Table 1: JVAL Photolysis reactions (version from March 29, 2021)

#	reaction	reference for spectrum
J1000	$O_2 \xrightarrow{h\nu} O + O$	Sander et al. (2011), Lyman-alpha from Chabrillat and Kockarts (1997) and Chabrillat and Kockarts (1998), Schumann-Runge band parameterization from Koppers and Murtagh (1996)
J1001b	$O_3 \stackrel{h\nu}{\rightarrow} O(^3P) + O_2$	Sander et al. (2011)
J1001a	$O_3 \stackrel{h\nu}{\rightarrow} O(^1D) + O_2$	Sander et al. (2011)
J2101	$H_2O_2 \stackrel{h\nu}{\rightarrow} OH + OH$	Sander et al. (2011)
J3101	$NO_2 \stackrel{h\nu}{\rightarrow} NO + O$	Sander et al. (2011)
J3103a	$NO_3 \stackrel{h\nu}{\rightarrow} NO_2 + O$	Sander et al. (2011)
J3103b	$NO_3 \stackrel{h\nu}{\rightarrow} NO + O_2$	Sander et al. (2011)
J3104	$N_2O_5 \stackrel{h\nu}{\rightarrow} NO_2 + NO_3$	Sander et al. (2011)
J3201	$HNO_3 \xrightarrow{h\nu} products$	Sander et al. (2011)
J3202	$HNO_4 \xrightarrow{h\nu} products$	Sander et al. (2011), IR overtones from Roehl et al. (2002)
J42004	$PAN \xrightarrow{h\nu} products$	Sander et al. (2011)
J3200	$HONO \xrightarrow{h\nu} products$	Sander et al. (2011)
J4100	$\text{CH}_3\text{OOH} \xrightarrow{\text{h}\nu} \text{products}$	Sander et al. (2011) up to 405 nm, Matthews et al. (2005) above 600 nm, zero in between
J41001a	$\text{HCHO} \stackrel{\text{h}\nu}{\rightarrow} \text{CO} + \text{H}_2$	Sander et al. (2011), quantum yields at $300~\mathrm{K}$ and $1~\mathrm{atm}$
J41001b	$\text{HCHO} \stackrel{\text{h}\nu}{\rightarrow} \text{CHO} + \text{H}$	Sander et al. (2011), quantum yields at 300 K and 1 atm
J42002	$\text{CH}_3\text{CO}_3\text{H} \stackrel{\text{h}\nu}{\to} \text{ products}$	Sander et al. (2011)
J42001a	$\text{CH}_3\text{CHO} \xrightarrow{\text{h}\nu} \text{CH}_3 + \text{CHO}$	Sander et al. (2011)
J43001	$\text{CH}_3\text{COCH}_3 \stackrel{\text{h}\nu}{\rightarrow} \text{ products}$	Hardcoded from old JVAL code. Pressure dependent.
J43003	$MGLYOX \xrightarrow{h\nu} products$	Hardcoded from old JVAL code. Pressure dependent.
