## PATENT SPECIFICATION

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Inventors: JOHN EDWARD DOWNES and DONALD WILLIAM RONALD HEADFORD

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## COMPLETE SPECIFICATION

## Bipyridylium Compounds and Herbicidal Compositions containing them

We, IMPERIAL CHEMICAL INDUSTRIES LIMITED, a British Company of Imperial Chemical House, Millbank, London, S.W.1, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:—

This invention relates to new bipyridylium quaternary salts, to processes for their preparation and to herbicidal compositions containing them.

U.K. Specification No. 813,532 discloses and claims the use as herbicides of compounds of the formula:—

where R and R<sup>1</sup> are alkyl radicals of not more than four carbon atoms which are substituted with a carbamyl or N-substituted carbamyl group; and X<sup>-</sup> is an anion.

We have now found, according to the present invention, that certain N-substituted carbamyl compounds of the formula shown above show selective herbicidal activity against broad-leafed species compared with their activity against grass species. The selective compounds are those of the formula:—

wherein the two radicals attached to the nitrogen atoms in the bipyridyl nucleus are identical and where  $R_1$  and  $R_2$  together with the nitrogen atom to which they are attached constitute a pyrrolidine, piperidine, hexamethyleneimine or morpholine ring optionally having a methyl radical as substituent; or where  $R_1$  and  $R_2$  are each separate, identical radicals and are alkoxyalkyl radicals of 2, 3 or 4 carbon atoms or cyanoalkyl radicals of 2 or 3 carbon atoms; and  $X^-$  is an anion. The anion  $X^-$  can be a monovalent ion, for example chloride, iodide, or bromide, or a divalent ion — where of course the symbol  $2X^-$  in the above structural formula represents a single ion rather than two monovalent ions. The choice of anion depends to a large extent upon the solubility of the respective salts in water and upon the ease with which the salts can be prepared.

The invention also consists in a process for the preparation of the compounds of the invention, in which 4,4¹-bipyridyl or a mono-quaternary salt of the formula:—

$$R_2$$
 N-co-cH<sub>2</sub>- N+ N  $X^{-1}$ 

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where R<sub>1</sub>, R<sub>2</sub> and X<sup>-</sup> have the meanings given above, is quaternised with a suitable quaternising agent. The process can be performed by simply heating a mixture of 4,41-bipyridyl (or the mono-quaternary salt as the case may be) with the quaternising agent. However, in some instances, especially where operating on a large scale, it may 5 be more convenient to use a suitable liquid as solvent or diluent for the reactants. 5 The invention includes herbicidal compositions containing as active ingredient a compound of the invention, and a carrier for the active ingredient. The composition can, for example, be a liquid composition obtained by dissolving or dispersing the active ingredient in a suitable liquid diluent, for example water or a suitable organic liquid. The herbicidal composition alternatively can be in the form of a powder in which 10 10 the active ingredient is in admixture with a powder diluent. The compositions can contain a wetting or dispersing agent to facilitate their use as spray compositions, or indeed can contain any other type of substance known to the art as being suitable as an adjuvant in herbicidal compositions, for instance stickers, binders, corrosion inhibitors 15 15 and colouring agents. Any wetting or dispersing agent used in the herbicidal compositions should preferably be a non-ionic surface-active compound, for instance an ethylene oxide-octyl phenol condensate, or a cationic compound, so as to avoid any undesirable interaction between the surface-active compound and the bipyridylium salt which might take place 20 if an anionic compound were to be used as surface-active compound. 20 The invention further consists in a method of killing, desiccating or otherwise severely damaging undesirable vegetation, in which the vegetation is treated with an effective amount of a compound or herbicidal composition of this invention. The invention is illustrated by the following Examples. 25 EXAMPLE 1 25 N,N-Di-\beta-ethoxyethyl chloroacetamide (29.5 g.) and 4,41-bipyridyl (8.7 g.) were heated and stirred together at 100°C, until a homogeneous liquid was obtained. The temperature of this liquid was then raised to 140°C, at which temperature a vigorous reaction took place and the mixture solidified. After heating the solidified mixture for 30 minutes, the solid was crushed, washed with acetone and recrystallised from 30 30 methanol/acetone, affording 1,1<sup>1</sup>-bis-(di-β-ethoxyethylcarbamyl-methyl) 4,4<sup>1</sup>-bipyridylium dichloride monohydrate. The melting point of the product was recorded as 244°C. EXAMPLES 2—12 A number of other substituted 4,41-bipyridylium dichlorides have been made by 35 35 the process described in Example 1 above, but using various N,N-di-substituted chloroacetamides or cyclic chloroacetamides instead of the N,N-di-\beta-ethoxyethyl chloroacetamide used in that Example. These other bipyridylium salts, which are set out in the table below as Examples 2-12, are all believed to have structures corresponding to the structural formula: -40 40 +N-CH2-CO-N 2CL N-CO-CH<sub>2</sub>-N+ and were obtained by a process in which there was used a chloroacetamide of the formula: CL-CH2-CO-N R. 45 In each instance the chloroacetamide and 4,41-bipyridyl were used in a molar ratio in 45 the range 1:1 to 1,1:1. The products of Examples 2-12 are identified in the table by the meaning given to the group

