

0.1 Final exam equation sheet

Layer energy equation: $\frac{dE}{dt} = I_{\downarrow} + I_{\uparrow}$ (1)

Solar constant: $S = \frac{S_0}{4}(1 - \alpha)$ (2)

Total grey body flux $I = \varepsilon \sigma T^4$ (3)

where $\sigma = 5.67 \times 10^{-8} \text{Wm}^{-2}\text{K}^{-4}$

transmissivity tr: $I_{\text{transmitted}} = \text{tr} I_0$ (4)

reflectivity α $I_{\text{reflected}} = \alpha I_0$ (5)

absorbtivity abs $I_{\text{absorbed}} = \text{abs} I_0$ (6)

Kirchoff's law $\varepsilon = \text{abs}$ (7)

CO_2 radiative forcing $\Delta F = (3.8 \text{Wm}^{-2}) \frac{\ln(\text{newp CO}_2 / \text{origp CO}_2)}{\ln(2)}$ (8)

Conservation of Energy: $\alpha I_0 + \text{abs} I_0 + \text{tr} I_0 = I_0$ (9)

moist static energy: $h_m = c_p T + l_v w_v + gz$ (10)

moist adiabatic lapse rate: $\Gamma = \frac{dT}{dz} = \frac{-g}{c_p + l_v \frac{dw_v}{dT}}$ (11)

hydrostatic balance: $dp = -\rho g dz$ (12)

mass in a layer in kg/m^2 : $M = \int_{z_1}^{z_2} \rho(z) dz$ (13)

energy in an ocean layer: $\Delta E = \rho_w D c_w \Delta T$ (14)

Conservation of energy for layer: $\frac{d\Delta E}{dt} = \Delta F$ (15)

change of temperature for an ocean layer: $\frac{d\Delta T}{dt} = \frac{\Delta F}{\rho_w c_w D}$ (16)

Planck feedback: $\frac{dI_G}{dT} = \frac{d(-\sigma T^4)}{dT} = f_{\text{planck}} = -4\sigma T^3 = -1/\lambda$ (17)

Conservation of energy with feedback: $\frac{\Delta E}{dt} = \Delta F - 4\sigma T^3 \Delta T$ (18)

Climate adjustment to abrupt forcing: $\Delta T(t) = \lambda \Delta F (1 - e^{-t/\tau})$ (19)

Climate adjustment timescale: $\tau = \rho_w c_w D \lambda$ (20)

Climate sensitivity: $\Delta T = \lambda \Delta F$ (21)

Climate mean temperature budget: $\rho_w c_w D \frac{dT}{dt} = \Delta F + \sum f_n \Delta T$ (22)

Climate feedback factor: $f_n = \frac{\Delta R}{\Delta T} = \left(\frac{\Delta R}{\Delta \text{climate}} \right) \left(\frac{\Delta \text{climate}}{\Delta T} \right)$ (23)

Climate sensitivity with feedbacks: $\lambda = -\frac{1}{\sum f_n}$ (24)

oxygen isotopes: $\delta^{18}\text{O} = \left[\frac{(^{18}\text{O}/^{16}\text{O})_{\text{sample}}}{(^{18}\text{O}/^{16}\text{O})_{\text{standard}}} - 1 \right] \times 1000$ (25)

Carbon Isotope Mass Balance: $(M_{\text{final}})(\delta_{13}C_{\text{final}})$
 $= (M_{\text{initial}})(\delta_{13}C_{\text{initial}})$
 $+ (M_{\text{added}})(\delta_{13}C_{\text{added}})$ (26)

Carbon Isotopic Excursion CIE: $\delta_{13}C_{\text{final}} - \delta_{13}C_{\text{initial}}$ (27)

$M_{\text{added}} = -\text{CIE} \times \frac{M_{\text{initial}}}{(\delta_{13}C_{\text{final}} - \delta_{13}C_{\text{added}})}$ (28)

Kaya identity: $I = P \times A \times EI \times CI$ (29)

0.2 Final exam constants

$\sigma = 5.67 \times 10^{-8} \text{Wm}^{-2}\text{K}^{-4}$ (30)

$c_p = 1004 \text{Jkg}^{-1} \text{K}^{-1}$ (31)

$c_w = 4186 \text{Jkg}^{-1} \text{K}^{-1}$ (32)

$\rho_w = 1000 \text{kgm}^{-3}$ (33)

$l_v = 2.5 \times 10^6 \text{Jkg}^{-1}$ (34)

1 ppm $\text{CO}_2 = 2.1 \text{Gtonnes C}$ (35)

$M_{\text{CO}_2} = 44 \text{kg kmole}^{-1}$ (36)

$M_{^{12}\text{C}} = 12 \text{kg kmole}^{-1}$ (37)