J6201	$HOCl \xrightarrow{h\nu} OH + Cl$	Sander et al. (2011)
J6101	$OClO \xrightarrow{h\nu} products$	Sander et al. (2011), value at 204 K
J6100	$\text{Cl}_2\text{O}_2 \stackrel{\text{h}\nu}{\to} \text{Cl} + \text{ClO}_2$	Sander et al. (2011)
J6301a	$ClNO_3 \stackrel{h\nu}{\rightarrow} Cl + NO_3$	Sander et al. (2011)
J6300	$ClNO_2 \stackrel{h\nu}{\rightarrow} products$	Ghosh et al. (2012)
J6000	$Cl_2 \stackrel{h\nu}{\rightarrow} 2Cl$	Sander et al. (2011)
J7100	$BrO \xrightarrow{h\nu} Br + O$	Sander et al. (2011)
J7200	$HOBr \xrightarrow{h\nu} OH + Br$	Sander et al. (2011)
J7600	$BrCl \xrightarrow{h\nu} Br + Cl$	Sander et al. (2011), based on formula by Maric et al. (1994)
J7301	$BrNO_3 \xrightarrow{h\nu} products$	Sander et al. (2011)
J7300	$BrNO_2 \xrightarrow{h\nu} products$	Sander et al. (2011)
J7000	$\operatorname{Br}_2 \stackrel{\mathrm{h}_{\nu}}{\to} \operatorname{products}$	Sander et al. (2011)
J6401	$CCl_4 \xrightarrow{h\nu} products$	Sander et al. (2011)
J6400	$CH_3Cl \xrightarrow{h\nu} products$	Sander et al. (2011)
J6402	$\text{CH}_3\text{CCl}_3 \stackrel{\text{h}\nu}{\to} \text{ products}$	Sander et al. (2011)
J6500	$CFCl_3 \xrightarrow{h\nu} products$	Sander et al. (2011), formula for temperature-dependence from DeMore et al. (1997)
J6501	$CF_2Cl_2 \xrightarrow{h\nu} \text{ products}$	Sander et al. (2011), formula for temperature-dependence from DeMore et al. (1997)
J7400	$CH_3Br \xrightarrow{h\nu} products$	Sander et al. (2011)

Table 1: Photolysis reactions (... continued)

#	reaction	reference for spectrum
	$CF_2ClBr \xrightarrow{h\nu} products$	Sander et al. (2011)
J7500	$CF_3Br \xrightarrow{h\nu} products$	Sander et al. (2011)
J8401	$CH_3I \xrightarrow{h\nu} \text{ products}$	Sander et al. (2011), using data at 298 K, temperature dependence not considered
J8402	$C_3H_7I \xrightarrow{h\nu} \text{ products}$	Sander et al. (2011)
J8403	$\text{CH}_2\text{ClI} \xrightarrow{\text{h}\nu} \text{products}$	Sander et al. (2011), using data for 298 K, temperature dependence not considered
J8400	$\mathrm{CH_2I_2} \overset{\mathrm{h}\nu}{ o} \mathrm{\ products}$	Sander et al. (2011), using data for 298 K, temperature dependence not considered
J8100	$IO \xrightarrow{h\nu} I + O$	Sander et al. (2011)
J8200	$HOI \xrightarrow{h\nu} products$	Sander et al. (2011)
J8000	$I_2 \stackrel{h\nu}{\rightarrow} 2I$	Keller-Rudek et al. (2013), based on Sander et al. (2006)
J8600	$ICl \xrightarrow{h\nu} products$	Sander et al. (2011), values shown as " $<$ 1" in their Table 4H-10 were set to 0
J8700	$\operatorname{IBr} \stackrel{h\nu}{\to} \operatorname{products}$	Sander et al. (2011)
J8300	$INO_2 \xrightarrow{h\nu} products$	Sander et al. (2011)
J8301	$INO_3 \stackrel{h\nu}{\rightarrow} products$	Sander et al. (2011)
	$\mathrm{SO}_2 \overset{\mathrm{h} u}{ o} \ \mathrm{SO}_2^*$	Danielache et al. (2008), quantum yield for dissociation is unknown.
	$SO_3 \xrightarrow{h\nu} products$	Sander et al. (2011)
J9000	$OCS \xrightarrow{h\nu} products$	Sander et al. (2011)
J2100	$H_2O \xrightarrow{h\nu} H + OH$	Sander et al. (2011)
J3100	$N_2O \xrightarrow{h\nu} N_2 + O(^1D)$	Sander et al. (2011)
J3102	$NO \xrightarrow{h\nu} N + O$	Hardcoded from old JVAL code.