and the melting point shown against the products are those obtained using a sample

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of the product which had been purified by recrystallisation, melting in each instance

being accompanied by decomposition of the sample under test.

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The substances obtained as products of the Examples have been tested for herbicidal activity against both broad-leafed plant species and grasses, and the results are expressed in the right-hand columns of the table. Under the heading "Herbicidal activity" is shown the amounts (in lb./acre) of the various compounds, expressed in terms of their respective cations, found necessary to obtain a 50% kill of the species tested 14 days after application to the test plants. The broad-leafed and grass species (referred to in the table under the respective headings "B" and "G") were as follows:—

Broad-lea	fed species	Grass spec	ies	
Sugar beet Mustard Kale Clover Redshank Mayweed	(Beta vulgaris) (Sinapis alba) (Brassica oleracea) (Trifolium pratense) (Polygonum persicaria (Matricaria inodora)	Barley Cocksfoot	(Triticum vulgare) (Avena fatua) (Hordeum sativum) (Dactylis glomerata)	10

The term "50% kill" used above refers to the species treated having been damaged to a total extent of 50% compared with untreated control plants; it does not necessarily mean that 50% of the plants treated had been killed. The assessment of the herbicidal effect of the substances tested was made as follows:-

Batches of plants of each of the broad-leafed and grass species were sprayed with aqueous solutions of the test substances at a range of concentrations. The herbicidal damage caused to each batch was visually assessed 14 days after spraying as a percentage, taking undamaged control plants as 0% and completely dead plants as 100%. The mean values obtained against all the broad-leafed species and the grass species at each particular concentration was then calculated and these mean values were plotted against concentration on a logarithmic scale. From the resulting graph the concentration of each substance required to produce 50% kill was then read off.

The figures given in the extreme right-hand column of the table under "Index of Selectivity" are the ratios of the adjacent figures against broad-leafed and grass species and it will be seen that several of the compounds show a most marked selective action against broad-leafed species.

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f. f.			03	[5	30	300	84	32	102
Index of Selec- tivity			>420	>215	>30	₩ 			¥ 
	Herbicidal activity	ტ	>28	>6.5	>5.0	25	>24	1.6	6.0
	Herbicid	я	0.067	0.030	0.17	0.083	0.28	0.050	0.059
Products of Examples	m.p. (decomp.) (°C.)		244	258	320	301	284	295	302
	Water of	sation	H <sub>2</sub> 0	$2H_2O$	2H20	O°H	$2\mathrm{H}_2\mathrm{O}$	$\mathrm{H_2O}$	$O_2^{-}$
	N. N.	9. <sup>4</sup>	-N(CH <sub>2</sub> -CH <sub>2</sub> -0-C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>	-N(CH <sub>2</sub> -CH <sub>2</sub> -0-CH <sub>3</sub> ) <sub>2</sub>	-N(CH <sub>2</sub> -CN) <sub>2</sub>		F Z F	C <sub>Z</sub>	o o
	Hwowh	No.	,I	63	က	4	ιO	9	7

