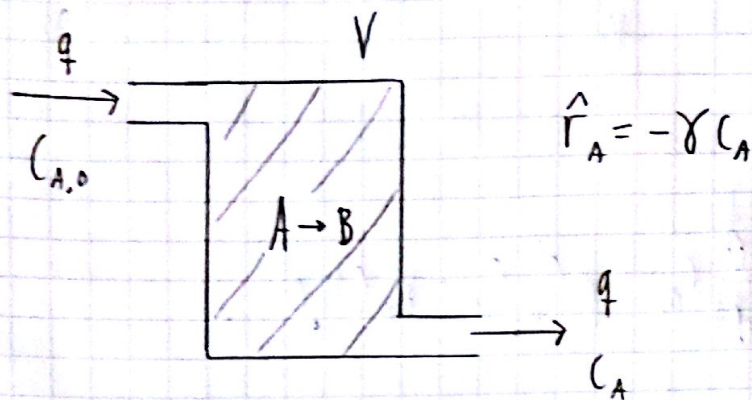
$$\frac{d}{dt} \iiint_{\partial V_c} c_k \vec{U}_k \cdot \vec{n} dA + \iiint_{V_c} \hat{r}_k dV$$

Ex.



$$\frac{d}{dt} n_k = \frac{d}{dt} (c_k V) = q_1 c_{1,k} - q_2 c_k - \hat{r}_k V$$

Exs.



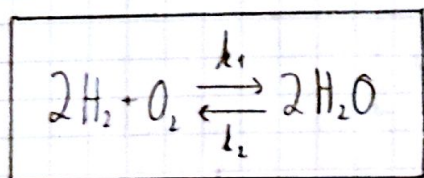
$$\frac{d}{dt}(C_A V) = q C_{A,0} - q C_A - \gamma C_A V$$

$$\boxed{\frac{d}{dt} C_A = -\left(\frac{q}{V} + \gamma\right) C_A + \frac{q}{V} C_{A,0}}$$

Generell:

$$\hat{r}_n = f(C_1, C_2, \dots, C_n, T)$$

Exs.

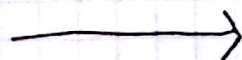


$$k_1 = \gamma_1 C_{\text{H}_2}^2 C_{\text{O}_2}$$

$$k_2 = \gamma_2 C_{\text{H}_2\text{O}}^2$$

$$\hat{r}_{\text{O}_2} = -\gamma_1 C_{\text{H}_2}^2 C_{\text{O}_2} + \gamma_2 C_{\text{H}_2\text{O}}^2$$

$$\hat{r}_{\text{H}_2} = 2 \hat{r}_{\text{O}_2}, \quad \hat{r}_{\text{H}_2\text{O}} = -2 \hat{r}_{\text{O}_2}$$

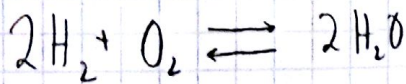


$$\frac{d}{dt} \iiint_{V_k} c_k dV = - \iint_{\partial V_k} c_k \vec{v} \cdot \vec{n} dA + \iiint_{V_k} \hat{r}_k dV$$

$$\frac{d}{dt} c_k = \hat{r}_k$$

$$\frac{d}{dt} c_{O_2} = -\gamma_{H_2} c_{O_2} + \gamma_{H_2O}$$

Stoichiometrische Koeffizienten



② ① ②

$$\Gamma = \begin{bmatrix} -2 \\ -1 \\ 2 \end{bmatrix}$$

$$\underline{c} = \begin{bmatrix} c_{H_2} \\ c_{O_2} \\ c_{H_2O} \end{bmatrix}$$

$$\frac{d}{dt} \underline{c} = \Gamma \hat{r}(\underline{c})$$

→

$$\hat{r}(\underline{c}) = \hat{r}_{O_2}$$