J41002	$CO_2 \stackrel{h\nu}{\rightarrow} CO + O$	Shemansky (1972), Lyman-alpha from Inn et al. (1953)
J6200	$HCl \xrightarrow{h\nu} H + Cl$	Sander et al. (2011)
J7603	$\mathrm{CHCl_2Br} \stackrel{\mathrm{h}\nu}{\to} \mathrm{products}$	Sander et al. (2011)
J7604	$\text{CHClBr}_2 \stackrel{\text{h}\nu}{\rightarrow} \text{ products}$	Sander et al. (2011)
J7602	$CH_2ClBr \xrightarrow{h\nu} products$	Sander et al. (2011)
J7401	$\mathrm{CH_2Br_2} \stackrel{\mathrm{h}\nu}{\to} \mathrm{products}$	Sander et al. (2011)
J7402	$CHBr_3 \xrightarrow{h\nu} products$	Sander et al. (2011), formula for temperature-dependence not only used for 290-340 nm but also for > 340 nm.
J9002	$SF_6 \xrightarrow{h\nu} products$	Lyman-alpha from Ravishankara et al. (1993)
J6301b	$CINO_3 \xrightarrow{h\nu} CIO + NO_2$	Sander et al. (2011)
J44008	$MACR \xrightarrow{h\nu} products$	Hardcoded from old JVAL code. Pressure dependent.
J44001	$MVK \xrightarrow{h\nu} products$	Hardcoded from old JVAL code. Pressure dependent.
J42008	$\text{CHOCHO} \stackrel{\text{h}\nu}{\rightarrow} \text{ 2CHO}$	Hardcoded from old JVAL code. Pressure dependent.
J42005	$HOCH_2CHO \xrightarrow{h\nu} products$	Sander et al. (2011)
J41003	$\mathrm{CH_4} \stackrel{\mathrm{h}\nu}{ o} \mathrm{products}$	Lyman-alpha from Fig. 1 of Turco (1975)
	$H_2SO_4 \stackrel{h\nu}{\rightarrow} SO_3 + H_2O$	Hardcoded from old JVAL code.
	$C_3O_2 \xrightarrow{h\nu} \text{ products}$	Stephan Kessel, pers. comm.
J41005	$\text{CH}_3\text{NO}_3 \stackrel{\text{h}\nu}{\to} \text{ products}$	Sander et al. (2011)
J41006	$CH_3O_2NO_2 \xrightarrow{h\nu} products$	Atkinson et al. (2006)
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Table 1: Photolysis reactions (... continued)

#	reaction	reference for spectrum
J41004	$\text{CH}_3\text{ONO} \xrightarrow{\text{h}\nu} \text{products}$	Sander et al. (2011), using $\varphi = 0.76$ for all wavelengths
J41008	$CH_3O_2 \xrightarrow{h\nu} products$	Sander et al. (2011)
J41009	$\text{HCOOH} \xrightarrow{\text{h}\nu} \text{ products}$	Sander et al. (2011)
$\rm J6500dc01$	$\text{CHF}_2\text{Cl} \stackrel{\text{h}\nu}{\to} \text{ products}$	Sander et al. (2011)
J42019	$C_2H_5NO_3 \xrightarrow{h\nu} \text{ products}$	Atkinson et al. (2006)
J43007	$NOA \xrightarrow{h\nu} products$	Barnes et al. (1993)
J44025	$3-\text{nitrooxy}-2-\text{butanone} \xrightarrow{\text{h}\nu} \text{ products}$	Barnes et al. (1993)
J47403	BENZAL $\stackrel{\text{h}\nu}{\rightarrow}$ HCO + C ₆ H ₅	Wallington et al. (2018)
	$3-\text{Me}-2-\text{nitrophenol} \xrightarrow{\text{h}\nu} \text{HONO} + \text{products}$	Chen et al. (2011)