TABLE I—Continued

	<del> </del>	Ť		·		
Index	tivity	229	>221	>104	>24	>24
lal activity	ტ	6.1	>6.5	>8.0	8	<b>%</b>
	æ	0.027	0.029	0.077	0.33	0.33
m.p. (decomp.) (°C.)		303	318	298	282	271
Water of crystalli- sation		2H <sub>2</sub> O	2H <sub>2</sub> O	2H2O	$0_{2}$ H $_{2}$ O	Anhydrous
X. X.	€F.c		II		( )	-N(CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>
Example	No.	∞	6	10	Ξ	12
	Water of m.p. Herbicidal activity	Water of m.p. crystallication (decomp.)	Example No. R. Water of crystallismos, $({}^{\circ}C, {}^{\circ}C, {}^{\circ})$ Herbicidal activity crystallismos, $({}^{\circ}C, {}^{\circ}C, {}^{\circ})$ B G	Example No.  8 $C_{CH_3}$	Example         Crystallity crystallity activity         Water of crystallity (°C.)         m.p. (°C.)         Herbicidal activity           8         -N.         2HaO         303         0.027         6.1           9         -N.         2HaO         318         0.029         >6.5           10         -N.         2HaO         298         0.077         >8.0	Example       Carter of Carystalling activity       Water of Carystalling activity       Mater of Carystalling activity       Mater of Carystalling activity       Mater of Carystalling activity       Mater of Carystalling activity       B       G         8       -N       2H <sub>2</sub> O       303       0.027       6.1         9       -N       2H <sub>2</sub> O       318       0.027       6.1         10       -N       2H <sub>2</sub> O       318       0.029       >6.5         11       -N       3H <sub>2</sub> O       282       0.33       >8

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The compounds of the invention have shown anti-fungal activity in in vitro tests when applied as an aqueous solution to spores of wheat leaf rust (Puccinia recondita).

WHAT WE CLAIM IS:-

1. A 4,41-bipyridylium quaternary salt of the formula: —

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wherein the two radicals attached to the two nitrogen atoms in the bipyridyl nucleus are identical and where  $R_1$  and  $R_2$  together with the nitrogen atom to which they are attached constitute a pyrrolidine, piperidine, hexamethyleneimine or morpholine ring; or  $R_1$  and  $R_2$  are each separate, identical radicals and are alkoxyalkyl radicals of 2, 3 or 4 carbon atoms or cyanoalkyl radicals of 2 or 3 carbon atoms; and  $X^-$  is an anion.

2. A compound according to Claim 1, in which R<sub>1</sub> and R<sub>2</sub> together with the nitrogen atom to which they are attached constitute a piperidine or morpholine ring having a methyl radical as substituent.

3. A compound of the formula shown in Claim 1 above, in which the group

$$\begin{array}{c}
R_1 \\
R_2
\end{array}$$

is a piperidine ring.

4. A compound of the formula shown in Claim 1 above, in which the group

is a 2-methyl piperidine ring.

5. A compound of the formula shown in Claim 1 above, in which the group 20

is a 2,6-dimethyl piperidine ring.

6. A compound of the formula shown in Claim 1 above, in which the group

25 is a morpholine ring.

7. A compound of the formula shown in Claim 1 above, in which the group

is a 2,6-dimethyl morpholine ring.

8. A compound of the formula shown in Claim 1 above, in which the group



is a 3,5-dimethyl morpholine ring.

9. A compound of the formula shown in Claim 1 above, in which the group

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is a di(\beta-methoxyethyl)amino group.

10. A compound of the formula shown in Claim 1 above, in which the group



is a di(\beta-ethoxyethyl)amino group.

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11. A compound according to any of the preceding claims, in which the anion  $\mathbf{X}^-$  is a halogen ion.

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12. A process for the preparation of a compound of any of the preceding claims, in which 4,41-bipyridyl or a mono-quaternary salt of the formula:—

$$R_1$$
 $R_2$ 
 $N-CO-CH_2-N+$ 
 $N$ 
 $X$ 

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where R<sub>1</sub>, R<sub>2</sub> and X<sup>-</sup> have the meanings given to them in Claim 1 above, is treated with an appropriate quaternising agent.

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13. A 4,41-bipyridylium salt obtained by the process of Claim 13.

14. The process for the preparation of 1,1<sup>1</sup>-bis-(carbamylmethyl) 4,4<sup>1</sup>-bipyridylium quaternary salts substantially as described in Example 1.

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15. The product of any one of Examples 1—12.

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16. A herbicidal composition comprising as active ingredient a compound of any of Claims 1—11, 13 and 15, and a carrier for the said active ingredient.

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17. A method of killing, desiccating or severely damaging undesirable vegetation, in which the vegetation is treated with an effective amount of a compound of any of Claims 1—11, 13 and 15 or with a herbicidal composition of Claim 16.

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BERTRAM F. DREW, Agent for the Applicants.

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