J46405	$2-\text{nitrophenol} \xrightarrow{h\nu} \text{HONO} + \text{products}$	Chen et al. (2011)
J42001b	$\text{CH}_3\text{CHO} \xrightarrow{\text{h}\nu} \text{CH}_2 = \text{CHOH}$	Andrews et al. (2012)
J43018	$\text{CH}_3\text{COCO}_2\text{H} \stackrel{\text{h}\nu}{\to} \text{ products}$	Sander et al. (2011)
J44038	$(\mathrm{CH_3})_2\mathrm{CHCHO} \stackrel{\mathrm{h}\nu}{\to} (\mathrm{CH_3})_2\mathrm{CH} + \mathrm{CHO}$	Allan et al. (2007)
J43025a	$\text{CH}_3\text{CH}_2\text{CHO} \xrightarrow{\text{h}\nu} \text{CH}_3\text{CH}_2 + \text{CHO}$	Allan et al. (2007)
J43025b	$\text{CH}_3\text{CH}_2\text{CHO} \stackrel{\text{h}\nu}{\to} \text{CH}_3\text{CHCHOH}$	Zhou et al. (2008)
J44037a	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CHO} \xrightarrow{\text{h}\nu} \text{CH}_3\text{CH}_2\text{CH}_2 + \text{CHO}$	Allan et al. (2007)
J44037b	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CHO} \xrightarrow{\text{h}\nu} \text{CH}_2\text{CH}_2 + \text{CH}_2\text{CHOH}$	Zhou et al. (2008)
	$2,4$ -pentanedione $\stackrel{\text{h}\nu}{\rightarrow}$ products	Messaadia et al. (2015)
J40203a	$PINAL \xrightarrow{h\nu} C96O2 + CHO$	Allan et al. (2007)
J40203b	$PINAL \xrightarrow{h\nu} PINENOL$	Andrews et al. (2012)
m J6500dc02	$CF_2ClCFCl_2 \xrightarrow{h\nu} products$	Sander et al. (2011)
m J6500dc03	$\text{CH}_3\text{CFCl}_2 \stackrel{\text{h}\nu}{\to} \text{products}$	Sander et al. (2011)
m J6500dc05	$CF_3CF_2Cl \xrightarrow{h\nu} products$	Sander et al. (2011)
J6500dc04	$CF_2ClCF_2Cl \xrightarrow{h\nu} products$	Sander et al. (2011)
m J6400dc02	$\text{CHCl}_3 \stackrel{\text{h}\nu}{\to} \text{ products}$	Sander et al. (2011)
m J6400dc01	$\mathrm{CH_2Cl_2} \overset{\mathrm{h}\nu}{\to} \mathrm{products}$	Sander et al. (2011)
	$\mathrm{HO_2} \stackrel{\mathrm{h}\nu}{\to} \mathrm{OH} + \mathrm{O3P}$	Sander et al. (2011)
	$ClO \xrightarrow{h\nu} O3P + Cl$	Sander et al. (2011)
J42022	$ \text{HOOCCOOH} \xrightarrow{\text{h}\nu} \text{CO}_2 + 0.72 \text{HCOOH} + 0.28 \text{CO} + 0.28 \text{H}_2 \text{O} $	
J????	$\text{Cl}_2\text{O} \stackrel{\text{h}\nu}{\to} \text{Cl} + \text{ClO}$	Atkinson et al. (2007)
J????	$\text{Cl}_2\text{O}_3 \stackrel{\text{h}\nu}{\to} \text{ products}$	Atkinson et al. (2007)
J????	$ClNO \xrightarrow{h\nu} Cl + NO$	Atkinson et al. (2007)
J????	$ClONO \xrightarrow{h\nu} Cl + NO_2$	Atkinson et al. (2007)

References

- Allan, W., Struthers, H., and Lowe, D. C.: Methane carbon isotope effects caused by atomic chlorine in the marine boundary layer: Global model results compared with Southern Hemisphere measurements, J. Geophys. Res., 112D, D04306, doi:10.1029/2006JD007369, 2007.
- Andrews, D. U., Heazlewood, B. R., Maccarone, A. T., Conroy, T., Payne, R. J., Jordan, M. J. T., and Kable, S. H.: Photo-tautomerization of acetaldehyde to vinyl alcohol: a potential route to tropospheric acids, Science, 337, 1203–1206, doi:10.1126/science.1220712, 2012.
- Atkinson, R., Baulch, D. L., Cox, R. A., Crowley, J. N., Hampson, R. F., Hynes, R. G., Jenkin, M. E., Rossi, M. J., Troe, J., and IUPAC Subcommittee: Evaluated kinetic and photochemical data for atmospheric chemistry: Volume II gas phase reactions of organic species, Atmos. Chem. Phys., 6, 3625–4055, doi:10.5194/ACP-6-3625-2006, 2006.
- Atkinson, R., Baulch, D. L., Cox, R. A., Crowley, J. N., Hampson, R. F., Hynes, R. G., Jenkin, M. E., Rossi, M. J., and Troe, J.: Evaluated kinetic and photochemical data for atmospheric chemistry: Volume III gas phase reactions of inorganic halogens, Atmos. Chem. Phys., 7, 981–1191, doi:10.5194/ACP-7-981-2007, 2007.
- Barnes, I., Becker, K. H., and Zhu, T.: Near UV absorption-spectra and photolysis products of difunctional organic nitrates possible importance as NO_x reservoirs, J. Atmos. Chem., 17, 353–373, doi:10.1007/BF00696854, 1993.
- Chabrillat, S. and Kockarts, G.: Simple parameterization of the absorption of the solar Lyman-alpha line, Geophys. Res. Lett., 24, 2659–2662, doi:10.1029/97GL52690, 1997.
- Chabrillat, S. and Kockarts, G.: Correction to "Simple parameterization of the absorption of the solar Lyman-alpha line", Geophys. Res. Lett., 25, 79, doi:10.1029/97GL03569, 1998.
- Chen, J., Wenger, J. C., and Venables, D. S.: Near-ultraviolet absorption cross sections of nitrophenols and their potential influence on tropospheric oxidation capacity, J. Phys. Chem. A, 115, 12235–12242, doi:10.1021/jp206929r, 2011.
- Danielache, S. O., Eskebjerg, C., Johnson, M. S., Ueno, Y., and Yoshida, N.: High-precision spectroscopy of ³²S, ³³S, and ³⁴S sulfur dioxide: Ultraviolet absorption cross sections and isotope effects, J. Geophys. Res., 113D, D17314, doi:10.1029/2007JD009695, 2008.
- DeMore, W. B., Sander, S. P., Golden, D. M., Hampson, R. F., Kurylo, M. J., Howard, C. J., Ravishankara, A. R., Kolb, C. E., and Molina, M. J.: Chemical kinetics and photochemical data for use in stratospheric modeling. Evaluation number 12, JPL Publication 97-4, Jet Propulsion Laboratory, Pasadena, CA, 1997.
- Ghosh, B., Papanastasiou, D. K., Talukdar, R. K., Roberts, J. M., and Burkholder, J. B.: Nitryl chloride (ClNO₂): UV/Vis absorption spectrum between 210 and 296 K and O(³P) quantum yield at 193 and 248 nm, J. Phys. Chem. A, 116, 5796–5805, doi:10.1021/JP207389Y, 2012.
- Inn, E. C. Y., Watanabe, K., and Zelikoff, M.: Absorption coefficients of gases in the vacuum ultraviolet. Part III. CO₂, J. Chem. Phys., 21, 1648–1650, doi:10.1063/1.1698637, 1953.
- Keller-Rudek, H., Moortgat, G. K., Sander, R., and Sörensen, R.: The MPI-Mainz UV/VIS spectral atlas of gaseous molecules of atmospheric interest, Earth Syst. Sci. Data, 5, 365–373, doi:10.5194/ESSD-5-365-2013, 2013.
- Koppers, G. A. A. and Murtagh, D. P.: Model studies of the influence of O₂ photodissociation parameterizations in the Schumann-Runge bands on ozone related photolysis in the upper atmosphere, Ann. Geophys., 14, 68–79, 1996.
- Maric, D., Burrows, J. P., and Moortgat, G. K.: A study of the UV-visible absorption spectra of Br_2 and BrCl, J. Photochem. Photobiol. A: Chem., 83, 179–192, doi:10.1016/1010-6030(94)03823-6, 1994.
- Matthews, J., Sinha, A., and Francisco, J. S.: The importance of weak absorption features in promoting tropospheric radical production, Proc. Natl. Acad. Sci. USA, 102, 7449–7452, doi:10.1073/PNAS.0502687102, 2005.
- Messaadia, L., Dib, G. E., Ferhati, A., and Chakir, A.: UV-visible spectra and gas-phase rate coefficients for the reaction of 2,3-pentanedione and 2,4-pentanedione with OH radicals, Chem. Phys. Lett., 626, 73–79, doi:10.1016/j.cplett.2015.02.032, 2015.
- Ravishankara, A. R., Solomon, S., Turnipseed, A. A., and Warren, R. F.: Atmospheric lifetimes of long-lived halogenated species, Science, 259, 194–199, doi:10.1126/SCIENCE.259.5092.194, 1993.
- Roehl, C. M., Nizkorodov, S. A., Zhang, H., Blake, G. A., and Wennberg, P. O.: Photodissociation of peroxynitric acid in the near-IR, J. Phys. Chem. A, 106, 3766–3772, doi:10.1021/JP013536V, 2002.
- Sander, S. P., Friedl, R. R., Golden, D. M., Kurylo, M. J., Moortgat, G. K., Keller-Rudek, H., Wine, P. H., Ravishankara, A. R., Kolb, C. E., Molina, M. J., Finlayson-Pitts, B. J., Huie, R. E., and Orkin, V. L.: Chemical Kinetics and Photochemical Data for Use in Atmospheric Studies, Evaluation Number 15, JPL Publication 06-2, Jet Propulsion Laboratory, Pasadena, CA, http://jpldataeval.jpl.nasa.gov, 2006.

- Sander, S. P., Abbatt, J., Barker, J. R., Burkholder, J. B., Friedl, R. R., Golden, D. M., Huie, R. E., Kolb, C. E., Kurylo, M. J., Moortgat, G. K., Orkin, V. L., and Wine, P. H.: Chemical Kinetics and Photochemical Data for Use in Atmospheric Studies, Evaluation No. 17, JPL Publication 10-6, Jet Propulsion Laboratory, Pasadena, http://jpldataeval.jpl.nasa.gov, 2011.
- Shemansky, D. E.: CO₂ extinction coefficient 1700–3000 Å, J. Chem. Phys., 56, 1582–1587, doi:10.1063/1.1677408, 1972.
- Turco, R. P.: Photodissociation rates in the atmosphere below 100 km, Geophys. Surv., 2, 153–192, doi:10.1007/BF01447907, 1975.
- Wallington, T. J., Ammann, M., Cox, R. A., Crowley, J. N., Herrmann, H., Jenkin, M. E., McNeill, V., Mellouki, A., Rossi, M. J., and Troe, J.: IUPAC Task group on atmospheric chemical kinetic data evaluation: Evaluated kinetic data, http://iupac.pole-ether.fr, 2018.
- Zhou, X., Davis, A. J., Kieber, D. J., Keene, W. C., Maben, J. R., Maring, H., Dahl, E. E., Izaguirre, M. A., Sander, R., and Smoydzin, L.: Photochemical production of hydroxyl radical and hydroperoxides in water extracts of nascent marine aerosols produced by bursting bubbles from Sargasso seawater, Geophys. Res. Lett., 35, L20803, doi:10.1029/2008GL035418, 2008.